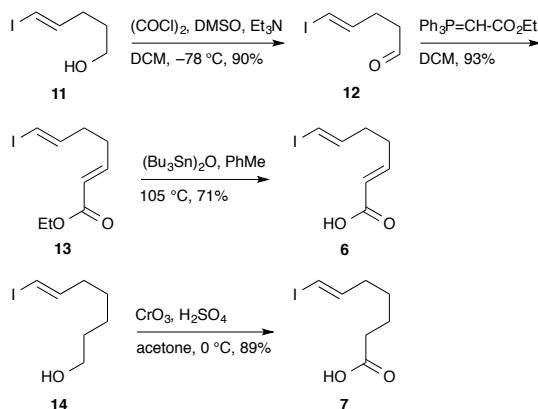


**A Concise Formal Total Synthesis of Lactimidomycin**

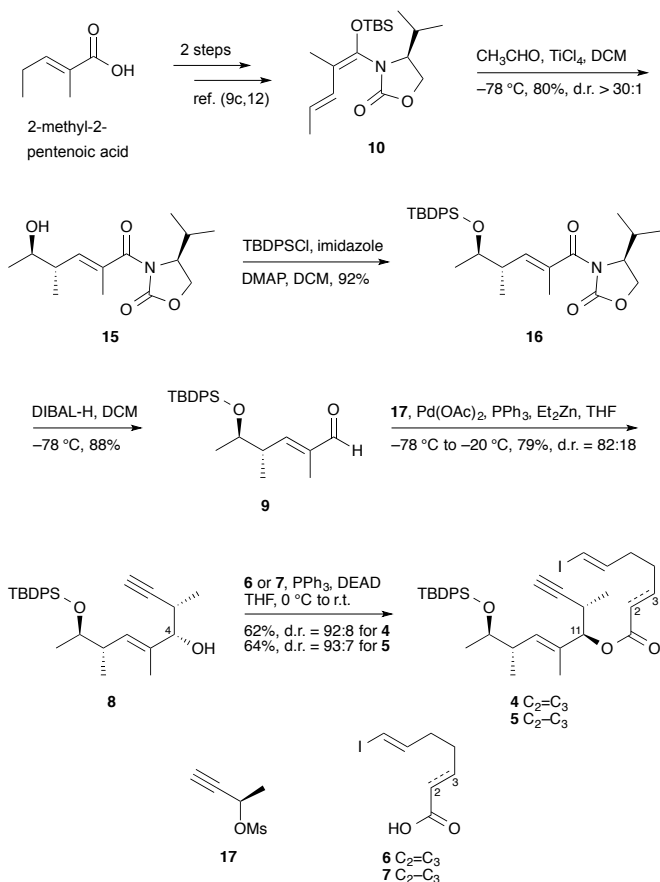
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could be obtained from iodides **4/5** via our coupling/reduction tandem reaction. Esters **4/5** could be readily prepared from propargyl alcohol **8** and the carboxylic acids **6/7** using Mitsunobu conditions. Enantioenriched alcohol **8** was envisioned to be obtained from aldehyde **9** using Marshall's asymmetric propargylation reaction.⁸ An *anti*-selective Kobayashi vinylogous aldol reaction of compound **10** could deliver aldehyde **9**.⁹

The preparation of acid **6** from alcohol **11**¹⁰ is shown in Scheme 2 and involved a 3-step sequence including Swern oxidation, Wittig olefination and tin-promoted ester cleavage. Common basic saponification conditions failed to convert ester **13** to acid **6**. Acid **7** was prepared from alcohol **14**¹¹ via Jones oxidation.

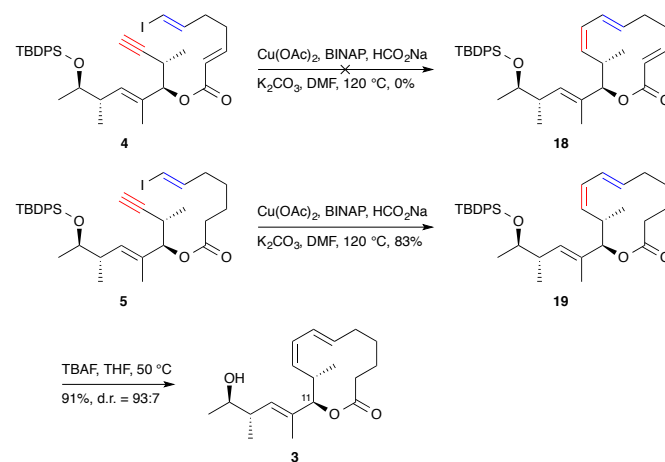


Scheme 2 Preparation of acids **6** and **7**.



Scheme 3 Synthesis of ester **4/5**.

The syntheses of esters **4/5** started from the commercially available (*E*)-2-methyl-2-pentenoic acid, which was converted to the known vinylketene silyl *N,O*-acetal **10** in two steps following literature reports (Scheme 3).^{9c, 12} Kobayashi's vinylogous aldol reaction of compound **10** with acetaldehyde afforded alcohol **15**. Maintaining the low reaction temperature was crucial for achieving high diastereoselectivity (d.r. > 30:1). Silyl protection of alcohol **15** followed by reductive removal of the chiral auxiliary provided aldehyde **9**. The aldehyde underwent Marshall's propargylation reaction with chiral allenylzinc reagent *in situ* formed from mesylate **17**¹³ to furnish alcohol **8** as an inseparable mixture of epimers. The diastereomeric ratio was 82:18 in favor of the (4*S*)-alcohol.¹⁵ Another version of the Marshall propargylation using a chiral allenylstannane¹⁴ gave the (4*R*)-alcohol as the major stereoisomer, however, with low selectivity (d.r. = 56:44). Finally, esters **4** and **5** were obtained by Mitsunobu esterification of alcohol **8** with acids **6/7**, along with 20% of the dehydration products. A diastereomeric ratio of 90:10, preferring the (1*R*)-ester (lactimidomycin numbering), was observed, which is presumably due to the higher propensity of the minor (4*R*)-epimer of alcohol **8** to generate the dehydration products. The esters **4/5**, we then subjected to the macrocyclization method based on the Castro-Stephens coupling and the *in situ* alkyne reduction (Scheme 4). The reaction of ester **4** resulted in complete decomposition to form a complex, unidentifiable mixture. We hypothesized that the *E*-double bond might reduce the flexibility of the molecular so that it could not adapt a conformation in which the reacting centers were close enough to facilitate the macrocyclization. In contrast, ester **5** without the conjugated double bond reacted smoothly to produce the expected lactone **19** in a yield of 83%. Completion of the formal total synthesis of lactimidomycin was achieved after desilylation of compound **19** to yield the known alcohol **3** and its C₁₁-epimer (d.r. = 93:7, separable). Spectra data and optical rotation ($[\alpha]_D^{22} = -85$ (c 0.10, CHCl₃)) of the major isomer were in excellent agreement with those of the same compound reported by Fürstner ($[\alpha]_D^{20} = -86$ (c 0.9, CHCl₃)).^{6a}



Scheme 4 Macrocyclization and deprotection.

In summary, the synthesis of macrolactone **3**, an advanced intermediate in the synthesis of lactimidomycin, has been realized from commercially available starting materials in a total of nine steps (longest linear sequence) using minimal functional group protection. This is to our knowledge the shortest route to lactimidomycin intermediates with similar structure. The key step was a copper-catalyzed ene-yne coupling/alkyne reduction tandem reaction that formed the 12-

membered *E,Z*-diene macrocycle. Other featured reactions included a Kobayashi vinylogous aldol reaction and a Marshall propargylation reaction to generate the four stereocenters.

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reported in Reference 8, and the fact that it led to the correct final product.

Notes and references

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† Electronic Supplementary Information (ESI) available: Experimental procedures, compound characterizations, and NMR spectra. See DOI: 10.1039/c000000x/

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