

# Analytical Methods

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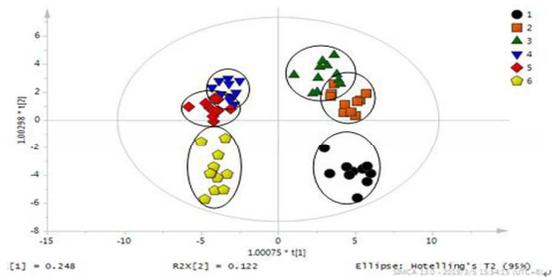


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The variation of root exudates from the Cd hyperaccumulator *Sedum alfredii* under different Cd exposed concentration and time was researched.

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4 **Metabolic profiling analysis of root exudates from the Cd**  
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7 **hyperaccumulator *Sedum alfredii* under different Cd exposed**  
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9 **concentration and time**  
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22

23 **Abstract**  
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26 Components of the Cd hyperaccumulator *Sedum alfredii* root exudates were  
27 surveyed by gas chromatography-mass spectrometry (GC-MS), the variation of root  
28 exudates from *S. alfredii* under different Cd exposed concentration and time was  
29 explored by metabonomics analysis, and the probable effect mechanism of *S. alfredii*  
30 tolerates or accumulates the heavy metal Cd was discussed. The root exudates were  
31 collected after 0, 5 and 10  $\mu\text{mol/L}$  Cd treated for 4 and 8 days. The collected solution  
32 was lyophilized and eluted with methanol, after derivatization with methoxyamine  
33 hydrochloride and N-methyl-N-trifluoroacetamide, the samples were analyzed by  
34 GC-MS. Principal component analysis (PCA) and orthogonal partial least-squares  
35 discrimination analysis (OPLS-DA) were carried out for pattern recognition and a  
36 clear separation among the different treatments was achieved. 15 compounds resulting  
37 in the separation among the different treatments were found and identified. And the  
38 changing tendencies in the relative content of these 15 compounds under the different  
39 treatments were explored. These results indicated that the Cd hyperaccumulator *S.*  
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4 *alfredii* could be able to adjust the secretion of root exudates to tolerate or accumulate  
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7 the heavy metal Cd.

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9 **Key words:** hyperaccumulator; *Sedum alfredii*; root exudates; GC-MS;  
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11 metabonomics  
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## 14 15 16 17 **Introduction**

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20 Plant root exudates are plant metabolites that are released to the root surface or  
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22 into the rhizosphere to enhance plant nutrient uptake or cope with  
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24 environment stresses, such as low-molecular-weight organic acids, amino acids, fatty  
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26 acids and sugars.<sup>1-4</sup> They can be modifying the pH and Eh of the rhizosphere,  
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28 chelating, complexing and depositing with heavy metals, altering the numbers and the  
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30 activity of rhizospheric microbes. Through these ways, root exudates can change the  
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32 chemical existence of heavy metals and increase their bio-availability.<sup>5</sup>  
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38 Root exudates play an important role in the process of phytoremediation as an  
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40 emerging green and in-situ remediation technology using plants to absorb, accumulate,  
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42 stabilize or volatilize contaminant from soil.<sup>6-8</sup> However, previous studies were  
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44 mostly focused upon the roles of root exudates, the changes in the total amount of  
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46 the dissolved organic matter (DOM) or the dissolved organic carbon (DOC),  
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48 or some specific organic acids and amino acids,<sup>9-12</sup> the components and variation of  
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50 root exudates of hyperaccumulators under different stresses were rarely revealed.  
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58 In this study, the hyperaccumulator *Sedum alfredii*, a Zn/Cd hyperaccumulator  
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60 native to China, with large biomass, rapid growth and asexual propagation,<sup>13-14</sup> was

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4 cultured in Cd stressed nutrition solutions. The analytical technique of gas  
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6 chromatography- mass spectrometry (GC-MS), a mature method and has long been  
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8 used for metabolite profiling of plant extracts,<sup>15-17</sup> was used to analyze the root  
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10 exudates released from *S. alfredii*. And the metabonomics analysis method, a high  
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12 throughput and unbiased comprehensive analysis method to study the dynamic change  
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14 of endogenous metabolism of a biological system that is stimulated or disturbed,<sup>18</sup>  
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16 was used to analyze the variation of root exudates from *S. alfredii* under the different  
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18 Cd exposed concentration and time. Finally, we found out the compounds resulting in  
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20 the separation among the different treatments. Then through the change trends of  
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22 these compounds, we explored the possible mechanism of *S. alfredii* tolerates or  
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24 accumulates the heavy metal Cd.  
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## 33 **Material and methods**

### 34 **Chemicals and instruments**

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36 Methanol (HPLC grade, Fisher), pyridine (HPLC grade, Sinopharm),  
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38 methoxamine hydrochloride (Sigma) and N-methyl-N-trimethylsilyl trifluoroacetamide  
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40 (MSTFA, Sigma) were employed in the study. Compositions of the nutrient solution  
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42 and CdCl<sub>2</sub> were purchased from Sinopharm.  
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50 GC-MS (TRACE GC Ultra-PolarisQ, ThermoFisher, an ion trap mass  
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52 spectrometry), nitrogen purging instrument (MG-2200, TOKYO RIKAKIKAI  
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54 CO.LTD), and vacuum freeze drying system (FDU-1100, TOKYO RIKAKIKAI  
55  
56 CO.LTD) were used in this study.  
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### **Plant material and growth conditions**

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4 The plant materials of *S. alfredii* were collected from an old Pb/Zn mining area  
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6 in Quzhou City, Zhejiang Province, China. The sampling site is located at 118°, 56'  
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8 east longitude and 29°, 17' north latitude. It refers to the previous study for more  
9  
10 information about this plant.<sup>13</sup> After having collected the plants, the rest of the  
11  
12 experiments were undertaken in our laboratory. The shoot tops of *S. alfredii* were cut  
13  
14 and cultured in a greenhouse in Shenyang University for 2 months. Healthy and  
15  
16 uniform *S. alfredii* seedlings were selected and planted in the basal nutrient solution.  
17  
18 The nutrition solution used was the half-strength Hoagland-Arnon solution,<sup>19</sup> which  
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20 comprised of 3 mM KNO<sub>3</sub>, 0.5 mM NH<sub>4</sub>H<sub>2</sub>PO<sub>4</sub>, 2.0 mM Ca(NO<sub>3</sub>)<sub>2</sub>, 1.0 mM  
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22 MgSO<sub>4</sub>·7H<sub>2</sub>O, 4.5 μM MnCl<sub>2</sub>·4H<sub>2</sub>O, 23 μM H<sub>3</sub>BO<sub>3</sub>, 0.4 μM ZnSO<sub>4</sub>·7H<sub>2</sub>O, 0.15 μM  
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24 CuSO<sub>4</sub>·5H<sub>2</sub>O, 0.05 μM H<sub>2</sub>MoO<sub>4</sub>·H<sub>2</sub>O, and 22 μM EDTA-Fe. The nutrient solution  
25  
26 was aerated continuously and renewed every 4 days, with its pH was adjusted to 6.0  
27  
28 using 0.1 M NaOH or HCl every day. The plants were grown under greenhouse  
29  
30 conditions with natural light. The temperature varied from 10 to 20 °C. Until the  
31  
32 relatively flourishing roots grow out, also were two weeks of pre-culture, *S. alfredii*  
33  
34 were selected for 3 Cd treatments: 0 (control), 5, 10 μM Cd, and Cd was supplied as  
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36 CdCl<sub>2</sub>. There were 33 pots (1 piece per pot) in total, with 11 replicates for each Cd  
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38 treatment.  
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### 51 52 **Collection of root exudates**

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54 After having grown for 4 and 8 days in the nutrient solution spiked with Cd salts  
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56 without renewal, the plants were transplanted to sterilized pots with 50 mL deionized  
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58 water per pot to collect root exudates for 6 h. Sample preparation, derivatization and  
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4 the GC-MS analysis were based on Suzuki et al.(2009)<sup>16</sup> and Lisec et al.(2006).<sup>20</sup> The  
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7 root exudates from each pot were frozen in liquid nitrogen and freeze-dried for 2 days.  
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10 The dried residue was resuspended in 100 mL of deionized water and freeze-dried  
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12 again. The dried residue was redissolved in 10 mL of cold MeOH, then blown to  
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14 dryness under a gentle nitrogen flow, and reconstituted in 1 mL of n-hexane used for  
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17 the GC-MS analysis.  
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### 19 20 **GC-MS analysis of root exudates**

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23 Samples were derivatized by 40  $\mu\text{L}$  of methoxyamine hydrochloride (20 mg  
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25  $\text{mL}^{-1}$  in pyridine, 2 h, 37  $^{\circ}\text{C}$ ) and 70  $\mu\text{L}$  N-methyl-N-(trimethylsilyl)  
26  
27 trifluoroacetamide (MSTFA) (30 min, 37  $^{\circ}\text{C}$ ). 1  $\mu\text{L}$  of the sample was injected into  
28  
29 the GC in the splitless mode. The GC analysis was carried out on a TR-5MS capillary  
30  
31 column (30 m, 0.25 $\mu\text{m}$ , 0.25 mm, Thermo Fisher, USA). The injection, interface and  
32  
33 ion-source temperatures were adjusted to 230, 250 and 210  $^{\circ}\text{C}$ , respectively. The gas  
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35 flow rate was 1  $\text{mL min}^{-1}$ , the column temperature was held for 1 min at 70  $^{\circ}\text{C}$ , 6 min  
36  
37 ramp to 76  $^{\circ}\text{C}$ , 50 min ramp to 330  $^{\circ}\text{C}$ , 10 min at 330  $^{\circ}\text{C}$ . The column end was  
38  
39 introduced into an ion trap mass spectrometer. Mass spectra were recorded at 2 scans  
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41  $\text{s}^{-1}$  with a  $m/z$  50-600 scanning range.  
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### 49 50 **Validation of the assay method**

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52 Because of the inhomogeneity of the plant growth, a large number of the same  
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54 sample of root exudates is difficult to acquire, and this assay method is not aimed at  
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56 the specific target compounds, the validation of this assay method is only involved the  
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58 instrument performance. A post-preparative sample of root exudates was selected as  
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4 the quality control (QC) sample to monitor the reproducibility and stability of the  
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7 GC-MS analysis method. The intra-day precision was determined by detecting the QC  
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10 sample seven times on one day. The inter-day precision was determined on six  
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12 continuous days. During the analysis, QC sample is inserted at intervals and analyzed  
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15 once every 10 samples during the analytical sequence to test the system stability.  
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### 17 **Data analysis**

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20 The raw GC-MS chromatogram was automatically analyzed using  
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23 the automatic mass spectral deconvolution and identification system (AMDIS, version  
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26 2.71) which obtained from NIST, and compared with the database of metabolites in  
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29 plants (Fiehn and Golm Metabolome Database (GMD)). If the similarity was greater  
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32 than 70%, the compounds were being identified. Then use the metabolomics  
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35 ion-based data extraction algorithm (MET-IDEA, version 2.08) to extract and process  
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38 the AMDIS output, 68 compounds were detected in one GC-MS scan. The parameters  
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41 for MET-IDEA were: (i) chromatography: GC; average peak width, 0.1; minimum  
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44 peak width, 0.3; maximum peak width, 6; peak start/stop slope, 1.5; adjusted retention  
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47 time accuracy, 0.95; peak overload factor, 0.3; (ii) mass spec: trap; mass accuracy, 0.1;  
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50 mass range, 0.5; (iii) AMDIS: exclude ion list, 73, 147, 281, 341, 415; lower mass  
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53 limit, 50; ions per component, 1. The peaks of 11.09 min which detected in all  
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56 samples were used for retention time calibration.

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59 Normalize the peak area of the identified root exudates, the normalization was  
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according to each compound based on the first appeared by the following and the  
normalization method is that the peak area values were divided by the average of the

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4 compound which first appeared, then import to the statistics software SIMCA-P 13.0.  
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7 The principal component analysis (PCA) and orthogonal partial least-squares  
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9 discrimination analysis (OPLS-DA) were used to analyze and explain the variation of  
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11 root exudates from *S. alfredii* under the Cd stress.  
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## 14 **Results**

### 15 **The results of GC-MS analysis**

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18 The GC-MS total ion chromatogram of root exudates from the hyperaccumulator  
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20 *S. alfredii* at 0, 5 and 10  $\mu\text{mol/L}$  Cd treatment for 4 and 8 days, as show in Fig.1.  
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23 Through analyzed the GC-MS data, 68 compounds were detected and identified.  
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26 Including lactic acid, oxalic acid, succinic acid and other low-molecular-weight  
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28 organic acids, l-valine, l-alanine, l-serine, l-glycine and other amino acids, xylose,  
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30 fructose, glucose and other sugars, dodecanol, ribitol, d-pinitol, cholesterol and other  
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32 alcohols, and other small-molecule metabolites, such as phosphoric acid, lauric acid  
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34 and oleic acid.  
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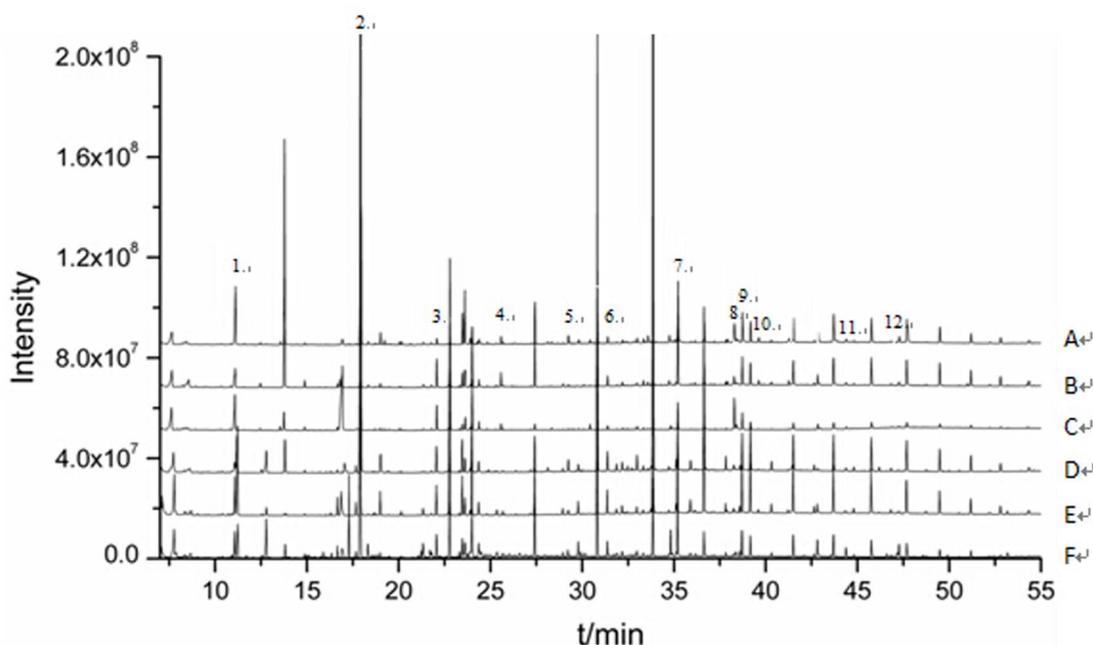


Fig.1 GC-MS TIC chromatogram of root exudates of the hyperaccumulator *S. alfredii*

A. 0  $\mu\text{mol/L}$  Cd treatment for 4 days; B. 5  $\mu\text{mol/L}$  Cd treatment for 4 days; C. 10  $\mu\text{mol/L}$  Cd treatment for 4 days; D. 0  $\mu\text{mol/L}$  Cd treatment for 8 days; E. 5  $\mu\text{mol/L}$  Cd treatment for 8 days; F. 10  $\mu\text{mol/L}$  Cd treatment for 8 days

Some identified compounds : 1. Lactic acid -2TMS; 2. Glycerol-3TMS; 3.

Putrescine-4TMS; 4. Threonic acid-4TMS; 5. Ribitol-5TMS; 6. Tetradecanoic

acid-1TMS; 7. Hexadecanoic acid-1TMS; 8. Oleic acid-1TMS; 9. Octadecanoic

acid-1TMS; 10. n-Docosane; 11. 1-Monohexadecanoylglycerol-2TMS; 12.

Trehalose-8TMS.

### Assay validation

The intra-day precision, the absolute deviation of retention time and the relative standard deviation (RSD) of peak area, for the identified peaks was 0.01-0.15 min and 1.8-6.7 %, respectively. The inter-day precision, the absolute deviation of retention time and the RSD of peak area, for the identified peaks was 0.01-0.25 min and

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4 3.2-8.6 %. The system stability, the absolute deviation of retention time and the RSD  
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7 of peak area, for the identified peaks was 0.01-0.3 min and 4.3-8.9 %, respectively.  
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10 These indicators were all within acceptable levels and tolerances. All these results  
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12 obtained indicated that the method was robust with good repeatability and stability.  
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### 15 **Pattern recognition analysis of identified root exudates**

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17 The identified root exudates were analyzed using the unsupervised principal  
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19 component analysis (PCA) method and supervised orthogonal partial least-squares  
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21 discrimination analysis (OPLS-DA) method.  
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25 The results of the PCA on the identified root exudates were shown in Fig.2A.  
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28 The representative points of the samples were mapped in the space spanned by the  
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30 first two principal components PC1 versus PC2. This scores plot is illustrated a  
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32 reasonable clustering appearing according to Cd exposed concentration and time.  
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PCA unravelled the existence of differences in the composition of root exudates  
(24.8% of the variance was captured by the first PC) from *S. alfredii* under different  
Cd exposed concentration and time.

In order to enhance the separation effect obtained from the PCA model, the  
supervised clustering method OPLS-DA was used. The results of the OPLS-DA on  
the identified root exudates were shown in Fig.2B. As you can be seen in the scores  
plot, a better separation was attained after OPLS-DA for the identified root exudates  
under different Cd exposed concentration and time.

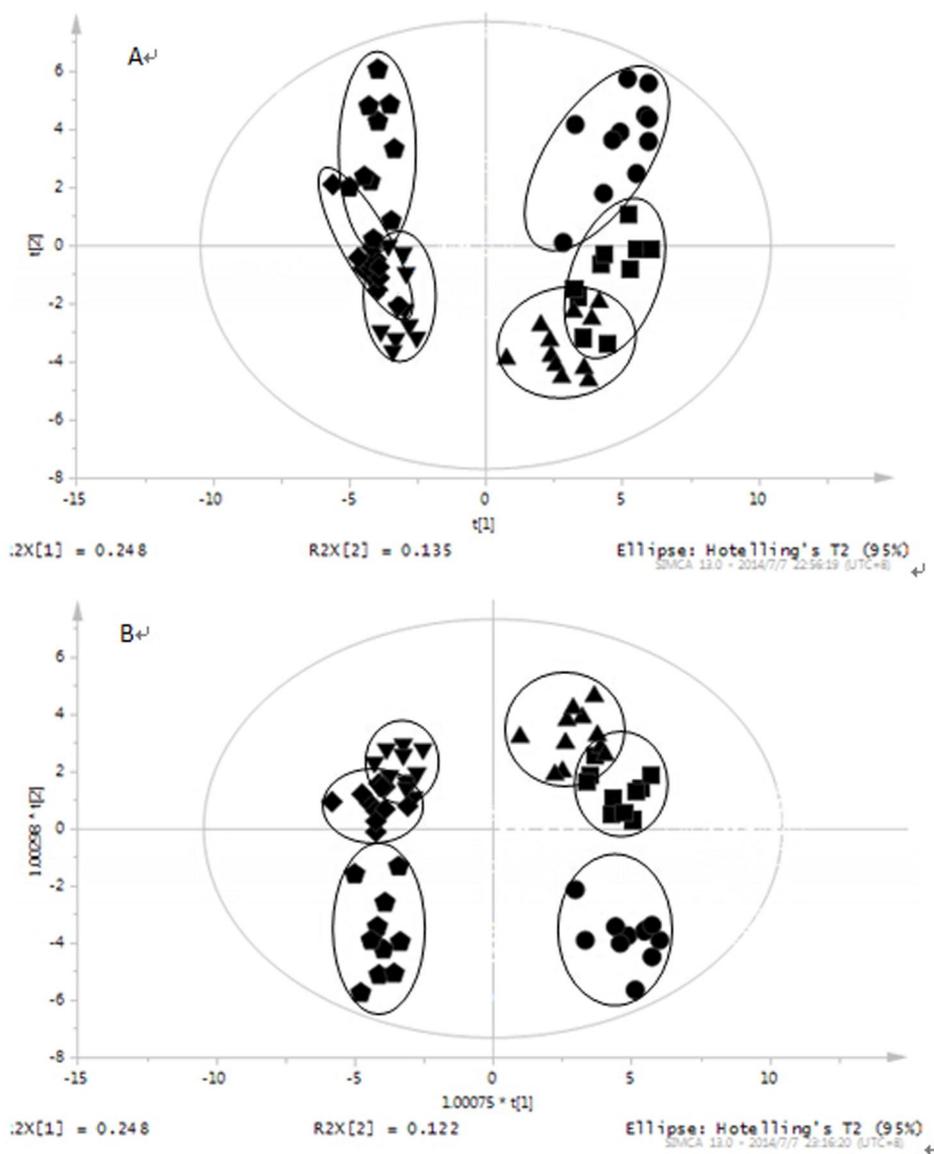
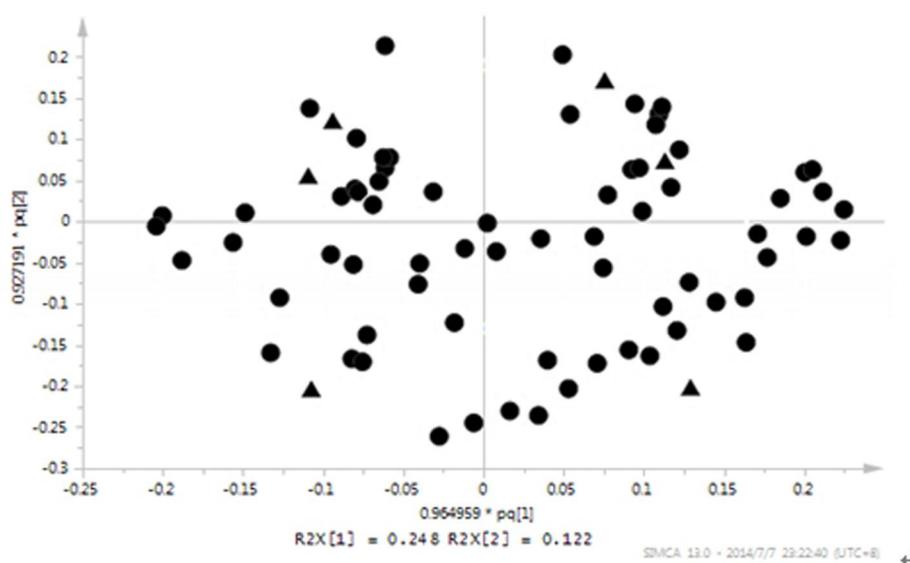


Fig.2 Principal component analysis (PCA) and orthogonal partial least-squares discrimination analysis (OPLS-DA) scores plots of root exudates of the hyperaccumulator *S. alfredii*

A. PCA scores plots, (○) 0  $\mu\text{mol/L}$  Cd treatment for 4 days, (□) 5  $\mu\text{mol/L}$  Cd treatment for 4 days, (△) 10  $\mu\text{mol/L}$  Cd treatment for 4 days, (◇) 0  $\mu\text{mol/L}$  Cd treatment for 8 days, (●) 5  $\mu\text{mol/L}$  Cd treatment for 8 days, (◆) 10  $\mu\text{mol/L}$  Cd treatment for 8 days; B. OPLS-DA scores plots

To find out the compounds of the identified root exudates resulting in the

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4 separation among the different Cd exposed concentration and time, the loadings plot  
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6 of the related OPLS-DA model was conducted (Fig.3). Combined with the loadings  
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8 plot, the variable importance factor (VIP, larger than 1, are the most relevant for  
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10 explaining the group) values of the OPLS-DA and analysis of variance (ANOVA,  
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12  $p < 0.05$ ), 15 compounds caused the separation among the different Cd exposed  
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14 concentration and time were found (Table 1).  
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39 Fig.3 Orthogonal partial least-squares discrimination analysis (OPLS-DA) loading  
40 plots of root exudates of the hyperaccumulator *S. alfredii*  
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42 ( )X, the root exudate variable; ( )Y, the group variable  
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Table 1 Potential biomarkers in root exudates of the hyperaccumulator *S. alfredii*

Compounds	0 $\mu\text{mol/L Cd}$	5 $\mu\text{mol/L Cd}$	10 $\mu\text{mol/L Cd}$	0 $\mu\text{mol/L Cd}$	5 $\mu\text{mol/L Cd}$	10 $\mu\text{mol/L Cd}$
	treatment for	treatment for	treatment for	treatment for	treatment for	treatment for
	4 days( n=11 )	4 days( n=11 )	4 days( n=11 )	8 days( n=11 )	8 days( n=11 )	8 days( n=11 )
Oxalic acid	1.00 $\pm$ 0.16	1.09 $\pm$ 0.11	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.48 $\pm$ 0.11	0.00 $\pm$ 0.00
2-Hydroxyacetic acid	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	1.00 $\pm$ 0.15	0.00 $\pm$ 0.00
Octanol	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	1.00 $\pm$ 0.41
Benzoic acid	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	1.00 $\pm$ 0.08	0.00 $\pm$ 0.00
2-Hydroxypentanoic acid	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	1.00 $\pm$ 0.33	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00
Succinic acid	1.00 $\pm$ 0.31	0.83 $\pm$ 0.09	0.77 $\pm$ 0.11	0.83 $\pm$ 0.11	0.83 $\pm$ 0.11	0.00 $\pm$ 0.00
Fumaric acid	0.00 $\pm$ 0.00	1.00 $\pm$ 0.12	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00
L-Serine	0.00 $\pm$ 0.00	1.00 $\pm$ 0.23	0.43 $\pm$ 0.12	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00
Decanoic acid	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	1.00 $\pm$ 0.12	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00
Putrescine	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	1.00 $\pm$ 0.95
Fructose	1.00 $\pm$ 0.21	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	1.60 $\pm$ 0.54	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00
Mannitol	1.00 $\pm$ 0.18	0.30 $\pm$ 0.04	0.10 $\pm$ 0.03	0.08 $\pm$ 0.02	0.04 $\pm$ 0.02	0.00 $\pm$ 0.00
1-Monooctadecanoylglycerol	1.00 $\pm$ 0.25	0.73 $\pm$ 0.46	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.43 $\pm$ 0.09	1.87 $\pm$ 0.85
Octacosanol	1.00 $\pm$ 0.25	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00
beta-Sitosterol	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	0.00 $\pm$ 0.00	1.00 $\pm$ 0.37	0.00 $\pm$ 0.00

The value is the ratio of the peak area of the compound which have the same retention time among the different treatments, mean  $\pm$

standard deviation.

### Change trends of the relative content of the potential biomarkers

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4 Under the Cd treated for 4 and 8 days, low concentration Cd can promote the  
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6 secretion of oxalic acid, high concentration Cd can inhibit its secretion. But compared  
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8 with 4 days, the secretion of oxalic acid reduced significantly at 8 days. This change  
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10 trend indicated that oxalic acid might play an important role in mobilizing Cd. At the  
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12 low Cd exposed concentration, the secretion of oxalic acid could increase the  
13  
14 available Cd contents in the matrix, and then increased the adsorption and  
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16 accumulation of Cd. At the high Cd exposed concentration, through reducing the  
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18 secretion of oxalic acid to avoid Cd poisoning. Meanwhile, the secretion of oxalic  
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20 acid is greatly influenced by the plant growth cycle.  
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28 When Cd treated for 4 days, the secretion of 1-monooctadecanoylglycerol  
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30 reduced with an increase in the Cd exposed level, but increased when Cd treated for 8  
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32 days. This indicated that 1-monooctadecanoylglycerol may be attributed to  
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34 mobilization of Cd. At the beginning of the Cd treatment, through reducing the  
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36 secretion of 1-monooctadecanoylglycerol to decrease the available Cd contents in the  
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38 matrix, and then reduced the toxic effect. With the increase of treatment duration, *S.*  
39  
40 *alfredii* adapted the Cd stresses, and then through increasing the secretion of  
41  
42 1-monooctadecanoylglycerol to mobilize and accumulate Cd.  
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50 At the beginning of the Cd treatment, 2-hydroxyacetic acid, benzoic acid and  
51  
52 beta-sitosterol are not secreted. But with the increase of treatment duration, namely  
53  
54 Cd treated for 8 days, these three compounds are secreted in the low exposed  
55  
56 concentration (5  $\mu\text{mol/L}$  Cd). But these three compounds are not secreted in the high  
57  
58 exposed concentration (10  $\mu\text{mol/L}$  Cd). This indicated that these three compounds  
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4 maybe play an important role in mobilizing Cd. At 5  $\mu\text{mol/L}$  Cd treatment, through  
5  
6 secreting these three compounds to mobilize Cd, and then adsorb and accumulate Cd.  
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10 But at 10  $\mu\text{mol/L}$  Cd treatment, though not secreting these three compounds to  
11  
12 decrease the available Cd contents in the matrix, and then reduce the toxic effect.  
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15 Octanol and putrescine are secreted only at 10  $\mu\text{mol/L}$  Cd treatment for 8 days.

16  
17 This might be because octanol and putrescine can stabilize Cd. Through the secretion  
18  
19 of octanol and putrescine to reduce the toxic effect that *S. alfredii* growth under the  
20  
21 high concentration for a long time.  
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25 At the beginning of the Cd treatment, Cd can promote the secretion of fumaric  
26  
27 acid and l-serine. But at the high Cd exposed concentration (10  $\mu\text{mol/L}$  Cd), the  
28  
29 secretion of these two compounds are reduced. And when Cd treated for 8 days, *S.*  
30  
31 *alfredii* not secreted fumaric acid and l-serine. This implied that fumaric acid and  
32  
33 l-serine might play an important role in mobilizing Cd. The low concentration Cd and  
34  
35 short time treatment can stimulate the secretion of fumaric acid and l-serine to  
36  
37 mobilize Cd and increase the available Cd contents, and then promote the  
38  
39 accumulation of Cd. The high concentration Cd or long time treatment can inhibit the  
40  
41 secretion of fumaric acid and l-serine to decrease the available Cd contents and reduce  
42  
43 the toxic effect.  
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51 Decanoic acid and 2-hydroxypentanoic acid are only secreted without Cd  
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53 treatment for 8 days. This showed that the secretion of decanoic acid and  
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55 2-hydroxypentanoic acid is related to the plant growth cycle. And due to the secretion  
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57 of these two compounds reduced to zero when Cd treated for 8 days, this implied that  
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4 decanoic acid and 2-hydroxypentanoic acid may be can mobilize Cd. When Cd treated,  
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7 *S. alfredii* decreased the secretion of these two compounds to reduce the available Cd  
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9 contents and the toxic effect.  
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11  
12 Fructose and octacosanol are only secreted without Cd treatment. When Cd  
13  
14 treated, the secretion of these two compounds decreased to zero. This implied that  
15  
16 fructose and octacosanol may be able to mobilize Cd. When Cd treated, *S. alfredii*  
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18 decreased the secretion of these two compounds to reduce the available Cd contents  
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20 and the toxic effect.  
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26 When Cd treated for 4 days, the secretion of mannitol and succinic acid were  
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28 reduced with an increase in the concentration of Cd treatments. Also, the secretion of  
29  
30 mannitol and succinic acid were reduced with an increase in the concentration of Cd  
31  
32 treatments when Cd treated for 8 days. However, compared with 4 days, the secretion  
33  
34 of these two compounds were lower when Cd treated for 8 days. This indicated that  
35  
36 the secretion of mannitol and succinic acid is related to the plant growth cycle and the  
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38 Cd stresses.  
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## 43 44 **Discussions**

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47 In this study, through the PCA and OPLS-DA of the identified root exudates, we  
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49 detected that the quantity or composition of root exudates released from *S. alfredii*  
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51 under different Cd exposed concentration and time are obviously different. These  
52  
53 results were accord with previous studies that the quantity and composition of root  
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55 exudates could be influenced by many factors including the soil structure, the  
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57 presence of microorganisms, the plant species as well as their developmental stage,  
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4 nutritional status and environment stresses.<sup>21, 22</sup>  
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7 Through the analysis of the loadings plot, the VIP values of OPLS-DA and  
8 ANOVA, 15 compounds resulting in the separation among different Cd exposed  
9 concentration and time were found out. This indicated that these 15 compounds might  
10 play a main role in tolerating or accumulating heavy metal Cd. In these potential  
11 biomarkers, most are the organic acids and amino acids. Organic acids and amino  
12 acids play an extremely important role in the process of hyperaccumulator tolerates  
13 and accumulates the heavy metal.<sup>23-25</sup> Other potential biomarkers were rarely involved  
14 in the previous studies.  
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28 The change trends of the relative content of these potential biomarkers revealing  
29 their probable roles in the area of *S. alfredii* tolerates or accumulates the heavy metal  
30 Cd. Oxalic acid, 2-hydroxyacetic acid, benzoic acid, fumaric acid, decanoic acid,  
31 z-hydroxypentanoic acid, l-serine, fructose, octacosanol, beta-sitosterol and  
32 1-monooctadecanoylglycerol might be mobilizing Cd. Octanol and putrescine may be  
33 able to stabilize Cd.  
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44 Oxalic acid can enhanced the adsorption and desorption of Cd on goethite, and  
45 can via an oxalate-bridge between the surface and the metallic cation to form a  
46 Cd-oxalate complex.<sup>26</sup> Organic acid, especially oxalic acid and citric acid, has strong  
47 metal-chelating properties and the ligand-promoted mechanism was the main  
48 mechanism of mineral dissolution, and indicated that oxalic acid was the main  
49 mineral-transforming agent.<sup>27</sup> The addition of exogenous oxalic acid increased the Cd  
50 accumulation in *Boehmeria nivea* (L.) Gaud. and improved the Cd translocation  
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4 efficiency from root to shoot.<sup>28</sup> So oxalic acid is a very important root secreted  
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7 substances, it can mobilize Cd around of rhizosphere, play an important role in the  
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10 phytoremediation of Cd polluted soil.

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12 Fumaric acid was found in the root exudates from the arsenic hyperaccumulator  
13  
14 *Pteris vittata*<sup>29</sup>, 2-hydroxyacetic acid was detected in the tobacco and sunflower  
15  
16 rhizosphere soils when they treated with Cd<sup>30</sup>, benzoic acid was found in the root  
17  
18 exudates of maize crop under 500 ppm Cr treatment<sup>31</sup> and l-serine was detected in the  
19  
20 leaf of *Thlaspi* hyperaccumulators under Ni and Zn stress condition<sup>32</sup>, but because of  
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22 their concentration are little, did not arouse enough attention and in-depth analysis  
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28 from past researchers. In this study, we do not pay attention to their absolute  
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30 concentrations, through metabolomics analysis we found that these compounds are  
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32 the potential biomarkers which resulted in the separation among different Cd  
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34 treatments, so the role of these compounds in the phytoremediation of heavy metal  
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36 polluted soil should arouse our attention.  
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42 Under heavy metal stress, the secretion of citric acid, malic acid and tartaric acid  
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44 was increased<sup>33,34</sup>. Citric acid and malic acid can form complexes with Ni<sup>35,36</sup>, and  
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46 the addition of exogenous citric acid can promote uptake of Cd by indian mustard<sup>37</sup>.  
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48 But in this study, these compounds were not selected as the potential biomarkers.  
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50 Perhaps they were not detected or identified in this method, or the changes of their  
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52 concentrations were not the main factors which resulting in the separation among  
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54 different Cd treatments.  
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Previous researches of root exudates of hyperaccumulator did not pay attention

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4 to these compounds, such as decanoic acid, 2-hydroxypentanoic acid, fructose,  
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7 octanol, and so on. But in this study, we thought that they may play a role in the  
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10 process of hyperaccumulator tolerates and accumulates the heavy metal by  
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12 metabonomics analysis, and we presumed their role, mobilize or stabilize heavy  
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15 metal.

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18 To our knowledge, the components of plant root exudates are multitudinous and  
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20 complex.<sup>38-40</sup> But in this study, only 68 compounds were identified. Mainly it is  
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23 because the limitation of the experiment method, the sensitivity of GC-MS and the  
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26 library of MS. In the future, we will use the more excellent methods such as  
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29 UPLC-Q-TOF or NMR to analyze the components of root exudates and verify the  
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32 role of these potential biomarkers by leaching experiments and pot soil culture  
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35 experiments. Then thoroughly research the role and mechanisms of root exudates  
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38 from hyperaccumulators that can tolerate and accumulate heavy metals.

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