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1           **Magnetic scaffolds of polycaprolactone with functionalized magnetite**  
2           **nanoparticles: Physicochemical, mechanical, and biological properties**  
3                           **effective for bone regeneration**

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**Abstract**

Magnetic scaffolds have gained significant attention for the disease treatment and tissue repair. Here we focus on magnetic nanocomposites scaffolds made of poly(caprolactone) (PCL) and magnetite nanoparticles (MNPs) for bone repair. The physico-chemical, mechanical, and magnetic properties of the scaffolds, the in vitro cell responses, and the in vivo tissue compatibility were examined in-depth to find the effectiveness for uses as bone scaffolds. The MNPs, produced by a surfactant-mediation process, were well-distributed within the PCL matrix to enable homogeneous nanocomposites. The PCL-MNP scaffolds showed excellent magnetic properties, preserving the superparamagnetic behavior. Incorporation of MNPs greatly improved the hydrophilicity and water swelling of scaffolds. Acellular apatite forming ability test revealed a higher mineral induction on the magnetic scaffolds than on the PCL scaffold. The mechanical stiffness increased significantly with the addition of MNPs, when tested under both static and dynamic compressed wet conditions. The initial cell adhesion to the magnetic scaffolds was substantially improved by ~1.4-fold with respect to the pure PCL scaffold, enabling earlier cellular proliferation confluence. The cell mineralization, as assessed by the quantification of calcium deposits, was significantly enhanced on the magnetic scaffolds. The magnetic scaffolds, subcutaneously implanted in rats for 2 weeks, revealed favorable tissue compatibility, with substantial fibroblastic cell invasion and neoblood vessel formation while exerting minimal inflammatory reactions. The results, demonstrating excellent physico-chemical, magnetic, mechanical and biological properties of the PCL-MNP scaffolds, support the potential use of the magnetic scaffolds for bone repair and regeneration.

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**Keywords:** Bone scaffolds; Magnetic nanoparticles, Nanocomposites; Bone cell responses; Bone repair

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## 56 1. Introduction

57

58 Magnetic nanoparticles (MNPs) have shown great promise in the biomedical fields due to their special  
59 physical properties, which include dimensions at the nanoscale and unique reaction to magnetic  
60 forces, providing favorable cell and tissue responses. Over the past decade, biomedical applications  
61 of MNPs have ranged from hyperthermia therapy to magnetic resonance imaging and even to  
62 magnetic drug delivery.<sup>1,2</sup> In the meanwhile, very few studies have been reported on the use of MNPs  
63 as a scaffold for tissue engineering. Scaffolds, as the key component in tissue engineering, aid in the  
64 adhesion and spreading of cells, support their growth and subsequently trigger their development to  
65 specific tissues. Therefore, engineering a scaffold with properties to improve these cellular processes  
66 is the key issue in scaffold development for tissue engineering.

67 Recently, MNPs have been introduced as the nanocomponent that can be incorporated  
68 within polymeric scaffolds to provide additional magnetic properties to the scaffolds.<sup>3-6</sup> The  
69 incorporation of inorganic nanoparticles including MNPs is considered a promising strategy to  
70 produce biopolymer-based bone scaffolds with properties more suitable for bone repair and  
71 regeneration, in terms of mechanical and biological aspects.<sup>7-9</sup> In particular, the MNPs incorporated in  
72 the scaffolds are considered to play a number of important roles in the stimulation and alteration of  
73 cellular responses which are favorable for bone formation and disease treatments. Under magnetic  
74 fields, the magneto-mechanical induction of bone cells or the temperature-induced hyperthermia  
75 therapy of cancerous cells are the possible reasons that explain the usefulness of the scaffolds.<sup>7,8</sup>

76 To this end, here we develop MNPs-incorporated magnetic biopolymer scaffolds for the  
77 purpose of bone repair. The surface-functionalized MNPs were added at small quantities to  
78 polycaprolactone (PCL) and a salt-leaching method was introduced to produce foam scaffolds. The  
79 physico-chemical, mechanical, and magnetic properties of the PCL-MNP scaffolds were investigated,  
80 and the bone cell responses as well as the tissue compatibility in rats were assessed to find further  
81 usefulness for bone tissue engineering.

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## 85 2. Experimental procedures

86

### 87 2.1. Preparation of MNPs and PCL-MNP scaffolds

88 MNPs were synthesized using a method developed by Sun et al.<sup>9</sup> Briefly, MNPs (10 nm in diameter)  
89 were prepared by mixing 3.5318 g Fe(acac)<sub>3</sub> (iron(III) acetylacetonate), 3.9123 g 1,2-hexadecanediol,  
90 10 ml of oleic acid, 10 ml of oleylamine, and 40 ml of benzyl ether under a flow of nitrogen gas. The  
91 mixture was preheated to reflux at 200 °C for 30 min while stirring, and then heated to 300 °C for an  
92 additional 2 h under a nitrogen atmosphere. The black-brown mixture was allowed to cool to room  
93 temperature and 50 ml of ethanol was added to the precipitate. The products were collected by  
94 centrifugation at 10,000 rpm for 5 min and then washed 4 times with ethanol and dried at 50 °C. The  
95 morphology and magnetic properties of the MNPs were characterized using transmission electron  
96 microscopy (TEM, 7100 microscope, JEOL, USA) and a vibrating sample magnetometer (VSM,  
97 Quantum Design MPMS-XL7, USA).

98 For the preparation of PCL-MNP scaffolds, 10% w/v of PCL (~ 80 kDa, Sigma-Aldrich, USA)  
99 was first dissolved in chloroform, and the MNPs were then added to the PCL solution. The  
100 concentrations of MNPs in PCL solutions were prepared at 0, 5, and 10 wt%, which designated as the  
101 PCL, PCL-MNP5, and PCL-MNP10, respectively. The mixture solutions were ultrasonicated to ensure  
102 they were homogeneous and stable. The NaCl particles sieved (200-500 µm in diameter) were  
103 poured into a cylindrical plastic mould and then packed tightly by a manual pressing, after which the  
104 mixture solutions were poured dropwise into the NaCl-filled mold, and then followed by freezing at -70  
105 °C and freeze-drying for 3 days. The resulting samples were washed with distilled water for 10 min (x  
106 9) while agitating at 100 rpm to leach out the salt completely, and then dried again.

107

### 108 2.2. Characterizations of PCL-MNP scaffolds

109 Morphologies of the scaffolds were observed by scanning electron microscopy (SEM; S-3000H  
110 Hitachi, Japan), and the atomic composition was analyzed with energy dispersive spectroscopy (EDS).  
111 The phase of the scaffolds was characterized by X-ray diffraction (XRD; Rigaku, USA). The scaffolds  
112 were scanned in the range of diffraction angle  $2\theta = 10-60^\circ$  at a rate of  $2^\circ \text{ min}^{-1}$  with a step width of  
113  $0.02^\circ$ ,  $2\theta$  using Cu K $\alpha$ 1 radiation at 40 kV and 40 mA current strength. Fourier transformed infrared

114 (FT-IR; Perkin-Elmer, USA) spectroscopy was used to observe the chemical status of the scaffolds.  
115 Thermogravimetric analysis (TGA) was carried out to analyze the thermal behavior and compositional  
116 fraction of the scaffolds. The samples were heated from room temperature to 500 °C at a heating ratio  
117 of 10 °C/min under nitrogen atmosphere.

118 The capacity of the scaffolds to take up water<sup>10</sup> or ethanol<sup>11</sup> was measured by the weight  
119 change before and after the soaking tests, as follows:  $\Delta W_s (\%) = ((W_s - W_o) / W_o) \times 100$ , where  $W_o$  and  
120  $W_s$  are the scaffold weights before and after the soaking, respectively. The porosity and density of the  
121 scaffolds were measured using a mercury porosimeter (PM33, Quantachrome, USA). The specific  
122 surface area was analyzed by the Brunauer-Emmett-Teller (BET) method under nitrogen gas. The  
123 hydrophilicity of the scaffolds was investigated by measuring the water contact angle using a  
124 Phoenix300 analyzer. Water droplet images made on the scaffold surface were obtained using a  
125 viewing system until equilibrium was reached at 25 °C. Typical images of the water droplet at an  
126 equilibrium state were taken for each sample, and five samples were tested for each group.

127 The apatite forming ability of the scaffolds was investigated immersing in a concentrated  
128 SBF (2 x SBF), which was used to speed up the apatite induction process and thus to shorten the  
129 evaluation periods of the apatite forming ability of bioactive materials.<sup>12</sup> Each sample (5 mm in  
130 diameter and 3 mm in thickness) was contained in 10 ml of 2 x SBF and then incubated at 37 °C for  
131 the given periods (1, 3, and 7 days). At each time, samples were collected, washed with distilled  
132 water, and dried at room temperature. Their apatite-forming abilities were analyzed using SEM and  
133 XRD.

134 The magnetic properties of the samples were measured by a VSM in an applied magnetic  
135 field of  $\pm 20$  kOe at room temperature<sup>13</sup>, in terms of saturation magnetization and hysteresis loops.  
136 VSM was calibrated using a standard reference (high purity nickel sphere), supplied with the  
137 instrument.

138

### 139 **2.3. Mechanical evaluation of scaffolds**

140 The mechanical properties of the PCL-MNP scaffolds were measured by a dynamic mechanical  
141 analysis (DMA, DMA25N, MetraVib, USA) under static and dynamic compression in wet conditions,  
142 as reported elsewhere.<sup>14</sup> A cylindrical sample (8 mm in diameter and 16 mm in height) was first  
143 soaked completely with PBS for 1 day. For a static test, samples were measured at a constant

144 compressive load while recording the strain with respect to time. Three different samples were tested  
145 at each condition, and obtained values were averaged.

146 Next, dynamic was performed on a parallel plate configuration. Mechanical spectrometry was  
147 monitored using dynamic frequency sweep with frequencies ranging from 0.5 to 10 Hz for 10 min at  
148 room temperature. The storage modulus ( $E'$ ) and loss modulus ( $E''$ ) were recorded. The tangent delta  
149 was evaluated from the ratio of  $E''/E'$ .

150

#### 151 **2.4. *In vitro* cell adhesion, growth and mineralization tests**

152 MC3T3-E1 cells (ATCC; American type culture collection) were cultured in  $\alpha$ -Minimal Essential  
153 Medium ( $\alpha$ -MEM, Gibco, USA) supplemented with 10% of fetal bovine serum (FBS, Hyclone, Thermo  
154 Scientific, USA), 100 U/mL of penicillin, and 100 $\mu$ l of streptomycin at 37 °C in a humidified atmosphere  
155 containing 5% of CO<sub>2</sub> for 7 days. 10<sup>5</sup> cells were seeded onto the scaffold (5 mm in diameter x 3 mm in  
156 height). After 1 day, the scaffold moved to each well of another plate, and the cell adhesion rate on the  
157 scaffold was measured by a cell counting kit (CCK). The cell proliferation was further assessed by the  
158 CCK method, for the culture periods of 3, 7, 14, and 21 days. For SEM observation of cells, samples  
159 were fixed in a solution of 2.5% glutaraldehyde, dehydrated in a gradient series of ethanol solutions,  
160 dried at room temperature, and coated with a thin layer of platinum.

161 The cellular mineralization was assessed by the Alizarin red assay (ARS; Sigma Aldrich,  
162 USA). After culturing for 14, 21 and 28 days, the cells were fixed with 70% ethanol for 1 h at 4 °C, and  
163 then immersed in 2% w/v of aqueous ARS solution (pH 4.1~4.3) for 30 min at room temperature. After  
164 several washes with distilled water, these stained samples were removed and eluted with 10% w/v of  
165 cetyl pyridinium chloride (CPC) in 10 mM sodium phosphate (pH 7) for 1 h. The absorbance of eluents  
166 was then read using a micro-plate reader at 595 nm after normalization with the total amount of  
167 mitochondrial dehydrogenase in the cells for the consistent quantitative assay of each sample solution.

168

#### 169 **2.5. *In vivo* tissue compatibility test**

170 The *in vivo* experiments were approved by Dankook University Institutional Animal Care and Use  
171 Committee, Korea. Three ten-week-old male Sprague-Dawley rats weighing 250-350 g were used.  
172 Experimental scaffold groups were PCL, PCL-MNP5, and PCL-MNP10, and four samples (5 mm in

173 diameter x 3 mm in height) per group were implanted. Prior to implantation, the scaffolds were  
174 sterilized with ethylene oxide gas. The animals were anesthetized with an intramuscular injection of  
175 80 mg/kg ketamine and 10 mg/kg xylazine of body weight. The skin on the dorsal region of the rat  
176 was shaved and asepsis of the operatory area was achieved with povidone and 70% ethanol. A 2 cm  
177 long linear incision was made in the skin using a # 10 blade mounted onto a bard-parker scalpel. Four  
178 small subcutaneous implant sites were made by blunt dissection with Halsted-mosquito hemostatic  
179 forceps on the back side in a lateral direction from the spine of each rat. Scaffolds were inserted into  
180 the prepared area, away from the incision point. The incision was subsequently sutured with 4-0 non-  
181 absorbable monofilament suture material (Dafilon®, B. Braun, Germany). During and after surgery,  
182 the rats were kept warm under observation until recovered from the anesthesia, after which the rats  
183 were housed, one rat per cage. The animals were kept on a 12 h light / 12 h dark schedule in the  
184 cages and provided with standard pellet food and water *ad libitum*.

185 Two weeks after the implantation, the animals were sacrificed by cervical dislocation. The  
186 tissue samples harvested for histologic analysis were immediately immersed in 4% buffered  
187 formaldehyde for 24 h at room temperature, and dehydrated in a series of graded ethanol. The  
188 specimens were bisected and embedded in paraffin. Paraffin blocks were serially sectioned at 5  $\mu$ m  
189 thicknesses along the longitudinal axis using a rotary microtome. The slides were classically stained  
190 with hematoxylin and eosin (HE) or Masson's trichrome (MT) stain, and were then observed with a  
191 light microscope for biocompatibility and vessel formation. Histological scores, given the points  
192 indexing absent (1), mild (2), moderate (3), and severe (4) degrees, obtained from both stained slides  
193 included the extent of inflammatory response, thickness of fibrous capsule, presence of blood vessel,  
194 and proliferation of fibroblasts.

195

## 196 **2.6. Statistical analysis**

197 Data were presented as the mean  $\pm$  one standard deviation. The statistical analysis was  
198 performed using a one-way ANOVA test by comparing between independent sample groups.  
199 Significance was considered at  $p < 0.05$  or  $p < 0.01$ .

200



## 201 **3. Results and discussion**

202

### 203 **3.1. Characteristics of PCL-MNP scaffolds**

204 As the magnetic nanocomponent, we prepared surface-functionalized magnetite ( $\text{Fe}_3\text{O}_4$ )  
205 nanoparticles. Magnetite is well-known as a class of magnetic iron oxide materials that has a cubic  
206 crystal form<sup>15</sup> and the nanoparticle form tends to agglomerate in the aqueous environment owing to  
207 the short-range van der Waals forces.<sup>16</sup> Therefore, the improvement of dispersibility in aqueous media  
208 has been one of the key issues in the development of MNPs. Our surface-functionalized MNPs were  
209 prepared by the reaction of  $\text{Fe}(\text{acac})_3$  with surfactants such as alcohols, oleic acids, and oleylamine at  
210 high temperature.<sup>17</sup> From a typical TEM micrograph (**Fig. 1a**), spherical and well-dispersed MNPs  
211 were observed with a narrow size distribution (an average size of 10.7 nm). The magnetic hysteresis  
212 curve of the as-synthesized MNPs measured by VSM at room temperature is shown in **Fig. 1b**. The  
213 MNPs have a saturation magnetization of  $72.1 \text{ emu g}^{-1}$ . The VSM curve passed through the zero point,  
214 and remnant magnetization and coercivity were not observed in the VSM curve, indicating that the  
215 superparamagnetism of the MNPs was preserved at room temperature. This is very consistent with  
216 the previous results.<sup>9, 18, 19</sup>

217 The surface-functionalized MNPs were then incorporated within the PCL scaffolds at  
218 different contents. The surface-functionalized MNPs were well dispersed in the chloroform solution,  
219 which was used to dissolve PCL. The porous scaffold of PCL-MNP scaffolds was then achieved via a  
220 salt leaching technique. The pore structure properties such as pore size, pore distribution, and  
221 porosity can affect the physicochemical properties of the porous scaffolds and the cellular  
222 behaviors.<sup>20-22</sup> **Fig. 2** shows SEM micrographs and EDS mapping images of the scaffolds on a cross-  
223 sectional view. All the scaffolds showed a well-developed pore structure without significant difference  
224 in the pore morphology. The macropores larger than  $250 \mu\text{m}$  are known to be suitable for cell  
225 penetration and engraftment whilst those smaller than  $100 \mu\text{m}$  restrict cellular infiltration within the  
226 pores.<sup>23</sup> According to the reports on the effects of pore size on osteoblast activity, pores larger than  
227  $300 \mu\text{m}$  were preferred for the induction of osteogenesis.<sup>23</sup> Therefore, the pores ranging from 250 to  
228  $500 \mu\text{m}$  implemented in this study will be suitable for bone tissue engineering applications. The EDS  
229 mapping revealed higher Fe signals in the MNP-incorporated scaffolds, and the signal distribution was  
230 observed to be uniform.

231 The porosity of the scaffolds, first estimated using the ethanol replacement test, was shown  
232 to be 65~70 %, which was in similar range to the porosity measured by the mercury intrusion  
233 porosimetry. Other properties such as bulk density and skeletal density could also be obtained by this  
234 method,<sup>24, 25</sup> as summarized in **Table 1**. While the porosity was similar among the scaffolds, the  
235 density level increased with increasing MNP content, which was due to the higher density of inorganic  
236 MNPs than PCL. The surface area of the scaffolds, measured by BET analysis, was shown to  
237 increase with increasing MNPs. Of note, the surface area of scaffolds was not dependent on porosity;  
238 rather, the value increased with increasing the MNPs content, which might be interpreted that the  
239 MNPs evenly embedded within the PCL polymeric matrix, with their nano-sized characteristic, should  
240 improve the surface area of the scaffolds.

241 The physico-chemical properties of the PCL-MNP scaffolds were further confirmed. The  
242 phase of the scaffolds was examined by XRD analysis (**Fig. 3a**). The MNPs showed diffraction peaks  
243 at  $2\theta \approx 31^\circ, 36^\circ, 43^\circ, 54^\circ, 57^\circ,$  and  $63^\circ$ , typical of bulk magnetite  $\text{Fe}_3\text{O}_4$ .<sup>26</sup> The average particle size,  
244 as calculated by Scherrer equation<sup>27</sup> for the strongest diffraction peak (311), was  $10.7 \pm 0.019$  nm,  
245 close to the size determined by TEM image. The magnetite peaks of the PCL-MNP scaffolds were  
246 more clearly observed with an increase in MNPs. The chemical bond structure of the scaffolds, as  
247 revealed by the FT-IR spectrum (**Fig. 3b**), showed typical bands related to PCL and MNPs, including  
248 a distinctive band at  $578 \text{ cm}^{-1}$  assigned to the Fe-O bond vibration of MNPs.<sup>28, 29</sup> This band became  
249 sharper in the PCL-MNP scaffolds with an increase in MNP content. Typical PCL vibration bands of  
250 C=O and C-O stretching were observed at  $1720$  and  $1293 \text{ cm}^{-1}$ , respectively. In addition, a weak and  
251 broad O-H stretching band of PCL was assigned to alcohol groups at  $3153\text{-}3640 \text{ cm}^{-1}$ . The thermal  
252 behavior of the scaffolds was monitored by TGA (**Fig. 3c**). The TGA curve of pure MNP showed a  
253 certain level of weight loss ( $\sim 10\%$ ), which was presumably due to the residual organic phases. While  
254 a typical thermal decomposition of the pure PCL was shown at almost  $400^\circ\text{C}$  (99% loss), the PCL-  
255 MNP scaffolds left certain levels of weight, which was ascribed to the presence of the MNP  
256 component in the scaffolds. Taking the weight losses of pure MNP and PCL also into consideration,  
257 the contents of MNPs within the PCL-MNP5 and PCL-MNP10 scaffolds were approximately 4.71 and  
258 10.01 wt.%, respectively, which was nearly consistent with the contents of MNPs incorporated in the  
259 preparation of the scaffolds.

260

### 261 3.2. Water affinity, swelling, and apatite forming ability of scaffolds

262 Water affinity results of the scaffolds are shown in **Fig. 4**. While the pure PCL scaffold showed a high  
263 degree of hydrophobicity (contact angle of as high as  $85^\circ$ ), the PCL-MNP scaffolds became more  
264 hydrophilic ( $61^\circ$  for 5% and  $47^\circ$  for 10% MNP) as shown in **Fig. 4a**. This was due to the existence of  
265 MNPs, more specifically, the carboxyl groups present on the surface of MNPs. In fact, during the salt  
266 leaching process, the surface of MNPs is possibly carboxylated. Jadhav et al. suggested that oleic  
267 acid, used as a surfactant to cap MNPs, electrostatically interacts with salt ions, resulting in  
268 dissociation of the COOH groups of oleic acid into  $\text{COO}^-$  and  $\text{H}^+$ , and consequently improving the  
269 hydrophilicity of MNPs.<sup>30</sup> We also found that the water dispersibility of the surfactant-capped MNPs  
270 markedly increased after the treatment with sodium chloride solution (data not shown).

271 As a result of this enhanced hydrophilicity, the PCL-MNP scaffolds showed an excellent  
272 water uptake capacity. The water uptake, measured for a period of up to 24 h, showed a significant  
273 difference between samples (**Fig. 4b**). The water uptake of PCL-MNP scaffolds quickly occurred  
274 within a few hours to reach almost saturation levels. The water uptake increased with increasing MNP  
275 content. As a result, after 24 h, the water uptake was recorded as ~1440% for PCL, 1870% for PCL-  
276 MNP5, and 2850% for PCL-MNP10 scaffolds. The water uptake capacity of the PCL-MNP10 scaffold  
277 was almost two-fold higher than that of the pure PCL scaffold. Furthermore, the scaffolds were  
278 observed to swell apparently after the water uptake. The increased volume of scaffolds, as optically  
279 measured (**Fig. 4c**), was in the order: 5 % in PCL < 8 % in PCL-MNP5 < 11 % PCL-MNP10, signifying  
280 a two-fold increase with 10% MNP addition to PCL. In fact, the water uptake level was due to both the  
281 pore-filling of water and the swelling of scaffolds. When considering the similar porosity levels for all  
282 scaffolds, the difference in water uptake capacity was primarily a result of the swollen (dimension-  
283 changed) property, i.e., swelling a scaffold through taking-up water molecules within the MNP-  
284 dispersed hydrophilic polymeric network should reflect the substantially increased water uptake  
285 behavior of the magnetic scaffolds.

286 The *in vitro* acellular mineralization behavior of PCL-MNP scaffolds was evaluated after  
287 immersion in SBF. **Fig. 5** shows the SEM morphologies and XRD patterns of the scaffolds after the  
288 SBF-immersion. The PCL-MNP5 and PCL-MNP10 scaffolds began to form mineral nanocrystallites on  
289 the surface as early as day 1, and the mineral phase covered the entire surface at day 3. After 7 days,  
290 the mineral crystallites had grown substantially (**Fig. 5a**). On the other hand, the pure PCL scaffold

291 started to show mineral formation at day 3, and then surface coverage at day 7 with much smaller  
292 crystals than those observed in the PCL-MNP scaffolds. The XRD patterns strongly supported the  
293 SEM results based on the characteristic peaks of HA crystal ( $2\theta \approx 26^\circ$  and  $32^\circ$ ) (**Fig. 5b**). It was thus  
294 clear that the MNP-incorporation enhanced the mineralization behavior of the scaffolds in SBF. This is  
295 due primarily to the surface-carboxylated MNPs distributed within the scaffolds. The calcium ions in  
296 the medium would be better attracted to the negatively-charged scaffold surface, and subsequently  
297 attract the phosphate ions to form mineral nuclei for crystallization.<sup>31, 32</sup> While the information on this  
298 behavior in SBF is somewhat limited because no cells were engaged in and the condition was not  
299 dynamic, the acellular mineralization results enable forecasting of the possible surface reactions and  
300 bone bioactivity of the magnetic scaffolds useful for bone regeneration purpose. As the biological  
301 properties of the scaffolds we also assessed the cell and tissue compatibility.

302

### 303 **3.3. Magnetic and mechanical properties of scaffolds**

304 MNPs produce a magnetic field surrounding the tissue following implantation, providing the matrix  
305 scaffolds physical conditions possibly beneficial for the native cell and tissue microenvironments.<sup>33</sup>  
306 For instance, the recent idea to utilize magnetic scaffolds in tissue engineering has been suggested  
307 for an alternative mediation of the angiogenesis process *in vivo*.<sup>34</sup> Here we examined the  
308 magnetization of the produced PCL-MNP magnetic scaffolds. The magnetic force of the PCL-MNP  
309 scaffolds will be dominated by the MNPs dispersed in the matrix. Magnetic properties of dry PCL-  
310 MNP scaffolds at room temperature were characterized from their typical hysteresis curves measured  
311 by VSM. **Fig. 6a** shows the magnetization *versus* magnetic field curves of PCL-MNP scaffolds. The  
312 hysteresis loops of PCL-MNP scaffolds showed a similar tendency to that of MNPs, and the  
313 magnetization value of PCL-MNP scaffolds was linearly proportionate to the MNP content (1.6 and 3.1  
314  $\text{emu g}^{-1}$ , respectively, with 5% and 10% MNP content). Similarly to the hysteresis loop of pure MNPs,  
315 coercivity and remanence were not observed in the curves of the PCL-MNP5 and PCL-MNP10  
316 scaffolds, which indicates the superparamagnetism of the scaffolds.<sup>35</sup> Field-dependent magnetic  
317 moments were nearly linear up to 0.5 kOe, reaching around 80% of the saturation magnetization  
318 value, which is obtained when materials can maximally reach the magnetization under the sufficient  
319 magnetic field.<sup>36</sup> Therefore, maximal magnetizations could be easily achieved in the magnetic  
320 scaffolds with only a small change in the external magnetic field. **Fig. 6b** shows the magnetic

321 interaction of the scaffolds by measuring distances between a permanent magnet and a magnetized  
322 scaffold.<sup>37</sup> The distance for pure PCL scaffolds was zero, while distances for magnetized PCL-MNP5  
323 and PCL-MNP10 scaffolds were 4.43 and 7.23 mm, respectively, which indicates the external  
324 magnetic field-guided distances of the magnetized scaffolds. Additive images show the magnet-  
325 attached scaffolds. Based on the magnetic properties, the PCL-MNP scaffolds were considered to  
326 preserve the superparamagnetic behavior of incorporated MNPs, although the saturation  
327 magnetization in the magnetic scaffolds was relatively small due to the small concentrations of MNPs.  
328 Further studies warrant the elucidation of the *in vivo* efficacy of the magnetic scaffolds in stimulating  
329 tissue repair or disease treatment through the *in situ* generated magnetism effects.

330         The mechanical behaviors of the PCL-MNP scaffolds were also examined under static and  
331 dynamic conditions using wet samples, as shown in **Figs. 7** and **8**, respectively. The mechanical  
332 properties are another important consideration of scaffolds targeting for bone repair and regeneration.  
333 **Fig. 7a** shows the typical stress-strain curves of the scaffolds under a static compressive load. All  
334 three scaffolds exhibit similar behavior. The stress value continues to increase with increasing strain,  
335 and the stress increasing rate increases during compression, which is generally observed in the  
336 porous materials in the course of densification and pore collapse under a compressive load. The  
337 incorporation of MNPs in the scaffolds recorded increased stress levels over the entire strain range,  
338 indicating higher resistance to deformation under a compressive load. The inset shows an initial linear  
339 region of stress-strain curves (within 2% strain), and the elastic modulus was obtained from the initial  
340 linear slope (**Fig. 7b**). The elastic modulus of PCL, PCL-MNP5, and PCL-MNP10 was 1.2, 1.4, and  
341 2.4 MPa, respectively, demonstrating that the MNPs distributed in the matrix played a significant role  
342 in stiffening the scaffolds.

343         Along with the static mechanical test, a dynamic mechanical analysis was further over a  
344 frequency range from 1 to 10 Hz performed under a constant strain.<sup>38</sup> The storage modulus ( $E'$ ;  
345 indicating the material elastic response to stress), loss modulus ( $E''$ ; indicating the material viscous  
346 response to stress),<sup>39</sup> and tangent delta ( $E''/E'$ ) were recorded, as shown in **Fig. 8**. The scaffolds  
347 exhibited little frequency-dependence for all the properties. The  $E'$  values were much higher than the  
348  $E''$  values, by four orders of magnitude, indicating that the PCL-MNP scaffolds have the ability to store  
349 elastic energy without a significant loss of polymeric viscosity. Importantly, the increase in MNP  
350 content significantly increased the storage modulus of the scaffolds, a coherent result of the static test.

351 There was little difference in the tangent delta among the scaffolds, as both storage and loss moduli  
352 increased similarly with MNPs incorporation. It is considered that the improved hydrophilicity and thus  
353 higher swelling of scaffolds, due to the MNP-incorporation, results in such change in modulus values.  
354 Although the MNPs themselves would render the PCL polymer network much stiffer, the increased  
355 volume (thus distance between polymer chains) in water and the interspaced water molecules should  
356 compensate the rigidity.

357

### 358 **3.4. *In vitro* osteoblastic responses to magnetic scaffolds**

359 The *in vitro* cell responses to the magnetic scaffolds were first assessed by the cell adhesion.  
360 **Fig. 9a** shows SEM micrographs of cell-attached PCL and PCL-MNP scaffolds performed at 1, 3, and  
361 24 h after cell seeding. Granule-like MC3T3-E1 cells were well revealed on the PCL-MNP5 and PCL-  
362 MNP10 scaffolds at 3 h after cell seeding, and the cells were found more abundantly in those  
363 nanocomposite scaffolds than those on the pure PCL scaffold. In order to quantify the cell adhesion  
364 level, CCK measurement was conducted, as shown in **Fig. 9b**. There was no significant difference of  
365 time-dependent cell adhesion rate in the pure PCL scaffold. On the contrary, in the PCL-MNP5 and  
366 PCL-MNP10 scaffolds, cell adhesion rapidly increased at 3 h after cell seeding and thereafter slightly  
367 increased up to 24 h. At 3 h, the cell adhesion level on the PCL-MNP10 scaffold was approximately  
368 1.4-fold higher than that on pure PCL scaffold.

369 The cell morphology, proliferation, and differentiation *in vitro* were further investigated. **Fig.**  
370 **10a** shows SEM micrographs of cells grown on the scaffolds for 14 days. All the scaffolds were highly  
371 populated with cells showing a number of cells and cellular products revealed throughout the scaffolds.  
372 There appeared to be little difference in the cell grown morphologies among the scaffolds. The  
373 proliferation of cells cultured on the scaffolds was then monitored for up to 21 days, as shown in **Fig.**  
374 **10b**. Cells grew rapidly up to 14 days for all scaffolds. Long-term cultures of cells over 14 to 21 days  
375 did not increase the cellular population, which was considered to be related to the confluence of cells  
376 on the scaffold surface and/or to the switch of the cellular proliferative potential dominantly into a  
377 differentiation. Interestingly, the cell proliferation rate, particularly after 7 days, was higher on the pure  
378 PCL scaffold than on the PCL-MNP scaffolds, with significant differences noticed at days 14 and 21  
379 (1.3~1.5-fold difference). It is thus considered that the cell proliferation rate from day 7 to 14 was  
380 higher on pure PCL. It is presumed that the cells on the PCL-MNP scaffolds might experience more

381 rapid proliferation-to-differentiation switch to undergo more active osteogenic differentiation processes.

382 As the final stage of osteogenic differentiation, the cellular mineralization is always  
383 considered and is of special importance. We analyzed the cellular mineralization behavior on the  
384 scaffolds by means of quantification of calcium deposits. For this, the staining of ARS that selectively  
385 binds to calcium was performed.<sup>40</sup> **Fig. 11a** shows the ARS quantified calcium deposit level on the  
386 cells cultured on each scaffold. While little difference was shown among scaffolds at days 7 and 14, a  
387 significant difference was noticed at day 21, which was a relatively prolonged culture period. The  
388 calcium level on the PCL-MNP10 scaffold was approximately 2.8-fold higher than that on the pure  
389 PCL scaffold. The mineral deposits in the PCL-MNP10 scaffold at day 21 were analyzed by EDS  
390 (**Figs. 11b**). The EDS mapping revealed high signals of Ca (in green) and P (in blue) with a Ca/P ratio  
391 of 1.7, being similar to that of stoichiometric HA. From these results, it is also evident that the cell  
392 mineralization was significantly enhanced on the magnetic scaffolds, indicating that the surface-  
393 functionalized MNPs in the scaffolds helped cellular osteogenesis and the final stage of mineralization.  
394 Although here we assessed the final stage of osteogenesis, more in-depth investigation into  
395 osteogenic behaviors including osteogenic gene expressions and protein syntheses will be warranted  
396 as further studies to elucidate the improved cellular mineralization. It is considered that the cells  
397 entering into an osteogenic differentiation could produce sufficient levels of bone matrix proteins,  
398 which are critically involved in subsequent cellular mineralization.

399 At this point, the reason for the improvement of cellular proliferation and osteogenic differentiation by  
400 the magnetic scaffolds needs to be discussed. The significant improvement in cell adhesion on the  
401 PCL-MNP magnetic scaffolds may first be attributed in part to the hydrophilic nature of the scaffolds  
402 that improves the affinity of proteins and cells.<sup>41</sup> It has frequently been reported that the hydrophilic  
403 modification of the PCL surfaces enhanced the early cell adhesion.<sup>42,43</sup> This improved initial cell  
404 adhesion event will affect subsequent cell proliferation, differentiation, and matrix production for  
405 cellular mineralization. Along with the improved hydrophilicity, the magnetism-related stimulation of  
406 cell behaviors should also be importantly considered. Several studies have also reported the  
407 influential role of MNPs incorporated within biomaterials and scaffolds in the cell proliferation and  
408 osteoblastic differentiation *in vitro*.<sup>44,45</sup> The MNPs were suggested to be a sort of single magnetic  
409 domain on the nanoscale, leading to significant alterations in ion channels of cell membrane that  
410 might be influential on the cell proliferation and differentiation behaviors.<sup>46,47</sup> The nanoscale-generated

411 magnetism-effect can be strengthened with increasing the content of MNPs, which would have  
412 stronger effects on the *in vitro* outcomes.

413

### 414 **3.5. *In vivo* tissue compatibility of magnetic scaffolds**

415 *In vivo* tissue responses of the scaffolds were evaluated after subcutaneous implantation in the dorsal  
416 region of rats for 2 weeks, as a preliminary study on the tissue compatibility of the magnetic scaffolds.

417 The recovery from the anesthesia was uneventful after surgery, and all animals showed normal  
418 healing process without material-related complications and remained in good health during the study

419 period. At 2 weeks after surgery, the harvested samples surrounding the tissues did not show any  
420 macroscopic redness or inflammatory signs. The histological photographs of samples are shown in

421 **Fig. 12**, and the microscopic tissue responses including the intensity of fibrous capsule formation,  
422 inflammatory response, capillary formation, and a granulation tissue rich in fibroblast, are scored in

423 **Table 2**. No significant differences were observed between the three scaffold groups. Mild to  
424 moderate fibroblastic and angioblastic proliferation was observed in all the samples. Fibrous tissue

425 encapsulation was shown in the scaffolds, which were minimally degraded. All of the scaffolds were  
426 filled with fibroblasts. While mild inflammatory response occurred in only a small part around the

427 samples, overall no rejection reactions were found after implantation in the animals. The newly  
428 formed tissue showed good integration to the adjacent tissues in all the groups. Organized fibrous

429 granulation tissue formed spaces between the scaffold and adjacent connective tissue and muscle.  
430 Fibroblasts existed within the adjacent fibrous capsules around these scaffold systems.

431 Collectively, the PCL-MNP magnetic scaffolds showed excellent tissue compatibility in rat  
432 subcutaneous model for 2 weeks, and the information delivers a minimal guideline of the possible use

433 of the developed scaffolds for further biomedical applications. Therefore, more in-depth *in vivo* studies  
434 using bone regeneration models for longer implantation periods are needed to elucidate the efficacy

435 of the magnetic scaffolds. While this issue remains as further study, the results on *in vitro* cellular  
436 responses to the scaffolds in the initial adhesion, proliferation and mineralization, as well as the

437 favorable physic-chemical properties including high water affinity and swelling behavior, and improved  
438 mechanical properties support the usefulness of the magnetic scaffolds for the bone repair and

439 regeneration.

440



441 **4. Conclusion**

442 The surface-functionalized MNPs incorporated within PCL at small contents (up to 10 wt.%)  
443 significantly improved the hydrophilicity and mechanical properties, while enabling superparamagnetic  
444 behaviors of the scaffolds. The PCL-MNP scaffolds were shown to promote apatite forming ability and  
445 stimulate cellular adhesion and mineralization while exhibiting good tissue compatibility. Although  
446 more in-depth studies on bone targeted efficacy are needed, the results demonstrated that the PCL-  
447 MNP magnetic scaffolds hold great promise for use in bone tissue engineering.

448

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## 537 List of Tables and Figures

538

539

540 **Table 1.** Porosity and density of the scaffolds analyzed by mercury intrusion porosimetry, and the  
541 surface area measured by BET.

542 **Table 2.** Histopathologic assessments of the tissue responses in the scaffolds at 2 weeks after  
543 implantation in rat subcutaneous tissue. Four samples were implanted for each group, and the  
544 implanted samples were rated according to the cell numbers/intensity of the lesions; 0: not observed  
545 or very low numbers of infiltrated inflammatory cells, 1: mild changes or low numbers of infiltrated  
546 inflammatory cells, 2: moderate changes or intermediate numbers of infiltrated inflammatory cells, 3:  
547 severe changes/numbers of infiltrated inflammatory cells. FC: fibrous capsule, IR: inflammatory  
548 response, BV: blood vessel, FB: fibroblasts.

549

550 **Fig. 1.** Characteristics of surface-functionalized MNPs; (a) TEM and (b) hysteresis loop.

551 **Fig. 2.** SEM morphologies of PCL, PCL-MNP5, and PCL-MNP10 scaffolds showing a highly porous  
552 structure, and their EDS mapping images representing MNP (Fe) distribution in the scaffold.

553 **Fig. 3.** Characteristics of the scaffolds; (a) XRD pattern, (b) FT-IR spectrum, and (c) TG analysis of  
554 PCL, PCL-MNP5, and PCL-MNP10 scaffolds.

555 **Fig. 4.** (a) Water contact angle, (b) water uptake capacity, and (c) swollen volume of PCL, PCL-MNP5,  
556 and PCL-MNP10 scaffolds.

557 **Fig. 5.** Time- and MNP-dependent apatite-forming ability of the scaffolds; (a) SEM morphology and (b)  
558 XRD pattern of PCL, PCL-MNP5, and PCL-MNP10 scaffolds with SBF immersion for 1, 3, and 7 days.  
559 Scale bar is 5  $\mu\text{m}$  in SEM images. Star-marked diffraction peaks at  $2\theta \approx 26^\circ$  and  $32^\circ$  represent  
560 apatite-characteristic bands in XRD.

561 **Fig. 6.** (a) Field-dependent magnetization curves of PCL, PCL-MNP5, and PCL-MNP10 scaffolds  
562 measured at room temperature, and (b) magnetic distances between a permanent magnet and these  
563 magnetized scaffolds under a static magnetic field. Pictures show magnetic interactions of the  
564 scaffolds with a permanent magnet.

565 **Fig. 7.** Static mechanical properties of PCL, PCL-MNP5, and PCL-MNP10 scaffolds analyzed by (a)  
566 typical compressive stress-strain curves and (b) elastic moduli obtained from initial linear regions  
567 shown in inset of (a).

568 **Fig. 8.** Dynamic mechanical analyses of storage modulus ( $E'$ ), loss modulus ( $E''$ ), tangent delta  
569 ( $E''/E'$ ), and their average values recorded in PCL, PCL-MNP5, and PCL-MNP10 scaffolds performed  
570 over a frequency sweep.

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573 **Fig. 10.** Osteoblastic cell proliferation on the scaffolds; (a) SEM observation at a 14-day culture and (b)  
574 CCK-kit assay of the cells at 3-, 7-, 14-, and 21-day culture.  $*p < 0.05$  and  $**p < 0.01$ , by a one-way  
575 ANOVA test.

576 **Fig. 11.** Cellular mineralization assay, by (a) ARS quantitative measurement of calcium deposits at 7-,  
577 14-, and 21-day cultures, (b) EDS mapping of Ca (green) and P (blue) element deposited on cells  
578 cultured on PCL-MNP10 scaffolds during 21 days, and EDS spectra of Ca and P signals.  $*p < 0.05$   
579 and  $**p < 0.01$ , by a one-way ANOVA test.

580 **Fig. 12.** Histological stains of the implanted scaffolds for 2 weeks in rat subcutaneous tissue. All  
581 groups show similar regenerative patterns of thin capsule with mild inflammatory responses around  
582 scaffolds. The biological host response against the groups was minimal. F: Fibrous capsule, P:  
583 polymer, and R: residual MNP particles

584

585

586 Table 1  
587

	porosimetry with mercury			BET with N <sub>2</sub> gas	
	Porosity (%)	Skeletal Density (g/cm <sup>3</sup> )	Bulk Density (g/cm <sup>3</sup> )	Density	Surface area(m <sup>2</sup> /g)
PCL	64.5 ± 3.54	0.104 ± 0.0006	0.0364 ± 0.003		5.67 ± 1.32
PCL-MNP5	74.58 ± 1.95	0.248 ± 0.030	0.0551 ± 0.007		4.76 ± 2.15
PCL-MNP10	70.9 ± 9.75	0.3705 ± 0.304	0.0620 ± 0.011		7.86 ± 0.44

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597 Table 2  
598

	PCL				PCL-MNP5				PCL-MNP10			
	FC	IR	BV	FB	FC	IR	BV	FB	FC	IR	BV	FB
# 1	1	1	1	2	1	1	1	2	1	1	1	2
# 2	1	1	1	2	1	1	1	2	1	2	1	2
# 3	2	2	1	2	1	2	1	2	2	1	1	2
# 4	2	1	1	2	1	2	1	2	2	1	1	2

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600



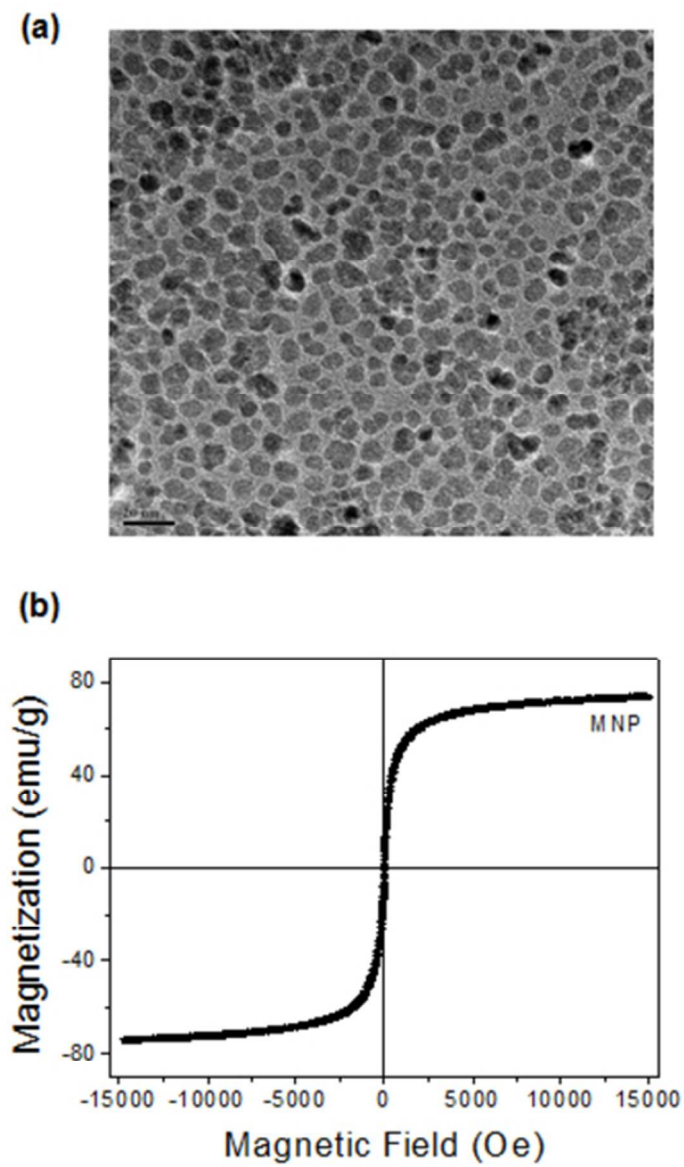


Fig. 1. Characteristics of surface-functionalized MNPs; (a) TEM and (b) hysteresis loop. 18x30mm (600 x 600 DPI)

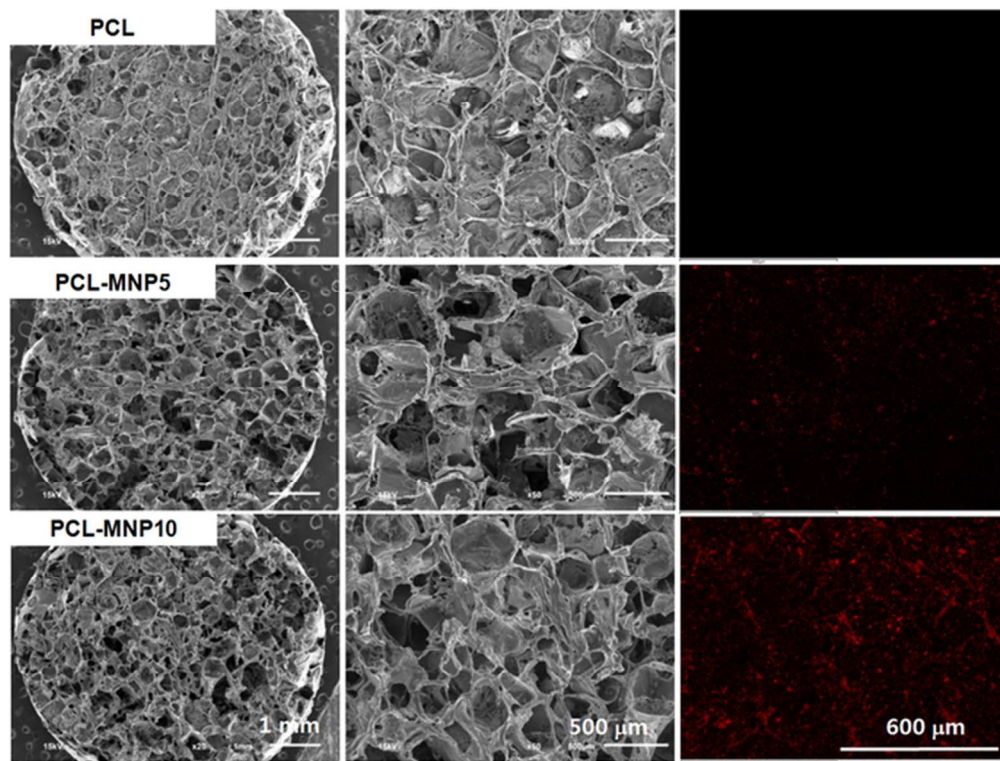


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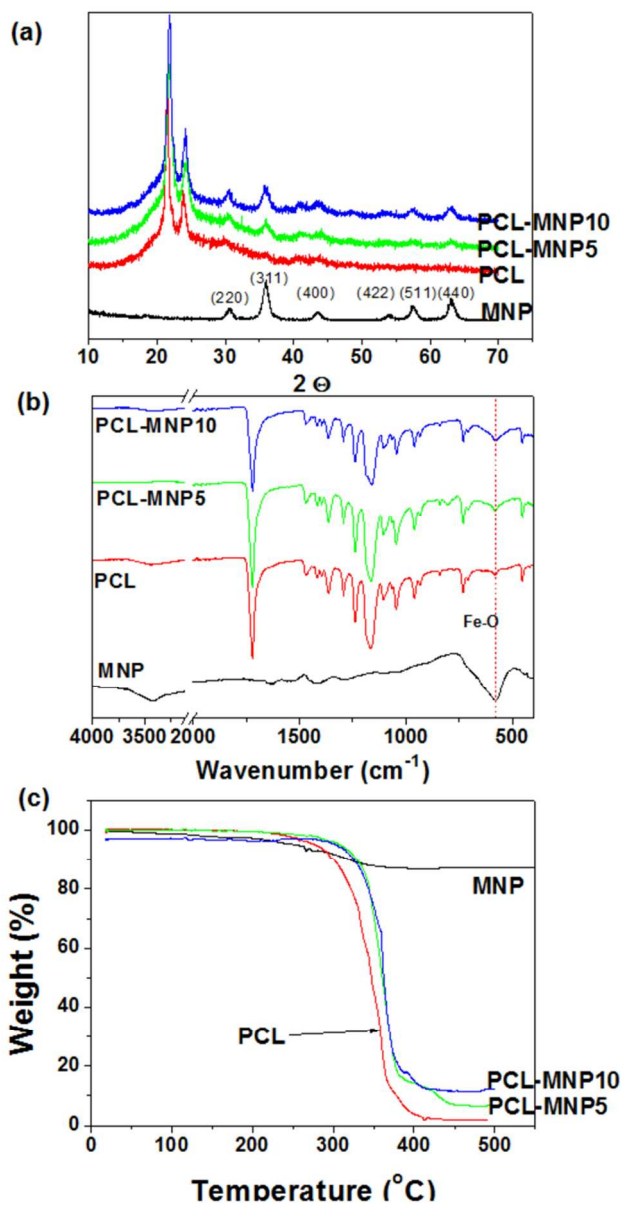


Fig. 3. Characteristics of the scaffolds; (a) XRD pattern, (b) FT-IR spectrum, and (c) TG analysis of PCL, PCL-MNP5, and PCL-MNP10 scaffolds.  
31x59mm (600 x 600 DPI)

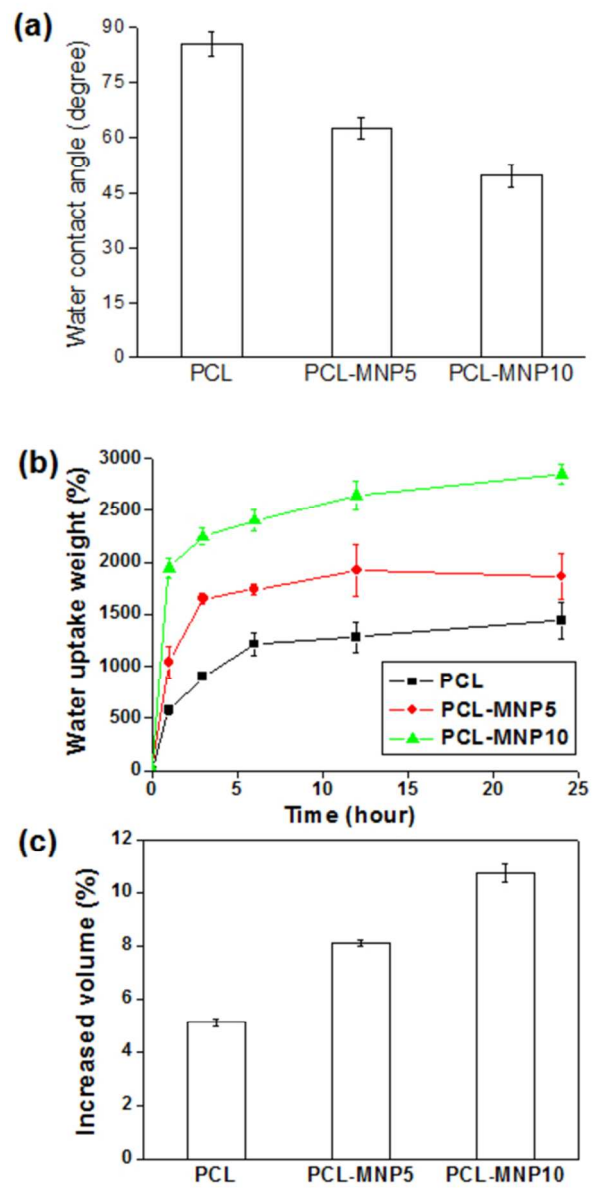


Fig. 4. (a) Water contact angle, (b) water uptake capacity, and (c) swollen volume of PCL, PCL-MNP5, and PCL-MNP10 scaffolds. 25x50mm (600 x 600 DPI)

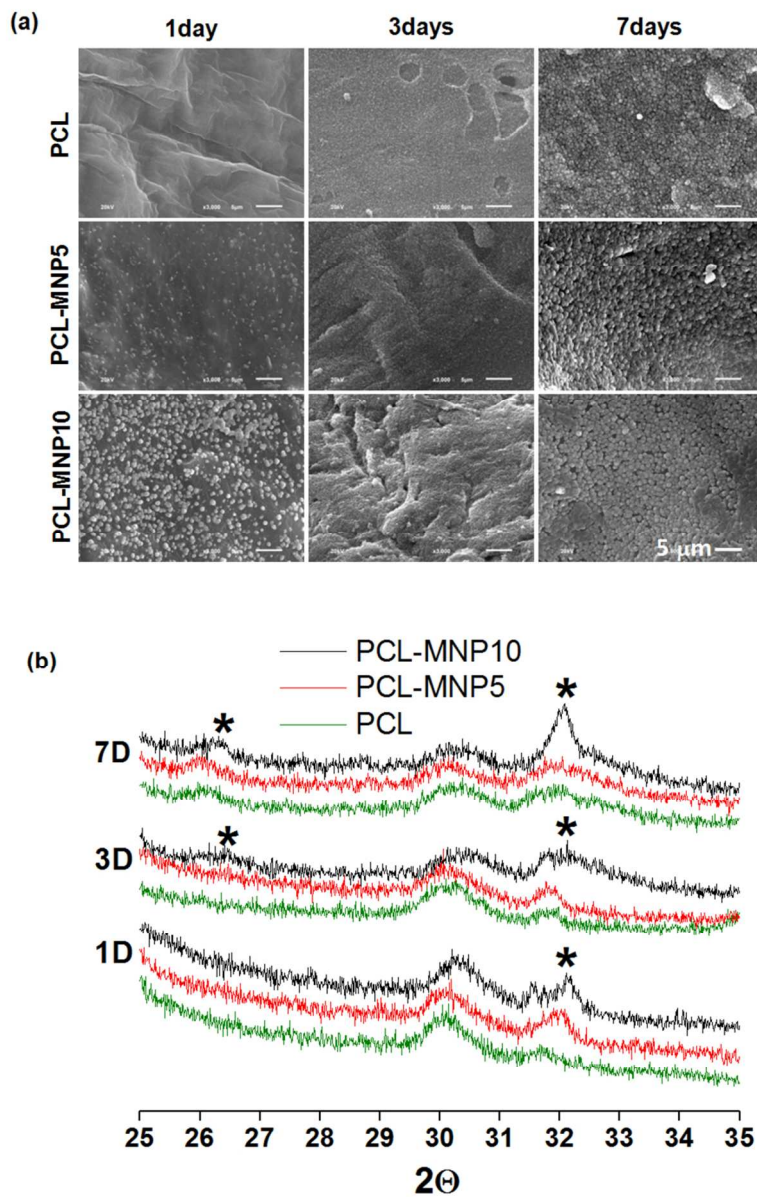


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43x68mm (600 x 600 DPI)

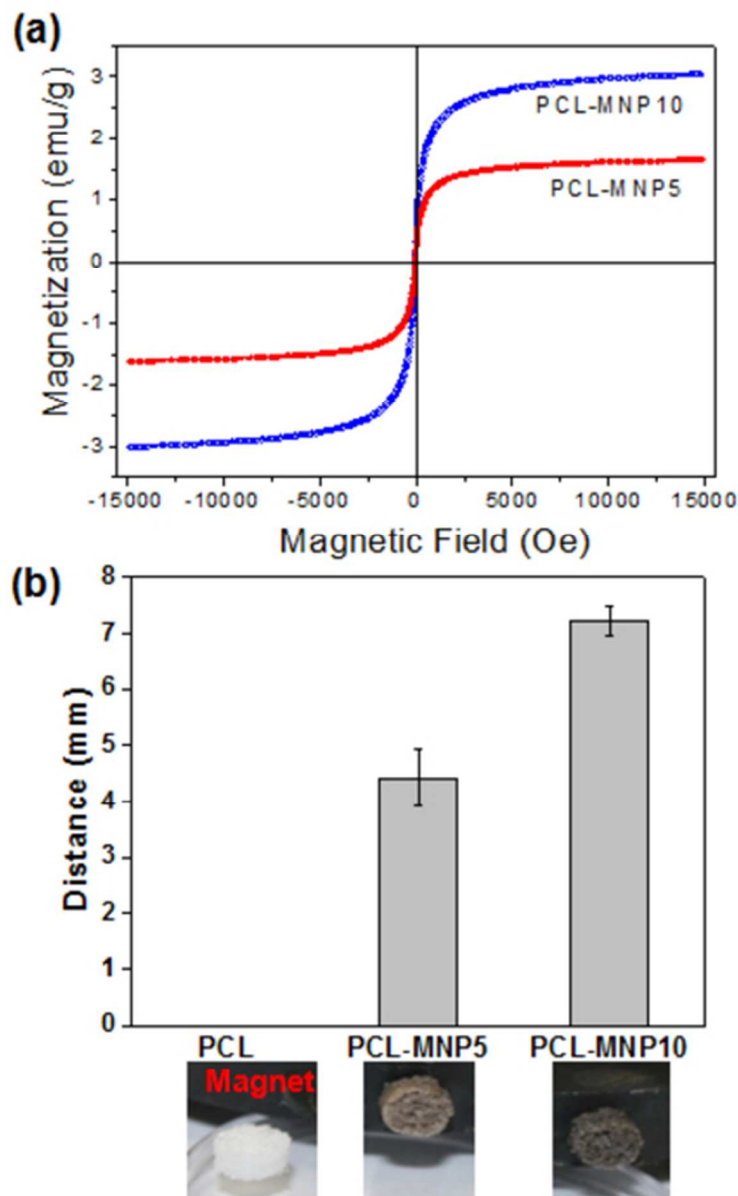


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18x29mm (600 x 600 DPI)

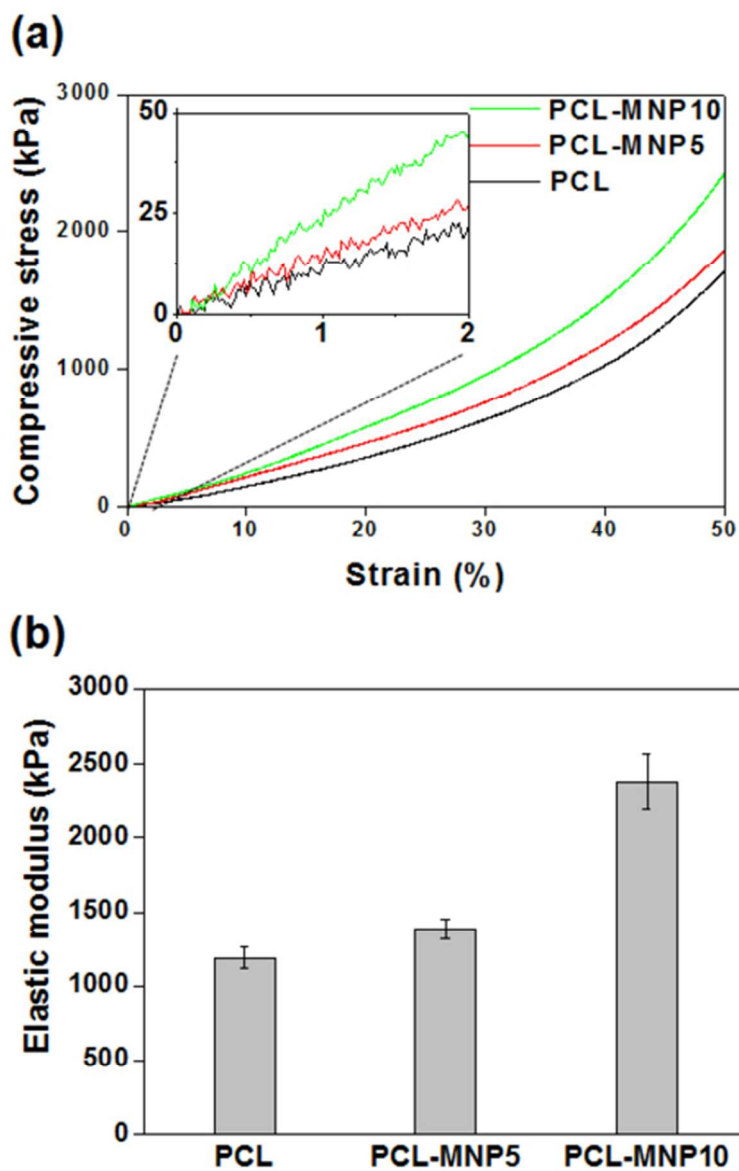


Fig. 7. Static mechanical properties of PCL, PCL-MNP5, and PCL-MNP10 scaffolds analyzed by (a) typical compressive stress-strain curves and (b) elastic moduli obtained from initial linear regions shown in inset of (a).

23x34mm (600 x 600 DPI)

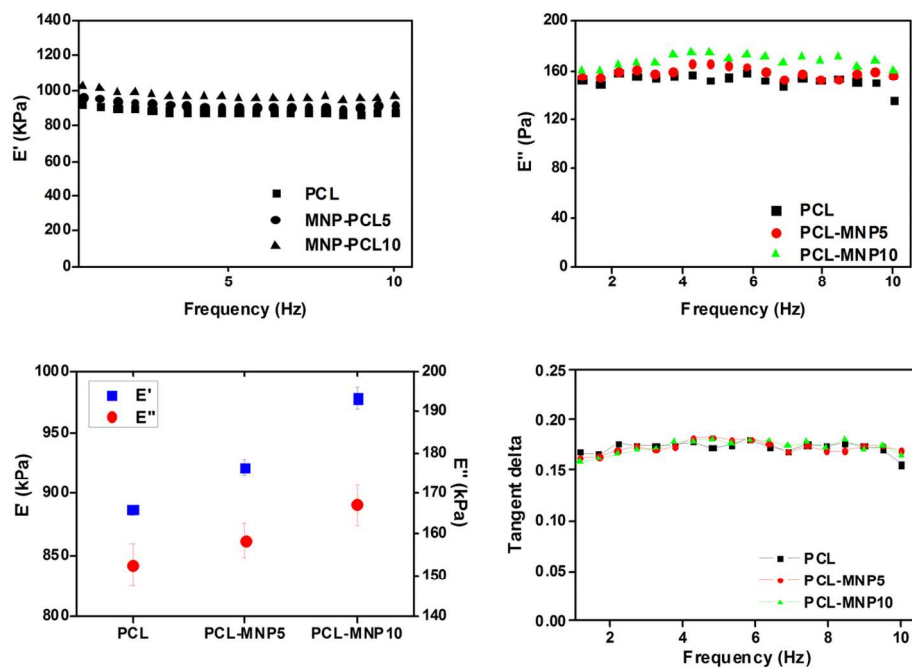


Fig. 8. Dynamic mechanical analyses of storage modulus ( $E'$ ), loss modulus ( $E''$ ), tangent delta ( $E''/E'$ ), and their average values recorded in PCL, PCL-MNP5, and PCL-MNP10 scaffolds performed over a frequency sweep.

122x85mm (300 x 300 DPI)



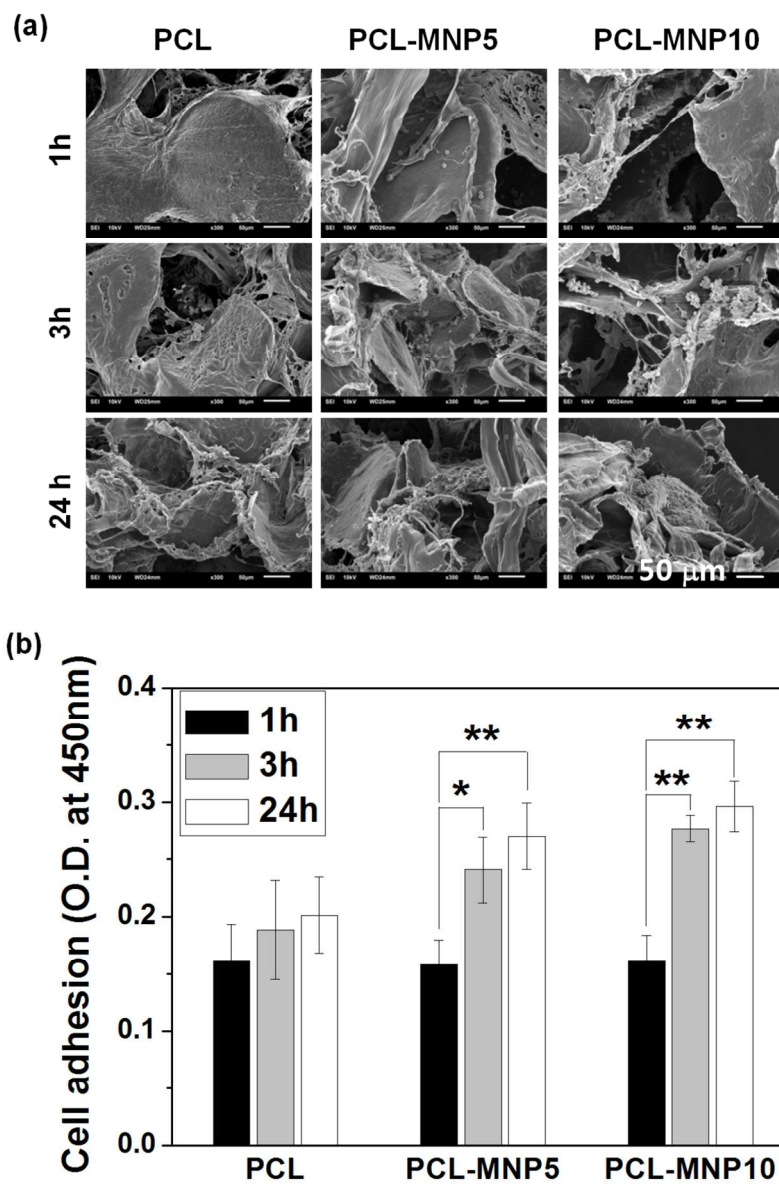


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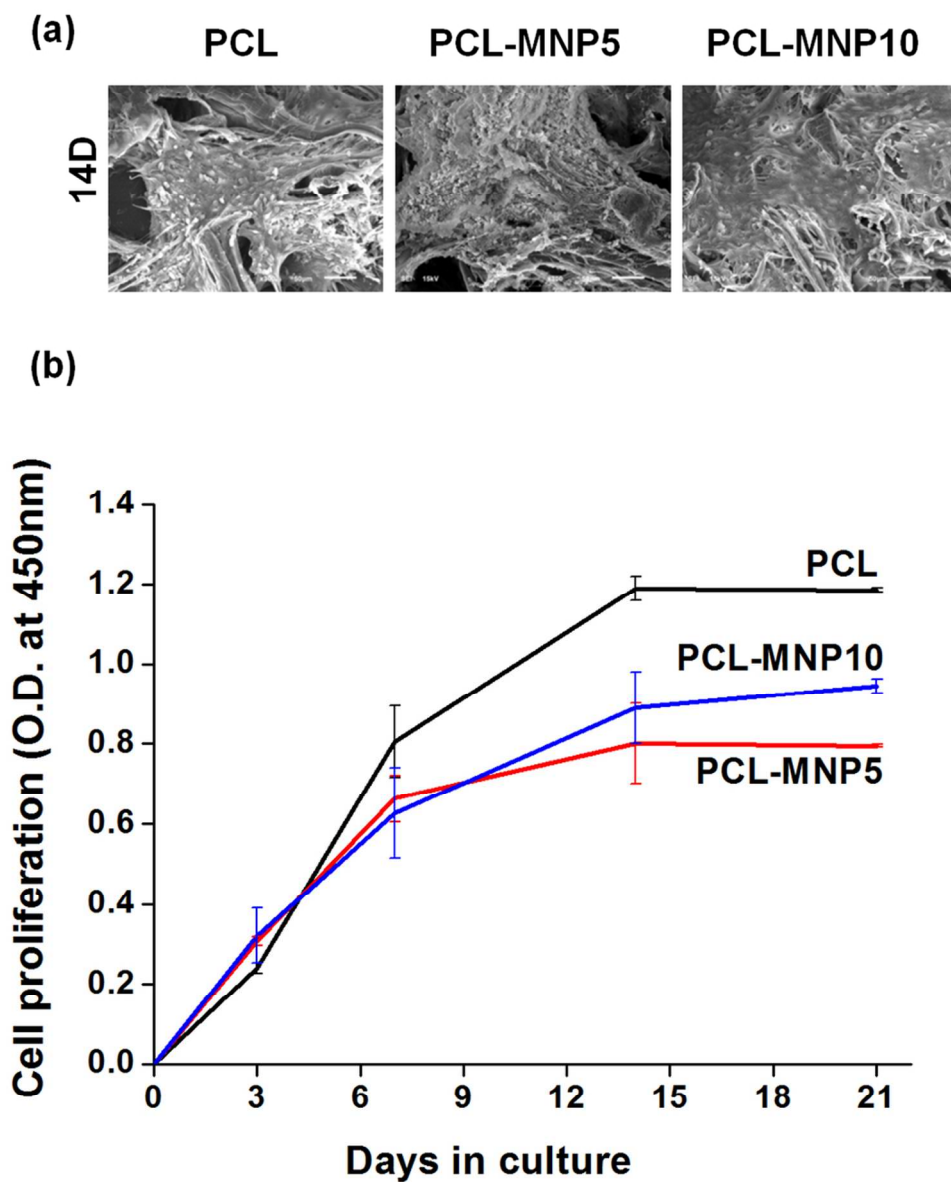


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43x52mm (600 x 600 DPI)

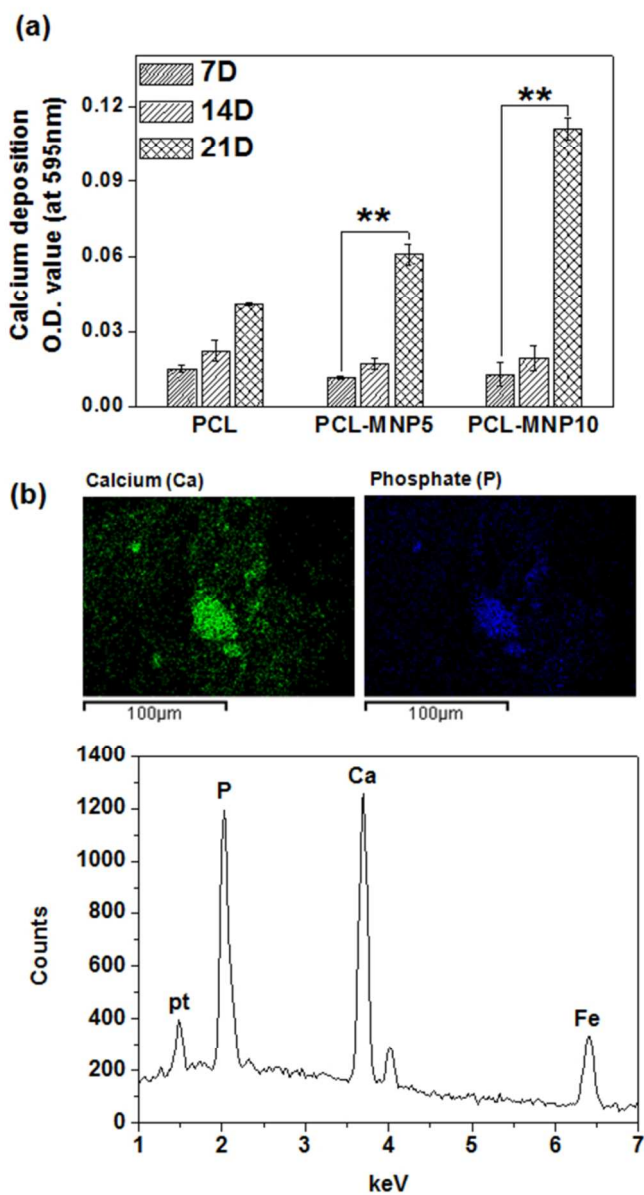


Fig. 11. Cellular mineralization assay, by (a) ARS quantitative measurement of calcium deposits at 7-, 14-, and 21-day cultures, (b) EDS mapping of Ca (green) and P (blue) element deposited on cells cultured on PCL-MNP10 scaffolds during 21 days, and EDS spectra of Ca and P signals. \* $p < 0.05$  and \*\* $p < 0.01$ , by a one-way ANOVA test.

27x49mm (600 x 600 DPI)

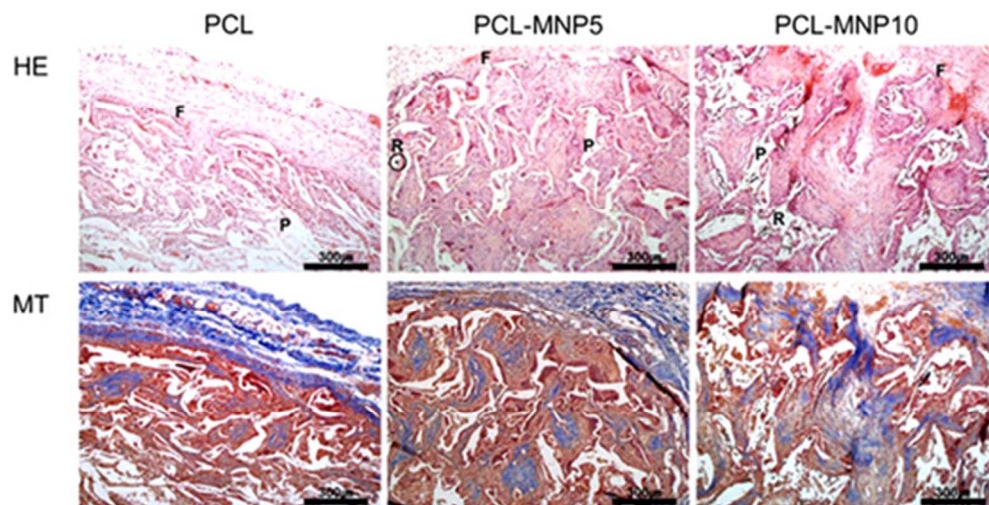


Fig. 12. Histological stains of the implanted scaffolds for 2 weeks in rat subcutaneous tissue. All groups show similar regenerative patterns of thin capsule with mild inflammatory responses around scaffolds. The biological host response against the groups was minimal. F: Fibrous capsule, P: polymer, and R: residual MNP particles  
22x11mm (600 x 600 DPI)