Journal of Materials Chemistry B

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.





Journal of Material Chemistry B

ARTICLE

A Layered Drug Nanovehicle toward Targeted Cancer Imaging and Therapy

Received 00th January 20xx, Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/

Shanyue Guan,^a Ruizheng Liang,*^a Chunyang Li,^a Dan Yan,*^b Min Wei,*^a David G. Evans^a and Xue Duan^a

A layered drug nanovehicle was fabricated *via* the co-intercalation of doxorubicin (DOX) and folic acid (FA) into the gallery of layered double hydroxides (LDHs). This supermolecular nanovehicle (denoted as DOX-FA/LDHs) demonstrates excellent fluorescence imaging and targeted therapy toward cancer cells. The nanovehicle shows uniform platelet morphology with average diameter of ~171 nm. The unique host-guest interactions lead to a high dispersion of DOX, and *in vitro* tests reveal a legible and strong fluorescence imaging for the sample of DOX-FA/LDHs. In addition, the DOX-FA/LDHs material produces a high anticancer activity toward HepG2 cells but rather low cytotoxicity to the normal cells (LO2 cells), as a result of the over-expression of FA to cancer cells. This work provides a facile approach for the design and preparation of drug nanovehicle with significantly enhanced biocompatibility, diagnosis and targeted therapy, which can be potentially applied in medical imaging and chemotherapy.

Introduction

Over the decades, as one of the most important therapy methods, chemotherapy has been extensively applied in the cancer treatment, owing to its high efficiency and convenience compared with other treatment approaches. 1,2 Howerer, conventional chemotherapeutic agents normally show the following disadvantages: (a) the non-specific recognition, (b) poor biocompatibility and solubility, (c) inefficient cellular internalization.^{3,4} This often leads to the non-specific uptake by the normal cells and causes obvious side effects. 5-7 In order to resolve the issues mentioned above, incorporation of chemotherapeutic agents within a suitable nanocarrier is a preferred alternative. To date, multifunctional nanocarriers have attracted considerable attention as a platform to deliver and release drugs. For instance, Fe₃O₄ nanoparticles (NPs), ^{8,9} SiO₂ NPs, ^{10,11} Au NPs, ¹²⁻¹⁶ block polymers ¹⁷⁻²⁰ and micelles ²¹⁻²⁴ have been widely studied as drug carriers for cancer therapy. However, they generally suffer from complicated preparation process, low efficiency of uptake or structure uncontrollability. As a result, combination of diagnosis and targeting therapy into one system with facile preparation and superior biocompatibility remains a challenge goal.^{25,26}

Layered double hydroxides (LDHs) is one kind of inorganic

generally lavered material expressed $_xM^{3+}_{X}(OH)_2](A^{n-})_{x/n}\cdot mH_2O$, which consist of cationic brucite-like layers and exchangeable interlayer anions. 27-29 By virtue of this unique structure, they have been widely explored as inorganicbiology composite materials for drug/gene delivery. 30-35 Moreover, due to the electropositivity of LDHs, it tends to accelerate cellular internalization and improve cellular uptake of drugs. Doxorubicin (DOX) is an approved chemotherapeutic drug in current clinical applications while folic acid (FA) possesses the targeting capability owing to its over-expression toward cancer cells. 15,36,37 Therefore, the incorporation of DOX and FA into the interlayer gallery of LDHs would exhibit the following advantages: (i) the host-guest interactions can improve the stability and hydrophilicity of DOX, resulting in an enhancement in drug permeability/retention;³⁸ (ii) the intrinsic fluorescence of DOX³⁹ endows the composite material with the capability of fluorescence imaging. In addition, the targeting ability of FA toward cancer cells would increase the drug uptake at the cancerous site and depress cytotoxicity to normal cells.

In this work, we report a layered anti-cancer nanovehicle DOX-FA/LDHs by incorporation of DOX and FA into the LDHs gallery, which shows extraordinarily good anticancer behavior, low cytotoxicity, as well as excellent targeted ability. XRD and UV-vis spectroscopy confirm that DOX and FA molecules are co-intercalated in the interlayer region of LDHs matrix successfully. The DOX-FA/LDHs material displays uniform nanoplatelet morphology with an average diameter of ~171 nm. *In vitro* tests performed with HepG2 cells demonstrate both a fluorescence imaging and anticancer performance of DOX-FA/LDHs, with a rather low cytotoxicity to the normal

[&]quot;State Key Laboratory of Chemical Resource Engineering, Beijing University of Chemical Technology, Beijing 100029, P. R. China.

UV-vis spectroscopy co-intercalated in successfully. The

^{*} E-mail: liangruizheng2000@163.com; weimin@mail.buct.edu.cn; Fax: (+86)10-6442-5385

^b Beijing Shijitan Hospital Capital Medical University Beijing, 100038, P. R. China *E-mail: <u>yd277@126.com</u>

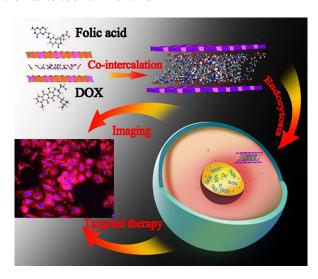
Electronic Supplementary Information (ESI) available. See DOI: 10.1039/x0xx00000x

ARTICLE Journal Name

cells (performed with LO2 cells). Moreover, the DOX-FA/LDHs nanovehicle displays a high storage stability, good biocompatibility and targeting capability, which would guarantee its practical applications.

Results and discussion

Monodisperse DOX-FA/LDHs was synthesized by the separate nucleation and aging steps (SNAS) method⁴⁰ reported by our group, via co-intercalation of DOX and FA into the gallery of LDHs. Scheme 1 illustrates the fabrication and cancer cells mediated endocytosis of DOX-FA/LDHs nanovehicle. In order to determine the optimal DOX loading, a series of DOX(x%)-FA(80%)/LDHs composite materials were prepared and their XRD patterns are shown in Fig. 1a. For the LDH precursor, its XRD pattern reveals a series of (00/) reflections with the (003) basal spacing of 0.73 nm (2 θ 12.14°), indicating a typical CO_3^{2-} LDHs (Fig. S1a). After the co-incorporation of DOX and FA into the LDHs gallery, the (003) reflection of DOX(x%)-FA(80%)/LDHs composites moves from 2θ 12.14° to the low angle (3.86°-4.82°) and the corresponding basal spacing expands from 0.73 nm to 1.83-2.28 nm (Table S1), indicating the intercalation of FA and DOX.



Scheme. 1 Schematic illustration of DOX-FA/LDHs composite material as a nanovehicle for imaging and therapy.

The co-intercalation of FA and DOX was studied by the FT-IR spectra (Fig. S1b). Compared with the spectrum of pristine LDHs, the new band at 1611 cm $^{-1}$ is attributed to the –C=N stretching vibration of the pterin ring in FA 41 while the bands at 1184 cm $^{-1}$ and 1497 cm $^{-1}$ correspond to the $\delta(\text{CH}_3\text{O}-)$ and $\delta(\text{N}-\text{H})$ stretching vibration of DOX, which indicates the conjugation of FA and DOX. The intensity of DOX characteristic bands (1184 and 1497 cm $^{-1}$) increases gradually along with the enhancement of DOX loading (from 0.5% to 3%). Furthermore, the chemical compositions of these DOX(x%)-FA(80%)/LDHs composites were measured by inductively coupled plasma atomic emission spectroscopy (ICP-AES) and listed in Table S2.

SEM images (Fig. 1c and Fig. S2) show that the DOX(x%)-FA(80%)/LDHs samples display uniform plate-like morphology. The hydrodynamic diameter (Fig. S3, Table S3) of these samples is also determined and the particle size gradually increases along with the enhancement of DOX loading from 0.5% to 3%. It has been reported that the clathrin-mediated endocytosis of LDHs nanovehicle is forbidden if the particle size is beyond ~300 nm. ⁴² Taking into account the colloid stability and cellular uptake, the sample of DOX(2%)-FA(80%)/LDHs (~171 nm) was chosen for the following study. HRTEM image of DOX(2%)-FA(80%)/LDHs (Fig. 1d) shows an uniform plate-like morphology, with the lattice fringe of 0.15 nm attributed to the (110) plane of an LDHs phase.

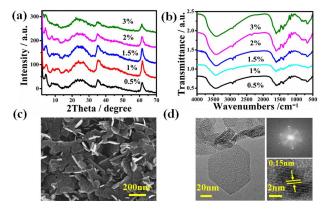


Fig. 1 (a) XRD patterns and (b) FTIR spectra of DOX(x%)-FA(80%)/LDHs samples (x ranges from 0.5% to 3%). (c) SEM image and (d) HRTEM image of DOX(2%)-FA(80%)/LDHs; the insets in (d) display the lattice fringe and Fourier transform image.

The intercalated structure was further investigated by the UV-vis absorption spectrometry (Fig. 2a). The pristine DOX shows a broad UV absorption with characteristic peaks at 233 nm and 488 nm. After the intercalation into LDHs gallery, an obvious red-shift is observed with the typical absorption peak at 556 nm. The red-shift can be ascribed to the collapse of the aggregation state of DOX owing to the host-guest interaction. For the DOX-FA/LDHs sample, an additional peak at 285 nm is observed compared with DOX/LDHs, originating from the absorption of FA. Owing to the inherent fluorescence properties of DOX, both DOX/LDHs and DOX-FA/LDHs are endowed with fluorescence capability, and Fig. 2b displays the photoluminescence (PL) spectra of DOX, DOX/LDHs and DOX-FA/LDHs with the same DOX loading. It should be noted that the fluorescence intensity of DOX/LDHs and DOX-FA/LDHs decreases after the intercalation relative to pristine DOX, as a result of the variation of microenvironment. Moreover, the Zeta potential of these samples was measured in aqueous solutions. The DOX/LDHs is positively-charged with a Zeta potential of +21.2 mV; while DOX-FA/LDHs is negativelycharged with -24.1 mV (Fig. 2c), indicating the additional loading of FA anions. For the DOX/LDHs, it undergoes passive type of internalization based on EPR effect, followed by clathrin-mediated endocytosis similar to previously reported Journal Name ARTICLE

drug-LDH hybrids.⁴³ The positively charged DOX/LDHs nanoparticles are favorable for the uptake of cells with negatively charged cell membrane, accounting for its passive targeting ability. However, the DOX-FA/LDHs nanovehicle conjugated with folic acid can be internalized inside the cell via a folate receptor (FR)-mediated active cancer targeting followed by cell uptake. 44 Thus, the surface negative charge of DOX-FA/LDHs may impose somewhat influence but can not significantly affect its cellular uptake. In addition, the sample of DOX-FA/LDHs undergoes a slow and controlled release of DOX with a 69.44% release ratio within 48 h (Fig. S4). We further investigated the stability of DOX and DOX-FA/LDHs by UV-vis absorption spectroscopy (Fig. 2d) at various time points. The absorbance of pristine DOX decreases sharply from 1 day to 15 days, and a loss of 62% is found. In contrast, for the DOX-FA/LDHs sample, it presents a relatively high stability at room temperature, with a slow decline of 35% after 15 days. Therefore, the intercalation of DOX and FA into the LDHs gallery can improve its stability to a large extent, which is desirable for storage and further clinical application.

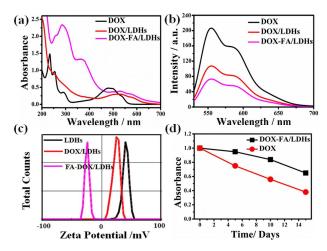


Fig. 2 (a) UV-vis absorption spectra and (b) photoluminescence spectra of DOX, DOX/LDHs and DOX-FA/LDHs. (c) Zeta potential of LDHs, DOX/LDHs and DOX-FA/LDHs. (d) Stability tests of DOX and DOX-FA/LDHs.

In order to shed light on the orientation and arrangement of the co-intercalated DOX and FA in LDHs at atomic-level, molecular dynamics (MD) simulations were performed over the sample of DOX(2%)-FA(80%)/LDHs. The structural model contains 96 Mg and 48 Al so as to keep the charge balance, and Fig. 3 displays the model geometry after the MD simulation. The simulated d_{003} value is 2.10 nm, in approximate agreement with the XRD data (2.28 nm). According to the simulated results, the DOX and FA molecules are slantwise oriented in the interlayer region, and the tilt angles of the tail vectors (as defined in Fig. S5) in DOX and FA with respect to the host layer are calculated to be θ_1 = 45.92° and θ_2 = 70.40°, respectively. The distance between the carboxyl group in FA and the hydroxyl group in DOX is as short as \sim 2.0 Å (Fig. 3b), within the interaction range of hydrogen bonding.

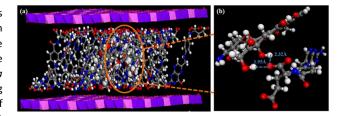


Fig. 3 The model geometry of DOX(2%)-FA(80%)/LDHs after MD simulation.

Efficient internalization of cancer cells toward the drug is essential for the cancer therapy, and the drug intake was firstly investigated through fluorescence imaging. HepG2 cells were used to incubate with LDHs, DOX, DOX/LDHs, DOX-FA/LDHs respectively for 24 h, and their fluorescence images were recorded. As shown in Fig. 4a-c, no fluorescence is observed when the cells are treated with pristine LDHs. For pure DOX, a weak red signal is detected (Fig. 4e), indicating the intake of DOX by HepG2 cells. As shown in Fig. 4f, the overlapped red fluorescence signal of DOX and the blue signal of DAPI can be detected, manifesting that DOX is located in the cell cytoplasm. In the case of DOX/LDHs, a relatively strong fluorescence intensity is observed, indicating an effective cell uptake of the DOX/LDHs (Fig. 4g-i). In addition, compared with pristine DOX and DOX/LDHs, the fluorescence intensity of DOX-FA/LDHs is significantly heightened with further introduction of FA (Fig. 4j-l), indicating the best intake performance owing to the targeting ability of FA toward HepG2 cells. A similar result is obtained in KB cancer cells (Fig. S6), demonstrating the targeting uptake ability of DOX-FA/LDHs.

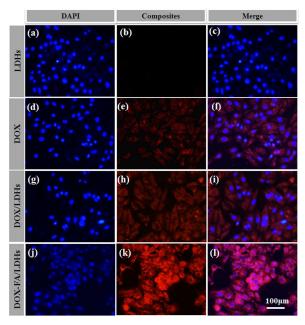


Fig. 4 Fluorescence imaging of HepG2 incubated with LDHs, DOX, DOX/LDHs and DOX-FA/LDHs, respectively.

ARTICLE Journal Name

The anticancer activity of DOX-FA/LDHs was further studied by in vitro tests performed with HepG2 cells and normal liver LO2 cells. The HepG2 cells were incubated with DOX, DOX(2%)/LDHs and DOX(2%)-FA(80%)/LDHs with equivalent DOX concentration ranging from 1.25 to 20 µg/mL for 24 h, washed thoroughly with PBS, and then determined by the MTT assay. As shown in Fig. 5a, a significant anticancer effect occurs and enhances gradually along with the increase of dosage from 1.25 to 20 $\mu g/mL$. The half maximal inhibitory concentration (IC₅₀) of DOX is 4.36 μ g/mL, which is less than that of DOX(2%)/LDHs (19.21 μ g/mL), and DOX(2%)-FA(80%)/LDHs (7.14 μg/mL). This is due to the rapid uptake of DOX while both the DOX(2%)/LDHs and DOX(2%)-FA(80%)/LDHs undergo an uptake-delivery-release procedure. Moreover, compared with the HepG2 cells treated with DOX(2%)/LDHs (IC₅₀ = 19.21 μ g/mL), the anticancer efficacy of DOX(2%)-FA(80%)/LDHs (IC₅₀ = $7.14 \mu g/mL$) is remarkably enhanced via conjugation with FA, demonstrating the effectiveness of FA toward specifically targeting HepG2 cells. Furthermore, we investigated the cytotoxicity of the DOX, DOX(2%)/LDHs and DOX(2%)-FA(80%)/LDHs with various drug concentration toward LO2 cells. Based on the MTT assay, the viability reveals that LO2 cells show a strong tolerance to DOX(2%)-FA(80%)/LDHs with an IC $_{50}$ as high as 22.81 µg/mL, which is 3.19 times larger than that of HepG2 cells (7.14 μg/mL). In contrast, the pristine DOX displays a much stronger cytotoxicity toward LO2 cells (IC $_{50}$ = 1.31 $\mu g/mL$) than HepG2 cells (IC₅₀ = $4.36 \mu g/mL$). Therefore, the sample of DOX(2%)-FA(80%)/LDHs exhibits an enhanced biocompatibility compared with pristine DOX.

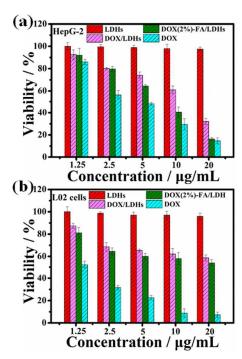


Fig. 5 Viability tests of (a) HepG2 and (b) L02 treated with LDHs, DOX/LDH, DOX(2%)-FA(80%)/LDHs and DOX for 24 h.

In order to make a comparison of DOX, DOX(2%)/LDHs and DOX(2%)-FA(80%)/LDHs, specific anticancer efficacy (SAE) is denoted as the difference value of viability between LO2 and HepG2 cells. The DOX(2%)-FA(80%)/LDHs drug exhibits a rather large SAE (17.71%), superior to that of the DOX(2%)/LDHs (-1.15%) and DOX (-20.71%) with the equivalent DOX concentration of 10 µg/mL. The results verify that DOX(2%)-FA(80%)/LDHs plays a positive role in anticancer activity while pristine DOX shows an obvious side-effect. We further investigated the reason for the different viability of HepG2 and LO2 cells toward DOX(2%)-FA(80%)/LDHs, by monitoring the cell uptake of drug at different time point. As shown in Fig. S7, the fluorescence intensity of HepG2 cells is much stronger than that of LO2 cells after 3 h incubation, indicating the fast internalization of HepG2 cells toward DOX(2%)-FA(80%)/LDHs. Moreover, DOX(2%)-FA(80%)/LDHs shows a time-dependent internalization behavior by HepG2 cells, and the accumulation maximum is observed after 24 h. In contrast, the fluorescence intensity of LO2 cells decreases significantly after 24 h incubation. The results above demonstrate that the overexpression of FA receptor toward HepG2 cells promotes the uptake of DOX(2%)-FA(80%)/LDHs, accounting for its excellent anticancer effectiveness.

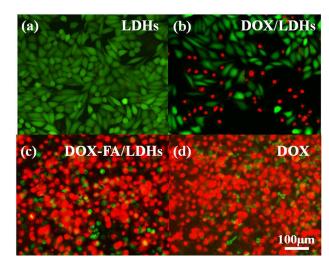


Fig. 6 Fluorescence imaging of LDHs, DOX/LDHs, DOX-FA/LDHs and DOX, respectively. Live/dead HepG2 cells are green/red (Calcein AM/PI), respectively.

To visualize the anticancer effect of drug, the dead and live HepG2 cells were stained with propidium iodide (PI) and Calcein-AM, respectively. The fluorescence microscopy image shows that HepG2 cells treated with LDHs (Fig. 6a) display no obvious apoptosis after 24 h incubation. The addition of DOX/LDHs (10 μ g/ml) causes partial apoptosis (Fig. 6b), indicating a weak anticancer performance. In contrast, the HepG2 cells treated with DOX(2%)-FA(80%)/LDHs (10 μ g/ml) exhibit an intense PI signal (Fig. 6c), demonstrating a predominant apoptosis after 24 h incubation. HepG2 cells treated with DOX (10 μ g/ml) also display a similar PI signal (Fig. 6d). Although DOX(2%)-FA(80%)/LDHs and DOX exhibit close

Journal Name ARTICLE

anticancer effectiveness, the former shows a weak cytotoxicity and largely enhanced biocompatibility, as a result of the incorporation of LDHs nanovehicle.

Conclusions

In summary, a supermolecular nanovehicle based on the intercalation of FA and DOX into the LDHs gallery was successfully fabricated. The DOX-FA/LDHs shows a platelet-like morphology with particle size of ~171 nm. *In vitro* experiments show that the DOX-FA/LDHs exhibits good imaging property and obviously increased targeting uptake. Moreover, the DOX-FA/LDHs nanovehicle produces a strong suppression on the proliferation of HepG2 cells while a highly decreased toxicity to the normal LO2 cells as a result of the over-expression of FA toward HepG2 cells. The enhanced biocompatibility would guarantee its application in cancer imaging and therapy.

Experimental

Reagents and Materials

DOX and FA were purchased from Sigma-Aldrich Company. Analytical grade chemicals including $Mg(NO_3)_2 \cdot 6H_2O$, $Al(NO_3)_3 \cdot 9H_2O$, NaOH, acetone and ethanol were purchased from Aladdin company and used without further purification. DMEM (Dulbecco's modified eagle medium), FBS (Fetal bovin serum), PBS (Phosphate buffer solution) were purchased from Beijing Solarbio Science and Technology Company. Deionized water was utilized throughout the whole experimental processes.

Synthesis of the composite

DOX-FA/LDHs composite was prepared according to the SNAS method reported by our group previously. 31 Typically, 40 ml of solution A (Mg(NO₃)₂·6H₂O: 0.003 mol, Al(NO₃)₃·9H₂O: 0.0015 mol and DOX: 3.0×10^{-5} mol) and 40 ml of solution B (NaOH: 0.01 mol and FA: 0.0012 mol) were simultaneously added to a colloid mill with rotor speed of 3000 rpm and mixed for 1 min. The colloid suspension was transferred into a Teflon-lined stainless steel autoclave. After hydrothermal treatment at 100 °C for 24 h, the product was centrifuged 3 times with deionized water and ethanol, respectively, followed by drying in the oven at 60 °C overnight.

Fluorescence imaging studies of the composites

HepG2 cells were grown and expanded in Dulbecco's modified Eagle's medium (DMEM) culture medium supplemented with 10% fetal bovine serum (FBS) and 1% penicillin-streptomycin at 37 °C under a 5% CO $_2$ atmosphere. After reaching 80–90% confluence, the HepG2 cells were washed with PBS, afterwards detached from the flask by addition of 1.0 mL of 0.25% trypsin for 1–3 min at 37 °C. To study the cellular uptake of DOX, DOX/LDHs and DOX(2%)-FA(80%)/LDHs, HepG2 cells (1×10^4 cells/well) were firstly seeded in a 96-well plate and cultured in a humid 5% CO $_2$ atmosphere for 24 h at 37 °C. Then pristine DOX, DOX/LDHs and DOX(2%)-FA(80%)/LDHs (equivalent DOX:

 $10~\mu g/mL)$ were added into the wells and further incubated for 24 h. Subsequently, the cells were washed three times with PBS followed by stained with 3 mg/mL DAPI for 20 min and further washing with PBS. Finally, a fluorescence microscopy was used to determine the drug uptake through the fluorescence intensity of DOX.

In vitro cell assay

To study the integrated performance of DOX, DOX/LDHs and DOX(2%)-FA(80%)/LDHs, HepG2 cells and L02 cells were used to evaluate their anti-cancer activity and cytotoxicity respectively. Specifically, HepG2 cells and L02 cells $(1\times10^4~{\rm cells/well})$ were seeded into two 96-well plates respectively. After the incubation of 24 h, pristine DOX, DOX/LDHs and DOX(2%)-FA(80%)/LDHs with concentration ranging from 1.25–20 µg/mL were added into the wells and further incubated for 24 h. After washing three times with PBS, the colorimetric 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) was used to determine the cell viability.

Sample Characterization

Powder X-ray diffraction (XRD) patterns were recorded by a Rigaku XRD-6000 diffractometer, using Cu K α radiation (λ = 0.15418 nm) at 40 kV, 30 mA, with the step of $0.04^{\circ}/2\theta$ in the range from 3 to 70°. UV-vis absorption spectra were collected in the range from 200-700 nm on a Shimadzu U-3000 spectrophotometer, with the slit width of 1.0 nm. The photoluminescence spectra were tested on a RF-5301PC fluorospectrophotometer with the excitation wavelength at 490 nm. Zeta potential and dynamic lighting scatting (DLS) diameter were conducted with photon spectroscopy (PCS, Nanosizer Nano ZS. Instruments). The chemical compositions were measured by inductively coupled plasma (ICP) emission spectroscopy (Shimadzu ICPS-7500). The morphology of composites was investigated by scanning electron microscope (SEM; Zeiss SUPRA 55) with the accelerating voltage of 20 kV. Transmission electron microscopy (TEM) images were recorded with JEOL JEM-2100 high resolution transmission electron microscopes; the accelerating voltage was 200 kV. The Fourier transform infrared spectra (FT-IR) were obtained using a Vector 22 (Bruker) spectrophotometer using the KBr pellet technique in the range 4000-400 cm⁻¹ with 2 cm⁻¹ resolution. The fluorescence imaging photograph was obtained by Nikon Ti-s fluorescence microscope with 40 folds enlargement.

Supporting Information

XRD pattern and FT-IR spectra are displayed in Fig. S1. The SEM images and DLS results are shown in Fig. S2 and Fig. S3. Distributions of tilt angle θ_1 of DOX and θ_2 of FA with respect to hydroxide sheets in the MD simulation models are presented in Fig. S5. Fluorescence images of L02 and HepG2 incubated with DOX-FA/LDHs for 3 h and 24 h are indicated in Fig. S7.

ARTICLE Journal Name

Acknowledgements

This work was supported by the 973 Program (Grant No. 2014CB932103), the National Natural Science Foundation of China (NSFC), the Innovation and Promotion Project of Beijing University of Chemical Technology and the Fundamental Research Funds for the Central Universities (YS 1406).

Notes and references

- 1 J. Zhang, Y.-C. Liang, X. Lin, X. Zhu, L. Yan, S. Li, X. Yang, G. Zhu, A. L. Rogach and P. K. Yu, ACS Nano, 2015, 9, 9741–9756.
- Z. Zhang, J. Wang and C. Chen, Adv. Mater., 2013, 25, 3869–3880.
- 3 R. Liang, R. Tian, L. Ma, L. Zhang, Y. Hu, J. Wang, M. Wei, D. Yan, D. G. Evans and X. Duan, Adv. Funct. Mater., 2014, 24, 3144–3151.
- 4 Y. Wang, W. Li, Y. Yang, Q. Zeng, K.-H. Wong, X. Li and T. Chen, J. Mater. Chem. B, 2015, DOI: 10.1039/c5tb01929j.
- 5 W.-W. Qi, H.-Y. Yu, H. Guo, J. Lou, Z.-M. Wang, P. Liu, A. Sapin-Minet, P. Maincent, X.-C. Hong, X.-M. Hu and Y.-L. Xiao, Mol. Pharm., 2015, 12, 675–683.
- 6 L. Chen, Y. Xue, X. Xia, M. Song, J. Huang, H. Zhang, B. Yu, S. Long, Y. Liu, L. Liu, S. Huang and F. Yu, J. Mater. Chem. B, 2015, 3, 8949–8962.
- 7 K. Li, Q. Su, W. Yuan, B. Tian, B. Shen, Y. Li, W. Feng and F. Li, ACS Appl. Mater. Interfaces, 2015, 7, 12278–12286.
- 8 J. Li, Y. Hu, J. Yang, P. Wei, W. Sun, M. Shen, G. Zhang and X. Shi, Biomaterials, 2015, 38, 10–21.
- J. Zeng, L. Jing, Y. Hou, M. Jiao, R. Qiao, Q. Jia, C. Liu, F. Fang,
 H. Lei and M. Gao, Adv. Mater., 2014, 26, 2694–2698.
- 10 Y. Wang and H. Gu, Adv. Mater., 2015, 27, 576-585.
- 11 Y. Liu, J. Bai, X. Jia, X. Jiang and Z. Guo, ACS Appl. Mater. Interfaces, 2015, 7, 112–121.
- 12 T. Chen, S. Xu, T. Zhao, L. Zhu, D. Wei, Y. Li, H. Zhang and C. Zhao, ACS Appl. Mater. Interfaces, 2012, 4, 5766–5774.
- 13 P. L. Truong, X. Ma and S. J. Sim, *Nanoscale*, 2014, 6, 2307–2315.
- 14 H. Wang, L. Zheng, C. Peng, M. Shen, X. Shi and G. Zhang, Biomaterials, 2013, 34, 470–480.
- 15 H.-X. Xia, X.-Q. Yang, J.-T. Song, J. Chen, M.-Z. Zhang, D.-M. Yan, L. Zhang, M.-Y. Qin, L.-Y. Bai, Y.-D. Zhao and Z.-Y. Ma, J. Mater. Chem. B, 2014, 2, 1945–1953.
- 16 J. Yang, M. H. Yao, M. S. Du, R. M. Jin, D. H. Zhao, J. Ma, Z. Y. Ma, Y. D. Zhao and B. Liu, Chem Commun., 2015, 51, 2569–2572.
- 17 J. Cui, Y. Yan, Y. Wang and F. Caruso, *Adv. Funct. Mater.*, 2012, **22**, 4718–4723.
- 18 R. Dorresteijn, N. Billecke, M. Schwendy, S. Pütz, M. Bonn, S. H. Parekh, M. Klapper and K. Müllen, Adv. Funct. Mater., 2014, 24, 4026–4033.
- 19 F. Li, X. Zhao, H. Wang, R. Zhao, T. Ji, H. Ren, G. J. Anderson, G. Nie and J. Hao, *Adv. Funct. Mater.*, 2015, **25**, 788–798.
- H. Deng, X. Zhao, J. Liu, L. Deng, J. Zhang, J. Liu and A. Dong, J. Mater. Chem. B, 2015, DOI: 10.1039/c5tb01939g.
- 21 X. Hu, J. Hu, J. Tian, Z. Ge, G. Zhang, K. Luo and S. Liu, *J. Am. Chem. Soc.*, 2013, **135**, 17617–17629.
- 22 P. Huang, D. Wang, Y. Su, W. Huang, Y. Zhou, D. Cui, X. Zhu and D. Yan, *J. Am. Chem. Soc.*, 2014, **136**, 11748–11756.
- 23 C. Y. Sun, Y. C. Ma, Z. Y. Cao, D. D. Li, F. Fan, J. X. Wang, W. Tao and X. Z. Yang, ACS Appl. Mater. Interfaces, 2014, 6, 22709–22718.
- 24 M. H. Xiong, Y. Bao, X. Z. Yang, Y. C. Wang, B. Sun and J. Wang, J. Am. Chem. Soc., 2012, 134, 4355–4362.

- 25 J. Yu, C. Yang, J. Li, Y. Ding, L. Zhang, M. Z. Yousaf, J. Lin, R. Pang, L. Wei, L. Xu, F. Sheng, C. Li, G. Li, L. Zhao and Y. Hou, Adv. Mater., 2014, 26, 4114–4120.
- 26 H. Wang, C. A. Thorling, X. Liang, K. R. Bridle, J. E. Grice, Y. Zhu, D. H. G. Crawford, Z. P. Xu, X. Liu and M. S. Roberts, J. Mater. Chem. B, 2015, 3, 939–958.
- 27 J. R. Rees, C. S. Burden and A. M. Fogg, J. Solid State Chem., 2015, 224, 36–39.
- 28 L. N. Ribeiro, A. C. Alcantara, M. Darder, P. Aranda, F. M. Araujo-Moreira and E. Ruiz-Hitzky, *Int. J. Pharm.*, 2014, **463**, 1–9.
- 29 D.-H. Park, G. Choi and J.-H. Choy, Struct. Bond., 2015, 166, 137–175.
- 30 J.-H. Choy, S.-Y. Kwak, Y.-J. Jeong and J.-S. Park, *Angew. Chem. Int. Ed.* 2000, **39**, 4041–4045.
- 31 D.-H. Park, J.-E. Kim, J.-M. Oh, Y.-G. Shul and J.-H. Choy, *J. Am. Chem. Soc.*, 2010, **132**, 16735–16736.
- 32 Q. Wang and D. O'Hare, Chem. Rev., 2012, 112, 4124–4155.
- 33 K. Khorsandi, R. Hosseinzadeh and M. Fateh, RSC Adv., 2015, 5, 93987–93994.
- 34 L. Yan, W. Chen, X. Zhu, L. Huang, Z. Wang, G. Zhu, V. A. L. Roy, K. N. Yu and X. Chen, *Chem. Comm.*, 2013, 49, 10938–10940.
- 35 S.-J. Choi, J.-M. Oh, H.-E. Chung, S.-H. Hong, I.-H. Kim and J.-H. Choy, *Curr. Pharm. Design*, 2013, **19**, 7196–7202.
- 36 J. M. Shen, X. M. Guan, X. Y. Liu, J. F. Lan, T. Cheng and H. X. Zhang, *Bioconjugate Chem.*, 2012, **23**, 1010–1021.
- 37 Y.-S. Yoon, B.-I. Lee, K. S. Lee, G. H. Im, S.-H. Byeon, J. H. Lee and I. S. Lee, *Adv. Funct. Mater.*, 2009, **19**, 3375–3380.
- 38 L. Wang, H. Xing, S. Zhang, Q. Ren, L. Pan, K. Zhang, W. Bu, X. Zheng, L. Zhou, W. Peng, Y. Hua and J. Shi, *Biomaterials*, 2013, 34, 3390–3401.
- 39 C. Yu, M. Zhou, X. Zhang, W. Wei, X. Chen and X. Zhang, Nanoscale, 2015, 7, 5683–5690.
- 40 Y. Zhao, F. Li, R. Zhang, D. G. Evans and X. Duan, *Chem. Mater.*, 2002, **14**, 4286–4291.
- 41 S. Cui, D. Yin, Y. Chen, Y. Di, H. Chen, Y. Ma, S. Achilefu and Y. Gu, *ACS Nano*, 2012, **7**, 676–688.
- 42 J.-M. Oh, S.-J. Choi, G.-E. Lee, J.-E. Kim and J.-H. Choy, *Chem. Asian J.*, 2009, **4**, 67–73.
- 43 J.-M. Oh, S.-J. Choi, S.-T. Kim and J.-H. Choy, *Bioconjugate Chem.*, 2006, **17**, 1411–1417.
- 44 J.-M. Oh, S.-J. Choi, G.-E. Lee, S.-H. Han and J.-H. Choy, *Adv. Funct. Mater.*, 2009, **19**, 1617–1624.

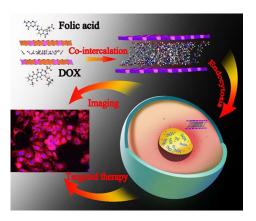
A Layered Drug Agent Nanovehicle toward Targeted Cancer Imaging and Therapy

Shanyue Guan,^a Ruizheng Liang,*^a Chunyang Li,^a Dan Yan,*^b Min Wei,*^a David G.

Evans^a and Xue Duan^a

- a. State Key Laboratory of Chemical Resource Engineering, Beijing University of Chemical Technology, Beijing 100029, P. R. China
- b. Beijing Shijitan Hospital. Capital Medical University Beijing 100038, P. R. China

The table of contents entry



A layered drug nanovehicle with superior anticancer performance was fabricated *via* the co-intercalation of doxorubicin (DOX) and folic acid (FA) into the gallery of layered double hydroxides (LDHs), which can be potentially applied in medical imaging/therapy.