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Complete List of Authors:	<p>Liu, Run; Sichuan Normal University, College of Chemistry and Materials Science</p> <p>Pan, Linlin; Sichuan Normal University, College of Chemistry and Materials Science</p> <p>Peng, Silu; Sichuan Normal University, College of Chemistry and Materials Science Chengdu</p> <p>Qin, Lili; Sichuan Normal University, College of Chemistry and Materials Science</p> <p>Bi, Jian; Sichuan Normal University, College of Chemistry and Materials Science</p> <p>Wu, Jiangtao; Sichuan Normal University, Chemistry and Materials Science</p> <p>Wu, Hua; Donghua University, Applied Physics; Simon Fraser University, Chemistry</p> <p>Ye, Zuo-Guang; Simon Fraser University, Chemistry</p>

Magnetolectric effect in a cubic ferrimagnetic spinel

LiFe_5O_8 with high coupling temperature

Run Liu,¹ Linlin Pan,¹ Silu Peng,¹ Lili Qin,¹ Jian Bi,¹ Jiangtao Wu,^{1†} Hua Wu,^{2,3†} and Zuo-Guang Ye^{3†}

¹. College of Chemistry and Materials Science, Sichuan Normal University, Chengdu 610068, China

². Department of Applied Physics, Donghua University, 201620 Shanghai, China

³. Department of Chemistry & 4D LABS, Simon Fraser University, Burnaby, BC, V5A 1S6, Canada.

† Corresponding authors. E-mails: Jiangtao Wu (jtwu@sicnu.edu.cn), Hua Wu (wuhua@dhu.edu.cn) and Zuo-Guang Ye (zye@sfu.ca)

Abstract: We report an effective magnetolectric (ME) coupling phenomenon in cubic ferrimagnetic spinel LiFe_5O_8 , with the command of its polarization by an applied magnetic field. A hysteretic response between magnetic field and electric polarization was observed. It is found that LiFe_5O_8 exhibits ME coupling that becomes significant at 250 K and increases with decreasing temperature. In contrast to many other spinel oxides that were reported to exhibit ME effect at very low temperatures (typically below 30 K), the bulk LiFe_5O_8 demonstrates a ME coefficient of $2 \text{ mV}\cdot\text{Oe}^{-1}\cdot\text{cm}^{-1}$ at 120 K. This material exhibits the highest ME coupling temperature among the magnetolectric spinel and related materials so far reported. This ME effect is attributed to the ordered arrangement of magnetic ions (Fe^{3+}) which are located in a noncentrosymmetric coordination environment. Our result provides a

1 new venue to search for single-phase magnetoelectric materials with high coupling
2 temperatures that could be used in potential technological applications.

3 **Keywords:** LiFe₅O₈, spinel magnetoelectrics, magnetoelectric effect

4 1 INTRODUCTION

5 Magnetoelectric (ME) materials that exhibit a cross coupling between magnetic
6 and electric order parameters have gained increasing interest due to their significant
7 potential for new-generation electronic and spintronic devices¹⁻⁶. Materials with a
8 spinel structure and general formula of AB₂O₄ (where A and B are metal ions, such as
9 Co²⁺, Zn²⁺, Fe³⁺, and Cr³⁺) have become promising ME materials due to their widely
10 tunable crystal structure and intricate magnetic/electric properties. Spinel typically
11 have a high permeability value with high resistivity that prevents the eddy current
12 energy losses, making them a good candidate for magnetoelectrics. However, at
13 present, only a few spinel compounds were reported to exhibit ME effect which only
14 occurred at very low temperatures, such as ZnCr₂Se₄ (20 K)⁷, CoCr₂O₄ (27 K),⁸
15 CoAl₂O₄ (3 K),⁹ and MnGa₂O₄ (20 K)¹⁰. Therefore, developing higher-temperature
16 and ideally room-temperature ME materials with spinel structure is a challenging task
17 which may lead to technological applications of spinel ME materials in advanced
18 devices and help better understand the microscopic mechanisms underlying the
19 structure-ME property relationship.

20 This work we explore a material from the family of spinel oxides, LiFe₅O₈, that
21 exhibits effective ME effect at relatively high temperatures (120 - 300 K). LiFe₅O₈
22 crystallizes in a B-site-ordered spinel structure with a cubic symmetry of *P*4₃32 at

1 room temperature.¹¹ Fig. 1(a) shows the schematic of the crystal structure in which all
2 Li^+ ions occupy the octahedral B-sites, whereas Fe^{3+} ions in high-spin state ($S = 5/2$)
3 occupy the slightly distorted octahedral B-sites (Fe_2) and tetrahedral A-sites (Fe_1) with
4 a 3:2 ratio. We consider LiFe_5O_8 to be a potential ME material at high temperatures
5 based on the following reasoning. Firstly, according to Rado's calculation,¹² a
6 magnetic field dependence of the magnetically induced electric polarization should
7 exist in LiFe_5O_8 , and moreover, Gridnev provided some indirect evidence of the ME
8 effect in a LiFe_5O_8 single crystal at low temperature.¹³ Secondly, LiFe_5O_8 crystallizes
9 in the $P4_332$ space group, in which the first and second order ME effects are allowed
10 in principle.¹⁴ Finally, LiFe_5O_8 possesses a large magnetization, which favors a strong
11 ME coupling effect.¹⁵ All of these features render LiFe_5O_8 a possible candidate as ME
12 material near room temperature.

13 In this work, bulk cubic spinel LiFe_5O_8 was synthesized and its ME property was
14 then systematically studied. It is found that bulk LiFe_5O_8 exhibits a hysteretic ME
15 signal at room temperature (300 K). The ME properties of LiFe_5O_8 could be attributed
16 to the ordered arrangement of Fe^{3+} ions located in a noncentrosymmetric coordination
17 environment along the [110] direction;¹⁶ this interpretation was supported by previous
18 theoretical studies^{10, 17}. Our results demonstrate that spinel LiFe_5O_8 exhibits the ME
19 effect at relatively high temperatures and provide a new pathway to search for new
20 room temperature ME materials.

21 **2 EXPERIMENTAL**

22 LiFe_5O_8 powders were prepared by solid-state reaction method. In brief, $\alpha\text{-Fe}_2\text{O}_3$

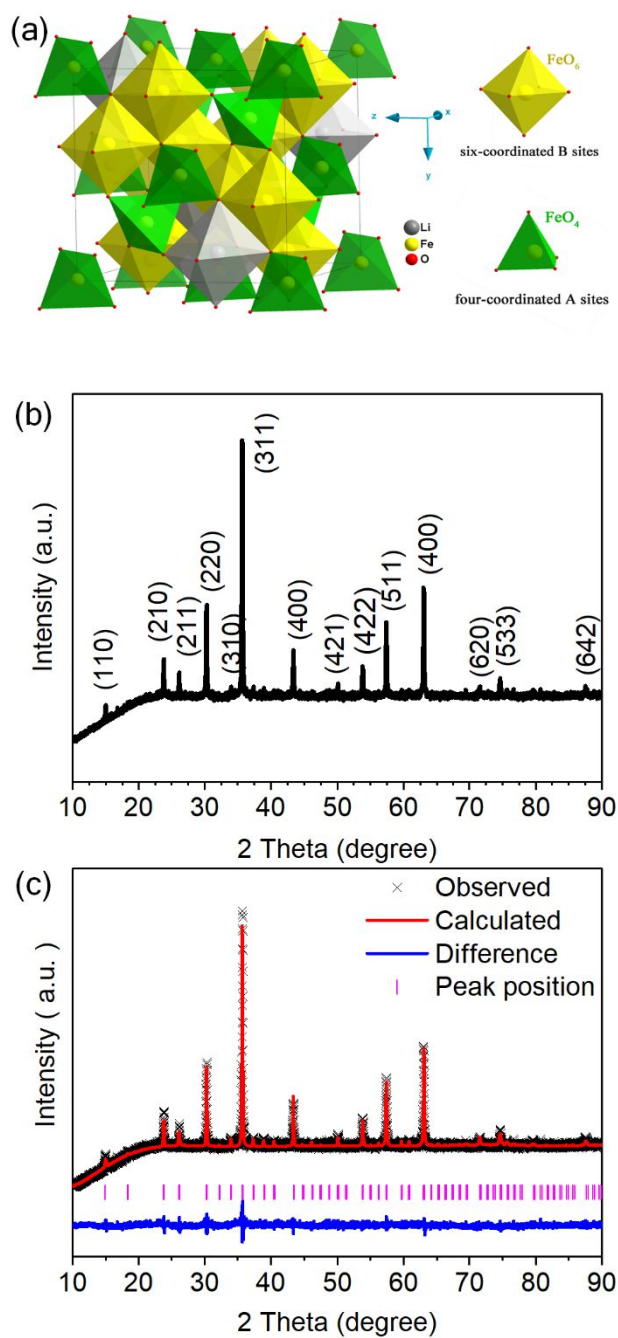
1 and Li_2CO_3 were milled and heated at 800 °C for 5 h ($\alpha\text{-Fe}_2\text{O}_3$ and Li_2CO_3 were dried
2 first at 200 °C for 7 h). The as-prepared LiFe_5O_8 powders were ground into fine
3 powders and pressed into pellets with 6 wt.% polyvinyl alcohol as binding agent. The
4 pellets were sintered in a furnace at 950 °C for 5 h in ambient atmosphere to form
5 dense ceramics. Both surfaces of the sintered ceramics were finely polished and
6 coated with silver paste electrodes.

7 X-ray powder diffraction (XRD) patterns were obtained using a SmartLab
8 diffractometer with Cu K_α radiation (Rigaku, Japan). Magnetic properties were
9 measured using a vibrating sample magnetometer (VersaLab and PPMS, Quantum
10 Design, Inc., USA). Magnetodielectric (magnetocapacitance) effect is commonly
11 considered to be a kind of ME behavior, which can also be exemplified by the
12 combination of magnetoresistance and the Maxwell-Wagner effect¹⁸. However, these
13 effects are not from true ME coupling phenomena. In this work, the ME properties
14 were characterized by measuring the direct ME voltage signal induced by the
15 magnetic field, i.e., $E(H_{ac})^2$. The ME effect can be quantitatively described by the ME
16 coefficient $\alpha_E = \Delta E / \Delta H_{ac}$. This parameter represents the variation in the ME voltage
17 signal (ΔE) under a small *ac* magnetic field ΔH_{ac} (1 Oe, 100 kHz) superimposed on a
18 *dc* magnetic field H_{dc} (-20 to 20 kOe) with the *dc* and *ac* magnetic fields being
19 perpendicular to the surface of the sample. The induced ME voltage signal ΔE and
20 phase difference were measured and recorded by a lock-in amplifier locked at the H_{ac}
21 excitation frequency and phase (SR830, SRS Inc., USA). The induced Faraday
22 electromotive force was eliminated from the raw data. Typically, the induced Faraday

1 electromotive force is about 1 μV (whereas $\alpha_E > 15 \mu\text{V}$).

2 **3 RESULTS AND DISCUSSION**

3 Room-temperature XRD analysis was performed to study the crystal structures of
4 LiFe_5O_8 . The XRD pattern in Fig. 1(b) can be indexed to a cubic structure with the
5 noncentrosymmetric space group of $P4_332$. The XRD data at room temperature were
6 refined by the Rietveld method using GSAS software to confirm the crystal structure
7 and obtain the atom positions of LiFe_5O_8 [Fig. 1(c)]. The Rietveld refinement results
8 (red line) fit well the observed XRD data (black cross), and confirmed that LiFe_5O_8
9 possesses a cubic spinel structure with lattice parameters $a = b = c = 8.3237 \text{ \AA}$. The
10 statistical parameters (χ^2 , R_p , and R_{wp}) are 0.663, 2.35%, and 1.83%, respectively. The
11 low values of reliability factors indicate the high accuracy of the structural analysis.
12 Table I lists the parameters of atomic coordinates, where x/a , y/b , and z/c represent the
13 spatial coordinate axis and O_1^{2-} and O_2^{2-} refer to the oxygen ions connecting two Fe^{3+}
14 ions and connecting Li^+ and Fe^{3+} ions, respectively. Our results show that the Fe^{3+}
15 ions of this spinel oxide occupy both the slightly distorted octahedral B-site and the
16 tetrahedral A-site, leading to their displacements from the center of the octahedron
17 and tetrahedron. This finding is in accordance with the published results^{11, 16}.



1

2 FIG. 1. (a) Schematic of the crystal structure of LiFe_5O_8 . (b) XRD patterns of LiFe_5O_8

3 samples. (c) Observed XRD data (black cross), calculated data (red line), and

4 difference (blue line) between the observed data and refinement results of LiFe_5O_8 .

5 The pink lines refer to the position of the Bragg's reflection peak of this compound.

6

1 TABLE I. Refined structural positions of cations/anions in LiFe_5O_8

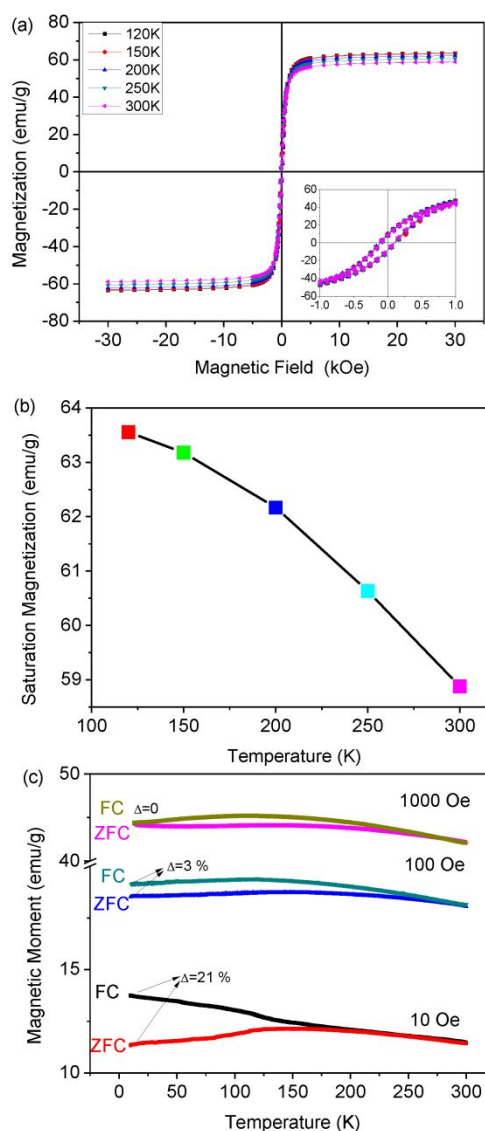
Ions	Wyckoff position	x/a	y/b	z/c
Li^+	4b	0.6250(0)	0.6250(0)	0.6250(0)
Fe_1^{3+}	8c	0.9981(4)	0.9981(4)	0.9981(4)
Fe_2^{3+}	12d	0.1250(0)	0.3681(4)	0.8819(4)
O_1^{2-}	8c	0.3825(15)	0.3825(15)	0.3825(15)
O_2^{2-}	24e	0.1233(10)	0.1318(10)	0.3725(10)

2

3 Fig. 2(a) shows the magnetization-magnetic field (M - H) relation measured at the
4 temperatures of at 300, 250, 200, 150 and 120 K. As shown in Figure 2 (a), LiFe_5O_8
5 displays similar slim magnetic hysteresis loops at these temperatures. The coercive
6 field H_c is 100 Oe and the remnant magnetization is about 9 emu/g, indicating that
7 LiFe_5O_8 is magnetically soft. The values of the saturation magnetization M_s are found
8 to be 63.6, 63.2, 62.2, 60.3 and 58.9 emu/g at 120, 150, 200, 250, and 300 K,
9 respectively, as shown in Figure 2 (b). The magnetism originates from the antiparallel
10 spin alignment between Fe_2 and Fe_1 .¹⁵ The strong and intrinsic magnetic interaction in
11 LiFe_5O_8 at 300 K is favorable for developing ME effects.¹⁹ Fig. 2(c) exhibit the
12 zero-field-cooled (ZFC) and field-cooled (FC) magnetization behavior of LiFe_5O_8 at
13 the field of 10, 100 and 1000 Oe between 300 and 10 K. It can be seen that under the
14 fields of 10 and 100 Oe, the ZFC and FC curves start to split at 170 and 230 K,
15 respectively, and the difference in the magnetic moments between ZFC and FC
16 increases as the temperature decreases, more significantly at lower field. At 10 K, the

1 relative difference in the magnetic moments reaches 21 % and 3 % under 10 and 100
2 Oe, respectively. When the external field increases to 1000 Oe, the splitting of the
3 ZFC/FC curves can be observed at about 210 K, with the difference in magnetic
4 moments increasing upon cooling to 120 K and then decreasing upon further cooling
5 before reaching zero at 10 K. These variations of magnetization upon ZFC and FC at
6 different fields represent the typical behavior of a spin glass system^{20, 21}, which can
7 be attributed to the competition between the glassy spin interactions and the Zeeman
8 effect. In a spin glass system, the magnetization first increases upon cooling until the
9 random interactions cause the spins to freeze out in random directions, resulting in a
10 decrease in magnetization upon further cooling. The temperature at which the
11 magnetization starts to decrease under weak fields (10 Oe in this work) is typically
12 called the "blocking temperature". Accordingly, the blocking temperature for LiFe_5O_8
13 is found to be about 120 K at which the ZFC curve began to decrease because of the
14 spin freezing. On the other hand, an external magnetic field tends to align the spins
15 parallel to the field, giving rise to an increasing magnetization. Therefore, the ZFC/FC
16 curves split normally. A strong enough magnetic field can gradually destroy the spin
17 glass state, so the ZFC/FC curves tend to overlap with each other with a smaller split.

18



1

2 FIG. 2 (a) Magnetic hysteresis loops for LiFe_5O_8 measured at 120, 150, 200, 250
 3 and 300 K. Inset shows the hysteretic behavior of magnetization between -1 kOe and
 4 1 kOe. (b) Variation of the saturation magnetization with temperature. (c) Variations
 5 of the ZFC and FC magnetization measured between 300 and 10 K at a field of 10,
 6 100 and 1000 Oe.

7

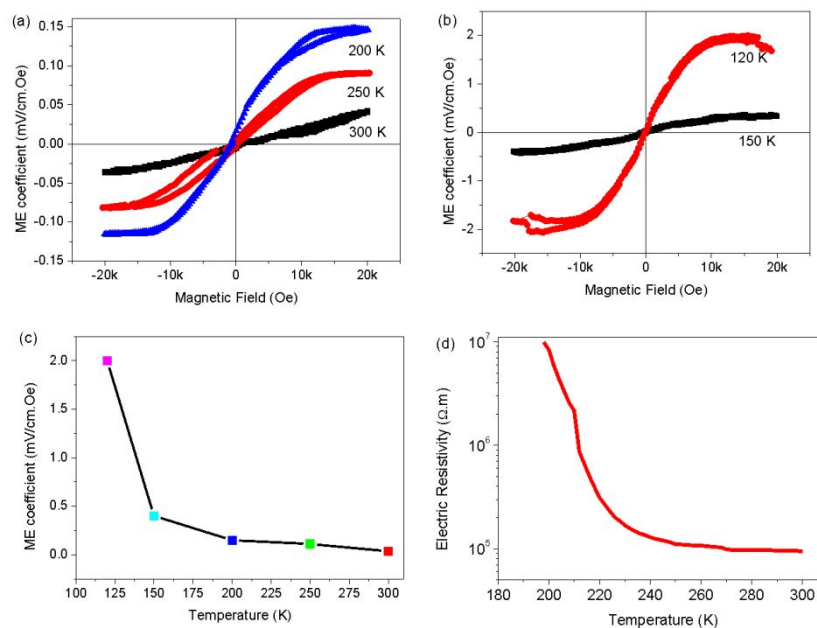
8 The ME effect of LiFe_5O_8 was measured in the form of ME coefficient (α_E). Figs.
 9 3(a) and (b) display the variation of α_E as a function of magnetic field (α_E-H) at
 10 various temperatures. Obviously, α_E exhibits a significant dependence on external

1 magnetic field upon cooling (from 300 to 120 K), thereby indicating a coupling
2 between the magnetic and electrical subsystems. The variation of α_E as a function of
3 magnetic field between 300 and 120 K looks similar to the M - H loop, demonstrating a
4 sharp increase when the field varies from -10 kOe to 10 kOe and a sharp decrease
5 when the field changes from 10 kOe to -10 kOe. It reaches saturation at the fields of
6 10 kOe to 20 kOe (and -10 kOe to -20 kOe). The magnetization reaches saturation at
7 about 4 kOe at various temperatures between 300 and 120 K (Fig 4 a), while the ME
8 coupling curves reach “saturation” at the fields of 10 - 15 kOe at temperatures below
9 250 K. At 300 K, the ME coefficient seems to increase with increasing magnetic field
10 up to 20 kOe. It is known that in artificial (composite) ME materials, for example, the
11 PZT/Co composite, the saturation magnetization field typically equals the saturation
12 ME field because the ME effect mainly originates from the interface strain via
13 magnetostrictive effect which in turn generates electric polarization through
14 piezoelectric effect. For single-phase ME materials like LiFe_5O_8 , however, the
15 dependences of the magnetization and ME coefficient on magnetic field and their
16 respective saturation fields are not necessarily the same because the ME effect results
17 from an intrinsic mechanism through the interaction of magnetic field with the spins
18 of the ordered magnetic ions (Fe^{3+}) located in a noncentrosymmetric coordination
19 environment, which in turn causes the changes in electric polarization. In addition, the
20 competition between the two crystalline anisotropies (tetrahedral and octahedral sites)
21 and the shape anisotropies may result in the rotation of the easy direction of
22 magnetization under an increasing magnetic field, causing the ME effect to saturate at

1 a higher field than the saturating field of the M-H relation. The difference in the
2 magnetization saturation field and ME saturation field is commonly found in other
3 single-phase materials. For example, in Dy-doped BiFeO₃ (Bi_{0.88}Dy_{0.12}Fe_{0.97}Ti_{0.03}O₃),
4 the saturation magnetization field is 25 Oe, while the saturation ME field is about 100
5 Oe² and in Sm-doped BiFeO₃, these two fields 20 kOe and less than 1 kOe,
6 respectively²². The nearly linear increase of ME coefficient with magnetic field at 300
7 K may be due to its weak ME effect, like in some other magnetoelectrics which do not
8 show any “saturation” in the ME effect and their ME coefficient increases linearly
9 with magnetic field, such as Bi_{0.68}La_{0.32}Fe_{0.655}Mn_{0.025}Ti_{0.32}O₃²³ and
10 Bi_{0.87}La_{0.05}Tb_{0.08}FeO₃²⁴.

11 Fig. 3(c) depicts the temperature dependence of α_E measured at the magnetic field
12 of 15 kOe. α_E increases with the decrease in temperature and its value reaches 2
13 mV·Oe⁻¹·cm⁻¹ at 120 K (0.11 ps/m), which indicates an significant ME effect at a
14 relatively high coupling temperature. Figure 3(d) shows the temperature - electricity
15 curve form 300 K to 200 K. It can be seen from Figure 3(d) that the resistivity at 300
16 K is in the order of 10⁵ Ohm.m and it increases with the decrease of temperature, first
17 slowly below room temperature and then sharply below 250 K. This result
18 corroborates our statement that spinels, as represented here by LiFe₅O₈, exhibit high
19 resistivity, making them good candidates as ME materials. With the decrease of
20 temperature, $\rho(T)$ shows a slow rise near room temperature (270-300 K) and followed
21 by a steep increase below ~250 K, suggests that the resistivity is substantially
22 enhanced with the decrease of temperature. It is interesting to note that the

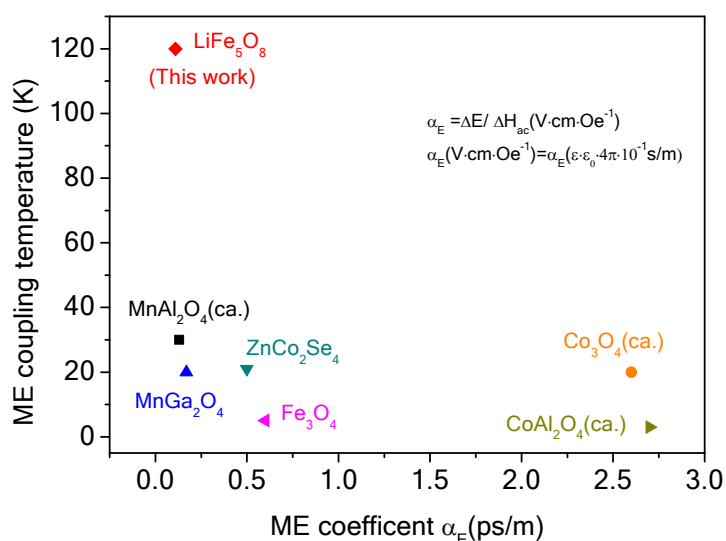
1 temperature dependences of the ME coefficient and resistivity show the same trend:
 2 both decreasing sharply with the increase of temperature. This suggests that the
 3 decrease of the measured ME coefficient values (α_E) at temperatures above 200 K
 4 may be partially caused by the loss of magnetically induced charge (polarization) due
 5 to a higher conductivity. So, the decreasing values of the measured α_E at temperatures
 6 above 120 K may be caused by two effects: i) the weakening of ME coupling and ii)
 7 the loss of magnetically induced charge (polarization) due to higher conductivity at
 8 high temperatures.



9
 10 FIG. 3. (a), (b) Variation of ME coefficient as a function of the external magnetic
 11 field measured at various temperatures for LiFe_5O_8 and (c) temperature dependence of
 12 the maximum value of ME coefficient. (d). Temperature dependence of *dc* electric
 13 resistivity.

14
 15 Although some spinels were reported to exhibit magnetoelectric effects, such as

1 CoCr_2O_4 (26 K),²⁵ MnCr_2O_4 (17 K),²⁶ Mn_3O_4 (40 K),²⁷ CdV_2O_4 (33 K),²⁸ CdCr_2S_4 (75
 2 K)²⁹, only the ME coefficients of a few compounds have been measured. Figure 4
 3 presents the reported ME coefficients of spinel compounds together with the value of
 4 LiFe_5O_8 measured in this work. It can be clearly seen that, compared with the spinel
 5 materials that display ME effect at low temperatures, including CoAl_2O_4 (calculated),⁹
 6 Fe_3O_4 nanoparticles,³⁰ ZnGr_2Se_4 ,⁷ Co_3O_4 and MnAl_2O_4 (calculated),¹⁰ the bulk
 7 LiFe_5O_8 exhibits a decent ME effect at a much higher coupling temperature. Note that
 8 although Co_3O_4 and CoAl_2O_4 reportedly exhibited high ME effect, their α_E values
 9 were calculated only and have not been measured experimentally. Moreover, the
 10 extreme toxicity of Co element and the low coupling temperatures make them
 11 unsuitable for ME applications.



12

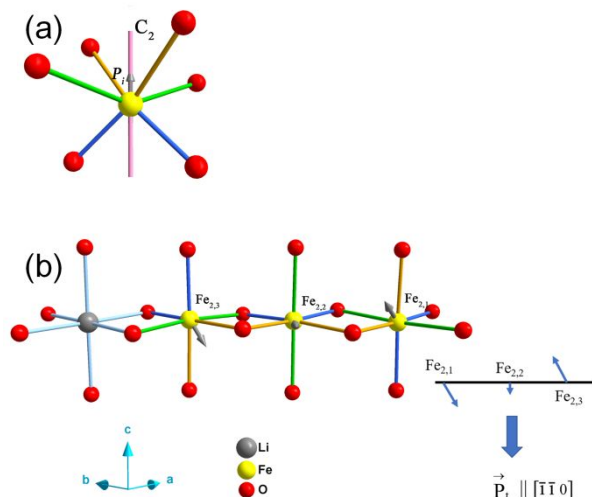
13 FIG. 4. Ordering temperature and ME coefficient (α_E) of reported spinel
 14 magnetoelectrics and LiFe_5O_8 of this work. Note that the α_E values of CoAl_2O_4 ,

15

Co_3O_4 and MnAl_2O_4 were calculated values only.

16

1 We believe that the aforementioned ME coupling in LiFe_5O_8 arises the
2 noncentrosymmetric crystal structure and the ordered arrangement of Fe^{3+} ions. From
3 the view of the crystal symmetry of $P4_332$, LiFe_5O_8 has a spinel structure with cations
4 ordering in the octahedral position in the $[110]$ direction with every cubic unit cell
5 containing three Fe^{3+} ions ($\text{Fe}_{2,1}$, $\text{Fe}_{2,2}$, and $\text{Fe}_{2,3}$) and one Li^+ ion in the (110) plane
6 [Fig. 5(b)]. In accordance with the crystal chemistry, the Fe^{3+} ion occupying the
7 slightly distorted B-sites exhibits a nonzero local electric polarization P_i , which is
8 represented by a gray arrow in Fig. 5(a), through the two-fold rotational axis
9 symmetry (C_2).³¹ Thereby, the three Fe^{3+} ions in the (110) plane exhibit electric
10 dipoles $\vec{P}_{1,2}$, $\vec{P}_{2,2}$ and $\vec{P}_{2,3}$ (gray arrows) with equal magnitude and different
11 directions along $[\bar{1}0\bar{1}]$, $[\bar{1}\bar{1}0]$, and $[0\bar{1}1]$, respectively [Fig. 5(b)]. The electric
12 polarization of the unit cell arises from the nonzero total dipolar moment \vec{P}_t , which
13 can be expressed as $\vec{P}_t = \vec{P}_{2,1} + \vec{P}_{2,2} + \vec{P}_{2,3}$, and is along the $[\bar{1}\bar{1}0]$ direction and
14 perpendicular to the (110) plane [Fig. 5(b)]. As for the magnetic interaction, Fe^{3+} has
15 electron configuration $(t_{2g})^3(e_g)^2$ that provides five unpaired electrons at the high-spin
16 state, constituting the unquenched orbital spin momentum. Herein, the spin \vec{S} of Fe^{3+}
17 ions occupying the octahedral site will modify the electric polarization due to spin-
18 orbital coupling, which results in the spin dependence of the electric polarization and
19 the coupling between polarization and magnetization, giving rise to the ME effects.
20



1

2 FIG. 5. (a) Fe³⁺ on a site of two-fold (C₂) symmetry formed by a distorted oxygen
 3 octahedron. The rotational C₂ symmetry axis is indicated (Note: the same color of
 4 bond represents the same bond length). (b) Three Fe³⁺ ions in the unit cell with their
 5 respective electric dipole moments (gray arrows).

6 4 CONCLUSIONS

7 In this study, the ME effect of the cubic spinel LiFe₅O₈ was investigated and the
 8 ME coefficient was measured through the command of polarization by an applied dc
 9 magnetic field (H_{dc}) between 120 and 300 K using a lock-in amplifier. It is found that
 10 LiFe₅O₈ exhibits ME coupling that becomes significant at 250 K and increases with
 11 decreasing temperature. In contrast to many other spinel oxides that were reported to
 12 exhibit ME effect at very low temperatures (typically below 30 K), the bulk LiFe₅O₈
 13 demonstrates a decent ME effect at much higher ordering temperatures with the
 14 maximum ME coefficient reaching 2 mV·Oe⁻¹·cm⁻¹ at a temperature as high as 120 K.
 15 The Fe³⁺ ions at a site of noncentrosymmetric coordination and the cation ordering are
 16 believed to give rise to the coupling between the local electric polarization and the

1 spin via spin-orbital interaction, leading to the observed magnetoelectricity in
2 LiFe_5O_8 .

3 The ME effect in LiFe_5O_8 could be further enhanced by improving the material
4 quality through chemical modification and/or physical means, such as annealing,
5 doping, single crystal growth, thin film deposition, etc. This work reveals that
6 LiFe_5O_8 as a prototype of spinel materials with high magnetic phase transition
7 temperature and noncentrosymmetric structure possesses the possibility to become a
8 promising material with significant ME coupling at relatively high temperatures,
9 potentially useful for new-generation electronic and spintronic device. It also provides
10 insights into the microscopic mechanisms of ME effect in spinel which will be helpful
11 in designing new single-phase ME materials.

12 **Acknowledgments**

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19 **References**

- 20 1. T. Amrillah, Y. Bitla, K. Shin, T. Yang, Y. H. Hsieh, Y. Y. Chiou, H. J. Liu, T. H. Do, D. Su, Y.
21 C. Chen, S. U. Jen, L. Q. Chen, K. H. Kim, J. Y. Juang and Y. H. Chu, *ACS Nano*, 2017, **11**,
22 6122-6130.
- 23 2. L. Pan, Q. Yuan, Z. Liao, L. Qin, J. Bi, D. Gao, J. Wu, H. Wu and Z.-G. Ye, *J. Alloys Compd.*,
24 2018, **762**, 184-189.
- 25 3. J. A. Mundy, C. M. Brooks, M. E. Holtz, J. A. Moyer, H. Das, A. F. Rebola, J. T. Heron, J. D.
26 Clarkson, S. M. Disseler, Z. Liu, A. Farhan, R. Held, R. Hovden, E. Padgett, Q. Mao, H. Paik, R.
27 Misra, L. F. Kourkoutis, E. Arenholz, A. Scholl, J. A. Borchers, W. D. Ratcliff, R. Ramesh, C. J.
28 Fennie, P. Schiffer, D. A. Muller and D. G. Schlom, *Nature*, 2016, **537**, 523-527.
- 29 4. H. Xi, X. Qian, M. C. Lu, L. Mei, S. Rupprecht, Q. X. Yang and Q. M. Zhang, *Sci. Rep.*, 2016, **6**,

- 1 29740.
- 2 5. J. Wu, Z. Shi, J. Xu, N. Li, Z. Zheng, H. Geng, Z. Xie and L. Zheng, *Appl. Phys. Lett.*, 2012, **101**,
- 3 122903.
- 4 6. M. M. Sang-Wook Cheong, *Nat. Mater.*, 2007, **6**, 13-20.
- 5 7. H. Murakawa, Y. Onose, K. Ohgushi, S. Ishiwata and Y. Tokura, *J. Phys. Sco. Jpn*, 2008, **77**,
- 6 043709.
- 7 8. Y. Yamasaki, S. Miyasaka, Y. Kaneko, J. P. He, T. Arima and Y. Tokura, *Phys. Rev. Lett.*, 2006,
- 8 **96**, 207204.
- 9 9. S. Ghara, N. V. Ter-Oganessian and A. Sundaresan, *Phys. Rev. B*, 2017, **95**, 094404.
- 10 10. R. Saha, S. Ghara, E. Suard, D. H. Jang, K. H. Kim, N. V. Ter-Oganessian and A. Sundaresan,
- 11 *Phys. Rev. B*, 2016, **94**, 014428.
- 12 11. P. B. Braun, *Nature*, 1952, **170**, 1123–1123.
- 13 12. G. T. Rado, J. M. Ferrari and J. P. Remeika, *J. Appl. Phys.*, 1978, **49**, 1953.
- 14 13. V. N. Gridnev, B. B. Krichevtsov, V. V. Pavlov and R. V. Pisarev, *J. Exp. Thero. Phys.*, 1997, **65**,
- 15 68-73.
- 16 14. M. Mercier, G. Velleaud and J. Puvinel, *Physica*, 1977, 86-88.
- 17 15. R. Zhang, M. Liu, L. Lu, S.-B. Mi and H. Wang, *J. Mater. Chem. C*, 2015, **3**, 5598-5602.
- 18 16. J. L. Dormann, A. Tomas and M. Nogues, *Phys. Stat. Sol. (A)*, 1983, **77**, 611–618.
- 19 17. V. P. Sakhnenko and N. V. Ter-Oganessian, *J. Phys.: Condens. Matter*, 2012, **24**, 266002.
- 20 18. G. Catalan, *Appl. Phys. Lett.*, 2006, **88**, 102902.
- 21 19. H. Wu and W. Z. Shen, *Solid State Comm.*, 2005, **133**, 487-491.
- 22 20. W. Luo, S. R. Nagel, T. F. Rosenbaum and R. E. Rosensweig, *Phys. Rev. Lett.*, 1991, **67**,
- 23 2721-2724.
- 24 21. K. Binder, *Rev. Mod. Phys.*, 1986, **58**, 801-976.
- 25 22. V. Sreenivas Puli, D. Kumar Pradhan, S. Gollapudi, I. Coondoo, N. Panwar, S. Adireddy, D. B.
- 26 Chrissey and R. S. Katiyar, *J. Magn. Magn. Mater*, 2014, **369**, 9-13.
- 27 23. C. M. Fernandez-Posada, A. Castro, J. M. Kiat, F. Porcher, O. Pena, M. Alguero and H. Amarin,
- 28 *Nat. Commun.*, 2016, **7**, 12772.
- 29 24. Q. H. Jiang, J. Ma, Y. H. Lin, C.-W. Nan, Z. Shi and Z. J. Shen, *Appl. Phys. Lett.*, 2007, **91**,
- 30 022914.
- 31 25. S. Tiwari and D. Sa, *J. Phys.: Condens. Matter*, 2010, **22**, 225903.
- 32 26. N. Mufti, G. R. Blake and T. T. M. Palstra, *J. Magn. Magn. Mater*, 2009, **321**, 1767-1769.
- 33 27. T. Suzuki and T. Katsufuji, *Journal of Physics: Conference Series*, 2009, **150**, 042195.
- 34 28. G. Giovannetti, A. Stroppa, S. Picozzi, D. Baldomir, V. Pardo, S. Blanco-Canosa, F. Rivadulla, S.
- 35 Jodlauk, D. Niermann, J. Rohrkamp, T. Lorenz, S. Streltsov, D. I. Khomskii and J. Hemberger, *Phys.*
- 36 *Rev. B*, 2011, **83**, 060402(R).
- 37 29. P. L. J. Hemberger, R. Fichtl, H.-A. Krug von Nidda, and V. T. A. Loidl, *Nature*, 2005, **434**,
- 38 364-367.
- 39 30. K. Yoo, B. G. Jeon, S. H. Chun, D. R. Patil, Y. J. Lim, S. H. Noh, J. Gil, J. Cheon and K. H. Kim,
- 40 *Nano Lett.*, 2016, **16**, 7408-7413.
- 41 31. S.J.Marin, M.O'Keeffe and D.E.Partin, *J. Solid State Chem.*, 1994, **113**, 413-419.

42
43