

ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

Cite this: DOI: 10.1039/coxx00000x

www.rsc.org/xxxxxx

ARTICLE TYPE

Hierarchically porous anatase TiO₂ coated-WO₃ 2D IO bilayer film and photochromic properties

Hua Li,^{ab} Huazhong Wu,^c JiajiaXiao,^b YanliSu,^b Jacques Robichaud,^a Ralf Brüning,^d and Yahia Djaoued^{*a}

Received (in XXX, XXX) Xth XXXXXXXXX 20XX, Accepted Xth XXXXXXXXX 20XX

DOI: 10.1039/b000000x

A hierarchically porous anatase TiO₂ coated-WO₃ 2D inverse opal (IO) bilayer film was fabricated on ITO glass using a layer by layer route with a hierarchically porous TiO₂ top layer and an ordered super-macroporous WO₃ 2D IO bottom layer. This novel TiO₂ coated-WO₃ 2D IO bilayer film was evaluated for photochromic applications.

In recent years, there has been much interest in inorganic photochromic materials due to their promising applications in display, imaging, and solar energy conversion^{1,2}. Transition-metal oxides, such as WO₃, TiO₂, MoO₃, show color change on exposure to either sunlight or UV irradiation. Photochromism of WO₃ has attracted widespread interest. However, as a wide bandgap semiconductor, WO₃ has a short life expectancy and shows poor photochromic properties, reacting to UV light only ($\lambda < 380$ nm). In order to increase its photochromic response, researchers have developed various routes, such as cathode activation³, coupling with a narrow bandgap semiconductor⁴ or with a noble metal⁵. Among these routes, coupling with TiO₂ nanoparticles attracted much attention since nano-TiO₂ greatly increases the photochromic properties of WO₃^{6,7}.

In order to extend the absorption of TiO₂ from UV to the visible light region, many methods⁸⁻¹¹ were developed, such as impurity doping^{12,13} or coupling with semiconductors¹⁴⁻¹⁶. Unfortunately, whatever the kind of doping, it was found that there is a critical doping concentration which limits the efficiency of such photochromic devices¹³. Coupling TiO₂ with semiconductors of different energy level requires good matching of the conduction band and valence band of the two semiconductors^{14,17}. If TiO₂ is coupled with a dye sensitizer, then high porosity and surface area are important for the sensitizer to be sufficiently adsorbed and for the paths of electron transport to be electronically interconnected¹⁸⁻²⁰.

In this communication, we present a novel designed bilayer film coupling WO₃ and TiO₂ with a highly developed porous structure. In such materials, the ordered 2D WO₃ IO functions as both coupling semiconductor and modifier of the absorption bands of a dye sensitizer, while the hierarchically porous TiO₂ coating provides an efficient area for the adsorption of the sensitizer and paths for electron transport. Such a bilayer film, for which the main steps of the fabrication procedure are illustrated schematically in Figure 1, exhibits good photochromic properties. In previous works, following the air-water interface method, sodium dodecylsulfate (SDS) solution at 2 % was used to

facilitate the PS spheres self-assembly on the surface of water^{21,22}. On the other hand, it is known that the efficiency of a photochromic device is dependent on a highly crystallized dielectric material¹⁵. Unfortunately, the presence of cation impurities such as Na⁺ from SDS is known to hinder the crystallization of the dielectric material. For example, tungsten could crystallize to Na₂WO₄ instead of WO₃, deteriorating the intercalation/deintercalation properties of a device (see ESI, Figure S1).

In order to avoid the undesired effects of the SDS surfactant on both the synthesis and the photochromic properties of the devices, a novel method was brought forward by combining aspects of the vortical surface method²³ with the air-water interface method²⁴. Consequently, in this work, the 2D PS opal film was self-assembled without the use of a surfactant within a Teflon ring²³ (6 cm in diameter) floating at the surface of water. In such a process, once the 2D PS opal film is self-assembled, the water is exchanged until the solution under the PS opal becomes clear. Subsequently, a WO₃ ethanol/aqueous solution is injected under the PS opal. WO₃ penetrates into the interstitial spaces between the PS spheres from below. Then an ITO glass substrate is inserted under the floating film and the solution is slowly sucked out to deposit the PS/WO₃ opal composite monolayer onto the ITO substrate.

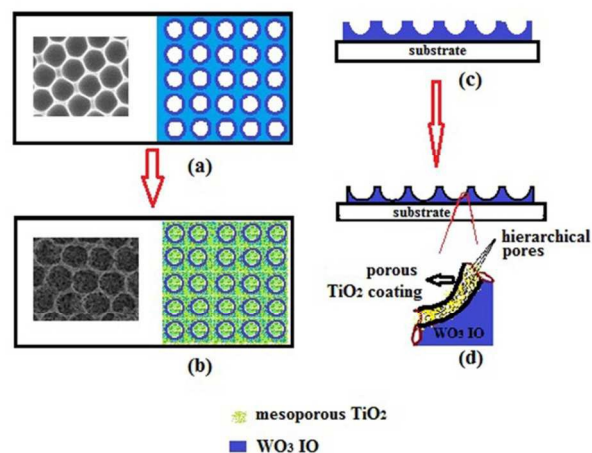


Figure 1 Schematic representation of the main steps in the fabrication of hierarchically porous anatase TiO₂ coated-WO₃ 2D IO composite bilayer film. (a) & (c) WO₃ 2D IO film was prepared by wet-injection strategy. (b) & (d) hierarchically porous TiO₂ was coated onto the WO₃ 2D IO.

A WO_3 IO film is obtained (Figure 1a and 1c) by the removal of the PS spheres. After heating to assure good adhesion between the WO_3 IO and the ITO substrate, a TiO_2 precursor colloidal solution is used to dip-coat onto the WO_3 2D IO. A final hierarchically porous anatase TiO_2 coated on the WO_3 2D IO is obtained after calcination and burning of the P123 template (see experimental details in ESI). Figures 2 and 3 show hierarchically porous anatase TiO_2 coated- WO_3 2D IO bilayer films templated from 500 and 750 nm PS spheres, respectively, as well as the results after the intermediate fabrication steps. A clean honeycomb structure with smooth bottom is evident in the WO_3 2D IO (Figures 2(d) and 3(d)). The honeycomb structure has circular upper-end openings which give us the chance to coat another layer onto it without destroying the array structure. After coating with TiO_2 , the bilayer films have a porous internal surface within the honeycomb macropores (Figures 2 (a) and 3(a)). The pores of the TiO_2 layer are around 10-100 nm in diameter (Figures 2(a), 3(a), and S2). Their hierarchical structure has a BET surface area of $125 \text{ m}^2/\text{g}$ and a BJH pore volume of $0.34 \text{ cm}^3/\text{g}$ (Figure S3).

The dip-coating parameters are crucial in the present experiment. When the withdrawing speed was as slow as 60 mm/min, the TiO_2 coating became too thick, hiding the 2D macropore array structure of the WO_3 2D IO monolayer (Figure S4). A 75 mm/min withdrawal speed, was the appropriate speed to obtain the TiO_2 hierarchically porous coating and keep the WO_3 2D IO structure. From the cross sectional view (Figure S5 left), one can see that the thickness of the bilayer film templated from 500 nm PS spheres is around 380 nm, with 150 nm at the bottom of the cavity left by the removal of the PS spheres. This cavity is approximately 230 nm deep. The width of the macropore is around 450 nm in diameter and its wall is around 50 nm thick. Corresponding EDS analysis (Figure S5 right) shows that the molar ratio of Ti to W is around 0.46.

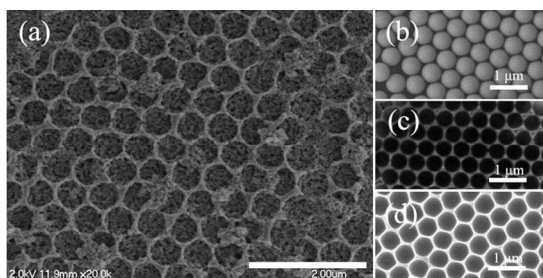


Figure 2 SEM images of hierarchically porous anatase TiO_2 coated- WO_3 2D IO composite bilayer (WT500) (a); PS opal (b); WO_3/PS composite film (c); and WO_3 2D IO (d) templated by 500 nm PS spheres.

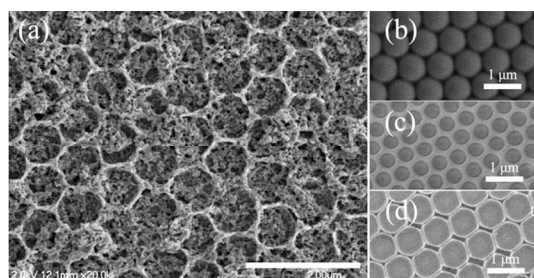


Figure 3 SEM images of hierarchically porous anatase TiO_2 coated- WO_3 2D IO composite bilayer (WT750) (a); PS opal (b); WO_3/PS composite film (c); and WO_3 2D IO (d) templated by 750 nm PS spheres.

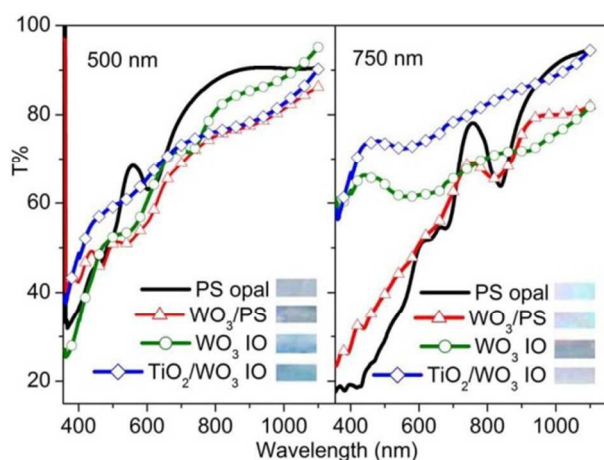


Figure 4 UV-vis-NIR transmittance spectra of PS opal films (solid line), WO_3/PS opal composite films (triangles), WO_3 IO films (circles) and TiO_2 coated- WO_3 2D IO composite bilayer (diamond) prepared from the PS spheres of 500 nm (left) and 750 nm (right). Insets show the optical reflection images from PS opal, PS/ WO_3 opal composite, WO_3 IO films, and TiO_2 coated- WO_3 2D IO composite bilayer on ITO substrates.

Figure 4 shows the UV-vis-NIR transmittance spectra of PS opal, WO_3/PS opal composite, annealed WO_3 IO monolayers and TiO_2 coated- WO_3 2D IO composite films prepared, along with their corresponding optical reflection images (see Figure 4 insets). The different colors depend on the wavelength absorbed by the photonic bandgap. All the films templated from 500 nm PS spheres, show a blue color. In contrast, colors of the films templated from 750 nm PS spheres changed from pale blue to violet.

Figure S6 shows the XRD patterns of the TiO_2 coated- WO_3 IO composite films templated from 500 and 750 nm PS spheres, along with the TiO_2 - WO_3 composite powder obtained by scraping the film templated from 500 nm PS spheres from the substrate. The tungsten oxide peaks in the WO_3 - TiO_2 composite samples are identified as monoclinic- WO_3 (pdf#43-1035, vertical solid bars). TiO_2 in the anatase form is also present (dotted vertical bars, pdf#21-1272). In order to determine the lattice parameters a , b , and c , a series of Gaussian peaks, superimposed on an overall quadratic background, were fitted to the experimental diffraction pattern. The peak width Δq , assumed to be the same for all reflections, is treated as a fitting parameter. After correction for the instrumental resolution, this width is used to estimate the crystallite size using the Scherrer method, with the assumption that crystallite to crystallite variations of the lattice parameter do not contribute to the peak width. The summary of XRD measurements on our samples is reported in Table S1. The average crystallite sizes are $22.4 \pm 1.7 \text{ nm}$ for m- WO_3 and $9.2 \pm 0.7 \text{ nm}$ for anatase TiO_2 .

The Raman spectra (Figure S7) of the TiO_2 coated- WO_3 2D IO composite films show peaks at 806 , 713 cm^{-1} ($\nu(\text{O}-\text{W}-\text{O})$), 326 , 273 cm^{-1} ($\delta(\text{O}-\text{W}-\text{O})$), 136 , 82.4 , 73 cm^{-1} (lattice modes), which are characteristic features of the monoclinic tungsten oxide (m- WO_3).

The mode emerging at 148.6 cm^{-1} is due to the crystallization of the TiO_2 in the anatase form. This anatase main band is high-frequency shifted and broadened with respect to the single-crystal

route. Ideas were borrowed from the ‘vortical surface method’ and ‘air-water interface method’ to develop a new procedure to prepare the WO₃ 2D IO bottom layer. Dip-coating was used to produce the TiO₂ top layer.

The films featured a hierarchically porous TiO₂ structure with pores from 10 to 100 nm and homogeneous macropores inherited from the WO₃ 2D IO. Such bilayer films show a well crystallized anatase TiO₂ top layer with an m-WO₃ under layer, which is favourable for fabrication of a photochromic device. The novel device was evaluated for prospective use in smart window applications with a high absorption in NIR range and mild absorption in the visible region.

ACKNOWLEDGEMENTS. The financial support of National Natural Science Foundation of China (grant No.21301123), the National Science and Engineering Research Council (NSERC) of Canada, the John R. Evans Leaders Fund of Canada Foundation for Innovation (CFI), and of the Research Assistantships Initiative of New Brunswick Innovation Fund (NBIF) is gratefully acknowledged. We also greatly thank the help in micro-Raman characteristic from Tianjin Gangdong SCI&Tech. Development Co. LTD.

Notes and references

^a Corresponding Author, Laboratoire de Recherche en Matériaux et Micro-spectroscopies Raman et FTIR, Université de Moncton-Campus de Shippagan, Shippagan, NB, E8S1P6, Canada. Fax: +1506 336 3434; Tel: +1506 336 3412; E-mail: *Corresponding Author, Yahia Djaoued, Yahia.djaoued@umoncton.ca

^b Department of Inorganic Materials, College of Chemistry Chemical Engineering and Materials Science, Soochow University, 199 Renai Road, Suzhou 215123, Jiangsu Province, PR China,

^c Department of Chemistry and Chemical Engineering, MinJiang University, Fuzhou, Fujian Province 350108, PR China

^d Department of Chemistry and Chemical Engineering, MinJiang University, Fuzhou, Fujian Province, 350108, PR China

^e Physics Department, Mount Allison University, Sackville, NB, E4L1E4, Canada

† Electronic Supplementary Information (ESI) available: [Experimental details; figures including SEM, Raman, XRD, HRTEM]. See DOI: 10.1039/b000000x/

- S. Songara, M. K. Patra, M. Manoth, L. Saini, V. Gupta, G. S. Gowd, S. R. Vadera and N. Kumar, *Journal of Photochemistry and photobiology A: Chemistry*, 2010, **209**, 68.
- B. Chi, L. Zhao, T. Jin, *J. Phys. Chem. C*, 2007, **111**, 6189.
- J. N. Yao, K. Hashimoto, K. Hashimoto, A. Fujishima, *Nature*, 1992, **355**, 624.
- (a) J. C. Santato, M. Ulmann, J. Augustynski, *Adv. Mater.*, 2001, **13**, 511; (b) M. A. Quevedo-Lopez, R. Ramirez-Bon, R. A. Orozco-Teran, O. Mendoza-Gonzalez, *Thin Solid Film*, 1999, **343**, 202.
- T. He, Y. Ma, Y. A. Cao, P. Jiang, W. S. Yang, J. N. Yao, *Langmuir*, 2001, **17**, 8024.
- (a) T. He, Y. Ma, Y. A. Cao, H. M. Liu, G. J. Zhang, W. S. Yang, J. N. Yao, *J. Phys. Chem. B*, 2002, **106**, 12670; (b) M. J. Taylor, *Phys. Lett. A*, 1968, **27**, 32.
- H. Huang, S. X. Lu, W. K. Zhang, Y. P. Gan, C. T. Wang, L. Yu, X. Y. Tao, *Journal of Physics and Chemistry of Solids*, 2009, **70**(3-4), 745.
- Y. Pang, W. Feng, J. Chen, Y. Liu, W. Cai, *J. Mater. Sci. Technol.*, 2007, **23**, 477.
- D. Kaczmarek, J. Domaradzki, A. Borkowska, *Thin Solid Films*, 2007, **515**, 6347.
- Y. Zhang, X. Xiong, Y. Han, X. Zhang, F. Shen, S. Deng, H. Xiao, X. Yang, G. Yang, H. Peng, *Chemosphere*, 2012, **88**, 145.
- Q. Li, J. K. Shang, *J. Am. Ceram. Soc.*, 2008, **91**(2), 660.

- S. U. M. Khan, M. Al-Shahry, W. B. Jr. Ingler, *Science*, 2002, **297**, 2243.
- (a) C. He, Y. Yu, X. Hu, A. Larbot, *Appl. Surf. Sci.*, 2002, **200**, 239. (b) C. M. Teh, A. R. Mohamed, *J. Alloys Compd.*, 2011, **509**, 1648.
- H. Zhang, G. Chen, D. W. Behnemann, *J. Mater. Chem.*, 2009, **19**, 5089.
- Y. Djaoued, S. Balaji, N. Beaudoin, *J Sol-Gel Sci. Technol.*, 2013, **65**(3), 374.
- S. Kuang, L. Yang, S. Luo, Q. Cai, *Appl. Surf. Sci.*, 2009, **255**, 7385.
- R. Daghri, P. Drogui, D. Robert, *Ind. Eng. Chem. Res.*, 2013, **52**(10), 3581.
- E. S. Kwak, W. Lee, N.-G. Park, J. Kim, H. Lee, *Adv. Funct. Mater.*, 2009, **19**, 1093.
- M. J. Cass, A. B. Walker, D. Martinez, L. M. Peter, *J. Phys. Chem. B*, 2005, **109**, 5100.
- P. Wang, S. M. Zakeerudin, J. d E. Moser, M. K. Nazeeruddin, T. Sekiguchi, M. Gratzel, *Nat. Mater.*, 2003, **2**, 498.
- J. Yu, Q. Yan, D. Shen, *ACS Appl. Mat. Inter.*, 2010, **2**, 1922.
- H. Li, G. Vienneau, M. Jones, B. Subramanian, J. Robichaud, Y. Djaoued, *J. Mater. Chem. C*, 2014, **2**, 7804
- F. Pan, J. Zhang, C. Cai, T. Wang, *Langmuir*, 2006, **22**(17), 7101.
- J. Yu, Q. Yan, D. Shen, *Applied Materials & Interfaces*, 2010, **2**(7), 1922.
- S. Balaji, Y. Djaoued, J. Robichaud, *J. Raman Spectrosc.*, 2006, **37**, 1416.
- L. E. Depero, IN.Sora, C. Perego, L. Sangaletti, G. Sberveglieri, *Sens Actuators B*, 1996, **31**, 19.
- UO. Krasovec, A. Georg, A. Georg, V. Wittwer, J. Luther, *Sol Energy Mater Sol Cells*, 2004, **84**, 369.
- S. Balaji, Y. Djaoued, A.-S. Albert, R. Bruning, N. Beaudoin, J. Robichaud, *J. Mater. Chem.*, 2011, **21**, 3940.