Organic & Biomolecular Chemistry

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/obc

Chemistry

PAPER

Cite this: DOI: 10.1039/c1ob00000x

www.rsc.org/obc

Enantioselective Cascade Double Michael Addition of 3-Nitro-2*H*-Chromenes and Acyclic Enones: Efficient Synthesis of Functionalized Tricyclic Chroman Derivatives

Jun-Hua Li and Da-Ming Du*

s Received (in XXX, XXX) Xth XXXXXXXX 2015, Accepted Xth XXXXXXXX 2015 DOI: 10.1039/c1ob000000x

An efficient protocol for the asymmetric construction of enantiomerically enriched tetrahydro-6Hbenzo[c]chromenes, tetrahydro-6H-benzo[c]thiochromene or hexahydrophenanthrene and their derivatives have been developed. The corresponding products were obtained by the cascade double

¹⁰ Michael addition of 3-nitro-2*H*-chromenes and its derivatives with α , β -unsaturated ketones catalyzed by a combination of a quinine-derived primary amine and benzoic acid. Through this methodology, the desired products could be obtained in moderate to good yield (up to 90%), with excellent diastereoselectivities (up to >25:1 dr) and moderate to excellent enantioselectivities (up to 95% *ee*).

Introduction

- ¹⁵ Chroman derivatives as an important class of compounds, were widely found in natural compounds and synthetic analogues exhibiting various biological activities,¹ and great progress in the synthesis of those kinds of skeletons via an organocatalytic cascade reaction has been made in recent years.² Among various
- ²⁰ kinds of the chroman skeletons, the tricyclic compounds, tetrahydro-6*H*-benzo[*c*]chromene system occurs widely in a variety of natural products exhibiting various biological activities, such as Nabilone, (+)-Conicol, secretase inhibitors, the cannabinoid receptor CB1 and so on (Figure 1).³ At the same
- ²⁵ time its similar structure tetrahydro-6*H*-benzo[*c*]thiochromen or hexahydrophenanthrene constitutes the basic skeleton of many naturally occurring and synthetic compounds possessing a number of interesting biological properties (Figure 1).⁴
- In this regard, the tetrahydro-6*H*-benzo[*c*]chromene, ³⁰ tetrahydro-6*H*-benzo[*c*]thiochromene and hexahydrophenanthrene derivatives have received much attention in synthetic endeavors, and recently several elegant approaches have been established for the asymmetric construction of these highly valuable heterocyclic architectures (Scheme 1). Few years ³⁵ ago, Hong and co-workers successively reported an
- organocatalytic domino condensation of 2-((*E*)-2- nitrovinyl)

⁴⁵ CB1 full agonist estrogen receptor modulator SHIP1 modulator Figure 1. Representative examples of biologically active compounds containing tetrahydro-6*H*-benzo[*c*]chromenes or hexahydrophenanthrene core.

benzene-1,4-diol and α,β-unsaturated aldehydes to synthesis ⁵⁰ skeleton of hexahydro-6*H*-benzo[*c*]chromenes.⁵ Later, groups of Li and Wang also synthesis skeleton of tetrahydro-6*H*benzo[*c*]chromene derivatives through an organocatalytic cascade reaction of 2-hydroxychalcones⁶ or (*E*)-2-hydroxyaryl-2-oxobut-3-enoate⁷ derivatives with α,β-unsaturated aldehydes. In addition, ⁵⁵ Hong and co-workers also reported the asymmetric synthesis of

the tetrahydro-6*H*-benzo[*c*]chromen-6-ones⁸ and hexahydrophenanthrene⁹ derivatives through the organocatalytic cascade reactions. Of course, other groups also have done a lot of good works in the synthesis of these highly valuable heterocyclic ⁶⁰ architectures.¹⁰ Although excellent works have been done for the



mistry Accepted Manusc rganic & Biomole

School of Chemical Engineering and Environment, Beijing Institute of Technology, Beijing 100081, People's Republic of China.

⁴⁰ **Corresponding author, E-mail: <u>dudm@bit.edu.cn;</u> Tel: +86 10 68914985.*

Electronic Supplementary Information (ESI) available: [Copies of $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectra of new compounds, and HPLC chromatograms]. See DOI: 10.1039/b000000x/

This journal is © The Royal Society of Chemistry 2015

asymmetric construction of these valuable heterocyclic architectures, some improvements still need to be made. For example, the substrate scopes of the reactions were still limited, especially for the hexahydrophenanthrene. and as far as we know, the directly asymmetric synthesis of tetrahydro-6*H*-

s the directly asymmetric synthesis of tetrahydrobenzo[c]thiochromenes has not been reported yet.

3-Nitro-2H-chromenes, which can be easily prepared from salicylaldehyde and nitroethylene¹¹ are valuable intermediates for the synthesis of chroman derivatives, and recently many complex 10 compounds has been constructed by the 3-nitro-2H-chromenes and its derivatives.¹² Herein, we would like to present the first organocatalytic Michael/Michael asymmetric cascade cycloaddition reaction of 3-nitro-2*H*-chromenes and α , β unsaturated ketones to provide the tetrahydro-6H-15 benzo [c] chromenes in moderate to good yield, with excellent diastereoselectivities and moderate enantioselectivities. Moreover,

the tetrahydro-6H-benzo[c]thiochromene and hexahydrophenanthrene could also be obtained under this asymmetric catalytic system.



Scheme 1. Cascade organocatalytic reactions for the synthesis of tetrahydro-6*H*-benzo[*c*]chromene or hexahydrophenanthrene derivatives.

25 Results and discussion

In recently years, quinine or quinidine alkaloid-derived primary amine with an organic protonic acid was proved to be a powerful catalyst to activate unsaturated ketones to complete chiral induction.¹³ With this in mind, we firstly examine the cascade

- ³⁰ Michael/Michael reaction of 3-nitro-2*H*-chromene **1a** and α , β unsaturated ketones **2a**, using 20 mol% quinine-derived primary amine **I** as catalyst and 30 mol% benzoic acid **A1** as an additive (Table 1, entry 1). We were pleased that the reaction proceeded smoothly to provide the desired product **3a** in moderate yield ³⁵ (63%) with good enantioselectivity (73% *ee*) and excellent
- diastereoselectivity (96:4 dr). Then cinchonidine-derived primary amine **II** was also used as the catalyst, and the same enantiomer product was obtained with the same enantioselectivity and slightly lower yield and diastereoselectivity (Table 1, entry 2). In 40 consideration of the tiny gap of the result, quinine-derived

primary amine **I** was selected for further optimization. Then several commonly used solvent or mixed solvent were screened to further improve the reaction efficiency (Table 3, entries 3–8). Chloroform was found to be the most optimal solvent, affording 45 the desired product **3a** in 67% yield with 99:1 dr and 77% *ee*

- ⁴⁵ the desired product 3a in 07% yield with 99.1 dr and 77% ee (Table 1, entry 7). Meanwhile, when THF was used as the solvent, no product was observed after 96 h. Afterwards, except the benzoic acid, different benzoic acid derivatives were also used as the additives to explore the influence of protonic acid on the ⁵⁰ reactivity of this transformation, but no better results were obtained (Table 1, entry 10–13). To our surprise, when no additive was added into the reaction, the opposite enantiomer product was obtained, at the same time the yield and enantioselectivity were reduced dramatically (Table 1, entry 9). ⁵⁵ This result indicate that the additives play an important role in the
- cascade Michael/Michael reaction, and it was more like a kind of cocatalyst than just the role of protonic acid to adjust the pH. We also investigated the effect of temperature. When the temperature of the reaction was changed from room temperature to 40 °C, the 60 same yield was obtained, but the enantioselectivity and diastereoselectivity became lower (Table 1, entry 14). After select the benzoic acid A1 as an additive and chloroform as the solvent, the quinidine-derived primary amine III was used as the catalyst for this cascade reaction, and the opposite enantiomer was 65 obtained. Surprisingly, although the similar diastereoselectivity was obtained, both the yield and enantioselectivity were lower than its pseudoenantiomer, quinine-derived primary amine, as the catalyst (Table 1, entry 15). Moreover, other primary amines such as 1,2-diphenylethylenediamine (IV), 1,2-diaminocyclohexane 70 (V), and L-proline (VI) were also used as the catalysts, but no better results were obtained (Table 1, entries 16-18). Overall, the quinine-derived primary amine I as catalyst and 30 mol% A1 as an additive turn out to be the optimal catalytic system for the synthesis of tetrahydro-6H-benzo[c]chromene in chloroform at 75 room temperature.

Having established the optimal reaction conditions, the scope of the organocatalytic cascade Michael/Michael addition reaction was explored by varying the 3-nitro-2*H*-chromenes **1** and α , β -unsaturated ketones **2** as summarized in Table 2. The reactions proceeded smoothly to give the corresponding adducts in moderate to good yields with good to excellent diastereoselectivities and moderate to good enantioselectivities.



Figure 2. The screened catalysts and additives.

This journal is © The Royal Society of Chemistry 2015

^{2 |} Org. Biomol. Chem., 2015, 13, 00-00

Table 1 Optimization of the reaction conditions.^a

	NO ₂ + Br		catalyst (20 additive (30 solvent,	0 mol%) 0 mol%) rt			
1a	2a				3a		
Entry	Catalyst	Additive	Solvent	Yield ^b (%)	dr ^c	<i>ee^d</i> (%)	
1	Ι	A1	CH_2Cl_2	63	25:1	73	
2	Π	A1	CH_2Cl_2	62	20:1	73	
3	Ι	A1	PhMe	77	>25:1	71	
4	Ι	A1	xylene	73	20:1	69	
5	Ι	A1	CH ₃ CN	33	4:1	67	
6	Ι	A1	THF	trace	-	-	
7	Ι	A1	CHCl ₃	67	>25:1	77	
8	Ι	A1	CHCl ₃ / PhMe(1:1)	70	25:1	73	
9 ^f	Ι	-	CHCl ₃	28	20:1	10	
10	Ι	A2	CHCl ₃	61	>25:1	76	
11	Ι	A3	CHCl ₃	49	>25:1	67	
12	Ι	A4	CHCl ₃	69	>25:1	74	
13	Ι	A5	CHCl ₃	66	>25:1	76	
14^e	Ι	A1	CHCl ₃	67	20:1	74	
15 ^f	III	A1	CHCl ₃	58	>25:1	56	
16	IV	A1	CHCl ₃	39	>25:1	69	
17 ^f	\mathbf{V}	A1	CHCl ₃	49	25:1	49	
18	VI	A1	CHCl ₃	trace	-	-	

^{*a*} Unless noted otherwise, Reactions were carried out with **1a** (0.21 mmol), **2a** (0.20 mmol), catalyst loading (20 mol%) and additive (30 mol%) in s solvent (1.0 mL) at room temperature for 96 h. ^{*b*} Isolated yield after

column chromatography purification.^c Determined by NMR spectroscopy analysis ^d Determined by chiral HPLC analysis.^e The reaction was performed at 40 °C ^f The opposite enantiomer.

- ¹⁰ Firstly, different substituents of α,β -unsaturated ketones were explored. Even similar results were obtained, but there are still some slightly difference between electron-donating substituents and electron-withdrawing substituents. Products with electron-donating substituents on the *ortho*, *meta* or *para* position
- 15 of α,β-unsaturated ketones obtain the good enantioselectivities, which were better than products with other substituents (Table 3, entries 3–4, 8 and 10–11). Moreover, when the α,β-unsaturated ketones with nitro-group on the *para* position was used as the substrate, the lowest enantioselectivity was obtained (Table 3,
- ²⁰ entry 5). Although good enantioselectivity of product with methoxy group on the *ortho* position of α , β -unsaturated ketone was obtained, the diastereoselectivity of product **3h** was only 10:1 (Table 3, entry 8). At the same time, excellent diastereoselectivity (>25:1) was obtained by all of other products,
- ²⁵ which represent that the electron-donating substituent on the *ortho* position of α,β -unsaturated ketone could lower the diastereoselectivity of the product. As for the yields, the yield of product **31** (90%) was obvious better than other products, this might indicate that the reactivity of α , β -unsaturated ketones ³⁰ derived from fused ring was higher for this reaction.

Next, different substituents or atoms on the 3-nitro-2Hchromenes were explored. When the substrates with electronwithdrawing substituents on the R¹ group were used (Table 3, entries 13–16), excellent diastereoselectivities and moderate ³⁵ enantioselectivities were obtained, but the yield of products were obvious lower. On the other hand, When the substrates with electron-donating substituents on the R¹ group were used (Table 3, entries 17 and 18), the yield of the products were much better, which indicate that, just like the α , β -unsaturated ketones, the ⁴⁰ reactivity of 3-nitro-2*H*-chromenes with electron-donating substituents was also higher. To evaluate the synthetic potential of this reaction system, the oxygen atom of 3-nitro-2*H*-chromene was replaced by a methylene or a sulfur atom, which provide an easy acess to the hexahydrophenanthrene (Table 3, entry 19) or ⁴⁵ tetrahydro-6*H*-benzo[*c*]thiochromene (Table 3, entry 20). The desired products **3s** and **3t** can be obtained in moderate yields (64% and 74%) with high to excellent enantioselectivities (81% and 95% *ee*) and excellent diastereoselectivities (>25:1 dr).

During the exploration of the scope of this cascade ⁵⁰ Michale/Michael reaction, 2-nitro-3*H*-benzo[*f*]chromene (**1h**) was also used as the substrate to afford the tetrahydro-1*H*dibenzo[*c*,*f*]chromen-2-one (**3u**). Firstly, the reaction was taken under optimal reaction conditions and the quinine-derived primary amine **I** was used as the catalyst. To our surprise, very low enantioselectivity was obtained (26% *ee*). After a series of analysis, we believe that the fused ring structure of substrate **1h** have a special steric effect between the catalyst **I** and additive **A1**. So the opposite enantiomer of the catalyst **I**, quinidine-derived primary amine **III**, was used as the catalyst to provide a different ⁶⁰ steric effect for the reaction, and the enantioselectivity of enantiomer product *ent*-**3u** was enhanced to 78% *ee*.



65 Scheme 2. The reaction of 2-nitro-3*H*-benzo[*f*]chromene with 4-(4-chlorophenyl)but-3-en-2-one under different catalysts.

To determine the absolute configuration of the cascade reaction products, structure of **3m** was assigned unambiguously by single-crystal X-ray analysis to be (6aR,7S,10aS),¹⁴ and the ⁷⁰ ORTEP structure is shown in Figure 1.



Figure 2. X-ray crystal structure of 3m.

PAPER

Organic & Biomolecular Chemistry

Cite this: DOI: 10.1039/c1ob00000x

www.rsc.org/obc

 Table 2 Substrate scope of the asymmetric cascade Michael/Michael addition^a

	$R^{1} \xrightarrow{I_{1}} X \xrightarrow{NO_{2}} + R^{2} \xrightarrow{O} \xrightarrow{I (20 \text{ mol}\%)} R^{1} \xrightarrow{I_{1} (20 \text{ mol}\%)} R^{1} \xrightarrow{H} R^{2}$										
		1a-i	2a-l		3a-t						
Entry	\mathbf{R}^1	Х	\mathbb{R}^2	Product	Yield ^b (%)	$\mathrm{dr}^{c}(\%)$	ee^d (%)				
1	Н (1а)	0	$4-BrC_{6}H_{4}(2a)$	3a	67	>25:1	77				
2	H (1a)	О	$4-ClC_6H_4(\mathbf{2b})$	3b	67	>25:1	77				
3	H (1a)	О	$4-MeC_{6}H_{4}(2c)$	3c	57	>25:1	79				
4	H (1a)	О	$4\text{-}MeOC_6H_4(2\mathbf{d})$	3d	75	>25:1	81				
5	H (1a)	0	$4-O_2NC_6H_4(2e)$	3e	66	>25:1	74				
6	H (1a)	О	Ph (2f)	3f	61	>25:1	78				
7	H (1a)	0	$2\text{-BrC}_{6}\text{H}_{4}(2\mathbf{g})$	3g	85	>25:1	75				
8	H (1a)	О	$2\text{-MeOC}_6\text{H}_4(2\mathbf{h})$	3h	78	10:1	80				
9	H (1a)	О	$3-O_2NC_6H_4(2i)$	3i	64	>25:1	77				
10	H (1a)	0	$3,4-(MeO)_2C_6H_3(2j)$	3ј	69	>25:1	83				
11	H (1a)	0	$4-(CH_3)_2NC_6H_4(2\mathbf{k})$	3k	78	>25:1	83				
12	H (1a)	0	1-naphthyl (21)	31	90	>25:1	79				
13	6-Br (1b)	0	$4-ClC_6H_4(\mathbf{2b})$	3m	53	>25:1	78				
14	6-Cl (1c)	0	$4-ClC_6H_4(\mathbf{2b})$	3n	59	>25:1	73				
15	6,8-Br ₂ (1d)	О	$4-ClC_6H_4(\mathbf{2b})$	30	45	>25:1	65				
16	6,8-Cl ₂ (1e)	0	$4\text{-}\mathrm{ClC}_6\mathrm{H}_4(\mathbf{2b})$	3p	54	>25:1	68				
17	8-MeO (1f)	О	$4\text{-}\mathrm{ClC}_6\mathrm{H}_4(\mathbf{2b})$	3q	80	>25:1	73				
18	8-EtO (1g)	О	$4\text{-}\mathrm{ClC}_6\mathrm{H}_4(\mathbf{2b})$	3r	81	>25:1	71				
19	H (1h)	CH_2	$4\text{-}\mathrm{Br}\mathrm{C}_{6}\mathrm{H}_{4}(\mathbf{2a})$	3 s	64	>25:1	81				
20	H (1i)	S	$4\text{-}\mathrm{Br}\mathrm{C}_{6}\mathrm{H}_{4}(\mathbf{2a})$	3t	74	>25:1	95				

^{*a*} Unless noted otherwise, reactions were carried out with **1** (0.21 mmol), **2** (0.20 mmol) catalyst **I** (20 mol%) and additive (30 mol%) in CH₃Cl (1.0 mL) at ⁵ room temperature for 96 h. ^{*b*} Isolated yield after column chromatography purification. ^{*c*} Determined by NMR spectroscopy analysis ^{*d*} Determined by chiral HPLC analysis.

Page 5 of Organic & Biomolecular Chemistry

PAPER

Cite this: DOI: 10.1039/c1ob00000x

www.rsc.org/obc



Inspired by the previous study of this catalytic system,^{13a,13b,13f} ⁵ a possible transition state model is proposed and shown in Figure 3. The activated dienamine ion intermediate **A** was first generated by the intercept of the primary amine catalyst to unsaturated ketone **2**. Then an intermolecular Michael addition was initiated between the prochiral carbon nucleophile **A** and 3-nitro-2*H*-

¹⁰ chromene **1**. The resulting intermediate **B** would further selectively engage itself in an intramolecular, iminium-catalyzed conjugate addition. Thus the tricyclic chroman derivatives were obtained through the cascade double Michael addition.

15 Conclusions

In summary, we have successfully developed an efficient protocol for the construction of enantiomerically enriched tetrahydro-6H-benzo[c]chromenes, tetrahydro-6H-benzo[c]thiochromene and hexahydrophenanthrene via the

- ²⁰ asymmetric organocatalytic cascade Michael/Michael reaction of 3-nitro-2*H*-chromenes and α , β -unsaturated ketones, which is catalyzed by a combination of a quinine-derived primary amine and benzoic acid. Through this new methodology, the corresponding chiral tricyclic chroman derivatives could be
- ²⁵ obtained in moderate to good yield (up to 90%) with excellent diastereoselectivities (up to >25:1 dr) and moderate to excellent enantioselectivities (up to 95% *ee*). The structure was confirmed by single crystal X-ray analysis of cascade adduct **3m**. Further studies on organocatalytic enantioselective synthesis of ³⁰ pharmaceutical intermediates or biologically active compounds

are ongoing in our laboratory.

Experimental

General information

Commercially available compounds were used without further ³⁵ purification. Column chromatography was carried out using silica gel (200–300 mesh). Melting points were measured with a XT-4 melting point apparatus without correction. The ¹H NMR spectra were recorded with a Bruker Avance 400 MHz spectrometer, while ¹³C NMR spectra were recorded at 100 MHz. Infrared

- ⁴⁰ spectra were obtained with a Nicolet Magna IR-560 spectrometer. The high resolution ESI-MS spectra were obtained with Bruker APEX IV Fourier transform mass spectrometer. Optical rotations were measured with a Krüss P8000 polarimeter at the indicated concentration with unit g per 100 mL. The enantiomeric excesses
- ⁴⁵ of the products were determined by chiral HPLC using Agilent 1200 LC instrument on Daicel Chiralpak AD-H or IB columns.

Materials

The substrates $1^{11c,15}$ and 2^{13g} were synthesized following the reported procedures.

50 General procedure for the enantioselective Michael addition reaction

A mixture of 3-nitro-2*H*-chromene **1** (0.21 mmol), α , β unsaturated ketone **2** (0.20 mmol), catalyst **I** (13.0 mg, 0.04 mmol, 20 mol%) and benzoic acid (7.4 mg, 0.06 mmol, 30 mol%) in 55 chloroform (1.0 mL) was stirred at room temperature for 96 h. Then the mixture was concentrated and purified by silica gel column chromatography (with ethyl acetate-petroleum ether as the eluent) to afford the desired products **3**.

60 (6a*R*,7*S*,10a*S*)-7-(4-bromophenyl)-6a-nitro-7,8,10,10a-

tetrahydro-6*H***-benzo[***c***]chromen-9(6***aH*)-**one** (**3a**). The product **3a** was obtained according to the general procedure as a white solid (54.2 mg, 67% yield), m.p. 66–71 °C. HPLC (Daicel Chiralpak AD-H column, *n*-hexane-2-propanol 80:20, flow rate 65 1.0 mL/min, detection at 254 nm): $t_{major} = 16.4$ min, $t_{minor} = 15.2$ min, 77% *ee*; $[\alpha]_D^{20} = -4.2$ (*c* 2.3, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃): δ 7.51–7.48 (m, 2H, ArH), 7.23–7.18 (m, 2H, ArH), 7.09–7.06 (m, 2H, ArH), 7.01 (dt, $J_1 = 1.2$ Hz, $J_2 = 7.6$ Hz, 1H, ArH), 6.93 (d, J = 8.0 Hz, 1H, ArH), 4.58 (d, J = 11.6 Hz, 1H, 70 CH), 4.30–4.23 (m, 2H, CH + CH), 3.54 (dd, $J_1 = 4.6$ Hz, $J_2 =$

- 13.0 Hz, 1H, CH), 3.41–3.30 (m, 2H, CH₂), 2.88 (dd, J_1 = 8.2 Hz, J_2 = 16.6 Hz, 1H, CH), 2.54 (dd, J_1 = 4.4 Hz, J_2 = 16.0 Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 205.8, 152.2, 134.4, 132.2, 129.9, 129.1, 128.4, 123.0, 122.6, 120.2, 117.5, 87.2, 65.2,
- $_{75}$ 43.4, 42.8, 41.4, 39.1 ppm; IR (KBr): $\tilde{\nu}~1717,~1585,~1543,~1488,~1456,~1409,~1340,~1276,~1259,~1228,~1075,~837,~751,~506~cm^{-1};~HRMS~(ESI):~m/z~calcd.~for~C_{19}H_{16}BrNNaO_4~[M~+~Na]^+~424.01549,~found~424.01628.$

This journal is $\ensuremath{\mathbb{C}}$ The Royal Society of Chemistry 2015

(6aR,7S,10aS)-7-(4-chlorophenyl)-6a-nitro-7,8,10,10atetrahydro-6H-benzo[c]chromen-9(6aH)-one (3b). The product 3b was obtained according to the general procedure as a white solid (47.6 mg, 67% yield), m.p. 170–173 °C. HPLC (Daicel

- ⁵ Chiralpak AD-H column, hexane–2-propanol 80:20, flow rate 1.0 mL/min, detection at 254 nm): $t_{\text{major}} = 15.2 \text{ min}$, $t_{\text{minor}} = 13.2 \text{ min}$, 77% *ee*; $[\alpha]_{\text{D}}^{20} = -1.4$ (*c* 1.9, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃): δ 7.34 (d, *J* = 8.0 Hz, 2H, ArH), 7.23–7.19 (m, 2H, ArH), 7.14 (d, *J* = 8.4 Hz, 2H, ArH), 7.02 (t, *J* = 7.6 Hz, 1H, ArH), 6.93
- ¹⁰ (d, J = 8.0 Hz, 1H, ArH), 4.58 (d, J = 11.6 Hz, 1H, CH), 4.30–4.24 (m, 2H, CH+CH),3.55 (dd, $J_1 = 4.4$ Hz, $J_2 = 12.8$ Hz, 1H, CH), 3.41–3.01 (m, 2H, CH₂) 2.89 (dd, $J_1 = 8.2$ Hz, $J_2 = 16.6$ Hz, 1H, CH), 2.54 (dd, $J_1 = 4.4$ Hz, $J_2 = 15.6$ Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 205.9, 152.2, 134.8, 133.9,
- ¹⁵ 129.5, 129.3, 129.0, 128.4, 122.5, 120.2, 117.5, 87.3, 65.2, 43.4, 42.7, 41.5, 39.0 ppm; IR (KBr): $\bar{\nu}$ 1724, 1689, 1546, 1489, 1456, 1280, 1249, 1229, 1125, 1093, 1068, 1014, 839, 811, 7753, 707, 666, 541, 514 cm⁻¹; HRMS (ESI): *m/z* calcd. for C₁₉H₁₆CINNaO₄ [M + Na]⁺ 380.06601, found 380.06605.
- 20

(6aR,7S,10aS)-6a-nitro-7-(p-tolyl)-7,8,10,10a-tetrahydro-6H-benzo[c]chromen-9(6aH)-one (3c). The product 3c was obtained according to the general procedure as a white solid (38.6 mg, 57% yield), m.p. 63-68 °C. HPLC (Daicel Chiralpak AD-H 25 column, n-hexane-2-propanol 80:20, flow rate 1.0 mL/min, detection at 254 nm): $t_{\text{major}} = 12.1 \text{ min}, t_{\text{minor}} = 10.7 \text{ min}, 79\% ee;$ $[\alpha]_{D}^{20} = +7.6$ (c 1.6, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃): δ 7.21–7.15 (m, 4H, ArH), 7.06 (d, J = 8.0 Hz, 2H, ArH) 7.00 (dt, J₁ = 1.2 Hz, J₂ = 7.6 Hz, 1H, ArH), 6.92 (d, J = 8.0 Hz, ArH, 1H) ³⁰ 4.66 (dd, $J_1 = 1.0$ H, $J_2 = 11.8$ Hz, 1H, CH), 4.31–4.25 (m, 2H, CH+CH), 3.50 (dd, J_1 = 4.0 Hz, J_2 = 12.8 Hz, 1H, CH), 3.39 (dd, $J_1 = 12.8$ H, $J_2 = 15.6$ Hz, 1H, CH), 3.31 (dd, $J_1 = 5.4$ H, $J_2 =$ 16.6 Hz, 1H, CH), 2.84 (dd, J₁ = 8.8 Hz, J₂ = 16.6 Hz, 1H, CH), 2.56 (dd, $J_1 = 4.0$ Hz, $J_2 = 15.6$ Hz, 1H, CH), 2.34 (s, 3H, CH₃) ³⁵ ppm; ¹³C NMR (100 MHz, CDCl₃): δ 206.5, 152.4, 138.7, 132.4, 129.8, 128.9, 128.5, 128.0, 122.4, 120.4, 117.5, 87.8, 68.2, 43.9, 43.3, 41.7, 38.5, 21.1 ppm; IR (KBr): v 2313, 1715, 1585, 1544, 1514, 1489, 1456, 12776, 1228, 1218, 1070, 819, 751, 503 cm⁻¹; HRMS (ESI): m/z calcd. for C₂₀H₂₀NO₄ [M + H]⁺ 338.13868, 40 found 338.13912.

(6aR,7S,10aS)-7-(4-methoxyphenyl)-6a-nitro-7,8,10,10atetrahydro-6H-benzo[c]chromen-9(6aH)-one (3d). The product

- **3d** was obtained according to the general procedure as a white solid (53.3 mg, 75% yield), m.p. 53–59 °C. HPLC (Daicel Chiralpak AD-H column, *n*-hexane–2-propanol 80:20, flow rate 1.0 mL/min, detection at 254 nm): $t_{major} = 18.0$ min, $t_{minor} = 16.4$ min, 81% *ee*; $[\alpha]_D^{20} = +7.9$ (*c* 1.5, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃): δ 7.22–7.17 (m, 2H, ArH), 7.12–7.08 (m, 2H, ArH), 7.00
- ⁵⁰ (td, $J_1 = 1.2$ Hz, $J_2 = 7.6$ Hz, 1H, ArH), 6.93–6.86 (m, 3H,ArH), 4.64 (dd, $J_1 = 0.8$ Hz, $J_2 = 12.0$ Hz, 1H, CH), 4.31–4.24 (m, 2H, CH+CH), 3.79 (s, 3H, CH₃), 3.49 (dd, $J_1 = 4.0$ Hz, $J_2 = 12.8$ Hz, 1H, CH), 3.37 (dd, $J_1 = 12.8$ H, $J_2 = 15.8$ Hz, 1H, CH), 3.31 (dd, $J_1 = 5.6$ H, $J_2 = 16.4$ Hz, 1H, CH), 2.85 (dd, $J_1 = 9.0$ Hz, $J_2 = 16.4$
- ⁵⁵ Hz, 1H, CH), 2.56 (dd, J_1 = 4.0 Hz, J_2 = 15.8 Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, CDCl₃): δ (100 MHz, CDCl₃): δ 206.5, 159.7, 152.4, 129.2, 128.9, 128.5, 127.3, 122.4, 120.4, 117.4, 114.4, 87.8, 65.2, 55.2, 43.8, 42.9, 41.8, 38.5 ppm; IR (KBr): \tilde{v} 1716, 1610, 1543, 1513, 1489, 1457, 1251, 1228, 1218, 1181, ⁶⁰ 1069, 1028, 835, 752, 510 cm⁻¹; HRMS (ESI): *m/z* calcd. for
- $C_{20}H_{20}NO_5 [M + H]^+$ 354.13360, found 354.13367; calcd. for $C_{20}H_{19}NNaO_5 [M + Na]^+$ 376.11554, found 376.11514.

(6aR,7S,10aS)-6a-nitro-7-(4-nitrophenyl)-7,8,10,10a-

- ⁶⁵ tetrahydro-6*H*-benzo[*c*]chromen-9(6*aH*)-one (3e). The product 3e was obtained according to the general procedure as a white solid (48.7 mg, 66% yield), m.p. 206–216 °C. HPLC (Daicel Chiralpak AD-H column, *n*-hexane–2-propanol 80:20, flow rate 1.0 mL/min, detection at 254 nm): *t*_{major} = 37.5 min, *t*_{minor} = 31.0 ⁷⁰ min, 74% *ee*; [α]_D²⁰ = -11.9 (*c* 1.4, CH₂Cl₂); ¹H NMR (400 MHz, acetone-*d*₆): δ 8.25 (d, *J* = 8.8 Hz, 2H, ArH), 7.58 (d, *J* = 8.4 Hz, 2H, ArH), 7.38 (d, *J* = 7.6 Hz, 1H, ArH), 7.21–7.17 (m, 1H, ArH),
- 7.01–6.98 (m, 1H, ArH), 6.86 (d, J = 8.0 Hz, 1H, ArH), 5.06 (dd, $J_1 = 2.4$ Hz, $J_2 = 12.8$ Hz, 1H, CH), 4.73–4.66 (m, 2H, CH+CH), 75 4.07 (dd, $J_1 = 4.4$ Hz, $J_2 = 11.2$ Hz, 1H, CH), 3.35 (dd, $J_1 = 11.2$
- Hz, $J_2 = 16.4$ Hz, 1H, CH), 3.12 (dd, $J_1 = 5.2$ Hz, $J_2 = 17.2$ Hz, 1H, CH), 2.96 (dd, $J_1 = 12.8$ Hz, $J_2 = 17.2$ Hz, 1H, CH), 2.73 (dd, $J_1 = 4.4$ Hz, $J_2 = 16.4$ Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, acetone- d_6): δ 206.8, 154.3, 149.8, 146.3, 131.7, 131.2, 130.4, 125.5 (dd), $J_2 = 16.2$ Hz, $J_2 = 16.4$ Hz, 149.8, 146.3, 131.7, 131.2, 130.4,
- 80 125.5, 123.8, 123.1, 118.7, 92.0, 66.7, 46.8, 45.4, 42.7, 37.9 ppm; IR (KBr): $\tilde{\nu}$ 1731, 1714, 1681, 1546, 1519, 1493, 1351, 1275, 1073, 857, 851, 765, 750, 695 cm^{-1}; HRMS (ESI): *m/z* calcd. For $C_{19}H_{16}N_2NaO_6\left[M + Na\right]^+$ 391.09006, found 391.08874.
- (6aR,7S,10aS)-6a-nitro-7-phenyl-7,8,10,10a-tetrahydro-6H-85 benzo[c]chromen-9(6aH)-one (3f). The product 3f was obtained according to the general procedure as a white solid (39.5 mg, 61% yield), m.p. 166-170 °C. (HPLC on Daicel Chiralpak AD-H column, n-hexane-2-propanol 80:20, flow rate 1.0 mL/min, 90 detection at 254 nm): $t_{\text{major}} = 15.1 \text{ min}, t_{\text{minor}} = 12.0 \text{ min}, 78\% ee.$ $[\alpha]_{D}^{20} = +12.2 \ (c \ 1.7, \ CH_{2}Cl_{2}); \ ^{1}H \ NMR \ (400 \ MHz, \ CDCl_{3}): \delta$ 7.39–7.33 (m, 3H, ArH), 7.22–7.17 (m, 4H, ArH), 7.00 (t, J = 7.6 Hz, 1H, ArH), 6.92 (d, J = 8.0 Hz, 1H, ArH), 4.66 (d, J = 11.6 Hz, 1H, ArH), 4.31–4.26 (m, 2H, CH+CH), 3.54 (dd, $J_1 = 4.0$ Hz, J_2 $_{95}$ = 12.8 Hz, 1H, CH), 3.41 (dd, J_1 = 12.8 Hz, J_2 = 15.8 Hz, 1H, CH), 3.31 (dd, J₁ = 5.6 Hz, J₂ = 16.4 Hz, 1H, CH), 2.85 (dd, J₁ = 9.0 Hz, $J_2 = 16.4$ Hz, 1H, CH), 2.57 (dd, $J_1 = 4.0$ Hz, $J_2 = 15.8$ Hz, 1H, CH) ppm; 13 C NMR (100 MHz, CDCl₃): δ 206.4, 152.3, 135.4, 129.1, 128.9, 128.8, 128.5, 128.1, 122.4, 120.3, 117.4, 100 87.7, 65.2, 43.8, 43.5, 41.5, 38.5 ppm; IR (KBr): v 1726, 1690, 1681, 1547, 1489, 1455, 1422, 1341, 1278, 1249, 1227, 1217, 1065, 757, 702, 542, 515 cm⁻¹; HRMS (ESI): *m/z* calcd. for $C_{19}H_{17}NNaO_4[M + H]^+$ 346.10498, found 346.10497.

(6aS,7S,10aS)-7-(2-bromophenyl)-6a-nitro-7,8,10,10a-105 tetrahydro-6H-benzo[c]chromen-9(6aH)-one (3g). The product 3g was obtained according to the general procedure as white solid (68 mg, 85% yield), m.p. 116-122 °C. HPLC (Daicel Chiralpak AD-H column, n-hexane-2-propanol 80:20, flow rate 1.0 mL/min, 110 detection at 254 nm): $t_{major} = 21.5 \text{ min}, t_{minor} = 12.1 \text{ min}, 75\% ee;$ $[\alpha]_{D}^{20} = +93.1$ (c 2.9, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃): δ 7.65 (dd, $J_1 = 1.6$ Hz, $J_2 = 8.0$ Hz, 1H, ArH), 7.32 (td, $J_1 = 1.2$ Hz, $J_2 = 7.6$ Hz, 1H, ArH), 7.24–7.16 (m, 3H, ArH), 7.07 (dd, $J_1 = 1.6$ Hz, J₂ = 7.6 Hz, 1H, ArH), 7.01–6.97 (m, 1H, ArH), 6.89–6.87 115 (m, 1H, ArH), 5.06 (dd, $J_1 = 2.0$ Hz, $J_2 = 12.8$ Hz, 1H, CH), 4.51 $(dd, J_1 = 2.6 Hz, J_2 = 13.6 Hz, 1 H, CH), 4.20 (d, J = 12.4 Hz, 1H)$ CH), 4.06 (dd, $J_1 = 3.2$ Hz, $J_2 = 14.0$ Hz, 1H, CH), 3.35 (dd, $J_1 =$ 14.0 Hz, $J_2 = 17.2$ Hz, 1H, CH), 3.12 (dd, $J_1 = 4.4$ Hz, $J_2 = 17.6$ Hz, 1H, CH), 2.78 (dd, J₁ = 13.6 Hz, J₂ = 17.6 Hz, 1H, CH), 2.50 $_{120}$ (dd, $J_1 = 3.2$ Hz, $J_2 = 17.2$ Hz, 1H, CH) ppm; 13 C NMR (100 MHz, CDCl₃): δ 206.3, 152.7, 135.1, 133.5, 130.3, 128.9, 128.8, 128.7, 127.7, 125.2, 122.2, 119.5, 117.5, 91.0, 65.2, 44.8, 41.7, 41.4, 36.3 ppm; IR (KBr): v 1723, 1586, 1543, 1489, 1342, 1247, 1215, 1067, 1023, 908, 756, 725, 671, 608, 543, 508, 463 cm^{-1} ; 125 HRMS (ESI): m/z calcd. for $C_{19}H_{16}BrNNaO_4 [M + Na]^+$ 424.01549, found 424.01482.

6 | Org. Biomol. Chem., 2015, **13**, 00–00

This journal is © The Royal Society of Chemistry 2015

(6a*R*,7*S*,10a*S*)-7-(2-methoxyphenyl)-6a-nitro-7,8,10,10atetrahydro-6*H*-benzo[*c*]chromen-9(6a*H*)-one (3h). The product 3h was obtained according to the general procedure as a white solid (55.3 mg, 78% yield), m.p. 149–152 °C. HPLC (Daicel Chiralpak AD-H column, *n*-hexane–2-propanol 80:20, flow rate 1.0 mL/min, detection at 254 nm): $t_{major} = 24.8$ min, $t_{minor} = 12.1$ min, 80% *ee*; $[\alpha]_D^{20} = +95.4$ (*c* 1.8, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃): δ 7.33–7.28 (m, 1H, ArH), 7.17–7.12 (m, 2H, ArH), 10 7.00–6.90 (m, 4H, ArH), 6.85 (d, *J* = 8.4 Hz, 1H, ArH), 5.03 (d, *J* = 12.4 Hz, 1H, CH), 4.47 (dd, $J_1 = 2.4$ Hz, $J_2 = 13.6$ Hz, 1H, CH), 4.14 (d, *J* = 10.4 Hz, 1H, CH), 3.91 (d, *J* = 11.6 Hz, 1H, CH), 3.85 (s, 3H, OCH₃), 3.41 (t, *J* = 15.2 Hz, 1H, CH), 3.06 (dd, $J_1 =$ 4.2 Hz, $J_2 = 17.6$ Hz, 1H, CH), 2.69 (dd, $J_1 = 13.6$ Hz, $J_2 = 17.6$

- ¹⁵ Hz, 1H, CH), 2.46 (dd, $J_1 = 2.8$ Hz, $J_2 = 17.2$ Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 205.5, 157.6, 156.5, 152.6, 129.8, 128.9, 128.7, 123.9, 121.9, 121.5, 117.3, 111.2, 110.6, 90.8, 77.2, 65.0, 55.2, 45.3, 41.0, 36.1 ppm; IR (KBr): \tilde{v} 1724, 1539, 1491, 1459, 1250, 1234, 1214, 1064, 1026, 753, 727, 488 cm⁻¹; HRMS ²⁰ (ESI): *m/z* calcd. for C₂₀H₂₀NO₅ [M + H]⁺ 354.13360, found
- 354.13361.

(6aR,7S,10aS)-6a-nitro-7-(3-nitrophenyl)-7,8,10,10a-

- **tetrahydro-6***H***-benzo**[*c*]**chromen-9(6***aH***)-one (3i**). The product ²⁵ 3i was obtained according to the general procedure as a white solid (47.4 mg, 64% yield), m. p. 72–76 °C. (HPLC on Daicel Chiralpak AD-H column, *n*-hexane–2-propanol 80:20, flow rate 1.0 mL/min, detection at 254nm): $t_{major} = 25.8$ min, $t_{minor} = 21.4$ min, 77% *ee*; $[\alpha]_D^{20} = +6.1$ (*c* 1.8, CH₂Cl₂); ¹H NMR (400 MHz, ³⁰ acetone-*d*₆): δ 8.25 (d, *J* = 7.6 Hz, 1H, ArH), 8.14 (s, 1H, ArH), 7.75–7.68 (m, 2H, ArH), 7.38 (d, *J* = 7.6 Hz, 1H, ArH), 7.18 (t, *J* = 7.6 Hz, 1H, ArH), 6.99 (t, *J* = 7.6 Hz, 1H, ArH), 6.86 (d, *J* = 8.4 Hz, 1H, ArH), 5.06 (d, *J* = 12.8 Hz, 1H, CH), 4.72–4.65 (m, 2H, CH+CH), 4.09 (dd, *J*₁ = 4.0 Hz, *J*₂ = 11.4 Hz, 1H, CH), 3.40 (dd, ³⁵ *J*₁ = 11.4 Hz, *J*₂ = 16.4 Hz, 1H, CH), 3.13 (dd, *J*₁ = 4.8 Hz, *J*₂ = 17.2 Hz, 1H, CH), 2.97 (dd, *J*₁ = 13.0 Hz, *J*₂ = 17.2 Hz, 1H, CH),
- 2.74 (dd, $J_1 = 4.0$ Hz, $J_2 = 16.4$ Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, acetone- d_6): δ 206.9, 154.3, 150.2, 141.0, 136.5, 132.1, 131.2, 130.4, 125.2, 125.1, 123.8, 123.0, 118.7, 92.0, 66.6, 46.7, 40 45.2, 42.8, 37.9 ppm; IR (KBr): \tilde{v} 1718, 1528, 1489, 1457, 1347,
- 1250, 1220, 1069, 810, 758, 730, 687 cm⁻¹; HRMS (ESI): m/z calcd. for C₁₉H₁₆N₂NaO₆ [M + Na]⁺ 391.09006, found 391.08919.

(6aR,7S,10aS)-7-(3,4-dimethoxyphenyl)-6a-nitro-

45 7,8,10,10a-tetrahydro-6*H*-benzo[*c*]chromen-9(6a*H*)-one (**3i**). The product **3j** was obtained according to the general procedure as a white solid (53.2 mg, 69% yield), m. p. 78-83 °C. HPLC (Daicel Chiralpak IB column, n-hexane-2-propanol 65:35, flow rate 1.0 mL/min, detection at 254 nm): $t_{major} = 36.6 \text{ min}, t_{minor} =$ $_{50}$ 21.0 min, 83% ee; $[\alpha]_{D}^{20} = +8.3$ (c 1.7, CH₂Cl₂); ¹H NMR (400 MHz, acetone- d_6): δ 7.35 (d, J = 7.6 Hz, 1H, ArH), 7.16 (t, J =7.6 Hz, 1H, ArH), 6.97 (t, J = 7.4 Hz, 1H, ArH), 6.90 (d, J = 8.0 Hz, 1H, ArH), 6.83 (d, J = 8.0 Hz, 1H, ArH), 6.79–6.75 (m, 2H, ArH), 5.03 (dd, J₁ = 1.8 Hz, J₂ = 12.8 Hz, 1H, CH), 4.63 (d, J = 55 12.4 Hz, 1H, CH), 4.56 (dd, $J_1 = 4.0$ Hz, $J_2 = 12.8$ Hz, 1H, CH), 3.79 (s, 6H, OCH₃+OCH₃), 3.74 (dd, $J_1 = 4.4$ Hz, $J_2 = 10.8$ Hz, 1H, CH), 3.25 (dd, $J_1 = 10.8$ Hz, $J_2 = 16.4$ Hz, 1H, CH), 3.07 (dd, $J_1 = 5.2$ Hz, $J_2 = 17.2$ Hz, 1H, CH), 2.89 (dd, $J_1 = 12.8$ Hz, $J_2 =$ 16.8 Hz, 1H, CH), 2.64 (dd, $J_1 = 4.4$ Hz, $J_2 = 16.4$ Hz, 1H, CH) ⁶⁰ ppm; ¹³C NMR (100 MHz, acetone- d_6): δ 207.8, 154.4, 151.4, 151.2, 131.2, 131.0, 130.3, 123.6, 123.5, 122.4, 118.7, 114.0, 113.6, 91.9, 66.9, 57.1, 57.0, 47.0, 45.7, 43.6, 37.9 ppm; IR (KBr): \tilde{v} 1720, 1587, 1544, 1518, 1491, 1460, 1347, 1246, 1147, 1072, 1023, 811, 759, 640, 476 cm⁻¹; HRMS (ESI): *m/z* calcd. for

 ${}^{_{65}}C_{21}H_{22}NO_6\left[M\,+\,H\right]^+$ 384.14416, found 384.14378; calcd. for $C_{21}H_{21}NNaO_6\left[M\,+\,Na\right]^+$ 406.12611, found 406.12592.

(6aR,7S,10aS)-7-(4-(dimethylamino)phenyl)-6a-nitro-7,8,10,10a-tetrahydro-6*H*-benzo[*c*]chromen-9(6a*H*)-one (3k).

- 70 The product 3k was obtained according to the general procedure as a white solid (56.8 mg, 78% yield), m.p.: 67-71 °C. HPLC (Daicel Chiralpak IB column, n-hexane-2-propanol 65:35, flow rate 1.0 mL/min, detection at 254 nm): $t_{major} = 21.0 \text{ min}, t_{minor} =$ 16.7 min, 83% ee; $[\alpha]_{D}^{20} = +14.0$ (c 1.1, CH₂Cl₂); ¹H NMR (400 ⁷⁵ MHz, acetone- d_6): δ 7.35 (d, J = 7.6 Hz, 1H, ArH), 7.18–7.14 (m, 1H, ArH), 7.02 (d, J = 8.4 Hz, 2H, ArH), 6.96 (t, J = 7.6 Hz, 1H, ArH), 6.83 (d, J = 8.4 Hz, 1H, ArH), 6.68 (d, J = 8.8 Hz, 2H, ArH), 5.00 (dd, $J_1 = 2.0$ Hz, $J_2 = 12.4$ Hz, 1H, CH), 4.66 (d, J =12.4 Hz, 1H, CH), 4.55–4.50 (m, 1H, CH), 3.67 (dd, $J_1 = 4.4$ Hz, $_{80} J_2 = 10.4 \text{ Hz}, 1\text{H}, \text{CH}), 3.18 \text{ (dd}, J_1 = 10.4 \text{ Hz}, J_2 = 16.4 \text{ Hz}, 1\text{H},$ CH), 3.06 (dd, $J_1 = 5.2$ Hz, $J_2 = 16.8$ Hz, 1H, CH), 2.93 (s, 6H, $2CH_3$), 2.90–2.80 (m, 1H, CH), 2.62 (dd, $J_1 = 4.4$ Hz, $J_2 = 16.4$ Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, acetone- d_6): δ 207.8, 154.4, 152.4, 131.2, 130.7, 130.2, 125.7, 123.8, 123.5, 118.7, 85 114.1, 91.9, 67.1, 47.1, 45.6, 43.8, 41.3, 37.9 ppm; IR (KBr): v
- 1709, 1611, 1548, 1523, 1488, 1348, 1225, 1042, 943, 823, 763, 613, 567, 525, 424 cm⁻¹; HRMS (ESI): m/z calcd. for $C_{21}H_{23}$. $N_2O_4 [M + H]^+$ 367.16523, found 367.16423.

90 (6aR,7S,10aS)-7-(naphthalen-1-yl)-6a-nitro-7,8,10,10a-

tetrahydro-6H-benzo[c]chromen-9(6aH)-one (3l). The product 31 was obtained according to the general procedure as a white solid (67.6 mg, 90% yield), m.p.: 99-102 °C. HPLC (Daicel Chiralpak AD-H column, n-hexane-2-propanol 80:20, flow rate 95 1.0 mL/min, detection at 254 nm): $t_{major} = 45.2 \text{ min}, t_{minor} = 16.0$ min, 79% ee; $[\alpha]_{D}^{20} = +121.0$ (c 2.6, CH₂Cl₂); ¹H NMR (400 MHz, acetone- d_6): δ 8.37 (d, J = 8.4 Hz, 1H, ArH), 7.97 (d, J =8.0 Hz, 1H, ArH), 7.91 (d, J = 8.0 Hz, 1H, ArH), 7.64–7.55 (m, 2H, ArH), 7.48 (t, J = 7.6 Hz, 1H, ArH), 7.41–7.38 (m, 2H, ArH), ¹⁰⁰ 7.15 (t, J = 8.0 Hz, 1H, ArH), 6.98 (t, J = 7.6 Hz, 1H, ArH), 6.81 $(d, J = 8.4 \text{ Hz}, 1\text{H}, \text{ArH}), 4.98 (dd, J_1 = 1.6 \text{ Hz}, J_2 = 12.4 \text{ Hz}, 1\text{H},$ CH), 4.90–4.80 (m, 3H, CH+CH₂), 3.48 (dd, $J_1 = 11.6$ Hz, $J_2 =$ 16.4 Hz, 1H, CH), 3.19-3.07 (m, 2H, CH₂), 2.68 (dd, $J_1 = 3.6$ Hz, $J_2 = 16.8$ Hz, 1H, ArH) ppm; ¹³C NMR (100 MHz, acetone-d₆): δ 105 207.9, 154.4, 135.9, 135.5, 133.8, 131.3, 131.0, 130.7, 130.3, 128.4, 127.8, 127.4, 126.7, 124.7, 123.7, 123.2, 118.6, 93.6, 67.3, 47.0, 44.1, 38.8, 37.9 ppm; IR (KBr): v 1722, 1586, 1544, 1491, 1459, 1343, 1275, 1259, 1230, 1214, 1070, 802, 780, 753 cm^{-1} ; HRMS (ESI): m/z calcd. for C₂₃H₁₉NNaO₄ [M + Na]⁺ 396.12063, 110 found 396.12062.

(6aR,7S,10aS)-2-bromo-7-(4-chlorophenyl)-6a-nitro-7,8,10,10a-tetrahydro-6*H*-benzo[*c*]chromen-9(6a*H*)-one (3m).

The product **3m** was obtained according to the general procedure The product **3m** was obtained according to the general procedure (Daicel Chiralpak IB column, *n*-hexane–2-propanol 65:35, flow rate 1.0 mL/min, detection at 254nm): $t_{major} = 22.0$ min, $t_{minor} =$ 17.6 min, 78% *ee*; $[\alpha]_D^{20} = +10.1$ (*c* 0.67, CH₂Cl₂); ¹H NMR (400 MHz, acetone-d₆): δ 7.58 (d, J = 2.4 Hz, 1H, ArH), 7.42–7.39 (m, 120 2H, ArH), 7.32 (dd, $J_1 = 2.4$ Hz, $J_2 = 8.8$ Hz, 1H, ArH), 7.28–7.25 (m, 2H, ArH), 6.82 (d, J = 8.8 Hz, 1H, ArH), 5.07 (dd, $J_1 = 2.4$ Hz, $J_2 = 12.8$ Hz, 1H, CH), 4.74 (dd, $J_1 = 2.4$ Hz, $J_2 = 12.8$ Hz, 1H, CH), 4.65 (dd, $J_1 = 4.6$ Hz, $J_2 = 12.8$ Hz, 1H, CH), 3.92–3.88 (m, 1H, CH), 3.24–3.18 (m, 1H, CH), 3.12 (dd, $J_1 = 5.2$ Hz, $J_2 =$ 125 17.2 Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, acetone- d_6): δ 206.8, 153.6, 137.7, 135.7, 133.7, 133.3, 131.9, 130.7, 126.0,

Org. Biomol. Chem., 2015, 13, 00-00 | 7

This journal is © The Royal Society of Chemistry 2015

120.8, 115.2, 91.7, 67.0, 46.8, 45.4, 43.2, 37.5 ppm; IR (KBr): $\tilde{\nu}$ 1723, 1547, 1479, 1409, 1339, 1279, 1251, 1092, 1067, 1014, 830, 813, 749, 692, 613, 510 cm⁻¹; HRMS (ESI): *m/z* calcd. for C₁₉H₁₅BrClNNaO₄ [M + Na]⁺ 457.97652, found 457.97672.

(6aR,7S,10aS)-2-chloro-7-(4-chlorophenyl)-6a-nitro-

7,8,10,10a-tetrahydro-6*H*-benzo[*c*]chromen-9(6a*H*)-one (3n). The product 3n was obtained according to the general procedure as a white solid (46.2 mg, 59% yield), m.p. 192-196 °C. HPLC 10 (Daicel Chiralpak IB column, n-hexane-2-propanol 65:35, flow rate 1.0 mL/min, detection at 254 nm): $t_{major} = 21.6 \text{ min}, t_{minor} =$ 16.6 min, 73% *ee*; $[\alpha]_{\rm D}^{20}$ = +11.8 (*c* 1.4, CH₂Cl₂); ¹H NMR (400 MHz, acetone- d_6): δ 7.44 (d, J = 2.8 Hz, 1H, ArH), 7.40 (d, J =8.4 Hz, 2H, ArH), 7.26 (d, J = 8.8 Hz, 2H, ArH), 7.19 (dd, $J_1 =$ $_{15}$ 2.4 Hz, $J_2 = 8.8$ Hz, 1H, ArH), 6.87 (d, J = 8.8 Hz, 1H, ArH), 5.06 (dd, $J_1 = 2.0$ Hz, $J_2 = 12.4$ Hz, 1H, CH), 4.74 (J = 12.8 Hz, 1H, CH), 4.66–4.62 (m, 1H, CH), 3.90 (dd, $J_1 = 4.4$ Hz, $J_2 = 10.4$ Hz, 1H, CH), 3.21 (dd, J₁ = 10.4 Hz, J₂ = 16.8 Hz, 1H, CH), 3.12 $(dd, J_1 = 4.8 Hz, J_2 = 17.2 Hz, 1H, CH), 2.94 (dd, J_1 = 12.8 Hz, J_2)$ ²⁰ = 17.2 Hz, 1H, CH), 2.71 (dd, *J*₁ = 4.4 Hz, *J*₂ = 16.4 Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, acetone-d₆): δ 206.8, 153.2, 137.8, 135.7, 131.9, 130.8, 130.7, 130.4, 128.0, 125.5, 120.4, 91.7, 67.1, 46.8, 45.4, 43.2, 37.6 ppm; IR (KBr): v 1722, 1540, 1488, 1411,

1339, 1250, 1089, 1010, 844, 817, 749, 690, 638, 613, 505, 423 ²⁵ cm⁻¹; HRMS (ESI): m/z calcd. for $C_{19}H_{15}Cl_2NNaO_4 [M + Na]^+$ 414.02703, found 414.02719.

(6aR,7S,10aS)-2,4-dibromo-7-(4-chlorophenyl)-6a-nitro-

- **7,8,10,10a-tetrahydro-6H-benzo**[*c*]**chromen-9(6aH)-one** (30). ³⁰ The product **30** was obtained according to the general procedure as a white solid (46.9 mg, 45% yield), m.p. 205–208 °C. HPLC (Daicel Chiralpak IB column, *n*-hexane–2-propanol 65:35, flow rate 1.0 mL/min, detection at 254 nm): $t_{\text{major}} = 20.0 \text{ min}, t_{\text{minor}} =$ 17.6 min, 65% *ee*; $[\alpha]_D^{20} = +1.5$ (*c* 1.6, CH₂Cl₂); ¹H NMR (400 ³⁵ MHz, acetone-*d*₆): δ 7.62 (s, 2H, ArH), 7.41 (d, *J* = 8.4 Hz, 2H,
- ArH), 7.27 (d, J = 8.8 Hz, 2H, ArH), 5.23 (dd, $J_1 = 2.0$ Hz, $J_2 = 12.4$ Hz, 1H, CH), 4.89 (d, J = 12.8 Hz, 1H, CH), 4.73–4.68 (m, 1H, CH), 3.95 (dd, $J_1 = 4.4$ Hz, $J_2 = 9.8$ Hz, 1H, CH), 3.23–3.11 (m, 2H, CH₂), 2.99 (dd, $J_1 = 13.2$ Hz, $J_2 = 17.2$ Hz, 1H, CH),
- ⁴⁰ 2.84–2.74 (m, 1H, CH) ppm; ¹³C NMR (100 MHz, acetone-*d*₆): δ 206.5, 150.6, 137.6, 135.9, 135.8, 133.3, 131.9, 130.7, 127.3, 115.1, 113.2, 91.7, 67.8, 46.8, 45.4, 43.2, 37.6 ppm; IR (KBr): \tilde{v} 1722, 1548, 1461, 1412, 1276, 1262, 1245, 1227, 1170, 1090, 1064, 1013, 866, 825, 750, 725, 671, 609, 531, 472 cm⁻¹; HRMS ⁴⁵ (ESI): *m/z* calcd. for C₁₉H₁₄Br₂CINNaO₄ [M + Na]⁺ 535.88703,
 - found 535.88661. (6aR,7S,10aS)-2,4-dichloro-7-(4-chlorophenyl)-6a-nitro-

(6ax, 75, 10a5) - 2, 4-dichioro-7-(4-chiorophenyi)-6a-hitro-7,8,10,10a-tetrahydro-6*H*-benzo[*c*]chromen-9(6a*H*)-one (3p).

- ⁵⁰ The product **3p** was obtained according to the general procedure as a white solid (46.4 mg, 54% yield), m.p. 211–220 °C. HPLC (Daicel Chiralpak IB column, *n*-hexane–2-propanol 65:35, flow rate 1.0 mL/min, detection at 254 nm): $t_{\text{major}} = 19.7$ min, $t_{\text{minor}} =$ 16.6 min, 68% *ee*; $[\alpha]_D^{20} = +2.1$ (*c* 2.0, CH₂Cl₂); ¹H NMR (400
- ⁵⁵ MHz, acetone- d_6): δ 7.45 (d, J = 2.4 Hz, 1H, ArH), 7.41 (d, J = 8.8 Hz, 2H, ArH), 7.36 (d, J = 2.4 Hz, 1H, ArH), 7.27 (d, J = 8.4 Hz, 2H, ArH), 5.23 (dd, $J_1 = 2.0$ Hz, $J_2 = 12.4$ Hz, 1H, CH), 4.89 (d, J = 12.4 Hz, 1H, CH), 4.72–4.68 (m, 1H, CH), 3.96 (dd, $J_1 = 4.6$ Hz, $J_2 = 9.8$ Hz, 1H, CH), 3.23–3.11 (m, 2H, CH₂), 2.99 (dd,
- ⁶⁰ J₁ = 12.8 Hz, J₂ = 17.2 Hz, 1H, CH), 2.77 (dd, J₁ = 4.6 Hz, J₂ = 16.8 Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, acetone-d₆): δ 206.5, 149.2, 137.6, 135.8, 132.0, 130.8, 130.4, 129.7, 127.8, 127.0, 124.2, 91.7, 67.7, 46.8, 45.4, 43.2, 37.7 ppm; IR (KBr): \tilde{v}

1724, 1549, 1469, 1412, 1248, 1226, 1091, 1066, 1035, 1013, 65 864, 846, 825, 620, 535 cm⁻¹; HRMS (ESI): *m/z* calcd. for $C_{19}H_{14}Cl_3NNaO_4 [M + H]^+$ 447.98806, found 447.98771.

(6aR,7S,10aS)-7-(4-chlorophenyl)-4-methoxy-6a-nitro-

- **7,8,10,10a-tetrahydro-6H-benzo**[*c*]**chromen-9(6aH)-one** (3**q**). ⁷⁰ The product 3**q** was obtained according to the general procedure as a white solid (62.1 mg, 80% yield), m.p. 201–205 °C. HPLC (Daicel Chiralpak AD-H column (*n*-hexane–2-propanol 80:20, flow rate 1.0 mL/min, detection at 254 nm): $t_{\text{major}} = 19.8$ min, $t_{\text{minor}} = 24.7$ min, 73% *ee*; $[\alpha]_{\text{D}}^{20} = -17.6$ (*c* 2.4, CH₂Cl₂); ¹H 75 NMR (400 MHz, acetone-*d*₆): δ 7.40 (d, J = 8.4 Hz, 2H, ArH),
- ⁷⁵ NMR (400 MHz, acetone- d_6): δ 7.40 (d, J = 8.4 Hz, 2H, ArH), 7.27 (d, J = 8.4 Hz, 2H, ArH), 6.93–6.88 (m, 2H, ArH), 6.86–6.82 (m, 1H, ArH), 5.05 (dd, $J_1 = 2.0$ Hz, $J_2 = 12.4$ Hz, 1H, CH), 4.64 (d, J = 12.4 Hz, 1H, CH), 4.60–4.56 (m, 1H, CH), 3.87 (dd, $J_1 = 4.4$ Hz, $J_2 = 10.8$ Hz, 1H, CH), 3.77 (s, 3H, CH₃), 3.24
- ⁸⁰ (dd, $J_1 = 10.8$ Hz, $J_2 = 16.4$ Hz, 1H, CH), 3.08 (dd, $J_1 = 4.8$ Hz, $J_2 = 17.2$ Hz, 1H, CH), 2.91 (dd, $J_1 = 12.8$ Hz, $J_2 = 17.2$ Hz, 1H, CH), 2.67 (dd, $J_1 = 4.4$ Hz, $J_2 = 16.4$ Hz, 1H, CH) ppm; ¹³C NMR (100 MHz, acetone- d_6): δ 207.2, 150.5, 144.1, 137.9, 135.7, 131.9, 130.7, 124.0, 123.3, 122.3, 112.4, 91.8, 66.7, 57.0, 47.0, 45.3,
- $_{85}$ 43.1, 37.8 ppm; IR (KBr): $\tilde{\nu}$ 1727, 1547, 1481, 1411, 1334, 1262, 1220, 1081, 1064, 1015, 829, 816, 788, 741, 666, 537 cm $^{-1}$; HRMS (ESI): m/z calcd. for $C_{20}H_{18}CINNaO_5$ [M + Na] $^+$ 410.07657, found 410.07655.

(6a*R*,7*S*,10a*S*)-7-(4-chlorophenyl)-4-ethoxy-6a-nitro-

- **7,8,10,10a-tetrahydro-6H-benzo**[*c*]**chromen-9(6aH)-one** (**3r).** The product **3r** was obtained according to the general procedure as a white solid (64.7 mg, 81% yield), m.p. 76–80 °C. HPLC (Daicel Chiralpak AD-H column, *n*-hexane–2-propanol 80:20, 95 flow rate 1.0 mL/min, detection at 254 nm): $t_{major} = 14.7$ min, $t_{minor} = 22.3$ min, 71% *ee*; $[\alpha]_D^{20} = -16.2$ (*c* 2.3, CH₂Cl₂); ¹H NMR (400 MHz, acetone-*d*₆): δ 7.42–7.38 (m, 2H, ArH), 7.28–7.25 (m, 2H, ArH), 6.92–6.86 (m, 2H, ArH), 6.83–6.78 (m, 1H, ArH), 5.06 (dd, $J_1 = 2.4$ Hz, $J_2 = 12.4$ Hz, 1H, CH), 4.63 (d, *J* 00 = 12.4 Hz, 1H, CH), 4.60–4.56 (m, 1H, CH), 4.03–3.96 (m, 2H, CH₂), 3.86 (dd, $J_1 = 4.4$ Hz, $J_2 = 10.8$ Hz, 1H, CH), 3.24 (dd, $J_1 =$
- 10.4 Hz, $J_2 = 16.4$ Hz, 1H, CH), 3.07 (dd, $J_1 = 4.8$ Hz, $J_2 = 17.2$ Hz, 1H, CH), 2.90 (dd, $J_1 = 13.2$ Hz, $J_2 = 17.2$ Hz, 1H, CH), 2.90 (dd, $J_1 = 13.2$ Hz, $J_2 = 17.2$ Hz, 1H, CH), 2.68 (dd, $J_1 = 4.4$ Hz, $J_2 = 16.4$ Hz, 1H, CH), 1.34 (t, J = 6.8 Hz, 3H, 105 CH₃) ppm; ¹³C NMR (100 MHz, acetone- d_6): δ 207.3, 149.7,
- ¹⁰⁵ CH3) ppH, ¹⁰⁵ C Wirk (100 WH2, accone a_{61} . b 207.3, 149.7, 144.3, 137.9, 135.6, 131.9, 130.6, 124.0, 123.3, 122.3, 113.6, 91.8, 66.7, 65.8, 47.0, 45.3, 43.1, 37.8, 16.1 ppm; IR (KBr): $\bar{\nu}$ 1715, 1586, 1544, 1487, 1473, 1338, 1261, 1212, 1091, 1013, 829, 764, 750 cm⁻¹; HRMS (ESI): m/z calcd. for C₂₁H₂₀ClNNaO₅ ¹¹⁰ [M + Na]⁺ 424.09222, found 424.09249.

(1S,4aS,10aR)-1-(4-bromophenyl)-10a-nitro-

1,4,4a,9,10,10a-hexahydrophenanthren-3(*2H*)-one (3s). The product **3s** was obtained according to the general procedure as ¹¹⁵ white solid (50.9 mg, 64% yield), m.p. 74–78 °C. HPLC (Daicel Chiralpak AD-H column, *n*-hexane–2-propanol 80:20, flow rate 1.0 mL/min, detection at 254 nm): $t_{major} = 12.9$ min, $t_{minor} = 15.0$ min, 81% *ee*; $[\alpha]_D^{20} = +9.3$ (*c* 1.8, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃): δ 7.49–7.42 (m, 2H, ArH), 7.18–7.06 (m, 4H, ArH), 6.97 (d, *J* = 8.8 Hz, 2H, ArH), 4.26 (dd, *J*₁ = 5.2 Hz, *J*₂ = 13.2 Hz, 1H, CH), 3.56 (dd, *J*₁ = 6.0 Hz, *J*₂ = 6.8 Hz, 1H, CH), 3.04 (dd, *J*₁ = 6.8 Hz, *J*₂ = 16.8 Hz, 1H, CH), 2.58 (dd, *J*₁ = 13.2 Hz, *J*₂ = 17.2 Hz, 1H, CH), 2.52–2.42 (m, 1H, CH) pm; ¹³C NMR (100 MHz, 125 CDCl₃): δ 207.0, 136.1, 135.0, 132.6, 132.1, 129.7, 129.1, 128.5, 127.2, 126.9, 122.7, 92.5, 49.4, 47.2, 42.0, 38.6, 28.0, 25.3 ppm.

This journal is © The Royal Society of Chemistry 2015

Organic & Biomolecular Chemistry Accepted Manuscript

70

IR (KBr): \tilde{v} 1715, 1537, 1489, 1453, 1441, 1409, 1351, 12776, 1226, 1120, 1109, 1076, 1009, 908, 832, 806, 753, 735, 649, 613, 516, 446 cm⁻¹; HRMS (ESI): *m*/*z* calcd. for C₂₀H₁₈BrNNaO₃ [M + Na]⁺ 422.03623, found 422.03681.

(6aR,7S,10aS)-7-(4-bromophenyl)-6a-nitro-7,8,10,10atetrahydro-6*H*-benzo[*c*]thiochromen-9(6a*H*)-one (3t). The product 3t was obtained according to the general procedure as a white solid (62.1 mg, 74% yield), m.p. 180–182 °C. HPLC

- ¹⁰ (Daicel Chiralpak AD-H column, *n*-hexane–2-propanol 80:20, flow rate 1.0 mL/min, detection at 254 nm): $t_{major} = 16.3$ min, $t_{minor} = 27.4$ min, 95% *ee*; $[\alpha]_D^{20} = +37.6$ (*c* 1.9, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃): δ 7.46 (d, J = 8.8 Hz, 2H, ArH), 7.21–7.12 (m, 4H, ArH), 6.97 (d, J = 8.4 Hz, 2H, ArH), 4.16 (dd,
- ¹⁵ $J_1 = 6.8$ Hz, $J_2 = 10.8$ Hz, 1H, CH), 3.95 (dd, $J_1 = 6.0$ Hz, $J_2 = 7.6$ Hz, 1H, CH), 3.63 (d, J = 14.0 Hz, 1H, CH), 3.56 (d, J = 14.0 Hz, 1H, CH), 3.08 (dd, $J_1 = 7.8$ Hz, $J_2 = 17.4$ Hz, 1H, CH), 2.90–2.78 (m, 3H, CH+CH₂) ppm; ¹³C NMR (400 MHz, CDCl₃): δ 206.0, 135.5, 134.6, 132.3, 131.5, 129.9, 129.6, 127.70, 127.69, 126.4,

25 (4S,4aR,12cS)-4-(4-chlorophenyl)-4a-nitro-3,4,4a,5tetrahydro-1H-dibenzo[c,f]chromen-2(12cH)-one (e

- tetrahydro-1*H*-dibenzo[*c*,*f*]chromen-2(12*cH*)-one (*ent*-3**u**). The product *ent*-3**u** was obtained according to the general procedure using catalyst **III** as a white solid (49.4 mg, 61% yield), m.p. 206–211 °C. HPLC (Daicel Chiralpak AD-H column, *n*-
- ³⁰ hexane–2-propanol 80:20, flow rate 1.0 mL/min, detection at 254 nm): $t_{major} = 21.9$ min, $t_{minor} = 32.8$ min, 78% *ee*; ¹H NMR (400 MHz, acetone- d_6): δ 8.16 (d, J = 8.4 Hz, 1H, ArH), 7.84 (d, J = 8.0 Hz, 1H, ArH), 7.75 (d, J = 8.8 Hz, 1H, ArH), 7.57 (t, J = 7.6 Hz, 1H, ArH), 7.44–7.33 (m, 5H, ArH), 7.04 (d, J = 8.8 Hz, 1H,
- ³⁵ ArH), 5.27–5.20 (m, 2H, CH₂), 5.00 (d, J = 12.4 Hz, 1H, CH), 3.92 (dd, $J_1 = 4.6$ Hz, $J_2 = 9.8$ Hz, 1H, CH), 3.44 (dd, $J_1 = 9.8$ Hz, $J_2 = 16.6$ Hz, 1H, CH), 3.31 (dd, $J_1 = 4.2$ Hz, $J_2 = 18.0$ Hz, 1H, CH), 2.89 (dd, $J_1 = 13.6$ Hz, $J_2 = 18.0$ Hz, 1H, CH), 2.81 (d, $J_1 = 4.6$ Hz, $J_2 = 16.6$ Hz, 1H, CH) ppm; ¹³C NMR (100 MHz,
- ⁴⁰ acetone- d_6): δ 206.8, 151.8, 138.0, 135.7, 133.5, 132.0, 131.7, 131.4, 130.7, 130.3, 129.1, 125.8, 123.5, 120.1, 114.5, 91.7, 66.5, 45.8, 45.7, 43.3, 34.5 ppm; IR (KBr): \tilde{v} 1720, 1620, 1599, 1543, 1471, 1413, 1343, 1233, 1112, 1014, 821, 763, 442 cm⁻¹; HRMS (ESI): m/z calcd. for C₂₃H₁₉ClNO₄ [M + H]⁺ 408.09971, found 409, 10055 cm⁻¹ cm⁺¹ cm⁺¹ cm⁻¹ cm⁺¹ cm
- ⁴⁵ 408.10005; *m/z* calcd. for $C_{23}H_{18}CINNaO_4 [M + Na]^+$ 430.08166, found 430.08193.

Acknowledgements

We are grateful for financial support from the National Natural Science Foundation of China (grant No 21272024).

50 Notes and references

- For selected references, see: (a) E. E. Schweizer and O. Meeder-Nycz, Chromenes, Chromanes, and Chromones, Wiley-Interscience, New York, **1977**; (b) K. C. Nicolaou, J. A. Pfefferkorn, A. J. Roecker, G. Q. Cao, S. Barluenga and H. J. Mitchell, *J. Am. Chem. Soc.* 2000,
- 122, 9939; (c) G. Zeni and R. C. Larock, Chem. Rev. 2004, 104, 2285; (d) H. C. Shen, Tetrahedron 2009, 65, 3931; (e) S. R. Trenor, A. R. Shultz, B. J. Love and T. E. Long, Chem. Rev. 2004, 104, 3059; (f) S. R. Trenor, A. R. Shultz, B. J. Love and T. E. Long, Chem. Rev. 2004, 104, 3059; (g) A. Coi, A. M. Bianucci, V.
- 60 Calderone, L. Testai, M. Digiacomo, S. Rapposelli and A. Balsamo, *Bioorg. Med. Chem.* 2009, **17**, 5565.

- 2 For selected reviews on synthesis of chroman derivatives with organocatalysis, see: (a) C. F. Nising and S. Bräse, *Chem. Soc. Rev.* 2012, **41**, 988; (b) C. Bhanja, S. Jena, S. Nayak and S. Mohapatra,
- Beilstein J. Org. Chem. 2012, 8, 1668; (c) M. G. Núnez, P. García, R. F. Moro and D. Díez, *Tetrahedron* 2010, 66, 2089; For selected examples on synthesis of chroman derivatives with organocatalysis, see: (a) W. Hou, B. Zheng, J. Chen and Y. Peng, Org. Lett. 2012, 14, 2378; (b) Z.-X. Jia, Y.-C. Luo, X.-N. Cheng, P.-F. Xu and Y.-C. Gu,
- J. Org. Chem. 2013, 78, 6488; (c) Z.-K. Fu, J.-Y. Pan, D.-C. Xu and J.-W. Xie, RSC Adv. 2014, 4, 51548; (d) A.-B. Xia, C. Wu, T. Wang, Y.-P. Zhang, X.-H. Du, A.-G. Zhong, D.-Q. Xu and Z.-Y. Xua, Adv. Synth. Catal. 2014, 356, 1753; (e) H. Shen, K.-F. Yang, Z.-H. Shi, J.-X. Jiang, G.-Q. Lai and L.-W. Xu, Eur. J. Org. Chem. 2011, 5031;
- (f) C. Liu, X. Zhang, R. Wang and W. Wang, Org. Lett. 2010, 12, 4948; (g) Y.-F. Wang, W. Zhang, S.-P. Luo, B.-L. Li, A.-B. Xia, A.-G. Zhong and D.-Q. Xu, Chem. Asian J., 2009, 4, 1834; (h) X. Liu and Y. Lu, Org. Biomol. Chem. 2010, 8, 4063; (i) J. Wang, H. Xie, H. Li, L. Zu and W. Wang, Angew. Chem. Int. Ed. 2008, 47, 4177;
 (j) Z. Du, C. Zhou, Y. Gao, Q. Ren, K. Zhang, H. Cheng, W. Wang
- and J. Wang, Org. Biomol. Chem. 2012, 10, 36.
 For selected examples, see: (a) R. Q. Skrabek, L. Galimova, K. Ethans and D. Perry, J. Pain 2008, 9, 164; (b) L.-J. Cheng, J.-H. Xie, Y. Chen, L.-X. Wang and Q.-L. Zhou, Org. Lett. 2013, 4, 764; (c) D.
- D. Dixon, M. A. Tius, G. A. Thakur, H. Zhou, *Org. Dett. Delts*, *J. P. Commun. V. G.*Shukla, Y. Peng and A. Makriyannis, *Bioorg. Med. Chem. Lett.* 2012, 22, 5322; (*d*) G. Ogawa, M. A. Tius, H. Zhou, S. P. Nikas, A. Halikhedkar, S. Mallipeddi and A. Makriyannis, *J. Med. Chem.* 2015, 58, 3104; (*e*) G. Appendino, S. Gibbons, A. Giana, A. Pagani,
 G. Grassi, M. Stavri, E. Smith and M. M. Rahman, *J. Nat. Prod.* 2008, 71, 1427.
- For selected examples, see: (a) M. Lloyd, M. Tom, H. Curtis, K. Vladimir, S. Rudong, P. Sam, B. Paul, P. Jeremy and B. N. Aini, *PCT Int. Appl.* 2011; (b) K. Konrad, H. Cecilia, N. Marita, G. Mikael, L. Ye, S. Andrei, W. Robert R. and R. Ronald W, *PCT Int. Appl.* 2002.
- 5 (a) P. Kotame, B.-C. Hong and J.-H. Liao, *Tetrahedron Lett.* 2009, **50**, 704; (b) B.-C. Hong, P. Kotame, C.-W. Tsai and J.-H. Liao, *Org. Lett.* 2010, **12**, 776.
- 100 6 L. Liu, Y. Zhu, K. Huang, B. Wang, W. Chang and J. Li, *Eur. J. Org. Chem.* 2014, 342.
- 7 Z.-C. Geng, S.-Y. Zhang, N.-K. Li, N. Li, J. Chen, H.-Y. Li and X.-W. Wang, J. Org. Chem. 2014, 79, 10772.
- 8 B.-C. Hong, P. Kotame and J.-H. Liao, *Org. Biomol. Chem.* 2011, 9, 382.
 - 9 A. Raja, B.-C. Hong and G.-H. Lee, Org. Lett. 2014, 16, 5756.
- For selected examples, see: (a) M. Rueping, J. Dufour and M. S. Maji, Chem. Commun. 2012, 48, 3406; (b) E. L. Pearson, L. C. H. Kwan, C. I. Turner, G. A. Jones, A. C. Willis, M. N. Paddon-Row and M. S.
- Sherburn, J. Org. Chem. 2006, 71, 6099; (c) D.-H. Jhuo, B.-C. Hong,
 C.-W. Chang and G.-H. Lee, Org. Lett. 2014, 16, 2724; (d) Y.-C.
 Chen, Synlett, 2008, 1919; (e) J.-W. Xie, W. Chen, R. Li, M. Zeng,
 W. Du, L. Yue, Y.-C. Chen, Y. Wu, J. Zhu and J.-G. Deng, Angew.
 Chem., Int. Ed. 2007, 46, 389; (f) F. Gläser. M. C. Bröhmer, T.
- Hurrle, M. Nieger and S. Bräse, *Eur. J. Org. Chem.* 2015, 1516; (g)
 R. Girotti, A. Marrocchi, L. Minuti, O. Piermatti, F. Pizzo and L. Vaccaro, *J. Org. Chem.*, 2006, **71**, 70; (*h*) Z.-W. Guo, X.-S. Li, W.-D. Zhu and J.-W. Xie, *Eur. J. Org. Chem.* 2012, 6924.
- For selected references on synthesis of 3-nitro-2*H*-chromenes, see: (*a*)
 A. N. Mamdouh, K. Rafik and G. Gerald, *Synth. Commun.* 1990, 20, 783; (*b*) R. Koussini and A. S. Al-Shihri, *Jordan J. Chem.*, 2008, 3, 103. (*c*) A. H. Clark, J. D. McCorvy, V. J. Watts and D. E. Nichols. *Bioorg. Med. Chem.* 2011, 19, 5420.

For selected examples, see: (a) W.-Y. Chen, L. Ouyang, R.-Y. Chen and X.-S. Li, *Tetrahedron Lett.*, 2010, **51**, 3972; (b) S.-Z. Nie, Z.-P. Hu, Y.-N. Xuan, J.-J. Wang, X.-M. Li and M. Yan, *Tetrahedron: Asymmetry*, 2010, **21**, 2055; (c) W.-Y. Chen, P. Li, J.-W. Xie and X.-S. Li, *Catal. Commun.*, 2011, **12**, 502; (d) Y. Jia, W. Yang and D.-M. Du, *Org. Biomol. Chem.*, 2012, **10**, 4739; (e) Y. Jia and D.-M. Du, *Adv.* 2013, **3**, 1970; (f) W. Yang, H.-X. He, Y. Gao and D.-M. Du, *Adv. Synth. Catal.* 2013, **355**, 3670; (g) J.-H. Li and D.-M. Du, *Org. Biomol. Chem.*, 2015, **13**, 5636; (h) F. Tan, C. Xiao, H.-G.

²⁰¹⁵ Org. Biomol. Chem., 2015, **13**, 00–00 | **9**

Cheng, W. Wu, K.-R. Ding and W.-J. Xiao, *Chem. Asian J.*, 2012, **7**, 493; (*i*) F. Tan, L.-Q. Lu, Q.-Q. Yang, W. Guo, Q. Bian, J.-R. Chen and W.-J. Xiao, *Chem. Eur. J.*, 2014, **20**, 3415;

- For selected examples, see: (a) G. Bencivenni, L.-Y. Wu, A. Mazzanti,
 B. Giannichi, F. Pesciaioli, M.-P. Song, G. Bartoli and P. Melchiorre, Angew. Chem., Int. Ed. 2009, 48, 7200; (b) L.-Y. Wu, G. Bencivenni,
- M. Mancinelli, A. Mazzanti, G. Bartoli and P. Melchiorre, Angew. Chem. Int. Ed. 2009, 48, 7196; (c) D.-F. Yu, Y. Wang and P.-F. Xu, Adv. Synth. Catal. 2011, 353, 2960; (d) J.-X. Zhang, N.-K. Li, Z.-M.
 Liu, X.-F. Huang, Z.-C. Geng and X.-W. Wang, Adv. Synth. Catal.
- Liu, X.-F. Huang, Z.-C. Geng and X.-W. Wang, *Adv. Synth. Catal.* 2013, **355**, 797; (e) H.-L. Cui and F. Tanaka, *Chem. Eur. J.* 2013, **19**, 6213; (f) J. Liang, Q. Chen, L. Liu, X. Jiang and R. Wang, *Org. Biomol. Chem.* 2013, **11**, 1441. (g) Y. Liu, T.-R. Kang, Q.-Z. Liu, L.-M. Chen, Y.-C. Wang, J. Liu, Y.-M. Xie, J.-L. Yang and L. He, *Org. Lett.* 2013, **15**, 6090.
- 14. Crystallographic data for compound-**3m** (CCDC-1406419) has been provided as CIF file in supporting information.
- 15. S. Maity, T. Naveen, U. Sharma and D. Maiti, Org. Lett., 2013, 15, 3384.