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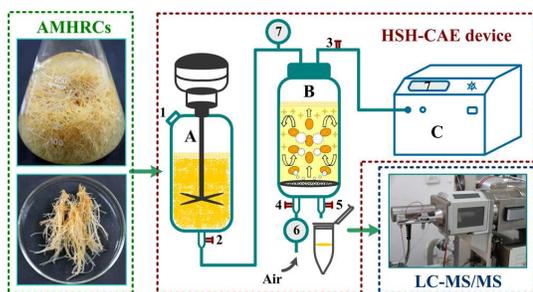


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HSH-CAE-LC-MS/MS opened up a new avenue for the direct determination of secondary metabolite profiles from fresh plant *in vitro* cultures.

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Direct determination of astragalosides and isoflavonoids from fresh *Astragalus membranaceus* hairy root cultures by high speed homogenization coupled with cavitation-accelerated extraction followed by liquid chromatography-tandem mass spectrometry

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26 **Abstract**

27 A direct analysis approach for plant *in vitro* cultures, namely high speed
28 homogenization coupled with cavitation-accelerated extraction (HSH-CAE) followed
29 by liquid chromatography-tandem mass spectrometry (LC-MS/MS), was developed
30 for simultaneous determination of six astragalosides and five isoflavonoids in
31 *Astragalus membranaceus* hairy root cultures (AMHRCs). In comparison to reported
32 soxhlet extraction (SE) and ultrasound-assisted extraction (UAE) methods, the
33 proposed sample preparation procedure (HSH-CAE) offers significant improvements
34 with regard to simplicity in operation (elimination of biomass drying and grinding),
35 high efficiency, enhanced yield and green aspects in terms of saving energy cost and
36 minimizing the generation of waste. Additionally, the HSH-CAE mechanism was
37 clarified via cytohistological studies of samples at cellular/tissular levels. Moreover,
38 the established LC-MS/MS method provided linearity with correlation coefficients
39 above 0.9991, limit of detections (LODs) below 1.77 ng mL⁻¹, relative standard
40 deviations (RSDs) below 6.01%, and recoveries above 96.84%. Furthermore, the
41 proposed HSH-CAE-LC-MS/MS method was also successfully applied for screening
42 high-productive AMHRCs. Overall, this study opened up a new avenue for the direct
43 determination of secondary metabolic profiles from fresh plant *in vitro* cultures, which
44 was valuable for improving the quality control of plant cell/organ cultures and shed
45 light on the metabolomics analysis from biological samples.

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51 1. Introduction

52 *Astragalus membranaceus* (Fisch.) Bunge roots (Radix astragali) are commonly
53 used in United States, Europe and Asian countries as traditional medicines for the
54 treatment of cardiovascular disease, cancers, diabetes mellitus, nephritis, leukemia,
55 hypertension and hyperhidrosis, or as health-promoting foodstuffs (typically soups
56 and teas) to enhance the human immune system and to reinforce the body vital energy
57 ¹⁻⁴. However, the limited supply (endemic plant species for East Asia) together with
58 the over-exploitation constitutes the most important hurdles for developing Radix
59 astragali-based drugs or dietary supplements ^{5,6}. To address these issues, it has
60 become feasible to use plant *in vitro* cultures for the large-scale production of valuable
61 phytochemicals from this species ⁴. In this context, *A. membranaceus* hairy roots
62 cultures (AMHRCs) generated from the genetic transformation of *Agrobacterium*
63 *rhizogenes*, have emerged as attractive alternatives to produce active compounds
64 effectively, economically and in an environmentally friendly way ⁶⁻⁹.

65 Astragalosides and isoflavonoids, the principal active ingredients of Radix
66 astragali, possess versatile biological activities as diverse as anti-tumor, cardiogenic,
67 immunomodulatory, neuroprotective, antiviral, antioxidant, anti-inflammatory,
68 antiperspirant, hepatoprotective, antihypertensive, anti-fatigue, anti-osteoarthritis
69 effects, etc. ^{4,10-12}. Accordingly, it is necessary to establish a valid analytical method
70 to characterize the major pharmaceuticals (astragalosides and isoflavonoids) and
71 efficiently control the quality of AMHRCs. It is known the fact that the analysis of
72 metabolites in plant *in vitro* cultures is a challenging task because of their chemical
73 diversity, usually low abundance and variability within different cell/organ lines ¹³.
74 Liquid chromatography-mass spectrometry (LC-MS/MS) offers excellent sensitivity
75 and selectivity, combined with the ability to elucidate or confirm chemical structures

76 of target constituents in complex biological samples based on their exact MS/MS
77 fragment patterns^{14,15}. Sample preparation is still the most tedious and
78 time-consuming step, which is recognized as the main bottleneck of the analytical
79 process¹⁶.

80 Extraction of secondary metabolites from plant cell/organ cultures is always
81 limited by their high water contents. Conventional sample preparation methods such
82 as soxhlet extraction (SE) and ultrasound-assisted extraction (UAE), are available for
83 phytochemicals extraction from plant *in vitro* cultures, but they generally require high
84 energy consumption for dewatering and drying pretreatment along with long duration
85 and low efficiency^{17,18}. Therefore, development of new sample preparation strategies
86 for HRCs that eliminate biomass drying and enhance extraction efficiency can lead to
87 significant energy and cost savings. Moreover, selection of a particular sample
88 preparation method should depend on the simplicity of the extraction technique and
89 its convenience. High speed homogenization (HSH) has been an effective sample
90 pretreatment technique, which can facilitate the destruction of fresh plant materials for
91 a better access to intracellular substances^{19,20}. In addition, cavitation-accelerated
92 extraction (CAE) developed by our laboratory, is a simple, environmentally friendly
93 and efficient technology, which has been successfully applied in the extraction and
94 analysis of bioactive ingredients from several medicinal plants^{21,22}. However, CAE
95 has never been used for the extraction of phytochemicals from plant *in vitro* cultures.

96 In this work, HSH coupled with CAE (HSH-CAE) followed by LC-MS/MS was
97 proposed for the direct determination of six astragalosides and five isoflavonoids in
98 AMHRCs. The diagram of work flow is shown in Fig. 1. Various influential
99 parameters of the proposed sample preparation method were optimized systematically.
100 Subsequently, the superiority of HSH-CAE was evaluated as compared to

101 conventional methods in terms of extraction efficiency and green aspects. Moreover,
102 cytohistological studies of samples before and after extraction were performed to
103 clarify the extraction mechanism. Furthermore, a sensitive and accurate LC-MS/MS
104 method with selected reaction monitoring (SRM) model for simultaneous
105 quali-quantitative analysis of eleven target compounds was successfully established.
106 The method validation of the proposed approach was also investigated. Eventually,
107 the developed analytical method was performed for screening high-productive
108 AMHRCs among eight candidates.

109 **2. Materials and methods**

110 **2.1. Materials and reagents**

111 Eight *A. membranaceus* hairy root lines (I–VIII) were successfully induced via
112 the genetic transformation of *A. rhizogenes* LBA9402 in our laboratory. Eight
113 AMHRCs (I–VIII) were initiated by culturing 1.5 g (fresh weight, FW) of different
114 hairy root lines in 250 mL Erlenmeyer's flasks containing 150 mL of Murashige and
115 Skoog (MS)-based liquid medium (pH 5.8) supplemented with 30 g L⁻¹ sucrose and 1
116 g L⁻¹ casein hydrolyzate but without NH₄NO₃, and incubated on a rotary shaker (100
117 rpm) in the dark at 25 ± 1 °C. AMHRCs were harvested by filtration after 4 weeks of
118 cultivation, and then rinsed with tap and distilled water. Moisture contents of eight
119 AMHRCs (I–VIII) were pre-determined for the further quantitative analysis.

120 Astragalosides and isoflavonoids standards including including astragaloside I
121 (AG I), astragaloside II (AG II), isoastragaloside II (IAG II), astragaloside III (AG
122 III), astragaloside IV (AG IV), cycloastragenol (CY), calycosin-7-O-β-D-glucoside
123 (CAG), ononin (ON), astraisoflavan-7-O-β-D-glucoside (ASG), calycosin (CA) and
124 formononetin (FO) were purchased from Weikeqi Biological Technology Co. Ltd.

125 (Sichuan province, China). Other reagents of either analytical or optical grade were
126 obtained from Beijing Chemical Reagents Co. (Beijing, China).

127 **2.2. HSH-CAE procedure**

128 **2.2.1. Extraction process**

129 The HSH-CAE instrument was designed and manufactured by our laboratory. As
130 the diagram illustrated in Fig. 1, the apparatus consists mainly of a high speed
131 homogenizer (A), a cavitation chamber (B) and a vacuum pump (C). During the
132 extraction process, the fresh AMHRCs VI (5.0 g, FW) were initially added into the
133 homogenizer from an inlet (1) while all the valves were kept closed. After HSH
134 treatment, valves (2) and (3) and the vacuum pump were turned on successively.
135 Meanwhile, the homogenates and extraction solvents were introduced automatically
136 into the cavitation chamber by the generated negative pressure. Subsequently, the
137 valve (2) was turned off but the valve (4) was turned on, and then the continuous air
138 flow was introduced into the cavitation chamber for CAE process. The vacuum degree
139 of the extraction system was monitored by the throttling gauge (6) and the pressure
140 gauge (7). Additionally, a sieve plate set in the bottom of the cavitation chamber was
141 utilized to generate cavitations with different intensities and characteristics. After the
142 extraction, valves (3) and (4) and the vacuum pump were turned off successively. The
143 vacuum in the cavitation chamber was released by adjusting the valve (2). Eventually,
144 the extraction solvent was filtered through the sieve plate under gravity, and collected
145 via the valve (5). The obtained solution was then centrifuged and filtered through a
146 0.22 μm nylon membrane for LC-MS/MS analysis.

147 **2.2.2. Experimental design**

148 To achieve the optimum efficiency by HSH-CAE, Box-Behnken design (BBD)²³
149 was applied to survey the effect of four key independent variables at three levels

150 (negative pressure -0.06 to -0.09 MPa, homogenization time 30 to 60 s, liquid/solid
151 ratio 3 to 8, and extraction time 10 to 30 min) on the dependent variable (the sum
152 yields of AG I, AG II, IAG II, AG III, AG IV, CY, CAG, ON, ASG, CA and FO).
153 Liquid/solid ratio was calculated based on the fresh weight of hairy roots. The actual
154 and coded levels of the independent variables used in the experimental design are
155 summarized in [Table S1](#). The experiment data were analyzed statistically with
156 Design-Expert 7.0 software (State-Ease, Inc., Minneapolis MN). Analyses of variance
157 (ANOVA) were performed to calculate and simulate the optimal values of the tested
158 parameters.

159 **2.3. Conventional procedures**

160 The washed AMHRCs VI were dried in a vacuum drier at 60 °C till a constant
161 weight, and then the dry materials were ground to fine powders and extracted by the
162 reported soxhlet extraction (SE) and ultrasound-assisted extraction (UAE) methods
163 with slight modifications^{17, 18}. For SE: root powders (0.5 g) were placed in a Soxhlet
164 apparatus and extracted with 80% ethanol solution (25 mL) at 90 °C for 4 h. For UAE:
165 root powders (0.5 g) were extracted with 80% ethanol solution (25 mL) in an
166 ultrasonic bath (KQ-250DB, Kun-shan Ultrasonic Instrument Co. Ltd., China) for 120
167 min. After the extraction of each method, the obtained solutions were centrifuged and
168 filtered through a 0.22 µm nylon membrane for LC-MS/MS analysis.

169 **2.4. LC-MS/MS analysis**

170 An Agilent 1100 series HPLC (Agilent, San Jose, USA) coupled to an API 3000
171 triple tandem quadrupole MS (Applied Biosystems, Concord, Canada) equipped with
172 a Phenomenex Gemini C18 110A reversed-phase column (250 mm × 4.6 mm I.D., 5
173 µm) was applied for the analysis of target compounds from AMHRCs. The binary
174 mobile phase consisted of acetonitrile (A) and 0.01% formic acid aqueous solution (B)

175 using the gradient program as follows: 0–3 min, 45% (A); 3–8 min, 45–50% (A);
176 8–13 min, 50–60% (A); 13–18 min, 60–65% (A); and 18–20 min, 65–45% (A). The
177 column temperature was maintained at 30 °C, the flow rate 1.0 mL min⁻¹ and the
178 injection volume 10 µL.

179 All mass spectra of target analyses were acquired in SRM mode with
180 electrospray ionization (ESI) source operating in the negative ion mode. The universal
181 parameters were as follows: nebulising gas, curtain gas and collision gas set as 12, 10
182 and 6 a.u. (arbitrary units), respectively; ion source temperature 300 °C; ion spray
183 voltage -4500 V; focusing potential and entrance potential set as -75 and -10 V,
184 respectively. To obtain the highest response for each analyte, the specific parameters
185 for acquiring the optimal precursor/product ion combinations including declustering
186 potential (DP), collision energy (CE) and collision cell exit potential (CXP) were
187 optimized and summarized in [Table 1](#). Contents of target compounds were calculated
188 by corresponding calibration curves based on the dry weight of roots. For fresh
189 AMHRCs, the dry weights were obtained through converting fresh weights by the aid
190 of their moisture contents.

191 **2.5. Statistical analysis**

192 All experiments in this work were conducted for three times. Results were
193 expressed as means ± standard deviations. The data were statistically analyzed using
194 the SPSS statistical software, version 17.0 (SPSS Inc, Chicago, Illinois, USA).
195 Differences between means were determined by analysis of variance (ANOVA) with
196 Duncan's test on the level of significance declared at $P < 0.05$.

197 **3. Results and discussion**

198 **3.1. Optimization of HSH-CAE conditions**

199 The rationale for application of the HSH pretreatment is various, the most

200 evident being the ability of a high speed homogenizer (12000 rpm) to handle fresh
201 plant materials in a continuous stream with turbulence, shear stress and friction, which
202 can affect morphological changes in plant matrix that enhances the following
203 extraction process^{19, 20}. Additionally, the cavitation of CAE in this work is generated
204 by negative pressure via vacuum pump, and air is introduced continuously into the
205 extraction vessel along with the collapse of bubbles that creates intensive
206 cavitation-collision, turbulence, suspension and interface effects for disrupting plant
207 cells and accelerating mass transfer^{21, 22}.

208 According to results of our preliminary experiments (data not shown), 80%
209 ethanol solution was adopted for the simultaneous extraction of six astragalosides and
210 five isoflavonoids. Considering that the numbers of experiments necessary for
211 optimizing extraction conditions can be reduced by statistical experimental design, the
212 homogenization time, negative pressure, liquid/solid ratio and extraction time on the
213 yield of total target analytes were optimized by BBD combined with response surface
214 methodology (RSM)²⁴. The experimental design matrix and all the related data are
215 illustrated in [Table S1](#).

216 **3.1.1 Fitting the mathematical model**

217 ANOVA results of the built quadratic model are presented in [Table S2](#). A highly
218 significant level of the model ($P < 0.0001$), a not significant “lack of fit” ($P > 0.05$)
219 and a desirable determination coefficient ($R^2 = 0.9843$) suggested that the built
220 mathematical model was precise and applicable. Moreover, the factors with the
221 significant effects ($P < 0.05$) on the dependent variable of the model were the linear
222 terms of X_1 , X_2 , X_3 and X_4 , interaction term of X_1X_4 , X_2X_4 and X_3X_4 , and quadratic
223 terms of X_1^2 , X_2^2 , X_3^2 and X_4^2 . However, the interaction terms of X_1X_3 , X_1X_3 and X_2X_3
224 were insignificant ($P > 0.05$) and could be negligible. The second-order polynomial

225 model was applied to express the extraction efficiency of total target analytes as the
226 following equation:

$$227 \quad Y = 2.79 - 1.11X_1 + 0.13X_2 - 0.092X_3 + 0.043X_4 + 0.065X_1X_4 - 0.08X_2X_4 \\ - 0.11X_3X_4 - 0.52X_1^2 - 0.095X_2^2 - 0.3X_3^2 - 0.067X_4^2$$

228 Where Y was the yield of total target analytes; X_1 was the negative pressure (MPa); X_2
229 was the homogenization time (s); X_3 was liquid/solid ratio (mL g^{-1}); and X_4 was the
230 extraction time (min).

231 3.1.2. Analysis of the response surface

232 As shown in Fig. 2A, both negative pressure and liquid/solid ratio exhibited
233 significantly double effect on the yield of target analytes, and it demonstrated that
234 negative pressure around -0.08 MPa and liquid/solid ratio around 5 resulted in high
235 recovery. As presented in Fig. 2B, the yield of target analytes increased with the
236 decrease of negative pressure from -0.06 to -0.08 MPa at a fixed homogenization time,
237 but decreased significantly afterwards. This phenomenon is closely related to the
238 cavitation effect of CAE that can promote the extraction of intracellular substances by
239 means of disrupting plant cells and accelerating mass transfer^{21,22}. Factually, the
240 cavitation effect increases initially with the decrease of negative pressure, but
241 diminishes expeditiously with the further decrease of negative pressure due to the lack
242 of air to form cavitation bubbles²¹. As exhibited in Fig. 2C, with increasing
243 liquid/solid ratio from 3 to 5 at a given extraction time, the yield of target analytes
244 increased accordingly, however, the further increase in liquid/solid ratio resulted in a
245 significant decrease in the recovery. The inadequate solvent can promote mass transfer
246 barrier as the distribution of intracellular compounds is concentrated in certain regions
247 which limits the movement of them out of cell matrix. However, the excessive solvent
248 can consume the cavitation energy of CAE thus resulting in the poor extraction

249 efficiency. It was observed from Fig. 2D that the yield of target analytes increased
250 with the extension of homogenization time from 30 to 54 s at a fixed liquid/solid ratio
251 but increased slightly afterwards. The extended homogenization time did not improve
252 extraction yield significantly as the target ingredients may have leached out from
253 sample matrices before extraction. As seen from Fig. 3E and F, at a given negative
254 pressure or homogenization time, the yield of target analytes increased with extended
255 time initially but stabilized beyond 22 min, which was likely ascribable to the
256 exhaustion of target compounds in AMHRCs.

257 **3.1.3. Verification of the predictive model**

258 Based on the mathematical model built, the optimal experimental parameters
259 were as follows: negative pressure -0.077 MPa, homogenization time 55.7 s,
260 liquid/solid ratio 5.21 and extraction time 19.28 min. Considering the actual operation,
261 the homogenization time, liquid/solid ratio and extraction time were modified slightly
262 to 56 s, 5.2 and 19.3 min, respectively. To validate the reliability of the theoretical
263 model prediction, three sequential experiments were performed under optimal
264 parameters. The yield of total target analytes was $2.81 \pm 0.09 \text{ mg g}^{-1}$ from the actual
265 experiments, which was a good fit for the value (2.85 mg g^{-1}) forecasted by the
266 regression model. Therefore, the optimal extraction conditions obtained were reliable
267 and practical.

268 **3.2. HSH-CAE predominance**

269 In the previous reports, conventional SE and UAE methods were utilized for the
270 extraction of phytochemicals from dried plant hairy root cultures^{17,18}. Therefore, the
271 superiority of the proposed HSH-CAE approach was evaluated against these
272 traditional methods in this work. The obvious advantages of HSH-CAE were mainly
273 reflected as follows: simplicity in operation (elimination of biomass drying and

274 grinding), highest efficiency (20.2 min as against 240 min of SE and 120 min of
275 UAE), improvable yield (2.81 mg g⁻¹ as against 2.69 mg g⁻¹ of SE and 2.57 mg g⁻¹ of
276 SE), lowest energy cost (0.021 kWh mg⁻¹ as against 15.94 kWh mg⁻¹ of SE and 15.32
277 kWh mg⁻¹ of UAE) and minimal CO₂ generation (0.017 Kg mg⁻¹ as against 12.75 Kg
278 mg⁻¹ of SE and 12.26 Kg mg⁻¹ for UAE). The energy consumption was determined
279 with a Wattmeter based on the extraction of 1 mg total target analytes. The calculation
280 of CO₂ ejected was made according to the previous literature ²⁵.

281 Conventional mechanical grinding technology for dried biomasses always causes
282 the local overheating of materials thus leading to the thermal degradation of some
283 susceptible compounds. However, HSH technique can effectively pulverize fresh
284 plant materials in a continuous slurry stream and avoid the localized increased
285 temperatures. In addition, traditional extraction methods always need long-term
286 heating at high temperature, which will result in the degradation of thermolabile
287 compounds. Conversely, CAE was performed with air flow at room temperature, thus,
288 the thermal degradation of sensitive analytes could be reduced or prevented. These
289 beneficial properties of HSH-CAE contributed to an increase in extraction yields of
290 target compounds. Consequently, the proposed HSH-CAE is a simple, low cost, green
291 and effective method for the direct and augmentation extraction of astragalosides and
292 isoflavonoids from fresh AMHRCs.

293 **3.3. HSH-CAE mechanism**

294 Since cell walls and membranes present formidable barriers to permeation by
295 extraction solvents, cells have to be disrupted prior to extraction. Herein, the
296 cytohistological analysis of AMHRCs samples before and after each procedure was
297 performed to clarify the HSH-CAE mechanism. The micrograph of untreated
298 AMHRCs is shown in Fig. 3A. After HSH (Fig. 3B), AMHRCs were obviously

299 dispersed from an intact organ into numerous cells (arrows). After CAE (Fig. 3C),
300 AMHRCs cells exhibited an evident rupture of cellular matrix (arrows), and a
301 movement of intracellular substances into the solvent media could be thus envisaged.
302 During extraction process, the intensive cavitation-collision effect along with the
303 collapse of bubbles can penetrate the surface of cellular matrix, thus resulting in the
304 effective disintegration of plant cells. The conclusive phenomenon in this study
305 suggested that HSH-CAE was an efficient hyphenated sample preparation technique
306 for the extraction of intracellular phytochemicals from plant *in vitro* cultures.

307 **3.4. Establishment of LC-MS/MS method**

308 **3.4.1. Optimization of LC conditions**

309 The fundamental basis of an efficient chromatographic process is the
310 development of a suitable mobile phase with an appropriate elution program, which
311 can achieve the best possible resolution and ionization of analytes in LC-MS/MS
312 analysis. According to our previous report, the acetonitrile-water mobile phase is the
313 best choice of separation and ionization of astragalosides and isoflavonoids^{8,9}.
314 Therefore, the mobile phase composition was simplified by using an
315 acetonitrile-water mixture for all analytes in this work. Additionally, the presence of
316 acid could improve the chromatographic behavior and reduce the peak tailing in the
317 present study. On the other hand, the addition of acid was not beneficial for the
318 deprotonation of analytes in the following ESI-MS/MS process with negative ion
319 mode. Taking into account the balance between the above two aspects, a solvent
320 system consisting of acetonitrile and 0.01% formic acid aqueous solution, which
321 could provide the satisfactory baseline stability and ionization efficiency, was
322 ultimately selected as mobile phase. Moreover, the developed gradient elution
323 program as described in Section 2.4 offered a short run time (20 min) and sufficient

324 resolution of eleven target analytes with very little matrix effects in the following
325 ESI-MS/MS analysis.

326 **3.4.2. Optimization of ESI-MS/MS parameters**

327 In our previous study, the ESI-MS/MS measurements of astragalosides in
328 positive ionization mode could provide higher response sensitivity as against those in
329 negative ionization mode⁸. However, the signals of isoflavonoids were hardly
330 detected in positive ionization mode but could be easily caught in negative ionization
331 mode⁹. For the simultaneous determination of all target analytes, the ESI-MS/MS
332 with negative-ion mode was eventually chosen to analyze the eleven target
333 compounds in spite of sacrificing the detection sensitivity of astragalosides. To obtain
334 the most informative fragmentation spectrum in ESI-MS/MS with SRM model,
335 several critical parameters including DP, CE and CXP on the signal intensities of all
336 target analytes were investigated systematically. Results of these parameters were
337 optimized manually, acquired and summarized in [Table 1](#).

338 Under the optimal conditions, the product ions mass spectra of CAG, ON, ASG,
339 AG IV, AG III, CA, AG II, IAG II, AG I, FO and CY are presented in [Fig. 4B–L](#),
340 respectively. The ESI-MS/MS analysis of CAG produced a precursor ion of m/z 445.4
341 ($[M-H]^-$), which produced a fragmentation pattern dominated by an ion at m/z 283.0
342 ($[(M-H-glucose)]^-$) ([Fig. 4B](#)). Therefore, the mass transition pattern m/z 445.4 \rightarrow 283.0
343 with the highest intensity was chosen for the identification and quantification of CAG.
344 Likewise, the SRM transitions at m/z 371.0 \rightarrow 356.1, m/z 429.1 \rightarrow 266.9, m/z
345 463.3 \rightarrow 301.1, m/z 783.6 \rightarrow 160.9, m/z 283.1 \rightarrow 268.0, m/z 825.6 \rightarrow 765.6, m/z
346 868.7 \rightarrow 807.4, m/z 267.1 \rightarrow 252.0, m/z 489.6 \rightarrow 383.3 and m/z 463.3 \rightarrow 301.1 were
347 selected for monitoring ON, ASG, AG IV & AG III, CA, AG II & IAG II, AG I, FO
348 and CY, respectively. The representative total ion chromatogram with SRM model of

349 standard mixture is shown in Fig. 4A. Obviously, the established LC-MS/MS method
350 achieved a rapid separation of all target compounds without sacrificing resolution.

351 **3.5. Method validation**

352 As the results summarized in Table 2, all calibration curves exhibited an
353 excellent linearity ($R^2 \geq 0.9991$) within the range of tested concentrations. Limit of
354 detections (LODs) for all target analytes were less than 1.77 ng mL^{-1} . Relative
355 standard deviations (RSDs) of intra- and inter-day measurements for the retention
356 time of all target analytes were less than 0.58% and 0.84%, respectively, and for the
357 peak area were less than 3.88% and 6.01%, respectively. Recoveries of standard
358 additions of all target analytes were ranging from 96.84% to 104.76%. Overall, the
359 aforementioned data indicated that the present method possessed good accuracy and
360 sensitivity for the quantification of target astragalosides and isoflavonoids in
361 AMHRCs.

362 **3.6. Application for screening high-productive AMHRCs**

363 Owing to the uncertainty of *A. rhizogenes* T-DNA integration into the host plant
364 genome, different *A. membranaceus* hairy root lines derived often show considerable
365 diverse biosynthesis patterns of secondary metabolites²⁶. Therefore, the proposed
366 HSH-CAE-LC-MS/MS method was applied for the determination of target
367 astragalosides and isoflavonoids in eight candidate AMHRCs (I–VIII) originated from
368 distinct hairy root lines. Quantitative results of eleven target analytes in different
369 cultures are presented in Table 3.

370 Obviously, various AMHRCs showed variations in contents of AG I, AG II, IAG
371 II, AG III, AG IV, CY, CAG, ON, ASG, CA and FO within the ranges of
372 914.22–1275.39, 564.61–707.61, 171.42–211.69, 151.65–198.55, 158.76–179.88,
373 13.94–17.69, 6.37–10.52, 4.18–6.07, 44.56–68.27, 63.97–97.26 and 51.89–65.93, μg

374 g^{-1} , respectively. Among them, the levels of astragalosides (AG I, AG II, IAG II, AG
375 III, AG IV and CY) in AMHRCs VI were significantly higher as compared to other
376 candidates, while AMHRCs II was categorized as the high-productive culture in terms
377 of isoflavonoids (CAG, ON, ASG, CA and FO) accumulation. The representative
378 LC-MS/MS total ion chromatogram of AMHRCs VI sample, and the corresponding
379 extracted ion chromatograms of CAG, ON, ASG, AG IV & AG III, CA, AG II & IAG
380 II, AG I, FO and CY are shown in Fig. 5A–J, respectively. This successful application
381 example indicated that the proposed analytical method is suitable for quality control
382 of AMHRCs or other plant *in vitro* cultures.

383 4. Conclusions

384 In the present study, a rapid, green and effective sample preparation and
385 analytical procedure for fresh plant *in vitro* cultures, i.e. HSH-CAE method followed
386 by LC-MS/MS detection, was developed and validated for the simultaneous
387 determination of six astragalosides (AG I, AG II, IAG II, AG III, AG IV and CY) and
388 five isoflavonoids (CAG, ON, ASG, CA and FO) in AMHRCs. Operational
389 conditions of HSH-CAE were optimized systematically. Compared with reported SE
390 and UAE methods, the proposed approach exhibited predominance of easy
391 manipulation, time-saving, high yield, low energy consumption and reduced waste.
392 Cytohistological investigations provided evidences of pronounced tissular/cellular
393 damages within HSH-CAE process. Moreover, the established LC-MS/MS method
394 was proved to have excellent linearity, precision, repeatability and reproducibility.
395 The validated HSH-CAE-LC-MS/MS method was also successfully applied for the
396 selection of high-productive AMHRCs. The observed beneficial effects exerted by the
397 proposed method in this work were valuable for the rapid and valid determination of
398 secondary metabolic profiles in AMHRCs or other plant *in vitro* cultures.

399

400 Acknowledgements

401 The authors gratefully acknowledge the financial supports by Application
402 Technology Research and Development Program of Harbin (2013AA3BS014),
403 Fundamental Research Funds for the Central Universities (2572014AA06),
404 Fundamental Research Funds for the Central Universities (2572015AA14), Special
405 Fund of National Natural Science Foundation of China (31270618), Key Program of
406 National Natural Science Foundation of China (81274010), and Heilongjiang
407 Province Outstanding Youth Fund (JC201101).

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475 **Figure Captions:**

476

477 **Fig. 1.** The work diagram of HSH-CAE-LC-MS/MS procedure.

478

479 **Fig. 2.** Response surfaces for target analytes extraction during HSH-CAE process: (A)
480 varying negative pressure and liquid/solid ratio; (B) varying negative pressure and
481 homogenization time; (C) varying liquid/solid ratio and extraction time; (D) varying
482 homogenization time and liquid/solid ratio; (E) varying extraction time and negative
483 pressure; (F) varying extraction time and homogenization time.

484

485 **Fig. 3.** Microscopic photographs of AMHRCs samples before extraction (A), after
486 HSH (B), and after CAE (C). Microscopic photographs were acquired via a light
487 microscope (Leica DM 4000B) equipped with a digital camera (Nikon DS-Ri1).

488

489 **Fig. 4.** LC-MS/MS total ion chromatogram with SRM model of standard mixture (A),
490 and the product ion mass spectra of CAG (B), ON (C), ASG (D), AG IV & AG III (E),
491 CA (F), AG II & IAG II (G), AG I (H), FO (I) and CY (J). The elution order of target
492 compounds as follows: 1. CAG, 2. ON, 3. ASG, 4. AG IV, 5. AG III, 6. CA, 7. AG II, 8.
493 IAG II, 9. AG I, 10. FO, 11. CY.

494

495 **Fig. 5.** LC-MS/MS total ion chromatogram with SRM model of AMHRCs sample (A),
496 and the corresponding extracted ion chromatograms of CAG (B), ON (C), ASG (D),
497 AG IV & AG III (E), CA (F), AG II & IAG II (G), AG I (H), FO (I) and CY (J). The
498 elution order of target compounds as follows: 1. CAG, 2. ON, 3. ASG, 4. AG IV, 5.
499 AG III, 6. CA, 7. AG II, 8. IAG II, 9. AG I, 10. FO, 11. CY.

Table 1. Optimized MS/MS parameters for the eleven target analytes^a.

Analytes	DP (V)	CE (V)	CXP (V)	SRM (amu)
AG I	-80	-34	-5	867.7 → 807.4
AG II & IAG II	-124	-36	-13	825.4 → 765.5
AG III & AG IV	-80	-54	-9	783.6 → 106.9
CY	-195	-40	-12	489.3 → 383.3
CAG	-20	-10	-5	445.2 → 283.0
ON	-31	-10	-5	428.8 → 266.9
ASG	-58	-23	-5	463.2 → 301.1
CA	-70	-24	-5	283.0 → 268.0
FO	-55	-31	-5	267.0 → 252.0

^a Other parameters: nebulising gas, curtain gas and collision gas set as 12, 10 and 6 a.u., respectively; ion source temperature 300 °C; ion spray voltage -4500 V; focusing potential and entrance potential set as -75 and -10 V, respectively.

Table 2. Calibration curves, LODs, precision and accuracy (recovery of standard addition) for the eleven target compounds as determined by the developed HSH-CAE-LC-MS/MS method.

Analytes	Calibration curves ^a	R^2	Linear range (ng mL ⁻¹)	LOD ^b (ng mL ⁻¹)	Intra-day RSD (%) ^c		Inter-day RSD (%)		Standard addition recovery (%) ^f Mean \pm RSD (n=3)
					Rt ^d	PA ^e	Rt	PA	
AG I	$Y = 986 X + 2570$	0.9991	8.64–8640	0.48	0.27	1.91	0.39	5.22	98.13 \pm 2.59
AG II	$Y = 263 X + 382$	1.0000	9.03–9030	0.86	0.38	2.35	0.21	4.96	99.51 \pm 3.07
IAG II	$Y = 212 X - 108$	1.0000	9.73–9730	1.50	0.16	3.88	0.37	6.01	103.27 \pm 1.83
AG III	$Y = 188 X + 656$	0.9999	9.36–9360	0.86	0.22	1.07	0.69	3.75	96.84 \pm 2.91
AG IV	$Y = 42.6 X + 293$	1.0000	8.82–8820	1.02	0.44	1.12	0.57	3.89	101.31 \pm 1.64
CY	$Y = 149 X - 175$	0.9996	8.36–8360	1.77	0.53	3.49	0.40	4.73	104.76 \pm 3.29
CAG	$Y = 1640 X + 14860$	1.0000	9.55–9550	0.11	0.29	2.77	0.73	5.15	99.02 \pm 2.43
ON	$Y = 6410 X + 32400$	1.0000	8.45–8450	0.08	0.58	3.02	0.65	5.86	97.87 \pm 1.52
ASG	$Y = 9900 X + 14600$	1.0000	8.91–8910	0.06	0.35	1.83	0.44	2.91	101.31 \pm 1.18
CA	$Y = 22500 X + 319000$	1.0000	9.01–9010	0.02	0.19	3.10	0.57	5.39	97.14 \pm 2.03
FO	$Y = 63700 X + 262000$	1.0000	8.82–8820	0.01	0.41	3.75	0.84	5.28	102.39 \pm 1.76

^a The calibration curves were constructed by plotting the peak areas versus the concentration (ng mL⁻¹) of each analyte, and each regression equation included eight data points; ^b LOD refers to the limit of detection; ^c RSD (%) = (SD/mean) \times 100; ^d Rt refers to the retention time for each analyte in the present LC-MS/MS method; ^e PA refers to the peak area for each analyte in the present LC-MS/MS method; ^f Three different spiking levels (10.57, 21.65 and 30.81 μ g g⁻¹ for AG I; 11.03, 20.79 and 33.66 μ g g⁻¹ for AG II; 9.79, 19.92 and 31.04 μ g g⁻¹ for IAG II; 11.46, 23.39 and 33.51 μ g g⁻¹ for AG III; 10.84, 20.73 and 29.56 μ g g⁻¹ for AG IV; 11.57, 23.84 and 33.25 μ g g⁻¹ for CY; 10.39, 22.55 and 31.26 μ g g⁻¹ for CAG; 9.61, 22.80 and 30.49 μ g g⁻¹ for ON; 11.97, 19.63 and 28.46 μ g g⁻¹ for ASG; 9.33, 20.86 and 30.72 μ g g⁻¹ for CA; and 10.65, 21.07 and 29.16 μ g g⁻¹ for FO) were applied in the standard addition recovery study; and the data are presented as the average of three determinations, where standard addition recovery (%) = (amount found – original amount)/amount spiked \times 100.

Table 3. Analysis of astragalosides and isoflavonoids in different AMHRCs via the developed HSH-CAE-LC-MS/MS method^a.

AMHRCs types ^b	Contents of astragalosides in AMHRCs ($\mu\text{g g}^{-1}$) ^c						Contents of isoflavonoids in AMHRCs ($\mu\text{g g}^{-1}$)				
	AG I	AG II	IAG II	AG III	AG IV	CY	CAG	ON	ASG	CA	FO
AMHRCs I	927.35 \pm 21.37	564.61 \pm 14.88	182.33 \pm 9.63	162.57 \pm 8.52	166.72 \pm 7.44	13.94 \pm 1.15	6.37 \pm 0.41	4.79 \pm 0.58	44.56 \pm 5.77	65.53 \pm 2.28	51.89 \pm 3.99
AMHRCs II	1216.83 \pm 33.05	637.51 \pm 17.52	171.42 \pm 11.23	158.69 \pm 11.66	163.96 \pm 9.93	15.84 \pm 0.94	10.52 \pm 0.66	5.79 \pm 0.13	68.27 \pm 3.72	97.26 \pm 5.64	65.93 \pm 1.57
AMHRCs III	914.22 \pm 28.69	567.56 \pm 23.35	185.19 \pm 8.45	155.84 \pm 14.74	165.55 \pm 13.50	14.67 \pm 1.33	7.58 \pm 0.38	5.03 \pm 0.27	44.91 \pm 5.40	69.84 \pm 8.13	57.62 \pm 5.46
AMHRCs IV	933.11 \pm 19.40	571.23 \pm 29.88	176.87 \pm 15.02	163.95 \pm 13.91	168.64 \pm 10.67	15.33 \pm 1.62	7.94 \pm 0.72	4.88 \pm 0.50	47.32 \pm 1.67	68.15 \pm 5.43	53.61 \pm 4.02
AMHRCs V	949.51 \pm 32.33	593.88 \pm 12.79	195.03 \pm 15.17	165.49 \pm 7.26	158.76 \pm 13.89	16.77 \pm 1.29	8.45 \pm 0.45	5.54 \pm 0.22	53.82 \pm 3.51	75.09 \pm 4.60	58.71 \pm 3.15
AMHRCs VI	1275.39 \pm 51.27	707.61 \pm 25.03	203.72 \pm 9.01	198.55 \pm 13.23	179.88 \pm 16.09	17.63 \pm 1.87	9.51 \pm 0.43	6.07 \pm 0.19	60.92 \pm 2.53	88.73 \pm 1.99	63.80 \pm 3.34
AMHRCs VII	1231.06 \pm 43.94	695.82 \pm 31.17	211.69 \pm 16.83	180.23 \pm 14.19	162.46 \pm 11.68	17.69 \pm 2.03	9.78 \pm 0.96	5.93 \pm 0.44	64.76 \pm 4.32	91.35 \pm 3.85	61.62 \pm 4.73
AMHRCs VIII	1184.22 \pm 35.52	610.75 \pm 18.93	184.26 \pm 10.75	151.65 \pm 8.33	166.43 \pm 7.27	15.29 \pm 1.41	6.49 \pm 0.63	4.18 \pm 0.35	49.28 \pm 2.27	63.97 \pm 4.41	56.35 \pm 2.13

^aOperational conditions of HSH-CAE were performed as follows: extraction solvent 80% ethanol, homogenization time 56 s, negative pressure -0.077 MPa, liquid/solid ratio 5.2 and extraction time 19.3 min; ^bEight candidate AMHRCs (I–VIII) were originated from eight distinct hairy root lines; ^cThe contents of analytes were calculated based on dry weights of AMHRCs which were obtained through converting fresh weights by the air-drying of their moisture contents.

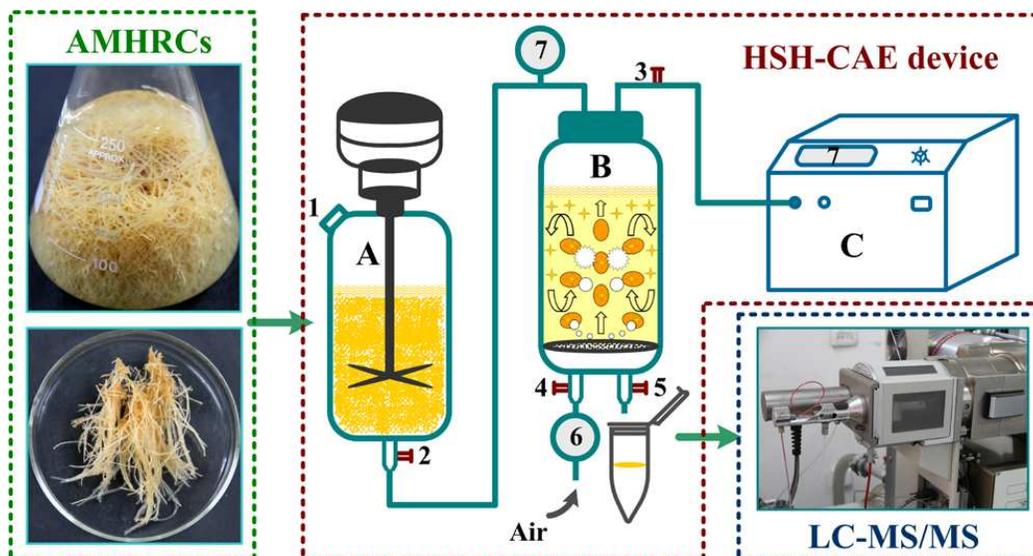


Fig. 1, Jiao et al.

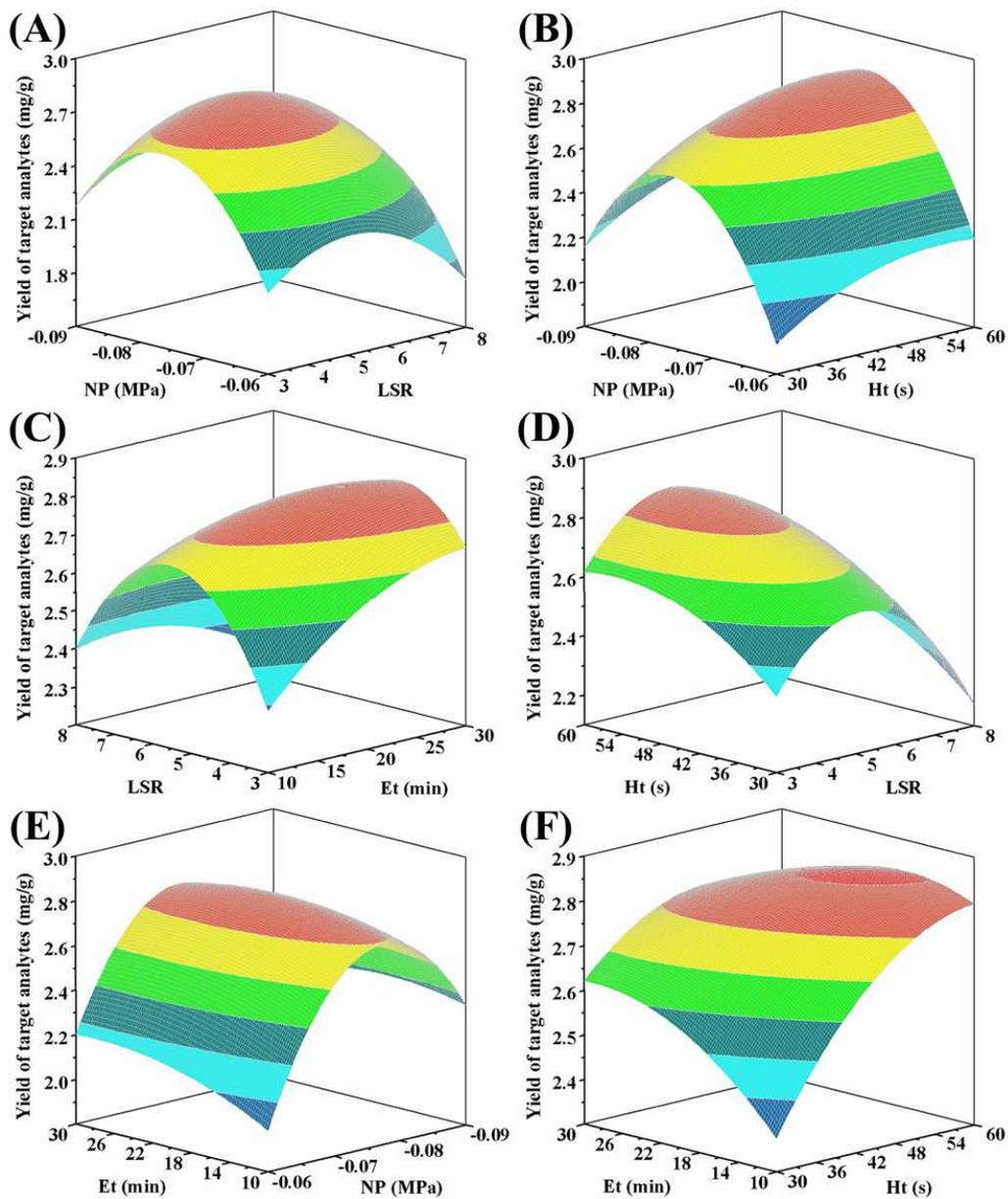


Fig. 2, Jiao et al.

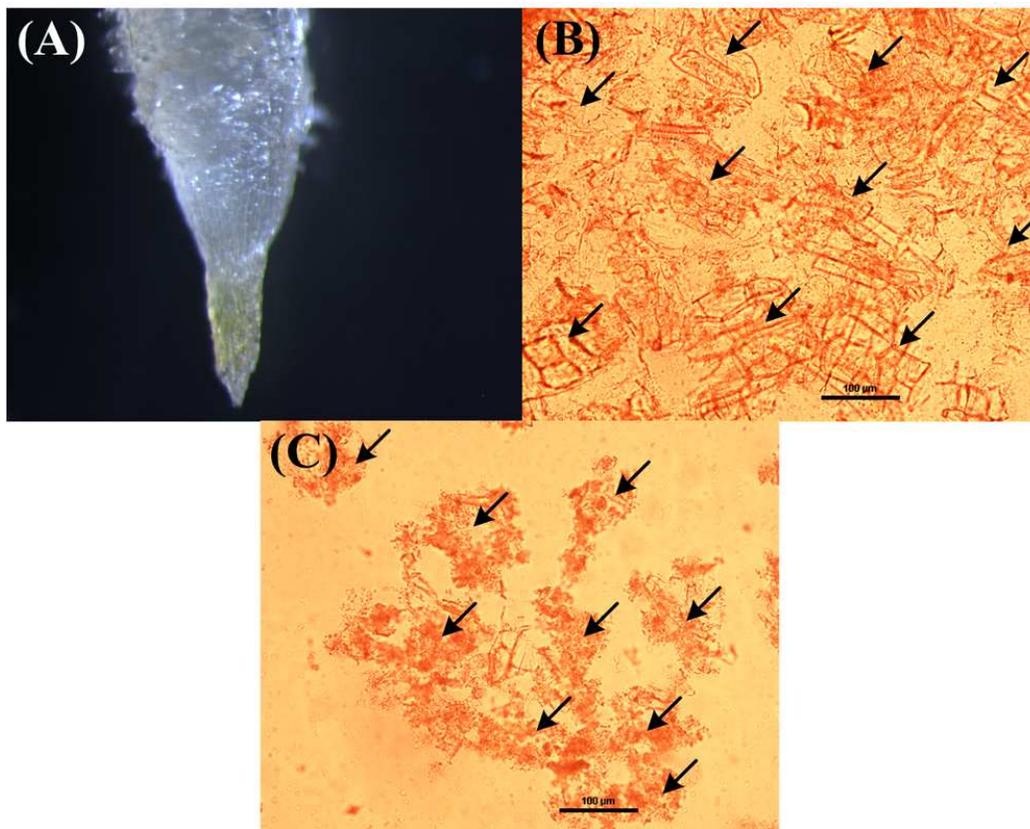


Fig. 3, Jiao et al.

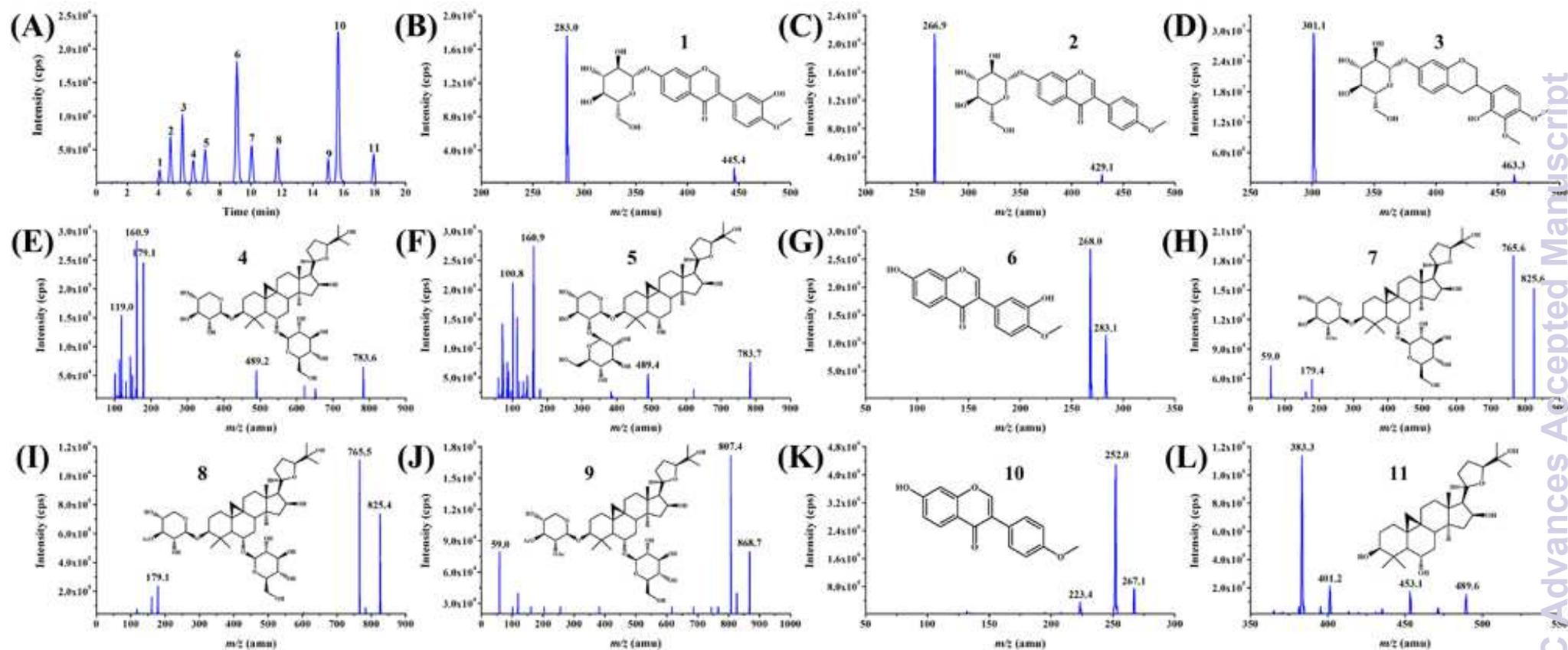


Fig. 4, Jiao et al.

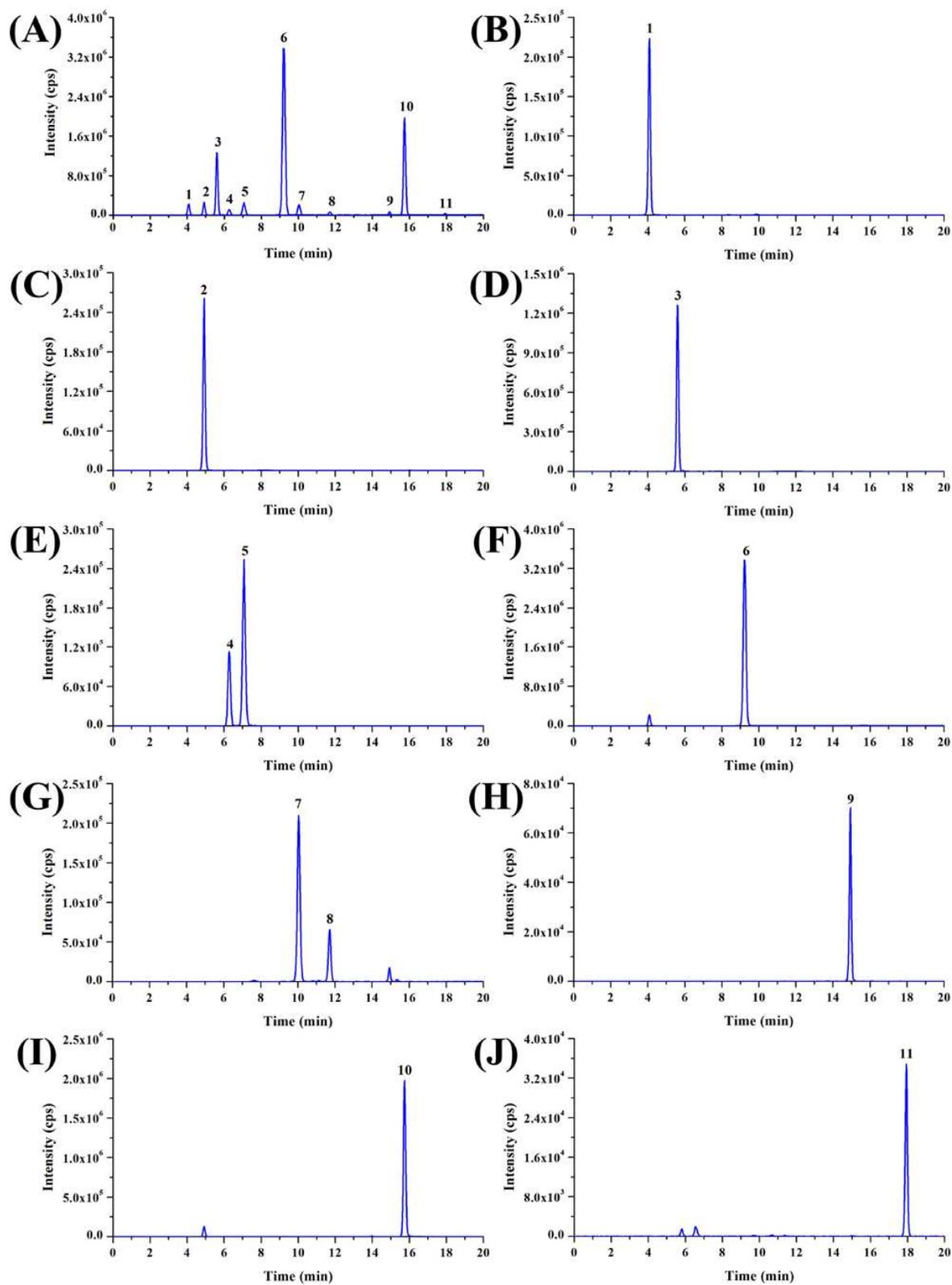


Fig. 5, Jiao et al.