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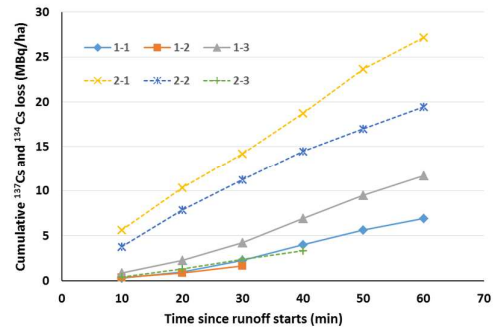
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First detailed investigation of radioactive cesium export process from upland fields under simulated rainfall in Fukushima.

Environmental impact statement

Heavy rainfalls can cause severe runoff and soil erosion from agricultural fields in Fukushima. The eroded sediments carried high concentration of radioactive caesium that could pose a radioactivity risk to the organisms in the river system.

1 Export of radioactive cesium from agricultural fields under
2 simulated rainfall in Fukushima.

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¹ The study was conducted during PT's visit to Tokyo University of Agriculture and Technology

17 **ABSTRACT**

18 In this study, we investigated the impact of rainfall on runoff, soil erosion and consequently
19 on the discharge of radioactive cesium in agricultural fields in Fukushima prefecture using
20 rainfall simulator. Simulated heavy rainfalls (50 mm/h) generated significant runoff and soil
21 erosion. The average concentration of radioactive cesium (the sum of ^{134}Cs and ^{137}Cs) in the
22 runoff sediments was ~3500 Bq/kg dry soil, more than double the concentrations measured in
23 the field soils which should be considered in studies using ^{137}Cs loss to estimate long-term
24 soil erosion. However, the estimated mass of cesium discharged through one runoff event
25 was less than 2% of the cesium inventory in the field. This suggested that cesium discharge
26 via soil erosion is not a significant factor in reducing the radioactivity of contaminated soils
27 in Fukushima prefecture. However, the eroded sediment carrying radioactive cesium will
28 deposit into the river systems and potentially pose a radioactivity risk for aquatic living
29 organisms.

30 **Keywords:** radioactive cesium, rainfall simulation, suspended solid, erosion, sediment

31

32 1. INTRODUCTION

33 The nuclear accident at the Fukushima Daiichi nuclear power plant (FDNPP) following the
34 massive earthquake and tsunami in March 2011 has released an enormous amount of
35 radionuclides into the atmosphere corresponding to the highest level (Level 7) on the
36 International Nuclear and Radiological Event Scale of the International Atomic Energy
37 Agency (IAEA). Decay of some short-lived radionuclides such as ^{131}I (half-life = 8 days (d)),
38 $^{129\text{m}}\text{Te}$ (34 d), and ^{136}Cs (13 d) and the remediation of contaminated soil has brought down
39 the level of radiation in Fukushima prefecture. However, concern remains over the
40 environmental radioactivity from relatively long-lived ^{134}Cs (2.07 years) and longer-lived
41 ^{137}Cs (30 years), especially for agricultural products grown in the region. It is widely believed
42 that people could be exposed to radioactivity through the consumption of ^{134}Cs and ^{137}Cs
43 contaminated food¹ due to the transfer of radioactive cesium into the biomass of agricultural
44 products.²

45
46 Many decontamination techniques have been employed in the contaminated zones in
47 Fukushima to reduce the external exposure in different surface areas. In agricultural lands,
48 current techniques include deep ploughing, topsoil removal and application of potassium
49 fertilizer with the average cost of about 1,500 yen/m² in the Special Decontamination Area
50 (area of high radioactive contamination where decontamination is implemented by the
51 national government).³ Most of the remediation efforts are focused on the Special
52 Decontamination Area where the recommended threshold for topsoil removal is equal or
53 higher than 5,000 Bq/kg (based on ^{137}Cs).³ In other areas where the radioactive level is lower
54 than the threshold, it would be too costly to apply soil removal techniques. It is important to
55 implement/facilitate any measures to reduce and/or completely remove the radioactivity in
56 these areas so that radioactivity could return to pre-accidental levels.

57

58 ^{137}Cs is strongly adsorbed to soil particles, especially clay and organic matter⁴ and thus can
59 be carried out of the field with runoff sediment. Soil erosion carrying ^{137}Cs from agricultural
60 lands could be considered a natural process to reduce the external dose of radioactivity in
61 farmland, especially when at least half of the deposition of ^{134}Cs and ^{137}Cs from FDNPP
62 occurred in the thin surface soil layer (0-4 cm)⁵ and downward movements of ^{134}Cs and ^{137}Cs
63 are restricted by strong adsorption behaviour.⁶ Recent studies have indicated that ^{134}Cs and
64 ^{137}Cs adsorbed to soil has been flushed by rainfall and snowmelt from the contaminated areas
65 possibly to the surrounding decontaminated farmlands and to the river system in large
66 quantities⁷ under severe weather conditions such as typhoon.^{8,9} However, little detail on the
67 relationship between the amount of radioactive cesium washed from contaminated farmlands
68 and the rainfall amount is reported and there are very few plot measurements to quantify soil
69 erosion and associated Cs export from cropland/paddies in this region.

70

71 In this study, we attempted to address this problem by i) investigating the level of soil erosion
72 and radioactive cesium discharge from farmlands in Fukushima prefecture using simulated
73 rainfalls and; ii) estimating the contribution of cesium discharge through runoff to the total
74 reduction of radioactivity in Fukushima farmlands.

75

76 **2. MATERIALS AND METHODS**

77 ***2.1. Study site***

78 This study was conducted in 2 upland fields in Nihonmatsu, Fukushima prefecture which are
79 about 30 km from FDNPP (Fig. 1). The study sites are not within the evacuation zone so easy
80 access was ensured during the study period. Based on the governmental air borne monitoring,

81 ^{134}Cs and ^{137}Cs depositions around the study site were 60-100 and 100-300 kBq/m²
 82 respectively on 16 November, 2012.¹⁰

83 The fields were ploughed before the experiment to ensure homogenous soil conditions. There
 84 was no rainfall in the area between ploughing and the rainfall simulation. Rainfall data were
 85 retrieved from the Japan Meteorological Agency website for Nihonmatsu meteorological
 86 station. The mean annual precipitation of the area is 1199 mm with typhoon season from
 87 September to October.

88 At each site, three plots of 5 m² (1m x 5m) were prepared inside the well-plough field using
 89 plastic borders (Fig. S1). The borders were driven ≈ 10 cm into the soil and were supported
 90 with additional soil at the base. Runoff at the downhill side of the plots was collected using a
 91 stainless steel V-shape drain covering the full width of the plot. The slopes were measured
 92 and initial surface soil samples (0–5cm) were taken at each individual plot before the rainfall
 93 simulation occurred. The slope and soil properties of the sites are listed in Table 1.

94

95 **Table 1.** Soil properties and field characteristics of experimental sites.

	Site 1	Site 2
Slope (%)	8.4	8.2
Moisture content (vol/vol)	0.29	0.27
Organic matter content (%)	3.14	4.24
Bulk density (g/cm ³)	0.96	0.94
^{134}Cs radioactivity (Bq/kg dry soil)	484	450
^{137}Cs radioactivity (Bq/kg dry soil)	1109	1062
Soil type (USDA classification)	Sandy loam	Sandy clay loam
Sand (%)	72	66
Silt (%)	23	6
Clay (%)	5	29
pH	6.68	6.17
Total K (mg/g)	0.005	0.01

96

97

98 **2.2. Rainfall simulation**

99 A swing type rainfall simulator was used in this experiment. The silicone nozzles were
100 mounted on a frame located approximately 1.6 m above the soil surface. To assess the rainfall
101 intensity delivered by the simulator, a plastic sheet was used to cover the experimental plot to
102 direct all simulated rainfall water into 10-L buckets and the time taken to fill the buckets was
103 measured.

104 The performance of the rainfall simulator was assessed by rain drop measurements of the
105 simulator after being calibrated at a rainfall intensity of 48.0 mm/hr. Drop sizes and rainfall
106 kinetic energy were measured using a laser drop-sizing gauge (LDG) consisting of a laser
107 transmitter and receiver (LX2-02; Keyence Corporation, Osaka, Japan), paired to an amplifier
108 (LX2-V10; Keyence Corporation, Osaka, Japan) and an A/D Converter (AXP-AD02; Adtek
109 System Science Co., Ltd., Yokohama, Japan) as described by Nanko et al.¹¹ Measurements
110 were performed for five minutes at three points, respectively: 1.25, 2.5, and 3.75 m from the
111 upstream edge on centre line of the plot.

112 The kinetic energy of natural rainfall in Japan was estimated from the equation with
113 parametric values of Japan derived from van Dijk et al.¹² The kinetic energy estimation
114 equation is expressed as:

$$115 \quad KE = 23.7I(1 - 0.51e^{-0.019I})$$

116 where KE is kinetic energy ($J m^{-2} h^{-1}$) and I is rainfall intensity ($mm h^{-1}$).

117

118 Field rainfall simulations were carried out in Site 1 in 15-16 November, 2013 and in Site 2 in
119 16-17 November 2013. Each simulation was designed to last for 60 min after the beginning
120 of the runoff event and a rainfall intensity of $50 mm h^{-1}$ was achieved. The value of the

121 rainfall intensity at 50mm h⁻¹ was selected (as calibrated) – with a return period of
 122 approximately 15 years. Lower rainfall intensity in Plot 2-3 at Site 2 was applied partly due
 123 to the shortage of water supply. Lack of water also shortened the simulation time in Plot 1-2
 124 at Site 1 (Table 2). During the simulation, all run-off water was collected and its volume was
 125 recorded. Two litres of run-off samples were taken every 10 min after run-off began. The
 126 samples were stored in plastic bottles and immediately transferred to the laboratory for
 127 analysis.

128 **Table 2.** Parameters of the rainfall simulations in different plots.

	Plot	Rainfall intensity (mm/h)	Rainfall amount (mm)	Runoff amount (mm)	Runoff coefficient (%)	Cumulative sediment loss (t/ha)
Site 1	1-1	50	55	26	47	1.41
	1-2*	50	23	5	22	0.42
	1-3	50	55	34	62	4.51
Site 2	2-1	50	55	42	76	7.18
	2-2	50	57	45	79	6.62
	2-3*	30	40	15	38	0.7

129 *Simulations during which water shortage occurred

130 **2.3. Sample analysis**

131 In the laboratory, run-off samples were filtered through a 1.1 µm glass microfiber filter paper
 132 (Whatman, GF/C) to separate the sediment phase from the liquid phase. Both soil and
 133 sediment samples were dried at 105 °C for two days. The sediment samples were regularly
 134 checked and stirred to ensure homogenous drying. Dry soil and sediment samples were
 135 grounded using a mortar and pestle, and preserved in tightly closed glass bottles prior to
 136 analysis.

137 About 5 g of dry sample was transferred to a plastic cup for measurement. The radioactivity
138 of ^{134}Cs and ^{137}Cs in the samples was determined using gamma-ray spectroscopy. Gamma-
139 ray emissions at energies of 604.7 keV (^{134}Cs) and 661.6 keV (^{137}Cs) were measured using a
140 high-purity germanium coaxial detector system (Ortec, GEM20-70) coupled to a multi-
141 channel analyzer (Ortec, DSPEC jr 2.0). The energy and efficiency calibrations for this
142 detector were performed using standard and background samples. All activities were decay-
143 corrected to the sampling date prior to statistical analyses. Particle size distribution of soil
144 and sediment samples were analysed using a laser diffraction particle size analyser (SALD-
145 2300, Shimadzu Co., Ltd., Kyoto, Japan).

146 Organic matter content of the samples was determined using loss-on-ignition (LOI) method.¹³
147 Briefly, around 3 g of air-dried sample were placed in a ceramic crucible of the electric
148 muffle furnace (Advantec. Ful230 FA) and heated to 105 °C for 12h. The sample was then
149 weighed to measure mass of dry sample. After that the dry sample was heated to 400 °C
150 overnight and weighed again. The value of organic matter content was calculated as the
151 difference between the initial and final weights of the dry sample divided by the initial dry
152 sample weight times 100%.

153

154 **3. RESULTS AND DISCUSSION**

155 ***3.1. Rainfall characteristics.***

156 During the performance assessment of the rainfall simulator, a total of 12,364 rain drops were
157 measured. The mean (\pm standard deviation) of the maximum drop diameter and mean median
158 drop size (D_{50}) at three points were 2.93 (\pm 0.03) mm and 1.63 (\pm 0.34) mm respectively. The
159 mean (\pm standard deviation) of rainfall intensity obtained from the LDG was 43.8 (\pm 11.1)
160 mmh^{-1} . The mean (\pm standard deviation) of rainfall kinetic energy was 702 (\pm 171) $\text{Jm}^{-2}\text{h}^{-1}$.

161 The kinetic energy of the rainfall simulator was compared to that of natural rainfall in Japan.
162 From the estimation, the kinetic energy produced by the rainfall simulator was equal to the
163 natural rainfall with intensity from 31 to 47 mm h⁻¹ which was considered acceptable for the
164 experiment.

165

166 **3.2. Runoff volume and soil erosion.**

167 3.2.1. Runoff volume

168 Rainfall simulations were carried out at the beginning of the dry season so there was no
169 significant rainfall before the experiment and soil moisture was considered lower than
170 infiltration capacity.

171 Runoff started within less than 10 minutes after the rainfall was initiated (average of 7
172 minutes at Site 1 and 8 minutes at Site 2), equivalent to less than 10 mm of rainfall. Such a
173 low threshold to generate runoff significantly increases the risk of runoff in this area. In
174 comparison, a similar rainfall simulation experiment on Andisol soil (light clay) in Tokyo
175 region required 30-50 minutes of rainfall with similar intensity (50 mm/h) to generate runoff
176 (unpublished data).

177 Fig. 2 showed that the runoff rates were high at both sites but they were stronger at Site 2
178 than at Site 1. The runoff rate increased quickly and plateaued when it approached the rainfall
179 rate (50 mm/h or 30 mm/h for plot 2-3). The runoff coefficients ranged up to a maximum of
180 62% in Site 1 (plot 1-3) while it reached 79% in Site 2 (plot 2-2) (Table 2).

181

182 3.2.2. Soil erosion

183 As shown in Fig. 3a, the concentrations of sediment in runoff water were relatively stable
184 throughout the rainfall event except at the time when runoff started (first flush effect). On
185 average, the sediment concentration measured in runoff water of Site 1 (sandy loam) was

186 significantly lower than that of Site 2 (sandy clay loam) ($p < 0.05$) but the effect should be
187 taken into account together with the sediment flux (sediment loss) from those plots. The
188 difference mostly caused by the lower sediment concentrations from plot 1-1 while sediment
189 concentrations from plot 1-3 were similar to those from 2-1 and 2-2 (data from 1-2 and 2-3
190 were excluded as the simulation condition were different). It is suggested that the variation in
191 field condition between the two Sites that caused such difference.

192 The rate of sediment loss in the experimental plots was shown in Fig. 3b. There was
193 significant difference between the erosion rates in two sites which was attributed mainly to
194 the difference in runoff rates between them (Fig. 2). The first flux effect also contributed
195 considerably to the difference in sediment loss as occurred in plot 2-1 and 2-2 compared to
196 other plots (Fig. 3b).

197 The maximum cumulative sediment loss at Site 2 was 7.18 t/ha with 42 mm of runoff amount
198 while the corresponding value at Site 1 was 4.51 t/ha with 34 mm of runoff amount (Table 2).
199 These values are comparable with the erosion rates reported for similar rainfall events¹⁴ and
200 fall within the range of the estimated annual soil erosion rates of 3.54 – 34 t/ha/year in
201 Japan.¹⁵ These estimations are reasonable since it came from a single rainfall event. Several
202 or longer heavy rains will probably increase the erosion rate.

203 Rainfall events of shorter duration (plot 1-2) or of lower intensity (plot 2-3) expectedly
204 produced a lower loss because they produced less runoff.

205 It should be noted that the runoff rate measured in this study was representative of the worst
206 case scenario since the fields at both sites were ploughed only a few days before the
207 experiment and had no vegetation cover available in all the plots. The bare and well ploughed
208 sandy loam/sandy clay loam soils were thus highly prone to runoff during high intensity
209 rainfall events.

210

211 ***3.3. The discharge of cesium through runoff/erosion.***

212 Cesium (^{134}Cs and ^{137}Cs) was discharged from the field to streams mainly through sediment
213 carried by runoff water because they are strongly adsorbed by organic matter or fine soil
214 particles.¹⁶ According to a previous study on the distribution of cesium in surrounding water
215 streams after the accident at FDNPP, the concentrations of cesium in the dissolved phase
216 were considered to have little impact on the total mass of cesium discharge and thus were not
217 measured in this study.¹⁷

218 The total concentrations of ^{134}Cs and ^{137}Cs in runoff sediments were presented in Fig. 4. The
219 concentrations of ^{134}Cs and ^{137}Cs in runoff sediment were higher (up to three folds) than the
220 average values measured in the plots before the experiment (Table 1). It could be attributed to
221 the smaller particle size of the runoff sediment as well as the higher concentration of organic
222 matter in the runoff sediments than those of the field soils (at both sites) as reported earlier.^{16,}
223 ¹⁸ In fact we observed a smaller average particle size of sediment samples than that of the
224 initial field soil samples (Fig. S2). We also found that there was a correlation between the
225 concentrations of cesium and the organic matter content of the sediments (Fig. 5) which was
226 in agreement with the results of previous studies.^{16, 19}

227 Cumulative losses of cesium (MBq/ha) during runoff were calculated and presented in Fig. 6.
228 As most cesium was carried by runoff sediment, the profile of cesium losses was very similar
229 to those of sediment as presented in Fig. 3b. The extent of cumulative losses of cesium from
230 the experiment plots was also similar to the extent of sediment losses in those plots (Fig. 3b
231 and Fig. 6).

232 Runoff volume and sediment concentrations are the two main factors that control the
233 discharge of cesium from the experimental plots. The findings of this study indicated that
234 compared to a large rainfall simulated in this study (~60 mm at 50 mm/h), shorter or smaller
235 rainfalls would result in a disproportionately smaller amount of cesium discharged to the

236 environment (see Fig. 6). Major rainfall events such as storms will certainly cause a surge in
237 the amount of cesium discharged to the environment as reported previously.²⁰

238

239 ***3.4. Implication for the field inventory of radioactive cesium and the*** 240 ***environmental fate and transport of cesium***

241 3.4.1. Effect of runoff on the inventory of radioactive cesium

242 The ratio of ^{134}Cs and ^{137}Cs measured in this study (Table 1) was expectedly in agreement
243 with the estimated ratio using natural decay rate (with the initial deposition ratio of
244 approximately 1:1).²¹ This ratio remained the same in the runoff sediment samples, i.e. that
245 the distribution of cesium in samples was relatively homogeneous. According to the
246 measurements in this study, the radioactivity of cesium has decreased by about 33% since the
247 nuclear incident with about 30% of the decrease due to the decay of ^{134}Cs .

248 During rainfall events, it was found that the average radioactivity of cesium in runoff
249 sediment (Bq/kg) was more than double that of the experimental field soil. The higher
250 radioactivity in runoff compared with field background means that the discharge of cesium
251 radioactivity through runoff may act as a way of land rehabilitation from cesium
252 contamination. However, compared to the total radioactivity of the field the maximum
253 amount of radioactive cesium discharged through runoff in this study was about 2% assuming
254 the depth of the contaminated soil layer of 10 cm due to the cultivation practice.⁹ Smaller and
255 shorter rainfalls would produce considerably lower discharge amounts. Therefore, runoff is
256 not considered an effective way to reduce the inventory of radioactive cesium in the field
257 (field decontamination). This finding is in agreement with data from previous study where no
258 evidence of change in the total inventory of ^{137}Cs could be found between the pre- and post-
259 rainy season of 2011 in different land uses in Fukushima prefecture.²² The same study also

260 suggested that rainfall events including intense rainfalls did not significantly modify the
261 profile of cesium with depth in the soil.²² Our estimation indicated that natural decay of
262 radioactive cesium, especially ¹³⁴Cs, will have larger impact on the field radioactivity than
263 rainfall runoff events (33% reduction by natural decay in 2.5 years against ~2% by one large
264 runoff event of 15-year return period).

265 It should be noted that the phenomenon that eroded soils contain higher concentrations of
266 ¹³⁷Cs than the bulk field soil would be considered in future studies where ¹³⁷Cs is used as
267 tracer to assess soil erosion.²³

268

269 3.4.2. Environmental fate and transport of cesium discharged from fields.

270 Soil erosion and subsequent sediment transport can play a major role in the dispersion of the
271 radioactive cesium in the natural environment because this element is strongly adsorbed to
272 the fine particle and the organic matter in the field.⁷ This study confirmed that the majority
273 of cesium mass transported out of the field was through sediment in runoff water. The
274 sediment eroded from fields loaded with radioactive cesium will be transported to the river
275 stream and deposited behind the dams before being released into the ocean.^{15, 17, 24, 25} Peak
276 levels of cesium radioactivity were observed in river water during the heavy rainfall events
277 with almost 100% contribution from particulate adsorbed cesium.²⁰ Ueda et al. reported that
278 the estimated discharge of radioactive cesium from river catchments in Fukushima prefecture
279 in 2011 was about 0.3-0.5% of the total cesium deposited on the catchments.²⁵ This reported
280 value was in agreement with the estimation from this study (~2%) considering that our data
281 was based on erosion mass from bare fields not from multiple land-use catchments
282 (consisting of e.g. forestry and meadow) in the previous study.²⁵

283 Although the discharge of radioactive cesium by soil erosion has no effect on the field
284 inventory of cesium, it may have significant consequences on water quality with high

285 concentrations of cesium (especially ^{137}Cs due to its long half-life) in runoff during extreme
286 rainfall events ¹⁵. Our experiments also confirmed that soil erosion was a way of
287 concentrating radioactive cesium in sediment, making the radioactivity of sediments found in
288 rivers higher than those measured in nearby soil. ²⁴ The flux of highly contaminated sediment
289 along rivers, especially during heavy rainfall events, will shift the load of radioactive soil
290 from upstream regions to coastal plains and eventually to the ocean. ^{8, 20} The peak load of
291 highly radioactive sediment in the rivers and coastal areas due to heavy rainfalls should be
292 considered carefully by local authorities to manage fishing and recreational activities in such
293 areas since contamination of sediment can also propagate to living organisms via changes in
294 food and habitat. ^{8, 9}

295

296 **4. CONCLUSIONS**

297 This study evaluated the impact of heavy rainfall to the runoff of radioactive cesium using
298 rainfall simulators. In the studied fields, a simulated heavy rainfall of 60 mm can cause
299 significant runoff with high runoff coefficients of up to 0.71. Results of sediment and
300 radioactive cesium analysis for runoff samples in this study can lead to the following
301 conclusions:

- 302 - Runoff caused by heavy rainfall can discharge large amount of radioactive cesium
303 from the field to the river system. In such a runoff event, almost all radioactive
304 cesium was transported out of the field via sediment.
- 305 - Concentrations of cesium in runoff sediment were several times higher than in the
306 field soil due to their higher organic matter content and smaller particle sizes.

307 - Soil erosion does not significantly change the field inventory of cesium in the field
308 but it may contribute significantly to the contamination of radioactive cesium in the
309 river system.

310

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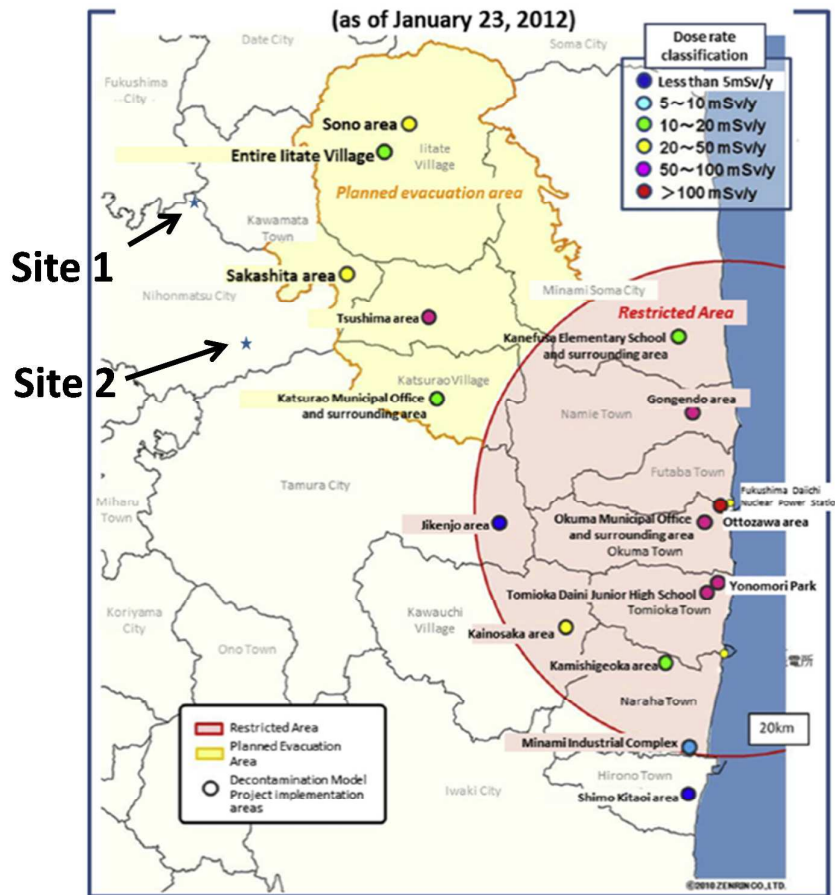
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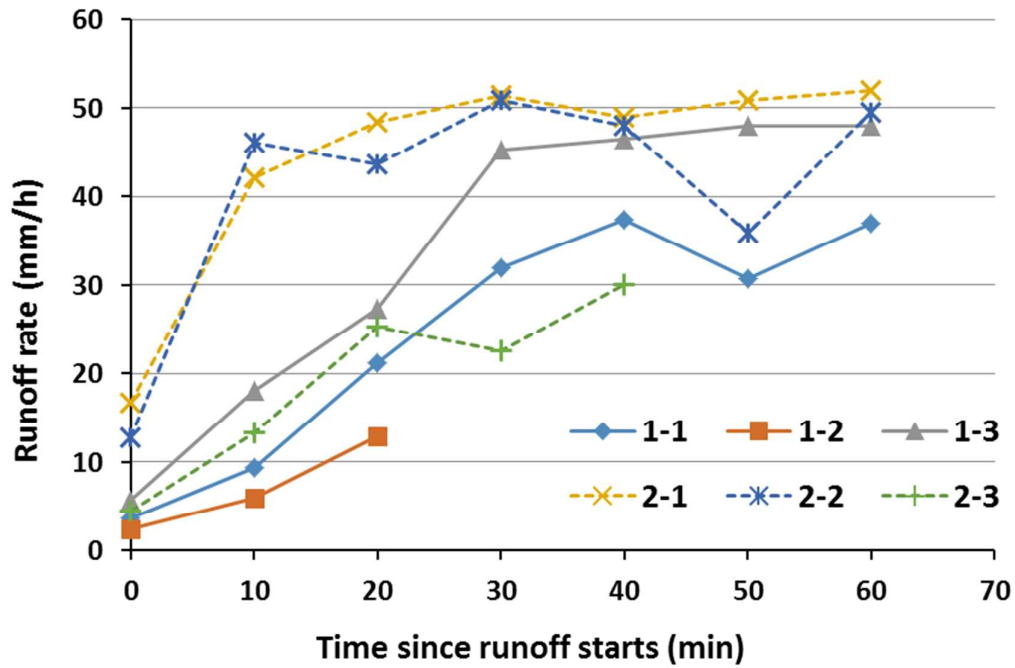
370 **Figure 1.** Location of the experimental sites in the Fukushima Prefecture (adapted from

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Hardie and McKinley, 2014).

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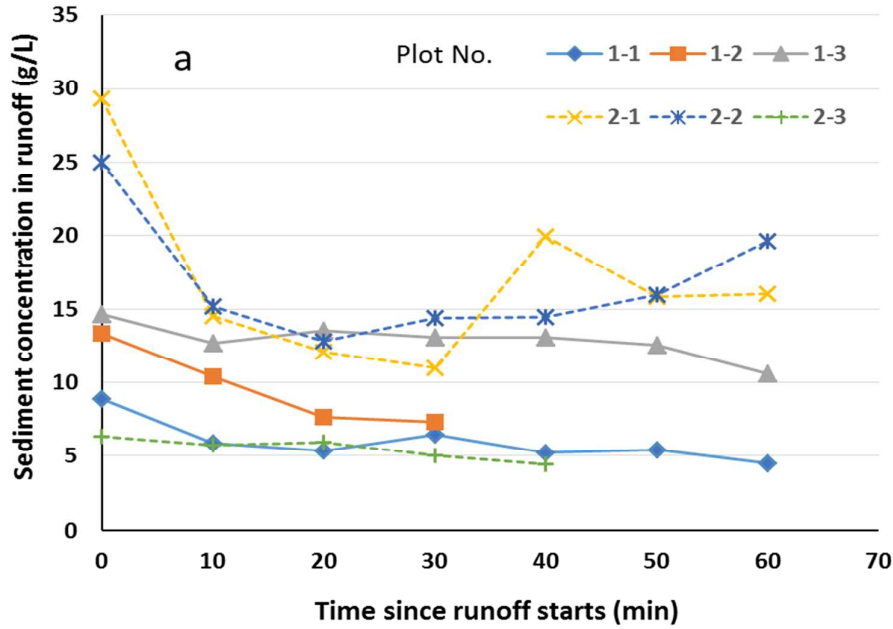
375 **Figure 2.** Runoff profile during the rainfall simulation in Site 1 (solid lines) and Site 2
376 (dashed lines).

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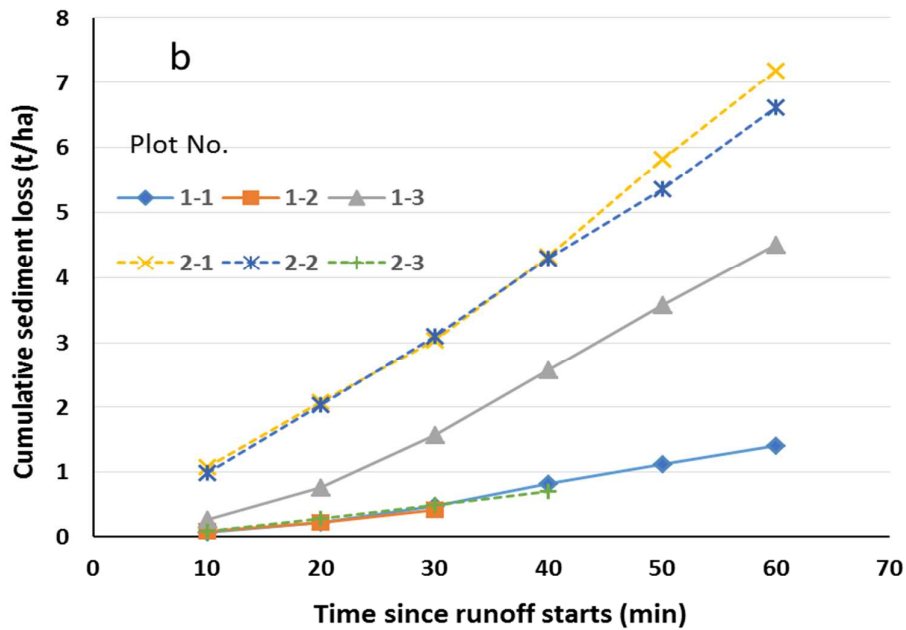
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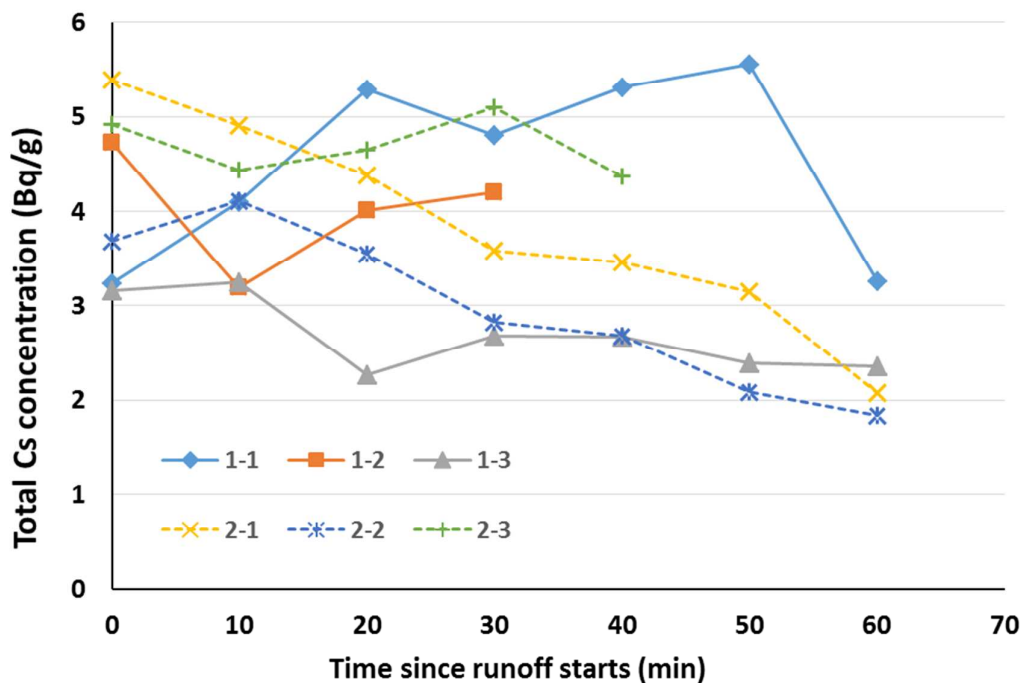
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385 **Figure 3.** Concentration (a) and cumulative loss (b) of sediment through runoff water during

386 the experiments.

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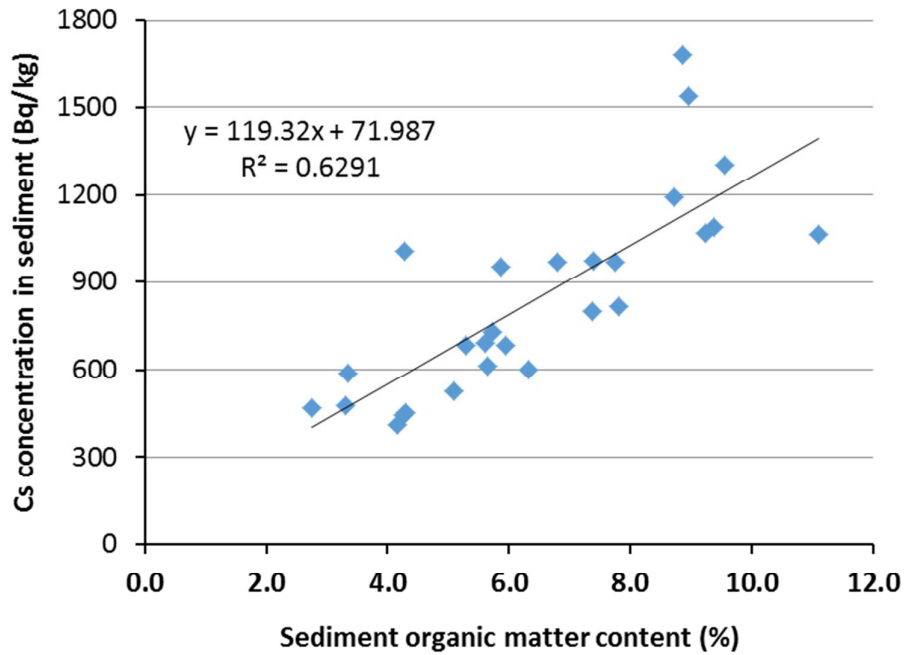
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390 **Figure 4.** Total concentration of ^{134}Cs and ^{137}Cs in sediment samples during the runoff events.

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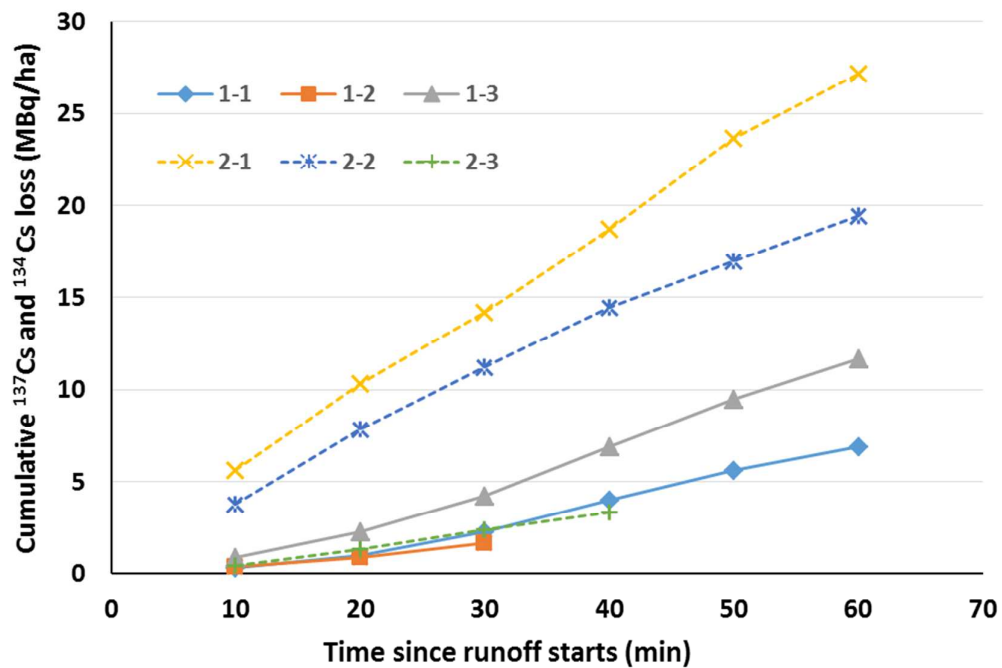


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394 **Figure 5.** Correlation between Cs concentration and organic matter content of sediment

395 samples in runoff water.



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397 **Figure 6.** Cumulative loss of radioactive caesium (^{134}Cs and ^{137}Cs) during the runoff events.

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