# **RSC Advances**



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. This Accepted Manuscript will be replaced by the edited, formatted and paginated article as soon as this is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/advances

## Journal Name

### **RSCPublishing**

### COMMUNICATION

Cite this: DOI: 10.1039/xoxxooooox

# Yb(OTf)<sub>3</sub> catalyzed [3+2] annulations of D–A cyclopropanes with $\beta$ -oxodithioesters: a regioselective synthesis of tetrahydrothiophenes

Received ooth January 2012, Accepted ooth January 2012 Shu-Wen Wang, Wei-Si Guo, Li-Rong Wen,\* and Ming Li\*

DOI: 10.1039/x0xx00000x

www.rsc.org/

# A new route to prepare tetrahydrothiophene derivatives from D–A cyclopropanes and $\beta$ -oxodithioesters catalyzed by Yb(OTf)<sub>3</sub> is reported. This is the first example using $\beta$ -oxodithioesters as dipolarophiles to react with D–A cyclopropanes. The method exhibits good regioselectivity.

The methodologies for C–S bond-forming have received extensive attention in transition-metal-catalyzed cross-coupling reactions<sup>1</sup> and metal-free radical-coupling reactions.<sup>2</sup> Organosulfur compounds play important roles in medicinal chemistry.<sup>3</sup> Among them, tetrahydrothiophenes are representative derivatives due to their important biological activities,<sup>4</sup> and as catalysts in asymmetric synthesis.<sup>5</sup> Although a variety of protocols have been reported by a number of organic or pharmaceutical chemists,<sup>6</sup> the development of novel and efficient methods to construct tetrahydrothiophenes with readily available starting materials and catalyst is still in high demand.

Recently, functionalized S,S-ketene acetals<sup>7</sup> and  $\beta$ -oxodithioesters (ODEs)<sup>8</sup> have received much attention in construction of heterocycles. ODEs have shown various chemical properties with intriguing five reactive centers. Three nucleophilic centers localize on the oxygen atom, sulphur atom and  $\alpha$ -carbon, and two electrophilic centers present on the carbonyl and thiocarbonyl groups. Due to the high reactivity, the reactions of ODEs with various bifunctional reagents could construct diverse heterocyclic compounds.

Donor–acceptor cyclopropanes (D–A cyclopropanes) are versatile building blocks in organic synthesis due to their high reactivity.<sup>9</sup> In the last decade, the annulation of D–A cyclopropanes with various dipolarophiles has become a powerful strategy for the construction of carbo- and heterocyclic compounds. A variety of dipolarophiles, including imines, carbonyls, alkenes, nitriles, and nitrones, have been employed to react with D–A cyclopropanes to construct pyrrolidine,<sup>10</sup> tetrahydrofuran,<sup>11</sup> cyclopentane,<sup>12</sup> pyrroline,<sup>13</sup> and oxazine<sup>14</sup> derivatives. Additionally, the intramolecular [3+2] cycloaddition of functionalized D–A cyclopropanes to construct structurally diverse carbo- or heterocyclic skeletons have been studied systematically.<sup>15</sup> However, there is no report on using ODEs as dipolarophiles to react with D–A cyclopropanes.





Scheme 1 The comparison of previous reactions and our work

Recently, Stoltz and coworkers<sup>16</sup> developed a methodology to synthesize tetrahydrothiophene derivatives from D-A cyclopropanes and isothiocyanates using stoichiometric Sn(OTf)2 (Scheme 1). A similar reaction was reported by Yang and coworkers<sup>17</sup> using 2.0 equiv AlCl<sub>3</sub> as a promoter. The development of catalytic methods that lead to the tetrahydrothiophenes from D-A cyclopropanes remains a challenge to synthetic chemists. In continuation of our interests in  $\beta$ -oxodithioesters<sup>18</sup> and constructing heterocycles,<sup>19</sup> herein, we report a novel catalytic methodology to synthesize tetrahydrothiophenes using  $\beta$ -oxodithioesters with D-A cyclopropanes. Notably, the tetrahydrothiophene derivatives 3 were generated regioselectively through attacking the cyclopropanes by

the thiocarbonyl group of  $\beta$ -oxodithioesters, and the cyclopentanes **4** which formed by the  $\alpha$ -carbon attacking were not observed. <sup>1</sup>H NMR spectrum of compound **3b** shows a singlet peak at  $\delta$  7.41 ppm is subject to the olefin proton, and a dd peak at  $\delta$  4.75 ppm is assignable to C5-H of the tetrahydrothiophene due to the coupling with adjacent CH<sub>2</sub>. Furthermore, the structures of **3** were confirmed by X-ray crystallographic analysis of **3g** and **3j** (see Figures S1 and S2 in SI).

Table 1	Optimization	of reaction	conditions <sup>a</sup>
---------	--------------	-------------	-------------------------

Ph	O S SMe <sup>+</sup>	Ph 2a	Et <u>conditions</u> DEt	EtOOC Ph-O	COOEt		
Entry	LA (mol %)	Solvent	Temp (°C)	Time (h)	Yield $(\%)^b$		
1	Yb(OTf) <sub>3</sub> (10)	DCM	reflux	14	80		
2	Yb(OTf) <sub>3</sub> (10)	DCE	reflux	2	93		
3	Yb(OTf) <sub>3</sub> (10)	MeNO <sub>2</sub>	reflux	2	mixture		
4	Yb(OTf) <sub>3</sub> (10)	MeCN	reflux	2	91		
5	Yb(OTf) <sub>3</sub> (10)	THF	reflux	2	56		
6	Yb(OTf) <sub>3</sub> (10)	Toluene	reflux	2	86		
7	InCl <sub>3</sub> (10)	DCE	reflux	6	28		
8	$Cu(OTf)_2(10)$	DCE	reflux	6	mixture		
9	Sc(OTf) <sub>3</sub> (10)	DCE	reflux	6	75		
10	Yb(OTf) <sub>3</sub> (5)	DCE	reflux	2	93		
11	$Yb(OTf)_3(2)$	DCE	reflux	2	48		
12	$Yb(OTf)_3(5)$	DCE	60 °C	12	76		
13		DCE	reflux	12	trace		
<sup><i>a</i></sup> Reaction conditions: The mixture of <b>1a</b> (0.2 mmol), <b>2a</b> (0.2 mmol) and solvent (2 mL) was stirred in a 25 mL flask. <sup><i>b</i></sup> Isolated yield. <sup><i>c</i></sup> no catalyst.							

Methyl 3-oxo-3-phenylpropanedithioate 1a and diethyl 2phenylcyclopropane-1,1-dicarboxylate 2a were selected as model substrates for the reaction conditions optimization (Table 1). Initially, in the presence of Yb(OTf)<sub>3</sub>, substrate 1a was converted to the tetrahydrothiophene 3a in 80% yield (entry 1). Then different solvents were screening (entries 2-6), and product 3a was formed up to 93% yield when DCE was used (entry 2). Subsequently, other catalysts, including InCl<sub>3</sub>, Cu(OTf)<sub>2</sub> and Sc(OTf)<sub>3</sub>, were tested (entries 7-9). Disappointedly, all the yields of 3a were lower than Yb(OTf)<sub>3</sub>. Finally, the equivalent of the catalyst was also tested (entries 10-11). When the amount of Yb(OTf)<sub>3</sub> was decreased to 5 mol %, the yield of product 3a was not reduced (entry 10). However, the yield of **3a** was dropped dramatically with further decreasing the catalyst loading to 2 mol % (entry 11). Further optimization of the conditions revealed that the yield was reduced to 76% when lower the temperature to 60 °C (entry 12), and the reaction could not proceed without a catalyst (entry 13).

With the optimal conditions in hand, the scope of the reaction was examined using a broad range of substituted ODEs 1 and D–A cyclopropane 2a (Table 2). ODEs with either electron-donating or electron-withdrawing groups on the phenyl group ( $\mathbb{R}^1$ ) showed similar reactivity and the corresponding tetrahydrothiophenes were formed in high yields (3b-3f). Additionally, the position of substituents on phenyl ring of 1 did not have significant influence on

the yields (**3g-3i**). Moreover, furyl and thiophenyl substituted ODEs **1** were tolerated in the reaction and the products (**3j** and **3k**) were isolated in excellent yields. The tetrahydrothiophene **3l** was also obtained with aliphatic ODEs as reactant, albeit in lower yield.

**Table 2** Synthesis of tetrahydrothiophenes **3a-31** from various  $\beta$ -oxodithioesters  $\mathbf{1}^{a,b}$ 



<sup>a</sup>Reaction conditions: **1** (0.2 mmol), **2** (0.2 mmol), Yb(OTf)<sub>3</sub> (0.01 mmol), DCE (2 mL), 80 °C, 2 h. <sup>b</sup>Isolated yield.

**Table 3** Synthesis of tetrahydrothiophenes **3m-3u** from various D-A cyclopropanes  $2^{a,b}$ 



<sup>&</sup>lt;sup>a</sup>Reaction conditions: **1** (0.2 mmol), **2** (0.2 mmol), Yb(OTf)<sub>3</sub> (0.01 mmol), DCE (2 mL), 80  $\degree$ , 7 h. <sup>b</sup>Isolated yield.

The scope of D–A cyclopropanes **2** was also examined with  $\beta$ -oxodithioester **1a** under the optimal conditions (Table 3). The ester group R<sup>2</sup> has no effect on the yields of the reaction (**3m** compared

with **3a**). Subsequently, the cyclopropanes bearing 4-fluoro, 4-chloro, 4-bromo, 4-methyl and 4-methoxy groups on the phenyl ring were employed and the desired tetrahydrothiophenes (3n-3r) were afforded in good to excellent yields. Furthermore, excellent yields were obtained when substrates with 2-chloro and highly electron-deficient 4-nitro groups were investigated (3s-3t), exhibiting the generality of the method. Unfortunately, no target product **3u** was obtained when furan substituted cyclopropanes was used as reactant.



Scheme 2 Synthetic study towards the vinyl substituted tetrahydrothiophene.

Encouraged by the above results, the substrate scope was extended to vinyl cyclopropane 2v (Scheme 2). Disappointingly, the reaction exhibited poor selectivity, and the desired vinyl-substituted tetrahydrothiophene 3v was obtained in only 12% yield. The major product was acyclic compound 5.



Scheme 3 Proposed reaction mechanism.

Based on the above experimental results, a possible domino-ringopening-cyclization (DROC)<sup>20</sup> mechanism is proposed in Scheme 3. Initially, the cyclopropane was activated by intimate ion pair, which was generated from 1,3-dicarbonyl group with ytterbium(III) triflate. Then, as nucleophile, the thiocarbonyl group of  $\beta$ -oxodithioesters not the  $\alpha$ -carbon attacked the activated cyclopropane **A** selectively to form the intermediate **B**. Finally, the tetrahydrothiophene was obtained through intramoleculer cyclization, followed by MeSH elimination.

### Conclusions

In conclusion, a Yb(OTf)<sub>3</sub>-catalyzed [3+2] annulation reaction of D– A cyclopropanes with  $\beta$ -oxodithioesters has been developed. This methodology has the following advantages: (1) good regioselectivity; (2) catalytic Yb(OTf)<sub>3</sub>; (3) readily available starting materials; (4) mild reaction conditions. Undoubtedly, this novel reaction is complementary to the [3+2] cycloaddition using D-A cyclopropanes. Efforts to develop new reactions base on  $\beta$ -oxodithioesters are currently underway.

#### Acknowledgements

This work was financially supported by the National Natural Science Foundation of China (21372137 and 21072110) and the Natural Science Foundation of Shandong Province (ZR2012BM003 and ZR2014BM006).

#### Notes and references

State Key Laboratory Base of Eco-Chemical Engineering, College of Chemistry and Molecular Engineering, Qingdao University of Science and Technology, Qingdao 266042, P. R. China.

E-mail: wenlirong@qust.edu.cn; liming928@qust.edu.cn

<sup>†</sup> Electronic Supplementary Information (ESI) available: Experimental procedures, full spectroscopic data for all new compounds, and crystal data for **3g** and **3j** (CIF). See DOI: 10.1039/c000000x/

- For a general review, see: (a) C. F. Lee, Y. C. Liu and S. S. Badsara, *Chem. Asian. J.*, 2014, 9, 706; for selected examples, see: (b) F. Y. Kwong and S. L. Buchwald, *Org. Lett.*, 2002, 4, 3517. (c) M. A. Fernandez-Rodr guez, Q. Shen and J. F. Hartwig, *J. Am. Chem. Soc.*, 2006, 128, 2180; (d) J. F. Hartwig, *Acc. Chem. Res.*, 2008, 41, 1534; (e) L. Chu, X. Yue and F. L. Qing, *Org. Lett.*, 2010, 12, 1644; (f) L. H. Zou, J. Reball, J. Mottweiler and C. Bolm, *Chem. Commun.*, 2012, 48, 11307; (g) F. J. Chen, G. Liao, X. Li, J. Wu and B. F. Shi, *Org. Lett.*, 2014, 16, 5644; (h) P. F. Wang, X. Q. Wang, J. J. Dai, Y. S. Feng and H. J. Xu, *Org. Lett.*, 2014, 16, 4586.
- (a) R. Y. Tang, Y. X. Xie, Y. L. Xie, J. N. Xiang and J. H. Li, *Chem. Commun.*, 2011, **47**, 12867; (b) S. S. Badsara, Y. C. Liu, P. A. Hsieh, J. W. Zeng, S. Y. Lu, Y. W. Liu and C. F. Lee, *Chem. Commun.*, 2014, **50**, 11374; (c) B. Du, B. Jin and P. Sun, *Org. Lett.*, 2014, **16**, 3032 (d) Y. Luo, X. Pan, C. Chen, L. Yao and J. Wu, *Chem. Commun.*, 2015, **51**, 180.
- 3 E. A. Ilardi, E. Vitaku and J. T. Njardarson, J. Med. Chem., 2014, 57, 2832.
- 4 (a) M. W. Harrold, R. A. Wallace, T. Farooqui, L. J. Wallace, N. Uretsky and D. D. Miller, *J. Med. Chem.*, 1989, 32, 874; (b) T. Ohtsuka, H. Kotaki, N. Nakayama, Y. Itezono, N. Shimma, T. Kudoh, T. Kuwahara, M. Arisawa and K. Yokose, *J. Antibiot.*, 1993, 46, 11; (c) N. J. Zhang, M. Tomizawa and J. E. Casida, *J. Org. Chem.*, 2004, 69, 876. (d) L. S. Jeong, S. A. Choe, P. H. Gunaga, H. O. Kim, H. W. Lee, S. K. Lee, D. K. Tosh, A. Patel, K. K. Palaniappan, Z. G. Gao, K. A. Jacobson and H. R. Moon, *J. Med. Chem.*, 2007, 50, 3159; (e) K. Haraguchi, H. Shimada, H. Tanaka, T. Hamasaki, M. Baba, E. A. Gullen, G. E. Dutschman and Y. C. Cheng, *J. Med. Chem.*, 2008, 51, 1885.
- 5 (a) K. Julienne, P. Metzner, V. Henryon and A. Greiner, J. Org. Chem., 1998, 63, 4532; (b) J. Zanardi, C. Leriverend, D. Aubert, K.

Journal Name

Page 4 of 4

Julienne and P. Metzner, J. Org. Chem., 2001, **66**, 5620; (c) A. Piccinini, S. A. Kavanagh, P. B. Connon and S. J. Connon, Org. Lett., 2010, **12**, 608; (d) M. T. Huang, H. Y. Wu and R. J. Chein, Chem. Commun., 2014, **50**, 1101.

- 6 (a) A. M. Ponce and L. E. Overman, J. Am. Chem. Soc., 2000, 122, 8672; (b) S. Brandau, E. Maerten and K. A. Jørgensen, J. Am. Chem. Soc., 2006, 128, 14986; (c) F. J. Robertson and J. Wu, J. Am. Chem. Soc., 2012, 134, 2775; (d) J. B. Ling, Y. Su, H. L. Zhu, G. Y. Wang and P. F. Xu, Org. Lett., 2012, 14, 1090; (e) Y. Su, J. B. Ling, S. Zhang and P. F. Xu, J. Org. Chem., 2013, 78, 11053; (f) S. Meninno, G. Croce and A. Lattanzi, Org. Lett., 2013, 15, 3436; (g) J. Song, J. Moss, D. C. Yang, Z. Guan and Y. H. He, RSC Adv., 2014, 4, 54032.
- For a general review, see: (a) L. Pan, X. Bi and Q. Liu, *Chem. Soc. Rev.*, 2013, 42, 1251; for selected examples, see: (b) Z. Fang, J. Liu, Q. Liu and X. Bi, *Angew. Chem., Int. Ed.*, 2014, 53, 7209; (c) Y. Liu, B. D. Barry, H. Yu, J. Liu, P. Liao and X. Bi, *Org. Lett.*, 2013, 15, 2608.
- For a general review, see: (a) M. S. Singh, G. C. Nandi and T. Chanda, *RSC Adv.*, 2013, 3, 14183; for selected examples, see: (b) S. Chowdhury, T. Chanda, S. Koley, N. Anand and M. S. Singh, *Org. Lett.*, 2014, 16, 5536; (c) S. Chowdhury, T. Chanda, S. Koley, B. J. Ramulu, R. C. F. Jones and M. S. Singh, *Org. Lett.*, 2013, 15, 5386; (d) B. J. Ramulu, T. Chanda, S. Chowdhury, G. C. Nandi and M. S. Singh, *RSC Adv.*, 2013, 3, 5345; (e) R. K. Verma, G. K. Verma, G. Shukla, A. Nagaraju and M. S. Singh, *ACS Comb. Sci.*, 2012, 14, 224.
- For a general reviews, see: (a) H. K. Grover, M. R. Emmett and M. A. Kerr, *Org. Biomol. Chem.*, 2015, 13, 655; (b) T. F. Schneider, J. Kaschel and D. B. Werz, *Angew. Chem.*, *Int.* Ed., 2014, 53, 5504; (c) C. A. Carson and M. A. Kerr, *Chem. Soc. Rev.*, 2009, 38, 3051.
- (a) C. A. Carson and M. A. Kerr, *J. Org. Chem.*, 2005, **70**, 8242; (b)
  S. K. Jackson, A. Karadeolian, A. B. Driega and M. A. Kerr, *J. Am. Chem. Soc.*, 2008, **130**, 4196; (c) A. T. Parsons, A. G. Smith, A. J. Neel and J. S. Johnson, *J. Am. Chem. Soc.*, 2010, **132**, 9688.
- 11 (a) P. D. Pohlhaus and J. S. Johnson, J. Am. Chem. Soc., 2005, 127, 16014; (b) F. Benfatti, F. de Nanteuil and J. Waser, Org. Lett., 2012, 14, 386.
- 12 (a) X. Qi and J. Ready, *Angew. Chem., Int. Ed.*, 2008, 47, 7068; (b) H. Xiong, H. Xu, S. Liao, Z. Xie and Y. Tang, *J. Am. Chem. Soc.*, 2013, 135, 7851.
- (a) G. Sathishkannan, K. Srinivasan, Org. Lett., 2011, 13, 6002. (b) B.
   Cui, J. Ren and Z. Wang, J. Org. Chem., 2014, 79, 790.
- 14 (a) I. S. Young and M. A. Kerr, *Angew. Chem., Int. Ed.*, 2003, 42, 3023; (b) M. D. Ganton and M. A. Kerr, *J. Org. Chem.*, 2004, 69, 8554. (c) M. P. Sibi, Z. Ma and C. P. Jasperse, *J. Am. Chem. Soc.*, 2005, 127, 5764.
- (a) S. Xing, Y. Li, Z. Li, C.Liu, J. Ren and Z. Wang, Angew. Chem., Int. Ed., 2011, 50, 12605; (b) Y. Bai, W. Tao, J. Ren, Z. Wang, Angew. Chem., Int. Ed., 2012, 51, 4112; (c) W. Zhu, J. Fang, Y. Liu, J. Ren, Z. Wang, Angew. Chem., Int. Ed., 2012, 52, 2032.
- 16 A. F. G. Goldberg, N. R. O'Connor, R. A. Craig II and B. M. Stoltz, Org. Lett., 2012, 14, 5314.
- 17 Y. X. Sun, G. S. Yang, Z. Chai, X. L. Mu and J. Chai, Org. Biomol. Chem., 2013, 11, 7859.
- 18 L. R. Wen, Z. R. Li, M. Li and H. Cao, Green Chem., 2012, 14, 707.
- (a) L. R. Wen, S. L. Li, J. Zhang and M. Li, *Green Chem.*, 2015, 17, 1581;
   (b) W. S. Guo, S. L. Li, L. Tang, M. Li and L. R. Wen, *Org.*

Lett., 2015, **17**, 1232; (c) M. Li, X. L. Lv, L. R. Wen and Z. Q. Hu, Org. Lett., 2013, **15**, 1262; (d) L. R. Wen, L. B. Men, T. He, G. J. Ji and M. Li, Chem. Eur. J., 2014, **20**, 5028; (e) S. W. Wang, W. S. Guo, L. R. Wen and M. Li, RSC Adv., 2014, **4**, 59218.

(a) M. K. Ghorai and D. P. Tiwari, *J. Org. Chem.*, 2010, **75**, 6173; (b)
M. K. Ghorai, Y. Nanaji and A. K. Yadav, *Org. Lett.*, 2011, **13**, 4256;
(c) M. K. Ghorai and D. P. Tiwari, *J. Org. Chem.*, 2013, **78**, 2617; (d)
M. K. Ghorai, R. Talukdar and D. P. Tiwari, *Chem. Commun.*, 2013, **49**, 8205.