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# **Abstract**



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#### **1. Introduction**



 Among all these pretreatment methods, alkali pretreatment has been studied thoroughly and most used. According to Zhu *et al.*<sup>12</sup>, a alkali pretreatment step with 84 the NaOH load of 5% (ambient temperature  $(20 \pm 0.5 \text{ C})$  for 24h), improved the 85 biogas yield of corn stover for 37.0%. You *et al.*<sup>13</sup> reported a 34.59% higher biogas 86 production from corn stover after pretreatment with 6% NaOH at 35  $\degree$ C for 3 h. Though alkali pretreatment has been considered as efficient pretreatment method for lignocellulosic substrates, there are still some shortages in alkali pretreatment, the

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 chemicals required might lead to increasing cost and environmental problems. In addition, the sodium introduced during alkali pretreatment could be an inhibiting 91 factor of anaerobic digestion<sup>14</sup> and a problem for utilization of fermentation residue as fertilizer. These lead to the requirement for an eco-friendly and economically feasible pretreatment of lignocellulosic substrates for anaerobic digestion.

 Recent studies have demonstrated that hydrolysis also can be enhanced by introducing limited amounts of oxygen (or air) directly into the anaerobic digester or 96 during a pretreatment step  $^{15}$ . According to Mshandete *et al.*<sup>16</sup>, Nine hours microaerobic pretreatment of sisal pulp prior to anaerobic digestion demonstrated a 26% higher methane yield compared to the sisal pulp without pretreatment. when 99 treating the compound of brown water and food waste,  $Lim$  and  $Wang$ <sup>17</sup> reported 100 10-21% higher methane yield with an oxygen load of 37.5 ml O<sub>2</sub> L<sup>-1</sup><sub>R</sub> d<sup>-1</sup>. According 101 to Fu *et al.* <sup>18</sup>, a thermophilic microaerobic pretreatment process at the oxygen loads 102 of 5 ml/g VS<sub>substrate</sub> improved the methane yield of corn straw for 16.24%.

 Alkali pretreatment is a traditional pretreatment method for lignocellulosic substrates. However, thermophilic microaerobic pretreatment is a completely new pretreatment. No studies have been carried out to investigate the effect of thermophilic microaerobic pretreatment on the anaerobic digestion of sugarcane bagasse. In this study, the effects of thermophilic microaerobic and alkali pretreatment on the AD of sugarcane bagasse were investigated. In addition, the structure change during pretreatment and the fermentative characteristics (*e.g.* methane yield, T90, lag-phase time and VS removal efficiency *etc*) of alkali and thermophilic microaerobic pretreated sugarcane bagasse were compared.

- **2 Materials and methods**
- **2.1 Substrate and inoculum**

 Inoculum used in this study was anaerobic sludge, which was obtained from a local wastewater treatment plant (Tuandao Water Treatment Plant, Qingdao, 116 Shandong province, China), and stored in a 4  $\mathbb C$  refrigerator until further use. The total solid (TS) and volatile solid (VS) of inoculum are 4.67% and 70.60% (based on TS), respectively. Substrate used in this study was sugarcane bagasse, which was collected from a sugar factory in Hainan province of China. The TS and VS of substrate are 29.67% and 96.24% (based on TS), respectively.

# **2.2 Microaerobic pretreatment of sugarcane bagasse**

 Thermophilic microaerobic pretreatment of sugarcane bagasse was carried out in 300 ml serum bottles with a working volume of 150 ml in duplicates. In this stage, 22g sugarcane bagasse and 20ml inoculum were mixed in bottles, and then deionized 125 water was added to reach a total volume of 150 ml. Each bottle was flushed with  $N_2$  for 5 min to replace the air, and then the bottles were closed with rubber stoppers. 31.4, 62.8, and 125.6 ml of oxygen at atmospheric pressure was injected to each 128 group with a syringe to reach the oxygen loads of 5, 10, 20 ml/g  $VS_{substrate}$  (marked as 129 TMP1, TMP2, TMP3). The bottles were placed in a shaking water bath at 55 °C with 120 rpm. The oxygen levels were measured by a gas chromatograph (SP 6890, Shandong Lunan Inc., China) every 4 hours until the oxygen was consumed completely.

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### **2.3 Alkali pretreatment of sugarcane bagasse**

 Alkali pretreatment of sugarcane bagasse was conducted in duplicates at ambient temperature for three days. During the alkali pretreatment, the NaOH dose was 2% of substrate (TS) and the loading rate was 65 g/L (TS of sugarcane bagasse loaded per liter effective volume of digester). The alkali pretreatment condition in this study was used in the sugar factory where we collected the sugarcane bagasse, which was also suggested by Zheng *et al.*<sup>19</sup> to be optimal in treating corn stover.

### **2.4 Batch anaerobic digestion tests**

 After thermophilic microaerobic pretreatment, the bottles were added with another 20ml anaerobic sludge and 30ml deionized water. The alkali pretreated sugarcane bagasse was transferred to 300 ml serum bottles, then 40ml anaerobic sludge and 138ml deionized water were added to reach a total volume of 200 ml. 22g untreated sugarcane bagasse, 40ml anaerobic sludge and 138ml deionized water were also mixed in bottles to test the biogas production from untreated sugarcane bagasse (marked as WP). Before anaerobic digestion, all the pH values were adjusted to 7.0 148 with 2 N NaOH and 2 N HCl, and then flushed with  $N_2$  for 5 min to replace the air, after that, the bottles were closed with rubber stoppers. All the bottles were placed in a 150 shaking water bath at 37  $\degree$ C with 110 rpm.

### **2.5. Structure analysis of solid fraction of sugarcane bagasse**

 Sugarcane bagasse samples were collected before and after pretreatment for the structure analysis. The structure analyses were conducted by a spectrum One FTIR system (The Nicolet iN10 IR Microscope) with a universal ATR (Attenuated Total  Reflection) accessory and wide angle X-ray diffraction, which was in accordance with 156 the reported methods  $^{18}$ .

### **2.6 Analytical methods**

 The biogas yield was measured by water displacement method. Biogas composition was measured by a gas chromatograph (SP 6890, Shandong Lunan Inc., China), equipped with a Porapak Q stainless steel column (180 cm long, 3 mm outer diameter) and a thermal conductivity detector. The temperatures of the injector, detector and oven were 50, 100 and 100 °C, respectively. The carrier gas was argon. TS, VS were 163 determined according to standard methods .

**3. Results and discussion**

 **3.1. The optimized oxygen loads during thermophilic microaerobic pretreatment** 

 When thermophilic microaerobic pretreatment is used as the pretreatment method, the oxygen load during TMP is a crucial parameter<sup>17, 21</sup>. Insufficient oxygen will not be strong enough to support the growth of facultative organisms. However, facultative organisms have higher growth rates and would out-compete strict anaerobes under high oxygen levels due to substrate competition. In addition, excessive oxygen may inhibit the activity of methanogens directly. In this study, the oxygen loads during thermophilic microaerobic pretreatment was investigated at the oxygen loads of 5, 10, 20 ml/g VSsubstrate. The methane yields of thermophilic microaerobic pretreated sugarcane bagasse are shown in Fig.1. Daily methane yields of thermophilic microaerobic pretreated sugarcane bagasse increased sharply at the fifth day of

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177 anaerobic digestion. The maximum daily methane yields of TMP1, TMP2 and TMP3 178 were obtained at  $9^{th}$ ,  $9^{th}$  and  $6^{th}$  day of AD, respectively. The cumulative methane 179 yields of thermophilic microaerobic pretreated sugarcane bagasse were ranged 180 between 196.5 and 229.6 ml/g  $VS_{substrate}$ , which were obtained at the oxygen loads of 181 20 and 10 ml/g  $VS_{substrate}$ , respectively. The maximum cumulative methane yield was 182 obtained at the oxygen loads of 10 ml/g  $VS_{\text{substrate}}$ , which was 29.28% higher than that 183 of WP. However, when the oxygen loads during TMP was 20 ml/g  $VS_{substrate}$ , the 184 cumulative methane yields decreased to 196.5 ml/g  $VS_{\text{substrate}}$ . This result was quite 185 accordance with what reported by Mshandete *et al.*<sup>16</sup> and Botheju *et al.*<sup>22</sup> Proper 186 oxygen loads (or the time exposed to oxygen) during microaerobic pretreatment is 187 crucial: microaerobic pretreatment would be beneficial for biogas production in a 188 proper condition, however, would be harmful in an improper condition.

# 189 **3.2 Comparisons of structural changes of sugarcane bagasse after thermophilic**

#### 190 **microaerobic and alkali pretreatment**

191 The ultimate purpose of pretreatment is to improve the methane yield or to 192 accelerate the anaerobic digestion process. On this basis, TMP2 was selected to make 193 a comparison with AP and WP.

194 3.2.1 *FT-IR* analysis of pretreated and untreated sugarcane bagasse

195 The result of *ATR FT-IR* spectroscopy was shown in Fig. 2. The peak near 3348 196  $\text{cm}^{-1}$  and 2900  $\text{cm}^{-1}$  represented wagging vibration in C-H and the O-H stretching of the hydrogen bonds of cellulose  $^{23, 24}$ . The absorption intensities of this two absorption 198 peaks was in the following order  $WP > AP > TMP$ , which means the cellulose of

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 sugarcane bagasse was partly disrupted during pretreatment. Moreover, thermophilic microaerobic pretreatment was more efficient at removal of cellulose. The band at  $1595 \text{ cm}^{-1}$  is attributed to aromatic ring stretching, which is associated with lignin removal. After alkali pretreatment the intensity of this peak was almost halved, which 203 was quite accordant with what reported by Sambusiti *et al.*<sup>25</sup>, alkali pretreatment is effective in altering the structure of lignin. However, thermophilic microaerobic 205 pretreatment almost had no effect on this peak. The band at  $1245 \text{ cm}^{-1}$  is attributed to C-O adsorption and has been proposed to be associated with the acetyl group in hemicelluloses. The intensity of this absorption peak of TMP decreased slightly. Relatively, the intensity of this absorption peak of AP dropped significantly, which means more hemicelluloses was disrupted during alkali pretreatment. The intensity of 210 the 900 cm<sup>-1</sup> is very sensitive to the amount of crystalline versus amorphous structure 211 of cellulose <sup>26</sup>. The intensity of this band was in the following order AP  $>$ WP  $>$ TMP, which means the crystalline structure after TMP was partly disrupted.

# 213 3.2.2 XRD analysis

214 The crystallinity of substrate is broadly accepted to be a negative factor for the 215 enzymatic hydrolysis of cellulose  $^{27}$ . The XRD analysis results were shown in Fig. 3 216 and Table.1. The crystallinity of sugarcane bagasse after TMP was decreased, which 217 was quite accorded with what reported by Fu *et al.*<sup>18</sup>. TMP was efficient in 218 crystallinity disruption. However, the crystallinity increased after AP, the results of 219 XRD analysis were quite accorded with the *FT-IR* analysis results. Increase of 220 crystallinity index after alkali pretreatment was also reported by Kumar *et al.*<sup>24</sup> and

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*Yao et al.* <sup>28</sup>. The greater hydrolysis of amorphous areas than crystalline areas, the removal of amorphous materials, such as lignin and acetyl groups might be the reason 223 for the increase of crystallinity after NaOH treatment  $^{28}$  $^{28}$  $^{28}$ .

 **3.3 Comparisons of fermentative characteristics between thermophilic microaerobic and alkali pretreated sugarcane bagasse**

 3.3.1 Methane yields of thermophilic microaerobic and alkali pretreated sugarcane bagasse during anaerobic digestion

 The methane-producing of sugarcane bagasse with thermophilic microaerobic and alkali pretreatment were shown in Fig.4. The maximum daily methane yield was obtained from TMP, which was 112.5% higher than that of untreated sample. The methane-producing peak of WP was 4 days later compared with those of TMP and AP, which means the methane-producing was accelerated after pretreatment. The maximum cumulative methane yield was obtained from the thermophilic microaerobic pretreated sugarcane bagasse and followed by the alkali pretreated sugarcane bagasse, which were 29.3% and 11.8% higher than that of untreated sample, respectively. As for the parameter of total methane yield, TMP was more efficient than AP. The total cumulative methane yield of TMP2 was 15.7% higher than that of AP. However, daily methane yield during the late stage of AD was tiny and it is not practical and economically feasible if the fermentation lasts too long. Therefore, the methane yield within the initial 40 days was also analyzed. The cumulative methane yields of AP, TMP2 and WP during the initial 40 days of AD were 165.1, 159 and 242 129.6 ml/g  $VS_{substrate}$ . The cumulative methane yield of AP during the initial 40 days

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 was 3.8% and 27.4% higher than those of TMP2 and WP, respectively, which means the methane-producing rate of AP during the initial 40 days was higher. AP and TMP2 obtained the same cumulative methane yield at the  $45<sup>th</sup>$  day of AD, after then, 246 the cumulative methane vield of TMP2 exceeded that of AP.

 The technical digestion time T90 is defined as the time consumed to achieve 90% 248 of maximum cumulative biogas production<sup>29</sup>. A shorter T90 means the substrate was consumed quickly, therefore, the anaerobic digestion system is more efficient. The T90 of AP, TMP2 and WP were 49, 59 and 56 days, respectively. The T90 of AP was 10 and 7 days less than those of TMP2 and WP, respectively. As for T90, AP was more efficient than TMP, which biogas-producing from alkali pretreated sugarcane bagasse was quicker.

3.3.2 The *modified Gompertz equation* analysis

 The *modified Gompertz equation* was usually employed to model the 256 methane-producing process<sup>30-32</sup>, which was written as following:

$$
P(t) = P \exp \left[-\exp \left(\frac{Rm*e}{P}(\lambda - t) + 1\right)\right]
$$

258 Where  $P$  (t) is the cumulative total methane yield (ml/g  $VS_{substrate}$ ),  $P$  is the total methane production potential (ml/g VSsubstrate), *R*m is the maximum 260 methane-producing rate (ml/d/g  $VS_{substrate}$ ),  $\lambda$  is the lag-phase time (d) and *t* is the elapsed time (d).

 The parameters of *modified Gompertz equation* fitting experimental data were shown 263 in table 2. The determination coefficient  $(R^2)$  ranged from 0.965 to 0.990, which indicated that methane-producing could well be explained by the *modified Gompertz*

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3.3.4 VS removal efficiency

 During the digestion process, volatile solids (VS) are degraded to a certain extent and converted into biogas and the degree of stabilization is often expressed as the 277 percent reduction in VS  $^{33}$ . The VS removal efficiencies of WP, AP and TMP were 54.48±0.35%, 60.65±0.91% and 63.93±0.62%, respectively. The maximum VS removal efficiency was obtained in TMP, which was 5.41% and 17.35% higher than those of AP and WP, respectively. The higher VS removal efficiency means more sugarcane bagasse was digested in TMP, which would be better for the reduction of fermentation residue.

**4. Conclusions**

 The effects of AP and TMP on the AD of sugarcane bagasse were investigated and compared in this study. Both AP and TMP are efficient pretreatment methods in AD of sugarcane bagasse. The oxygen load during TMP is crucial, the maximum cumulative methane yield of sugarcane bagasse was obtained at the oxygen load during TMP was

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288 10 ml/g  $VS_{substrate}$ . TMP is efficient in crystallinity disruption, lag-phase time, methane production and VS removal. AP was efficient in lignin removal, the technical digestion time and methane-producing rate. Compared with AP, which needs large amount of chemical reagent during pretreatment, TMP is more eco-friendly and economically feasible pretreatment method in AD of sugarcane bagasse.

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Table 1 crystallinity indices of untreated and pretreated sugarcane bagasse



Table 2 Parameters of *modified Gompertz equation* fitting experimental data

	Groups	$P$ (ml/g VS <sub>substrate</sub> )	$R_m$ (ml/d/g VS <sub>substrate</sub> )	$\lambda$ (d)	$\mathbb{R}^2$
	AP	$188.4 \pm 1.7$	$6.7 + 0.3$	$1.548 + 0.737$	0.983
	TMP <sub>2</sub>	$233.5 + 6.4$	$4.3 + 0.5$	$\boldsymbol{0}$	0.965
	<b>WP</b>	$174.5 + 2.0$	$4.4 + 0.2$	$3.819 \pm 0.966$	0.990
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