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COMMUNICATION

## An extremely rapid-response fluorescence probe for hydrogen peroxide and its applications in living cells

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**A turn-on fluorescence probe ACF for rapid detection of H<sub>2</sub>O<sub>2</sub> was constructed. The probe utilized 2-(azidomethyl)benzoyl group as new reaction site, which exhibited rapid response to H<sub>2</sub>O<sub>2</sub> with a 118-fold fluorescence enhancement within 5 min. The biological application of ACF was confirmed by fluorescence imaging H<sub>2</sub>O<sub>2</sub> in living cells.**

Hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), one of the most important reactive oxygen species (ROS), has been known as a harmful metabolic product and a component of the immune response to microbial invasion for a long time.<sup>1</sup> However, H<sub>2</sub>O<sub>2</sub> functions as an ubiquitous intracellular second messenger when it is generated at a low concentration (< 0.7 μM).<sup>2</sup> It stimulates cell proliferation,<sup>3</sup> differentiation,<sup>4</sup> and migration<sup>5</sup> by activating the signalling pathways. H<sub>2</sub>O<sub>2</sub> might be generated aberrantly and result in oxidative stress with stimulation by exogenous chemicals. However, H<sub>2</sub>O<sub>2</sub> is an oxidant unlike the classical second messenger.<sup>6</sup> The resulting H<sub>2</sub>O<sub>2</sub> and other ROS will attack cellular structures or biomolecules such as proteins,<sup>7</sup> liposomes,<sup>8</sup> and DNA,<sup>9</sup> which has been correlated with aging,<sup>10</sup> Alzheimer's disease,<sup>11</sup> and cancer.<sup>12</sup> Obviously, location and timing of H<sub>2</sub>O<sub>2</sub> generation in biological process is highly required and would provide important information to understand the function of H<sub>2</sub>O<sub>2</sub>.

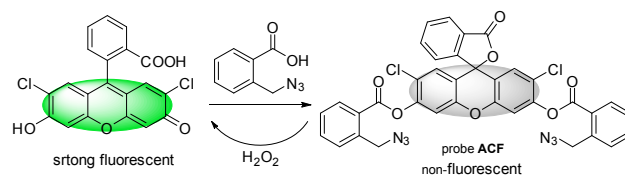
Currently, several chemical methods have been developed to detect intracellular H<sub>2</sub>O<sub>2</sub>, including mass assays,<sup>13</sup> proteomics assays,<sup>14</sup> and fluorescence-based assays.<sup>15–21</sup> Among these methods, fluorescence-based assays were useful because of their non-destructive features. A few fluorescent probes designed for H<sub>2</sub>O<sub>2</sub> detection have been reported since 2003.<sup>16a</sup> The first developed and most popular probes were a kind of boronate ester,<sup>16</sup> which have occupied more than a half of all the H<sub>2</sub>O<sub>2</sub> probes, including NIR, ratiometric, targetable, trappable and two-photon probes etc. Some of them have been successfully applied to monitor H<sub>2</sub>O<sub>2</sub> at physiological levels *in vitro* and *in vivo* and others have been used to explore the cellular mechanisms associated with H<sub>2</sub>O<sub>2</sub>. Chang group contributed a lot in this field.<sup>16b</sup> Another kind of fluorescent

probes developed for H<sub>2</sub>O<sub>2</sub> in the early stage contained arylsulfonyl esters as trap groups.<sup>17</sup> After that, several kinds of probes were designed based on unique H<sub>2</sub>O<sub>2</sub>-responsive sites such as diphenylphosphine,<sup>18</sup> α-diketone groups,<sup>19</sup> metal complexes<sup>20</sup> and some Chalcogen.<sup>21</sup> The efforts to find novel H<sub>2</sub>O<sub>2</sub>-responsive sites were still in progress.

Although several H<sub>2</sub>O<sub>2</sub>-responsive sites have been developed and a lot of fluorescence probes have been constructed, the reaction rates and fluorescence background levels of some fluorescence probes were generally not satisfactory for biological applications. When treated with H<sub>2</sub>O<sub>2</sub>, most of them had a long response time which became an important and complex issue for monitoring the H<sub>2</sub>O<sub>2</sub> concentration in living cells.<sup>22</sup> Only a few of them could respond to H<sub>2</sub>O<sub>2</sub> fast and selectively. It is especially important to develop rapid-response probes to monitor the H<sub>2</sub>O<sub>2</sub> in biological process.

Reduced fluorescent dyes such as 2',7'-dichlorodihydrofluorescein diacetate were commonly used as fluorescence probes for H<sub>2</sub>O<sub>2</sub>.<sup>23</sup> However, it still showed non-fluorescence if 2',7'-dichlorodihydrofluorescein diacetate was only oxidized to 2',7'-dichlorofluorescein diacetate.<sup>24</sup> It is obvious that the ester bond was broken during or after the oxidation. In this process, H<sub>2</sub>O<sub>2</sub> as a good nucleophile,<sup>16b</sup> might promote a nucleophilic substitution. So it is possible to construct a fluorescence probe for H<sub>2</sub>O<sub>2</sub> based on breaking a special ester bond. Therefore, finding a special ester bond and linking it with a fluorophore may be a feasible strategy.

Based on this strategy, **ACF** was synthesized by the reaction of 2',7'-dichlorofluorescein and 2-(azidomethyl)benzoyl acid (scheme 1). **ACF** exhibited almost no fluorescence (fluorescence quantum yield: Φ = 0.0024, in CH<sub>3</sub>OH/PBS buffer, 10 mM, pH = 7.4, 5/95, ESI<sup>+</sup>).



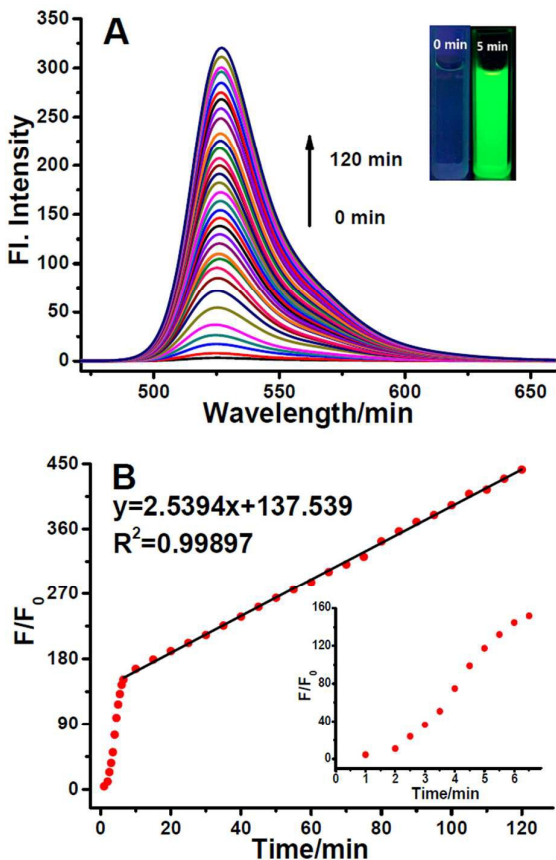
**Scheme 1** The synthesis of probe **ACF** and the response of **ACF** to H<sub>2</sub>O<sub>2</sub>

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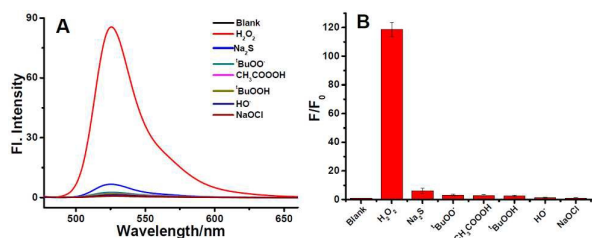
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When treated with  $\text{H}_2\text{O}_2$ , **ACF** showed extremely rapid response. Only in 5 min, the fluorescent intensity was increased by 118-fold (Fig. 1). The fluorescent intensity increased quite fast in the first 4 min, after which the increase rate fell a little. After 10 min, the fluorescent intensity was almost linear to the time with a correlation coefficient of 0.9990. And when extended to 120 min, the fluorescent intensity even increased by 441-fold (fluorescence quantum yield:  $\Phi = 0.6780$ , in  $\text{CH}_3\text{OH}/\text{PBS}$  buffer, 10 mM,  $\text{pH} = 7.4$ , 5/95, ESI<sup>†</sup>). The fluorescent intensities in the following experiments were recorded in 5 min due to the rapid-response of **ACF** for  $\text{H}_2\text{O}_2$ . Then the effect of pH on the fluorescence of **ACF** was evaluated, which showed it was stable from pH 6 to 8, even in the presence of  $\text{H}_2\text{O}_2$  (Fig. S1, ESI<sup>†</sup>).



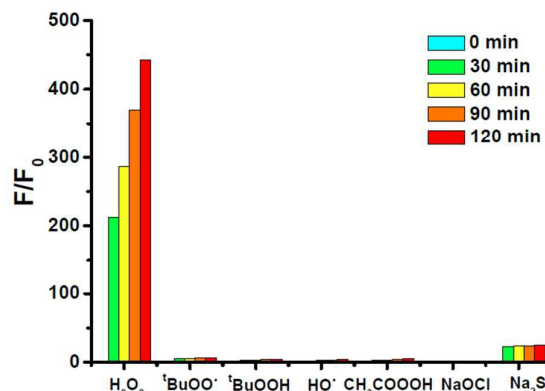
**Fig. 1** (A) Time-dependent fluorescence spectra of **ACF** ( $5 \mu\text{M}$ ) with  $\text{H}_2\text{O}_2$  (80 eq,  $400 \mu\text{M}$ ) in  $\text{CH}_3\text{OH}/\text{PBS}$  buffer (10 mM,  $\text{pH} = 7.4$ , 5/95). (B) Line chart. Inset: enlarged view of time area in the first 6 min.  $\lambda_{\text{ex}} = 450 \text{ nm}$ ,  $\lambda_{\text{em}} = 527 \text{ nm}$ . Slits: 5/5 nm.

To estimate the selectivity of probe **ACF** for  $\text{H}_2\text{O}_2$ , fluorescence responses to other ROS were examined. As shown in Fig. 2, significant fluorescence enhancement was observed after incubation with  $\text{H}_2\text{O}_2$  for 5 min. Compared to  $\text{H}_2\text{O}_2$ , other ROS induced only negligible fluorescence enhancements under the same condition, including *t*-BuOOH, hydroxy radical,  $\text{CH}_3\text{CO}_3\text{H}$ , and so on. Considering 2-azidomethylbenzoate had been used as a  $\text{H}_2\text{S}$  trap,<sup>25</sup>  $\text{H}_2\text{S}$  was also examined. It is true that fluorescence enhancement is observed after incubation with  $\text{H}_2\text{S}$ . But the increment was far less than that of  $\text{H}_2\text{O}_2$  (Fig. 2).



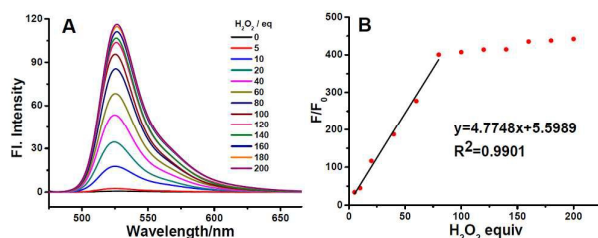
**Fig. 2** (A) Fluorescence response of **ACF** ( $5 \mu\text{M}$ ) incubated with ROS and  $\text{H}_2\text{S}$  (80 eq) in  $\text{CH}_3\text{OH}/\text{PBS}$  buffer (10 mM,  $\text{pH} = 7.4$ , 5/95) for 5 min. (B) Bar graph. 1. Blank; 2.  $\text{H}_2\text{O}_2$ ; 3.  $\text{Na}_2\text{S}$ ; 4. *t*-BuOO; 5.  $\text{CH}_3\text{CO}_3\text{H}$ ; 6. *t*-BuOOH; 7. HO; 8. NaOCl.  $\lambda_{\text{ex}} = 519 \text{ nm}$ ,  $\lambda_{\text{em}} = 527 \text{ nm}$ . Slits: 5/5 nm.

To further estimate the selectivity of probe **ACF** for  $\text{H}_2\text{O}_2$ , time-dependent fluorescence changes to ROS and  $\text{H}_2\text{S}$  were recorded. As shown in Fig. 3, the fluorescence enhancements for other ROS were still negligible in 120 min. While the enhancement for  $\text{H}_2\text{S}$  in 30 min was obvious. However, it was not strong enough to obstruct the detection of  $\text{H}_2\text{O}_2$  (the fluorescent intensity for  $\text{H}_2\text{S}$  to  $\text{H}_2\text{O}_2$  was 1:9.6). What was more, the fluorescent intensity for  $\text{H}_2\text{O}_2$  increased obviously with the time, while that for  $\text{H}_2\text{S}$  almost ceased. Thus, probe **ACF** showed high selectivity toward  $\text{H}_2\text{O}_2$ .



**Fig. 3** Time-dependent fluorescence spectra of **ACF** ( $5 \mu\text{M}$ ) with ROS and  $\text{H}_2\text{S}$  (80 eq,  $400 \mu\text{M}$ ) in  $\text{CH}_3\text{OH}/\text{PBS}$  buffer (10 mM,  $\text{pH} = 7.4$ , 5/95). Time points represent 30, 60, 90, and 120 min.  $\lambda_{\text{ex}} = 450 \text{ nm}$ ,  $\lambda_{\text{em}} = 527 \text{ nm}$ . Slits: 5/5 nm.

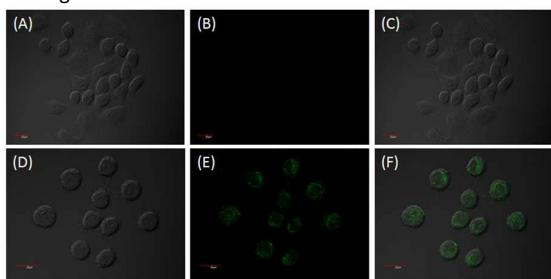
Subsequently, we examined the reactivity of **ACF** towards different concentrations of  $\text{H}_2\text{O}_2$  in  $\text{CH}_3\text{OH}/\text{PBS}$  buffer (10 mM,  $\text{pH} = 7.4$ , 5/95) at  $25^\circ\text{C}$ . As expected, after incubation with  $\text{H}_2\text{O}_2$ , the fluorescence intensity of probe **ACF** increased gradually as the increase of the  $\text{H}_2\text{O}_2$  amounts. Fig. 4 showed the fluorescence intensity of probe **ACF** increased almost linearly with the concentration of  $\text{H}_2\text{O}_2$  in the range of 50–400  $\mu\text{M}$ , and a correlation coefficient of 0.9901. Specifically, the detection limit of  $\text{H}_2\text{O}_2$  was determined to be 6.5 nM based on the  $3\sigma/\text{slope}$  method (ESI<sup>†</sup>), which was much lower than those of the reported probes. The good linearity indicated that probe **ACF** was able to qualitatively and quantitatively determine the level of  $\text{H}_2\text{O}_2$ .



**Fig. 4** (A) Fluorescence spectra of **ACF** (5  $\mu\text{M}$ ) incubated with different concentrations of  $\text{H}_2\text{O}_2$  (0–200 eq, 0–1.0 mM) in  $\text{CH}_3\text{OH}/\text{PBS}$  buffer (10 mM, pH = 7.4, 5/95) for 5 min. (B) Linear fitting chart.  $\lambda_{\text{ex}} = 450 \text{ nm}$ ,  $\lambda_{\text{em}} = 527 \text{ nm}$ . Slits: 5/5 nm.

Next, we carried out competition experiments in the presence of ROS and  $\text{H}_2\text{S}$  (Fig. S2, ESI<sup>†</sup>). **ACF** was still able to respond to  $\text{H}_2\text{O}_2$  with strong fluorescence enhancements in the presence of the interfering species. Moreover, the process of **ACF** for detection of  $\text{H}_2\text{O}_2$  was confirmed by the HRMS-ESI spectra. The mass signal for 2',7'-dichlorofluorescein ( $[\text{M} + \text{H}]^+$  calcd. for  $\text{C}_{20}\text{H}_{11}\text{Cl}_2\text{O}_5^+$ , 400.9978, found: 400.9962, Fig. S3, ESI<sup>†</sup>), was detected after probe **ACF** was incubated with  $\text{H}_2\text{O}_2$ . So the ester bond was possible broken by the nucleophilic substitution induced by  $\text{H}_2\text{O}_2$ . A controlled experiment showed that the probe would completely lose its effect if the azido group was instead by a hydrogen atom, which proved the importance of the azido group. Though the exact mechanism and the special effect of azido group were not clear, the in-depth mechanism study is in progress.

Encouraged by the above excellent results, we subsequently explored the potential applications of **ACF** for monitoring and imaging of  $\text{H}_2\text{O}_2$  in living cells. Firstly, the cytotoxicity of **ACF** was evaluated using A-549 cells and Raw 264.7 cells (obtained from the College of Life Science, Nankai University, Tianjin, China; Serum was purchased from Gibco) by MTT assay [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide, Fig. S4, ESI<sup>†</sup>]. Probe **ACF** showed almost no cytotoxicity in 0.1–30  $\mu\text{M}$  range to both of them, implying that the probe was suitable for bioimaging of  $\text{H}_2\text{O}_2$  in living cells. Finally, we assessed the application of the probe for monitoring and imaging of  $\text{H}_2\text{O}_2$  in living cells. HeLa cells (obtained from the College of Life Science, Nankai University, Tianjin, China) incubated with **ACF** (10  $\mu\text{M}$ ) in culture medium for 15 min at 37  $^\circ\text{C}$ , showed almost no fluorescence (Fig. 5B). However, if the cells were pre-treated with **ACF** (10  $\mu\text{M}$ ) for 15 min and then incubated with  $\text{H}_2\text{O}_2$  (10 eq, 100  $\mu\text{M}$ ) for 15 min, strong fluorescence was observed (Fig. 5E). The obvious fluorescent enhancement indicated that probe **ACF** could image  $\text{H}_2\text{O}_2$  in living cells. **ACF** responded to  $\text{H}_2\text{O}_2$  only in 15 min, making the detection get close to real-time monitoring.



**Fig. 5** Bright-field (A), fluorescence image (B) and the overlay (C) of Figure A and B of HeLa cells incubated with **ACF** (10  $\mu\text{M}$ ) for 15 min. Bright-field (D), fluorescence image (E) and the overlay (F) of Figure D and E of HeLa cells incubated with **ACF** (10  $\mu\text{M}$ ) for 15 min and

washed with PBS three times. After replacement of the medium, cells were incubated with  $\text{H}_2\text{O}_2$  (10 eq, 100  $\mu\text{M}$ ) for another 15 min.

## Conclusions

In conclusion, aiming at finding rapid-response fluorescent probes for detection of  $\text{H}_2\text{O}_2$ , a new probe 2',7'-dichloro-3',6'-bis(2-(azidomethyl)benzoate)fluorescein (**ACF**) was developed. **ACF** can rapidly respond to  $\text{H}_2\text{O}_2$  and offer highly sensitivity and selectivity by utilizing the unique chemical reactivity of the ester bonds and  $\text{H}_2\text{O}_2$ . Preliminary fluorescence imaging experiments indicate that **ACF** is a good fluorescent tool for rapidly monitoring  $\text{H}_2\text{O}_2$  in living cells. We believe the novel  $\text{H}_2\text{O}_2$  response site will be broadly used for quantitatively monitoring of  $\text{H}_2\text{O}_2$  in biological systems. Relevant studies on this strategy and its biological applications are underway.

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## An extremely rapid-response fluorescence probe for hydrogen peroxide and its applications in living cells

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An extremely rapid-response fluorescence probe for  $\text{H}_2\text{O}_2$  was constructed for monitoring the  $\text{H}_2\text{O}_2$  levels in biological process.

