# Journal of Materials Chemistry C

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

*Accepted Manuscripts* are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](http://www.rsc.org/Publishing/Journals/guidelines/AuthorGuidelines/JournalPolicy/accepted_manuscripts.asp).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](http://www.rsc.org/help/termsconditions.asp) and the Ethical quidelines still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/materialsC

Cite this: DOI: 10.1039/c0xx00000x

## **ARTICLE TYPE**

www.rsc.org/xxxxxx

### **Encapsulating carbon nanotubes with SiO2: a strategy of applying them in polymer nanocomposites with high mechanical strength and electrical insulation**

## **Xiaoliang Zeng,a,b Shuhui Yu,<sup>a</sup> \* Lei Ye,<sup>c</sup> Mingyang Li,<sup>a</sup> Zhilong Pan,<sup>a</sup> Rong Sun,<sup>a</sup>\* and Jianbin Xu<sup>c</sup>** 5

#### *Received (in XXX, XXX) Xth XXXXXXXXX 20XX, Accepted Xth XXXXXXXXX 20XX*  **DOI: 10.1039/b000000x**

Multiwalled carbon nanotubes (MWCNTs) have been widely used as mechanical reinforcement fillers for polymer during the past decades. However, high electrical conductivity of MWCNTs hampers their <sup>10</sup>applications in some special fields. In this study, the MWCNT was encapsulated with an insulating silicon

- oxide (SiO<sub>2</sub>) layer to form core-shell structure MWCNT@SiO<sub>2</sub> nanoparticles, which were used to fill bismaleimide-triazine (BT) resin. The obtained polymer nanocomposites possessed high mechanical strength, electrical insulation, improved thermal stability, and good optical transparency. These excellent properties were attributed to the strong interfacial interaction between  $MWCNT@SiO<sub>2</sub>$  and polymer, as
- $\mu$ s well as the suppression of electron transport by SiO<sub>2</sub> layer on MWCNT surface. The nanocomposites were employed to fabricate a printed circuit substrate, on which a frequency "flasher" circuit and the electrical components worked well. This work has demonstrated the possibility of using MWCNTs as mechanical reinforcement fillers in polymer nanocomposites, which simultaneously possess electrical insulation.

#### <sup>20</sup>**1. Introduction**

Polymer nanocomposites have received a great deal of attention both in science and engineering over recent decades in the development of advanced materials for a wide range of applications.<sup>1</sup> Carbon nanotubes (CNTs) have been considered to <sup>25</sup>be ideal reinforcement fillers for high-performance polymer nanocomposites due to their excellent mechanical properties, involving excellent Young's module with values greater than 1 TPa and tensile strength of  $63$  GPa.<sup>2</sup> A small addition amount of CNTs to polymers could result in superior mechanical properties,

 $30$  as previously reported.<sup>3-6</sup> On the other hand, owing to their intrinsic nature, the addition of CNTs simultaneously leads to enhanced electrical conductivity, even at low volume fractions<sup>7-11</sup>. The electrically conductive polymer nanocomposites are regarded as promising materials for use in sensors, actuators, and

35 electromagnetic shielding, etc.<sup>12-14</sup> However, high electrical conductivity hampers their application in some fields, such as printed circuit substrates, sealants for semiconductor devices and light emitting diodes, etc.

In order to utilize the outstanding mechanical strength of <sup>40</sup>CNTs while maintaining the electrical insulation of the polymer matrix, an effective solution is to cover the individual CNT with an insulating layer. 15-17 For example, Hayashida and coworkers

pioneered the design of polymer based CNTs nanocomposite with a high electrical resistance by generating an insulating polymer 45 layer on CNTs surface.<sup>18</sup> However, most of the former research <sup>19-</sup>

<sup>24</sup> mainly focused on the preparation, electrically and thermally conductive properties of the nanocomposites filled with insulated CNTs. The important properties including mechanical, dielectric, and optical properties, as well as the practical applications in 50 printed circuit substrates have been rarely investigated. Developing the CNTs filled polymer nanocomposites with high mechanical strength, electrical insulation and other excellent properties for practical applications still remains a challenge.

In this study, we developed a strategy for fabricating <sup>55</sup>functionalized MWCNTs filled polymer nanocomposites with high mechanical strength and electrical insulation. The Functionalized MWCNTs were prepared by encapsulating MWCNTs with  $SiO<sub>2</sub>$  and the process was driven by the formation of Si-O-C bonds. Bismaleimide-triazine (BT) resin was employed <sup>60</sup>as polymer matrix, which has been widely used in the printed circuit substrates due to its excellent thermal stability and good retention of mechanical properties at elevated temperatures.<sup>25, 26</sup> The results indicated that the obtained polymer nanocomposites possessed high mechanical strength, electrical insulation,  $65$  improved thermal stability, a low dielectric constant of  $~3.5$  (1) GHz) and good optical transparency. The nanocomposites have been successfully applied to fabricate a printed circuit substrate, on which a frequency "flasher" worked well. These unique properties were discussed based on the analysis of the microstructure and interfaces of the prepared nanocomposites.

#### **2. Experimental**

#### <sup>5</sup>**2.1 Materials**

Tetraethyl orthosilicate (TEOS), ammonia solution (NH4OH) (56 wt%), nitric acid (HNO<sub>3</sub>), absolute ethanol (CH<sub>3</sub>CH<sub>2</sub>OH) and methyl ethyl ketone were acquired from Sinopharm Chemical Reagent Co., Ltd. MWCNTs prepared by a chemical vapor <sup>10</sup>deposition (CVD) method were purchased from Carbon Nanotechnology Inc. China. Deionized water was produced by a Milli-QSystem (Millipore). 2,2'-bis (4-cyanatophenyl) propane (BCE, Heijang Kinlyuan Pharmaceutical Co., Ltd., China), 4,4' bismaleimidodiphenylmethane (BMI, Honghu Bismaleimide <sup>15</sup>Resin Factory, China), and 2, 2'diallyl bisphenol A (DBA, Wuxi

Resin Factory, China) were used as starting materials to prepare the BT resin, according to our previous work.<sup>27</sup> All of the aforementioned reagents were of analytical grade and were used as received.

#### <sup>20</sup>**2.2 Oxidation of MWCNTs**

MWCNTs  $(0.22 \text{ g})$  were modified by treating them in ~15.0 mol. $L^{-1}$  HNO<sub>3</sub> solution (40 mL) under reflux at 120 °C for 6 h. After cooling to room temperature, they were vacuum-filtered through a 0.45 mm polytetrafluoroethylene (PTFE) membrane <sup>25</sup>and washed with deionized water until the filtrate reached a neutral pH. The oxidized MWCNTs were dried under vacuum for 24 h.

#### **2.3 Preparation of MWCNT@SiO<sup>2</sup>**

The preparation of  $MWCNT@SiO<sub>2</sub>$  was based on MWCNTs-30 COOH via sol-gel reaction, as reported previously.<sup>16</sup> Oxidized MWCNTs (0.20) and ammonia solution (1.0 mL) were sonicated in ethanol/deionized water mixture (8:2 v/v, 100 mL) for 1 h. The pH of the mixture was kept at 11. In parallel, TEOS (2.0 mL) was sonicated in ethanol (40 mL) for 30 min. The TEOS/ethanol <sup>35</sup>mixture was added dropwise to the oxidized MWCNTs solution.

The reaction at 30 °C lasted for 2 h after addition of TEOS. The products were vacumm-filtered through a 0.45 µm PTFE membrane, and washed with absolute ethanol at least 5 times before drying at 80 °C for 12 h.

#### <sup>40</sup>**2.4 Preparation of polymer nanocomposites**

The polymer nanocomposites were fabricated by solution processing. A certain amount of  $MWCNT@SiO<sub>2</sub>$  particles dispersed in methyl ethyl ketone were ultrasonicated to form a stable colloid, followed by adding a certain amount of BT resin

- <sup>45</sup>and stirring for 4 h. Then a little hardener (2-ethyl-4 methylimidazole) was added and the mixture was stirred for another 2 h. Finally, the resultant solution mixture was coated onto the copper, and then dried at  $100 \degree C$  for 2 h to thoroughly evaporate the solvent. The nanocomposites were then cured at
- <sup>50</sup>150 °C for 2 h, 180 °C for 2 h, and 200 °C for 2 h in sequence. The thickness of the nanocomposites was about 30 µm. A series of BT/MWCNT $@SiO_2$  nanocomposites were prepared with the  $MWCNT@SiO<sub>2</sub>$  loading varied between 0.4 and 2.0 wt%. For comparison studies, pure BT resin and nanocomposites filled with

#### **2.5 Measurements**

The morphology of the MWCNT $@SiO_2$  was examined by transmission electron microscopy (TEM, G2 F20, FEI Tecnai) <sup>60</sup>with an accelerating voltage of 200 kV. The FTIR spectra were obtained on a Bruker Vertex 70 with pure KBr as the background from 400 and 4000 cm<sup>-1</sup>. UV-vis spectra were performed with a UV-vis spectrometer (UV-3600, Shimadzu). Raman spectra were obtained from 1200 to 3000 cm-1 using a Raman spectrometer <sup>65</sup>(Renishawin Via Raman microscope, excitation at 514 nm). The chemical state of the surface was characterized by X-ray photoelectron spectroscopy (XPS) on an Axis Ultra-DLD spectrometer (Kratos Analytical Ltd.), by using a monochromatized Al Ka source (1486.6 eV) operating at 15 kV  $70$  and 15 mA. The vacuum during analysis was in the range from 10<sup>-</sup>  $10$  to  $10^{-9}$  Pa. The Peakfit software was used to carry out the curve fitting. The volume resistivities of  $BT/MWCNT@SiO<sub>2</sub>$  and BT/MWCNT nanocomposites were measured using a highresistance meter (Agilent 4339B, Agilent Technologies) and <sup>75</sup>resistance meter (Keithley 2400, Keithley Instruments), respectively. The dielectric properties were measured with an impedance analyzer (Agilent 4991A) in the frequency range of  $10<sup>7</sup>$ -10<sup>9</sup> Hz. Static uniaxial in-plane tensile tests were conducted with a dynamic mechanical analyzer (DMA Q800, TA <sup>80</sup>Instruments). The samples were cut with a razor into rectangular strips of approximately  $4\times15$  mm<sup>2</sup> for mechanical testing. All tensile tests were conducted in controlled force rate mode with a preload of 0.001N and a force ramp rate of 1.0 N.min<sup>-1</sup>. SEM images were obtained using a field-emission scanning 85 electronmicroscope (Nova NanoSEM 450, FEI) combined with energy dispersive spectrometer (X-Max50, Oxford Instruments) with 5 or 10 kV accelerating voltage. Thermal stabilities of the nanocomposites were measured with a Thermogravimetric analysis (TGA, Thermal Analysis (TA) Instruments SDT-Q600 <sup>90</sup>analyzer) in the air environment. The temperature ramps were from 40 to 900  $^{\circ}$ C at a heating rate of 10  $^{\circ}$ C.min<sup>-1</sup>. A frequency "flasher" circuit was fabricated by bonding discrete electronic components to the MWCNT@SiO<sub>2</sub> nanocomposite film  $(100 \times 50)$ mm<sup>2</sup>) to demonstrate its potential application in printed circuit 95 substrates.

#### **3. Results and discussion**

#### **3.1 Characterization of MWCNT@SiO<sup>2</sup>**

The TEM examination (Figure 1a) reveals that there are almost no free  $SiO<sub>2</sub>$  particles in MWCNT@SiO<sub>2</sub> fillers. This is attributed 100 to the generation of oxygenated groups on MWCNTs by acid treatment, which facilitates the heterogeneous nucleation of  $SiO<sub>2</sub>$ on MWCNT surfaces while inhibiting its homogeneous nucleation. In other words, oxidization process of MWCNT promotes the formation of  $SiO<sub>2</sub>$  shell on its surface rather than 105 growing into free  $SiO<sub>2</sub>$  particles. The amorphous  $SiO<sub>2</sub>$  layer with  $\sim$  10 nm in thickness is uniformly coated on the surface of MWCNT, as shown in Figure 1b and c, and SEM images (Figure S1, ESI†). In addition, the energy dispersive X-ray spectroscopy (EDX) elemental mappings (Figure S2, ESI†) of carbon, silicon,  $110$  and oxygen further confirm that  $SiO<sub>2</sub>$  is uniformly coated onto

MWCNTs. The calculated  $SiO<sub>2</sub>$  weight loading in  $MWCNT@SiO<sub>2</sub>$  is 22.8%. Surface functionalization of MWCNT is expected to have a significant effect on its dispersibility in water. Indeed, the MWCNT $@SiO_2$  in water exhibits good <sup>5</sup>dispersion, whereas the pristine MWCNTs deposit at the bottom, as shown in Figure 2(d). Moreover, the MWCNT $@SiO_2$  shows higher light transmittance compared with MWCNTs-COOH with the same loading in water.





- 10 Figure 1. TEM images of (a) MWCNT@SiO<sub>2</sub>; (b) and (c) are the enlarged ones of the selected regions in (a); (d) photograph of pristine MWCNTs, CNTs-COOH and MWCNT@SiO<sub>2</sub> dispersed in water.
- The existence of functional groups in the MWCNT $@SiO_2$  is <sup>15</sup>evidenced by FTIR (Figure 2). The characteristic absorption bands of MWCNTs $@SiO_2$  at approximately 1109, 957, 798 and 476 cm<sup>-1</sup> are ascribed to  $v_{as}$  (Si-O-Si),  $v(Si$ -OH),  $v(Si$ -O-Si), and  $\delta$ (Si-O-Si), respectively, indicating the formation of SiO<sub>2</sub> on MWCNTs. The peak at  $3414 \text{ cm}^{-1}$  is assigned to the O-H <sup>20</sup>stretches of Si-OH, which may be beneficial to improve the
- $MWCNT@SiO<sub>2</sub>'s interfacial interaction with polymer.$



Figure 2. FTIR spectra of pristine MWCNTs, CNTs-COOH, and MWCNT@SiO<sub>2</sub>.

- 25 To confirm the formation of a covalent bond between  $SiO<sub>2</sub>$  and MWCNT, XPS experiments were carried out, as shown in Figure 3 (a). The spectrum of  $C_{1s}$  electron orbital (Figure 3(b)), indicates the existence of seven carbon components: the graphite (283.3 ev), the C<sub>sp2</sub> (284.8 ev), the C<sub>sp3</sub> (285.1 ev), the C-O bond (287.1 <sup>30</sup>ev), the C=O bond (287.8 ev), the O=C-O bond (289.0 ev), and the carbon of  $\pi$ - $\pi$ <sup>\*</sup> (289.1). Through the fitting of the O 1s electron orbital spectrum (Figure 3(c)), the -O-  $(533.0 \text{ eV})$  are observed. The  $SiO_2$  component at 104.2 ev in  $Si<sub>2p</sub>$  electron orbital spectrum can be observed (Figure 3(d)). The presence of  $SiO_2-C_2$  $35(101.1 \text{ eV})$  and  $SiO<sub>3</sub>$ -C (103.6 eV) bonds indicates the existence
- of a covalent bond (Si-O-C) between  $SiO<sub>2</sub>$  and MWCNT.







Figure 4. Raman spectra of pristine MWCNTs, CNTs-COOH, and MWCNTs@SiO<sub>2</sub>.

Figure 4 presents Raman spectra of the MWCNTs $@SiO_2$ . The <sup>5</sup>MWCNTs and functionalized MWCNTs (MWCNTs-COOH and MWCNT $\omega$ SiO<sub>2</sub>) exhibit a D band at around 1340 cm<sup>-1</sup>, which is induced by the disordered structure or  $sp<sup>3</sup>$  hybridized carbons and a G band at around 1570 cm<sup>-1</sup>, which is derived from the splitting of the  $E_{2g}$  stretching mode of graphite. The intensity ratio of D  $10$  band and G band (I<sub>D</sub>/I<sub>G</sub>) is used to characterize the defect quantity and monitor the functionalization of carbon materials. In our work, due to increasing defects and new edges introduced during the oxidative treatment, the value of  $I_D/I_G$  increased from 0.83 (pristine MWCNTs) to 0.94 (MWCNTs-COOH). After 15 encapsulating MWCNT with  $SiO_2$ ,  $I_D/I_G$  of CNT decreased to 0.82, even slightly lower than that of the pristine MWCNTs. These results can be interpreted in terms of some defect suppression of the lattice after coating  $SiO<sub>2</sub>$ .

#### **3.2 Properties of BT/MWCNT@SiO<sup>2</sup> nanocomposites**

#### <sup>20</sup>**Optical property**

Figure 5(a) presents photographs of the BT/MWCNTs $@SiO<sub>2</sub>$ and BT/MWCNTs nanocomposite films with different loadings. The  $BT/MWCNT@SiO<sub>2</sub>$  nanocomposite films are optically transparent at all filler loadings and their light transmittance at <sup>25</sup>800 nm is about 30%, which is slightly lower than that of the neat

- BT resin film (37%). In contrast, the light transmittance rate of the BT/MWCNTs nanocomposite films significantly decreases with increasing pristine MWCNTs loadings (Figure 5(b)). When the pristine MWCNTs loading is 1.6 wt%, the light transmittance
- <sup>30</sup>of the composite is only 5%. The enhanced optical transmittance compared with pristine MWCNTs should be originated from the uniform dispersion of MWCNT@SiO<sub>2</sub> in BT resin and the  $SiO<sub>2</sub>$ layer, which can prevent the light absorption of MWCNTs. The transparent property of materials is very important, especially for
- <sup>35</sup>the applications in flexible displays, colourful e-paper, organic light-emitting diodes (OLEDs) and organic photovoltaics (OPVs). It is assumed that this strategy may be extended to fabrication of nanocomposites filled with MWCNT@SiO<sub>2</sub> with higher transparency by proper choice of polymers, and thus
- 40 broaden the application of MWCNT $@SiO_2$ .



Figure 5. The effect of MWCNT $@SiO_2$  on the optical transparency and UV-vis transmittance of the nanocomposite films: (a) photographs showing the transparency of samples; (b) <sup>45</sup>UV-vis spectra of nanocomposites with different MWCNT and MWCNT@SiO<sub>2</sub> loadings.

#### **Electric and dielectric properties**

Figure 6 shows the volume resistivity of the  $BT/MWCNT@SiO<sub>2</sub>$  and  $BT/MWCNT$  nanocomposites with <sup>50</sup>different filler loadings. Addition of a small amount of pristine MWCNTs (0.4 wt%) leads to the transformation of the nanocomposite from insulator to semiconductor. Specifically, the volume resstivity declines from 1.32 ×10<sup>17</sup> $\Omega$ .cm for pure BT resins to  $6.14 \times 10^4$  Ω.cm for the nanocomposite containing 0.8  $55 \text{ wt\%}$  MWCNTs. In contrast, the insulate  $\text{SiO}_2$  coated on MWCNT surface exhibits a barrier effect on the electrons transformation. The volume resistivity of the nanocomposites containing  $MWCNT@SiO<sub>2</sub>$  is higher by approximately five orders of magnitude than that containing the pristine MWCNTs, at the 60 same loadings. When the MWCNT@SiO2 loading is 0.8 wt%, the measured volume resistivity of the nanocomposite is  $2.36 \times 10^9$ Ω.cm, indicating that it is still an insulator (>1.0×10<sup>9</sup> Ω.cm). However, further increase of MWCNT $@SiO_2$  loading leads to a decrease of volume resistivity to  $4.74 \times 10^7 \Omega$  cm, suggesting that 65 the nanocomposite is not insulated.



#### loadings.

- The miniaturization of electronic devices integrated in printed circuits substrates has resulted in rapid increase of the propagation delay caused by interconnection. In order to reduce the resistance-capacitance (RC) delay, low dielectric constant and loss of insulating materials are urgently desired.<sup>28</sup> As shown in
- <sup>10</sup>Figure 7 (a) and (b), all nanocomposites show similar dielectric stability on frequency and their dielectric constants increase linearly with loading amount. Compared to the pristine MWCNTs (Figure 7 (c)), MWCNT $@SiO_2$  leads to less improvement in the dielectric constant of BT resin. With the same loading  $(1.6 \text{ wt\%}),$
- 15 the dielectric constant of  $BT/MWCNT@SiO<sub>2</sub>$  nanocomposites increase from 3.0 for pure BT resin to 3.5, while it increases to 5.0 for BT/MWCNTs nanocomposites. On the other hand, it seems that the addition of  $MWCN@SiO<sub>2</sub>$  does not influence the dielectric loss of BT resin. In contrast, the dielectric loss of the
- <sup>20</sup>BT/MWCNTs nanocomposites (Figure 7(d)) increases with filler loading and reaches up to 0.04. We also find that adding more  $MWCNT@SiO$ <sub>2.0</sub> wt%) leads to increased dielectric loss (0.04) (Figure S3, ESI†), which may be attributed to the poorer dispersion of MWCNT $@SiO_2$  in BT resins, which causes the
- <sup>25</sup>regional formation of a conductive network of the CNT within the polymer matrix. There are two possible explanations for the discrepancy of dielectric properties between  $MWCNT@SiO<sub>2</sub>$  and MWCNT. First, the insulated  $SiO<sub>2</sub>$  layer on MWCNT limits the dipoles moment, resulting in low energy storage in an
- <sup>30</sup>electromagnetic field. Second, dielectric properties depend on the inter-action between inorganic and polymer. Because  $MWCNT@SiO<sub>2</sub>$  has stronger inter-action with BT resin than pure MWCNTs due to the existence of active -OH group,  $MWCNT@SiO<sub>2</sub>$  fillers have bigger restricting influence on the 35 orientation and relaxation of dipoles than MWCNTs in nanocomposites.



Figure 7. Dielectric constant and loss of  $BT/MWCNT@SiO<sub>2</sub>$  (a, b) and BT/MWCNT (c, d) with different filler loadings.

#### <sup>40</sup>**Mechanical properties**

BT resins.

Figure 8(a) shows the typical stress-strain curves of  $BT/MWCNT@SiO<sub>2</sub>$  nanocomposites. The stress strength (Figure  $8(b)$ ) and ultimate strain (Figure  $8(c)$ ) were extracted from the curve just before failure. Young's modulus (Figure 8(d)) was 45 calculated from the slope of the linear fit in the initial section of the curve (0.5 % strain). For the neat BT resin film, the stress strength, elongation and Young's modulus are 48.2 MPa, 2.6% and 1.6 GPa respectively. After the addition of pristine MWCNTs or MWCNT $\omega$ SiO<sub>2</sub>, the mechanical properties of BT resin <sup>50</sup>increase with the filler loadings, and the addition of 0.8 wt% fillers exhibits the highest stress strength and elongation. Adding more fillers brings about a slight decrease in stress strength and elongation. Such a phenomenon was also observed in other reports.<sup>29, 30</sup> There could be regional agglomeration of the filler 55 particles in the composite, which limits the efficiency of reinforcement. As compared to the pristine MWCNTs,  $MWCNT@SiO<sub>2</sub>$  leads to more pronounced improvement in the mechanical properties, except for the elongation values. The stress strength and Young's modulus of the nanocomposite 60 containing 0.8 wt% MWCNT@SiO<sub>2</sub> are 87.4 MPa and 3.18 GPa, respectively, which are higher than that of the BT/pristine MWCNTs. Figure 8(d) displays that the elongation values of  $BT/MWCNT@SiO<sub>2</sub>$  are lower than those of the  $BT/MWCNTs$ composites, but still higher than that of pure BT resin. The results 65 indicate strong interfacial interaction between MWCNT@SiO<sub>2</sub> and BT resin, which leads to difficult deformation of the nanocomposites under the tensile loading. Furthermore, as a control experiment, we also measured the mechanical properties of the composite filled with 0.8 wt% glass fibers which are the <sup>70</sup>conventional composites used as printed circuit substrate materials. As shown in Figure 8, its tensile strength, elongation and Young's modulus are 63.0 MPa, 2.4 GPa, and 3.1%, respectively, which are all lower that of the nanocomposite filled with 0.8 wt%  $MWCNT@SiO<sub>2</sub>$ . The control experiments with 75 pristine MWCNTs and glass fibers confirm that  $MWCNT@SiO<sub>2</sub>$ plays the key role for enhancement of mechanical properties of



Figure 8. (a) typical stress-strain curves of BT resin based nanocomposites with different loadings of MWCNT@SiO<sub>2</sub> or MWCNTs, (b) absolute value of tensile strength, (c) percentage <sup>5</sup>elongation, and (d) Young's modulus of the nanocomposites.

In order to compare the mechanical properties of the nanocomposites filled with functionalized CNTs, we define the rate of increase of tensile strength  $(\sigma)$  and Young's modulus (Y) to the CNTs' weight fraction (W). The equations are as followed:

$$
\frac{d\sigma}{dW} = \frac{\sigma_c - \sigma_P}{W} \quad (1)
$$

$$
\frac{dY}{dW} = \frac{Y_c - Y_P}{W} \quad (2)
$$

Table 1 summarizes recent typical studies on the effects of functionalized CNTs on the mechanical properties of polymer nanocomposites. These results indicate that the addition of <sup>15</sup>functionalized CNTs can enhance the tensile strength and Young's modulus, and different systems possess different degree of mechanical improvement. In our research, the dσ/dW and dY/dW are as high as 4900 and 200, respectively, which are higher than most reported values. Therefore, we believe that the <sup>20</sup>improvement of 81.2% and 98.5% in both tensile strength and Young's modulus in our results is significant.



Figure 9. SEM fractographs of the pure BT resin and  $BT/MWCNTs@SiO<sub>2</sub> nanocomposites: (a) the pure BT resin; (b)$ 25 0.8 wt% MWCNT $\omega$ SiO<sub>2</sub>; (c) 0.8 wt% Pristine MWCNTs and (d) 1.6 wt% MWCNT $@SiO<sub>2</sub>$ .





<sup>30</sup> PMMA: Poly(methyl methacrylate); f-SWNTs: functionalized single-walled carbon nanotubes; f-MWCNTs: functionalized multiwalled carbon nanotubes.

As we all know, the mechanical reinforcement of MWCNT filled nanocomposites strongly depends on the extent of load transfer between the matrix and MWCNTs.<sup>31, 40, 41</sup> For 35 MWCNT@SiO<sub>2</sub>, a number of OH groups have been verified to be attached to its surface, while pristine MWCNTs lack surface functional groups, as indicated in Figure 2. Therefore, the  $MWCNT@SiO<sub>2</sub>$  has stronger interfacial interactions with BT resins than pristine MWCNTs. These innate advantage results in 40 efficient stress transfer between the MWCNT $@SiO_2$  and BT

matrix, which could avoid the deformation or fracture of the materials under external force. On the other hand, the dispersion of nanofiller in matrix is another key factor that affects the reinforcement. As shown in SEM images in Figure 9(a), the pure <sup>45</sup>BT resin exhibits a smooth fracture surface. After the addition of MWCNT@SiO<sub>2</sub> (0.8 wt%) into BT resin, MWCNT@SiO<sub>2</sub> are well-dispersed in the nanocomposites, as shown in Figure 9(b). In contrast, the agglomeration or voids can be clearly seen in BT/MWCNTs nanocomposite with the same nanofiller loading

(Figure 9(c)). Owing to the good dispersion of MWCNT $\omega$ SiO<sub>2</sub> in the matrix, the large surface area enables more interaction sites at the polymer/nanofiller interphase, leading to the improved mechanical strength. With the higher loading of 1.6 wt%, the <sup>5</sup>composite exhibits a significantly increased viscosity, so that porosity is easily introduced to the composite, as shown in Figure 9(d). The tensile strength of composites is very sensitive to such void defects. Therefore, a slightly decreased tensile strength is observed for the composite with the addition of 1.6 wt% 10 MWCNT@SiO2.

**Thermal stability** 



Figure 10. TGA curves of the pure BT resin and nanocomposites containing different MWCNT@SiO<sub>2</sub> loadings. The inset shows  $15$  the  $T_{d5\%}$  values for all samples.

Thermal stability is one of the most important properties of the nanocomposites to assess their processing and service-life performance in the applications of printed circuit substrates. Figure 10 shows the TGA curves for the pure BT resin and BT/  $_{20}$  MWCNT@SiO<sub>2</sub> nanocomposites. The nanocomposites are stable up to 350 °C and then completely decompose at above 700 °C, leaving no residue. The 5 wt% weight loss temperatures  $(T_{d5\%})$ for all samples are extracted and summarized in the inset of Figure 10. The values of  $T_{d5\%}$  increase from 335.2 °C for the pure 25 BT resin to 353.7  $\degree$ C for the nanocomposite containing 1.6 wt%  $MWCNT@SiO<sub>2</sub>$ , indicating the improved thermal stability. The

- stabilization effect of MWCNTs@SiO<sub>2</sub> is mainly attributed to good polymer matrix-MWCNT@SiO<sub>2</sub> interaction, as well as barrier effect of MWCNT@SiO<sub>2</sub>. Furthermore, the growth of T<sub>d5%</sub>
- 30 slows down when MWCNT@SiO<sub>2</sub> is more than 0.8 wt%, due to the existence of agglomerations in the composite, as shown in the SEM image (Figure 9d).

#### **4. Application in frequency ''flasher''circuit**

In order to demonstrate the potential application of the BT/  $35$  MWCNT@SiO<sub>2</sub> nanocomposites in printed circuit substrates, we fabricated a frequency "flasher" circuit by bonding discrete electronic components to our nanocomposite substrate. The  $MWCNT@SiO<sub>2</sub>$  loading in the nanocomposite is 0.8 wt%, which leads to the maximum values in mechanical properties as shown

<sup>40</sup>in the above results. One should note that the BT loaded with 0.8 wt% pristine MWCNTs cannot be used as the printed circuit substrate, because of its low electrical resistivity. In addition, the

pure BT resin is also not the ideal circuit substrate because of its low mechanical strength, which will lead to easy break during the 45 fabrication of the circuit.



Figure 11. Applications of the nanocomposite in printed circuit substrates: (a) circuit principle diagram of FFC. (b, c) the optical images of FFC working before and after bending; (d) optical

images of two LEDs flashed alternately.

Figure 11(a) presents the circuit principle diagram. The circuit comprises of four resistors (R1, R2, R3, and R4), one capacitor, one power (6 V), one 555 timer integrated circuit (NE555) and two LEDs (green (D1) and red (D2)). The detailed <sup>55</sup>manufacturing process of the frequency "flasher" circuits follows that of the common printed circuit boards (PCB). The copper (Cu) clad laminate containing MWCNT@SiO<sub>2</sub> was prepared, so that we can use copper wire as electrical circuit. Figure S4 shows the fabrication process of the copper clad laminate. The electrical 60 circuit was fabricated by UV photoengraving method (Figure S5, ESI†). The circuit could be turned on by connecting the negative and positive electrodes (Figure 11(b)). The working printed circuit substrate demonstrates excellent flexibility. As shown in Figure 11(c), it is working well after bending to a certain extent. <sup>65</sup>When the circuit is on, the two LEDs flash alternately at a fixed frequency (Figure 11(d)). A movie of the operation of the LED chip can be found in the electronic supplementary information (see Movie S1, ESI†).

#### **5. Conclusions**

We have demonstrated a strategy for fabricating MWCNTs based polymer nanocomposites with high mechanical strength and electrical insulation by encapsulating MWCNT with  $SiO<sub>2</sub>$  $(MWCNT@SiO<sub>2</sub>)$ . With only 0.8 wt% MWCNT@SiO<sub>2</sub>, the resultant nanocomposites exhibits 81.2% and 98.5 % increases in

tensile strength and Young's modulus (compared to pure BT resin), respectively. The volume resistivity of the nanocomposites containing MWCNT@SiO<sub>2</sub> is higher by approximately five orders of magnitude than that containing the pristine MWCNTs at

- s the same loadings. Addition of MWCNT@SiO<sub>2</sub> hardly affects the dielectric properties and optical transparency of BT resin. Furthermore, the values of  $T_{d5%}$  increase with increase of MWCNT@SiO<sub>2</sub>, and reaches 353.7 °C for the nanocomposite containing 1.6 wt% MWCNT@SiO<sub>2</sub>, which is 18.5 °C higher
- 10 than that of pure BT resin. These excellent properties are attributed to  $MWCNT@SiO<sub>2</sub>'s$  strong interfacial interactions with BT resin and high electrical insulation of the  $SiO_2$ -coated MWCNTs. The  $BT/MWCNT@SiO<sub>2</sub>$  nanocomposites have been successfully demonstrated as a printed circuit substrate, on which
- <sup>15</sup>a frequency "flasher" circuit and the electrical components worked well. The concept of using  $MWCNT@SiO<sub>2</sub>$  as fillers in this study will also be applicable to a wide range of other polymer nanocomposites.

#### **ACKNOWLEDGMENT**

The authors acknowledge the financial support from National Natural Science Foundation of China (No.51377157), Guangdong and Shenzhen Innovative Research Team Program (No. 2011D052 and KYPT20121228160843692) and Shenzhen Electronic Packaging Materials Engineering Laboratory <sup>25</sup>(No.2012-372).

#### **Notes and references**

<sup>a</sup> Shenzhen Institutes of Advanced Technology, Chinese Academy of *Sciences.1068 Xueyuan Avenue, Shenzhen University Town, Shenzhen, China* 

- <sup>30</sup> <sup>*b*</sup> College of Advanced Technology, University of Chinese Academy of *Sciences, China.*
- *c Department of Electronics Engineering, The Chinese University of Hong Kong, Hong Kong, China*

\* Corresponding authors. Email: Shuhui Yu, sh.yu@siat.ac.cn; Rong Sun, 35 rong.sun@siat.ac.cn.

- † Electronic Supplementary Information (ESI) available: [SEM images of MWCNT@SiO<sub>2</sub>, EDX elemental mapping of  $MWCNT@SiO<sub>2</sub>$ , dielectric properteis of BT/2.0 wt% MWCNT@SiO<sup>2</sup> nanocomposite, preparation of copper clad
- 40 laminate containing MWCNT@SiO<sub>2</sub>, fabrication of electrical circuit by subtractive process, and a video that shows the operation of the frequency "flasher" circuits.]. See DOI: 10.1039/b000000x/.
- 1. M. Naffakh, A. M. Díez-Pascual, C. Marco, G. J. Ellis and M. A.
- <sup>45</sup>Gómez-Fatou, *Progress in Polymer Science*, 2013, **38**, 1163- 1231.
- 2. J. N. Coleman, U. Khan and Y. K. Gun'ko, *Advanced Materials*, 2006, **18**, 689-706.
- 3. N. Roy, R. Sengupta and A. K. Bhowmick, *Progress in Polymer*  <sup>50</sup>*Science*, 2012, **37**, 781-819.
- 4. A. M. Diez-Pascual and D. Gascon, *ACS applied materials & interfaces*, 2013, **5**, 12107-12119.
- 5. Y. Ma, P. L. Chiu, A. Serrano, S. R. Ali, A. M. Chen and H. He, *Journal of the American Chemical Society*, 2008, **130**, 7921-
- <sup>55</sup>7928.
- 6. N. G. Sahoo, S. Rana, J. W. Cho, L. Li and S. H. Chan, *Progress in*

*Polymer Science*, 2010, **35**, 837-867.

- 7. C. Min, X. Shen, Z. Shi, L. Chen and Z. Xu, *Polymer-Plastics Technology and Engineering*, 2010, **49**, 1172-1181.
- <sup>60</sup>8. T. Souier, C. Maragliano, M. Stefancich and M. Chiesa, *Carbon*, 2013, **64**, 150-157.
	- 9. J. Zhang, D. Jiang, H.-X. Peng and F. Qin, *Carbon*, 2013, **63**, 125-132.
	- 10. E. Y. Li and N. Marzari, *ACS Nano*, 2011, **5**, 9726-9736.
	- 11. S. L. Zhang, S. B. Yin, C. R. Rong, P. F. Huo, Z. H. Jiang and <sup>65</sup>G. B. Wang, *Eur. Polym. J.*, 2013, **49**, 3125-3134.
	- 12. X. Sun, H. Sun, H. Li and H. Peng, *Advanced materials (Deerfield Beach, Fla.)*, 2013, **25**, 5153-5176.
	- 13. S. Desbief, N. Hergue, O. Douheret, M. Surin, P. Dubois, Y. Geerts, R. Lazzaroni and P. Leclere, *Nanoscale*, 2012, **4**, <sup>70</sup>2705-2712.
	- 14. Y. Q. Li, Y. A. Samad, K. Polychronopoulou, S. M. Alhassan and K. Liao, *Scientific Reports*, 2014, **4**, 4652.
	- 15. K. Ding, B. Hu, Y. Xie, G. An, R. Tao, H. Zhang and Z. Liu, *Journal of Materials Chemistry*, 2009, **19**, 3725-3731.
- <sup>75</sup>16. A. J. Paula, D. Stefani, A. G. Souza Filho, Y. A. Kim, M. Endo and O. L. Alves, *Chemistry-a European Journal*, 2011, **17**, 3228-3237.
	- 17. S. W. Kim, T. Kim, Y. S. Kim, H. S. Choi, H. J. Lim, S. J. Yang and C. R. Park, *Carbon*, 2012, **50**, 3-33.
- <sup>80</sup>18. K. Hayashida and H. Tanaka, *Advanced Functional Materials*, 2012, **22**, 2338-2344.
	- 19. S. K. Yadav, I. J. Kim, H. J. Kim, J. Kim, S. M. Hong and C. M. Koo, *Journal of Materials Chemistry C*, 2013, **1**, 5463- 5470.
- <sup>85</sup>20. T. Morishita, M. Matsushita, Y. Katagiri and K. Fukumori, *Journal of Materials Chemistry*, 2011, **21**, 5610-5614.
	- 21. W. Cui, F. Du, J. Zhao, W. Zhang, Y. Yang, X. Xie and Y.-W. Mai, *Carbon*, 2011, **49**, 495-500.
- 22. Y. Zhang, S. Xiao, Q. Wang, S. Liu, Z. Qiao, Z. Chi, J. Xu and <sup>90</sup>J. Economy, *Journal of Materials Chemistry*, 2011, **21**, 14563- 14568.
	- 23. J. Di, Z. Yong, Z. Yao, X. Liu, X. Shen, B. Sun, Z. Zhao, H. He and Q. Li, *Small*, 2013, **9**, 148-155.
- 24. A. Gohier, B. Laik, K.-H. Kim, J.-L. Maurice, J.-P. Pereira-<sup>95</sup>Ramos, C. S. Cojocaru and V. Pierre Tran, *Advanced Materials*, 2012, **24**, 2592-2597.
	- 25. X. Zeng, S. Yu and R. Sun, *J. Appl. Polym. Sci.*, 2013, **128**, 1353-1359.
	- 26. X. Zeng, S. Yu, R. Sun and R. Du, *Materials Chemistry and*  <sup>100</sup>*Physics*, 2011, **131**, 387-392.
	- 27. X. L. Zeng, S. H. Yu and R. Sun, *J. Appl. Polym. Sci.*, 2013, **128**, 1353-1359.
	- 28. C. Yuan, K. Jin, K. Li, S. Diao, J. Tong and Q. Fang, *Advanced Materials*, 2013, **25**, 4875-4878.
- <sup>105</sup>29. J. Wang, Z. Shi, Y. Ge, Y. Wang, J. Fan and J. Yin, *Journal of Materials Chemistry*, 2012, **22**, 17663-17670.
	- 30. L. An, Y. Pan, X. Shen, H. Lu and Y. Yang, *Journal of Materials Chemistry*, 2008, **18**, 4928-4941.
- 31. J. M. Gonzalez-Dominguez, A. Anson-Casaos, A. M. Diez-110 Pascual, B. Ashrafi, M. Naffakh, D. Backman, H. Stadler, A. Johnston, M. Gomez and M. Teresa Martinez, *Acs Applied Materials & Interfaces*, 2011, **3**, 1441-1450.



2006, **39**, 6565-6573.



Functionalized multiwalled carbon nanotubes (MWCNTs)-silicon oxide (SiO2) core-shell nanoparticles filled polymer nanocomposites with high mechanical strength and electrical insulation were developed, and employed to fabricate a printed circuit substrate.