

# RSC Advances



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

*Accepted Manuscripts* are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. This *Accepted Manuscript* will be replaced by the edited, formatted and paginated article as soon as this is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



25 the enrichment of quorum sensing auto-inducer relating bacteria genera (21%) and the  
26 stable production of EPS were suggested as main reasons for the positive effect of  
27 L-tyrosine on the granulation and stability of AGS.

28

29 **Keywords:** Aerobic granular sludge; granulation and stability; L-tyrosine effect;  
30 COD/N ratio

31

## 32 **1. Introduction**

33

34 As a special form of biofilm, aerobic granular sludge (AGS) has been recognized as a  
35 low energy and small footprint technology for substituting activated sludge processes  
36 for municipal and industrial wastewater treatment.<sup>1</sup> The merits of AGS, including  
37 improved settleability, high biomass retention, and high flexibility against changes in  
38 pollution and environmental conditions, have been intensively demonstrated.<sup>2</sup> These  
39 advantages ensure a high application potential of the AGS technology in wastewater  
40 treatment. Nonetheless, efficient start-up and stability of AGS process still remain  
41 critical issues.<sup>3</sup>

42

43 The extracellular polymeric substance (EPS) is known to affect the granulation  
44 process of AGS.<sup>4</sup> The main components proteins (PN) and polysaccharides (PS) of  
45 EPS are further found to be important in maintaining the properties of AGS.<sup>5</sup> The  
46 granular structure is supported by a backbone mainly composed of PS<sup>1,6</sup> and the PN

47 may improve AGS granulation and structure integrity due to enhanced surface  
48 hydrophobicity and the reduced surface negative charge.<sup>4,7</sup> Apart from PS and PN,  
49 alginate-like exopolysaccharides (ALE) successfully extracted from the EPS of AGS  
50 developed with synthetic wastewater has been confirmed its function of improving  
51 formation of AGS.<sup>8</sup> In addition, the EPSs also contain other components, such as  
52 lipids and humic, fulvic, amino acids, etc.<sup>1,4</sup> Their roles in EPS's contribution in the  
53 granulation and structural stability of AGS are unknown. Therefore, further  
54 investigation on possible contributing components of EPSs in AGS is deemed  
55 necessary to resolve critical issues.

56

57 Our recent study discovered a significant decrease in tyrosine-like compounds in the  
58 EPS extracted from the AGS that disintegrated when it exposed to a low COD/N ratio  
59 (=1) medium in a sequencing airlift bioreactor (SABR).<sup>9</sup> The effect of tyrosine has  
60 been investigated on formation of biofilm, and mainly focused on L and D-tyrosine.  
61 The L-tyrosine was found a precursor to *N*-acyl-tyrosine which is bacteria signaling  
62 controlling biofilm formation.<sup>10</sup> On the contrary, Kolodkin-Gal et al.<sup>11</sup> reported that  
63 D-amino acids including D-tyrosine could prevent biofilm formation. Mixed effects  
64 were also observed in "L" isomers of various amino acids, including L-tyrosine; some  
65 promoted while some inhibited biofilm formation.<sup>12</sup> An exploratory study on the role  
66 of L-tyrosine in aerobic sludge granulation while maintaining the structure integrity  
67 needs to be carried out. Aerobic sludge granulation and stability in a SABR dosed  
68 with L-tyrosine and that without L-tyrosine dosing as the control system were tested

69 in parallel. Close monitoring of the physical, chemical and biological characterization  
70 of both AGS granules were performed. Findings from this study would shed light on  
71 the role of EPS in the AGS formation and stability.

72

## 73 **2. Materials and Methods**

74

### 75 **2.1 Reactor and operation**

76

77 Aerobic sludge granulation was conducted in two identical SABRs (100 cm in height  
78 and 5 cm in diameter with a working volume of 1.1 L in each reactor). The SABRs  
79 were operated in parallel under the same conditions, one serving as the control reactor  
80 (Ra) free of L-tyrosine, while the other (Rb) tested the effect of L-tyrosine (Sigma  
81 Aldrich, St. Louis, MO, USA) dosed via its influent as a constant rate of 6 mg/L, as  
82 adopted from a biofilm formation control study.<sup>11,13</sup> A 2-L/min air flow rate was  
83 applied in each reactor, maintaining a superficial up-flow air velocity of 1.2 cm/s.<sup>14</sup>  
84 Both reactors had a 2.4-h operation cycle comprised of 6 min feeding, 120 min  
85 aeration, 5 min settling, 5 min decanting and 8 min idling. The volumetric exchange  
86 ratio of each reactor was set at 50%, corresponding to a hydraulic retention time  
87 (HRT) of 4.8 h. Details of the operation strategies are provided in Table A.1 in the  
88 Supplementary Information (SI).

89

### 90 **2.2 Feeding**

91

92 Both reactors were inoculated with 2 g/L of activated sludge taken from a local  
93 sewage treatment plant in Hong Kong. Synthetic wastewater was prepared by diluting  
94 the synthetic stock solution (Table A.2 in SI) with tap water to set the influent COD  
95 concentration at a typical level, i.e. 400 mg/L. Acetate, glucose and yeast comprised  
96 the source of organics. In order to decrease the COD/N ratio in the influent from 4  
97 (phase I) to 1 (Phase II), ammonium chloride was added from day 60 accordingly, in  
98 order to increase the influent ammonium concentration from 100 to 400 mg-N/L in  
99 Phase I and II respectively (Table A.3 in SI). The ratio of bicarbonate ( $\text{NaHCO}_3$ ) to  
100 ammonium-nitrogen (N) in the influent was fixed at 4 to maintain pH at 7.9. The  
101 ambient temperature in the laboratory was  $23 \pm 2$  °C.

102

### 103 2.3 Analytical methods

104

#### 105 2.3.1 Water quality and sludge physical property analysis

106

107 COD was determined following the Standard Methods.<sup>15</sup> Total nitrogen (TN) was  
108 measured by using a total organic carbon analyzer (TOC-VCPH, Shimadzu, Japan)  
109 equipped with a total nitrogen measurement unit (TNM-1, Shimadzu). A flow  
110 injection analyzer (QuikChem 8500, Lachat Instruments) was applied to measure the  
111 concentration of ammonium nitrogen while the nitrate and nitrite nitrogen were  
112 analyzed by using an ion chromatograph (HIC-20A super, Shimadzu). The physical

113 properties of granular sludge were characterized with mixed liquid suspended solid  
114 (MLSS), mixed liquor volatile suspended solids (MLVSS), sludge volumetric index  
115 (SVI), particle size distribution, morphology and structure, cohesion, and specific area.  
116 MLSS, MLVSS and SVI were measured every two days according to the Standard  
117 Method<sup>15</sup>. The size distribution of sludge was determined with a laser diffraction  
118 particle size analyzer (LSI3 320, Beckman Coulter). The sludge micro-structure and  
119 morphology were examined with a scanning electron microscope (SEM) (JSM 6300F,  
120 JEOL) after fixing the sludge sample over night at 4°C in 2% paraformaldehyde, 2%  
121 glutaraldehyde, and 1X phosphate-buffered saline (PBS) mixed solution and  
122 subsequent lyophilization. Cohesion tests followed the protocol reported by Wan et  
123 al.<sup>16</sup> with the following modifications: 1) measurement chamber geometry: closed  
124 with a volume of 520 mL, diameter 100 mm, depth 66 mm, paddle diameter 63 mm,  
125 and paddle height 24 mm placed at 1/3 of the chamber height from the bottom (paddle  
126 power number  $N_p$  of 2.9), and 2) a series of velocity gradients of 250, 13230 and 250  
127  $s^{-1}$  were applied. The specific area and total pore volume of sludge were also  
128 examined by using an automatic instrument (ASAP 2020, Micrometrics) and then  
129 determined from the Brunauer–Emmett–Teller (BET) method.<sup>17</sup>

130

### 131 2.3.2 EPS analysis

132

#### 133 *EPS content and component*

134

135 EPS was first extracted from the sludge using a formaldehyde-NaOH method.<sup>18</sup> The  
136 carbohydrate or polysaccharide (PS) in EPS was then quantified by using the  
137 phenol-sulfuric acid (PSA) method with glucose as the standard.<sup>19</sup> The protein (PN) in  
138 EPS was determined from the modified Lowry colorimetric method with bovine  
139 albumin serum as the standard.<sup>20</sup> A luminescence spectrometer (F-4500 FL  
140 Spectrophotometer, Hitachi) was used to examine the EEM spectra of extracted EPS  
141 for components identification with methodology applied as previously described<sup>9</sup>.

142

### 143 2.3.3 Microbial diversity analysis

144

145 A Power Soil DNA extraction kit (MO BIO Laboratories Inc.) was periodically  
146 applied to extract DNA from the sludge samples taken from both reactors. The 16S  
147 rDNA gene was amplified by PCR with the following steps: 94 °C for 5 min followed  
148 by 30 cycles of 94 °C for 30 s, 53 °C for 30 s and 72 °C for 45 s; and a final extension at  
149 72 °C for 10 min. Then 200ng of purified 16S rDNA amplicons from each sample were  
150 pooled and subjected to pyrosequencing using the ROCHE 454 FLX Titanium platform  
151 (Roche) at the National Human Genome Centre of China in Shanghai, China. The  
152 pyrosequencing methodology employed was the same as previously described<sup>9</sup>.

153

154 Low quality sequences were removed from raw sequence data by trimming the  
155 barcode tags and primer sequences. FASTA files were generated from the resulting  
156 sequences according to the barcodes of individual samples. The sequences were then



157 aligned using the software Mothur ver. 1.17.0<sup>21</sup> and the distance matrix was produced.  
158 Operational taxonomic units (OTU) were determined at 90, 95 and 97% similarities  
159 (Mothur v. 1.17.0). Rarefaction curves and diversity indices (ACE and Chao1) were  
160 determined from the calculated OTUs using the same software. In the  
161 taxonomy-based analysis, the representative sequences from each OTU were  
162 subjected to the RDP-II Classifier of the Ribosomal Database Project (RDP),<sup>22</sup> the  
163 National Centre for Biotechnology Information (NCBI) BLAST,<sup>23</sup> and the  
164 Greengenes Databases.<sup>24</sup> The relative abundance and occurrence of the tags assigned  
165 into these three samples were visualized as a heat map using the Multi Experiment  
166 Viewer (MeV) v4.8.1 software.

167

### 168 **3. Results**

169

#### 170 **3.1 Formation of aerobic granules**

171

172 Disintegration of AGS induced by low COD/N ratio (i.e., COD/N = 1) has been  
173 confirmed in our previous work<sup>9</sup>. Thus, to eliminate unexpected effects imposed by  
174 factors other from the L-tyrosine, both reactors were started up under an appropriate  
175 COD/N ratio of 4 in the formation phase (phase I). The seeding sludge was aerated for  
176 2 days in both reactors prior to the inoculation. A short settling time (5 min) was then  
177 applied in both reactors. MLSS was intensively monitored during the start-up period.  
178 In the control reactor, a decrease of MLSS was observed within the first 18 days

179 (from 2 to 1.4 g/L), but the sludge in the test reactor showed excellent settling  
180 properties, enabling a continuous increase in MLSS from initially 2 to 6.4 g/L at day  
181 76 (see Fig. 1a).

182

183 Figure 1b shows the mean particle diameter of the sludge in each reactor. The mean  
184 particle size in the test reactor reached 200  $\mu\text{m}$  after 7 days and then became stable at  
185  $2,150 \pm 20 \mu\text{m}$  after 32 days. The  $\text{SVI}_5$  of both reactors was initially 100 mL/g. After  
186 11 days' operation,  $\text{SVI}_5$  of the test reactor significantly decreased to 35 mL/g and  
187 subsequently remained at around 47 mL/g throughout the entire period. Conversely,  
188 the  $\text{SVI}_5$  of the control reactor stayed unchanged until day 39, and then gradually  
189 reduced to 76 mL/g after day 46. Apparently, granulation in the control reactor was  
190 much slower than that in the test reactor, though its sludge floc size was measured to  
191 be 200  $\mu\text{m}$  or above from day 14. The ratio of  $\text{SVI}_5$  to  $\text{SVI}_{30}$  is often used to evaluate  
192 the extent of granulation, and when the value is close to unity full granulation is  
193 recognized.<sup>25</sup> Figure 1c illustrates that the ratio of  $\text{SVI}_5/\text{SVI}_{30}$  was close to unity in  
194 the test reactor after 11 days, and 39 days were required for the control reactor, i.e. 28  
195 days longer than the test reactor.

196

### 197 3.2 Stability of aerobic granular sludge

198

199 In our previous stability study of AGS, the granules underwent a condition of  
200 stepwise decrease in COD/N ratio from 4 to 1, and eventually disintegration occurred

201 at a ratio of 1.<sup>9</sup> Accordingly, after full granulation was achieved in these two reactors,  
202 the COD/N ratio was decreased from 4 at day 60 to 1 at day 120 to investigate the  
203 effect of L-tyrosine on the stability of aerobic granules, during which the key physical,  
204 chemical and biological characteristics of both AGSs were monitored closely. The  
205 respective results are summarized below.

206

### 207 3.2.1 Particle size and settling ability

208

209 With a decrease of COD/N ratio from 4 to 1, the mean diameter of the granules in the  
210 test reactor remained stable at  $2100 \pm 100 \mu\text{m}$  throughout the operation, and in the  
211 control reactor significantly decreased by 90% from  $1600 \pm 50 \mu\text{m}$  initially to  $200$   
212  $\pm 18 \mu\text{m}$  after 60 days (see Fig. 2a).

213

214 Meanwhile, the  $\text{SVI}_5$  of the control reactor increased from  $55 \pm 5 \text{ mL/g}$  to  $110 \pm 10$   
215  $\text{mL/g}$ . However, that of the test reactor maintained at  $50 \pm 5 \text{ mL/g}$  till the end of the  
216 experiment. According to the ratio of  $\text{SVI}_5/\text{SVI}_{30}$ , AGS in the control reactor  
217 disintegrated after day 88, while the test reactor maintained the extent of granulation  
218 during the whole operation in Phase II with COD/N equal to 1 (see Fig. 2b). Due to  
219 the deterioration of the settling ability of AGS in the control reactor, half of the sludge  
220 was washed out, i.e. the MLSS concentration decreased from  $6 \text{ g/L}$  at day 94 to  $3 \text{ g/L}$   
221 at the end of the experiment. Comparatively, the MLSS in the test reactor increased  
222 from  $5.9$  to  $7.6 \text{ g/L}$  (see Fig. 2c).

223

224 3.2.2 Performance of the reactors

225

226 The ammonium nitrogen removal efficiencies of both reactors during the entire  
227 operation are shown in Fig. 3. Both reactors showed similar removal efficiency (more  
228 than 90%) during the granulation period (Phase I, COD/N=4). With the COD/N ratio  
229 further decreased from 4 to 1, the removal efficiency of both reactors was halved at  
230 day 60. The removal efficiency of the control reactor continually decreased to 17%  
231 eventually. Conversely, that of the test reactor gradually recovered to 75% at the end  
232 of experiment.

233

234 The maximum oxidation rate of the ammonium and nitrite in both reactors in Phases I  
235 (COD/N=4) and II (COD/N=1) were determined from the cycle tests conducted with  
236 a dissolved oxygen (DO) concentration fixed at 8.2 mg/L. As shown in Table 1, the  
237 rates of the test reactor were determined to be 8.4 and 3.0 mg N/g VSS·h, 50 and 43%  
238 higher than that of the control reactor (5.6 and 2.1 mg N/g VSS·h) in Phase I  
239 (COD/N=4). The maximum oxidation rate of ammonia increased to 19.4 in the  
240 control reactor and 21.8 mg N/g VSS·h in the test reactor while the ratio of COD/N  
241 decreased to 1 in Phase II. The nitrite oxidation rate in the test reactor maintained at  
242 2.9 mg N/g-VSS·h, but in the control reactor, it increased threefold to 6.1 mg  
243 N/g-VSS·h due to the disintegration of AGS in control reactor after the COD/N ratio  
244 decreasing to 1.

245

246 In phase I, TN loss in the control reactor ( $35 \pm 5\%$ ) was greater than that ( $30 \pm 5\%$ ) in  
247 the test reactor (see Fig. 3b), which could be ascribed to higher porosity of AGS in the  
248 testing reactor (Table 2) decreasing anaerobic zones in granules. When the COD/N  
249 ratio decreased from 4 to 1, the TN loss gradually reduced in this reactor due to  
250 disintegration of granules, however the TN loss in the test reactor still remained at  
251 between 20 and 30% when the ratio reached 1, with the stable porosity and surface  
252 area features of AGS maintained (see Table 2). The reasons for such stable properties  
253 of the granules is further discussed in Section 3.2.3. The COD removal efficiency of  
254 both reactors was maintained at more than 85% throughout the operation.

255

256 3.2.3 Physical characteristics of granules

257

258 The physical strength of the granules in both reactors was measured to illustrate their  
259 structural features and capability to withstand high abrasion and water shear forces.  
260 Figure 4 shows the cohesion results of the granules in both reactors. During Phase I  
261 (COD/N=4), no apparent coagulation and break-up of the granules were observed in  
262 the control reactor when the mixing forces (G value) was  $250\text{s}^{-1}$ ,  $13230\text{ s}^{-1}$  and  $250\text{ s}^{-1}$   
263 successively. This confirmed that the sludge in this reactor was of pure granular  
264 structure.<sup>16</sup> However, when the same G-values were applied to the test reactor,  
265 break-up and re-coagulation were observed (see Fig.4a) though the granules remained  
266 intact, indicating that flocculent sludge-like properties were incorporated in the  
267 granules in the test reactor. The same sludge behavior in the test reactor was observed

268 in Phase II (COD/N=1), with a complete dispersion occurring in the control reactor  
269 (see Fig. 4b).

270

271 The physical strength of the granules is usually negatively correlated with its  
272 porosity,<sup>26</sup> which determines substrate transport and oxygen diffusion within the  
273 granules. Hence the porosities of the granules in both reactors under different COD/N  
274 ratios were examined, as shown in Table 2. With the COD/N ratio decreased from 4  
275 to 1, the surface area (BET) and total pore volume of the granules in the control  
276 reactor clearly rose from 1.9 to 4.5 m<sup>2</sup>/g and 9.0 to 18.1 ml/kg-VSS respectively, in  
277 line with the results of the cohesion tests. Accordingly, the surface area and total pore  
278 volume of the granules in the test reactor were maintained at 3.5 m<sup>2</sup>/g and 13.0  
279 ml/kg-VSS, respectively.

280

#### 281 4. Discussion

282

##### 283 4.1 Effect of L-tyrosine on formation and stability of AGS

284

285 The production of EPSs, especially the PS and PN compounds, plays a key role in  
286 aerobic granulation and structure integrity via altering the physical-chemical  
287 properties of the cellular surface of AGS, such as hydrophobicity and charge.<sup>5,27</sup> The  
288 change of certain components in EPSs, e.g., tyrosine-like compounds fading out in the  
289 disintegrating AGS was revealed by Luo et al.<sup>9</sup> In the present study, the effect of  
290 L-tyrosine on AGS is to: accelerate granulation (stage I) with L-tyrosine at 6 mg/L,

291 acquiring AGS formation within 11 days, i.e. 5 days shorter than the fastest  
292 granulation period ever reported (see Table 3) and a stable reactor performance as  
293 well as structural integrity of AGS in terms of the changes in diameter and physical  
294 strength (Phase II).

295

296 4.2 The production of EPS

297

298 As shown in Figure 5, in Phase I (COD/N=4), the PS content of the granules in the  
299 test reactor reached 38 mg/ g-VSS, two times greater than that in the control reactor.

300 When the COD/N ratio reduced to 1 at day 60, the PS extracted from the granules in  
301 the test reactor slightly decreased to 35 mg/g-VSS; comparatively it significantly  
302 decreased by 50% to 10 mg/g-VSS in the control reactor, which in turn disintegrated  
303 the granules (Fig. 2). This indicated the importance of EPS in stability maintenance of  
304 the AGS. Stable secretion of EPS in test reactor can be ascribed to three reasons.

305 Firstly, the porous structure of the L-tyrosine-promoted granules. Higher porosity of  
306 the aerobic granules in the test reactor could relieve the limitation of mass transfer and  
307 thus maintained PS secretion in the center of the granules.<sup>1,4</sup> Secondly, the L-tyrosine  
308 as substrate can be utilized by bacteria to synthesize *N*-acyl-tyrosine which promotes  
309 the production of EPS<sup>10</sup>. Thirdly, the L-tyrosine is the main compound of the tyrosine  
310 kinase and tyrosine phosphatases which are vital to cell regulatory enzymes for a  
311 number of microbial processes, including EPS secretion.<sup>32</sup>

312

313 At the end of experiment, a 3D-EEM analysis was conducted to identify the different  
314 tyrosine-EPS in both reactors. The results indicate that the tyrosine protein-like  
315 (region A) substances were observed in granules of both reactors in phase I.  
316 Subsequently the tyrosine protein-like contours faded out in the granules of the  
317 control reactor after the COD/N ratio decreasing from 4 to 1, but can be detected in  
318 the testing reactor throughout the reactor operation (see Fig. A.1 in SI). In our  
319 pervious study, the tyrosine protein-like compound in EPS correlated with the  
320 disintegration of aerobic granules under extremely low COD/N ratio. Therefore, the  
321 accumulation of tyrosine protein-like would be one of the reasons for the integrity  
322 structure of aerobic granules in testing reactor under phase II.

323

324 4.3 The improvement of microbial community

325

326 The microbial communities of the granules in both reactors were analyzed and  
327 compared by 16S rDNA pyrosequencing at the end of the experiment. Approximately  
328 40,000 and 50,000 effective sequence tags were retrieved from the two types of  
329 granules (control and L-tyrosine-promoted). Sufficient sequencing was confirmed by  
330 the rarefaction curves (see Fig. A.2 in SI).

331

332 At the phylum level, proteobacteria in L-tyrosine-promoted granules accounted for 80%  
333 much higher than the control granules (57%). This phylum is reported to associate  
334 with the production of *N*-acyl-tyrosine which can promote production of EPS<sup>10</sup>. The



335 abundance of bacteroidetes (27%), which are responsible for secreting lectin-specific  
336 EPS (glycoconjugates) for cell attachment,<sup>33</sup> was 3.5 times higher in the  
337 L-tyrosine-promoted granules than in the control granules (8%).

338

339 At the genus level (Fig. 6), the abundance of *Flavobacterium* (3.3%), *Nitrosomonas*  
340 (3.0%) and *Thauera* (15%) in the L-tyrosine-promoted granules were much higher  
341 than those of the control granules (0.2, 1.6 and 0.7%, respectively). Meanwhile, all of  
342 the aforesaid genera have been shown to contain species that positively correlate with  
343 the quorum sensing (QS) auto-inducers during granulation.<sup>34</sup> The QS auto-inducers,  
344 especially *N*-acyl-homoserine-lactone (AHL), have been recently recognized as  
345 signaling molecules for biofilm development.<sup>34</sup> These genera could be driving or  
346 accelerating the granulation when L-tyrosine dosed. In addition, chemical similarities  
347 between AHL and *N*-acyl-tyrosine has been reported previously<sup>10</sup> and they both can  
348 enhance the formation of biofilm. However connections between L-tyrosine, AHL and  
349 these genera are not clear yet, and a further study is deserved.

350

351 The portion of *Flaviumibacter* was larger in the L-tyrosine-promoted granules (1.9 %)  
352 than the control granules (0.1 %). *Flaviumibacter* belongs to filamentous bacteria  
353 which can accelerate aerobic granulation with high porosity by offering a structural  
354 network to aggregate the cells together.<sup>35</sup> Therefore, the richness of *Flaviumibacter*  
355 with L-tyrosine dosing could be another cause for the high porosity of AGS in the test  
356 reactor.

357

358 AOB *Nitrosomonas* in the L-tyrosine-promoted and control granules accounted for 3  
359 and 1.5 % respectively. NOB *Nitrospira* and *Nitrobacter* were rare in both granules (<  
360 0.3%), and such low abundance of NOB reflects the accumulation of nitrite in both  
361 reactors. The minimum sludge retention times (SRTs) were calculated for both AOB  
362 and NOB, as shown in Table A.5 in SI. The mean SRT of both reactors was 8 days  
363 longer than the minimum SRT for AOB and NOB, i.e. 4.4 and 1.8 days. Tarre et al.,  
364 (2007) indicated an optimum pH range for AOB and NOB growth of 7.0 to 8.5 and  
365 7.3 to 7.5, respectively. The low NOB abundance is most likely caused by the influent  
366 pH of 7.5-8.0 in the reactors. Additionally, the highly porous structure of  
367 L-tyrosine-promoted granules is beneficial to the development of AOB abundance  
368 from which oxygen can easily penetrate into core area of the granules.

369

## 370 5. Conclusions

371

372 With dosing of L-tyrosine, test reactor achieved full granulation after 11 days'  
373 operation, and maintained structural integrity when decreased COD/N ratio from 4 to  
374 1. Granules in test reactor were determined more porous than that in the control  
375 reactor, and this property potentially mitigates the mass transfer limitation in the  
376 granules. The stable secretion of EPS and the enrichment of genera related to the  
377 secretion of QS auto-inducers and filamentous in the test reactor were suggested as  
378 the main reasons for the positive effect of tyrosine on the granulation and stability of

379 AGS.

380

381 **Acknowledgement**

382 This work is partly supported by China Natural Science Foundation (38000-41030553).

383

384

385 **Reference**

- 386 1. S. S. Adav, D. J. Lee, *Journal of Hazardous Materials*, 2008, **154**, 1120-1126.
- 387 2. M. K. de Kreuk, N. Kishida, M. C. M. van Loosdrecht, *Water Science &*  
388 *Technology*, 2007, **55**, 75-81.
- 389 3. K. Y. Show, D. J. Lee, J. H. Tay, *Applied Biochemistry and Biotechnology*, 2012,  
390 **167**, 1622-1640.
- 391 4. B. S. McSwain, R. L. Irvine, M. Hausner, P. A. Wilderer, *Applied and*  
392 *Environmental Microbiology*, 2005, **71**, 1051-1057.
- 393 5. J. E. Schmidt, B. K. Ahring, *Applied Microbiology and Biotechnology*, 1994, **42**,  
394 457-62.
- 395 6. B. Long, C. Z. Yang, W. H. Pu, J. K. Yang, Y. F. Shi, J. Wang, J. Bai, X. Y. Zhou,  
396 G. S. Jiang, C. Y. Li, F. B. Li, *Bioresource Technology*, 2014, **169**, 244-250.
- 397 7. L. Zhu, M. L. Lv, X. Dai, Y. W. Yu, H. Y. Qi, X. Y. Xu, *Bioresource Technology*,  
398 2012, **107**, 46-54.
- 399 8. Y. M. Lin, M. de Kreuk, M. C. M. van Loosdrecht, A. Adin, *Water Research*, 2010.  
400 **44**, 3355-3364.
- 401 9. J. H. Luo, T. W. Hao, L. Wei, H. R. Mackey, Z. Q. Lin, G. H. Chen, *Water*  
402 *Research*, 2014, **62**, 127-135.
- 403 10. J. W. Craig, M. A. Cherry, S. F. Brady, *Journal of Bacteriology*, 2011, **193**,  
404 5707-5715.
- 405 11. I. Kolodkin-Gal, D. Romero, S. Cao, J. Clardy, R. Kolter, R. Losick, *Science*,  
406 2010, **328**, 627-629.
- 407 12. S.N. Goh, A. Fernandez, S.Z. Ang, W.Y. Lau, D.L. Ng, E.S.G. Cheah, *Journal of*  
408 *Biology and Life Science*, 2013, **4**, 103-115.
- 409 13. H. J. Xu, Y. Liu, *Water Research*, 2011, **45**, 5796-5804.
- 410 14. J. H. Tay, Q. S. Liu, Y. Liu, *Water Science and Technology*, 2004, 49, 35-40.
- 411 15. APHA, AWWA, WEF, 21st ed, Washington, D.C. 2005.
- 412 16. J. F. Wan, I. Mozo, A. Filali, A. Line, Y. Bessiere, M. Sperandio, *Biochemical*  
413 *Engineering Journal*, 2011, 58-59, 69-78.
- 414 17. V. Fierro, F. V. Torne, D. Montane, A. Celzard, *Microporous and Mesoporous*  
415 *Materials*, 2008, **111**, 276-284.
- 416 18. H. Liu, H. H. Fang, *Journal of Biotechnology*, 2002, **95**, 249-56.
- 417 19. M. Dubois, K. A. Gilles, J. K. Hamilton, P. A. Rebers, F. Smith, *Analytical*  
418 *Chemistry*, 1956, **28**, 350-356.
- 419 20. B. Frølund, T. Griebe, P. H. Nielsen, *Applied Microbiology and Biotechnology*,  
420 1995, **43**, 755-761.
- 421 21. P. D. Schloss, S. L. Westcott, T. Ryabin, J. R. Hall, M. Hartmann, E. B. Hollister,  
422 R. A. Lesniewski, B. B. Oakley, D. H. Parks, C. J. Robinson, J. W. Sahl, B. Stres,  
423 G. G. Thallinger, D. J. van Horn, C. F. Weber, *Applied and Environmental*  
424 *Microbiology*, 2009, **75**, 7537-41.
- 425 22. J. R. Cole, Q. Wang, E. Cardenas, J. Fish, B. Chai, R. J. Farris, A. S.  
426 Kulam-Syed-Mohideen, D. M. McGarrell, T. Marsh, G. M. Garrity, J. M. Tiedje,  
427 *Nucleic Acids Research*, 2009, **37**, 141-145.

- 428 23. M. Johnson, I. Zaretskaya, Y. Raytselis, Y. Merezhuk, S. McGinnis, T. L. Madden,  
429 *Nucleic Acids Research*, 2008, **36**, W5–W9.
- 430 24. T. Z. de Santis, P. Hugenholtz, N. Larsen, M. Rojas, E. L. Brodie, K. Keller, T.  
431 Huber, D. Dalevi, P. Hu, G. L. Andersen, *Applied and Environmental*  
432 *Microbiology*, 2006, **72**, 5069–5072.
- 433 25. M. K. de Kreuk, J. J. Heijnen, M. C. M. van Loosdrecht, *Biotechnology and*  
434 *Bioengineering*, 2005, **90**, 761-9.
- 435 26. R. Lemaire, R.I. Webb, Z. G. Yuan, *ISME Journal*, 2008, **2**, 528–541.
- 436 27. Y. Liu, S. Y. Yang, J. H. Tay, Q. S. Liu, L. Qin, Y. Li, *Enzyme and Microbial*  
437 *Technology*, 2004, **34**, 371–379.
- 438 28. X. H. Wang, M. H. Diao, Y. Yang, Y. J. Shi, M. M. Gao, S. G. Wang, *Bioresource*  
439 *technology*, 2012, **110**, 105-110.
- 440 29. M. Pijuan, U. Werner, Z. G. Yuan, *Water Research*, 2011, **45**, 5075-5083.
- 441 30. L. Liu, D. W. Gao, Min. Zhang, F. Yuan, *Journal of Hazardous Materials*, 2010,  
442 **181**, 382-387.
- 443 31. X. M. Li, Q. Q. Liu, Q. Yang, L. Gao, G. M. Zeng, J. M. Hu, W. Zheng,  
444 *Bioresource Technology*, 2009, **100**, 64-67.
- 445 32. A. J. Standish, A. A. Salim, H. Zhang, R. J. Capon, R. Morona, *PLoS ONE*, 2012, **7**,  
446 5.
- 447 33. M. B. Chirstin, R. N. Thomas, M. F. Bernhard, A. Rudolf, *Systematic and Applied*  
448 *Microbiology*, 2013, **36**, 417-425.
- 449 34. C. H. Tan, K. S. Koh, C. Xie, M. Tay, Y. Zhou, R. Williams, W. J. Ng, S. A. Rice,  
450 S. Kjelleberg, *ISME Journal*, 2014, **8**, 1186-97
- 451 35. Y. M. Zheng, H. Q. Yu, S. J. Liu, *Chemosphere*, 2006, **63**, 1791–1800.

## LIST OF TABLES AND FIGURES

### TABLES

**Table 1.** Maximum oxidation rates (mg N/g VSS·h) of ammonium and nitrite

**Table 2.** Surface area and porosity of granules in phases I and II

**Table 3.** AGS granulation time vs different enhancing methods

### FIGURES

**Fig. 1** Profile of sludge indices under granulation period (Phase I) with COD/N ratio of 4 for: (a) MLSS, (b) mean diameter of granules, and (c) SVI<sub>5</sub> and SVI<sub>5</sub>/SVI<sub>30</sub>

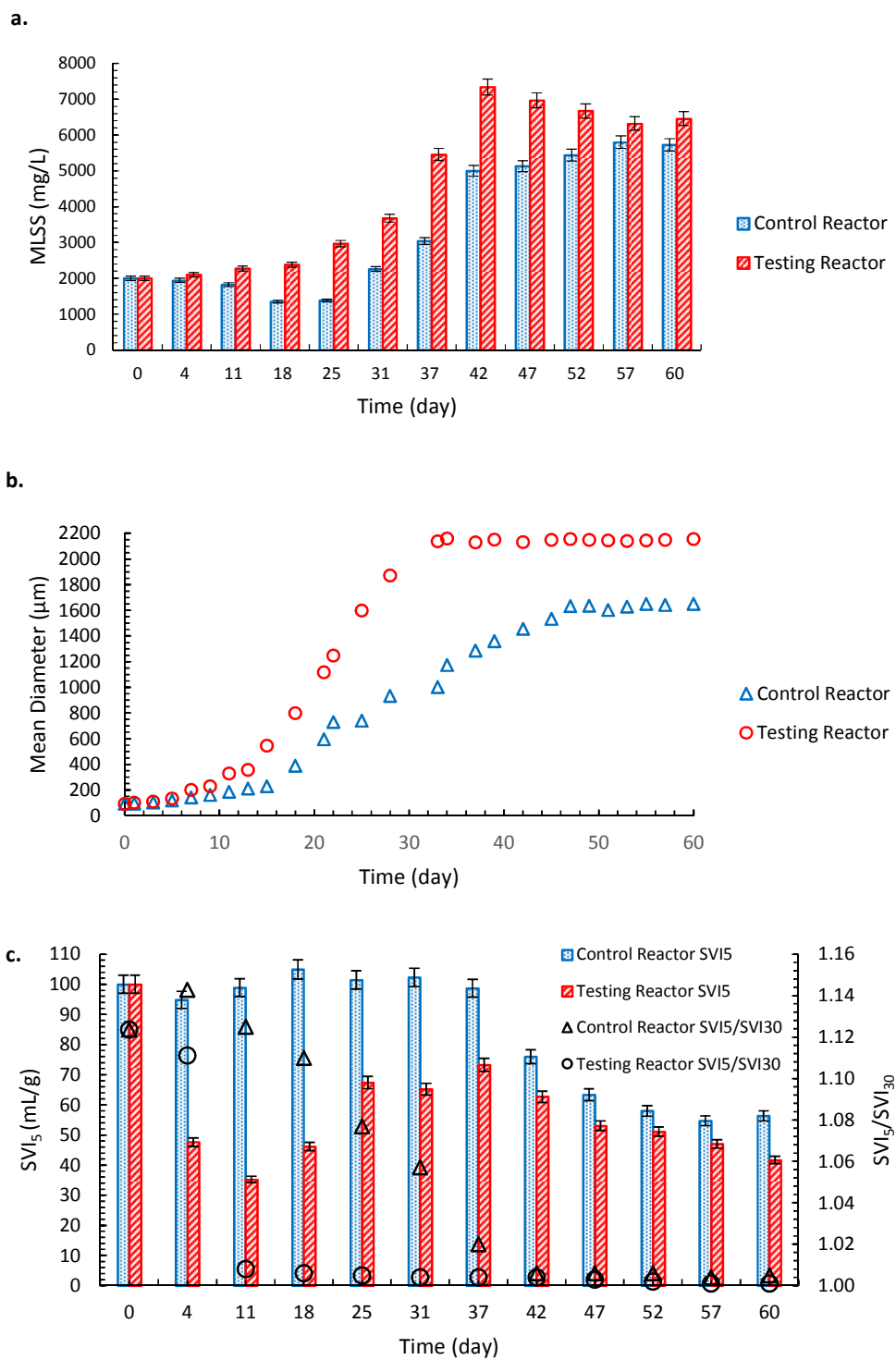
**Fig. 2** Profile of sludge indices in Phase II with COD/N =1: (a) mean diameter of granules, (b) SVI<sub>5</sub> and SVI<sub>5</sub>/SVI<sub>30</sub>, and (c) MLSS

**Fig.3** Profile of the removal efficiencies in both reactors during the whole operation for (a) ammonia and (b) TN

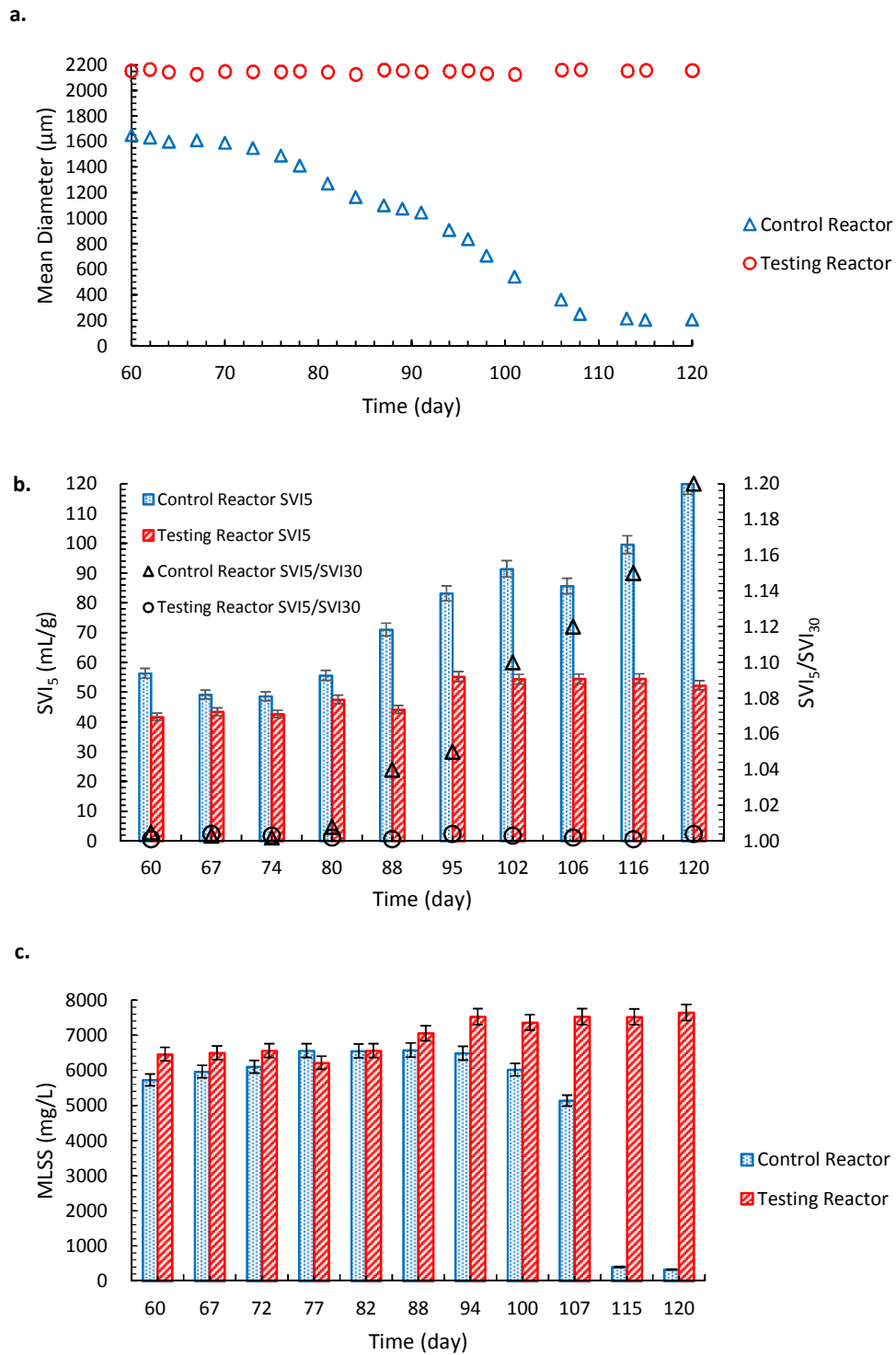
**Fig.4** Changes in the mean particle size of the granular sludge during the cohesion tests: (a) Phase I (COD/N=4), (b) Phase II (COD/N=1)

**Fig. 5** Profiles of EPS content of the granular sludge in (a) control reactor, (b) testing reactor

**Fig. 6** Taxonomic classification of bacterial 16s rDNA reads retrieved from sludge of the control and testing reactors at genera level

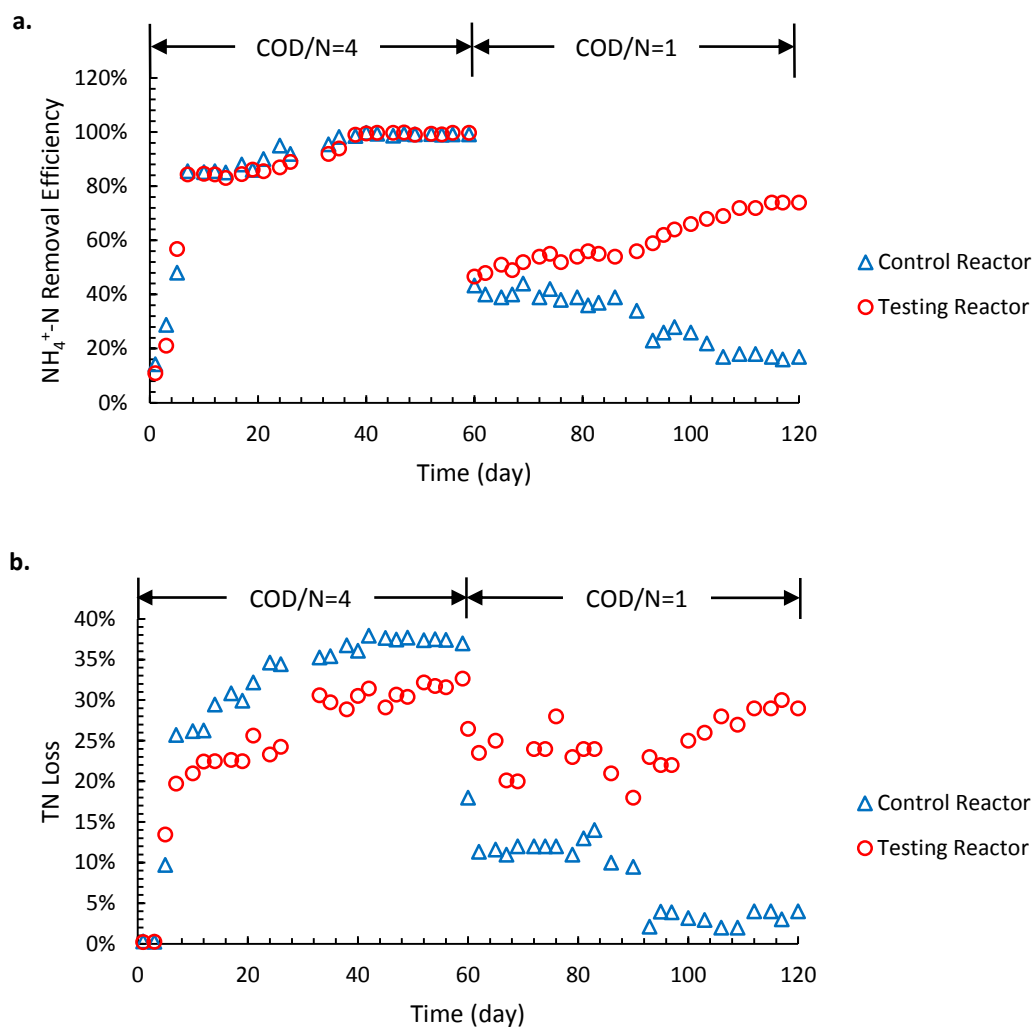


**Fig. 1** Profile of sludge indices under granulation period (Phase I) with COD/N ratio of 4 for: (a) MLSS, (b) mean diameter of granules, and (c) SVI<sub>5</sub> and SVI<sub>5</sub>/SVI<sub>30</sub>

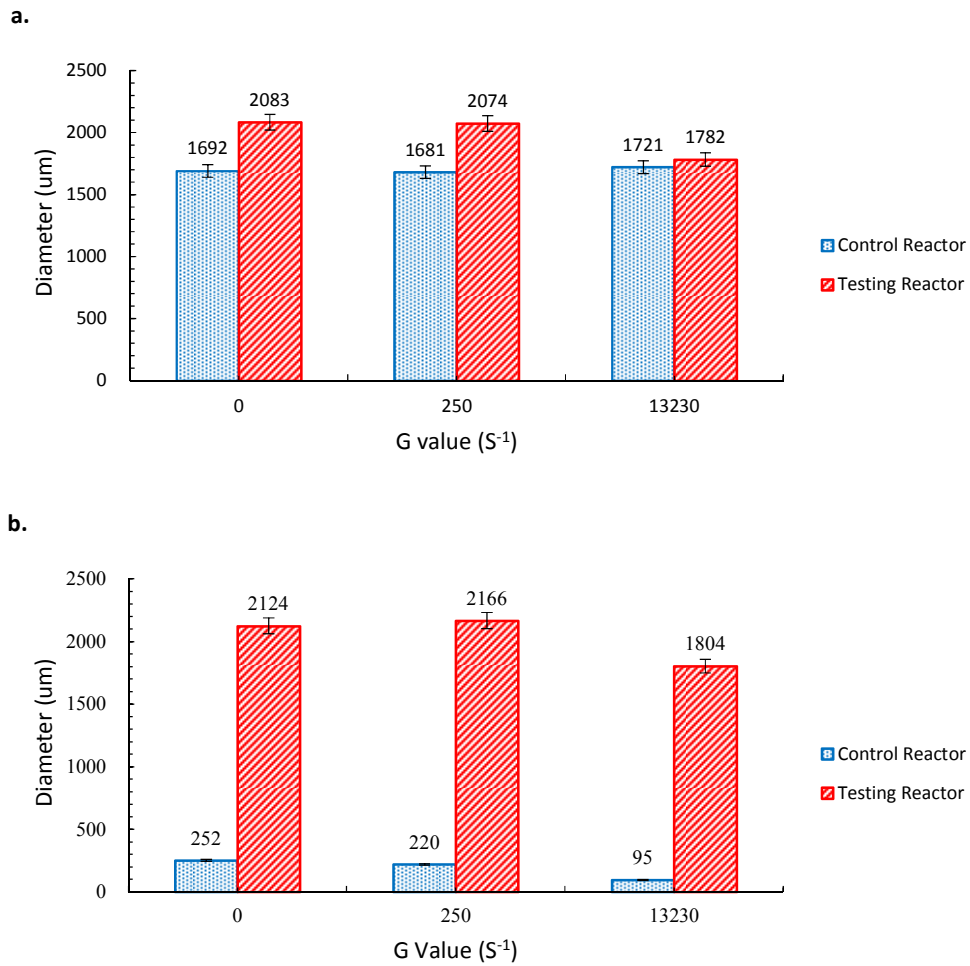


**Fig. 2** Profile of sludge indices in Phase II with COD/N =1: (a) mean diameter of granules, (b) SVI<sub>5</sub> and SVI<sub>5</sub>/SVI<sub>30</sub>, and (c) MLSS

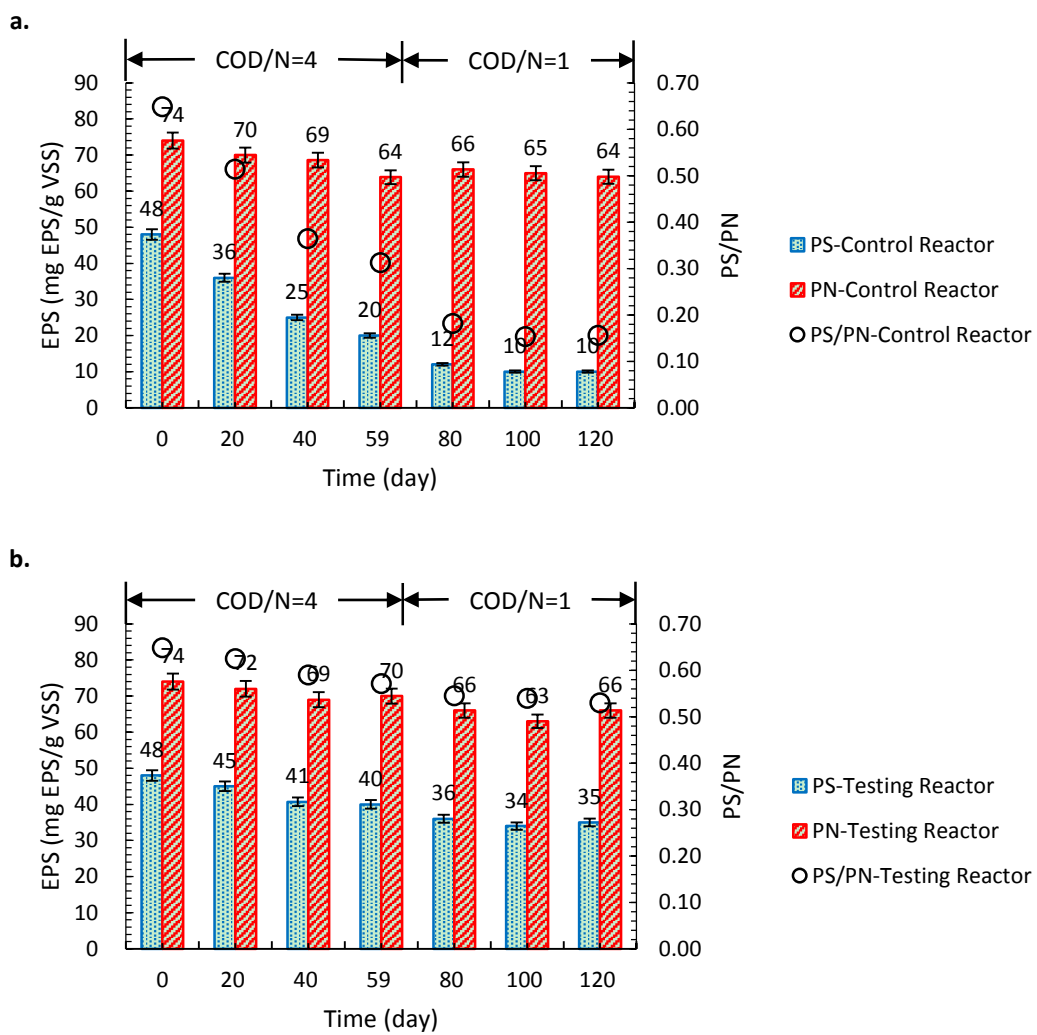




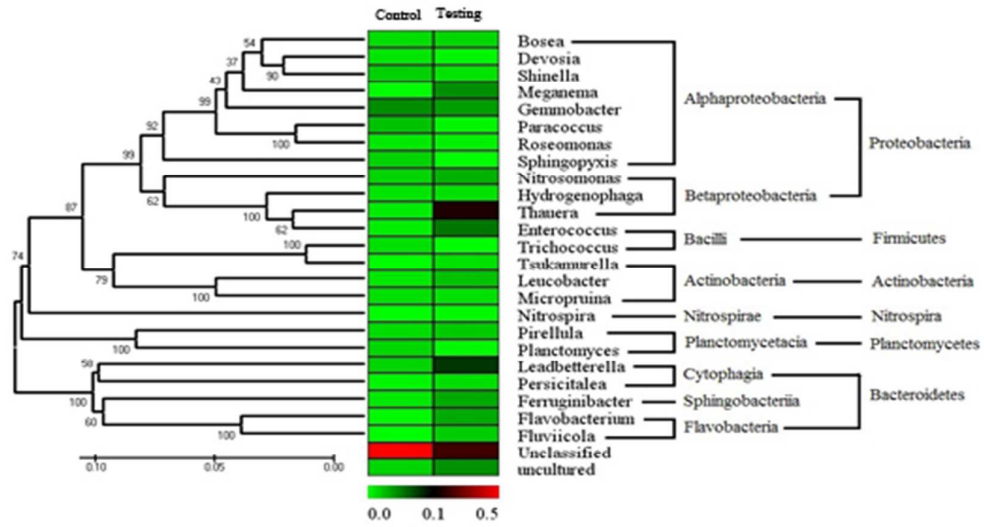
**Fig. 3** Profile of the removal efficiencies in both reactors during the whole operation for (a) ammonia and (b) TN



**Fig. 4** Changes in the mean particle size of the granular sludge during the cohesion tests: (a) Phase I (COD/N=4), (b) Phase II (COD/N=1)



**Fig. 5** Profiles of EPS content of the granular sludge in (a) control reactor, (b) testing reactor



**Fig. 6.** Taxonomic classification of bacterial 16s rDNA reads retrieved from sludge of the control and testing reactors at genera level

**Table 1.** Maximum oxidation rates (mg N/g VSS·h) of ammonium and nitrite

	Phase I, COD/N=4		Phase II, COD/N=1	
	Control Reactor	Testing Reactor	Control Reactor	Testing Reactor
Max. $\text{NH}_4^+$ oxidation rate	$5.6 \pm 0.2$	$8.4 \pm 0.3$	$19.4 \pm 0.5$	$21.8 \pm 0.5$
Max. $\text{NO}_2^-$ oxidation rate	$2.1 \pm 0.1$	$3.0 \pm 0.2$	$6.1 \pm 0.2$	$2.9 \pm 0.1$

**Table 2.** Surface area and porosity of granules in phases I and II

COD/N	BET Surface Area (m <sup>2</sup> /g)		Total Pore Volume (ml/kg-VSS)	
	Control Reactor	Testing Reactor	Control Reactor	Testing Reactor
	4	1.9±0.2	3.6±0.4	9.0±0.6
1	4.5±0.4	3.4±0.3	18.1±1.1	12.6±0.9

**Table 3.** AGS granulation time vs different enhancing methods

Strategies	granulation period (days)	References
Static magnetic field	25	28
50% crashed granules mixed	20	29
Ca <sup>2+</sup> augmentation	17	30
Mg <sup>2+</sup> augmentation	16	31
L-tyrosine	11	Present study