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ARTICLE

Amphiphilic Trimethylpyridylporphyrin-Fullerene (C₇₀) Dyad: An Efficient Photosensitizer under Hypoxia Condition

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Amphiphilic trimethylpyridylporphyrin-C₇₀ (PC₇₀) dyad with improved photosensitization has been successfully prepared. The PC₇₀ dyad forms liposomal nanostructure through molecular self-assembling. Increased absorption coefficient in visible region, good biocompatibility, and high photostability were observed on self-assembling structure. Surprisingly, in comparison with previously reported photosensitizer porphyrin, PC₇₀ exhibited enhanced photodynamic therapy (PDT) effect under hypoxia condition. Further investigation illustrated that PC₇₀ went through extremely long-life triplet state (211.3 μs) under hypoxia, which enabled the exiguous oxygen to approach and interact with the activated (³P-C₇₀)* more efficiently and produce more singlet oxygen. This would overcome the problems of existing photosensitizer as low PDT efficiency in cancerous tissue under hypoxia. The excellent properties of PC₇₀ dyad would make it promising phototherapeutic agents especially for the treatment of early- and late-stage cancers under the shallow and hypoxia tissues.

Introduction

Photodynamic therapy (PDT) has recently attracted much attention because of its fine controllability, improved selectivity, and low systemic toxicity.¹ Photosensitizers (PSs) as PDT agents could react with oxygen in tissues upon light irradiation to generate reactive oxygen species (ROS), which is highly toxic to tumor cells.² However, current applications of PDT agents such as hematoporphyrin,³ 5-Aminolevulinic acid (5-ALA),⁴ tin ethyl etiopurpurin (SnET₂),⁵ mono-L-aspartylchlorin e6 (Npe6),⁶ are generally restricted by their drawbacks of poor solubility and photostability, low production of singlet oxygen (¹O₂) at tumor hypoxic microenvironment.⁷ Therefore, it is still imperative to search the desired water-dispersible, highly photostable and efficient PDT agents, especially with high activity and biocompatibility under hypoxia tumor microenvironment.

Fullerene based nanomaterial has been applied in many areas, such as photoelectric device,⁸ solar cells,⁹ and MRI contrast agents.¹⁰ Furthermore, a new application of fullerene derivatives as excellent PS for PDT¹¹ was under studied due to its unique properties including good biocompatibility, facile multiple surface functionalization, efficient generation of ROS under visible light illumination and total metabolism from living organism.¹² Very recently, we reported a carboxylic acid functionalized fullerene which exhibited excellent PDT property under white-light illumination.¹³ As we know, the photodynamic activity of fullerenes can be further enhanced by conjugation with dye molecules¹⁴ (e.g. porphyrin or phthalocyanine, which are standards of traditional PSs and have a natural proclivity to accumulate in cancer tissues.¹⁵ Notably,

activity of most PDT agents based on the concentration of the oxygen in the tissues,¹⁶ while the hypoxic tumor microenvironment with low oxygen level brought down the PDT efficiency of these PSs. Viviana Rivarola's group reported a PS (porphyrin-C₆₀) that could produce singlet oxygen under anaerobic condition, however, the synthesized porphyrin-C₆₀ exhibited poor solubility and further formulation process with liposome was required for application.¹⁷

Herein, we prepared an amphiphilic photosensitizer, trimethylpyridylporphyrin-C₇₀ (PC₇₀) dyad, which could form a ring structure like liposome by self-assemble. The as-prepared PC₇₀ showed enhanced absorption cross-section in visible region, good water-dispersibility, high photostability, and favorable biocompatibility. Significantly, the present PC₇₀ exhibited extraordinary photodynamic effect even under hypoxic condition upon light irradiation. These excellent properties enable PC₇₀ function as an improved PDT agent at low level of oxygen. Moreover, a possible mechanism has been proposed by investigating the transient absorption spectroscopy of PC₇₀ according to laser flash photolysis.

Results and discussion

Characterization of PC₇₀ and D-TMPyP

D-TMPyP was prepared according to established procedure,¹⁸ and the synthetic route of D-TMPyP was outlined in Scheme S1 (SI). D-TMPyP reacted with C₇₀ through one step reaction to achieve C₇₀-TMPyP according to our previously reported method.¹⁹ After purification by flash chromatography, the chemical structure of products were confirmed by ¹H NMR

(DMSO-*d*₆) and MALDI-TOF-MS (α -cyano-4-hydroxy cinnamic acid as the matrix) (Fig.S1-S7). The obtained final product was mixed with methyl tosylate and refluxed in argon atmosphere and then passed through an anion-exchange resin repeatedly to yield Trismethylpyridylporphyrin-C₇₀ as a chloride salt (PC₇₀). The absorption of PC₇₀ and D-TMPyP were investigated (Fig. 1A). The molar absorption coefficient of Soret band of PC₇₀ was less than that of D-TMPyP, and behaved a slightly red-shifted, which indicated that a partial of electron density migrated from the D-TMPyP ring to the fullerene entity. As a result, the prolonged lifetime of triplet of PC₇₀ was observed as shown below. Fluorescence spectra were investigated in Fig S8-S9.²⁰ The fluorescence intensity of PC₇₀ was decreased compared to D-TMPyP which indicated the interaction between D-TMPyP and C₇₀. Under irradiation of white light at 17 mW·cm⁻² for 10, 30, 60, 90, 120 or 150 min, the absorbance intensity of PC₇₀ decreased slowly, while that of both D-TMPyP and PpIX decreased dramatically (Fig. 1B). The result indicated that PC₇₀ possessed better photostability than D-TMPyP and PpIX. In addition, the ¹O₂ quantum yield of the PC₇₀ was measured to be ca. 42% at white light (Fig. S10) with Rose Bengal as a standard photosensitizer.

The diameters of the D-TMPyP and PC₇₀ were measured by dynamic light scattering (Nano-ZS ZEN3600, Malvern Instruments, Germany). Hydrodynamic size of PC₇₀ (ca. 90 nm) is smaller than that of D-TMPyP (ca. 220 nm) (Fig. 1C). This size would facilitate the PC₇₀ to be taken up by cells. In addition, zeta potentials (ζ) of PC₇₀ and D-TMPyP were also measured (Fig. 1D). Only minute difference of ζ -potential of D-TMPyP and PC₇₀ was observed. The appropriate size and zeta potential may lead to cellular uptake toward nanoparticles more easily.²¹

Most interestingly, the self-assembling of amphiphilic PC₇₀ forms a ring circle structure similar to liposome, which was identified by transmission electron microscope (Fig. S11). The formed assembling structure exhibits diameter ca. 30 nm and may be further developed for drug delivery.

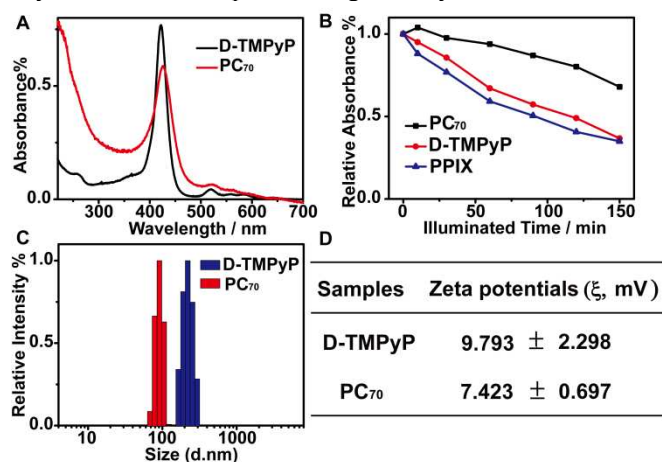


Fig. 1 (A) UV-Vis absorption Spectra of PC₇₀ and D-TMPyP in water; (B) The relative absorbance of PC₇₀, D-TMPyP and PpIX exposed to light irradiation for 10, 30, 60, 90, 120 and 150 min, respectively, at a power density of 17 mW·cm⁻²; (C) Size distribution of PC₇₀ and D-TMPyP; (D) Zeta potential of PC₇₀ and D-TMPyP (pH = 7.4).

Comparison of Cellular Uptake of PC₇₀ and D-TMPyP

The cellular uptake of sensitizer was evaluated by incubating 10 μ M of PC₇₀ with A549 cells at various time points.¹⁷ The

uptake of D-TMPyP by A549 cells was parallel tested as control group. In each case, the concentration of intracellular sensitizer was measured by UV-Vis absorption. PC₇₀ was rapidly internalized into A549 cells in 3 h and reached a plateau after incubation for 8 h (Fig. 2A). The maximum uptake value of PC₇₀ (~20 nmol·10⁶ cells) was three fold of that of D-TMPyP. This is due to the amphipathic property of PC₇₀ that make it more easily across the cytomembrane.

Another experiment was explored to study the mechanism of cellular uptake of PC₇₀ nanoparticles by incubating the A549 cells with PC₇₀ and ionic D-TMPyP at 4°C and 37°C, respectively. The result was shown in Fig. 2B, both PC₇₀ and D-TMPyP showed more cell uptaken at 37°C rather than 4°C. It was suggested that the cellular uptake of PC₇₀ and D-TMPyP nanoparticles was energy-dependent.²² In particularly, PC₇₀ and D-TMPyP showed a selective uptake toward A549 cells in comparison with HaCaT cells (Fig. 2C), which may contribute to the tumor-targeting of D-TMPyP moiety.²³

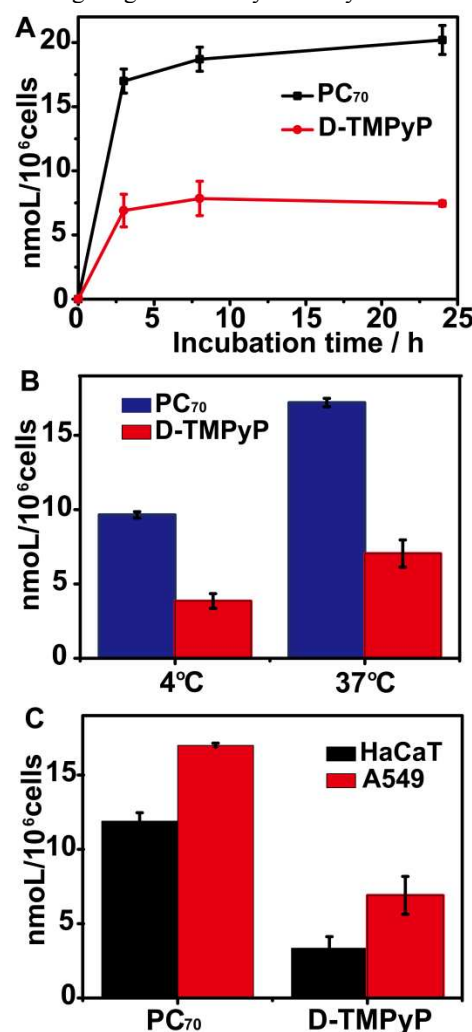


Fig. 2 (A) The comparison of uptake of PC₇₀ and D-TMPyP into A549 cells at different incubation time; (B) Comparison of uptake of PC₇₀ and D-TMPyP for 3 h incubation at 4°C and 37°C; (C) The comparison of uptake of PC₇₀ and D-TMPyP into A549 cells and HaCaT cells, respectively. Values represent mean \pm S.D. of three separated experiments.

Intracellular Location study of PC₇₀ in A549 cells

To confirm that PC₇₀ was internalized into cells, the localization of PC₇₀ in A549 cells was characterized by laser scanning confocal microscopy, and D-TMPyP was used as a control. Both of PC₇₀ and D-TMPyP emit red fluorescence upon excitation at 405 nm. As shown in Fig. 3A, the red fluorescence was distributed in most of cytoplasm which indicated that PC₇₀ mainly located in cytoplasm but not nonspecifically attached to the cell membrane. Whereas, red fluorescence of D-TMPyP significantly quenched (Fig. 3B) indicated that little D-TMPyP molecules were uptaken by A549 cells. This result coincided with the previously reported. For further confirmation, transmission electron microscopy (H-7650 TEM Hitachi Ltd., Japan) was carried out to characterize the intracellular distribution of PC₇₀. As shown in Fig. 3D, PC₇₀ (indicated by white arrows) was phagocytized into cell as a small cluster and mostly localized in the cytoplasm. In contrast, the control without treated with PC₇₀ didn't show any nanoparticles (Fig. 3C). The favorable uptake toward cancerous cells of PC₇₀ would facilitate its PDT efficiency.

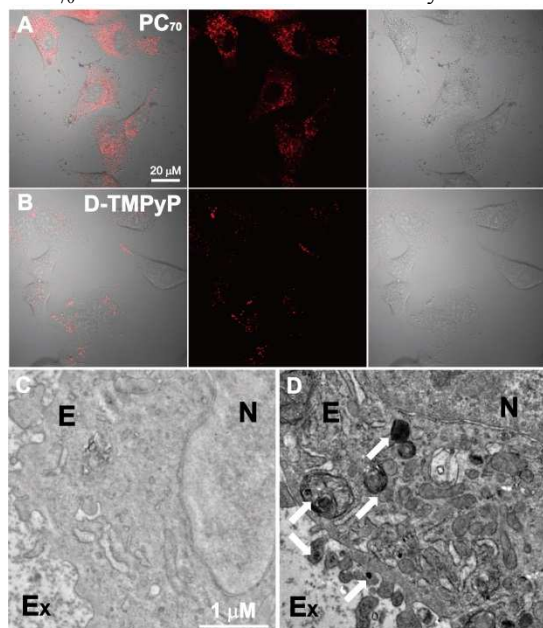


Fig. 3 (A) Confocal images of PC₇₀ uptake into A549 cells after 3 h incubation; (B) Confocal images of D-TMPyP treated under the same condition; the images of (A) and (B) from left to right represent merged fluorescence and optical images, fluorescence images and optical images, respectively. (C) TEM image of A549 cells without treatment of PC₇₀; (D) TEM image of A549 cells incubated with PC₇₀. N = nuclear region, E = endosome, Ex = extracellular region.

The Photodynamic Activity of PC₇₀ under Air Condition

According to the cellular uptake results, PC₇₀ was internalized into cells after incubation for 3 h. When upon light irradiation, PC₇₀ with extended π -conjugated system facilitated visible light absorption in the range from 400 to 700 nm, and produced ROS to kill cells. Importantly, the dark cytotoxicity of PC₇₀ was negligible for both cancer cells and normal cells, indicating the favorable biocompatibility of PC₇₀ (Fig. 4A). To investigate the photodynamic activity of PC₇₀ against cancer cell, we studied the viability of A549 cells *in vitro*. After exposure to white light for 5 min, 10 min and 20 min at a power density of 17 mW·cm⁻², the cellular viability was detected. The result indicated that both of PC₇₀ and D-TMPyP (Fig. S12) showed strong inhibitory

effects on cell viability even at low concentration of 1 μ M under light irradiation (Fig. 4B), the killing efficiency can reach 98%. As we expect, cellular toxicity could be improved by extending the irradiation time and increasing the PC₇₀ concentration (Fig. 4C). That's to say, the photodynamic activity of the PC₇₀ was dose- and irradiation time-dependent. Interestingly, the cell death was accompanied by membrane bleb. As shown in Fig. 4D, blebs appeared on several cells after irradiation for 1 min. During irradiation last for 5 min, blebs appeared on most of cellular membranes, and blebs appeared on nuclear membranes after irradiation for 10 min (Fig. S13). This results was similar to our previous reported.^{13a}

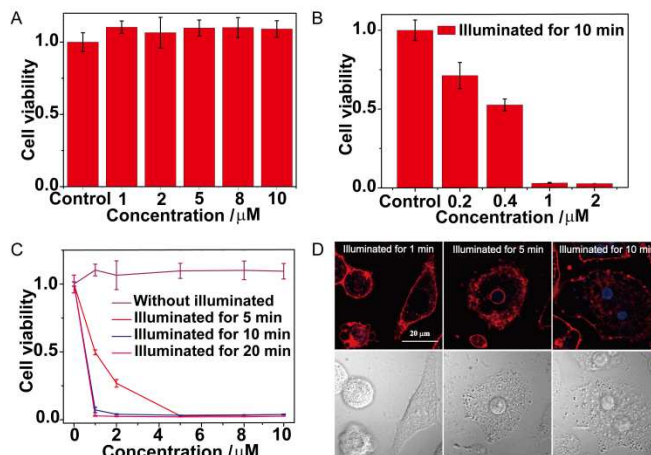


Fig. 4 (A) Cell viability of A549 cells incubated with different concentrations of PC₇₀ for 24 h in dark; (B) Cell viability of A549 cells incubated with PC₇₀ at gradient concentrations for 3 h and exposed to light irradiation for 10 min at a power density of 17 mW·cm⁻²; (C) Dose- and time- dependent PDT effects of PC₇₀ on the A549 cell viability; (D) Confocal images of A549 cells stained with Dil and Hoechst 33258 after treated with 2 μ M of PC₇₀ for 3 h and exposed to light irradiation for 1, 5 and 10 min, respectively. The above images represent fluorescence images and those at bottom represent optical images. Values represent mean \pm S.D. of three separated experiments.

The Photodynamic Activity of PC₇₀ under Nitrogen Condition

Most of PDT efficiency relies on the concentration of oxygen around the tumor tissues, however, the tumor microenvironment is usually hypoxic and acidic.²⁴ As a result, the hypoxic tumor microenvironment significantly hindered the efficiency of the PDT. In order to illustrate thoroughly the PDT efficiency of PC₇₀ under hypoxia, we further investigated the PDT efficiency of PC₇₀ under anaerobic condition with white light irradiation. After incubating with either PC₇₀ or D-TMPyP, A549 cells were exposed to light under nitrogen atmosphere. As shown in Fig. 5A, A549 cells in the presence of PC₇₀ showed a severe photodamage (80%), in contrast, D-TMPyP only showed limited damage (22%) to A549 cells. This result demonstrated that the PDT efficiency of PC₇₀ was higher than that of D-TMPyP in hypoxic environment. Furthermore, confocal microscopy experiment was carried out. When the A549 cells were incubated with PC₇₀, PI stained the cell nucleus transmembrane with red fluorescence, demonstrating that the cells damage severely (Fig. 5B). Meanwhile, most of A549 cells treated with D-TMPyP were not stained by PI, indicating that D-TMPyP didn't show significant cytotoxicity to A549 cells under the anaerobic condition. The remarkable

cytotoxicity of PC₇₀ to tumor cells makes it potential as an excellent photosensitizer, especially under anaerobic condition.

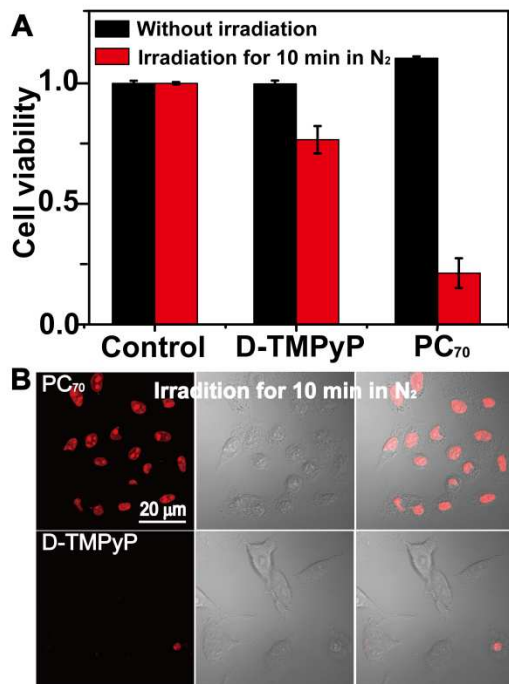


Fig. 5 (A) The comparison of Cell viability of A549 cells incubated with either PC₇₀ or D-TMPyP for 3 h and subsequent irradiation for 10 min at a power density of 17 mW·cm⁻² in nitrogen atmosphere. Cells without irradiation were used as a control. (B) Confocal images of A549 cells stained with PI after treated with 2 μM PC₇₀ for 3 h and subsequently exposed to light irradiation for 10 min at a power density of 17 mW·cm⁻² in nitrogen atmosphere. Cells treated with D-TMPyP were used as a control. Values represent mean ± S.D. of three separated experiments.

Mechanistic Study of Super Photodynamic Activity under Nitrogen Condition

In order to understand why PC₇₀ is more effective to kill cells than D-TMPyP under nitrogen condition, a series of experiments were carried out. Free radical species detection was executed to identify which kind of ROS was produced by PC₇₀ under light irradiation at the presence of different ROS scavenger.²⁵ A549 cells were incubated with PC₇₀ (2 μM) in dark for 3 h, then the medium were replaced by DMEM without phenol red containing either 10 mM sodium azide (NaN₃), or mannitol, or SOD (50 units). Then cells were irradiated under white light for 10 min. The results in Fig. 6A and 6B indicated that both of PC₇₀ and D-TMPyP induced cell death as a result of the generation of ¹O₂ under air condition. The photo irradiation experiments at presence of different ROS scavenger were also carried out under nitrogen condition. The results were similar to that of under air-saturated conditions (Fig. 6C and 6D). Cells incubation with NaN₃, mannitol, and SOD were used as control (Fig. S14).

To quantify the photosensitizing ability of PC₇₀ and D-TMPyP in producing singlet oxygen, the electron spin-resonance (ESR) spectroscopy was measured under air condition and nitrogen condition. 2,2,6,6-Tetramethylpiperidin (TEMP) was used as a radical scavenger.²⁶ As shown in Fig. S15A and S15B, under air condition little difference of PC₇₀ and D-TMPyP was observed. However, after purging with nitrogen for 20 min, the

ESR intensity treated with PC₇₀ is reduced to half of the original. Meanwhile, that treated with D-TMPyP nearly disappeared (Fig. S15C and S15D), suggesting that the PC₇₀ produce ROS at lower oxygen level more efficiently than D-TMPyP under hypoxic condition.

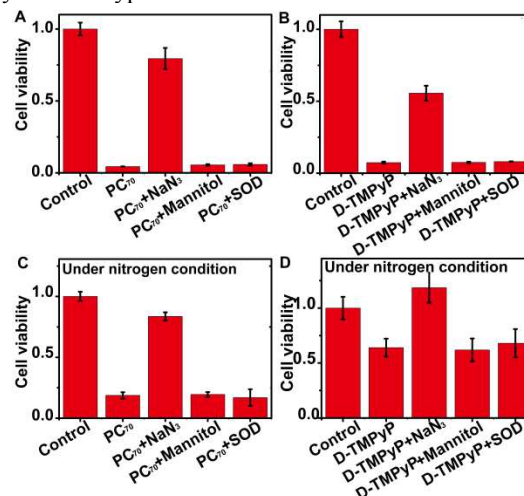


Fig. 6 (A) Viability of A549 cells incubated with 2 μM PC₇₀ for 3 h in presence of different ROS quenchers upon light irradiation for 10 min. (B) Viability of A549 cells incubated with 2 μM of D-TMPyP for 3 h in presence of different ROS quenchers upon light irradiation for 10 min. (C) Viability of A549 cells incubated with 2 μM PC₇₀ for 3 h in presence of different ROS quenchers upon light irradiation for 10 min under nitrogen condition. (D) Viability of A549 cells incubated with 2 μM D-TMPyP for 3 h in presence of different ROS quenchers upon light irradiation for 10 min under nitrogen condition. Values represent mean ± S.D. of three separated experiments.

The transient absorption spectroscopy based on laser flash photolysis was applied to detect the triplet excited state lifetime of D-TMPyP and PC₇₀, which was associated with the photosensitizing efficiency of D-TMPyP and PC₇₀ in producing singlet oxygen. As shown in Fig. 7, the temporal decay for the triplet absorption band at 480 nm was monitored.²⁷ Under air-saturated condition, as shown in Fig. 7A, the triplet lifetime of PC₇₀ (1.86 μs ± 0.01 μs) was slightly longer than that of D-TMPyP (1.64 μs ± 0.01 μs) under air-saturated condition. So there is no significant difference on the behavior of cell damage between PC₇₀ and D-TMPyP under air-saturated condition. However, under N₂-saturated condition, distinctive triplet decay behavior was exhibited for PC₇₀ and D-TMPyP, as shown in Fig. 7B. Unlike D-TMPyP triplet which had a single-exponential decay lifetime of 71.3 μs ± 0.5 μs, the PC₇₀ triplet followed a second-order exponential decay with lifetime of 62.8 μs ± 4.8 μs (32%) and 211.3 μs ± 7.0 μs (68%). The relaxation pathways of PC₇₀ triplet state, as illustrated in Fig. 7C, can reasonably explain the double-exponential result. The shorter lifetime of 62.8 μs, which was similar to 71.3 μs of D-TMPyP, was assigned to the decay of triplet state itself. In contrast, the longer lifetime of 211.3 μs can be assigned to the exciplex formed by energy transfer between excited D-TMPyP and ground state C₇₀. The formation of exciplex may be due to the longer lifetime of triplet which has larger probability formed at low oxygen concentration. It was the formation of exciplex that increased the triplet lifetime of PC₇₀. Under low oxygen concentration condition, the diffusion rate of oxygen

was low, the elongated triplet lifetime of PC₇₀ made it still possible to photosensitize the ground state of oxygen to produce singlet oxygen, whereas the shorter-lived triplet state of D-TMPyP had much less efficiency in photosensitization. This could explain why PC₇₀ retains PDT ability to kill cells under low oxygen concentration condition.

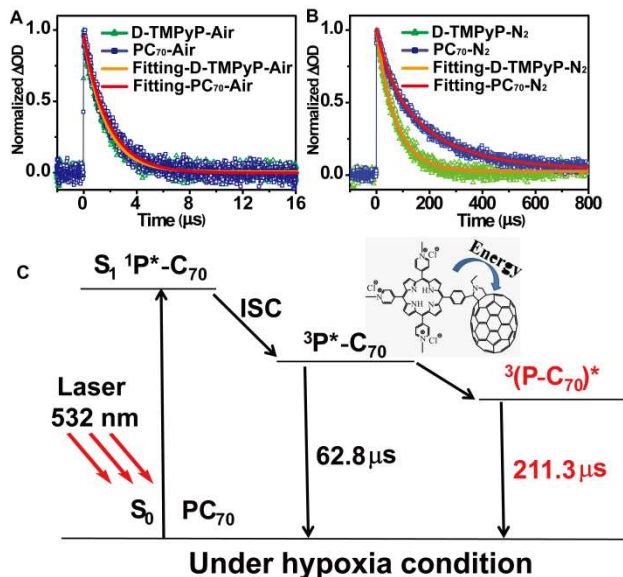


Fig. 7 The transient UV-visible absorption spectroscopies of D-TMPyP and PC₇₀ under different conditions. (A) Under air-saturated condition and (B) Under N₂-saturated condition. (C) The illustrated relaxation pathways of PC₇₀ triplet state.

Conclusions

In conclusion, the successfully synthesized photosensitizer (PC₇₀) based on trimethylpyridylporphyrin-C₇₀ dyad was investigated. The as-prepared PC₇₀ performed good water dispersibility, high photostability, and favorable biocompatibility. Significantly, the present PC₇₀ exhibits extraordinary photodynamic effects even under the hypoxia condition upon light irradiation. These excellent properties enable PC₇₀ facilitate the PDT of cells with lower level of oxygen. Further investigation illustrated that PC₇₀ possessed extremely long-life triplet state (211.3 μs) under hypoxia condition, which enabled the exiguous oxygen had enough diffusion time to reach the activated (³P-C₇₀)^{*} and interacted with each other to produce singlet state oxygen. This find may make PC₇₀ promising as an improved PDT agent and especially can be used in the treatment of early- and late-stage cancers under shallow and hypoxia tissues.

Experimental section

Preparation and Characterization of the water-soluble PC₇₀ and the water-soluble Trimethylpyridylporphyrin

Experimental details of the synthetic procedures and analysis of characterization are given in the Supporting Information.

¹O₂ quantum yield measurement via a chemical method.²⁸

Water soluble disodium 9, 10-anthracendipropionic acid (Na₂-ADPA) was used as a ¹O₂ trapping agent, and Rose Bengal (RB) was used as a standard photosensitizer. In the experiment, 100 μL of Na₂-ADPA solution (1 mg/mL) was added into 1.5 mL of

PC₇₀ solution. The resulting solution was irradiated under white light (400-700 nm) at a power density of 10 mW/cm². To eliminate the inner-filter effect, the absorption maxima of RB and PC₇₀ were adjusted to about 0.2 OD. The absorption of Na₂-ADPA at 380 nm was recorded at different irradiation time to get the decay rate of photosensitizing process. The ¹O₂ quantum yield of PC₇₀ in water (ΦPC₇₀) was calculated using the formula below.

$$\Phi_{PC_{70}} = \Phi_{RB} \times K_{PC_{70}} \times A_{RB} / (K_{RB} \times A_{PC_{70}})$$

Where, K_{PC₇₀} and K_{RB} are the decomposition rate constants of Na₂-ADPA by PC₇₀ and RB, respectively. A_{PC₇₀} and A_{RB} represent light absorption by PC₇₀ and the RB, respectively, which are determined by integration of the optical absorption bands in the wavelength range 400-700 nm. Φ_{RB} is the ¹O₂ quantum yield of RB, and Φ_{RB} = 75% in water. The measured ¹O₂ quantum yield of PC₇₀ in water is ca. 42%.

Cell Culture

The A549 cells and HaCaT cells were cultured with Dulbecco's modified Eagle's medium (DMEM) (Invitrogen, USA) supplemented with 10% fetal bovine serum (Hyclone Company, South Logan, UT), penicillin (100 μg·mL⁻¹), and streptomycin (100 μg·mL⁻¹) (Gibco, Grand Island, N. Y. USA) in 5% CO₂ at 37°C in a humidified incubator.

Assays of cellular uptake of PC₇₀ and D-TMPyP

As mentioned above, an appropriate number of A549 cells (ca. 5 × 10⁵ cells·cm⁻²) were incubated with PC₇₀ (10 μM) in the culture flasks at different incubation time points, such as 3 h, 8 h and 24 h in dark at 37°C. Then the culture medium was removed and the cells were washed several times with icy PBS solution. The cellular uptake was determined by adding 1.0 mL of 4% sodium dodecyl sulphate (SDS, Merck) to the culture flasks and subsequent incubating for 15 min in the dark at room temperature. The concentration of PC₇₀ was evaluated by measuring the UV-vis absorbance at 428 nm. The concentration of D-TMPyP was examined under the same conditions but measured the UV-vis absorbance at 422 nm. Each experiment was compared with a culture control without photosensitizer. The same experiment was carried out to compare the cellular uptake of PC₇₀ toward A549 cells and HaCaT cells through incubating 3 h in dark at 4°C.

TEM Observation

A549 cells incubated with PC₇₀ (2 μM) for 3 h and washed with Hank's balanced salt solution (HBSS) to remove the noninternalized nanoparticles. The obtained cells were fixed with 4% glutaraldehyde and 4% paraformaldehyde overnight at 4°C respectively. After that, cells were centrifuged at 1000 rpm for 3 min, and washed with PBS (1 mM) for three times. Afterwards the cells were post-fixed in 1% osmium tetroxide for 2 h at room temperature in 1 mm³ masses. After dehydrated by a graded series of ethanol and acetone the cells were embedded in Epon812. Ultramicrotome (UC6, Leica Ltd. Co, Germany) were used to cut ultrathin sections and transferred onto 200-mesh copper grids, which were stained with uranyl acetate and lead nitrate, then observed with a H-7650 TEM instrument (Hitachi Ltd., Japan).

PDT treatment and cell viability assay

Before cultured in 96-well plate for 24 h, cells were counted using a cell counter to control the cell density at ca. 5 × 10⁴ cells·cm⁻², and then incubated with PC₇₀ for 3 h in dark at 37°C.

After removing the culture medium, DMEM was added into the plate. The cells were subsequently irradiated with the M-visual light source (MVL-210, MEJIRO GENOSSEN, Japan) at 17 mW·cm⁻² for 10 minutes. Then fresh culture medium was added to replace the aged one and cultivated the cells for 24 h in the dark at 37°C. The same procedure without irradiation was carried out to determine the dark toxicity. Cytotoxicity was evaluated by a WST-8 assay with a Cell counting Kit-8 (CCK-8; DOJINDO, Kumamoto, Japan) which has characteristic absorbance at 450 nm. This can be read with a 96-well plate reader (iMarkmicroplate reader, Bio-RAD, USA) to determine the cell viability. During the experiment, the D-TMPyP was used as a control.

Confocal images after staining with Dil and Hoechst 33258

First, A549 cells were incubated with different concentration of PC₇₀ and D-TMPyP at 37°C, 5% CO₂ for 3 h, respectively. Then aged cell culture was replaced with fresh DMEM (without phenol red) and exposed to white light for 1 min, 5 min and 10 min, respectively, at a power density of 17 mW·cm⁻², the cellular viability was detected. After that, either Hoechst 33258 (1 mg·mL⁻¹) or Dil (1 mg·mL⁻¹) was added and incubated with the cells for 20 min before observation. The image was obtained by a FV 1000-IX81 confocal laser scanning microscope (Olympus, Japan).

Inhibition of ROS Generation

A549 cells with a cell density of ca. 5 × 10⁴ cells·cm⁻² were cultured in a 96-well plate for 24 h. To inhibit the ROS generation, A549 cells were incubated with PC₇₀ (2 μM) in dark for 3 h, then the aged medium was replaced by fresh medium containing either 10 mM sodium azide (NaN₃), or mannitol, or SOD (50 units) and cells were irradiated under white light for 10 min. After that, the medium containing free-radical scavenger was replaced. 1 day later, the cell viability was detected by CCK-8. Under the same conditions, several ROS scavenger (NaN₃, singlet oxygen (¹O₂) quencher, mannitol, a hydroxyl radical (·OH) quencher, and superoxide dismutase (SOD), a superoxide anion free radical (O₂^{·-}) quencher) were used as controls.

Evaluation of radical scavenging ability by ESR spectroscopy:

Studies on radical scavenging by PC₇₀ (10 μM) were performed by spin trapping of 2,2,6,6-Tetramethylpiperidin (TEMP). ESR spectra were measured with a Bruker ELEXSYSE 500 ESR spectrometer at 25°C. Singlet oxygen was generated from light irradiation (50 mW/cm²). D-TMPyP (10 μM) was used as a control. After purging with nitrogen for 20 min, the ESR intensity of TEMP containing either PC₇₀ or D-TMPyP was also detected.

PDT treatment under anaerobic condition

A549 cells were incubated with either PC₇₀ (2 μM) or D-TMPyP (2 μM) in dark for 3 h at 37°C as previously described but in nitrogen atmosphere, and then exposed to light irradiation for 10 min. The medium was replaced by the fresh DMEM (without phenol red). Cell viability was detected by CCK-8 after one day incubation. After removing the medium and washing by PBS for several times. The cells were stained with propidium iodide (PI) (8 μM) for 15 min before observation. The images were obtained by n FV 1000-IX81 confocal laser scanning microscope (Olympus, Japan).

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Notes and references

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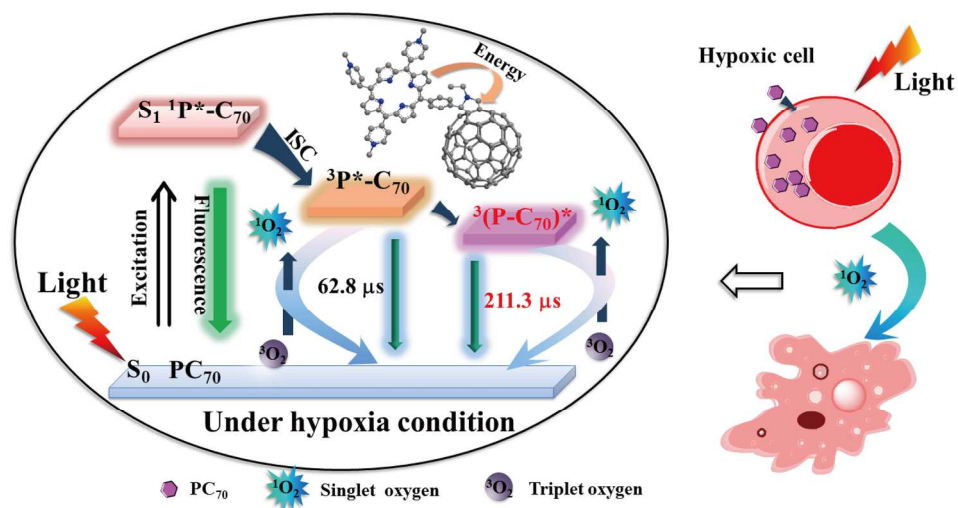
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- a) D. Dolmans, D. Fukumura, R. Jain, *Nat. Rev. Cancer* 2003, **3**, 380;
 - b) F. Schmitt, J. Freudenreich, N. Barry, L. Juillerat-Jeanneret, G. Suss-Fink, B. Therrien, *J. Am. Chem. Soc.*, 2012, **134**, 754; c) Y. Cakmak, S. Kolemen, S. Duman, Y. Dede, Y. Dolen, B. Kilic, Z. Kostereli, L. Yildirim, A. Dogan, D. Guc, E. Akkaya, *Angew. Chem. Int. Ed.*, 2011, **50**, 11937; d) J. Barreto, W. O'Malley, M. Kubeil, B. Graham, H. Stephan, L. Spiccia, *Adv. Mater.*, 2011, **23**, H18; e) J. Celli, B. Spring, I. Rizvi, C. Evans, K. Samkoe, S. Verma, B. Pogue, T. Hasan, *Chem. Rev.*, 2010, **110**, 2795; f) L. Li, M. Nurunnabi, M. Nafujjaman, Y. Jeong, Y. Lee, K. Huh, *J. Mater. Chem. B*, 2014, **2**, 2929.
- a) Z. Zhu, Z. Tang, J. Phillips, R. Yang, H. Wang, W. Tan, *J. Am. Chem. Soc.*, 2008, **130**, 10856; b) E. Shibu, S. Sugino, K. Ono, H. Saito, A. Nishioka, S. Yamamura, M. Sawada, Y. Nosaka, V. Biju, *Angew. Chem. Inter. Ed.*, 2013, **52**, 10559; c) C. Schweitzer, R. Schmidt, *Chem. Rev.*, 2003, **103**, 1685; d) S. Jin, L. Zhou, Z. Gu, G. Tian, L. Yan, W. Ren, W. Yin, X. Liu, X. Zhang, Z. Hu, Y. Zhao, *Nanoscale*, 2013, **5**, 11910; e) H. Kim, S. Mun, Y. Choi, *J. Mater. Chem. B*, 2013, **1**, 429.
- Y. Ren, R. Wang, Y. Liu, H. Guo, X. Zhou, X. Yuan, C. Liu, J. Tian, H. Yin, Y. Wang, *Biomaterials*, 2014, **35**, 2462.
- a) G. Berkovitch, D. Doron, A. Nudelman, Z. Malik, A. Rephaeli, *J. Med. Chem.*, 2008, **51**, 7356; b) F. Giuntini, L. Bourre, A. MacRobert, M. Wilson, I. Eggleston, *J. Med. Chem.*, 2009, **52**, 4026.
- M. Kaplan, R. Somers, R. Greenberg, et al. *J. Surg. Oncol.*, 1998, **67**, 121.
- a) A. Chan, M. Juarez, R. Allen, W. Volz, T. Albertson, *Photodermatol. Photo.*, 2005, **21**, 72; b) S. Taber, V. Finger, C. Coots, T. Wieman, *Clin. Cancer Res.*, 1998, **4**, 2741.
- a) S. Wang, P. Huang, L. Nie, R. Xing, D. Liu, Z. Wang, J. Lin, S. Chen, G. Niu, G. Lu, X. Chen, *Adv. Mater.*, 2013, **25**, 3055; b) S. Walenta, M. Wetterling, M. Lehrke, G. Schwickert, K. Sundfor, E. Rofstad, W. Mueller-Klieser, *Cancer Res.*, 2000, **60**, 916.
- a) H. Li, B. Tee, J. Cha, Y. Cui, J. Chung, S. Lee, Z. Bao, *J. Am. Chem. Soc.*, 2012, **134**, 2760; b) S. Sato, H. Nikawa, S. Seki, L.

- Wang, G. Luo, J. Lu, M. Haranaka, T. Tsuchiya, S. Nagase, T. Akasaka, *Angew. Chem. Int. Ed.* 2012, **51**, 1589.
- 9 a) X. Meng, W. Zhang, Z. Tan, Y. Li, Y. Ma, T. Wang, L. Jiang, C. Shu, C. Wang, *Adv. Funct. Mater.*, 2012, **22**, 2187; b) X. Meng, G. Zhao, Q. Xu, Z. Tan, Z. Zhang, L. Jiang, C. Shu, C. Wang, Y. Li, *Adv. Funct. Mater.*, 2014, **24**, 158.
- 10 a) C. Shu, C. Wang, J. Zhang, H. Gibson, H. Dorn, F. Corwin, P. Fatouros, T. Dennis, *Chem. Mater.*, 2008, **20**, 2106; b) R. Bolskar, A. Benedetto, L. Husebo, R. Price, E. Jackson, S. Wallace, L. Wilson, J. Alford, *J. Am. Chem. Soc.*, 2003, **125**, 5471.
- 11 a) J. Fan, G. Fang, F. Zeng, X. Wang, S. Wu, *Small*, 2013, **9**, 613; b) R. Mendes, A. Bachmatiuk, B. Buchner, G. Cunibertib, M. Rummeli, *J. Mater. Chem. B*, 2013, **1**, 401; c) W. Jang, D. Yim, I. Hwang, *J. Mater. Chem. B*, 2014, **2**, 2202; d) Y. Yamakoshi, N. Umezawa, A. Ryu, K. Arakane, N. Miyata, Y. Goda, T. Masumizu, T. Nagano, *J. Am. Chem. Soc.* 2003, **125**, 12803; e) P. Padmawar, T. Canteenwala, S. Verma, L. Tan, G. He, P. Prasad, L. Chiang, *Chem. Mater.* 2006, **18**, 4065.
- 12 a) F. Giacalone, N. Martin, *Chem. Rev.* 2006, **106**, 5136; b) A. Ikeda, M. Akiyama, T. Ogawa, T. Takeya, *ACS Med. Chem. Lett.* 2010, **1**, 115; c) D. McCluskey, T. Smith, P. Madasu, C. Coumbe, M. Mackey, P. Fulmer, J. Wynne, S. Stevenson, J. Phillips, *ACS Appl. Mater. Interfaces*, 2009, **1**, 882; d) E. Otake, S. Sakuma, K. Torii, A. Maeda, H. Ohi, S. Yano, A. Morita, *Photochem. Photobiol.*, 2010, **86**, 1356; e) M. Wang, S. Maragani, L. Huang, S. Jeon, T. Canteenwala, M. Hamblin, L. Chiang, *Eur. J. Med. Chem.* 2013, **63**, 170; f) P. Mroz, A. Pawlak, M. Satti, H. Lee, T. Wharton, H. Gali, T. Sarna, M. Hamblin, *Free Radic. Biol. Med.* 2007, **43**, 711.
- 13 a) Q. Liu, M. Guan, L. Xu, C. Shu, C. Jin, J. Zheng, X. Fang, Y. Yang, C. Wang, *Small*, 2012, **8**, 2070; b) Q. Liu, L. Xu, X. Zhang, N. Li, J. Zheng, M. Guan, X. Fang, C. Wang, C. Shu, *Chem. Asian J.* 2013, **8**, 2370.
- 14 a) X. Liu, M. Zheng, X. Kong, Y. Zhang, Q. Zeng, Z. Sun, W. Buma, H. Zhang, *Chem. Commun.* 2013, **49**, 3224; b) X. Wang, C. Yang, J. Chen, X. Yan, *Anal. Chem.* 2014, **86**, 3263; c) G. He, L. Tan, Q. Zheng, P. Prasad, *Chem. Rev.*, 2008, **108**, 1245.
- 15 a) T. Dougherty, J. Kaufman, A. Goldfarb, K. Weishaupt, D. Boyle, A. Mittleman, *Cancer Res.* 1978, **38**, 2628; b) R. Lipson, A. Olsen, *J. Natl. Cancer Inst.* 1961, **26**, 1; c) T. Dougherty, G. Grinnedy, R. Fiel, K. Weishaupt, D. Boyle, *J. Natl. Cancer Inst.* 1975, **55**, 115.
- 16 a) Q. Lin, C. Bao, Y. Yang, Q. Liang, D. Zhang, S. Cheng, L. Zhu, *Adv. Mater.* 2013, **25**, 1981; b) C. Jin, J. Lovell, J. Chen, G. Zheng, *ACS Nano.*, 2013, **7**, 2541; c) S. Wang, P. Huang, L. Nie, R. Xing, D. Liu, Z. Wang, J. Lin, S. Chen, G. Niu, G. Lu, X. Chen, *Adv. Mater.*, 2013, **25**, 3055.
- 17 M. Alvarez, C. Prucca, M. Milanesio, E. Durantini, V. Rivarola, *Int. J. Biochem. Cell Biol.*, 2006, **38**, 2092.
- 18 a), Y. Ishikawa, A. Yamashita, T. Uno, *Chem. Pharm. Bull.*, 2001, **49**, 287; b) K. Okuda, C. Abeta, T. Hirota, M. Mochizuki, T. Mashino, *Chem. Pharm. Bull.*, 2002, **50**, 985.
- 19 W. Xu, L. Feng, Y. Wu, T. Wang, J. Wu, J. Xiang, B. Li, L. Jiang, C. Shu, C. Wang, *Phys. Chem. Chem. Phys.* 2011, **13**, 428.
- 20 W. Lei, G. Jiang, Q. Zhou, Bao. Zhang and Xue. Wang, *Phys. Chem. Chem. Phys.* 2010, **12**, 13255.
- 21 M. Milanesio, M. Alvarez, J. Silber, V. Rivarola, E. Durantini, *Photochem. Photobiol. Sci.*, 2003, **2**, 926.
- 22 J. Li, F. Jiang, B. Yang, X. Song, Y. Liu, H. Yang, D. Cao, W. Shi, G. Chen, *Sci. Rep.*, 2013, **3**, 1998.
- 23 a) P. Kubat, K. Lang, P. Anzenbacher, *Biochim. Biophys. Acta.*, 2004, **1670**, 40; b) Y. Shibata, A. Matsumura, F. Yoshida, T. Yamamoto, K. Nakai, T. Nose, I. Sakata, S. Nakajima, *Cancer Lett.* 2001, **166**, 79; c) R. Song, Y. Kim, Y. Sohn, *J. Inorg. Biochem.* 2002, **89**, 83; d) A. Kano, Y. Taniwaki, I. Nakamura, N. Shimada, K. Moriyama, A. Maruyama, *J. Control Release* 2013, **167**, 315.
- 24 a) J. Brown, W. Wilson, *Nat. Rev. Cancer*, 2004, **4**, 437; b) X. Yue, C. Yanez, S. Yao, K. Belfield, *J. Am. Chem. Soc.*, 2013, **135**, 2112; c) P. Prasad, C. Gordijo, A. Abbasi, A. Maeda, A. Ip, A. Rauth, R. DaCosta, X. Wu, *ACS Nano*. 2014, **8**, 6510; d) N. Kong, K. Lin, H. Li, J. Chang, *J. Mater. Chem. B*, 2014, **2**, 1100.
- 25 a) S. Cherng, Q. Xia, L. Blankenship, J. Freeman, W. Wamer, P. Howard, P. Fu, *Chem. Res. Toxicol.* 2005, **18**, 129; b) R. Williams, T. Glinka, M. Flanagan, R. Gallegos, H. Coffman, D. Pei, *J. Am. Chem. Soc.*, 1992, **114**, 733; c) E. Rozhkova, I. Ulasov, B. Lai, N. Dimitrijevic, M. Lesniak, T. Rajh, *Nano. Lett.*, 2009, **9**, 3337; d) Y. Jin, J. Cowan, *J. Am. Chem. Soc.*, 2005, **127**, 8408; e) Q. An, C. Sun, D. Li, K. Xu, J. Guo, C. Wang, *ACS Appl. Mater. Interfaces*, 2013, **5**, 13248; f) S. Kruss, M. P. Landry, E. Vander Ende, B. Lima, N. Reuel, J. Zhang, J. Nelson, B. Mu, A. Hilmer, M. Strano, *J. Am. Chem. Soc.*, 2014, **136**, 713.
- 26 a) W. He, H. Kim, W. Wamer, D. Melka, J. Callahan, J. Yin, *J. Am. Chem. Soc.* 2014, **136**, 750; b) W. Zhang, X. Gong, C. Liu, Y. Piao, Y. Sun, G. Diao, *J. Mater. Chem. B*, 2014, **2**, 5107.
- 27 a) C. Zhang, J. Zhao, S. Wu, Z. Wang, W. Wu, J. Ma, S. Guo, L. Huang, *J. Am. Chem. Soc.*, 2013, **135**, 10566; b) F. Deschler, A. Sio, E. Hauff, P. Kutka, T. Sauermaann, H. Egelhaaf, J. Hauch, E. Como, *Adv. Funct. Mater.*, 2012, **22**, 1461.
- 28 J. Ge, M. Lan, B. Zhou, W. Liu, L. Guo, H. Wang, Q. Jia, G. Niu, X. Huang, H. Zhou, X. Meng, P. Wang, C. Lee, W. Zhang, X. Han, *Nat. Commun.*, 2014, **8**, 4596.



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