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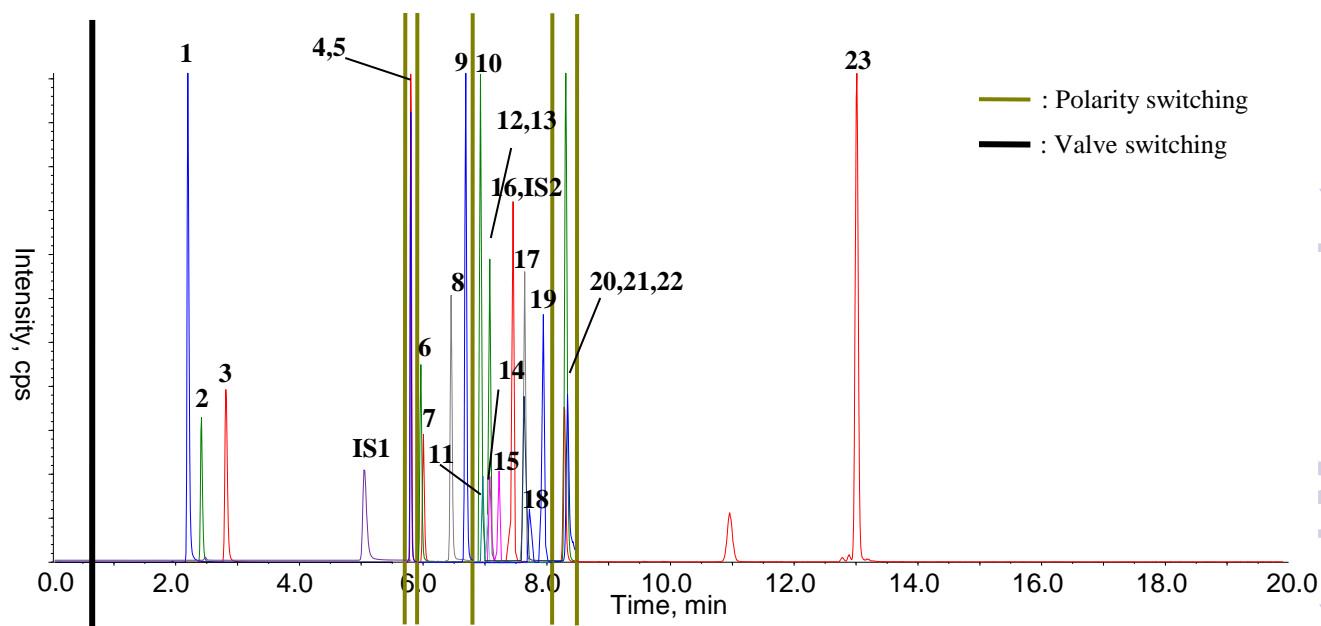
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## Graphic abstract



Simultaneous determination of ten aconite alkaloids and thirteen ginsenosides using online solid phase extraction hyphenated with polarity switching ultra-high performance liquid chromatography coupled with tandem mass spectrometry.

1 **Simultaneous determination of aconite alkaloids and**  
2 **ginsenosides using online solid phase extraction hyphenated**  
3 **with polarity switching ultra-high performance liquid**  
4 **chromatography coupled with tandem mass spectrometry**

5

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23

24 **ABSTRACT**

25 Aconite alkaloids and ginsenosides have been demonstrated as the effective  
26 constituents in Shenfu injection (SFI), which is a widely used Chinese herbal  
27 formulation prepared from red ginseng and processed aconite root. Quality control is  
28 quite critical to meet the increasing demand of SFI in market and to guarantee the  
29 safety in clinic, in particular regarding those toxic aconite alkaloids. Given the  
30 significant merits of online solid phase extraction (SPE) for sample preparation, an  
31 online SPE-based method was developed to simultaneously quantify ten aconite  
32 alkaloids and thirteen ginsenosides in SFI using UHPLC–MS/MS. Because of their  
33 distinct ionization behaviors, polarity switching was implemented in ion source  
34 domain with a scheduled program, and positive and negative ionization modes were  
35 applied for alkaloids and saponins, respectively. Quantitative terms of the proposed  
36 method with respect to linearity, limit of detection, lower limit of quantification,  
37 precision, and accuracy were evaluated, and the results indicated that the method is  
38 sensitive, convenient, and reliable. The developed method was successfully applied  
39 for the simultaneous determination of aconite alkaloids and ginsenosides in ten  
40 batches of SFI (SFI1–10). The validated method can be adopted as a meaningful tool  
41 for the quality control of SFI concerning aconite alkaloids and ginsenosides, and also  
42 it offers a reliable choice for the widely qualitative analysis of constituents in  
43 complex matrices being free from tedious sample preparation procedures.

44

45 **Keywords:** Online solid phase extraction; Polarity switching; Aconite alkaloids;  
46 Ginsenosides; Shenfu injection.

47

48

## 49 1. Introduction

50 Shenfu injection (SFI) is a modern Chinese medicine preparation derived from a  
51 traditional formulation, Shenfu decoction [1]. It is prepared from the extracts of red  
52 ginseng (steamed roots of *Panax ginseng*) and aconite (processed lateral roots of  
53 *Aconitum carmichaeli*) using multistage countercurrent extraction and macroporous  
54 resin adsorption technology [2]. In comparison with the decoction, SFI is  
55 advantageous at convenience for clinical use [3]. This Chinese medicine preparation  
56 used alone or combined with other routine treatments has been widely accepted as an  
57 effective therapeutic approach for chronic congestive heart failure in clinical  
58 practices [4]. SFI has also been extensively implemented as a complementary agent  
59 for the treatment of ischemic cardiomyopathy with heart insufficiency [5], septic  
60 shock [6], acute myocardial dysfunction [7], *etc.* In addition, SFI has been suggested  
61 it is beneficial for postoperative recovery after heart surgery [8, 9] and improvement  
62 of life quality of patients undergoing chemotherapy [10-12]. In recent years,  
63 mechanism investigations have revealed that the clinical outcomes of SFI mainly  
64 rely on accelerating energy metabolism and antioxidant capacity [13, 14],  
65 modulating apoptosis [13-15], blocking sodium channels [16], inhibiting expressions  
66 of TNF- $\alpha$ , IL-1 $\beta$ , and IL-6 [17, 18], and down-regulating the activity of NF- $\kappa$ B [18].  
67 As we known, aconite alkaloids and ginsenosides are the primary effective  
68 ingredients in SFI, and the modern pharmacological evaluations have demonstrated  
69 that ginsenosides offer determinant contribution for the vasodilator benefit of SFI  
70 [19], whereas the alkaloids play vital roles in cardiac electrophysiological effect for  
71 SFI by blocking ion channels [16].

72 Quality control is quite critical to meet the increasing demand of SFI in market  
73 and to guarantee the safety in clinic, in particular regarding those toxic aconite  
74 alkaloids. In comparison with the long history in clinical application, nonetheless,  
75 reports concerning the quality control of SFI have been seldom archived [3]. Natural  
76 variations in climate and preparation procedures usually give rise to batch-to-batch  
77 inconsistency of SFI, thus leading to safety risks, because diterpene alkaloids are of  
78 potential toxicity at the meanwhile of significant activity, even at low concentrations

79 [20, 21]. Therefore, it is crucial to propose sensitive and reliable analytical  
80 techniques to strictly control the concentration windows of aconite alkaloids in SFI  
81 products, as well as the ginsenosides.

82 Generally speaking, injection should be the isotonic solution of blood to avoid  
83 cell damage caused by the exposure of blood cells to too much or too little solute  
84 during intravenous administration; hence, it is not astonishing that our preliminary  
85 assays demonstrated a wealth of metal ions presented in SFI using inductively  
86 coupled plasma mass spectrometry (ICP-MS). As a consequence, it is critical to  
87 divert or remove those non-volatile and hydrophilic substances prior to quantitative  
88 analysis using ultra-high performance liquid chromatography hyphenated with  
89 tandem mass spectrometry (UHPLC-MS/MS), to avoid ion source contamination  
90 and signal suppression [22]. Liquid-liquid extraction (LLE) and offline solid phase  
91 extraction (SPE) generally involve tedious manual working procedures and/or  
92 relatively large amounts of solvent that must be removed prior to UHPLC-MS/MS  
93 analysis [22]; however, the superiorities of online SPE include high automatic,  
94 solvent-saving and efficiency. In view of those significant merits of online SPE for  
95 sample preparation, in current study, an online SPE-based method was developed  
96 and validated for the simultaneous determination of ten aconite alkaloids along with  
97 thirteen ginsenosides in SFI. Because of the distinct ionization properties between  
98 alkaloids and saponins, the detection of analytes was carried out by multiple reaction  
99 monitoring (MRM) scanning coupled with switching electrospray ion source polarity  
100 between positive and negative modes in a single run, and positive and negative  
101 ionization modes were applied for alkaloids and saponins, respectively.

102

## 103 **2. Materials and method**

### 104 **2.1. Materials**

105 Ginsenosides 20(S)-F1, 20(S)-F2, Rb1, Rb2, Rc, Rd, Re, Rf, Rg1, 20(S)-Rg3,  
106 20(R)-Rh1, Ro, and pseudo-ginsenoside F11 (F11), as well as ten diterpene alkaloids,  
107 namely songorine, neoline, talatisamine, benzoylmesaconine, benzoylaconine,  
108 benzoylhypaconine, aconitine, hypaconitine, lappaconite hydrobromide, and

109 yunaconitine were all obtained from Shanghai Standard Biotech Co. Ltd (Shanghai,  
110 China). Their chemical structures (Fig. 1) were further elucidated in the authors'  
111 laboratory by  $^1\text{H}$ -,  $^{13}\text{C}$ -NMR and MS analyses, and the purity of each standard  
112 compound was determined to be more than 98% by normalization of the peak areas  
113 detected with HPLC–DAD–MS/MS. Two internal standards (IS), including  
114 laurotetanine (IS1, purity>98%) for aconite alkaloids and tenuifolin (IS2,  
115 purity>98%) for ginsenosides were previously purified and identified from *Litsea*  
116 *cubeba* and *Polygala tenuifolia* in our lab, respectively.

117 A total of ten batches of SFI products (SFI1–10) were all supplied by Ya'an  
118 Sanjiu Pharmaceutical Co., Ltd (Sichuan, China).

119 Formic acid, ammonium formate, methanol, trifluoroacetic acid (TFA), and  
120 acetonitrile (ACN) were of HPLC grade and purchased from Merck (Darmstadt,  
121 Germany). Deionized water was prepared by Milli-Q plus water purification system  
122 (Millipore, Bedford, MA, USA). The other chemicals were of analytical grade and  
123 obtained commercially from Beijing Chemical Works (Beijing, China).

124

## 125 2.2. Sample preparation

126 Stock solutions of all analytes were prepared by dissolving each authentic  
127 compound in methanol at a concentration of approximately 1.0 mg/mL and then  
128 stored at 4°C until use. Serial standard mixtures were prepared by mixing all stock  
129 solutions and subsequently diluting using Milli-Q water containing both 1.0 ng/mL  
130 of IS1 and 1.0 ng/mL of IS2 to desired concentration levels. SFI products (SFI1–10)  
131 were 1000-fold diluted with Milli-Q water spiked with IS1 and IS2 (1.0 ng/mL for  
132 either). All standard solutions and SFI samples were filtered through 0.22  $\mu\text{m}$  filter  
133 membranes prior to online SPE–UHPLC–MS/MS measurement.

134

## 135 2.3. Online SPE–UHPLC–MS/MS analysis

136 The liquid chromatographic analysis was conducted on a Shimadzu UHPLC  
137 system (Shimadzu, Kyoto, Japan) consisting of two LC-20AD<sub>XR</sub> solvent delivery  
138 units, a LC-20AD pump, a SIL-20AC<sub>XR</sub> autosampler, a CTO-20AC column oven, a

139 SPD-M20A DAD detector, DGU-20A<sub>3R</sub> degasser, and a CBM-20A controller. An  
140 ABSciex 5500 Qtrap mass spectrometer (ABSciex, Foster City, CA, USA), mounted  
141 with an electrospray ionization (ESI) interface and a 6-port/2-channel valve, was  
142 used to connect with the UHPLC system. ABSciex Analyst Software package  
143 Version 1.6.2 was adopted to synchronize the whole system, and also for data  
144 acquisition and processing. The general layout of the online SPE hyphenated with  
145 UHPLC–MS/MS instrumentation setup proposed previously was introduced in  
146 current study with minor modifications [23], and the schematic is elucidated in Fig.  
147 2. An entire measurement was divided into two phases, the loading phase and the  
148 elution phase, by alternating the valve between A- and B-channel (Figs. 2A and 2B).

149 Both of the SPE column (Security Guard™, C<sub>18</sub> 3.0 × 4 mm I.D., Phenomenex,  
150 Torrance, CA, USA) and the analytical column (Kinetex-C<sub>18</sub> shell 4.6 × 100 mm I.D.,  
151 2.6 μm particle size, Phenomenex) were maintained at room temperature (25°C).  
152 The LC-20AD pump was responsible for delivering 2% aqueous acetonitrile  
153 containing 10 mmol/L ammonium formate to SPE column at a flow rate of 0.8  
154 mL/min aiming to facilitate the lipophilic constituents of the injected sample being  
155 trapped in the SPE column. After flushing hydrophilic substances at loading phase  
156 (A-channel for valve) for 30 seconds (Fig. 2A), the system was switched to elution  
157 phase (B-channel for valve, Fig. 2B) and maintained for another 19.5 minutes, and  
158 the lipophilic matrix components including the analytes were back-flushed from SPE  
159 column into the analytical column using a programmed gradient condition. The  
160 mobile phase was composed of 10 mmol/L ammonium formate-H<sub>2</sub>O (A) and 0.1%  
161 HCOOH-ACN (B) and delivered by the two LC-20AD<sub>XR</sub> pumps in gradient at a  
162 flow rate of 0.8 mL/min as follows: 0–3.0 min, 20% B; 3.0–6.0 min, 20%–40% B;  
163 6.0–10.0 min, 40% B; 10.0–15.0 min, 40%–60% B; 15.0–18.0 min, 60% B; and  
164 18.0–20.0 min, 20% B. The temperature of autosampler was set at 4°C to stabilize  
165 samples.

166 The mass spectrometer was operated in MRM mode. Ion optics was tuned using  
167 standard polypropylene glycol (PPG) dilution solvent. Nitrogen was used as the  
168 nebulizer, curtain, heater, and collision gas. The ion source was heated to 500°C, and



169 the ion-spray voltages were maintained at 5500 V and -4500 V for positive and  
170 negative ionization, respectively. Gas setting: nebulizer gas (GS1) as 55 psi, heater  
171 (GS2) as 55 psi and curtain gas (CUR) as 35 psi. Two precursor-to-product ion  
172 transitions were recorded for each analyte or IS. The polarity-switching schedule, the  
173 precursor-to-product ion transitions, optimized declustering potential values (DP),  
174 and collision energy values (CE) are elucidated in Table 1, while the dwell time,  
175 entrance potential (EP) and collision cell exit potential (CXP) levels of each ion  
176 transition were fixed at 120 ms, 10 V, and 12 V, respectively.

177 The Analyst software quantification module was used to achieve the quantitative  
178 process including peak detection, peak integration, and analyte quantification.  
179 Within the automated Analyst Classic integration algorithm the smoothing factor was  
180 set as 2 and the bunching factor as 1 for all monitored peaks.

181

## 182 2.4. Method validation

183 The method validation was carried out in term of internationally accepted  
184 criterion [24].

### 185 2.4.1. Linearity, LOQ and LOD

186 The linearity was assayed using external calibration curves with more than seven  
187 concentration levels for each analyte, and each level was conducted in triplicate.  
188 Calibration curve generation was performed by plotting the peak area ratio of an  
189 analyte and its corresponding IS against the theoretical concentration over the  
190 calibration concentration range. A 1/x weighting function was used for the linear  
191 regression of each analyte. The acceptance criterion for each calibration curve was a  
192 correlation coefficient ( $r$ ) greater than 0.99. Lower limit of quantification (LLOQ)  
193 and limit of determination (LOD) were termed as the concentration at a  
194 signal-to-noise ratio (S/N) of about 10 and 3, respectively.

### 195 2.4.2. Precision, repeatability, and stability

196 Intra- and inter-day variations were selected to achieve the precision assay for  
197 the proposed method. For intra-day variability evaluation, the mixed standard  
198 solutions at low, medium and high concentration levels were analyzed for six

199 replicates within a single day, while for inter-day assay, the solutions were examined  
200 in triplicate per day for consecutive three days. Variations were expressed by relative  
201 standard deviations (RSDs).

202 To evaluate the repeatability, six replicates of a selected sample (SFI1) were  
203 analyzed using the aforementioned method. The RSD value was calculated to  
204 measure the repeatability. In order to investigate the stability of the samples, the  
205 selected sample solution (SFI1) was analyzed every twelve hours within consecutive  
206 three days and the solution was stored at 4°C.

#### 207 2.4.3. Recovery assay

208 The recovery was used to assess the accuracy of the method. Known amounts  
209 (low, medium, and high concentration levels) of mix standards solution were added  
210 into a certain amount (0.10 mL) of selected SFI (SFI1). Afterwards, the combined  
211 solution was 1000-fold diluted with Milli-Q water fortified with IS1 and IS2 (1.0  
212 ng/mL for either) and subjected for online SPE hyphenated with UHPLC–MS/MS  
213 analysis.

214

### 215 3. Results and discussion

#### 216 3.1. Development of online SPE–UHPLC–MS/MS

217 Firstly, a Phenomenex guard column was selected from several SPE columns to  
218 carry out solid phase extraction. SPE column loading procedure was modified from  
219 the method published previously [23], and the time point for valve switching was  
220 fixed before the elution of songorine by directly introducing the SPE column effluent  
221 into mass spectrometer.

222 Several columns were screened in term of acceptable separation for those  
223 analytes, at least between alkaloids and ginsenosides, and Phenomenex Kinetex-C<sub>18</sub>  
224 shell column (100 mm × 4.6 mm i.d., particle size 2.6 µm) was finally chosen due to  
225 it is advantageous at peak capacity, peak shape, and back-pressure [25, 26], in  
226 comparison with ACE UltraCore 2.5 SuperC<sub>18</sub> column (150 mm × 3.0 mm i.d.,  
227 particle size 2.5 µm, Advance Chromatography Technologies Ltd., Aberdeen,  
228 Scotland), Phenomenex Kinetex-C<sub>18</sub> shell column (100 mm × 2.1 mm i.d., particle

229 size 2.6  $\mu\text{m}$ ), and Capcell core ADME column (2.1 mm  $\times$  150 mm i.d., particle size  
230 2.7  $\mu\text{m}$ , Shiseido, Tokyo, Japan).

231 For mass parameter optimization, each analyte stock solution was diluted to  
232 appropriate content (50–100 ng/mL) by 50% aqueous methanol containing 10  
233 mmol/L ammonium formate, and directly infused into the mass spectrometer at a  
234 rate of 10  $\mu\text{L}/\text{min}$  by the syringe pump. The investigated analytes and internal  
235 standards were firstly characterized according to their MS<sup>1</sup> and MS<sup>2</sup> spectra to  
236 ascertain their precursor-to-product ion transitions for quantitative analysis. The  
237 positive and negative ionization modes were compared, and the results proved that  
238 the negative mode could provide higher responses for ginsenosides and tenuifolin  
239 (IS2), while positive ionization was found to be more suitable for aconite alkaloids  
240 and laurotetanine (IS1). After careful optimization, the formate  
241 anion-to-deprotonated ion transitions were selected for most ginsenosides, whereas  
242 positive quasi-molecular ion generated the highest responses for all alkaloids. The  
243 prominent daughter ions of alkaloids were usually yield by the neutral loss of  
244 methanol, carbon monoxide or water. Two precursor-to-product ion transitions per  
245 compound (quantifier and qualifier ion pairs) were set, with the latter one (Table 1)  
246 being matched to the quantifier ion transition in respect to CE and DP values. Ion  
247 source gas flows and ion source temperature adopted the typical ranges for the  
248 UHPLC effluent.

249 Finally, mobile phase modifiers were screened among ammonium acetate, formic  
250 acid, acetic acid, TFA, and ammonium formate by comparing the overall response of  
251 the analytes. Formic acid and ammonium formate were finally introduced as the  
252 modifiers for organic and aqueous mobile phases, respectively, to enhance the  
253 ionization of both ginsenosides and alkaloids.

254 Above all, the optimized chromatography and mass spectrometry parameters are  
255 shown in “UHPLC–MS/MS analysis” section and in Table 1.

256

### 257 3.2. Mass fragmentation behaviors of ginsenosides and aconite alkaloids

258 Aconite alkaloids and ginsenosides were widely demonstrated as the dominant

259 effective components in SFI. As for the bioactive ingredients in *Panax quinquefolius*,  
260 *P. notoginseng* and *P. ginseng* [27, 28], the fragmentation patterns of ginsenosides  
261 have been well defined [29, 30]. In current study, most ginsenosides generated  
262 significant intensities of  $[M+HCOO]^-$  adduct ions, whereas the intensities of the  
263 expected deprotonated molecular ions ( $[M-H]^-$ ) were relatively low due to the  
264 introduction of ammonium formate and formic acid as the mobile phase additives  
265 (Fig. S1) [31]. When the formate anion was selected as the precursor ion to generate  
266  $MS^2$  spectra, neutral cleavage of formic acid (HCOOH, 46 u) occurred initially to  
267 yield the deprotonated molecular ion ( $[M-H]^-$ ). And then,  $[M-H]^-$  ion would  
268 successively expelled sugar residues, for instance glucose, rhamnose and xylose, by  
269 the dissociation of glucosidic bonds, and  $[aglycone-H]^-$  ( $m/z$  475 or 459) were  
270 generated finally; however, few further fragment was observed for  $[aglycone-H]^-$   
271 (Fig. S1).

272 On the other side, ionization and dissociation of aconite alkaloids occurred under  
273 positive ionization. Aside from songorine ( $C_{20}$ -type alkaloid), all the investigated  
274 alkaloids could be categorized into diterpene ( $C_{19}$ -type) alkaloids. It was reported  
275 that the cleavage of a methanol ( $CH_3OH$ , 32 u) moiety is the primary fragmentation  
276 pathway of aconite alkaloids [32]. When the protonated molecular ions ( $[M+H]^+$ ) of  
277 alkaloids were transmitted to collision induce dissociation (CID) cell, the neutral  
278 losses of 32 u were usually observed attributing to the presences of methoxy groups  
279 for most alkaloids (Fig. S1). Moreover, the dissociation of  $H_2O$  was also frequently  
280 observed for  $C_{19}$ -type alkaloids. In the case of songorine, the absence of methoxy  
281 substituent resulted in the absence of 32 u loss in the  $MS^2$  spectrum; instead,  
282 successive cleavages of  $H_2O$  (18 u) groups occurred as the prominent fragmentation  
283 pathways.

284

### 285 3.3. Method Validation

#### 286 3.3.1. Linearity, LLOQ and LOD

287 A weight of  $1/x$  was applied to minimize the relative error for the curve fitting.  
288 Correlation coefficients ( $r$ ) of calibration curves in all inter-run cases were higher

289 than 0.99 over the concentration ranges (Table 2). The LLOQs and LODs are  
290 elucidated at Table 2. Except for ginsenosides F1, 20(*R*)-Rh1, and 20(*S*)-Rg3, the  
291 LLOQs of all analytes were lower than 20 pg/mL, and the LODs were less than 10  
292 pg/mL. The values suggested great performances of linearity and sensitivity for the  
293 developed method.

### 294 3.3.2. Precision, repeatability and stability assays

295 For all targeted compounds, accuracies located at the range of 88.3–113.6% for  
296 all low, medium and high concentration levels. The RSDs of intra- and inter-day  
297 precisions were found lower than 14.9% for all analytes. Table 3 presents the results  
298 for precision assay. Those data indicated that the developed method is precise and  
299 accurate.

300 The repeatability presented as RSD ( $n = 6$ ) was between 5.39% and 13.7%, and  
301 the results of stability assay suggested that the samples could keep stable during  
302 online SPE–UHPLC–MS/MS measurements.

### 303 3.3.3. Recovery assay

304 Known amounts (low, medium and high concentration levels) of mix standards  
305 solution were added into 0.10 mL of SFI product (SFI1) prior to online  
306 SPE–UHPLC–MS/MS analysis. The recoveries were observed between 86.1% and  
307 112.9% for all analytes, which could satisfy the quantitative criteria for alkaloids and  
308 ginsenosides in complex matrices (Table 3).

309 In addition, the impacts from carryover and re-injection were also assessed and  
310 the results suggested that influences of these two items could be neglected due to the  
311 quite slight influence of them.

312

### 313 3.4. Determination of twenty-three analytes in SFI

314 The developed online SPE–UHPLC–MS/MS system was applied for  
315 simultaneous quantification of twenty-three analytes in ten batches of SFI products  
316 (SFI1–10). The typical chromatogram of SFI is shown in Fig. 3B, and all the  
317 determined contents are summarized in Table 4.

318 Overall, the contents of ginsenosides were much higher than those of aconite

319 alkaloids, which are in great coincidence with the findings archived in Ref. [3, 33];  
320 however, ginsenoside 20(*S*)-F1 and pseudo-ginsenoside F11 were not detected in all  
321 SFI samples. In particular, ginsenosides Rb1, Rg1, Rc, Rb2, Re, and 20(*S*)-Rg3 were  
322 determined as the abundant constituents, whereas the contents of ginsenosides Rd  
323 and 20(*R*)-Rh1 were slightly lower than the other ginsenosides. Although the toxicity  
324 of aconite alkaloids was widely mentioned, the contents of those compounds in SFI  
325 (lower than 1.0  $\mu\text{g}/\text{mL}$  for most alkaloids) ascertained the location at the safety  
326 window with a clinical dosage. In particular, trace distributions (lower than 0.1  
327  $\mu\text{g}/\text{mL}$  in most cases) were revealed for aconitine, yunaconitine, and hypaconitine,  
328 particularly lappaconite hydrobromide, which are sorted into diester-diterpenoid  
329 alkaloids and exhibit greater toxic possibility than those monoester-diterpenoid  
330 alkaloids [34]. The contents of both ginsenosides and aconite alkaloids showed  
331 relatively big variations (approximately 3 folds) among different batches.

332 In practice, a problem for the hyphenation of LC and MS is that nonvolatile  
333 substances cannot be introduced into the mass spectrometer, since the nonvolatile  
334 salts, particularly metal ions, dramatically degrade its performance [35–37]. In  
335 addition, the metals can also join with analytes having carboxyl, carbonyl, ether or  
336 ester group to form cluster adducts, which could result in irreproducible quantitative  
337 results. We preliminarily measured the contents of metals in SFI using Elan DRCII  
338 ICP-MS (Perkin Elmer, Waltham, MA, USA). The contents of  $\text{Na}^+$ , and  $\text{K}^+$  were  
339 measured as 628.0 and 62.6  $\mu\text{g}/\text{mL}$  in SFI, respectively, and the total contents of  
340 some other metals, such as Fe, Mg, and Ca, were determined more than 4  $\mu\text{g}/\text{mL}$ .  
341 Therefore, when SFI was directly injected into mass spectrometer, it is of a great risk  
342 for contaminating our system. We were reluctant to risk contaminating our system,  
343 which is used for many other purposes; thus, we did not have the opportunity to  
344 obtain information about the long-term influence of metal ions on mass spectrometer  
345 performance.

346 Despite that a couple of methods have been proposed to monitor ginsenosides  
347 and aconite alkaloids in Shenfu products using UHPLC–MS/MS, our current study is  
348 advantageous at synchronous determination, pretreatment-free and high sensitivity.

349 Further studies are ongoing in our laboratory on characterization of the  
350 pharmacokinetic profiles of SFI in rats, and the proposed online  
351 SPE–UHPLC–MS/MS system is expected to be an appropriate technique to directly  
352 analyze the biological samples.

353

#### 354 **4. Concluding remarks**

355 A novel sensitive and selective online SPE hyphenated with UHPLC–MS/MS  
356 method operating negative and positive switching mode in a single analysis process  
357 was developed and validated in terms of LOD, LLOQ, linearity, precision, accuracy  
358 and recovery assays. A total of twenty-three constituents, including ten aconite  
359 alkaloids and thirteen ginsenosides, were simultaneously determine in ten batches of  
360 SFI. Above all, the validated method not only provides a meaningful tool for the  
361 quality control of SFI, but also offers a reliable choice for the widely qualitative  
362 analysis of constituents in complex matrices without tedious sample preparation  
363 procedures.

364

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370

#### 371 **Appendix Supplementary data**

372 Supplementary data (Supplemental figures) associated with this article can be  
373 found, in the online version, at <http://dx.doi.org/.....>

374

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439

440 **Figure legends**

441 **Fig. 1** Chemical structures and molecular weights (M.W.) of and 23 analytes  
442 investigated in this study, including thirteen ginsenosides, 20(*S*)-F1, 20(*S*)-F2, Rb1,  
443 Rb2, Rc, Rd, Re, Rf, Rg1, 20(*S*)-Rg3, 20(*R*)-Rh1, Ro, and pseudo-ginsenoside F11  
444 (F11), as well as ten diterpene alkaloids, songorine, neoline, talatisamine,  
445 benzoylmesaconine, benzoylaconine, benzoylhypaconine, aconitine, hypaconitine,  
446 lappaconite hydrobromide, and yunaconitine.

447 **Fig. 2** Connectivity sketch of the six-port switching valve controlling the online SPE  
448 hyphenated with UHPLC–MS/MS step. Loading phase: the specimen aliquot  
449 delivered from the auto-sampler is captured onto the SPE column using 2% aqueous  
450 acetonitrile containing 10 mM ammonium formate to expel those hydrophilic and  
451 non-volatile substances, for example metal ions, while the valve is maintained at  
452 A-channel; elution phase: the specimen fraction adsorbed onto the SPE column,  
453 mainly containing ginsenosides, aconite alkaloids, and some other apolar substances,  
454 is eluted using a programmed gradient elution and subsequent to MS/MS detection,  
455 and the valve is maintained at B-channel. Details are described at Section 2.3. Online  
456 SPE–UHPLC–MS/MS analysis.

457 **Fig. 3** Representative multiple ion pairs extracted ion current (TIC) chromatograms  
458 of MRM mode of reference compounds mixture (A) and SFI product (SFI1, B). As  
459 described above, valve switching (marked with black line) occurs at 0.5 min, and  
460 polarity switching (marked with brown line) occurs for five times. Except for  
461 loading phase (0–0.5 min), each segment is magnified to make every signal visible,  
462 and the intensities of all the base peaks (highest signals) are elucidated. **1**, Songorine;  
463 **2**, Neoline; **3**, Talatisamine; **4**, Rg1; **5**, Re; **6**, Benzoylmesaconine; **7**, Lappaconite  
464 hydrobromide; **8**, Benzoylaconine; **9**, Benzoylhypaconine; **10**, F11; **11**, Rb1; **12**, Ro;  
465 **13**, Rf; **14**, Rc; **15**, Rb2; **16**, 20(*S*)-Rg2; **17**, Rd; **18**, 20(*R*)-Rh1; **19**, 20(*S*)-F1; **20**,  
466 Aconitine; **21**, Yunaconitine; **22**, Hypaconitine; **23**, 20(*S*)-Rg3.

467

468 **Table 1** The precursor-to-product ion transitions, declustering potential values (DP),  
 469 collision energy values (CE), retention times ( $t_R$ ) of the 23 targeted components and t  
 470 the polarity switching schedule.

period	duration	analyte	$t_R$ (min)	Ion transition Precursor>product <sup>a</sup>	DP (V)	CE (eV)
period 1 (positive)	5.70	Songorine	2.24	<b>358&gt;340</b> ; 358>322	100	39
		Neoline	2.46	<b>438&gt;420</b> ; 438>388	120	40
		Talatisamine	2.88	<b>422&gt;390</b> ; 422>358	120	39
		IS1	4.94	<b>328&gt;311</b> ; 328>280	100	17
period 2 (negative)	0.17	Rg1	5.78	<b>845&gt;845</b> ; 845>799	-100	-11
		Re	5.78	<b>991&gt;991</b> ; 991>945	-100	-11
period 3 (positive)	0.93	Benzoylmesaconine	5.95	<b>590&gt;540</b> ; 590>558	90	48
		Lappaconite hydrobromide	5.99	<b>585&gt;356</b> ; 585>324	60	46
		Benzoylaconine	6.44	<b>604&gt;554</b> ; 604>572	100	47
		Benzoylhypaconine	6.67	<b>574&gt;542</b> ; 547>510	103	47
period 4 (negative)	1.31	F11	6.91	<b>845&gt;845</b> ; 845>799	-100	-11
		Rb1	6.95	<b>1153&gt;1153</b> ; 1153>1107	-100	-15
		Ro	7.02	<b>1001&gt;1001</b> ; 1001>955	-100	-15
		Rf	7.06	<b>845&gt;845</b> ; 845>799	-100	-11
		Rc	7.06	<b>1123&gt;1123</b> ; 1123>1077	-100	-15
		Rb2	7.20	<b>1123&gt;1123</b> ; 1123>1077	-100	-15
		20(S)-Rg2	7.42	<b>829&gt;829</b> ; 829>783	-100	-11
		IS2	7.44	<b>679&gt;679</b> ; 679>455	-100	-15
		Rd	7.61	<b>991&gt;991</b> ; 991>945	-100	-15
period 5 (positive)	0.40	Aconitine	8.28	<b>646&gt;586</b> ; 646>526	120	44
		Yunaconitine	8.31	<b>660&gt;600</b> ; 660>568	107	42
		Hypaconitine	8.34	<b>616&gt;556</b> ; 616>524	130	44
		20(S)-Rg3	13.03	<b>829&gt;829</b> ; 829>783	-100	-11
period 6 (negative)	11.49					

471 <sup>a</sup>: two ion pairs were optimized for each analyte, and the ion transitions in bold were  
 472 adopted for quantitative analysis, while the other one was adopted as qualifier ion  
 473 pair.

474 **Table 2** Linear regression data, lower limits of quantification (LLOQs) and limits of  
 475 detection (LODs) for all targeted analytes.

Analyte	Linear regression data			LLOQ (pg/mL)	LOD (pg/mL)
	Regression equation	r	Linear range (ng/mL)		
Songorine	$y = 17.1 x - 0.341$	0.997	0.0816–20.4	6.53	1.31
Neoline	$y = 3.3 x + 0.265$	0.999	0.0190–59.5	7.62	1.52
Talatisamine	$y = 5.06 x - 0.165$	0.999	0.0176 – 55.0	7.04	1.41
Rg1	$y = 1.4 x + 3.61$	0.996	0.476 – 119.0	7.62	1.52
Re	$y = 0.542 x + 1.44$	0.999	0.0282 – 88.0	5.63	1.13
Benzoylmesaconine	$y = 1.79 x + 0.129$	0.997	0.0712 – 44.5	0.11	0.06
Lappaconite hydrobromide	$y = 1.82 x - 0.0461$	0.997	0.0728 – 45.5	1.16	0.58
Benzoylaconine	$y = 2.64 x - 0.000127$	0.999	0.0848 – 53.0	1.36	0.68
Benzoylhypaconine	$y = 3.89 x + 0.302$	0.999	0.0310 – 48.5	1.24	0.62
F11	$y = 0.607 x + 0.113$	0.997	0.336 – 84.0	5.38	1.08
Rb1	$y = 0.17 x + 0.0853$	0.990	0.195 – 122	7.81	1.56
Ro	$y = 0.818 x + 0.438$	0.998	0.128 – 80.0	2.56	1.02
Rf	$y = 0.387 x + 0.425$	0.997	0.340 – 85.0	5.44	2.72
Rc	$y = 0.0625 x + 0.0449$	0.996	0.0291 – 91.0	5.82	2.91
Rb2	$y = 0.0603 x + 0.0731$	0.990	0.0397 – 124	7.94	3.97
20(S)–Rg2	$y = 0.467 x + 1.12$	0.991	0.252 – 63.0	4.03	2.02
Rd	$y = 0.226 x + 0.324$	0.997	0.0784 – 98.0	6.27	3.14
20(R)–Rh1	$y = 1.67 x - 0.00622$	0.999	0.656 – 82.0	78.4	15.7
20(S)–F1	$y = 1.48 x + 0.0355$	0.999	0.157 – 98.0	392	157
Aconitine	$y = 1.95 x - 0.13$	0.996	0.0888 – 55.5	17.8	7.10
Yunaconitine	$y = 4.34 x - 0.17$	0.998	0.0944 – 59.0	18.9	7.55
Hypaconitine	$y = 2.17 x + 0.127$	0.996	0.0776 – 48.5	15.5	6.21
20(S)–Rg3	$y = 0.0997 x + 0.175$	0.993	0.108 – 67.5	108.0	54.0

476 **Table 3** Intra-, inter-day and recovery performance parameters of quality control  
 477 samples for all monitored analytes ( $n = 6$ ).

Analyte	Concentration (ng/mL)	Intra-day RSD (%)	Inter-day RSD (%)	Recovery (%)	RSD (%)
Songorine	low	3.76	12.3	97.9	4.31
	medium	8.05	4.53	93.2	3.24
	high	13.2	14.8	91.3	1.93
Neoline	low	8.31	14.4	95.5	2.17
	medium	7.69	3.06	96.1	2.64
	high	14.6	14.7	108.6	1.98
Talatisamine	low	5.91	11.5	96.8	3.33
	medium	7.43	4.08	103.4	4.12
	high	13.3	14.9	111.1	1.44
Rg1	low	9.73	11.8	106.6	9.87
	medium	11.3	13.5	109.2	10.2
	high	10.6	12.4	95.8	8.94
Re	low	10.7	12.5	97.1	9.25
	medium	12.3	13.6	99.3	9.88
	high	11.8	13.9	100.4	11.0
Benzoylmesaconine	low	9.58	9.08	96.8	2.40
	medium	7.33	5.08	101.2	1.35
	high	13.6	14.9	106.4	3.54
Lappaconite hydrobromide	low	5.45	10.6	98.3	4.56
	medium	7.55	4.48	97.9	1.97
	high	12.8	13.9	108.8	3.61
Benzoylaconine	low	6.09	11.9	99.6	1.88
	medium	8.46	5.83	98.9	1.43
	high	13.1	13.9	102.4	3.65
Benzoylhypaconine	low	6.44	9.62	103.1	1.49
	medium	6.78	3.70	101.7	5.23
	high	13.7	14.5	98.5	3.44
Rb1	low	12.1	13.9	86.1	8.95
	medium	9.80	11.7	89.9	9.13
	high	13.9	14.1	111.2	9.23
Ro	low	8.52	11.6	90.1	10.4
	medium	7.83	13.3	113.2	9.67
	high	11.6	14.6	98.5	8.91
Rf	low	11.1	12.9	101.7	10.3
	medium	10.4	13.1	111.6	11.1
	high	10.9	9.97	95.4	7.99
Rc	low	9.01	10.0	89.9	10.4
	medium	9.53	8.13	92.2	11.3
	high	11.3	7.46	98.8	10.6

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Rb2	low	7.84	11.3	112.9	8.93
	medium	6.53	9.37	107.1	10.5
	high	8.10	10.8	96.0	11.4
20(S)-Rg2	low	6.54	9.03	112.7	10.3
	medium	7.27	6.55	107.5	9.86
	high	8.43	10.9	110.9	10.6
Rd	low	9.35	14.0	95.5	9.65
	medium	10.3	12.8	93.1	9.73
	high	8.24	10.6	91.3	9.32
20(R)-Rh1	low	9.17	12.5	97.2	10.8
	medium	9.20	11.1	109.4	9.98
	high	11.3	12.4	108.8	10.0
Aconitine	low	5.19	2.34	98.4	3.21
	medium	7.59	2.59	99.1	4.36
	high	12.63	12.41	96.5	2.95
Yunaconitine	low	5.16	2.74	97.2	3.71
	medium	8.36	3.68	104.2	5.23
	high	12.65	12.62	106.6	4.51
Hypaconitine	low	10.32	7.22	102.3	2.55
	medium	9.28	3.36	95.8	1.98
	high	12.53	13.82	98.9	3.72
20(S)-Rg3	low	7.37	8.77	111.2	11.4
	medium	8.46	9.13	109.9	10.8
	high	11.4	13.2	91.1	10.0

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478

479

**Table 4** The contents ( $\mu\text{g/mL}$ ) of investigated compounds in the ten batches of Shenfu injection (SF11–10)

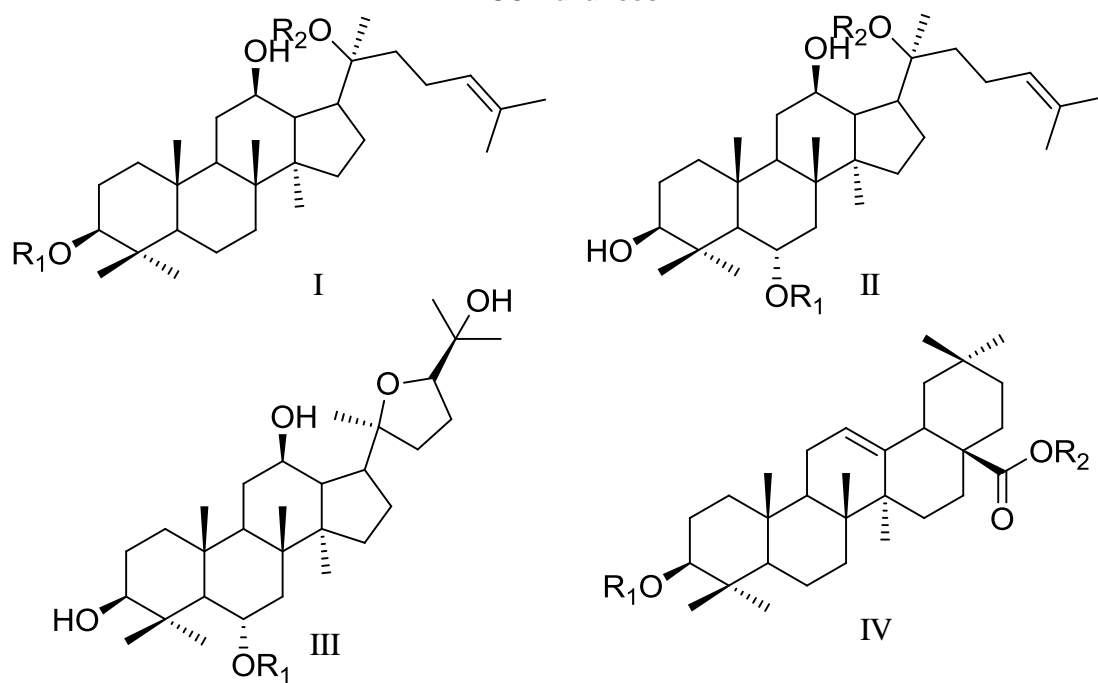
Sam	Contents																				
ples	1 <sup>a</sup>	2	3	4	5	6	7	8	9	11	12	13	14	15	16	17	18	20	21	22	23
SF1	0.162±	0.646±	0.103±	135.3±	74.2±	1.76±	0.026±	0.174±	0.528±	191.2±	69.0±	91.2±	48.7±5	76.0±	34.2±	37.4±	35.8±	0.070±	0.041±	0.027±	74.9±
	0.02	0.08	0.01	14.9	7.02	0.22	0.00	0.02	0.08	44.7	10.3	13.8	.73	14.6	5.64	3.12	3.72	0.00	0.00	0.01	12.8
SF2	0.145±	0.601±	0.094±	150.4±	81.7±	1.23±	0.026±	0.130±	0.323±	245.6±	51.8±	63.0±	114.4±	80.8±	32.5±	27.4±	36.0±	0.072±	0.041±	0.042±	65.4±
	0.04	0.22	0.05	9.38	4.9	0.40	0.00	0.05	0.13	31.2	8.17	4.06	11.0	12.4	6.69	2.34	4.40	0.00	0.00	0.03	5.14
SF3	0.116±	0.487±	0.066±	130.0±	75.6±	0.90±	0.026±	0.114±	0.411±	154.3±	40.6±	72.0±	79.7±1	49.0±	40.2±	31.4±	37.2±	0.070±	0.040±	0.029±	73.7±
	0.01	0.05	0.01	16.9	14.7	0.14	0.00	0.02	0.07	25.9	8.02	6.40	0.9	7.16	3.34	5.33	6.74	0.00	0.00	0.03	12.5
SF4	0.219±	0.919±	0.158±	107.6±	67.7±	1.60±	0.027±	0.209±	0.709±	136.6±	29.2±	55.2±	86.2±9	47.8±	49.6±	38.6±	36.6±	0.072±	0.041±	0.053±	49.6±
	0.08	0.38	0.07	7.91	7.42	0.56	0.00	0.08	0.29	25.9	4.10	6.98	.24	6.63	5.28	3.02	1.36	0.00	0.00	0.04	4.94
SF5	0.192±	0.776±	0.134±	122.8±	74.3±	1.37±	0.028±	0.179±	0.611±	137.5±	32.0±	54.4±	94.2±1	46.1±	55.8±	42.8±	37.2±	0.075±	0.044±	0.033±	56.4±
	0.06	0.30	0.07	6.68	8.28	0.47	0.00	0.06	0.24	26.3	5.60	5.26	2.2	6.17	4.48	5.44	2.90	0.01	0.01	0.03	8.76
SF6	0.188±	0.730±	0.123±	132.3±	84.1±	2.05±	0.027±	0.233±	0.856±	79.3±1	24.0±	82.8±	46.3±7	26.4±	39.7±	28.8±	30.6±	0.073±	0.043±	0.060±	29.6±
	0.05	0.22	0.05	22.0	15.7	0.70	0.00	0.07	0.29	1.1	2.6	4.90	.93	1.59	4.42	3.86	5.18	0.00	0.00	0.03	2.90
SF7	0.267±	1.23±0	0.272±	128.6±	79.6±	2.48±	0.027±	0.317±	1.11±0.	137.8±	34.8±	90.8±	65.8±1	64.7±	40.3±	46.8±	35.4±	0.076±	0.041±	0.093±	75.2±
	0.02	.08	0.03	17.8	16.8	0.20	0.00	0.03	11	19.9	1.44	8.43	8.4	3.66	7.51	3.09	4.98	0.00	0.00	0.02	10.3
SF8	0.148±	0.817±	0.160±	116.0±	57.6±	0.74±	0.027±	0.108±	0.278±	89.8±1	23.6±	77.0±	44.0±8	45.8±	24.2±	25.8±	26.2±	0.078±	0.042±	0.104±	36.9±
	0.02	0.19	0.03	14.3	8.99	0.13	0.00	0.02	0.04	4.5	4.64	8.90	.17	5.41	4.64	2.20	1.04	0.00	0.00	0.03	3.76
SF9	0.249±	0.663±	0.215±	146.9±	83.4±	1.74±	0.031±	0.213±	0.545±	193.5±	37.0±	67.4±	99.8±1	59.3±	43.4±	47.4±	38.6±	0.165±	0.051±	0.692±	48.2±
	0.08	0.05	0.02	11.7	10.2	1.13	0.00	0.08	0.30	20.1	1.38	8.97	8.3	7.13	3.11	6.74	7.60	0.06	0.00	0.51	6.39
SF10	0.112±	0.490±	0.117±	121.2±	78.4±	0.49±	0.038±	0.080±	0.161±	169.9±	47.2±	50.0±	95.5±1	55.5±	38.0±	31.2±	32.8±	0.091±	0.058±	0.094±	56.4±
	0.03	0.12	0.05	12.7	9.51	0.45	0.01	0.05	0.13	25.5	5.5	7.08	0.1	8.31	6.14	4.76	5.22	0.03	0.02	0.10	6.01



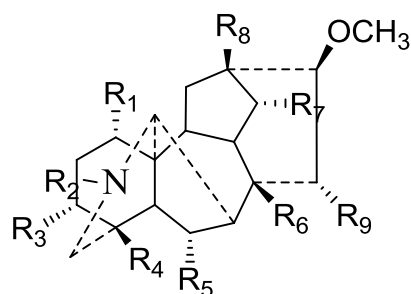
480 <sup>a</sup>: **1**, Songorine; **2**, Neoline; **3**, Talatisamine; **4**, Rg1; **5**, Re; **6**, Benzoylmesaconine; **7**, Lappaconite hydrobromide; **8**, Benzoylaconine; **9**,  
481 Benzoylhypaconine; **11**, Rb1; **12**, Ro; **13**, Rf; **14**, Rc; **15**, Rb2; **16**, 20(*S*)–Rg2; **17**, Rd; **18**, 20(*R*)–Rh1; **20**, Aconitine; **21**, Yunaconitine;  
482 **22**, Hypaconitine; **23**, 20(*S*)–Rg3.

483

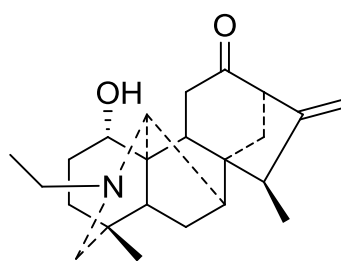
484



Compound	Skeleton	R <sub>1</sub>	R <sub>2</sub>	M.W.
Rb1	I	glc(2-1)glc	glc(4-1)glc	1108
Rb2	I	glc(2-1)glc	glc(4-1)ara	1078
Rc	I	glc(2-1)glc	glc(6-1)ara	1078
Rd	I	glc(2-1)glc	glc	946
20(S)-Rg3	I	glc(2-1)glc	H	784
20(S)-Rg2	II	glc(2-1)rha	H	784
20(S)-F1	II	H	glc	638
Re	II	glc(2-1)rha	glc	946
Rf	II	glc(2-1)glc	H	800
Rg1	II	glc	glc	800
20(R)-Rh1	II	glc	H	638
F11	III	glc(2-1)rha	-	800
Ro	IV	gluA(2-1)glc	glc	956



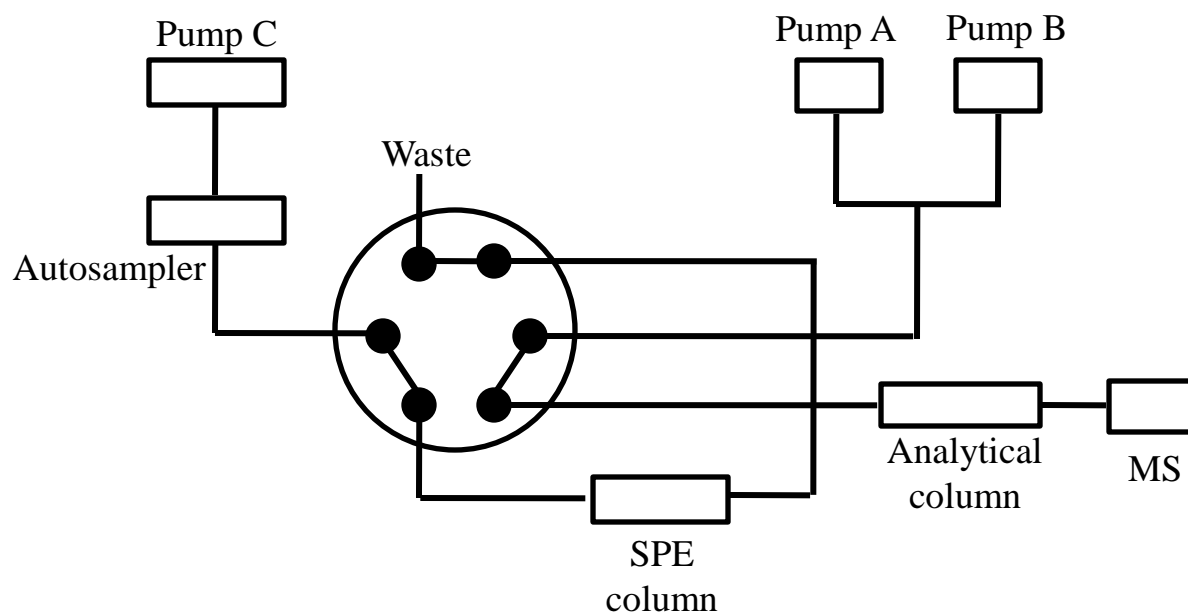
Skeleton of alkaloids



Songorine (M.W.: 357)

Compound	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	R <sub>7</sub>	R <sub>8</sub>	R <sub>9</sub>	M.W.
Aconitine	OCH <sub>3</sub>	C <sub>2</sub> H <sub>5</sub>	OH	CH <sub>2</sub> OCH <sub>3</sub>	OCH <sub>3</sub>	OCOCH <sub>3</sub>	OBz	OH	OH	645
Hypaconitine	OCH <sub>3</sub>	CH <sub>3</sub>	H	CH <sub>2</sub> OCH <sub>3</sub>	OCH <sub>3</sub>	OCOCH <sub>3</sub>	OBz	OH	OH	615
Benzoylaconitine	OCH <sub>3</sub>	C <sub>2</sub> H <sub>5</sub>	OH	CH <sub>2</sub> OCH <sub>3</sub>	OCH <sub>3</sub>	OH	OBz	OH	OH	603
Benzoylhypaconitine	OCH <sub>3</sub>	CH <sub>3</sub>	H	CH <sub>2</sub> OCH <sub>3</sub>	OCH <sub>3</sub>	OH	OBz	OH	OH	573
Benzoylmesaconitine	OCH <sub>3</sub>	CH <sub>3</sub>	OH	CH <sub>2</sub> OCH <sub>3</sub>	OCH <sub>3</sub>	OH	OBz	OH	OH	589
Neoline	OH	C <sub>2</sub> H <sub>5</sub>	H	CH <sub>2</sub> OCH <sub>3</sub>	OCH <sub>3</sub>	OH	OH	H	H	437
Talatisamine	OCH <sub>3</sub>	C <sub>2</sub> H <sub>5</sub>	H	CH <sub>2</sub> OCH <sub>3</sub>	H	OH	OH	H	H	421
Yunaconitine	OCH <sub>3</sub>	C <sub>2</sub> H <sub>5</sub>	OH	CH <sub>2</sub> OCH <sub>3</sub>	OCH <sub>3</sub>	OCOCH <sub>3</sub>	OBzOCH <sub>3</sub>	OH	H	659
Lappaconite Hydrobromide	OCH <sub>3</sub>	C <sub>2</sub> H <sub>5</sub>	H	OBzNHCOCCH <sub>3</sub>	H	OH	OCH <sub>3</sub>	H	H	584

## Loading phase



## Elution phase

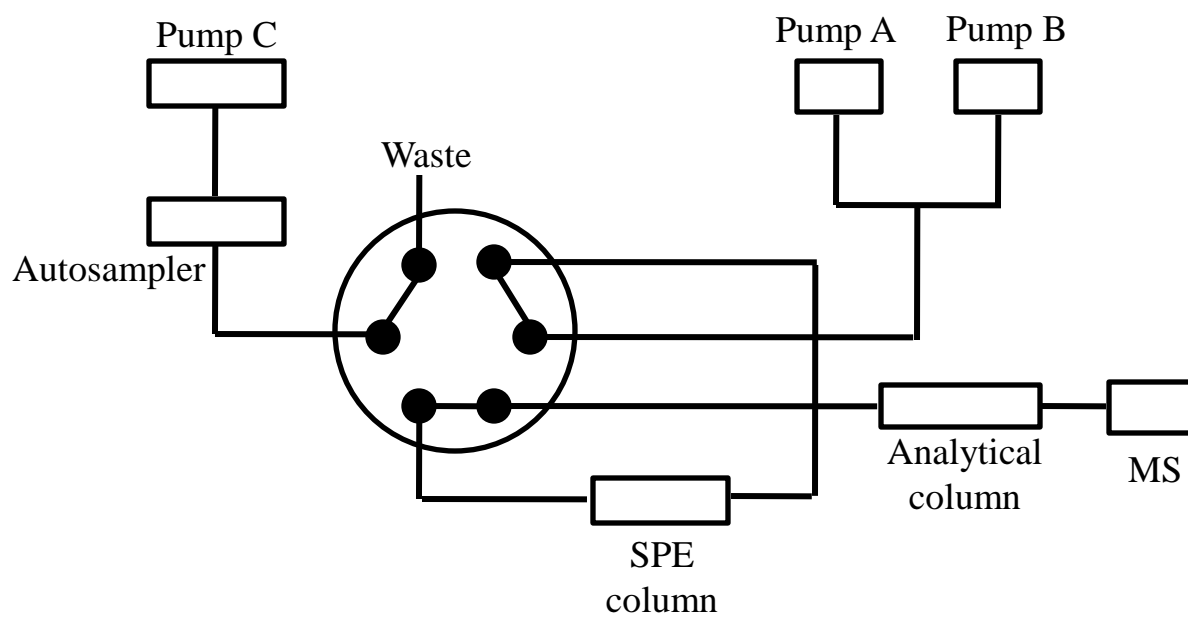
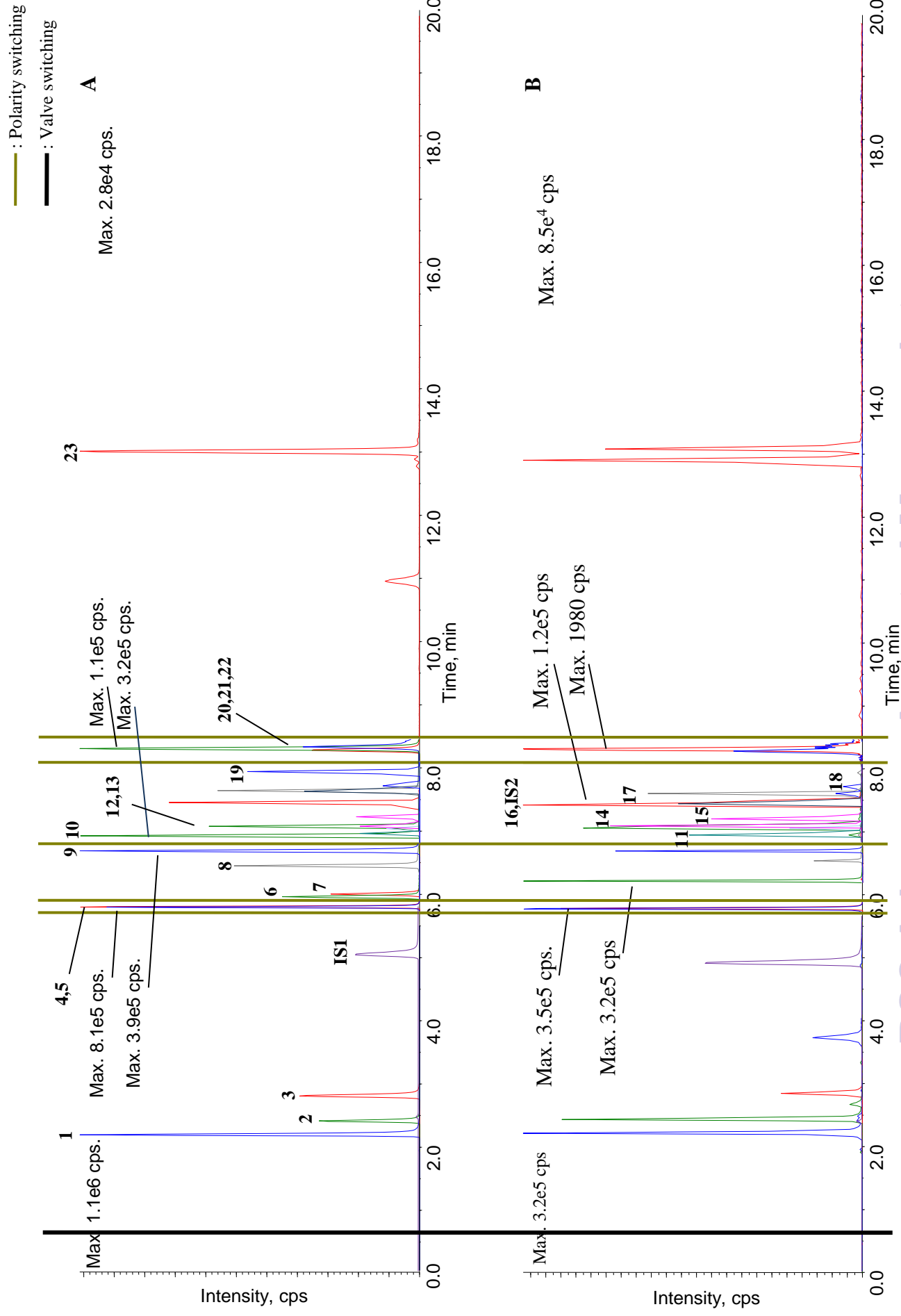


Fig. 2



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Fig.3