ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/chemcomm

Rational Design and Synthesis of Substrate-Product Analogue

Mohan Pal,^a Mandar Khanal,^a Ryan Marko,^a Srinath Thirumalairajan,^a

Inhibitors of α -Methylacyl-coenzyme A Racemase from

Mycobacterium tuberculosis†

and Stephen L. Bearne^{a,b,*}

ChemComm

COMMUNICATION

Received 00th January 20xx, Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/chemcomm

2,2-Bis(4-*i*-butylphenyl)propanoyl-CoA and 2,2-bis(4-*t*-butylphenyl)propanoyl-CoA are rationally designed, gem-disubstituted substrate-product analogues that competitively inhibit α -methylacylcoenzyme A racemase from *Mycobacterium tuberculosis* with K_i values of 16.9 ± 0.6 and 21 ± 4 μ M, respectively, exceeding the enzyme's affinity for the substrate by approximately 5-fold.

With the exception of utilizing compounds that mimic the electronic and/or geometric characteristics of the substrate(s) or product(s), or the intermediate(s) or transition state(s) formed during catalysis,¹⁻³ there are few general strategies for designing enzyme inhibitors. Our observation that mandelate racemase (MR, **Scheme 1**) is inhibited by benzilate⁴⁻⁶ suggested a new strategy for designing inhibitors of racemases and epimerases. This strategy utilizes substrate-product analogues that incorporate structural features of both the substrate and the product of an enzyme-catalyzed racemization or epimerization reaction. However, such analogues proved to be only modest inhibitors of glutamate, serine, and proline racemases, which have compact active sites.^{7,8} That the inhibition of these enzymes was not as effective as in the case of MR suggested that this inhibitor design strategy requires enzymes

Scheme 1. Similar reactions catalyzed by MR and MtMCR								
R ₂	R_1 R_3 R_3	R ₂	\bigcirc	R_1 I O^- R_3] 🛶	R ₂		
Е-В1:	HB ₂ –E	_ EB	1H	НВ ₂ –Е		+ E–B₁H	:B ₂ –E	
enzyme	substrate	•	R ₁	R ₂	R_3	B ₁	B ₂	
MR	mandelate	Э	OH	Н	0-	Lys 166	His 297	
<i>Mt</i> MCR	ibuprofenoyl-0	CoA	CH_3	<i>i</i> -Bu	SCoA	Asp 156	His 126	

^{a.} Department of Biochemistry & Molecular Biology, Dalhousie University, Halifax, NS, B3H 4R2, Canada. E-mail: sbearne@dal.ca

^{b.} Department of Chemistry, Dalhousie University, Halifax, NS,B3H 4R2, Canada.

+ Electronic Supplementary Information (ESI) available: Synthetic procedures, enzyme purification and assays, characterization of new compounds, and inhibition studies. See DOI: 10.1039/x0xx00000x with capacious and plastic active sites. To test this hypothesis, we rationally designed substrate-product analogue inhibitors of α -methylacyl-coenzyme A racemase from *Mycobacterium tuberculosis* (EC 5.1.99.4, *Mt*MCR), employing *gem*-disubstitution on the carbon-2 stereocenter of a substrate by the group undergoing motion (i.e., the acyl groups of *Mt*MCR substrates) during epimerization. Herein, we report that the novel, *gem*-disubstituted, substrate-product analogues **5** (Fig. 1 and Scheme 2) competitively inhibit *Mt*MCR with binding affinities commensurate with the affinities of inhibitors identified for the human homologue (*vide infra*), thereby demonstrating the utility of our inhibitor design strategy.

gem-Disubstituted inhibitors are reminiscent of bisubstrate analogue inhibitors. Such analogues combine structural features of the two substrates required by a target enzyme,^{1,9} and thereby often yield enhanced specificity and binding, the latter arising because of entropic effects.¹⁰ Similarly, substrate-product analogues for racemases and epimerases combine structural attributes of two ligands and enhanced binding affinity is anticipated due to the additional binding determinants, provided that the additional steric bulk is tolerated.

The X-ray crystal structure of MR complexed with the substrate analogue (S)-atrolactate revealed that the carboxylate, hydroxyl group, and ipso carbon of the phenyl ring of the substrate bind in the same plane (Fig. 1A),¹¹ so that the enantiomers of mandelate are likely bound in a mirror-image orientation (Fig. 1B).^{12,13} X-ray crystallographic studies⁴ and site-directed mutagenesis experiments⁵ suggest that the phenyl ring moves between two sub-sites (i.e., R- and Spockets, Fig. 1B) during catalysis. MR is competitively inhibited by benzilate ($K_i = 0.6 \text{ mM}$, cf. $K_s = 1.0 \text{ mM}$; Fig. 1C),¹⁴ indicating that the active site can accommodate the two phenyl rings of this substrate-product analogue. Thus gemdisubstituted inhibitors such as benzilate may be designed to present binding determinants simultaneously to the R- and Spockets. Indeed, we subsequently showed that 3,3,3-trifluoro-2-hydroxy-2-(trifluoromethyl)propanoate is a highly potent substrate-product analogue inhibitor of MR based on the substrate 3,3,3-trifluorolactate ($K_i = 27 \mu M$).^{15,16}

This journal is © The Royal Society of Chemistry 2015

COMMUNICATION



Fig. 1. Binding of ligands to MR and *Mt*MCR. (A) Superposition of the substrate analogue (*S*)-atrolactate (PDB 1MDR)¹¹ and the intermediate analogue BzH (PDB 3UXK)⁴ bound at the active site of MR. The active site Mg^{2+} is shown in space-filling representation. (B) Mirror-image binding of substrate enantiomers (superposed) results in three substituents on the α -carbon lying in the same plane (magenta dotted line) with the α -hydrogens being antipodal.^{12,13} Aryl rings attached to the α -carbon atom project linearly to define the *R*- and *S*-pockets of MR for binding the phenyl ring of *R*- and *S*-mandelate (where R₁ = OH, R₂ = H, and R₃ = O⁻), respectively. (C) Benzilate with its two phenyl groups binding at the active site of the C925/C264S/K166C triple mutant of MR.¹⁵ (D) *gem*-Diaryl substrate-product analogue inhibitors of *Mt*MCR. (E) Stereoview showing the superposition of the substrates (2*S*)- and (2*R*)-ibuprofenoyl-CoA bound at the active site of *Mt*MCR (PDB 2GCE).¹⁷ The CoA moiety and α -CH₃ remain fixed, while the 2-aryl moiety moves over a Met-rich surface (purple) during catalysis.

To test our hypothesis that enzymes with capacious active sites might be more amenable to inhibition by substrateproduct analogues, we applied our inhibitor design strategy to MtMCR. MtMCR is a cofactor-independent enzyme that catalyzes the 1,1-proton transfer reaction^{17,18} which effects the reversible epimerization of the CoA-thioesters derived from a variety of (2R)- and (2S)-methyl branched fatty acids,¹⁹ a number of C27 bile acid intermediates,²⁰ and 2-arylpropionic acids (e.g., ibuprofen, Scheme 1).^{21,22} The MtMCR-catalyzed epimerization of ibuprofenoyl-CoA is very similar to the racemization of mandelate catalyzed by MR (Scheme 1). Studies using 2-²H-labeled or 2-³H-labeled substrates^{17-19,22-24} and X-ray crystallographic studies on MtMCR indicate that the enzyme is a two-base epimerase and epimerization proceeds via an enol/enolate intermediate.^{17,23,25} Structural studies conducted by Wierenga and co-workers²³ on MtMCR complexed with (R,S)-ibuprofenoyl-CoA, (2S)-methylmyristoyl-CoA, and (2R)-methylmyristoyl-CoA revealed that the nonpolar acyl group moves through a Met-rich hydrophobic pocket during Page 2 of 5

catalysis (**Fig. 1E**), while the CoA and α -CH₃ groups remain bound at their respective binding sites.¹⁷ Substrate-product analogues could therefore be designed that are *gem*disubstituted with the hydrophobic group undergoing motion.

Our synthesis of such a substrate-product analogue inhibitor (5) is outlined in Scheme 2 (also see ESI). For the para-substituted species ($R_2 = i$ -butyl or t-butyl), gemdisubstitution was accomplished using the general synthesis of 2,2-diarylpropanoates described by Patai and Dayagi.²⁶ Ethyl pyruvate (1) was reacted with either *i*- or *t*-butylbenzene in concentrated sulfuric acid at -15 °C to afford 2c and 2d, respectively, after purification using silica gel chromatography. Vigorous stirring was required since the immiscible mixture of i- or t-butylbenzene and the ethyl pyruvate/sulfuric acid solution became quite viscous as the reaction proceeded. Conversion to 3c and 3d was achieved by hydrolysis using 5 M KOH in hexanes/MeOH. Finally, thioesters 5a – d were prepared by activating 3a - d with CDI to form 4a - d in dry DCM, followed by reaction with the trilithium salt of CoA in an aqueous bicarbonate solution.¹⁸ The resulting product was purified using semi-preparative solid-phase extraction to afford thioesters **5a** – **d** in yields of 26 – 48%.

Using a continuous circular dichroism (CD)-based assay and (2S)-ibuprofenoyl-CoA as the substrate,²⁷ we examined the inhibitory effect of **3a** – **d** and **5a** – **d** on *Mt*MCR activity. Recombinant MtMCR was overproduced as a fusion protein bearing an N-terminal hexahistidine tag and purified using nickel ion affinity chromatography (see ESI). Thioesters 5a - d are competitive inhibitors of MtMCR (Table 1), with 5c and 5d being the most effective inhibitors (Fig. 2) and binding with affinities ~5-fold greater than that exhibited for the substrate (2S)-ibuprofenoyl-CoA (assuming $K_m \approx K_s$). Clearly, MtMCR can simultaneously accommodate the two aryl groups of 5a - d. The geminal phenyl groups of 5a and 5b afforded a binding affinity that was weaker than that observed for the substrate by ~3-fold. The similar binding affinity for these two inhibitors suggests that the α -CH₃ group in **5b** does not contribute significantly to the binding affinity, perhaps because of the less efficient binding of the phenyl groups relative to larger hydrophobic acyl groups. However, Lloyd and co-workers found that (S)-2-methyldecanoyl-CoA and decanoyl-CoA exhibited K_m values of 614 and 225 μ M, respectively, also suggesting the methyl group does not contribute to ground



 $\mathbf{a},\,\mathsf{R}_1=\mathsf{R}_2=\mathsf{H};\,\mathbf{b},\,\mathsf{R}_1=\mathsf{CH}_3,\,\mathsf{R}_2=\mathsf{H};\,\mathbf{c},\,\mathsf{R}_1=\mathsf{CH}_3,\,\mathsf{R}_2=\mathit{i}\text{-butyl};\,\mathbf{d},\,\mathsf{R}_1=\mathsf{CH}_3,\,\mathsf{R}_2=\mathit{t}\text{-butyl}$

Journal Name

state binding by human α -methylacyl-CoA racemase (AMACR).²⁸ The addition of the *i*- or *t*-butyl groups in **5c** or **5d**, respectively, enhanced the inhibition ~20-fold, consistent with the presence of additional hydrophobic interactions within the large hydrophobic cavity that comprises the binding sites for the acyl moiety of substrate epimers.¹⁷ Comparison of the IC₅₀ values for the inhibition of *Mt*MCR by CoA and the free acids **3a** – **d** (**Table 1**) reveals that the presence of both the acyl and CoA moieties in the inhibitor are required for better binding. The surprising exception is **3c**, which was a reasonably effective inhibitor, binding with an affinity similar to that of the substrate. This observation suggests that inhibitors might be developed that either lack the CoA moiety or have only part of the CoA structure.

Since **5a** bears an α -proton similar to the substrate, *Mt*MCR could potentially catalyze exchange of this proton with deuterium from solvent D₂O. Using a concentration of *Mt*MCR exceeding that typically employed under assay conditions by at least 1000-fold, and monitoring the signals arising from the α proton and phenyl protons of **5a** by ¹H NMR spectroscopy, no *Mt*MCR-catalyzed exchange was observed over 24 h (see ESI). Thus, the bound orientation of **5a** places the α -proton in a location where it is not amenable to exchange. Such a binding orientation is consistent with the two phenyl rings occupying the binding sites for the 4-*i*-butylphenyl groups of the enantiomers of ibuprofenoyl-CoA (**Fig. 1E**) with the α -proton located in the binding pocket for the α -methyl group where it is not accessible to the Brønsted acid-base catalysts.

MtMCR shares a similar mechanism and substrate specificity with its homologue human AMACR (43% amino acid

TABLE 1. Inhibition	of MtMCR
---------------------	----------

compound	inhibition results			
-	IC ₅₀ , μM ^a	<i>Κ</i> i, μΜ ^b		
coenzyme A	$1.2 (\pm 0.2) \times 10^3$	nd ^c		
(S)-ibuprofen ^d	$39 (\pm 5) \times 10^3$	nd		
(2 <i>S</i>)-ibuprofenoyl-CoA	-	$106 \pm 15 (K_m)$		
diphenylacetic acid (3a) ^d	> 6.3 × 10 ^{3e}	nd		
diphenylacetyl-CoA (5a)	$1.1 (\pm 0.3) \times 10^3$	358 ± 39		
2,2-diphenylpropanoic acid (3b) ^d	42 (± 5) × 10^3	nd		
2,2-diphenylpropanoyl-CoA (5b)	610 ± 73	343 ± 72		
2,2-bis(4- <i>i</i> -butylphenyl)- propanoic acid (3c) ^d	316 ± 37	102 ± 9		
2,2-bis(4- <i>i</i> -butylphenyl)- propanoyl-CoA (5c)	39.5 ± 0.9	16.9 ± 0.6		
2,2-bis(4- <i>t</i> -butylphenyl)- propanoic acid (3d) ^d	_f	nd		
2,2-bis(4- <i>t</i> -butylphenyl)- propanoyl-CoA (5d)	44 ± 8	21 ± 4		

^aIC₅₀ values determined with [(25)-ibuprofenoyl-CoA] = 100 μM and [*Mt*MCR] = 0.18 nM. ^bValues reported are the averages of three independent trials and the errors are the standard deviations. ^cNot determined. ^dDetermined in the presence of 20% DMSO. ^eValue estimated assuming competitive inhibition with only 24 (± 3)% inhibition observed and [**3a**] = 20 mM. ^fOnly 15 (± 0.5)% inhibition observed at [**3d**] = 480 μM; however, it is likely that weak inhibition arises from solubility problems (see **Fig. 50S** in the ESI).





Fig 2. Inhibition of *Mt*MCR by **5c** and **5d**. (A) and (C) Representative Lineweaver-Burk plots showing the competitive inhibition of *Mt*MCR by **5c** and **5d**, respectively. Concentrations (μ M): **5c**: 0 (O), 12 (\triangle), 24 (\bigtriangledown), 36 (\diamond); **5d**: 0 (O), 16 (\triangle), 32 (\bigtriangledown), 48 (\diamond); (25)-ibuprofenoyl-CoA: 40, 80, 160, 360, 600, and 800; and *Mt*MCR: 0.18 nM (7.5 ng/mL). (B) and (D) Representative plots of the apparent K_m/V_{max} values (determined from direct fits of Eqn. 15 to initial velocity data (i.e., Michaelis-Menten plots; see **Figs. 56S** and **57S** in the ESI)) vs. concentrations of **5c** and **5d**, respectively.³⁰ K_i values (determined in triplicate) are given in **Table 1**.

sequence identity with MtMCR).²⁹ Elevated levels of AMACR have been associated with a variety of cancers,³¹ including prostate cancer (PCa)^{32,33} for which the enzyme serves as a biomarker.³³⁻³⁵ Knockdown of AMACR gene expression suppresses the growth of several PCa cell lines suggesting that AMACR may be a therapeutic target for PCa.^{29,36-39} Several efforts to develop inhibitors of AMACR have been reported. Analogues of the natural substrate branched chain α methylacyl-CoA thioesters possessing one or more β -fluorine atoms were synthesized as potential suicide substrates.40 These compounds, however, did not covalently modify the enzyme as anticipated but acted as competitive inhibitors with $K_{\rm i}$ values of 0.9 – 20 μ M ($K_{\rm m}/K_{\rm i}$ = 27–1.2). Morgenroth *et al.*⁴¹ synthesized 2-methylenacyl-CoA thioesters as AMACR binding ligands; e.g., (13E/12)-13-iodo-2-methylentridec-12-enoyl-CoA acted as a competitive inhibitor with a K_i value of 19 μ M (K_m/K_i = 1.5). One of the most potent inhibitors is N-dodecyl-Nmethylcarbamoyl-CoA with a K_i value of 98 nM ($K_m/K_i \approx 969$).⁴² Because these inhibitors closely resemble the structure of the substrates, one might anticipate that they would lack specificity. A high-throughput screen of a 5,000-compound library for inhibitors of AMACR activity led to the identification of several compounds that were not structural mimics of the substrates but inhibited AMACR with IC₅₀ values between 0.8 and 85 μ M ($K_m/K_i \approx 107 \& 1.0$, respectively).²⁹ Some of these compounds were irreversible inactivators of AMACR.

In summary, we have demonstrated that utilizing gemdisubstituted substrate-product analogues furnishes an

COMMUNICATION

effective strategy for designing inhibitors of racemases and epimerases with capacious active sites. This rational approach offers an alternative to the well-established strategy of using β -halo compounds^{43,44} to generate suicide substrates of racemases and epimerases catalyzing formation of enol/enolate intermediates, or to the strategies of designing an inhibitor based on the structure of the substrate(s), product(s), intermediate(s), or transition state(s) of enzyme-catalyzed reactions. The inhibitors described in the present work may furnish tools for studying cholesterol ester metabolism in M. tuberculosis,⁴⁵ or lead compounds for the development of AMACR inhibitors directed against PCa. The additional acyllike chain on such gem-disubstituted inhibitors may endow the inhibitor with specificity, obviating its recognition by other fatty acyl-CoA-metabolizing enzymes. Of course, in vivo studies will require structural alterations and prodrug strategies to circumvent hydrolysis of the thioester and the lack of cell permeability arising from the CoA moiety, respectively.^{46,47} Alternatively, the reasonable binding affinity exhibited by gemdisubstituted diaryl acids such as 3c may offer a means of overcoming the need for incorporation of the CoA moiety.

This work was supported by a Discovery Grant from the Natural Sciences and Engineering Research Council (NSERC) of Canada (S.L.B.), a grant from the Prostate Cancer Research Foundation (S.L.B.), and a Cancer Research Training Program studentship award (M.K.) from the Beatrice Hunter Cancer Research Institute. We thank Dr. Mike Lumsden (NMR-3) for his assistance in conducting NMR experiments.

Notes and references

- 1 A. Radzicka and R. Wolfenden, *Methods Enzymol.*, 1995, **249**, 284-312.
- 2 R. A. Copeland, *Evaluation of enzyme inhibitors in drug discovery: A guide for medicinal chemists and pharmacologists*, John Wiley and Sons, Inc., New York, 2013.
- 3 V. L. Schramm, ACS Chem. Biol., 2013, 8, 71-81.
- 4 A. D. Lietzan, M. Nagar, E. A. Pellmann, J. R. Bourque, S. L. Bearne and M. St. Maurice, *Biochemistry*, 2012, **51**, 1160-1170.
- 5 F. Siddiqi, J. R. Bourque, H. Jiang, M. Gardner, M. St. Maurice, C. Blouin and S. L. Bearne, *Biochemistry*, 2005, 44, 9013-9021.
- 6 M. St. Maurice and S. L. Bearne, *Biochemistry*, 2000, **39**, 13324-13335.
- 7 M. Harty, M. Nagar, L. Atkinson, C. M. LeGay, D. J. Derksen and S. L. Bearne, *Bioorg. Med. Chem. Lett.*, 2013, 23, 390-393.
- 8 M. Pal and S. L. Bearne, *Bioorg. Med. Chem. Lett.*, 2014, **24**, 1432-1436.
- 9 A. D. Broom, J. Med. Chem., 1989, 32, 2-7.
- 10 M. I. Page and W. P. Jencks, Proc. Natl. Acad. Sci. USA, 1971, 68, 1678-1683.
- 11 J. A. Landro, J. A. Gerlt, J. W. Kozarich, C. W. Koo, V. J. Shah, G. L. Kenyon, D. J. Neidhart, S. Fujita and G. A. Petsko, *Biochemistry*, 1994, 33, 635-643.
- 12 K. R. Hanson, Arch. Biochem. Biophys., 1981, 211, 575-588.
- 13 A. D. Mesecar and D. E. Koshland, Jr., *Nature*, 2000, **403**, 614-615.
- 14 M. St. Maurice and S. L. Bearne, *Biochemistry*, 2004, **43**, 2524-2532.
- 15 M. Nagar, A. D. Lietzan, M. St. Maurice and S. L. Bearne, *Biochemistry*, 2014, **53**, 1169-1178.
- 16 M. Nagar, A. Narmandakh, Y. Khalak and S. L. Bearne, *Biochemistry*, 2011, **50**, 8846-8852.
- 17 P. Bhaumik, W. Schmitz, A. Hassinen, J. K. Hiltunen, E. Conzelmann and R. K. Wierenga, *J. Mol. Biol.*, 2007, **367**, 1145-1161.

- 18 D. J. Darley, D. S. Butler, S. J. Prideaux, T. W. Thornton, A. D. Wilson, T. J. Woodman, M. D. Threadgill and M. D. Lloyd, *Org. Biomol. Chem.*, 2009, 7, 543-552.
- W. Schmitz, R. Fingerhut and E. Conzelmann, *Eur. J. Biochem.*, 1994, 222, 313-323.
- 20 D. W. Russell, Annu. Rev. Biochem., 2003, 72, 137-174.
- 21 C. S. Chen, W. R. Shieh, P. H. Lu, S. Harriman and C. Y. Chen, *Biochim. Biophys. Acta.*, 1991, **1078**, 411-417.
- 22 W.-R. Shieh and C.-S. Chen, J. Biol. Chem., 1993, 268, 3487-3493.
- K. Savolainen, P. Bhaumik, W. Schmitz, T. J. Kotti, E. Conzelmann, R. K. Wierenga and J. K. Hiltunen, *J. Biol. Chem.*, 2005, **280**, 12611-12620.
- 24 W. Schmitz, C. Albers, R. Fingerhut and E. Conzelmann, *Eur. J. Biochem.*, 1995, **231**, 815-822.
- 25 S. Sharma, P. Bhaumik, W. Schmitz, R. Venkatesan, J. K. Hiltunen, E. Conzelmann, A. H. Juffer and R. K. Wierenga, J. Phys. Chem. B, 2012, 116, 3619-3629.
- 26 S. Patai and S. Dayagi, J. Chem. Soc., 1958, 3058-3061.
- 27 D. Ouazia and S. L. Bearne, Anal. Biochem., 2010, 398, 45-51.
- 28 F. A. Sattar, D. J. Darley, F. Politano, T. J. Woodman, M. D. Threadgill and M. D. Lloyd, *Chem. Commun.*, 2010, 46, 3348-3350.
- 29 B. A. Wilson, H. Wang, B. A. Nacev, R. C. Mease, J. O. Liu, M. G. Pomper and W. B. Isaacs, *Mol. Cancer Ther.*, 2011, **10**, 825-838.
- 30 I. H. Segel, *Enzyme Kinetics*, John Wiley and Sons, Inc., New York, 1975.
- 31 A. Nassar, M. B. Amin, D. G. Sexton and C. Cohen, Appl. Immunohistochem. Mol. Morphol., 2005, 13, 252-255.
- 32 M. A. Rubin, M. Zhou, S. M. Dhanasekaran, S. Varambally, T. R. Barrette, M. G. Sanda, K. J. Pienta, D. Ghosh and A. M. Chinnaiyan, *JAMA*, 2002, **287**, 1662-1670.
- 33 Kumar-Sinha, R. B. Shah, B. Laxman, S. A. Tomlins, J. Harwood, W. Schmitz, E. Conzelmann, M. G. Sanda, J. T. Wei, M. A. Rubin and A. M. Chinnaiyan, Am. J. Pathol., 2004, 164, 787-793.
- 34 J. Luo, S. Zha, G. W.R., T. A. Dunn, J. L. Hicks, C. J. Bennett, C. M. Ewing, E. A. Platz, S. Ferdinandusse, R. J. Wanders, J. M. Trent, W. B. Isaacs and A. M. De Marzo, *Cancer Res.*, 2002, **62**, 2220-2226.
- 35 Z. Jiang, B. A. Woda, C. L. Wu and X. J. Yang, Am. J. Clin. Pathol., 2004, 122, 275-289.
- 36 S. Zha, S. Ferdinandusse, S. Denis, R. J. Wanders, C. M. Ewing, J. Luo, A. M. De Marzo and W. B. Isaacs, *Cancer Res.*, 2003, **63**, 7365-7376.
- 37 M. D. Lloyd, D. J. Darley, A. S. Wierzbicki and M. D. Threadgill, *FEBS J.*, 2008, **275**, 1089-1102.
- 38 K. Takahara, H. Azuma, T. Sakamoto, S. Kiyama, T. Inamoto, N. Ibuki, T. Nishida, H. Nomi, T. Ubai, N. Segawa and Y. Katsuoka, *Anticancer Res.*, 2009, **29**, 2497-2505.
- 39 M. D. Lloyd, M. Yevglevskis, G. L. Lee, P. J. Wood, M. D. Threadgill and T. J. Woodman, *Prog. Lipid Res.*, 2013, **52**, 220-230.
- 40 M. Yevglevskis, G. L. Lee, M. D. Threadgill, T. J. Woodman and M. D. Lloyd, Chem. Commun., 2014, 50, 14164-14166.
- 41 A. Morgenroth, E. A. Urusova, C. Dinger, E. Al-Momani, T. Kull, G. Glatting, H. Frauendorf, O. Jahn, F. M. Mottaghy, S. N. Reske and B. D. Zlatopolskiy, *Chem. Eur. J.*, 2011, **17**, 10144-10150.
- 42 A. J. Carnell, R. Kirk, M. Smith, S. McKenna, L. Y. Lian and R. Gibson, *ChemMedChem*, 2013, **8**, 1643-1647.
- 43 A. J. Carnell, I. Hale, S. Denis, R. J. Wanders, W. B. Isaacs, B. A. Wilson and S. Ferdinandusse, *J. Med. Chem.*, 2007, **50**, 2700-2707.
- 44 A. Fersht, *Structure and Mechanism in Protein Science*, W.H. Freeman and Co., New York, 1999.
- 45 R. Lu, W. Schmitz and N. S. Sampson, Biochemistry, 2015, 54, 5669.
- 46 Y. Hwang, S. Ganguly, A. K. Ho, D. C. Klein and P. A. Cole, *Bioorg. Med. Chem.*, 2007, **15**, 2147-2155.
- 47 K. Vong, I. S. Tam, X. Yan and K. Auclair, ACS Chem. Biol., 2012, 7, 470-475.

This journal is © The Royal Society of Chemistry 2015

Journal Name

Journal Name

Table of Contents Graphic



gem-Disubstituted substrate-product analogues competitively inhibit α -methylacyl-coenzyme A racemase from *Mycobacterium tuberculosis*, binding with affinities exceeding that of the substrate by ~5-fold.