

ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/chemcomm

COMMUNICATION

Fabrication of a high-performance dye-sensitized solar cell with 12.8% conversion efficiency using organic silyl-anchor dyes

Kenji Kakiage,^a Yohei Aoyama,^a Toru Yano,^{*a} Keiji Oya,^a Toru Kyomen^b and Minoru Hanaya^{*b}

Received (in XXX, XXX) XXth XXXXXXXX 2015, Accepted XXth XXXXXXXXXX 20XX

DOI: 10.1039/b000000x

The co-sensitization of organic silyl-anchor dyes in dye-sensitized solar cells (DSSCs) using carbazole and coumarin dyes with organosilicon tethers for binding to titanium dioxide has been examined. We have succeeded in fabricating a high-performance DSSC with a light-to-electric energy conversion efficiency of 12.8% under one sun simulated solar irradiation.

Dye-sensitized solar cells (DSSCs), which are composed of mesoporous nanocrystalline-TiO₂ thin layers modified by photosensitizing dyes as working electrodes, redox electrolytes and counter electrodes, have become a promising alternative photovoltaic technology to conventional inorganic solar cells because of their potential low costs, relatively high solar light-to-electric energy conversion efficiencies (η) and fine photovoltaic properties in low-light intensities.¹⁻³ In DSSCs, η of 12-13% under the simulated sunlight irradiation at one sun have been reported up to now by the photosensitization using polypyridyl and porphyrin complexes of metals such as ruthenium and zinc.²⁻⁷

Recently, we have reported a metal-free carbazole/alkyl-functionalized oligothiophene/alkoxysilyl-anchor moiety type compound, **ADEKA-1** (Fig. 1a), as the organic photosensitizing dye for DSSCs. The **ADEKA-1**-sensitized solar cell exhibited the highest η of 12.5% among the cells with organic dyes reported so far with the open-circuit photovoltage (V_{oc}) higher than 1 V by using a cobalt(III/II) tris(5-chloro-1,10-phenanthroline) complex ([Co(Cl-phen)₃]^{3+/2+}) as the redox electrolyte.⁴ Besides the high photovoltaic performance, the TiO₂ photoelectrode sensitized with **ADEKA-1** possesses much higher durability to mixed solvents containing nitrile and water than those with conventional carboxy-anchor dyes. In DSSCs co-sensitizations by using two kinds of dyes have been demonstrated to be effective to improve the energy conversion efficiencies,⁶⁻¹¹ and the durability of the photoelectrode allows co-adsorptions of plural sensitizing dyes to the electrode for producing the co-sensitization effect.

In this work, therefore, we examined the co-sensitization in the DSSCs with **ADEKA-1** by using the alkoxysilyl-anchor coumarin dye of **SFD-5** (Fig. 1b) to confirm the possibility of the co-sensitization and to improve the photovoltaic performance of the cells. While **ADEKA-1** has major absorption band in the visible region from 400 to 600 nm with the maximum molar absorption coefficient (ϵ_{max}) of 43,200 at λ_{max} = 498 nm, **SFD-5** has a strong absorption band due to the π - π^* transition from 370 to 480 nm (ϵ_{max} = 58,400 at λ_{max} = 440 nm) as shown in Fig. S1. The oxidation potential (E_{ox} : approximately the energy level of

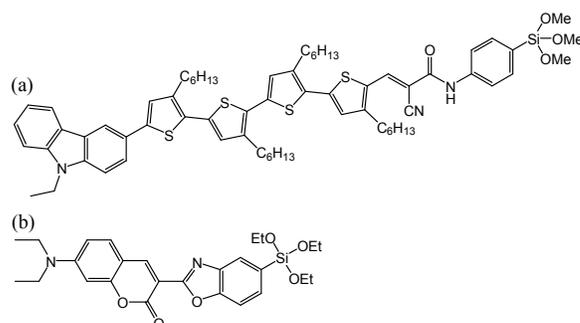


Fig. 1 Molecular structures of metal-free organic silyl-anchor dyes: (a) **ADEKA-1** and (b) **SFD-5**.

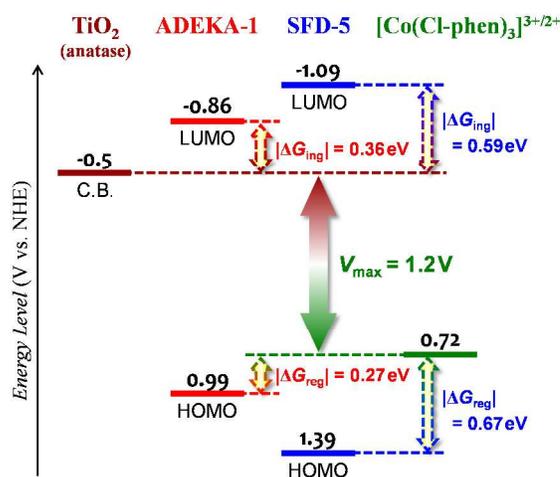


Fig. 2 Schematic energy diagram of the DSSC composed of anatase-TiO₂, **ADEKA-1**, **SFD-5** and [Co(Cl-phen)₃]^{3+/2+}.

HOMO) of **SFD-5** was estimated to be 1.39 V vs. NHE, which is more positive enough than the redox potential of 0.72 V vs. NHE of [Co(Cl-phen)₃]^{3+/2+} as is same in **ADEKA-1**.^{4,12,13} Schematic energy diagram of the DSSC is shown in Fig. 2. The positive E_{ox} of **SFD-5** allows the utilization of [Co(Cl-phen)₃]^{3+/2+} as the redox electrolyte in the cells to produce a high photovoltage. From the difference in the absorption characteristics and the positive E_{ox} , **SFD-5** is considered as an appropriate co-sensitizer for **ADEKA-1**.

We fabricated the cells, equipped with an antireflection film, composed of the TiO₂ electrodes photosensitized by **ADEKA-1**

with/without **SFD-5** and an experimentally optimized [Co(Cl-phen)₃]^{3+/2+} electrolyte solution. The *IPCE* spectra of the **ADEKA-1**-sensitized cells with and without **SFD-5** are shown in Fig. 3a. The cell with **SFD-5** exhibited better *IPCE* values than the cell without **SFD-5** in the range between 400 and 530 nm which is the light-absorption region of **SFD-5** (Fig. S1), and a maximum *IPCE* value of the cell with **SFD-5** was reached to 88%. On the other hand, *IPCE* values from 550 to 800 nm were almost same in the two cells. While in **ADEKA-1** the alkoxy-silyl-anchor moiety links to the chromophore (carbazole/alkyl-functionalized oligothiophene) via the phenyl-amide moiety, **SFD-5** has the alkoxy-silyl-anchor moiety bound directly to the coumarin-based chromophore and a higher excited-state oxidation potential (E_{ox}^* : approximately the energy level of LUMO) of -1.09 V vs. NHE than that of **ADEKA-1** (-0.86 V vs. NHE) (Fig. 1 and 2). Thus **SFD-5** is expected to have a higher electron injection efficiency from the light-excited state to the TiO₂ conduction band than **ADEKA-1**, and the improvement of the *IPCE* observed in the light-absorption region of **SFD-5** is considered as the result of the co-sensitization effect by **SFD-5**. The obtained results indicate also that the co-sensitization with **ADEKA-1** and **SFD-5** functioned effectively without the negative interaction between the two organic silyl-anchor dyes for the improvement of the photovoltaic performance of the cell.

The photovoltaic parameters, short-circuit photocurrent density (J_{sc}), V_{oc} , fill factor (*FF*) and η , of the **ADEKA-1**-sensitized cells with and without **SFD-5** were assessed from the photocurrent-voltage (*J-V*) curves under the simulated AM-1.5G solar irradiation with the one sun intensity of 100 mW cm⁻² (Fig. 3b). The results of the *J-V* measurements are summarized in Table S1. As expected from the *IPCE* spectra (Fig. 3a), the J_{sc} of the cells co-sensitized by **ADEKA-1** and **SFD-5** was improved to 16.0±0.1 mA cm⁻² from that of the cells photosensitized only by **ADEKA-1** of 15.6±0.1 mA cm⁻². The V_{oc} and *FF* values were almost the same in the two kinds of cells showing that **SFD-5** did not affect those photovoltaic properties. Due to the increment of the photocurrent, the η was improved from 12.5% to 12.8% by the co-sensitization of **ADEKA-1** and **SFD-5** under the one sun irradiation (AM-1.5G, 100 mW cm⁻²).

In conclusion, the examination of the co-sensitization using two metal-free organic silyl-anchor dyes of **ADEKA-1** and **SFD-5** showed that **SFD-5** works as co-sensitizer to **ADEKA-1** and we succeeded in obtaining the high *IPCE* up to 88%, V_{oc} over 1 V and 12.8% conversion efficiency in the DSSC containing the cobalt(III/II) complex electrolyte. The energy conversion efficiency achieved here is the highest among those reported so far for the DSSCs using metal-free organic photosensitizing dyes.^{2-7,14-16} The result shows the potential of organic silyl-anchor dyes^{4,12,13,17-19} as the photosensitizers for DSSCs and a possibility of co-sensitization by plural organic silyl-anchor dyes for further improvement of the photovoltaic performance of DSSCs. Detailed molecular design of organic silyl-anchor dyes and co-sensitization using those dyes are a promising way to obtain higher energy conversion performance in DSSCs, and such investigations are currently underway in our group.

This work was partly supported by the “Element Innovation” Project by Ministry of Education, Culture, Sports, Science & Technology in Japan.

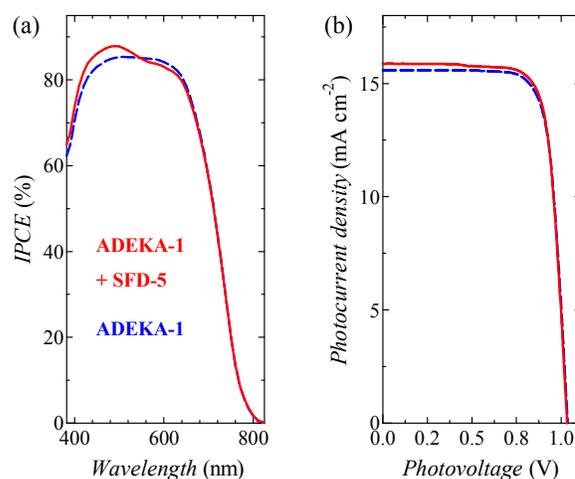


Fig. 3 (a) *IPCE* spectra and (b) *J-V* properties under the simulated one sun irradiation (AM-1.5G, 100 mW cm⁻²) of the cells photosensitized by **ADEKA-1** with/without **SFD-5**.

Notes and references

^a Environmental & Energy Materials Laboratory, ADEKA

CORPORATION, 7-2-35 Higashiogu, Arakawa, Tokyo 116-8554, Japan.

E-mail: yanotoru@adeka.co.jp

^b Division of Molecular Science, Graduate School of Science and Technology, Gunma University, 1-5-1 Tenjin-cho, Kiryu, Gunma 376-8515, Japan. E-mail: mhanaya@gunma-u.ac.jp

- B. O'Regan and M. Grätzel, *Nature*, 1991, **353**, 737.
- M. Grätzel, *Acc. Chem. Res.*, 2009, **42**, 1788.
- A. Hagfeldt, G. Boschloo, L. Sun, L. Kloo and H. Pettersson, *Chem. Rev.*, 2010, **110**, 6595.
- K. Kakiage, Y. Aoyama, T. Yano, T. Otsuka, T. Kyomen, M. Unno and M. Hanaya, *Chem. Commun.*, 2014, **50**, 6379.
- S. Mathew, A. Yella, P. Gao, R. Humphry-Baker, B. F. E. Curchod, N. Ashari-Astani, I. Tavernelli, U. Rothlisberger, M. K. Nazeeruddin and M. Grätzel, *Nat. Chem.*, 2014, **6**, 242.
- H. Ozawa, Y. Okuyama and H. Arakawa, *ChemPhysChem*, 2014, **15**, 1201.
- A. Yella, H.-W. Lee, H. N. Tsao, C. Yi, A. K. Chandiran, M. K. Nazeeruddin, E. W.-G. Diao, C.-Y. Yeh, S. M. Zakeeruddin and M. Grätzel, *Science*, 2011, **334**, 629.
- Y. Numata, S. Zhang, X. Yang and L. Han, *Chem. Lett.*, 2013, **42**, 1328.
- N. Shibayama, H. Ozawa, M. Abe, Y. Ooyama and H. Arakawa, *Chem. Commun.*, 2014, **50**, 6398.
- P. J. Holliman, M. L. Davies, A. Connell, B. Vaca Velasco and T. M. Watson, *Chem. Commun.*, 2010, **46**, 7256.
- P. J. Holliman, K. J. Al-Salihi, A. Connell, M. L. Davies, E. W. Jones and D. A. Worsley, *RSC Adv.*, 2014, **4**, 2515.
- K. Kakiage, T. Tokutome, S. Iwamoto, T. Kyomen and M. Hanaya, *Chem. Commun.*, 2013, **49**, 179.
- K. Kakiage, T. Kyomen and M. Hanaya, *Chem. Lett.*, 2013, **42**, 1520.
- M. Liang and J. Chen, *Chem. Soc. Rev.*, 2013, **42**, 3453.
- M. Zhang, Y. Wang, M. Xu, W. Ma, R. Li and P. Wang, *Energy Environ. Sci.*, 2013, **6**, 2944.
- S. Ahmad, E. Guillén, L. Kavan, M. Grätzel and M. K. Nazeeruddin, *Energy Environ. Sci.*, 2013, **6**, 3439.
- S. K. Matta, K. Kakiage, S. Makuta, A. Veamatahau, Y. Aoyama, T. Yano, M. Hanaya and Y. Tachibana, *J. Phys. Chem. C*, 2014, **118**, 28425.
- K. Kakiage, Y. Aoyama, M. Yamamura, T. Yano, M. Unno, T. Kyomen and M. Hanaya, *Silicon*, 2014, **6**, 123.
- K. Kakiage, M. Yamamura, E. Fujimura, T. Kyomen, M. Unno and M. Hanaya, *Chem. Lett.*, 2010, **39**, 260.