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## Compressed Sodalite-like MgH<sub>6</sub> as a Potential High-temperature Superconductor

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Recently, an experimental work reported a very high  $T_c$  of ~190K in hydrogen sulphide (H<sub>2</sub>S) at 200 GPa. The search of new superconductors with high superconducting critical temperatures in hydrogen-dominated materials has attracted significant attention. Here we predict a candidate phase of MgH<sub>6</sub> with a sodalite-like framework in conjunction with first-principles electronic structure calculations. The calculated formation enthalpy suggests that it is thermodynamically stable above 263 GPa relative to MgH<sub>2</sub> and solid hydrogen (H<sub>2</sub>). Moreover, the absence of imaginary frequency in phonon calculations implies that this MgH<sub>6</sub> structure is dynamically stable. Furthermore, our electron-phonon coupling calculation based on BCS theory indicates that this MgH<sub>6</sub> phase is a conventional superconductor with a high superconducting critical temperature of ~260 K under high pressure, which is even higher than that of the recently reported compressed H<sub>2</sub>S. The present results offer insights in understanding and designing new high-temperature superconductors.

### Introduction

The search of new superconductors with high critical temperatures ( $T_c$ ) has attracted great attention due to the discovery of superconductivity of mercury in 1911.<sup>1</sup> For conventional Bardeen-Cooper-Schrieffer (BCS) superconductors, MgB<sub>2</sub> was found to possess a  $T_c$  of 39 K.<sup>2</sup> However, there are no other conventional superconducting compounds to be found to have a higher  $T_c$  than that of MgB<sub>2</sub>. For other unconventional superconducting compounds, the last several decades have witnessed an exciting revolution in searching high-temperature superconductivity materials, such as cuprates and iron-based superconductivity materials, but their superconducting mechanism cannot be explained by Bardeen-Cooper-Schrieffer (BCS) theory and still raises a source of debate.<sup>3-5</sup> Ashcroft suggested the metallic phase of solid hydrogen should possess high-temperature conductivity.<sup>6</sup> Until now, solid hydrogen has not been found to become metallic until up to 360 GPa in laboratory.<sup>7</sup> It is noted that a recent theoretical work proposed that H<sub>2</sub>S may become a good superconductor above 100 GPa.<sup>8</sup> Soon after that a high-pressure electrical measurement well proved that the compressed H<sub>2</sub>S can become a superconductor with a very

high  $T_c$  of ~190K at 200 GPa.<sup>9</sup> On the other hand, another theoretical work also suggests that (H<sub>2</sub>S)<sub>2</sub>H<sub>2</sub> is a good superconductor with  $T_c$  of 191-204K at 200 GPa.<sup>10</sup> There have also been many theoretical efforts in studying the superconducting mechanism of H<sub>2</sub>S and investigating whether the compressed H<sub>2</sub>S is stable and dissociate into other Sulfur hydrides.<sup>11-18</sup> At present, the superconducting mechanism of compressed H<sub>2</sub>S is not quite clear and remains elusive. Therefore, investigations of other relevant hydrides are important as well as helpful in shedding light on the intricate superconducting mechanism and providing ideas for new potential high-temperature superconductor design.

Recently, CaH<sub>6</sub> has been predicted to have a good superconductivity of 235 K at 150 GPa.<sup>19</sup> This structure consists of a sodalite-like hydrogen cage filled with interstitial Calcium (Ca) atoms. Due to the similarity between Mg and Ca, Mg may also adopt a sodalite-like structure with hydrogen atoms. Therefore, it is interesting to explore the stability of sodalite MgH<sub>6</sub> structure and its potential superconducting behaviors. Moreover, another recent theoretical work also suggests that YH<sub>6</sub> adopts this sodalite structure with a  $T_c$  of 251-264 K at 120 GPa.<sup>20</sup> A recent study based on evolutionary structure searches suggested that MgH<sub>4</sub> can be stable with respect to decomposition into MgH<sub>2</sub> and H<sub>2</sub> near 100 GPa, and it remains the most stable stoichiometry until at least 200 GPa.<sup>21</sup>

In this work, we study the stability of MgH<sub>6</sub> relative to MgH<sub>2</sub> and H<sub>2</sub> on the basis of first-principle density functional theory. The calculations indicate that the MgH<sub>6</sub> is dynamically stable and thermodynamically stable above 263 GPa. Further electron-phonon coupling calculations imply that this MgH<sub>6</sub> phase possesses very good superconductivity with a high superconducting critical temperature of ~260K above 300 GPa. Our current results also show that the charge transfer

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between Mg and H atoms and H-H interactions are responsible for this good superconductivity. We find that the calculated  $T_c$  of this  $\text{MgH}_6$  phase increases up to above 300 GPa.

## Results

Inspired by our previous work<sup>19</sup> predicting high superconductivity of  $\text{CaH}_6$  where Hydrogen atoms adopt a sodalite-like framework that is responsible for high superconductivity, we have replaced Ca atoms by Mg atoms in this sodalite-like structure. This structure was optimized with fully relaxing the atoms and lattice parameters within first-principles electronic structure framework, which produces a body centered cubic (bcc) crystal structure and space group of  $Im\bar{3}m$  with 14 atoms per unit cell, in agreement with our previous work.<sup>19</sup> To investigate the thermodynamically stability, we have calculated the formation enthalpy relative to  $\text{MgH}_2$  and  $\text{H}_2$  (Fig. 1), where previously predicted  $P6_3/mmc$   $\text{Ni}_2\text{In}$ -type structure<sup>21</sup> was used for  $\text{MgH}_2$  and  $C2/c$  and  $Cmca-12$  structure<sup>22</sup> were used for solid hydrogen. Interestingly, this  $\text{MgH}_6$  is stable above 322 GPa which is higher than 150 GPa in  $\text{CaH}_6$ . This is not unreasonable since the light elements adopt low-pressure structures for heavy elements.<sup>23</sup> On the other hand,  $\text{MgH}_6$  is a hydrogen-dominated material where hydrogen is the lightest element and has a large zero-point (ZP) energy that may revise the stability range at high pressures. To account for the ZP effects, we estimated the ZP energies for  $\text{MgH}_6$ ,  $\text{MgH}_2$  and  $\text{H}_2$ , using a harmonic method.<sup>24,25</sup> As shown in the inset of Fig. 1, the inclusion of the ZP energies does shift down the stability of this  $\text{MgH}_6$  phase from 322 GPa to 263 GPa. This indicates that the ZP effect is further helpful to stabilize this newly predicted  $\text{MgH}_6$  structure.

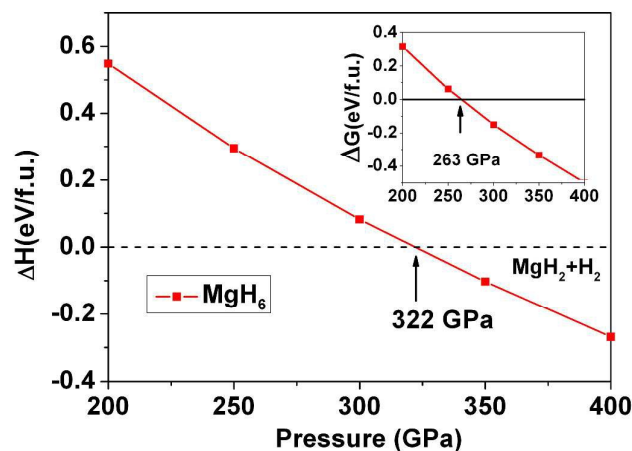


Fig. 1 The calculated formation enthalpy of  $\text{MgH}_6$  relative to  $\text{MgH}_2$  and  $\text{H}_2$ . The inset shows the formation energy of  $\text{MgH}_6$  with considering zero-point energy effects.

The sodalite-like  $\text{MgH}_6$  structure is shown in Fig. 2a, each hydrogen atom has four neighbor hydrogen atoms with distance of  $1.1\text{\AA}$  at 300 GPa. The H and Mg atom situate 2d position (0, 0.25, 0.5) and 2a position (0, 0, 0), respectively. The distance between hydrogen atoms is similar to the

distance in atomic phases of solid hydrogen (e.g.  $1.0\text{\AA}$  in Cs-IV structure of solid hydrogen at 500 GPa). It is noted that one previous theoretical work suggests that the atomic phases of solid hydrogen possess good superconductivity under high pressure.<sup>26</sup> The similar bonding behaviors suggest that  $\text{MgH}_6$  may be a good superconductor as well. To further gain insight into the interactions between hydrogen atoms, we have calculated the electron localization function (ELF).<sup>27</sup> The topological analysis of ELF is commonly used to determine the degree of electron localization and subsequently, the tendency to form two electron covalent bonds in molecules and solids. Represented in a convenient scale,  $\text{ELF}=1$  corresponds to perfect localization and at the low limit  $\text{ELF}=0.5$  reflects the behavior of a homogeneous electron gas. The contour plot of the ELF for this  $\text{MgH}_6$  structures is shown in Fig. 2b. The strong interaction between hydrogen atoms can be clearly seen. Moreover, we have carried out Bader calculations on this  $\text{MgH}_6$  structure using the quantum theory of atoms-in-molecules (AIM).<sup>28</sup> In the AIM theory, an atom within a solid is defined through the “zero-flux” condition of the electron density gradient. The results suggest that each Mg atom donates 1.63 electrons and each hydrogen atom attracts 0.27 electrons. This charge transfer may help enhance the interactions between hydrogen atoms.

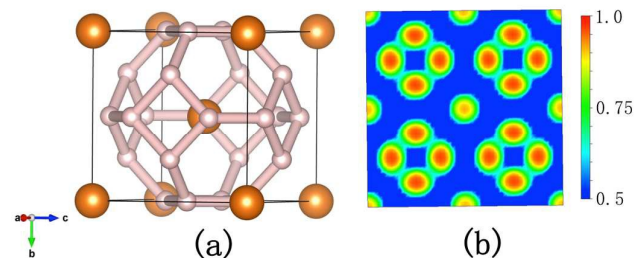


Fig. 2 (a) The structure of  $\text{MgH}_6$ . The small and big spheres represent H and Mg atoms, respectively. (b) The 2-dimensional electron localization function in {100} plane is shown with a  $2\times 2$  supercell at 300 GPa.

## Discussion

To investigate the electronic states, band structure for this  $\text{MgH}_6$  structures was calculated as shown in Fig. 3. The results clearly show that this phase is a metallic phase, which is in good agreement with the previous results.<sup>19</sup> The projected density of states in Fig. 3 show that the  $\text{H}_s$  orbit is also important for the contribution of this metallization. There are two bands crossing the Fermi level (Fig. 3). This results in a high electronic density states at the Fermi energy (Fig. 3). Moreover, there are two parallel bands along the H-N-P direction, which is helpful to form Fermi nesting that can induce strong electron-phonon coupling. The band dispersion along P- $\Gamma$ -N direction is parabolic-like and close to the Fermi level. One band crossing along the N-P direction with steep slopes indicates the existence of mobile electrons along these symmetry directions. The co-existence of electrons with large effective masses (flat band around  $\Gamma$  and P) and itinerant

electrons with high mobility (steep bands in H-N, N-P and P- $\Gamma$ ) suggested the possibility of high superconductivity.

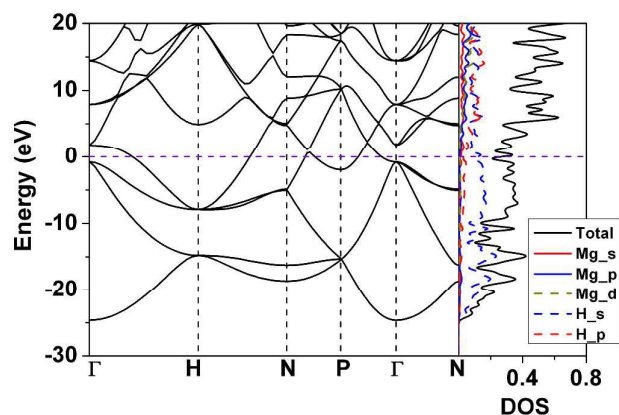


Fig. 3 The calculated band structure and density of states of MgH<sub>6</sub> structure at 300 GPa.

As mentioned above, the electronic band structure of MgH<sub>6</sub> shows the existence of sharp and flat bands crossing the Fermi energy level, indicative of its possible superconductivity. In order to investigate the superconductivity of MgH<sub>6</sub>, phonon band structures, phonon density of states (PHDOS) and Eliashberg spectral function  $\alpha^2F(\omega)$  for MgH<sub>6</sub> were carried out. As shown in Fig. 4. There is no imaginary frequency to be found in the whole Brillouin zone (BZ), confirming that it is a dynamically stable structure.

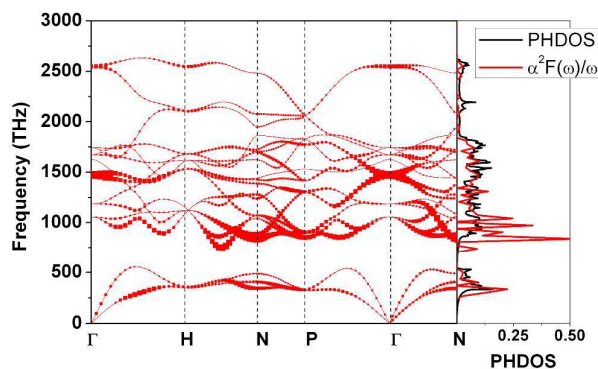


Fig. 4 The calculated phonon band structure, phonon density of states and Eliashberg spectral function of MgH<sub>6</sub> at 300 GPa.

Circles with a radius proportional to the electron-phonon coupling strength were also plotted in Fig. 4, to illustrate the contributions associated with different phonon modes. One can observe that nearly all phonon modes contribute to the whole Brillouin zone, reflecting a three-dimensional superconducting nature of the MgH<sub>6</sub> structure. Moreover, the superconductivity in MgH<sub>6</sub> is associated with the Kohn anomalies observed in the phonon dispersion (Fig. 4). At 300 GPa (Fig. 4b), the electron-phonon coupling parameter ( $\lambda$ ) for the MgH<sub>6</sub> structure is 3.29 with an average phonon frequency  $\ln(\omega)$  of 1450 K. Using the Allen Dynes equation,<sup>29</sup> which is an extension of the McMillan theory,<sup>30</sup> with a nominal Coulomb

pseudopotential parameter ( $\mu^*$ ) of 0.12 the estimated superconducting critical temperature  $T_c$  is 263 K.

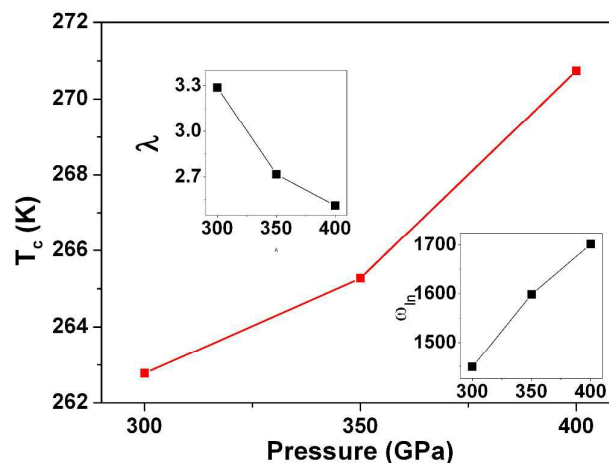


Fig. 5 Superconducting temperature ( $T_c$ ) vs. pressure. The coupling parameters, the average phonon frequencies  $\omega_{ln}$  and superconducting temperatures of MgH<sub>6</sub> as a function of pressure.

We have also studied the  $T_c$  of MgH<sub>6</sub> as a function of pressure. As shown in Fig. 5, the  $T_c$  increases in the pressure range of 300 to 400 GPa. Furthermore, it is found that the calculated  $\lambda$  decreases with increasing pressure, but the average frequency increases as expected. The competing between the  $\lambda$  and the average frequency results in the increased  $T_c$  with pressure. The origin of the superconductivity could be traced back by a comparison of the calculated Eliashberg spectral function ( $\alpha^2F(\omega)$ ) and PHDOS. As shown in Fig. 4b, 15% of the electron phonon coupling is contributed to the low-frequency vibrations in the region from 0 to 600  $\text{cm}^{-1}$ , which are mostly from the vibrations of Mg atoms. The remained 85% derives from high-frequency vibrations from 600 to 2600  $\text{cm}^{-1}$  which are predominately H-H stretching and bending modes. The superconducting critical temperature of 263 K at 300 GPa is higher than 39K<sup>2</sup> in MgB<sub>2</sub> and even 190K<sup>9</sup> in recently measured solid H<sub>2</sub>S.

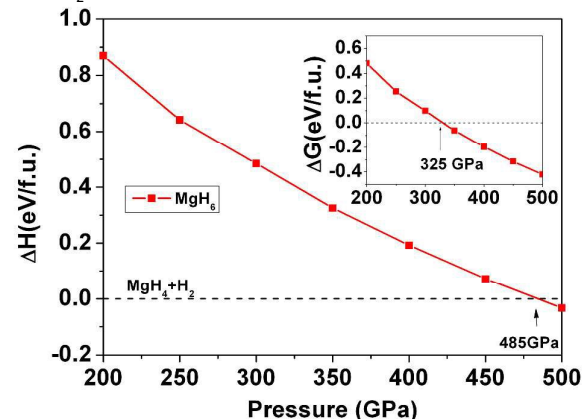


Fig. 6 The calculated formation enthalpy of MgH<sub>6</sub> relative to MgH<sub>2</sub> and H<sub>2</sub>. The inset shows the formation energy of MgH<sub>6</sub> with considering zero-point energy effects.

Finally, we should point out that our calculated phase transition pressure (263 GPa) of MgH<sub>6</sub> is higher than that ( $\sim$ 100



GPa) of MgH<sub>4</sub>,<sup>21</sup> which means that MgH<sub>4</sub> is more thermodynamically stable than MgH<sub>6</sub> at low pressures. Furthermore, we have carried out the calculations on formation enthalpy of MgH<sub>6</sub> relative to MgH<sub>4</sub> and H<sub>2</sub> as shown in Fig. 6. It shows that MgH<sub>6</sub> actually thermodynamically stable relative to MgH<sub>4</sub> and H<sub>2</sub> above 325 GPa with considering ZP effects. However, it is noteworthy that MgH<sub>6</sub> is thermodynamically stable above 263 GPa relative to MgH<sub>2</sub> and H<sub>2</sub>. Moreover, our phonon calculation confirms that is still a dynamically stable structure. These results suggest MgH<sub>6</sub> may be synthesized above 263 GPa with starting materials of MgH<sub>2</sub> and H<sub>2</sub>. We also envisage that our predicted MgH<sub>6</sub> structure could possess a very high superconducting temperature. This work will benefit the study of superconducting mechanism for conventional BCS superconductors.

## Conclusions

In summary, we have proposed a sodalite-like MgH<sub>6</sub> structure within first-principles electronic structure framework. This newly predicted MgH<sub>6</sub> structure was found to be thermodynamically stable above 263 GPa on the basis of the formation enthalpy calculations relative to MgH<sub>2</sub> and H<sub>2</sub>. The phonon calculations suggest that this MgH<sub>6</sub> structure is dynamically stable due to the absence of imaginary frequency in the whole BZ. Furthermore, our electron-phonon coupling calculations reveal that MgH<sub>6</sub> is a good BCS superconductor with a high superconducting critical temperature of ~260 K above 300 GPa. The electron-phonon calculations also suggest that the three-dimensional superconducting nature in this newly predicted MgH<sub>6</sub> structure. The calculated  $T_c$  of MgH<sub>6</sub> was found to increase at a pressure range from 300 to 400 GPa with the decrease of the calculated  $\lambda$  and the increase of average frequency under compression. Our finding will stimulate future experimental study on synthesis of Magnesium hydrides and explore its high- $T_c$  superconductivity under high pressure.

## Computational details

Structural optimizations and enthalpy calculations were carried out with the Vienna *ab initio* simulation (VASP)<sup>31-33</sup> program and projector-augmented plane wave (PAW)<sup>34</sup> potentials employing the Perdew–Burke–Ernzerhof (PBE)<sup>35</sup> functional. The valence configurations for the H and Mg potentials are  $1s^1$  and  $2s^2 2p^6 3s^2$ , respectively. A plane wave basis set with an energy cutoff of 800 eV was used. Dense k-point meshes were employed to sample the first Brillouin zone (BZ) to ensure the energies are converged within 5 meV/atom. The elements of the interatomic force constant (IFC) and electron-phonon coupling matrices were computed using the linear response method with a  $4 \times 4 \times 4$  q-point mesh and  $24 \times 24 \times 24$  k-point mesh for the first Brillouin zone integrations.<sup>36</sup> A plane wave basis set with an energy cutoff of 80 Ry was used.

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