

JAAS

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

1
2
3
4 1
5
6 2 **Determination of cadmium in geological samples by aerosol**
7
8
9 3 **dilution ICP-MS after inverse aqua regia extraction**
10
11 4
12
13
14 5

15
16 6 Qian Xu, Wei Guo*, Lanlan Jin, Qinghai Guo, Shenghong Hu*
17
18 7

19
20
21 8 *State Key Laboratory of Biogeology and Environmental Geology, China University of*
22
23

24 9 *Geosciences, Wuhan, 430074, P. R. China*
25
26 10
27
28 11
29
30 12
31
32 13
33
34 14
35
36 15
37
38 16
39 *Corresponding Author. Tel and Fax: +86-27-67883495;
40
41 17
42 Email address: Wei.Guo@cug.edu.cn (Wei Guo); Shhu@cug.edu.cn (Shenghong Hu).
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Abstract

The determination of cadmium (Cd) in geological samples by inductively coupled plasma mass spectrometry (ICP-MS) is suffered to significant Mo and/or Zr based oxide interference. We have developed a valid method for Cd determination using Ar aerosol dilution ICP-MS after extraction with inverse aqua regia. Over 90% of the Zr was removed in the extraction procedure, and the residual Zr-hydroxides and Mo-oxides or hydroxides were successfully eliminated by adding an amount of Ar to the sample aerosol prior to the plasma. Compared to the conventional mode without adding Ar, the amount of oxide and hydroxide ions formed in the plasma was reduced by up to 90%. The relative yields of the interfering oxides and hydroxides were as low as 0.005% (MoOH/Mo or MoO/Mo) and 0.007% (ZrOH/Zr). Under the optimized dilution gas flow rate (0.85 L min⁻¹) and carrier gas flow rate (0.24 L min⁻¹), the limit of detection (LOD,3σ) for ¹¹¹Cd was 1.6 ng g⁻¹. The proposed method was applied to the determination of Cd in 81 soil, sediment, and rock standard reference materials (SRMs). The results for 68 of these geological SRMs were in good agreement with the reference values. The Cd levels in 10 limestone SRMs (GSR-22, GSR-23, GSR-24, GSR-27, GSR-28, GUI-1, GUI-2, GUI-2, DIAN-1, DIAN-2, and DIAN-3) were reported for the first time, and reference Cd values for other 3 geological SRMs (GSD-7, GSD-19, and GSM-1) were updated by this method. The results showed that this method has great potential for Cd determination in geological samples.

40 Introduction

41 Cd is an important trace element in the study of environmental geochemistry, and is
42 implicated in environmental quality and human health¹ as it is a highly toxic metal,
43 even at very low concentrations^{2,3}. The abundance of cadmium is just 0.08 $\mu\text{g g}^{-1}$ in
44 the Earth's crust⁴. Therefore, the accurate determination of Cd in geological samples
45 is often a challenge due to its low concentration and the presence of complex sample
46 matrices⁵.

47 Inductively coupled plasma mass spectrometry (ICP-MS) has been shown to be
48 the most applicable technique for the determination trace elements in various
49 geological samples, due to its high sensitivity, multi-element capability, and wide
50 dynamic range.⁶⁻¹⁷ However, the measurement of all isotopes of Cd is interfered by
51 various isobaric ions and/or oxide or hydroxide ions of Pd, Sm, In, Zr, Mo, Ru, Nb,
52 and Y. Specifically, the two common isotopes of ¹¹¹Cd and ¹¹⁴Cd suffer interference
53 from the polyatomic ions ⁹⁵Mo¹⁶O⁺, ⁹⁴Mo¹⁶OH⁺, ⁹⁴Zr¹⁶OH⁺, ⁹⁸Mo¹⁶O⁺, ⁹⁶Mo¹⁸O⁺ and
54 ⁹⁷Mo¹⁶O⁺, and the isobaric ion ¹¹⁴Sn⁺ (See ESI† Table S1).¹⁸ In typical geological
55 samples the contents of these interfering elements are hundreds to thousands of times
56 higher than that of Cd, as the abundances of Zr and Mo are 132 and 0.8 $\mu\text{g g}^{-1}$,
57 respectively, in the Earth's crust,⁴ which causes erroneous results in Cd determination
58 by ICP-MS method.¹⁹ Several techniques can be used to minimize this interference,
59 such as solvent or solid phase extractions,^{20, 21} precipitation,⁵ volatile species
60 generation,²² electrothermal volatilization (ETV),^{23, 24} membrane desolvation,²⁵
61 addition of carbon-containing compounds to modify the argon plasma,²⁶ isotope

1
2
3
4 62 dilution,^{27, 28} and dynamic reaction cell techniques (DRC).^{20, 29} Although these
5
6 63 methods can remove the matrix elements and polyatomic ion interference, they are
7
8 64 either time consuming or require special instrumental set-ups. Recently, an alternative
9
10 65 method employing online dilution of the sample aerosol with Ar before it enters the
11
12 66 plasma has been shown to reduce oxide formation without significantly compromising
13
14 67 detection limits.³⁰ When a dilution gas was used, the aerosol entering the plasma
15
16 68 contains much less water (and acid) than a conventionally diluted sample, resulting in
17
18 69 much hotter plasma and as much as ten times lower oxide formation.³¹ This technique
19
20 70 was shown to alleviate interference from the oxides of lower atomic number
21
22 71 lanthanoids on the detection of the heavy-lanthanoids in the analysis of airborne
23
24 72 particulate matter, but only a limited level of interfering species (ca. 20-fold) was
25
26 73 addressed in this study.³¹

27
28
29
30
31
32
33
34 74 In this work, a method for the direct determination of trace-level Cd in
35
36 75 geological samples based on aerosol dilution ICP-MS (in combination with sample
37
38 76 pre-treatment by boiling in inverse aqua regia extraction) was developed, aiming to
39
40 77 eliminate the complicated polyatomic interference from Zr and Mo based oxides and
41
42 78 hydroxides. The optimization of the technique and its application to the determination
43
44 79 of trace Cd in a series of geological standard reference materials (SRMs) are
45
46 80 discussed in detail.

47
48
49
50
51 81

52 82 **Materials and methods**

53 54 55 56 57 83 **Instrumentation, reagents and standards**

1
2
3
4 84 An Agilent 7700x ICP-MS (Agilent Technologies, USA) with an aerosol dilution
5
6 85 system (Fig. 1), a MicroMist nebulizer and a Peltier cooled (2 °C) quartz Scott-type
7
8
9 86 double pass spray chamber was used for this work. The operating parameters for
10
11 87 ICP-MS were optimized prior to the determinations with a tuning solution containing
12
13 88 1.0 ng mL⁻¹ of Ce, Co, Li, Mg, Tl and Y, and the oxide production rate, calculated as
14
15 89 the ratio of CeO⁺/Ce⁺ was reduced from 3.0% to less than 0.3%. The optimized
16
17 90 instrument operating parameters are listed in Table 1. High purity water (18.2 MΩ
18
19 91 cm⁻¹) used for the preparation of all blanks, standards and samples was obtained from
20
21 92 a Millipore water purification system (Millipore, France). HNO₃ (99.999%), HCl
22
23 93 (99.999%), and HF (99.999%) were purchased from Alfa Aesar Ltd. (Tianjin). Single
24
25 94 element stock solutions (Cd, Mo, Zr, and Rh, *etc.*) were purchased from the National
26
27 95 Center for Analysis and Testing of Steel Materials (NCATSM, China). The accuracy
28
29 96 of the method was assessed by using two rock SRMs (andesite AGV-2 and basalt
30
31 97 BCR-2) obtained from the United State Geological Survey (USGS, USA), two SRMs
32
33 98 (metal-rich sediment SdAR-M2 and gabbro GSM-1) obtained from the International
34
35 99 Association of Geoanalysts (IAG, UK), and 77 Chinese geological SRMs (details
36
37 100 given in Table S2, see ESI†), from the Institute of Geophysical and Geochemical
38
39 101 Exploration (IGGE, China).

102 **Sample preparation by boiling inverse aqua regia extraction**

103 0.2000 g of the homogenized sample powder (200 mesh) was weighed into a 10 mL
104 polyethylene tube, to which 3.0 mL of a freshly prepared inverse aqua regia solution

1
2
3
4 105 (HNO₃:HCl=3:1) was added. The capped tube was then placed in a water bath at
5
6 106 95 °C for 2h. After cooling to room temperature, the solution was diluted to 10 mL
7
8 107 with high purity water. The solution was centrifuged to remove the undissolved solid
9
10 108 particles, and 1.0 mL of supernatant was transferred to a 10 mL polyethylene tube.
11
12 109 Water (4.0 mL) was added, and the tube was shaken vigorously. For comparison, a
13
14 110 conventional complete digestion method employing closed pressurized digestion with
15
16 111 a mixture of HF + HNO₃ was also used (see ESI† Table S3).^{32, 33}
17
18
19
20
21
22
23
24
25

26 113 **Results and discussion**

27 28 29 114 **Optimization of aerosol dilution ICP-MS after inverse aqua regia extraction**

30
31
32 115 ¹¹¹Cd was selected as the isotope for ICP-MS analysis because it is the only isotope of
33
34 116 Cd without isobaric interference. The major problem encountered in the determination
35
36 117 of ¹¹¹Cd is interference from Zr and Mo polyatomic ions (i.e., ⁹⁵Mo¹⁶O⁺, ⁹⁴Mo¹⁶OH⁺,
37
38 118 and ⁹⁴Zr¹⁶OH⁺).³⁴ In this study, the elimination of the interference from Zr/Mo-oxides
39
40 119 using aerosol dilution was investigated for the analysis of the geological samples. As
41
42 120 shown in Fig. 2, there is a significant decrease in the ratios of oxide species
43
44 121 (⁹⁵Mo¹⁶O⁺/⁹⁵Mo⁺, ⁹⁴Mo¹⁶OH⁺/⁹⁵Mo⁺, and ⁹⁴Zr¹⁶OH⁺/⁹⁴Zr⁺) formed in the plasma with
45
46 122 increasing gas dilution. At the optimized dilution gas rate (0.85 L min⁻¹), the ratios of
47
48 123 (MoOH⁺ + MoO⁺)/Mo⁺ and ZrOH⁺/Zr⁺ are 0.005% and 0.007%, respectively, which
49
50 124 is 10 times smaller than that observed with the conventional standard mode (i.e.,
51
52 125 dilution gas = 0) (Fig. 2). The signal contribution to m/z 111 from different elements
53
54
55
56
57
58
59
60

1
2
3
4 126 under the standard mode and the aerosol dilution mode was compared. As shown in
5
6 127 [Table 2](#), under standard mode, the signal intensities for m/z 111 are 12328 cps, 2987
7
8 128 cps, and 12080 cps for Cd (1 ng mL^{-1}), Mo (200 ng mL^{-1}) and Zr (1000 ng mL^{-1}),
9
10
11 129 respectively. Conversely, under aerosol dilution mode, the signal intensities for m/z
12
13
14 130 111 are 1754 cps (Cd), 24.8 cps (Mo), and 91.5 cps (Zr). Compared to the
15
16 131 conventional ICP-MS method (without aerosol dilution), the signal intensity of Cd
17
18 132 dropped by 7 times, but the signal background ratio (SBR) increased from 0.82 to
19
20
21 133 15.1 by addition of the dilution gas ([Table 2](#)). Aerosol dilution ICP-MS was evaluated
22
23
24 134 by the analysis of geological SRMs with high Zr and/or Mo contents (i.e. Zr content /
25
26 135 Cd content >3000 , and Mo content /Cd content >90). [Table 3](#) lists the results obtained
27
28
29 136 by aerosol dilution ICP-MS and standard mode ICP-MS following the conventional
30
31 137 complete digestion procedure. The Cd values obtained by aerosol dilution ICP-MS are
32
33
34 138 much closer to the certified values than those obtained by standard mode ICP-MS,
35
36 139 due to the reduction of the Zr or Mo based oxide levels. When the proposed aerosol
37
38
39 140 dilution ICP-MS technique coupled with the conventional complete digestion method
40
41 141 is employed, unsatisfactory results are obtained for all selected SRMs with a Zr
42
43
44 142 content /Cd content > 1500 . However, the result was satisfactory for the stream
45
46
47 143 sediment SRM (GSD-3a), which possesses a relatively low Zr content /Cd content
48
49 144 ratio (566). Most geological samples have a high Zr content (Zr content /Cd content
50
51 145 ratios > 1500) and a low Mo content (Mo content /Cd content ratios <100). Therefore,
52
53
54 146 use of the aerosol dilution technique can effectively reduce the Mo based
55
56
57 147 interferences, but it cannot completely eliminate Zr based interferences. Therefore,
58
59
60

1
2
3
4 148 further study should be conducted to obtain more accurate results.
5
6

7 149 In our recent paper, a pre-treatment procedure based on extraction with boiling
8
9
10 150 aqua regia was established to alleviate the interference from matrix Zr on As
11
12 151 determination in DRC ICP-MS.²⁹ As was extracted almost completely (>98%) but the
13
14 152 Zr recovery was <5%. This is because Zr is a lithophilic element which is strongly
15
16
17 153 bound in silicate minerals and cannot be dissolved out by this procedure.²⁹ A similar
18
19
20 154 result for Ag and Cd determination in environmental samples with low Zr content /Cd
21
22 155 content ratios was reported by Li *et al.*³⁵ To evaluate the feasibility of this preparation
23
24 156 method, three geological SRMs (soil GSS-4, stream sediment GSD-9, and rock
25
26
27 157 GSR-2) were extracted with boiling aqua regia (HCl: HNO₃ = 3:1) or inverse aqua
28
29
30 158 regia (HCl: HNO₃ = 1:3). Table 4 shows that the recoveries of Zr are less than 10%
31
32 159 using both extraction methods. However, much larger amounts of Zr (>95%) are
33
34
35 160 dissolved by the conventional complete digestion procedure, which results in the
36
37
38 161 positively biased results for Cd determination in standard mode ICP-MS. As can be
39
40 162 seen from Table 4, some little positive bias for Cd was observed using this extraction
41
42
43 163 procedure, which originates from the residual or dissolved Zr (<10%) and Mo
44
45 164 (>60%).
46

47 165 Residual Zr and Mo were completely eliminated using the proposed aerosol
48
49
50 166 dilution technique. It can be seen in Table 5 that the values obtained for the geological
51
52 167 SRMs containing high levels of matrix Zr and/or Mo using aerosol dilution ICP-MS
53
54
55 168 after inverse aqua regia extraction, without any correction, are in good agreement with
56
57
58 169 the certified values. Although the values of Zr content /Cd content for GSS-3 and Mo
59
60

1
2
3
4 170 content /Cd content for GSD-3 are high at 4100 and 920, respectively, our obtained
5
6 171 values for Cd content are 0.054 ± 0.001 and $0.093 \pm 0.013 \mu\text{g g}^{-1}$, which agree well
7
8 172 with the certified values of 0.060 ± 0.009 and $0.100 \pm 0.020 \mu\text{g g}^{-1}$, respectively. The
9
10 173 results clearly show that the proposed method based on aerosol dilution ICP-MS
11
12 174 combined with the boiling inverse aqua regia extraction is reliable. The extraction
13
14 175 procedure separates most of the Zr matrix (>90%) from the analyte, and the remaining
15
16 176 Zr (<10%) and Mo are eliminated by the subsequent aerosol dilution technique.
17
18
19
20
21 177
22
23

24 178 **Analysis of geological SRMs**

25
26
27 179 The signal background ratio (SBR) of Cd increases 200-fold (from 0.82 to 160, as
28
29 180 shown in Fig. S1, see ESI†) by our proposed method compared to conventional mode
30
31 181 ICP-QMS employing the complete digestion procedure. The limit of detection (LOD,
32
33 182 3σ) is 1.6 ng g^{-1} (taking into account the 250-fold dilution factor). This sensitivity is
34
35 183 sufficient for trace-level Cd determination of solid geological samples, in which the
36
37 184 Cd levels typically range from 0.02 to $2.0 \mu\text{g g}^{-1}$.
38
39
40
41

42 185 Two USGS SRMs (andesite AGV-2 and basalt BCR-2) and two IAG (metal-rich
43
44 186 sediment SdAR-M2 and gabbro GSM-1) were assessed using the proposed method, as
45
46 187 shown in Table 6. For comparison, the reference values and those reported in the
47
48 188 literature are listed. As shown in Table 6, the Cd contents determined for AGV-2 and
49
50 189 BCR-2 are in agreement with the USGS reference values. However, the Cd levels in
51
52 190 AGV-2 and BCR-2 obtained using the conventional ICP-QMS coupled with the
53
54 191 complete digestion procedure ($0.121 \mu\text{g g}^{-1}$ and $0.227 \mu\text{g g}^{-1}$ for AGV-2, $0.082 \mu\text{g g}^{-1}$

1
2
3
4 192 and $0.75 \mu\text{g g}^{-1}$ for BCR-2)^{36, 37} are significantly higher than the USGS reference
5
6 193 values ($0.061 \mu\text{g g}^{-1}$ and $0.140 \mu\text{g g}^{-1}$, respectively) and our reported values ($0.058 \pm$
7
8 194 $0.004 \mu\text{g g}^{-1}$ and $0.148 \pm 0.007 \mu\text{g g}^{-1}$, respectively). For the two IAG SRMs
9
10 195 SdAR-M2 and GSM-1, which were used as an international proficiency test for
11
12 196 analytical geochemistry laboratories in 2014 (*GeoPT36*),³⁸ our analysis yielded values
13
14 197 of $5.10 \pm 0.20 \mu\text{g g}^{-1}$ and $0.072 \pm 0.004 \mu\text{g g}^{-1}$, respectively. The value for SdAR-M2
15
16 198 is consistent to the IAG reference value ($5.10 \pm 0.19 \mu\text{g g}^{-1}$), but the Cd level
17
18 199 determined value for GSM-1 ($0.072 \pm 0.004 \mu\text{g g}^{-1}$) is lower than the
19
20 200 recommended value ($0.128 \pm 0.012 \mu\text{g g}^{-1}$). Ten separate aliquots of this SRM were
21
22 201 analysed over a period of three months, and consistent results (0.070 - $0.076 \mu\text{g g}^{-1}$)
23
24 202 were obtained (Fig. S2, see ESI†). GSM-1 (gabbro) is a basic intrusive rock, which
25
26 203 differs from BCR-2 (basalt), which is a basic extrusive rock. The efficiency of Cd
27
28 204 extraction from the different basic rocks may be influenced by its crystallization
29
30 205 process, and further study is required to explain this phenomenon.

31
32
33
34
35
36
37
38
39 206 The proposed method was also employed to determine Cd levels in 77 Chinese
40
41 207 geological SRMs, comprising 28 soils, 30 sediments, and 19 rocks. The measured Cd
42
43 208 levels in 65 geological samples are in good agreement with their certified values (Fig.
44
45 209 3). Although Cd is present at ultra-trace levels ($< 0.20 \mu\text{g g}^{-1}$) in 40 of the geological
46
47 210 SRMs, the measured values are consistent with the certified values (see inset of Fig 3).
48
49 211 Detailed results for these SRMs are given in Table S2 (see ESI†). Table 7 shows the
50
51 212 values determined for two stream sediments (GSD-7 and GSD-19), which are
52
53 213 inconsistent with the IGGE certified values. In addition, ten limestone SRMs
54
55
56
57
58
59
60

1
2
3
4 214 (GSR-22, GSR-23, GSR-24, GSR-27, GSR-28, GUI-1, GUI-2, DIAN-1, DIAN-2, and
5
6 215 DIAN-3) have no current reference values or literature values, and the recommended
7
8 216 values of these SRMs are provided by this work (Table 7). The results of repeated
9
10 217 analysis, in which the SRMs were extracted and re-analysed ten times over a period of
11
12 218 three months, are consistent (Fig. 4), which indicates that our recommended values
13
14 219 are reliable. This method also could be used for the determination of other trace
15
16 220 elements, such as Ag (Table S4, see ESI[†]).
17
18
19
20
21
22

221

222 **Conclusions**

223 A method for the determination of trace-level Cd in geological samples using aerosol
224 dilution ICP-MS after extraction with boiling inverse aqua regia was developed. The
225 significant interfering Zr or Mo based oxides and hydroxides was effectively reduced
226 by the proposed method, compared to conventional ICP-QMS coupled with the
227 complete digestion method, the SBR of Cd improved by two orders of magnitude.
228 The proposed method was employed to analyse 81 geological SRMs, and the results
229 of 68 samples were in good agreement with the certified or literature values. The Cd
230 levels of 13 other SRMs have been provided for the first time by this method. The
231 developed method shows great potential for the direct determination of trace levels of
232 Cd in geological samples.

233

234 **Acknowledgements**

1
2
3
4 235 This work was supported by the National Nature Science Foundation of China (No.
5
6 236 21207120), the National Key Scientific Instrument and Equipment Development
7
8 237 Projects of China (no. 2011YQ06010008), and the Fundamental Research Funds for
9
10 238 the China University of Geosciences (Wuhan) (no. CUGL140411).
11
12
13
14
15
16
17

239

240 **References**

- 241 1. J. Sastre, A. Sahuquillo, M. Vidal and G. Rauret, *Analytica Chimica Acta*, 2002,
22 242 **462**, 59-72.
23
24
25 243 2. B. J. Alloway, *Heavy metals in soils*, Springer Science & Business Media. 2013.
26
27 244 3. M. Shirani, A. Semnani, S. Habibollahi and H. Haddadi, *Journal of Analytical*
28
29 245 *Atomic Spectrometry*, 2015.
30
31 246 4. R. Rudnick and S. Gao, *Treatise on geochemistry*, 2003, **3**, 1-64.
32
33 247 5. K. Inagaki, A. Takatsu, A. Uchiumi, A. Nakama and K. Okamoto, *Journal of*
34
35 248 *Analytical Atomic Spectrometry*, 2001, **16**, 1370-1374.
36
37 249 6. J. Avivar, L. Ferrer, M. Casas and V. Cerda, *Journal Of Analytical Atomic*
38
39 250 *Spectrometry*, 2012, **27**, 327-334.
40
41 251 7. D. Wilhelms-Dick, T. Westerhold, U. Rohl, F. Wilhelms, C. Vogt, T. J. J.
42
43 252 Hanebuth, H. Rommermann, M. Kriews and S. Kasten, *Journal Of Analytical Atomic*
44
45 253 *Spectrometry*, 2012, **27**, 1574-1584.
46
47 254 8. V. Yilmaz, L. Rose, Z. Arslan and M. D. Little, *Journal Of Analytical Atomic*
48
49 255 *Spectrometry*, 2012, **27**, 1895-1902.
50
51 256 9. J. Liu, D. H. Zou, X. L. Yang, X. Z. Cheng and Q. Chen, *Analytical Methods*,
52
53
54
55
56
57
58
59
60

- 1
2
3
4 257 2013, **5**, 6774-6780.
5
6 258 10. T. Ohno, Y. Muramatsu, Y. Shikamori, C. Toyama, N. Okabe and H. Matsuzaki,
7
8 259 *Journal Of Analytical Atomic Spectrometry*, 2013, **28**, 1283-1287.
9
10
11 260 11. P. Foltynova, A. Bednarik, V. Kanicky and J. Preisler, *Journal Of Analytical*
12
13 261 *Atomic Spectrometry*, 2014, **29**, 1585-1590.
14
15
16 262 12. N. S. Horner and D. Beauchemin, *Journal Of Analytical Atomic Spectrometry*,
17
18 263 2014, **29**, 715-720.
19
20
21 264 13. A. Pullen, M. Ibanez-Mejia, G. E. Gehrels, J. C. Ibanez-Mejia and M. Pecha,
22
23 265 *Journal Of Analytical Atomic Spectrometry*, 2014, **29**, 971-980.
24
25
26 266 14. M. Tang, W. F. McDonough and R. Arevalo, *Journal Of Analytical Atomic*
27
28 267 *Spectrometry*, 2014, **29**, 1835-1843.
29
30
31 268 15. B. B. Chen, Y. L. Wu, X. Q. Guo, M. He and B. Hu, *Journal Of Analytical Atomic*
32
33 269 *Spectrometry*, 2015, **30**, 875-881.
34
35
36 270 16. W. Guo, S. H. Hu, Z. W. Wu, G. Y. Lan, L. L. Jin, X. G. Pang, J. C. Zhan, B.
37
38 271 Chen and Z. Y. Tang, *Journal Of Analytical Atomic Spectrometry*, 2015, **30**, 986-993.
39
40
41 272 17. R. Khan, Y. Yokozuka, S. Terai, N. Shirai and M. Ebihara, *Journal Of Analytical*
42
43 273 *Atomic Spectrometry*, 2015, **30**, 506-514.
44
45
46 274 18. T. W. May and R. H. Wiedmeyer, *Atomic Spectroscopy*, 1998, **19**, 150-155.
47
48
49 275 19. W. Guo, S. H. Hu, Y. F. Xiao, H. F. Zhang and X. J. Xie, *Chemosphere*, 2010, **81**,
50
51 276 1463-1468.
52
53
54 277 20. T. C. Duan, X. J. Song, P. R. Guo, H. F. Li, L. H. Pan, H. T. Chen and J. W. Xu,
55
56 278 *Journal Of Analytical Atomic Spectrometry*, 2007, **22**, 403-406.
57
58
59
60

- 1
2
3
4 279 21. P. H. Liao, S. J. Jiang and A. C. Sahayam, *Journal Of Analytical Atomic*
5
6 280 *Spectrometry*, 2012, **27**, 1518-1524.
7
8
9 281 22. W. N. Chen, S. J. Jiang, Y. L. Chen and A. C. Sahayam, *Analytica chimica acta*,
10
11 282 2015, **860**, 8-14.
12
13 283 23. S. M. Maia, D. Pozebon and A. J. Curtius, *Journal Of Analytical Atomic*
14
15 284 *Spectrometry*, 2003, **18**, 330-337.
16
17
18 285 24. M. L. Lin and S. J. Jiang, *Journal Of Analytical Atomic Spectrometry*, 2011, **26**,
19
20 286 1813-1818.
21
22 287 25. X. X. Duan and M. Regelous, *Journal Of Analytical Atomic Spectrometry*, 2014,
23
24 288 **29**, 2379-2387.
25
26
27 289 26. D. Karunasagar and J. Arunachalam, *Analytica chimica acta*, 2001, **441**, 291-296.
28
29
30 31 27. A. Makishima, H. Kitagawa and E. Nakamura, *Geostand Geoanal Res*, 2011, **35**,
32
33 34 57-67.
35
36 37 28. R. Q. Thompson and S. J. Christopher, *Analytical Methods*, 2013, **5**, 1346-1351.
38
39 39 29. W. Guo, S. H. Hu, J. Y. Zhang, L. L. Jin, X. J. Wang, Z. L. Zhu and H. F. Zhang,
40
41 42 294 *Journal Of Analytical Atomic Spectrometry*, 2011, **26**, 2076-2080.
43
44 45 30. S. Doker and M. Uslu, *Food Anal Method*, 2014, **7**, 683-689.
46
47 48 31. V. Celo, E. Dabek-Zlotorzynska, J. J. Zhao, I. Okonskaia and D. Bowman,
49 297 *Analytica chimica acta*, 2011, **706**, 89-96.
50
51 52 298 32. W. Guo, S. H. Hu, X. J. Wang, J. Y. Zhang, L. L. Jin, Z. L. Zhu and H. F. Zhang,
53
54 299 *Journal Of Analytical Atomic Spectrometry*, 2011, **26**, 1198-1203.
55
56 57 300 33. W. Guo, S. H. Hu, J. Y. Zhang and H. F. Zhang, *Sci Total Environ*, 2011, **409**,

1
2
3
4 301 2981-2986.
5

6 302 34. W. Guo, S. H. Hu, J. A. Zhao, S. S. Jin, W. J. Liu and H. F. Zhang, *Microchemical*
7
8 *Journal*, 2011, **97**, 154-159.
9

10 304 35. Z. X. Li, L. P. Zhou and F. Tan, *Microchim Acta*, 2006, **156**, 263-269.

11 305 36. Z. C. Hu and S. Gao, *Chem Geol*, 2008, **253**, 205-221.

12 306 37. S. K. Marx and B. S. Kamber, *Applied Geochemistry*, 2010, **25**, 1221-1237.

13 307 38. P. C. Webb, M. Thompson, P. J. Potts and S. A. Wilson, *GeoPT36-an international*
14
15
16
17
18
19
20
21
22 308 proficiency test for analytical geochemistry laboratories - report on round 36 / August

23 309 2014 (Gabbro, GSM-1). *International Association of Geoanalysts*: Unpublished
24
25
26
27 310 report.

28
29 311
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3
4 312

5
6 313 **Figure captions**

7
8
9 314

10
11
12 315 **Fig. 1.** Schematic diagram of our aerosol dilution system.

13
14
15 316 **Fig. 2.** Effects of dilution gas flow rate on oxide formation ratios ($(^{94}\text{Mo}^{16}\text{OH}^+ +$
16
17 $^{95}\text{Mo}^{16}\text{O}^+)/^{95}\text{Mo}^+$ and $^{94}\text{Zr}^{16}\text{OH}^+ / ^{94}\text{Zr}^+$) and the signal background ratio (SBR).
18
19

20
21
22 318 **Fig. 3.** Cd values determined for 65 geological SRMs by the proposed method vs.
23
24 319 their certified values. The inset shows the 40 geological SRMs containing ultra-trace
25
26 320 levels of Cd ($< 0.20 \mu\text{g g}^{-1}$).
27
28

29
30 321 **Fig. 4.** Stability of Cd values obtained through the proposed method in 12 geological
31
32 322 SRMs: two stream sediments: GSD-7 (a) and GSD-19 (b), and ten limestones SRMs:
33
34 323 GSR-22 (c), GSR-23 (d), GSR-24 (e), GSR-27 (f), GSR-28 (g), GUI-1 (h), GUI-2 (i),
35
36 324 DIAN-1 (j), DIAN-2 (k), and DIAN-3 (l). For each SRM, ten separate aliquots of the
37
38 325 samples were analysed over a period of three months.
39
40
41
42

43
44 326
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

327

328

Table 1. Instrument operating parameters

ICP-MS instrument	Agilent 7700x ICP-MS
RF Power, W	1450
Plasma gas flow, L min ⁻¹	15
Nebulizer gas flow, L min ⁻¹	0.24 for aerosol dilution mode; 1.09 for standard mode.
Dilution gas flow, L min ⁻¹	0.85 for aerosol dilution mode; 0 for standard mode.
Sampling depth ,mm	7.0
Sample uptake rate, mL min ⁻¹	0.3
Dwell time, ms	30
Sweeps	100
Readings	1
Replicates	3
¹⁴⁰ CeO/ ¹⁴⁰ Ce ratio (%)	< 0.3 for aerosol dilution mode; < 3.0 for standard mode.
Isotope monitored	¹¹¹ Cd

329

330

331

Table 2. Signal contribution to m/z 111 by different elements (n=3)

Signal intensity	Elements								Signal background ratio (SBR)
	Blank, pure water		Cd, 1 ng mL ⁻¹		Mo, 200 ng mL ⁻¹		Zr, 1000 ng mL ⁻¹		
	Intensity, cps	RSD, %	Intensity, cps	RSD, %	Intensity, cps	RSD, %	Intensity, cps	RSD, %	
Conventional ICP-MS	6.67	9.1	12328	0.9	2987	1.3	12080	1.2	0.82
Aerosol dilution ICP-MS	1.11	8.5	1754	1.2	24.8	6.5	91.5	5.6	15.1

336

337

Table 3. Results of Cd levels in six geological SRMs obtained through aerosol dilution ICP-MS after conventional complete digestion.

Geological SRMs	Zr/Cd	Mo/Cd	Cd, $\mu\text{g g}^{-1}$ (n = 3)		
			Aerosol dilution	Certified Value	Standard mode
			ICP-MS		ICP-MS
GSS-3, yellow-brown earth soil	4100	5	0.113 ± 0.001	0.060 ± 0.009	0.303 ± 0.006
GSS-6, red soil	1692	138	0.180 ± 0.010	0.130 ± 0.030	0.402 ± 0.012
GSS-22, soil	3923	10	0.113 ± 0.003	0.065 ± 0.012	0.330 ± 0.020
GSD-3, stream sediment	2200	920	0.128 ± 0.007	0.100 ± 0.020	0.425 ± 0.002
GSD-3a, stream sediment	566	96	0.520 ± 0.010	0.500 ± 0.060	0.770 ± 0.040
GSR-4, quartz sandstone	3567	13	0.094 ± 0.012	0.060 ± 0.016	0.340 ± 0.020

338

339

1
2
3
4
5
6
7 3408 **Table 4.** Results for Zr, Mo, and Cd levels in geological SRMs obtained through conventional ICP-MS without aerosol dilution using different
9
10 342 sample preparation methods.

Geological SRMs	Elements	Cd, $\mu\text{g g}^{-1}$ (n = 3)			Certified values
		Complete digestion	Extraction with aqua regia	Extraction with inverse aqua regia	
GSS-4, soil	Zr	531 ± 5	16.6 ± 0.4	23.0 ± 0.8	500 ± 42
	Mo	2.30 ± 0.4	2.60 ± 0.02	1.80 ± 0.10	2.60 ± 0.30
	Cd	1.05 ± 0.05	0.420 ± 0.010	0.393 ± 0.005	0.350 ± 0.06
GSD-9, sediment	Zr	387 ± 8	8.32 ± 0.04	10.1 ± 0.1	370 ± 20
	Mo	0.750 ± 0.010	0.537 ± 0.005	0.610 ± 0.010	0.640 ± 0.110
	Cd	0.680 ± 0.020	0.330 ± 0.01	0.320 ± 0.010	0.260 ± 0.040
GSR-2, rock	Zr	93.0 ± 2.0	9.70 ± 0.70	7.30 ± 0.40	99.0 ± 11.0
	Mo	0.600 ± 0.020	0.350 ± 0.020	0.351 ± 0.004	0.540 ± 0.090
	Cd	0.142 ± 0.005	0.088 ± 0.005	0.080 ± 0.003	0.061 ± 0.014

11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35 34336 344
37
38
39
40
41
42
43
44
45
46
47
48
49

Table 5. Results for Cd levels in six geological SRMs obtained through different ICP-MS methods after extraction with boiling inverse aqua regia.

Geological SRMs	Zr/Cd	Mo/Cd	Cd, $\mu\text{g g}^{-1}$ (n = 3)		
			Aerosol dilution ICP-MS	Certified value	Standard mode ICP-MS
GSS-3, yellow-brown earth soil	4100	5	0.054 ± 0.001	0.060 ± 0.009	0.069 ± 0.013
GSS-6, red soil	1692	138	0.134 ± 0.007	0.130 ± 0.030	0.179 ± 0.004
GSS-22, soil	3923	10	0.058 ± 0.003	0.065 ± 0.012	0.074 ± 0.002
GSD-3, stream sediment	2200	920	0.093 ± 0.013	0.100 ± 0.020	0.242 ± 0.006
GSD-3a, stream sediment	566	96	0.493 ± 0.005	0.500 ± 0.060	0.587 ± 0.005
GSR-4, quartz sandstone	3567	13	0.055 ± 0.005	0.060 ± 0.016	0.065 ± 0.005

350

351

352

Table 6. Cd levels in four geological SRMs ($\mu\text{g g}^{-1}$)

Geological SRMs	This study (N=10) ^a	Ref. ^b	Literature values
USGS AGV-2, andesite	0.058 ± 0.004	(0.061)	0.121 ^c , 0.082 ^d ,
USGS BCR-2, basalt	0.148 ± 0.007	(0.14)	0.227 ^c , 0.75 ^d
IAG SdAR-M2, sediment	5.10 ± 0.20	5.10 ± 0.19	/
IAG GSM-1, gabbro	0.072 ± 0.005	0.128 ± 0.012	/

353 ^a Mean \pm SD, the SD is the standard deviation for ten separate aliquots of the samples
 354 analyzed over a period of three months

355 ^b Reference value by USGS and IAG

356 ^c Literature value with convention inductively coupled plasma quadrupole mass
 357 spectrometry method by Marx and Kamber ³⁷

358 ^d Literature value with convention inductively coupled plasma quadrupole mass
 359 spectrometry method by Hu and Gao ³⁸

360

361

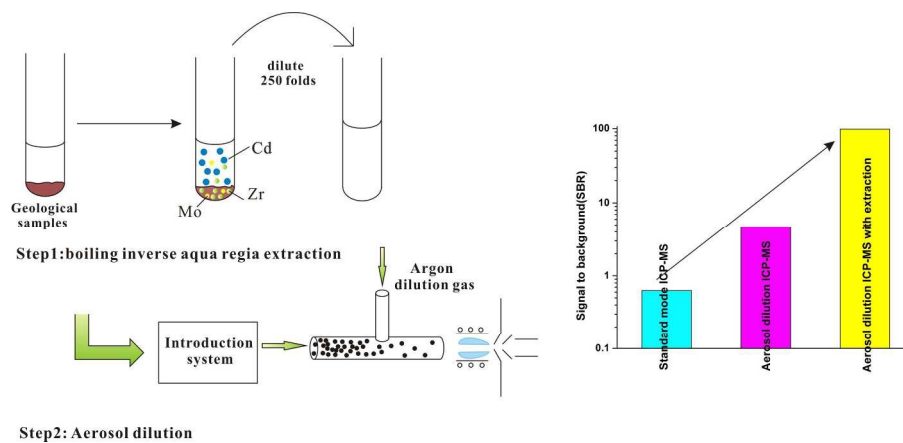
362 **Table 7.** Recommended Cd values for 12 geological SRMs determined in this work

363

 $(\mu\text{g g}^{-1})$

Geological SRMs	This work (N=10)	Reference values
GSD-7, stream sediment	1.20 ± 0.06	$1.05 \pm 0.06^{\text{a}}$
GSD-19, stream sediment	0.088 ± 0.007	$0.120 \pm 0.01^{\text{a}}$
GSR-22, limestone	0.121 ± 0.011	/ ^b
GSR-23, limestone	0.770 ± 0.040	/ ^b
GSR-24, limestone	0.063 ± 0.006	/ ^b
GSR-27, limestone	0.550 ± 0.050	/ ^b
GSR-28, limestone	0.510 ± 0.030	/ ^b
GUI-1, Limestone	0.042 ± 0.005	/ ^b
GUI-2, Limestone	0.160 ± 0.017	/ ^b
DIAN-1, Limestone	0.520 ± 0.030	/ ^b
DIAN-2, Limestone	0.366 ± 0.027	/ ^b
DIAN-3, Limestone	0.209 ± 0.015	/ ^b

364 ^a Recommend value365 ^b No reported value



Developed a valid method based on aerosol dilution ICP-MS after inverse aqua regia extraction for direct determination of trace levels of Cd in geological samples.
224x96mm (300 x 300 DPI)

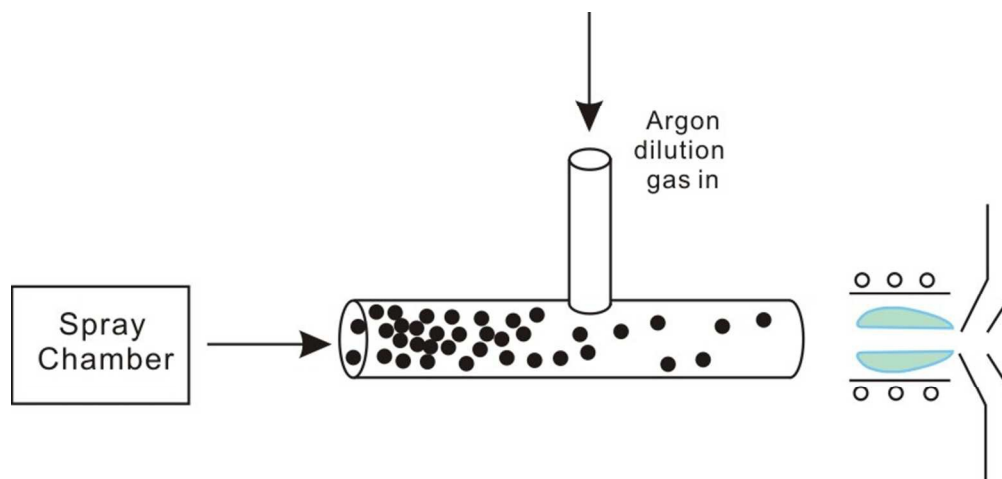


Fig. 1. Schematic diagram of our aerosol dilution system.
75x35mm (300 x 300 DPI)

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

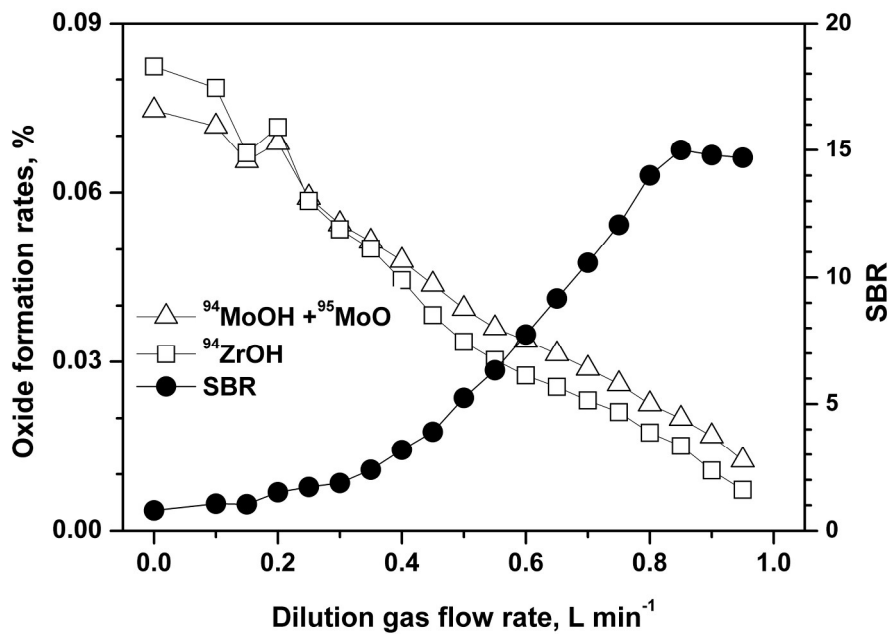


Fig. 2. Effects of dilution gas flow rate on oxide formation ratios ($(^{94}\text{Mo}16\text{OH} + ^{95}\text{Mo}16\text{O})/^{95}\text{Mo}+$ and $^{94}\text{Zr}16\text{OH}+/^{94}\text{Zr}+$) and the signal background ratio (SBR).
209x148mm (300 x 300 DPI)

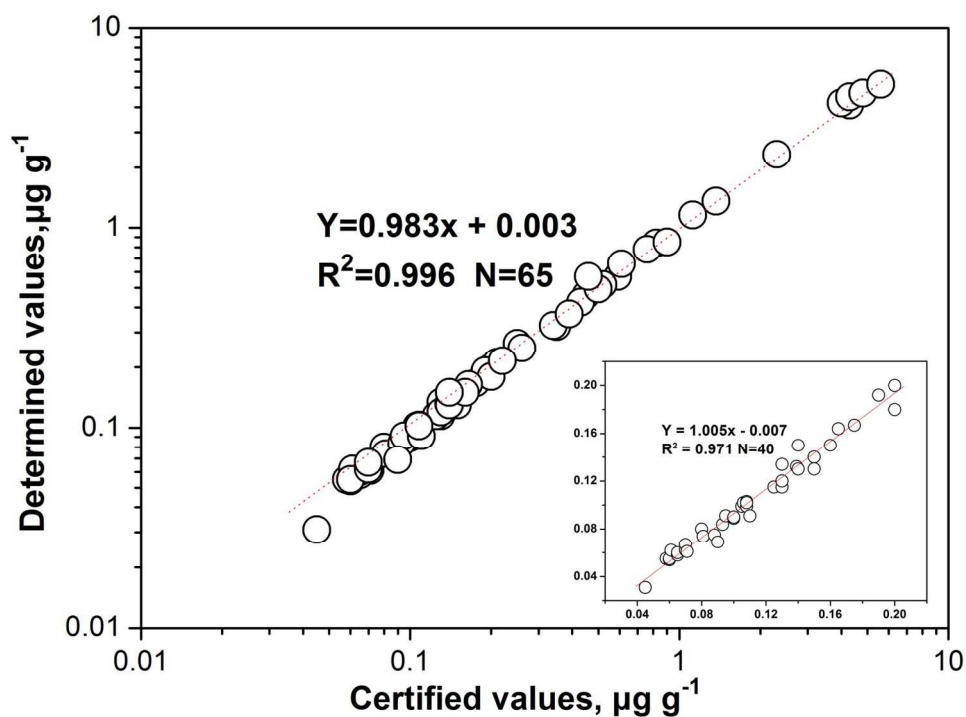


Fig. 3. Cd values determined for 65 geological SRMs by the proposed method vs. their certified values. The inset shows the 40 geological SRMs containing ultra-trace levels of Cd ($< 0.20 \mu\text{g g}^{-1}$).
156x118mm (300 x 300 DPI)

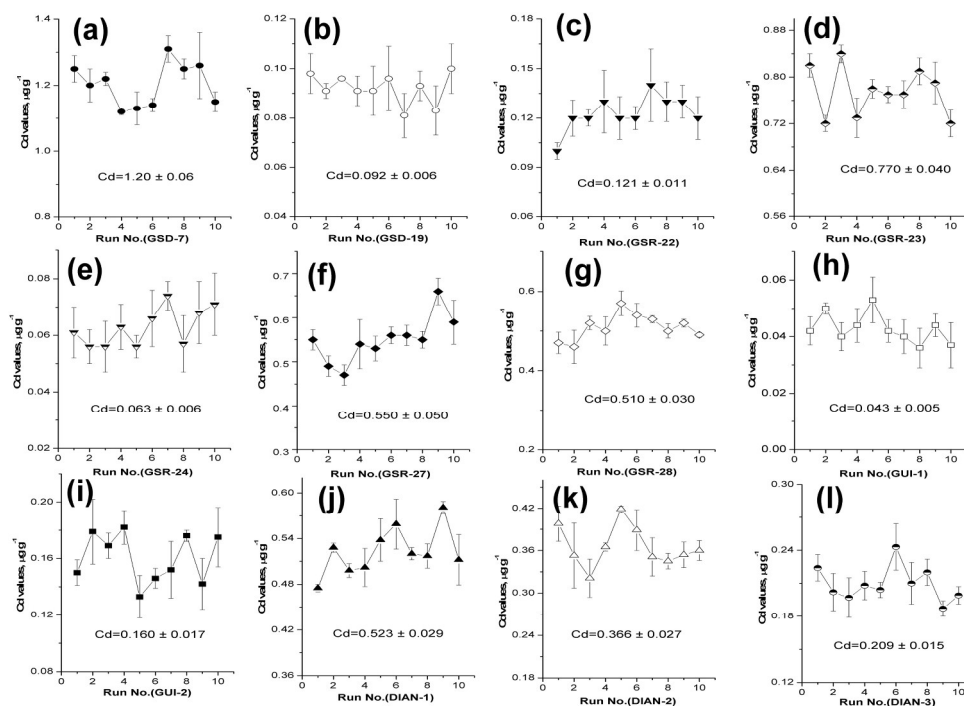


Fig. 4. Stability of Cd values obtained through the proposed method in 12 geological SRMs: two stream sediments: GSD-7 (a) and GSD-19 (b), and ten limestones SRMs: GSR-22 (c), GSR-23 (d), GSR-24 (e), GSR-27 (f), GSR-28 (g), GUI-1 (h), GUI-2 (i), DIAN-1 (j), DIAN-2 (k), and DIAN-3 (l). For each SRM, ten separate aliquots of the samples were analysed over a period of three months.
210x148mm (300 x 300 DPI)