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1 **Qualitative analysis of chiral alanine by UV-visible-shortwave near**  
2 **infrared diffuse reflectance spectroscopy combined with chemometrics**

3

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10 **Abstract:** UV-visible-shortwave near infrared diffuse reflectance  
11 spectroscopy (UV-vis-SWNIR DRS) combined with chemometrics are  
12 firstly investigated to discriminate enantiomers and their racemic  
13 compounds, using D-, L- and DL-alanine as model compounds. After  
14 optimizing the measuring conditions of powder particle sizes and distance  
15 between fiber probe and sample, discriminant partial least squares (PLS-DA)  
16 models were built with UV-vis-SWNIR DRS to implement the qualitative  
17 analysis of alanine chirality. As a result, under the optimized conditions of  
18 particle size sifted through 100-mesh and distance of 5.3 mm, an excellent  
19 discrimination of chirality with an accuracy of 100% were obtained, better  
20 than that of manufacturers with an accuracy of 86.67%. The results of this

21 study infer that UV-vis-SWNIR DRS combined with chemometrics can be a  
22 rapid, simple and noninvasive method for chiral analysis.

23 **Keywords:** UV-vis-SWNIR DRS; D-, L- and DL-alanine; qualitative  
24 analysis; chemometrics

## 26 **1 Introduction**

27 Separation and analysis of enantiomers are continuously active areas in  
28 chiral research, as individual enantiomers and racemates of chiral  
29 compounds have significant differences in pharmacological activity,  
30 pharmacokinetic characteristics and toxicological properties. It is hardly  
31 surprising that with the increasing production of chiral drugs, the  
32 pharmaceutical industry needs some more effective and rapid chiral  
33 analytical methods for drug quality control. Methods generally used to  
34 determine enantiomers are based on either separation or spectroscopic  
35 techniques. While separation-based methods, such as chromatography<sup>1,2</sup> and  
36 capillary electrophoresis<sup>3-5</sup> which often need complex sample pretreatment,  
37 are time-consuming and expensive. Circular dichroism as a  
38 spectrophotometry, NMR as a spectroscopy and MS as a spectrometry are  
39 also commonly used. However, circular dichroism is lower sensitive<sup>6</sup> and  
40 often performed in solution<sup>7,8</sup>. NMR<sup>9</sup> and MS<sup>10,11</sup> require the addition of

41 chiral reagents, besides they are both expensive and the latter is destructive.  
42 As for spectroscopic methods, as UV-visible spectroscopy<sup>12-14</sup>, UV  
43 spectroscopy<sup>15-19</sup>, Fluorescence spectroscopy<sup>20-22</sup>, Resonance Rayleigh  
44 Scattering spectroscopy<sup>23</sup> and Near-infrared spectroscopy<sup>6, 24</sup>, guest-host  
45 complexes were formed in solution which results in spectral shift between  
46 enantiomers consequently allows chiral analysis. However, guest-host  
47 complexation in solution not only makes the analytical procedure much  
48 complicated but also needs more extra cost.

49 On the other hand, diffuse reflectance infrared Fourier transform  
50 spectroscopy (DRIFT) <sup>25, 26</sup> and terahertz time domain spectroscopy  
51 (THz-TDS) <sup>27-29</sup> methods were reported for analysis of enantiomers in solid  
52 powder, but they demand to mix samples with KBr and magnesium oxide,  
53 respectively. The THz-TDS<sup>27-29</sup> also has some other shortcomings:  
54 relatively weak absorption intensities compared to those of mid-infrared  
55 spectra and UV-visible spectra and the interference from the strong  
56 absorption of water<sup>27</sup>.

57 Therefore, developing a simple-preprocessing, rapid and inexpensive  
58 chiral analytical technique is significant. UV-visible-shortwave near infrared  
59 diffuse reflectance spectroscopy, one of the spectral analysis techniques  
60 based on the measurement of the output light loading information of the

61 structure and composition of sample through multiple interactions of  
62 incident light with internal molecules, has been used for studying the  
63 surface structures of molecules dispersed vanadium oxide on various  
64 supports<sup>30</sup>, for determining the composition of mineral-organic mixes<sup>31</sup>, for  
65 the detection of *p*-aminophenol<sup>32</sup>, for the qualitative and simultaneous  
66 quantitative analysis of cimetidine polymorphs<sup>33</sup>. Above all of these imply  
67 that UV-vis-SWNIR DRS can be expected to be a nondestructive chiral  
68 analytical technique.

69 In our preliminary study, it was found that the UV-vis-SWNIR diffuse  
70 reflectance spectra showed a significant difference among L-alanine,  
71 D-alanine and DL-alanine. Moreover, according to literature survey, there is  
72 still no report on the study of UV-vis-SWNIR DRS for chiral analysis.  
73 Therefore, the aim of our study is to investigate the feasibility of applying  
74 UV-vis-SWNIR DRS coupled with chemometrics to achieve the analysis of  
75 D-, L- and DL-alanine, in order to develop a new strategy *i.e.* a convenient,  
76 fast, low-cost and nondestructive method for discrimination of enantiomers.

77 Accordingly, based on this study, we propose a new method to  
78 determine enantiomers of compounds with UV-vis-SWNIR DRS.

## 79 **2 Experimental**

### 80 **2.1 Reagents**

81 D-, L- and DL-alanine (purity $\geq$ 98%) were purchased from Kelong  
82 chemical reagent factory and Best-Reagent Company in Chengdu and  
83 Sinopharm chemical Reagent Co., Ltd in Shanghai, respectively. They were  
84 qualified by the polarimetry listed in the current Chinese Pharmacopoeia, in  
85 which their solutions of 0.05g/ml were prepared and used.

86 Each of the nine kinds of alanine products was milled and sifted  
87 through 60, 80, 100, 120 and 200-mesh sieve (0.300, 0.200, 0.150, 0.125,  
88 0.075 mm nominal diameters), and stored in shade for analysis. Among the  
89 particle samples, 200-mesh was used for X-ray diffraction patterns and 60,  
90 80, 100 and 120-mesh were used for study on UV-vis-SWNIR DRS.

## 91 **2.2 Instruments**

92 S3000 Fiber Optic Spectrometer (Race-Technology Co., Ltd,  
93 Hangzhou, China) equipped with a 3648-element linear silicon CCD array  
94 detector (Toshiba TCD 1305), a Y-type optical fiber probe with 100 cm in  
95 length and 0.4 mm in diameter, a light source (Oceans Optics Inc., USA)  
96 and a home-made sample cell made from dark gray PVC, was used to  
97 measure the UV-vis-SWNIR diffuse reflectance spectra within the  
98 wavelength region of 200-1100 nm.

99 WZZ-3 automatic polarimeter (Shen Guang Instrument Co., Ltd,  
100 Shanghai, China), Optical microscope (UOP Photoelectric Technology Co.,

101 Ltd, Chongqing, China) and X'Pert PRO powder diffractometer  
102 (PANalytical Company, Holland) with a Pixcel 1D detector and Cu  $K_{\alpha 1}$   
103 radiation in the range of 5-50°  $2\theta$  were used to measure specific rotation,  
104 micrographs and X-ray diffraction patterns of D-, L- and DL-alanine,  
105 respectively.

106 Differential Scanning Calorimeter (DSC) 8500 (Perkinelmer Company,  
107 USA) with a measured temperature range of -180 °C to 750 °C and  
108 extremely fast controlled scanning rates to 750°C/min was used to analyze  
109 the melting points of D-, L- and DL-alanine.

### 110 **2.3 Acquisition of UV-vis-SWNIR diffuse reflectance spectra**

111 Spectra were measured under the following conditions: in the range  
112 from 200 to 1100 nm, integral time of 397 ms and a resolution of 0.29 nm,  
113 with a spectralon as background reference. Each alanine sample of 0.18 g  
114 was filled in the sample cell, pressed by free fall impacts of a round rod  
115 from a same height. Then the optical probe was placed vertically on the  
116 upper surface of the sample powders. Each sample was measured three  
117 times and their average spectrum was treated as the final spectrum of the  
118 sample. And each kind of alanine had fifteen samples by re-filling a given  
119 amount of the same kind of alanine fifteen times.

### 120 **2.4 Data analysis**

121 The UnscramblerX version10.2 (Camo Process AS, Oslo, Norway) was  
122 applied to analyze the acquired UV-vis-SWNIR diffuse reflectance spectral  
123 data. This software supplies many data preprocessing methods such as mean  
124 center and scale, autoscaling, normalization (Nor), savitzky-golay  
125 smoothing (SG), multiplicative scatter correction (MSC), standard normal  
126 variate transformation (SNV), detrend, the first derivative (1D), the second  
127 derivative (2D) and so on. Then, discriminant partial least squares (PLS-DA)  
128 were utilized to establish models and make prediction.

129 PLS-DA is a multiple linear classification based on principal  
130 component analysis in which the dimension reduction of both the  
131 independent and dependent variable matrix are carried out. And in the  
132 model of PLS-DA, calibration set and test set of a number of samples are  
133 needed. Calibration set is used to establish the model in which test set is  
134 used for prediction. What's more, the optimized latent variables (LVs) is  
135 selected to obtain a good result when the smallest root mean standard error  
136 of cross validation and largest square correlation coefficient are given by  
137 leave-one-out cross validation method.

138 The index values of square correlation coefficient ( $R^2$ ), root mean  
139 square error of calibration (RMSEC), root mean square error of  
140 cross-validation (RMSECV), root mean square error of prediction (RMSEP),



141 ratio of performance deviation (RPD) and accuracy (A) are used to evaluate  
 142 the established models as shown in Eq.(1), (2), (3) and (4).

$$143 \quad R^2 = 1 - \frac{\sum_{i=1}^n (\hat{y}_i - y_i)^2}{\sum_{i=1}^n (y_i - \bar{y}_i)^2} \quad (1)$$

$$144 \quad RMSE = \sqrt{\frac{\sum_{i=1}^n (\hat{y}_i - y_i)^2}{n-1}} \quad (2)$$

$$145 \quad RPD = \left( \frac{\sigma}{RMSE} \right)_{prediction} = \left( \sqrt{\frac{1}{1-R^2}} \right)_{prediction} \quad (3)$$

$$146 \quad \sigma = \sqrt{\frac{1}{n} \sum_{i=1}^n (y_i - \bar{y}_i)^2}$$

$$147 \quad A = \frac{N_{correction}}{N_{total}} \times 100\% \quad (4)$$

148 Where n=numbers of samples;  $\hat{y}_i$  =actual values of sample  $i$ ;  
 149  $y_i$  =predicted values of sample  $i$ ;  $\bar{y}_i$  =mean of all the predicted values;  
 150  $N_{correction}$  =correct numbers of predicted samples;  $N_{total}$  =total numbers of  
 151 predicted samples.

152 Generally, the values of  $R^2$ , RMSEC, RMSECV and RMSEP close to 1,  
 153 0, 0 and 0 respectively, represent an accurate classification of models. And  
 154 based on the RPD statistics, the predicted accuracy of the model is  
 155 categorized into three states: accurate (RPD>2), moderate (1.4<RPD<2),  
 156 and poor (RPD<1.4) <sup>34, 35</sup>.

## 157 **3 Results and discussions**

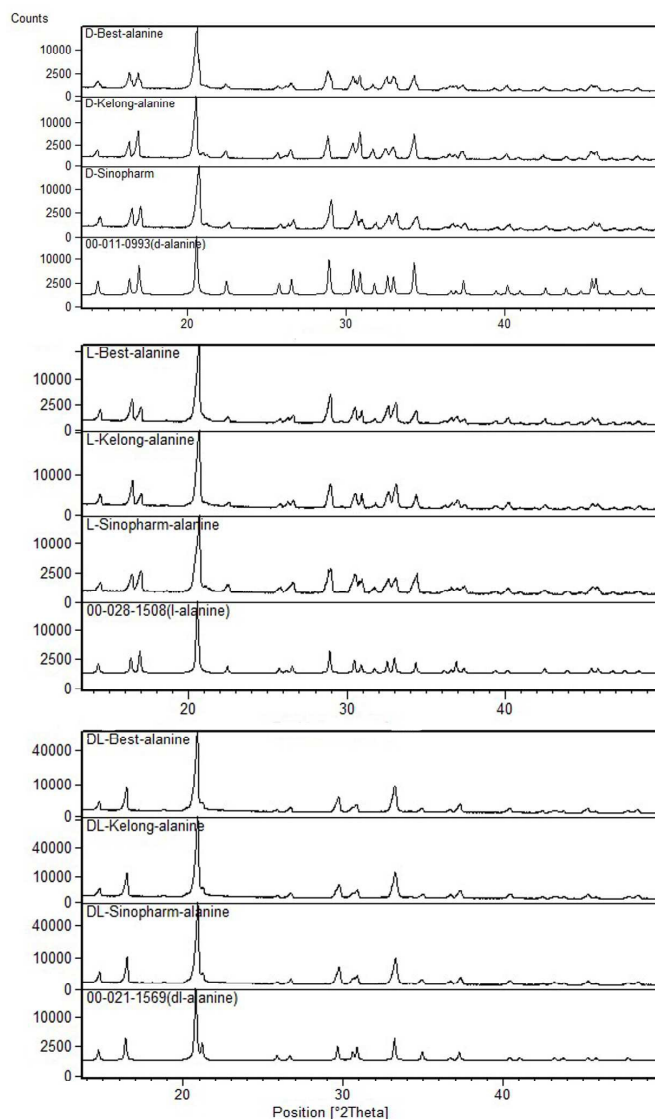
### 158 **3.1 Characterization of D-, L- and DL-alanine**

159 The nine kinds of alanine products with different chirality from three  
160 different manufactures were characterized by X-ray powder diffractometer  
161 and Differential Scanning Calorimetry (DSC) to learn their crystal structures  
162 and optical microscopy to learn their appearances.

#### 163 **3.1.1 X-ray diffraction patterns**

164 Powders of the nine kinds of alanine through 120-mesh were milled for  
165 two minutes and then measured by an X'Pert PRO diffractometer with an  
166 PIXcel 1D detector and Cu K $\alpha$ 1 radiation ( $\lambda=1.54056 \text{ \AA}$ , generator setting:  
167 40 kV, 40 mA). The X-ray diffraction data were collected at room  
168 temperature in the range of 5-50°  $2\theta$ , using a step size of 0.013°  $2\theta$  and a  
169 count time of 29 s per step. By searching and matching, the achieved X-ray  
170 diffraction patterns of the nine kinds of alanine are consistent with the  
171 standard X-ray diffraction patterns in the Powder Diffraction File (PDF)  
172 card in International Centre for Diffraction Data (ICDD) of D-, L- and  
173 DL-alanine as presented in Fig.1, and the detailed crystal information was  
174 displayed in Tab.1. As seen, firstly, they have different space groups.  
175 DL-alanine belongs to monoclinic while D- and L-alanine are orthorhombic.  
176 Secondly, in terms of unit cell, the length of a, b and c, volume (V) and

177 density ( $D_c$ ) also show difference in an extent. Even for the same chirality  
178 of alanine, after cell refinement, varieties are obtained in a, b, c, V and  $D_c$ .  
179 These difference prove that these nine kinds of alanine do have different  
180 crystal forms.



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182

Figure 1. X-ray diffraction patterns of DL-, D- and L-alanine.

183

Table 1. Crystal structure information of DL-, D- and L-alanine

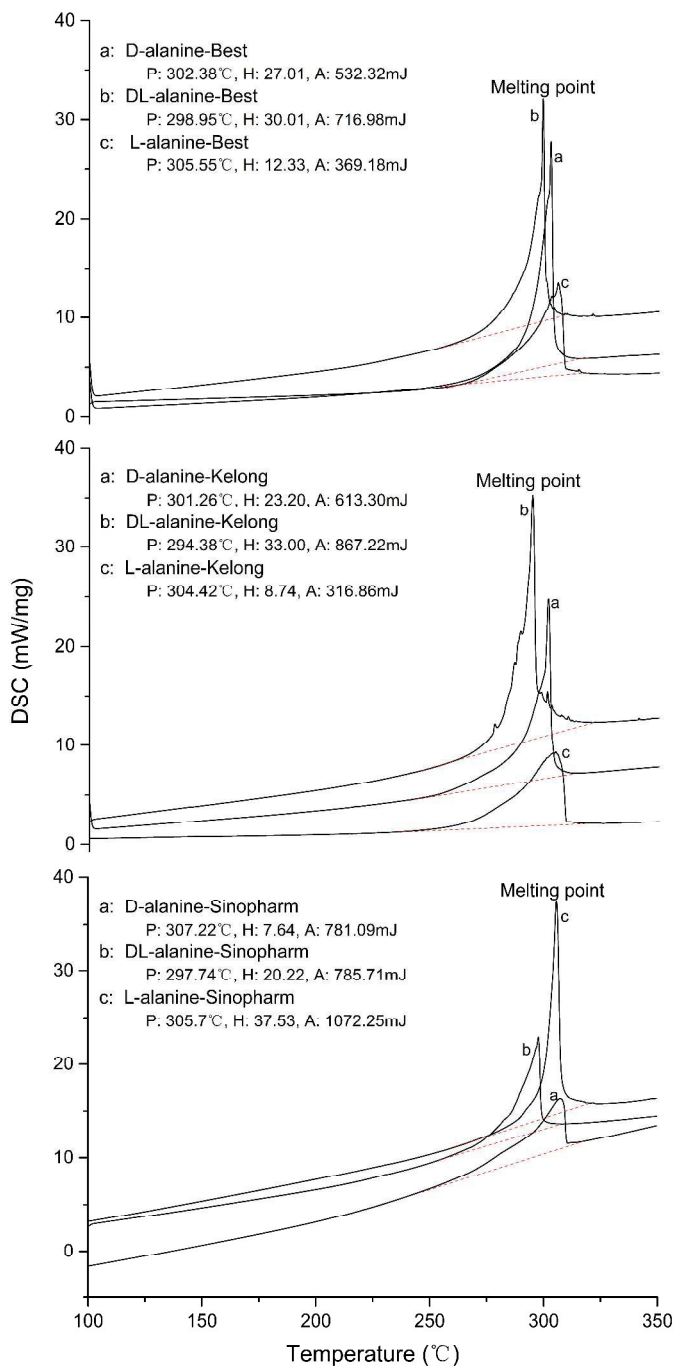
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	Alanine	Space group	a × b × c/nm	$\alpha=\beta=\gamma$	V/nm <sup>3</sup>	Dc/g.cm <sup>-3</sup>	Z
DL	PDF#-21-1569	Pna21	12.019×6.044×5.831		423.58	1.396	
	Best		12.017×6.000×6.016		433.76	1.364	
	Kelong		12.018×6.025×5.830		422.16	1.402	
	Sinopharm		12.020×5.987×5.840		420.33	1.408	
D	PDF#-11-0993	P212121	6.000×12.100×5.750		417.45	1.418	
	Best		6.020×12.102×5.778	90°	421.03	1.406	4
	Kelong		6.006×12.099×5.809		422.15	1.402	
	Sinopharm		5.995×12.050×5.788		418.13	1.415	
L	PDF#-28-1508	P212121	6.031×12.351×5.782		430.69	1.375	
L	Best		6.021×12.317×5.778		428.53	1.381	
	Kelong		6.020×12.324×5.784		429.10	1.379	
	Sinopharm		6.002×12.312×5.776		426.84	1.386	

### 185 3.1.2 Differential Scanning Calorimetry thermograms

186 The characteristics of the nine kinds of alanine were studied with  
187 Differential Scanning Calorimetry. The measurement was carried out in  
188 protecting nitrogen atmosphere with a flow rate of 20.0ml/min and a heating  
189 rate of 10°C/min within a temperature range of 100~350°C. Fig.2 shows the  
190 obtained thermograms. It is found that the nine kinds of alanine have  
191 different melting points, although they fell into the reported range of  
192 296~316°C. As for the alanine from Best and Kelong, the melting  
193 temperature of L-alanine is higher than D-alanine, while that of D-alanine is  
194 higher than DL-alanine. While as for Sinopharm, the melting point of  
195 D-alanine is higher than L-alanine. Besides, the height and area of these

196 peaks are also different in value. These differences indicate that the nine  
197 kinds of alanine have some different crystal structures.



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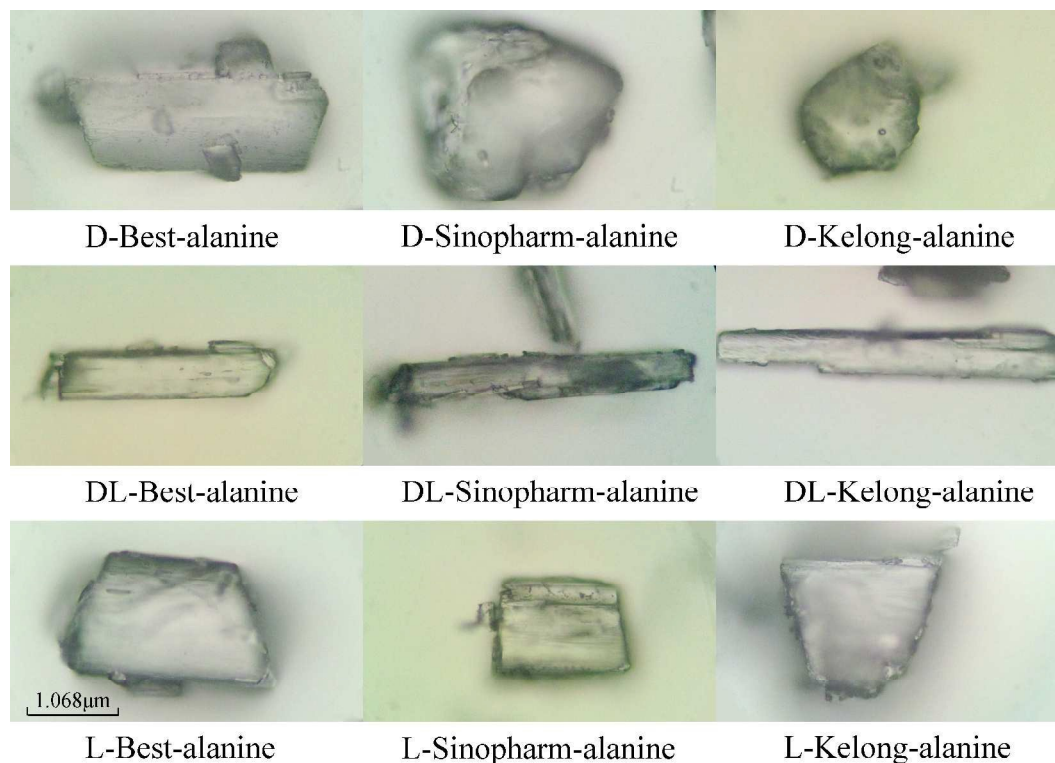
Figure 2. The DSC thermograms of DL-, D- and L-alanine.

200

(P: peak; H: peak height; A: peak area)

### 201 **3.1.3 Optical micrographs**

202       Optical microscopy was applied to look insight the micrographs of  
203 alanine products from the different manufacturers, since the size and shape  
204 of particles may differ with crystal forms and habits. The optical  
205 micrographs with  $10\times 40$  times magnification of the nine kinds of alanine  
206 products through 100-mesh were got as shown in Fig.3. Their size and  
207 shapes are different between different kinds of chirality and manufacturers.  
208 The D-alanine from Sinopharm and Kelong are near to irregular sphere  
209 while the one from Best is in the shape of cylindrical. All DL-alanine are in  
210 acicular shape besides the difference in size. As for the L-alanine, they are  
211 in a good cylinder with some difference between each other. These  
212 considerable differences demonstrate that they have different crystal habits.



213

214

Figure 3. Micrographs of DL-, D- and L-alanine with 10 $\times$ 40 times magnification.

215

All these reveal that D-, L- and DL-alanine have different crystal

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structures probably resulting in diverse absorption or reflectance of

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spectroscopy, which may allow UV-vis-SWNIR DRS to be applied for

218

classification.

219

### 3.2 Optimization of measuring conditions of UV-vis-SWNIR spectra

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Based on the Kubelka-Munk light scattering theory<sup>36,37</sup>, the absorption

221

intensities will increase along with the increasing of powder particle sizes.

222

Therefore, to establish an accurate model of discrimination in this research,

223

the effect of particle sizes was investigated with the original spectral data in

224

the region of 220-980 nm instead of 200-1100 nm in order to filter the noise

225 of machine, using the samples of D-, L- and DL-alanine from Best with the  
 226 size of 60, 80, 100 and 120-mesh.

227 45 spectral samples for each particle sizes were divided into a  
 228 calibration set and a prediction set by randomly. The calibration set had 30  
 229 spectra and prediction set had 15 spectra. The result of modeling PLS-DA is  
 230 shown in Tab.2. It indicates that the best size for UV-vis-SWNIR DRS is  
 231 100-mesh, resulted in the discrimination with a  $R^2$  of 0.9919 and RMSEP of  
 232 0.0400 for prediction.

233 Then, the distance of 3.8, 4.5, 5.3 and 6.1 mm between fiber probe and  
 234 sample was also investigated by modeling PLS-DA with original spectral  
 235 data under the condition of constant 100-mesh. The spectra of a same chiral  
 236 alanine differed very seldom with distance. The predicted result given in  
 237 Tab.2 displays that the farther distance the better performance of the model  
 238 within the distance of 4.5 mm, but beyond that there is no notable  
 239 difference.

240 The total results suggest that the optimum powder size and distance are  
 241 100-mesh and 5.3 mm, respectively.

242 Table 2. Optimization by modeling PLS-DA with UV-vis-SWNIR spectra

Optimum condition	LVs	Calibration		Cross-validation		Prediction		
Particle size (mesh)		$R^2$	RMSEC	$R^2$	RMSECV	$R^2$	RMSEP	RPD
60	6	0.9900	0.0816	0.9779	0.1216	0.9745	0.0090	6.7267



80	4	0.9801	0.1151	0.9671	0.1478	0.9511	0.4144	4.5222
120	8	0.9957	0.0533	0.9762	0.1497	0.9879	0.0124	9.0909
100	6	0.9941	0.0117	0.9861	0.0902	0.9919	0.0400	11.1111
Distance (mm)								
3.8	4	0.9284	0.2185	0.9049	0.2783	0.8658	0.5905	2.7298
4.5	6	0.9947	0.0596	0.9892	0.0896	0.9789	0.2412	6.8843
5.3	6	0.9953	0.0557	0.9905	0.0872	0.9767	0.2417	6.5512
6.1	6	0.9955	0.0550	0.9905	0.0798	0.9838	0.2750	7.8567

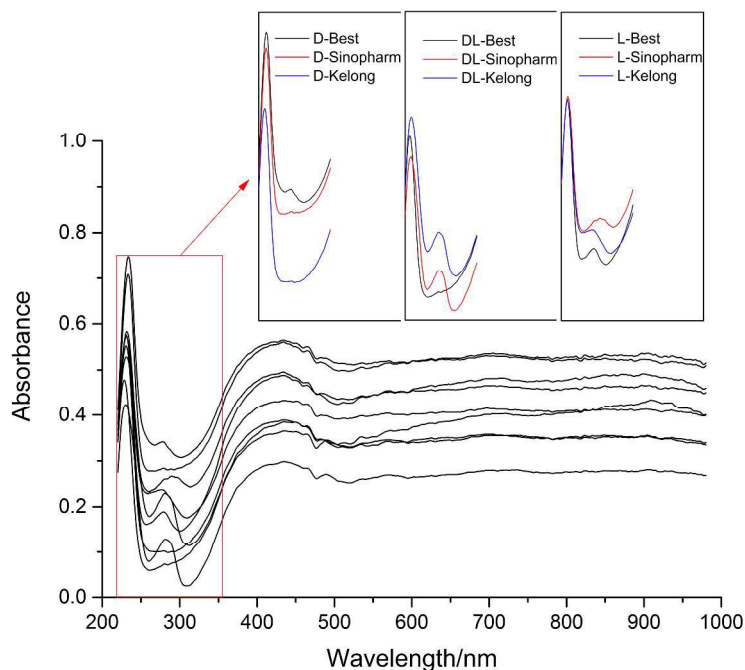
### 243 3.3 Qualitative analysis

#### 244 3.3.1 UV-vis-SWNIR spectra of nine kinds of alanine products

245 The spectra of nine kinds of alanine products were measured under the  
246 optimal conditions of 100-mesh and distance of 5.3 mm as presented in Fig.  
247 4 and each kind of alanine products has 15 spectral samples. In total, there  
248 are 135 spectra.

249 Among the spectra of nine kinds of alanine products there are big  
250 differences in the wavelength region of 227-233 nm and 250-350 nm,  
251 besides the differences in the total shifts in absorbance. The maximum peaks  
252 of alanine shifted slightly from 227 nm to 233 nm with an absorbance  
253 change from 0.4 to 0.75. Meanwhile, in the region of 250-350 nm, there was  
254 a big valley in which six kinds of alanine products have a notable peak with  
255 different peak position and absorbance, except for D-alanine from  
256 Sinopharm and Kelong and DL-alanine from Best. Although these  
257 differences are not enough in direct identification of chirality by eyes, when

258 combined with chemometrics they could be used to realize the above goal,  
259 as being proved subsequently.



260

261

Figure 4. UV-vis-SWNIR spectra of nine kinds of alanine products.

### 262 3.3.2 Classification basing on chirality

263 In this case, the nine kinds of alanine products were classified into  
264 three groups numbered with 1~3, according to their chirality but regardless  
265 of their manufacturing origins. The 135 spectra were split into a calibration  
266 set and a prediction set by randomly. The calibration set had 90 spectra and  
267 prediction set had 45 spectra. Modeling PLS-DA with several data  
268 preprocessing methods was carried out in the range of 220-980 nm. Tab.3  
269 gives the result that modeling PLS-DA without data preprocessing but just  
270 eliminating 3 outlier samples has a good discrimination with a better

271 prediction of a  $R^2$  of 0.9673 and a RMSEP of 0.1476 as shown in Fig.5.  
 272 Moreover, Table 4 and Fig.6A, B demonstrate in detail that the 9 kinds of  
 273 alanine products could be grouped into 3 classes with a 100% accuracy  
 274 according to their chirality. It turns out that it is feasible for UV-vis-SWNIR  
 275 DRS to implement the discrimination of chirality.

276 Table 3. Result of modeling PLS-DA of the 3 grouped alanine

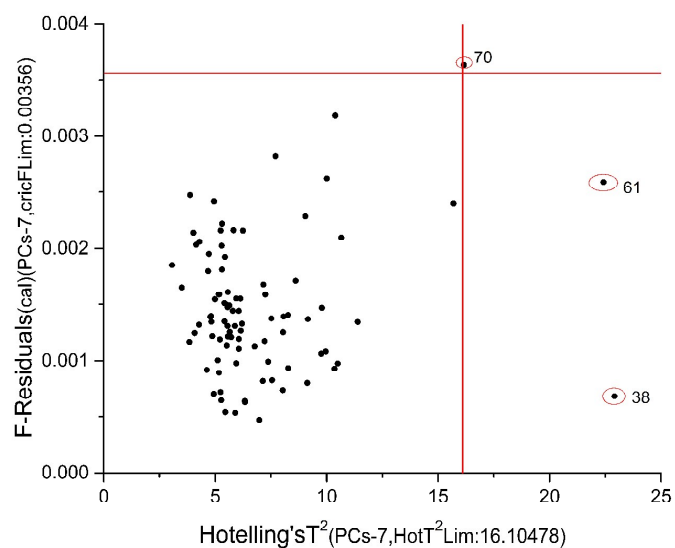
Data preprocessing		Origin	SG	SNV	MSC	Nor	Center&scale	Detrend
No eliminating outlier								
LVs		7	7	8	7	7	7	5
Calibration	$R^2$	0.9708	0.9708	0.9704	0.9610	0.9641	0.9701	0.9654
	RMSEC	0.1394	0.1396	0.1405	0.1613	0.1546	0.1411	0.1519
Cross-validation	$R^2$	0.9571	0.9570	0.9580	0.9476	0.9480	0.9553	0.9570
	RMSECV	0.1711	0.1713	0.1692	0.1889	0.1883	0.1745	0.1711
Prediction	$R^2$	0.9668	0.9667	0.9675	0.9661	0.9630	0.9657	0.9696
	RMSEP	0.1489	0.1491	0.1471	0.1504	0.1571	0.1512	0.1425
Accuracy%		100	100	100	100	100	100	100
Eliminating outlier		38,61,70	38,6,70	38,46,70	38,70	32,37,38,70	38,61,70	61
LVs		7	7	8	7	7	6	5
Calibration	$R^2$	0.9785	0.9785	0.9733	0.9640	0.9739	0.9753	0.9662
	RMSEC	0.1196	0.1196	0.1346	0.1554	0.1338	0.1283	0.1496
Cross-validation	$R^2$	0.9714	0.9714	0.9616	0.9526	0.9668	0.9653	0.9576
	RMSECV	0.1397	0.1397	0.1632	0.1804	0.1527	0.1538	0.1670
Prediction	$R^2$	0.9673	0.9670	0.9684	0.9675	0.9621	0.9620	0.9707
	RMSEP	0.1476	0.1485	0.1452	0.1472	0.1589	0.1591	0.1396
Accuracy%		100	100	100	100	100	100	100

277 Table 4. Prediction of the 3 grouped alanine by PLS-DA

samples	Numbered values	Predicted values	samples	Numbered values	Predicted values	samples	Numbered values	Predicted values
1	1	0.84	16	2	1.93	31	3	3.08
2	1	0.81	17	2	1.92	32	3	3.37
3	1	0.90	18	2	1.94	33	3	2.99
4	1	0.89	19	2	1.94	34	3	2.60
5	1	0.77	20	2	1.90	35	3	3.01

6	1	1.08	21	2	1.90	36	3	2.79
7	1	1.05	22	2	1.92	37	3	3.07
8	1	1.06	23	2	1.89	38	3	2.95
9	1	1.07	24	2	1.93	39	3	3.23
10	1	1.12	25	2	1.88	40	3	2.88
11	1	1.01	26	2	2.14	41	3	2.97
12	1	1.09	27	2	2.23	42	3	3.04
13	1	1.10	28	2	2.15	43	3	2.96
14	1	1.19	29	2	2.26	44	3	3.10
15	1	1.04	30	2	2.26	45	3	2.88
Threshold		1±0.5			2±0.5			3±0.5
Correct numbers <sup>a</sup>					45			
Accuracy%					100			

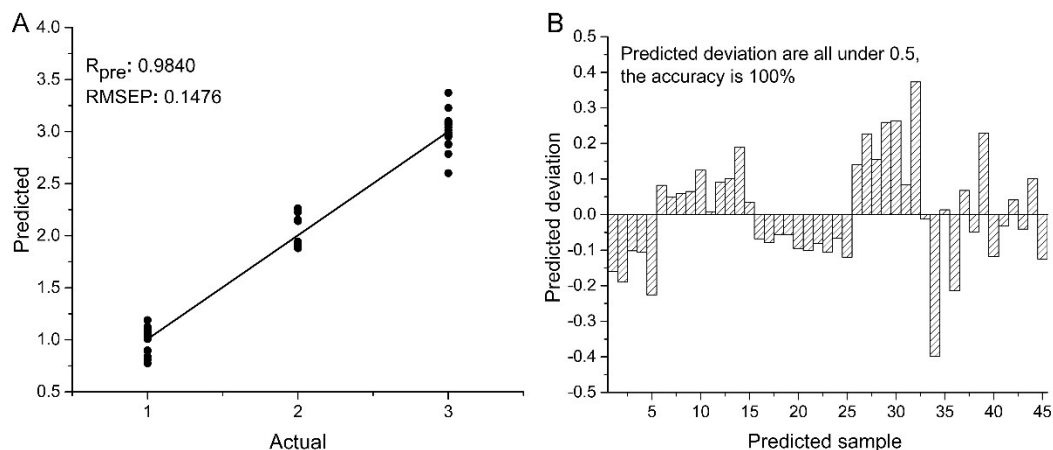
278 <sup>a</sup> Correct numbers means the correct numbers of predicted samples.



279

280

Figure 5. Diagram of outlier samples of the 3 grouped alanine.



281

282 Figure 6. Results of prediction of the 3 grouped alanine: A. diagram of actual and

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predicted values, B. deviation map of predicted samples.

284 **3.3.3 Classification basing on both chirality and manufacturing origins**

285 To see whether the observed differences in spectra are due to their  
 286 chirality or to different habits of D- and L-alanine in solid, classification  
 287 basing on both chirality and manufacturers was carried out. All alanine  
 288 samples were divided into nine groups according to both their chirality and  
 289 manufacturers, numbered with 1~9. The following data processing was the  
 290 same as the procedure mentioned above. As shown in Tab.5, the result of  
 291 modeling PLS-DA indicates that MSC has a better prediction with a  $R^2$  of  
 292 0.9787 and RMSEP of 0.3770 by eliminating 2 outlier samples shown in  
 293 Fig.7. The detailed information in Tab.6 and Fig.8A, B show that the  
 294 accuracy of the nine groups just reach 86.67% with six wrong  
 295 discrimination which is not very well compared with the 3 grouped alanine.

296

Table 5. Result of modeling PLS-DA of the 9 grouped alanine

Data preprocessing		Origin	SG	SNV	MSC	Nor	Center&scale	Detrend
No eliminating outlier								
LVs		7	7	7	7	6	7	6
Calibration	R <sup>2</sup>	0.9761	0.9760	0.9723	0.9769	0.9699	0.9755	0.9693
	RMSEC	0.3988	0.4003	0.4296	0.3926	0.4478	0.4040	0.4522
Cross-validation	R <sup>2</sup>	0.9652	0.9650	0.9627	0.9687	0.9517	.09635	0.9607
	RMSECV	0.4871	0.4886	0.5046	0.4619	0.5737	0.4990	0.5178
Prediction	R <sup>2</sup>	0.9768	0.9766	0.9797	0.9793	0.9738	0.9761	0.9715
	RMSEP	0.3932	0.3947	0.3681	0.3717	0.4183	0.3996	0.4359
Accuracy%		84.44	84.44	80	84.44	82.22	84.44	73.33
Eliminating outlier		38,61,70	38,61,70	38,70	38,70	21,32,37,38,70	38,61,70	37,38,89
LVs		7	7	7	7	6	7	
Calibration	R <sup>2</sup>	0.9800	0.9799	0.9748	0.9780	0.9744	0.9796	0.9617
	RMSEC	0.3683	0.3694	0.4131	0.3854	0.4215	0.3721	0.5063
Cross-validation	R <sup>2</sup>	0.9709	0.9708	0.9669	0.9711	0.9635	0.9696	0.9536
	RMSECV	0.4496	0.4506	0.4784	0.4471	0.5092	0.4596	0.5633
Prediction	R <sup>2</sup>	0.9760	0.9738	0.9760	0.9787	0.9611	0.9762	0.9609
	RMSEP	0.3997	0.4182	0.4004	0.3770	0.5095	0.3980	0.5105
Accuracy%		84.44	84.44	77.78	86.67	66.67	84.44	64.44

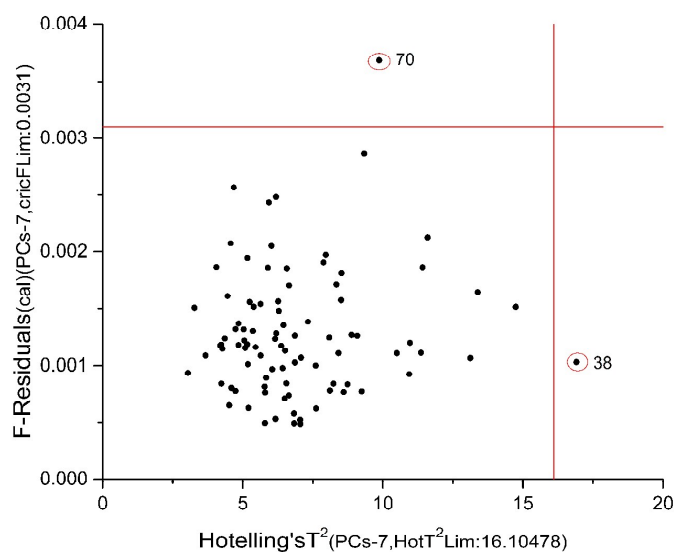
297

Table 6. Prediction of the 9 grouped alanine by PLS-DA

Samples	Numbered values	Predicted values	Threshold	Samples	Numbered values	Predicted values	Threshold
1	1	0.55	1±0.5	24	5	5.16	5±0.5
2	1	0.67		25	5	4.96	
3	1	0.81		26	6	6.21	6±0.5
4	1	0.54		27	6	6.33	
5	1	0.91		28	6	6.03	
6	2	1.81	2±0.5	29	6	6.43	
7	2	2.13		30	6	6.53	
8	2	2.07		31	7	7.06	7±0.5
9	2	2.30		32	7	7.52	
10	2	1.86		33	7	6.89	
11	3	2.52	3±0.5	34	7	6.05	
12	3	2.72		35	7	6.90	
13	3	2.92		36	8	7.23	8±0.5
14	3	2.75		37	8	8.26	
15	3	3.05		38	8	7.92	
16	4	4.36	4±0.5	39	8	8.89	
17	4	4.08		40	8	7.56	

18	4	4.47		41	9	8.85	9±0.5
19	4	4.09		42	9	9.37	
20	4	4.149		43	9	8.86	
21	5	4.96	5±0.5	44	9	9.36	
22	5	5.24		45	9	7.99	
23	5	4.94					
Correct numbers				39			
Accuracy%				86.67			

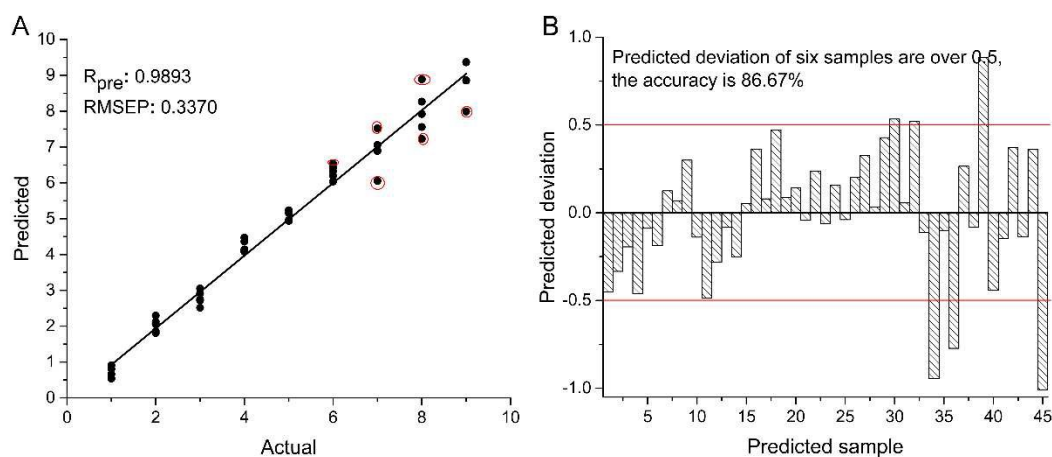
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Figure 7. Diagram of outlier samples of the 9 grouped alanine.



301

302

Figure 8. Results of prediction of the 9 grouped alanine: A. diagram of actual and

303

predicted values, B. deviation map of predicted samples.

### 304 **3.3.4 Comparison of classification by chirality with by manufacturers**

305 The above results that the accuracy of 100% obtained when the alanine  
306 was divided into 3 groups is much better than that of 86.67% when grouped  
307 into 9 groups imply that UV-vis-SWNIR DRS with chemometrics is more  
308 sensitive to chirality than to manufacturing origins. In other words, other  
309 than manufacturing origins, the chirality is the preferential factor for the  
310 difference of spectra between enantiomers and racemates of alanine.

### 311 **3.4 Inference of Principle**

312 It is well known that UV-vis spectra of enantiomers are the same in  
313 non-chiral media or solutions. But the situation in UV-vis reflectance  
314 spectra of alanine enantiomers in powder are different as observed in this  
315 work. This is likely due to their different crystal structures and habits. It also  
316 has been reported that different polymorph can display different  
317 ultraviolet-visible spectra<sup>38</sup>, different crystal structures can cause different  
318 UV-vis-SWNIR diffuse reflectance spectra<sup>33</sup>. Here a preliminary inference  
319 is made that UV-vis-SWNIR diffuse reflectance spectra of enantiomers in  
320 pure solid state may be different, because different configuration of  
321 enantiomer pairs can result in different crystal forms and habits. Although  
322 the differences are small, it forms the basis to apply UV-vis-SWNIR DRS  
323 combined with chemometrics on the identification of enantiomers.



#### 324 **4 Conclusion**

325 The feasibility of UV-vis-SWNIR DRS for qualitative analysis of  
326 enantiomers and racemates of chiral compound in solid state was first  
327 investigated and proved by using D-, L- and DL-alanine from three different  
328 manufactures as model compounds. The classification based on chirality has  
329 an accuracy of 100% which is better than that of 86.67% based on  
330 manufactures. It is obvious that UV-vis-SWNIR DRS is more sensitive to  
331 chirality of alanine. Hence, UV-vis-SWNIR diffuse reflectance spectroscopy  
332 as a novel qualitative analysis method for chiral compounds is proposed. It  
333 is simply sample-preprocessing, rapid, convenient and inexpensive and can  
334 be expected to apply in pharmaceutical industry. More work is needed for its  
335 development and better understanding of the effect of molecular  
336 configuration on diffuse reflectance spectra.

337

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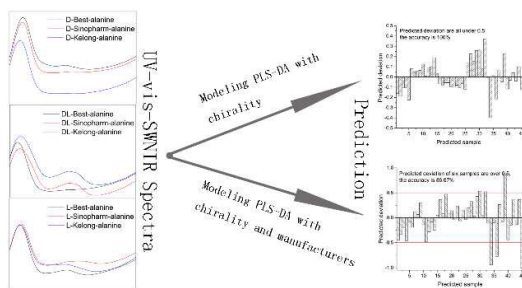
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UV-vis-SWNIR DRS is firstly and successfully used to discriminate the chirality of compounds in powder with a good accuracy.