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A Linear Mass Concentration Detector for Solvent Gradient Polymer Separations

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Abstract

Characterization of copolymers requires the measurement of two distributions—molecular weight (MW) and chemical composition (CC). Molecular weight distributions (MWD) are traditionally determined using size exclusion chromatography (SEC) run under isocratic solvent conditions. Chemical composition distributions (CCD) are often determined using liquid adsorption chromatography (LC) with solvent gradients. The use of solvent gradients, however, often limits options of compatible detectors. A gradient compatible, universal linear mass concentration detector is a longstanding unmet need. Many industrially-relevant polymers lack chromophores or other discriminating moieties requiring detectors with a universal response. Differential refractive index (dRI) is incompatible with gradient elution due to its small dynamic range. Charged aerosol detectors (CAD) and evaporative light scattering detectors (ELSD) are probably the most promising options for gradient elution detection, but both suffer from a nonlinear mass concentration response. Silicon photonic microring resonators are optical sensors that are responsive to changes in the local refractive index (RI). The substantial dynamic range of this technology makes it attractive for refractive index-based detection during solvent gradient elution. Previously, the microring resonator platform was used as a SEC detector to characterize the MWD of broadly dispersed polystyrene (PS) standards. In this study, we demonstrate the gradient compatibility of the microring resonator platform for polymer detection by quantifying the CCD of polymer blend components. Control experiments were run with UV and ELSD detection, highlighting the uniqueness of the platform as a linear mass concentration detector with a universal detector response.

Introduction

Characterizing polymeric samples by size, composition, structure, and purity can help solve many challenges in industrial polymer manufacturing. Comprehensive characterization of polymers requires a suite of analytical methods with the ultimate goal of establishing structure property relationships.¹ Continuing advances in polymer chemistry and advanced manufacturing have enabled commercial viability of increasingly complex polymer systems. This complexity demands continued advances in polymer characterization methods.² For instance, homopolymers such as polystyrene or high density polyethylene can be well characterized by molecular weight distributions (MWD).¹ The standard bearer for MWD determination is size exclusion chromatography (SEC) coupled to a UV/Visible (UV) or differential

refractometer (dRI) for linear mass concentration detection.^{3,4,5}

Nearly all modern product development in the polymer industry involves copolymer commercialization because of the need for a balance of properties in most new materials. Polymers with two or more incorporated monomers leads to an inherent chemical composition distribution (CCD). Much like

MWD, CCD dictates many physical properties of polymers including melt temperature, thermal stability, solubility, and mechanical properties such as tensile strength.^{1,6} As such, CCD measurements are a vital component of copolymer characterization.

Traditional techniques for characterizing chemical composition (CC) of copolymers include mass spectrometry (MS), various spectroscopic methods, and liquid chromatography (LC). MS and spectroscopic methods (e.g., NMR, FT-IR) provide an averaged measure of CC rather than a distribution. Given that copolymers are heterogeneous in both CC and MW, a compositional average does not provide the most comprehensive characterization.⁷ More informative distribution measurements require a separation of polymer components. Therefore, CC characterizations by MS or NMR often use preparative fractions obtained by SEC or LC, or these techniques are often hyphenated directly in-line with these separations methods.^{8,9,10,11,12} However, the complexity of these hyphenated techniques is such that gradient elution high performance liquid chromatography (HPLC) has emerged as the most commonly used approach for CCD measurements in practice.¹³

Gradient elution LC achieves separations based on CC and functions by increasing the elution strength of the mobile phase throughout the experiment. Analytes with a high affinity for the stationary phase adsorb onto the column and are eluted once there is a higher affinity for the mobile phase.¹⁴ The challenge,

1
2
3 however, with gradient LC is a lack of
4 available/compatible detectors.

5 The most common options for GPC
6 characterization of polymers include dRI, UV, charged
7 aerosol detectors (CAD)^{4,15,16}, and evaporative light
8 scattering detectors (ELSD);^{17,18,19,20} however, each of
9 these has limitations for certain applications. dRI lacks
10 the dynamic range to track the gradient baseline and
11 obviously any analytes eluting during the gradient. UV
12 detectors, widely used in gradient LC, give linear mass
13 concentration response, but many industrial polymers,
14 including polyolefins, polyacrylates, and polyalkoxides
15 lack chromophores.²¹ Because of their universal
16 response, CAD and ELSD are preferred as gradient
17 compatible detectors for polymer analysis. Both CAD
18 and ELSD nebulize and evaporate off the mobile phase
19 prior to detection, but they give a non-linear response
20 as a function of polymer concentration and solvent
21 composition. This non-linearity is a major limitation of
22 both CAD and ELSD for industrial polymer CCD as it
23 substantially complicates quantitative measurements.
24^{20,22, 23} It is important to mention that one approach to
25 overcoming ELSD non-linearity is actually linearizing
26 the response. This is often done by applying a
27 correction to the ELSD signal and has been shown to
28 give good correlation with isocratic GPC data
29 compared to traditional detectors.²⁴

30 Silicon photonic microring resonators are a type of
31 whispering gallery mode optical sensor that detects
32 small changes in refractive index (RI) at the sensor
33 surface.²⁵ The microring resonator platform offers a
34 very large RI dynamic range making the sensor
35 gradient compatible, as has been previously
36 demonstrated.²⁶ This is in contrast to conventional dRI
37 detectors, which are not gradient compatible because
38 of their small dynamic range. As a result, dRI detectors
39 can only be used for CCD type measurements as a
40 concentration detector for an isocratic dimension in
41 concert with multidimensional separations.²⁷

42 Typically, the microring resonator platform is
43 implemented as a biosensor for monitoring molecular
44 binding events on the sensor surface.^{25,28,29,30} Much
45 work has also been done interfacing similar optical
46 resonator detectors with various separation methods
47 such as gas chromatography³¹ and capillary
48 electrophoresis,^{32,33,34} and recently a microwave
49 interferometer was interfaced with HPLC for gradient
50 separations.³⁵ In addition, our previous work
51 demonstrated the applicability of microring resonator
52 platform for industrial polymer analysis. Here we
53 interfaced SEC separations with the microring
54 resonators to determine the MWD of broad range
55 polystyrene (PS) standards. These experiments were
56 performed in conjunction with separate dRI and UV
57 experiments to demonstrate the agreement of MWD
58 determined by the microrings to these conventional
59 detectors.³⁶

60 Herein, we demonstrate the quantification of CC
by utilizing the microring resonator platform as a RI
based gradient elution detector. Using poly(styrene-co-
methyl methacrylate) (PS-PMMA) copolymer
standards, the microring resonator platform was

calibrated for polystyrene (PS)/polymethacrylate
(PMMA) composition. The mass linearity of the
microring resonators was demonstrated and directly
compared to identical experiments performed with UV
and ELSD detection. Additionally, copolymer blends
were created using the same PS-PMMA standards and
these blends were analyzed using all three detection
methods. The composition calibration allowed for
identification of blend components, while mass
calibrations allowed for quantification of the abundance
of each component. These results demonstrate the
versatility and applicability of the microring resonator
platform for characterizing industrial polymers with a
single detector.

Experimental

Materials

High purity solvents were all purchased from
Sigma-Aldrich (St. Louis, MO). Poly(styrene-co-methyl
methacrylate) (PS-PMMA) standards were purchased
from Polymer Source, Inc. (Dorval, QC). Polystyrene
(PS) and polymethacrylate (PMMA) homopolymer
standards were purchased from Agilent (Santa Clara,
CA). All polymer standards were used as received.
Four 10 mg/mL PS-PMMA standards varying in PS
content (82%, 54%, 31%, and 14% mol PS, provided
by vendor) were prepared in chloroform and two 10
mg/mL homopolymer samples (70kDa PS and 70kDa
PMMA) were prepared in chloroform.

Microring Resonators

The microring resonator system (Maverick M1
optical scanning instrumentation) and sensor array
chips were purchased from Genalyte, Inc. (San Diego,
CA). Detailed descriptions of sensor fabrication and
instrument operation has been described elsewhere.²⁶

Microring resonators are ring-shaped optical
cavities of 30 μm diameter with adjacent linear
waveguides. 128 individually addressable microring
sensors are arranged in an array on a 4 mm x 6 mm
chip. Sensor chips have a protective photoresist
coating that is removed before use by successively
immersing chips in acetone and isopropanol baths,
followed by an acetone rinse.

Each individual microring is probed by an external
tunable cavity diode laser centered at 1550 nm. Optical
transmission is monitored as a function of wavelength,
and dips in transmittance signal occur at resonant
wavelengths defined by the following equation.

$$\lambda_r = \frac{2\pi r n_{eff}}{m} \quad (1)$$

where r is the ring radius, n_{eff} is the effective refractive
index, and m is a constant. As the refractive index
environment surrounding the sensor surface changes,
such as analyte elution or a switching mobile phase
composition, the resonant wavelength will shift. These
changes in resonance wavelengths correspond to

changes in n_{eff} which are monitored as a function of time and referred to as relative shift in units of delta picometers (Δpm).^{37,25}

Table 1: Gradient HPLC Method

| | Time (min) | Mobile Phase | |
|--------------------|------------|----------------------|-----|
| | | 95:5 Cyclohexane:THF | THF |
| Starting Condition | 0 | 95% | 5% |
| Isocratic hold | 10.5 | 95% | 5% |
| Gradient time | 30.5 | 10% | 90% |
| Purging | 40.5 | 10% | 90% |
| Re-conditioning | 60.5 | 95% | 5% |

HPLC

Chromatographic separations were performed on a Waters Alliance e2695 separation module (Milford, MA) equipped with a Waters 2489 UV/Visible detector and a Waters 2424 ELSD detector. A Kromasil (Bohus, Sweden) column was used for all separations. Column dimensions were 250 mm X 4.6 mm and the packing material was 5 μm silica particles with 60 Å pores.^{20,38,39,40} A 0.4 mL/min flow rate was maintained for a gradient of 95% cyclohexane (spiked with 5% tetrahydrofuran (THF)) to 90% THF (Table 1) and the column oven was held at 35°C. To increase mass on the column, multiple 5 μL injections of polymer were injected in isocratic mode using initial gradient conditions.^{41,42} This method was utilized to maintain small injections and prevent polymer breakthrough.⁴³ The Waters 2489 UV/Visible detector wavelength was set to 260 nm with a sampling frequency of 10 HZ. The Waters 2424 ELSD detector operated at a gain of 20, gas pressure of 20 psi, drift tube temperature of 50°C, and a nebulizer temperature of 12°C.

HPLC-Microring Resonator Interface

The microring resonator cartridge was assembled by placing the sensor chip on an anodized aluminum cartridge base followed by a polyethylene terephthalate (Mylar) gasket and a polytetrafluoroethylene (Teflon) cartridge top. The Mylar gasket and Teflon cartridge top direct fluid flow across the chip and the whole assembly is secured together by screws. 1/16" PEEK tubing with a 0.25 mm flangeless ¼-28 interface were used to couple the HPLC directly to the microring resonator cartridge, as

described previously.³⁶ For a diagram of the cartridge assembly see Figure S1 of the Supporting Information.

Once assembled the cartridge can be handled much like a flow cell, with a volume of approximately 2 μL , requiring no further handling of the sensor chip. Additionally the chip which is composed of silicon dioxide is robust enough to be used repeatedly without degraded sensor performance. Fouling is often an issue with surface based sensors especially with biological applications, however that is not an area of concern here since with LC methods analytes only come in contact with the sensor chip once they are in favorable solvents which mitigate any affinity they might have for the surface.

Data Analysis

Data analysis was carried out using custom software written in R (version 3.4.1) and RStudio (version 1.0153). Chromatograms were obtained for all the detection methods by plotting signal intensity as a function of time. Microring resonator data was obtained for each individual ring and individual ring responses were averaged to obtain the averaged response. This averaged raw data was then baseline corrected to account for the sloping baseline resulting from the continually changing mobile phase composition (Figure 1). The sloping baseline spans over a relative shift exceeding 800 Δpm , therefore peaks on a scale of 2-15 Δpm are initially obscured by the background baseline and overlapping traces (see Figure 1A-B). Peaks begin to become slightly visible by zooming in on the raw chromatogram, dashed lines are used to guide eye of peak location, (Figure 1B) but this is still not sufficient for any useful identification or quantification. An alternative plot of Figure 1B is presented with Figure S2, which allows for better visibility of peaks before correction. Baseline correction was necessary to remove the background baseline caused by the solvent gradient. Baseline correction was performed by fitting the sloping gradient observed in Figure 1B to a third order polynomial and subtracting the fit from the raw data (seen in Figure 1C). The resultant chromatogram was smoothed with a locally estimated scatterplot smoothing function (LOESS), a common smoothing function for time series data.⁴⁴ The final corrected microring resonator chromatogram resembles chromatograms obtained by more traditional detectors, as shown in Figure 1D. All chromatographic calculations were performed on

baseline corrected and smoothed data. Lastly, this baseline correction process is very robust since identical gradient methods will always observe the same RI change since this is a function of mobile phase, therefore baseline response is very reproducible making correction for this routine.

Results and Discussion

Gradient Separation of PS-PMMA Copolymers

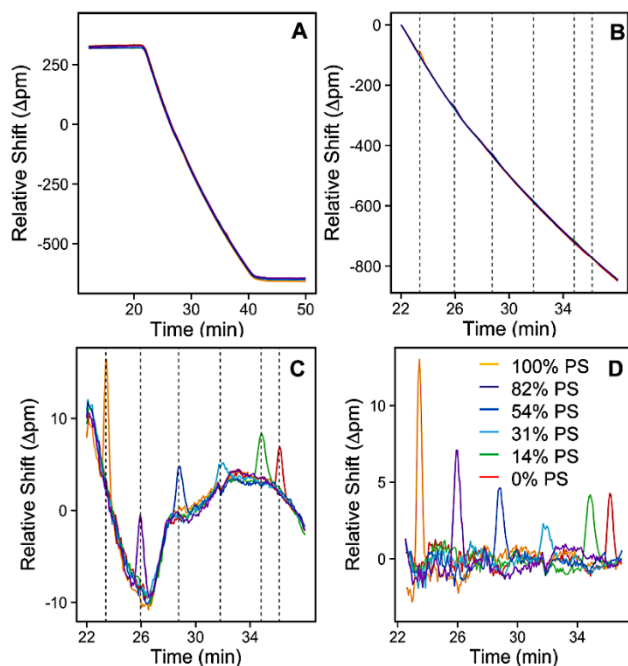


Figure 1: Data Treatment Process for Microring Resonators

A. Raw overlapping microring traces shows direct monitoring of gradient mixing through experiment since RI changes with changing mobile phase, this means gradient shape/slope is very reproducible for identical methods. **B.** A subset of **A** shows small peaks can be observed on the sloping baseline (peak location is indicated by dashed lines). This sloping baseline is fit to a third order polynomial which is then used to baseline correct the data. **C.** The fit obtained from the baseline observed in **B** is extrapolated and subtracted from the raw data. **D.** Then finally the subtracted chromatogram can be further corrected by applying a LOESS function.

The described gradient method outlined in Table 1 was applied to 0.15 mg injections of PS-PMMA copolymer samples prepared in chloroform, and varying in PS content covering the full range of 100%-0% moles PS (0% - 100% moles PMMA). Peaks begin to elute from the column with increasing THF content, with 100% PS eluting the earliest in the gradient given that it is the least polar. Therefore, elution order is highest to lowest PS content (lowest to highest PMMA content), with the 100% PMMA standard taking the longest to elute from the column. Detection of the

described separation was implemented using three detection methods; microring resonators (Figure 2A), ELSD (Figure 2B) and UV (Figure 2C). UV and microring resonators were connected in series, ELSD data was obtained in a separate experiment due to the destructive nature of the detection technique. A chart of the used flow path can be found in Figure S3 of the Supporting Information.

A comparison of the UV, ELSD, and microring resonator detectors is shown in Figure 2. The most notable difference between the ELSD and microring resonator detectors is increased noise of the microring resonator chromatograms, where the average signal to noise ratio of the microrings is approximately 4 orders

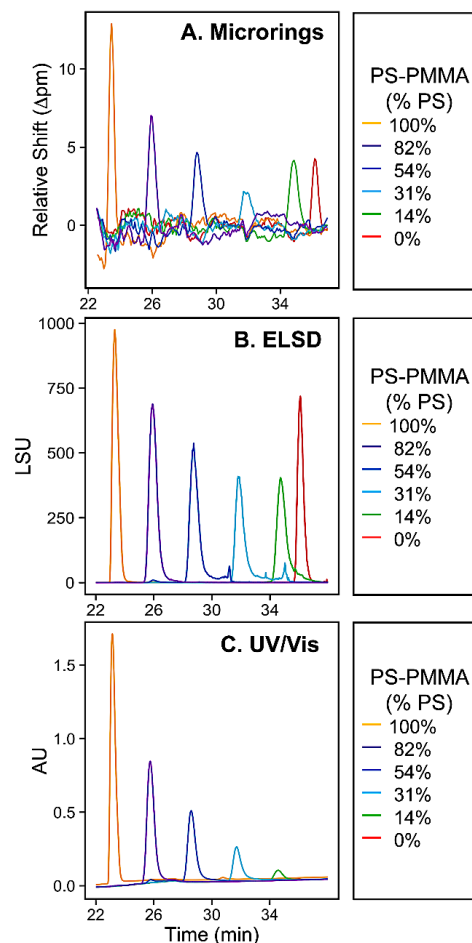


Figure 2: Gradient LC Chromatogram Comparison

Separations of PS-PMMA copolymers with a cyclohexane to THF gradient. Samples were prepared in chloroform at a concentration of 10 mg/mL and a mass of 0.15 mg was injected. Chromatograms were obtained by all three detectors **A.** microring resonator platform, **B.** Evaporative light scattering (ELSD), and **C.** UV/visible (UV).

of magnitude smaller than UV and ELSD. This increased level of noise was somewhat expected since the microring resonator platform is held under ambient conditions susceptible to temperature fluctuations, whereas the UV and ELSD detectors are far less sensitive to temperature variations. The impact of temperature on the microring resonator signal has been discussed elsewhere, however, it is estimated that a 0.1°C change in temperature results in ~4.5 Δpm change in microring resonator response (this estimate was determined in a related study).^{36,26}

Other comparisons that can be made are the differences of peak heights observed as the PS content changes. For the microring resonators (Figure 2A) this decrease in peak height is due to the changing RI contrast as PMMA content is increased. The respective RI (*n*) of PS, PMMA, cyclohexane and THF are as follows; 1.59,⁴⁵ 1.49,⁴⁶ 1.43, and 1.41, at 20 °C and 632.8 nm. The 31% PS-PMMA peak provided the smallest response from the microring resonators. This is caused by a combination of the RI contrast and the decreased sensitivity of the microring resonator platform for high molecular weights. With increasing molecular weight, the radius of gyration of a random polymer coil in solution also increases, and polymers with larger radii of gyration have portions that extend beyond the evanescent field of the sensor causing this sensitivity fall-off. This molecular weight dependence has been investigated and described previously.^{36,47} All PS-PMMA standards have a molecular weight within a range of 60-80 kDa, with the exception of 31% PS which has a molecular weight of 117 kDa. These were available choices at the time of purchase. As for the changing peak heights with the ELSD chromatogram (Figure 2B) this is most likely due to scattering differences as shape and size of analyte particle varies with different PS:PMMA ratios and elution solvent composition. Finally, for the UV chromatogram (Figure 2C) there is a consistent decrease in peak height as PMMA content is increased, corresponding to the decrease in the UV-active PS component. These visual observations of peak height/area are further supported by a more quantitative comparison of peak integrations for the three detectors' chromatograms (Figure S4).

Verification of Interface Integrity

Retention times were obtained from chromatograms of all three detection methods (microring resonators, ELSD, and UV) and representative chromatograms are presented in Figure 2. This was used to calibrate for copolymer composition, the % moles of PS were plotted against the PS/PMMA peak elution time for each polymer composition standard, as shown in Figure 3. Data points represent the average retention times for three replicated experiments (*n*=3). There is very little variance among individual retention times so error bars are small, this verifies interface robustness. Overlaying all three calibrations on the same axis highlights

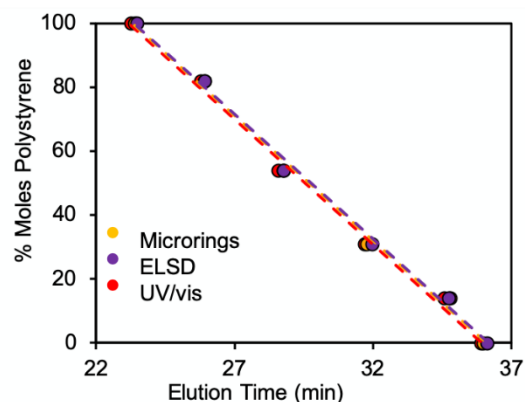


Figure 3: Copolymer composition versus elution times. Elution times from the chromatograms obtained in Figure 1 were plotted against % moles of polystyrene for each copolymer. Resulting in calibrations for copolymer composition. Plotting all three calibrations on the same axis show overlapping curves, verifying interface integrity such as no dead volume or delay between detectors.

indistinguishable slopes attributed by the agreement across detectors and lack of significant dead volume in the flow path. The fitting parameters of these curves can be found in Table 2. Additionally, these compositional calibrations can be useful for the identification of unknown PS-PMMA samples or blend samples that have multiple components or for computing the chemical composition distribution of a broad PS-PMMA copolymer of unknown composition.

Table 2: Fitting Parameters for Polystyrene Content Calibration

| | Detector | b x + c | | R ² |
|---|------------|---------|-------|----------------|
| | | b | c | |
| A | Microrings | -7.830 | 281.4 | 0.9954 |
| B | ELSD | -7.834 | 283.1 | 0.9962 |
| C | UV/vis | -7.829 | 282.5 | 0.9947 |

Mass Concentration Response Curves

Using the same PS-PMMA copolymer samples prepared in chloroform and the same experimental method discussed earlier, gradient separations were performed for various injected masses ranging from 0.15 to 0.75 mg for each detector.

Table 3: Linear Fitting Parameters for Mass Detection

| | Detector | Standard (%PS) | $b x + 0$ | | R^2 |
|---|------------|----------------|-----------|--|--------|
| | | | b | | |
| A | Microrings | 100 | 36.10 | | 0.9819 |
| | | 82 | 33.25 | | 0.9848 |
| | | 54 | 29.53 | | 0.9924 |
| | | 31 | 23.66 | | 0.9585 |
| B | UV/vis | 100 | 221.4 | | 0.9942 |
| | | 82 | 211.2 | | 0.9621 |
| | | 54 | 148.6 | | 0.9971 |
| | | 31 | 73.68 | | 0.9727 |

Table 4: ELSD Fitting Parameters for Mass Detection

| | Detector | Standard (%PS) | $b x^2 + c x + 0$ | | R^2 |
|---|----------|----------------|-------------------|-------|--------|
| | | | b | c | |
| C | ELSD | 100 | 38.35 | 110.3 | 0.9996 |
| | | 82 | 30.07 | 131.1 | 0.9953 |
| | | 54 | 72.11 | 73.34 | 0.9961 |
| | | 31 | 29.35 | 92.81 | 0.9921 |

Table 5: LOD and LOQ Comparison

| | Detector | Standard (%PS) | LOD (μg) | LOQ (μg) |
|---|------------|----------------|-----------------------|-----------------------|
| A | Microrings | 100 | 220 | 550 |
| | | 82 | 240 | 590 |
| | | 54 | 270 | 670 |
| | | 31 | 340 | 840 |
| B | UV | 100 | 0.0055 | 0.011 |
| | | 82 | 0.0057 | 0.011 |
| | | 54 | 0.0082 | 0.016 |
| | | 31 | 0.016 | 0.032 |
| C | ELSD | 100 | 0.0017 | 0.0043 |
| | | 82 | 0.0015 | 0.0036 |
| | | 54 | 0.0026 | 0.0065 |
| | | 31 | 0.0021 | 0.0051 |

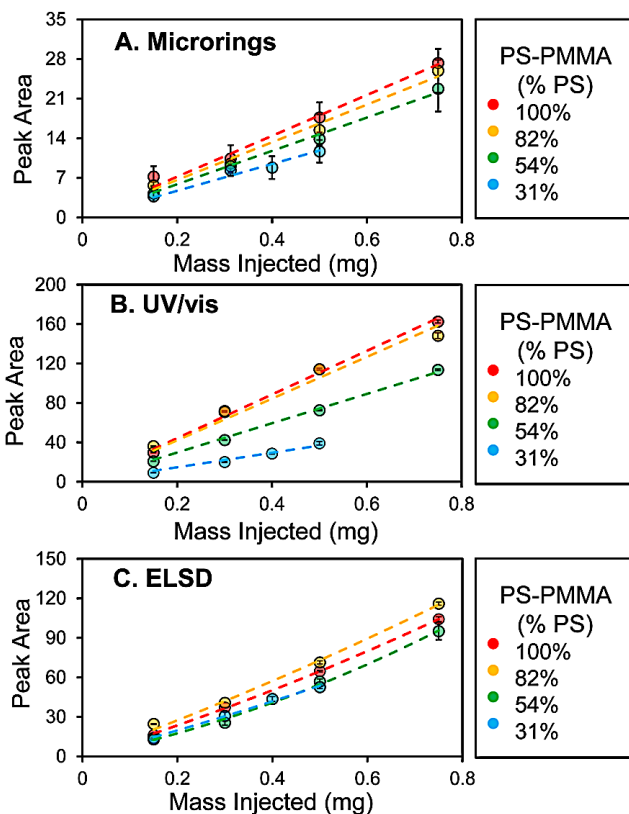


Figure 4: Mass Detection Calibrations

A. Repeating cyclohexane:THF gradient separations of PS-PMMA copolymers at 4 different injected masses for 4 standards demonstrated the linearity of the microring resonators. Plotting mass injected against peak area illustrates this linear correlation. **B.** Comparable linear correlation is also observed by UV/visible (UV) detection. **C.** However plotting mass injected against peak area for evaporative light scattering (ELSD) demonstrates the non-linearity of the detector. (Mass Range: 0.15-0.75 mg)

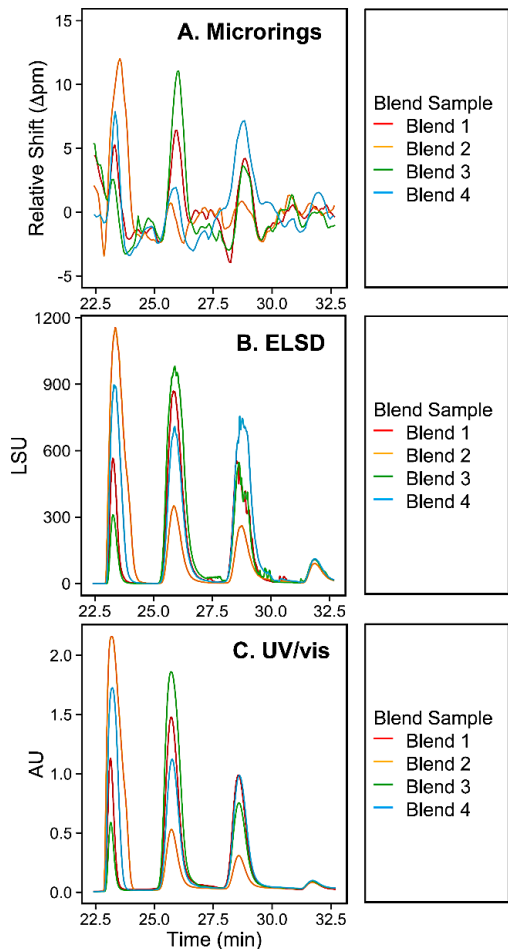


Figure 5: Copolymer blend analysis by various detectors

Polymer blends were made by mixing three PS-PMMA copolymers at various ratios all with a concentration of 11 mg/mL in chloroform. Separations were achieved based on composition using a cyclohexane:THF gradient. Chromatograms were obtained by detection with the **A.** microring resonators **B.** evaporative light scattering (ELSD) and **C.** UV/visible (UV).

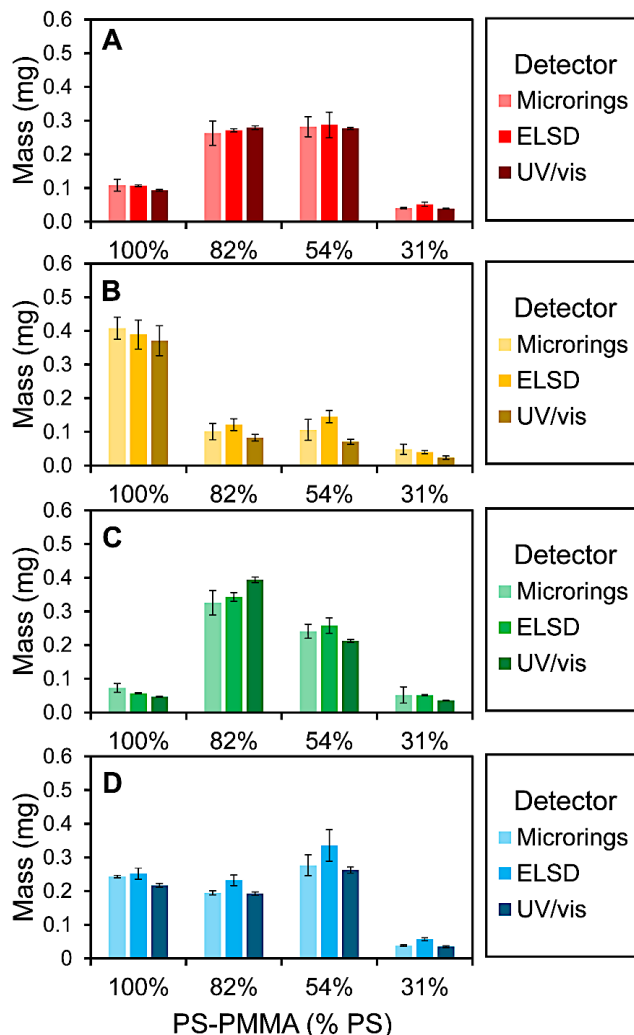


Figure 6: Quantitative Analysis of Polymer Blends

Integrating each peak area allowed for the quantification of mass detected for each component of the sample. This was done across all detectors allowing for a direct comparison, good correlation is observed since comparable mass values were obtained for each component by each method. Each histogram represents a different blend sample of the same three components **A.** Blend 1, **B.** Blend 2, **C.** Blend 3, and **D.** Blend 4.

Table 6: Polymer blend analysis, mass quantification

| | | Quantified Mass for Each Blend Component (mg) | | | | | | | |
|------------|-------|---|--------------------|-------------|--------------------|-------------|--------------------|-------------|--------------------|
| | | 100% PS | | 82% PS-PMMA | | 54% PS-PMMA | | 31% PS-PMMA | |
| Detector | Blend | Average | Standard Deviation | Average | Standard Deviation | Average | Standard Deviation | Average | Standard Deviation |
| Microrings | 1 | 0.11 | 0.02 | 0.26 | 0.04 | 0.28 | 0.03 | 0.040 | 0.002 |
| | 2 | 0.41 | 0.03 | 0.10 | 0.02 | 0.11 | 0.03 | 0.048 | 0.015 |
| | 3 | 0.072 | 0.013 | 0.33 | 0.04 | 0.24 | 0.02 | 0.05 | 0.02 |
| | 4 | 0.24 | 0.003 | 0.19 | 0.006 | 0.28 | 0.03 | 0.038 | 0.002 |
| UV | 1 | 0.093 | 0.003 | 0.28 | 0.005 | 0.28 | 0.003 | 0.038 | 0.0009 |
| | 2 | 0.37 | 0.04 | 0.08 | 0.01 | 0.070 | 0.007 | 0.024 | 0.005 |
| | 3 | 0.047 | 0.0011 | 0.39 | 0.008 | 0.21 | 0.004 | 0.036 | 0.0008 |
| | 4 | 0.22 | 0.006 | 0.19 | 0.005 | 0.26 | 0.01 | 0.035 | 0.002 |
| ELSD | 1 | 0.11 | 0.003 | 0.27 | 0.005 | 0.29 | 0.04 | 0.052 | 0.006 |
| | 2 | 0.39 | 0.04 | 0.12 | 0.02 | 0.15 | 0.02 | 0.040 | 0.005 |
| | 3 | 0.057 | 0.0013 | 0.34 | 0.013 | 0.26 | 0.02 | 0.051 | 0.002 |
| | 4 | 0.25 | 0.02 | 0.23 | 0.02 | 0.34 | 0.05 | 0.057 | 0.004 |

1
2
3 The 31% PS standard was measured over a narrower
4 mass range because of the pressure limits of the HPLC
5 system. Injecting too much mass of the later eluting
6 standards caused the HPLC system to go over
7 pressure. These standards are least soluble in the
8 initial gradient conditions, resulting in precipitation onto
9 the column and increased pressure. The mass
10 concentration response was investigated by plotting
11 mass injected against the integrated peak area for
12 each detection method allowing for mass calibrations
13 of 4 PS-PMMA standards, as seen in Figure 4. Here all
14 y intercepts were set to 0 as a means to normalize
15 across detectors. Comparing these mass calibration
16 curves demonstrates the linearity of the microring
17 resonator response (Figure 4A). The UV detector also
18 has a linear response for the 4 PS-PMMA standards
19 and slope offset correlates with PS content (Figure 4B).
20 The fitting parameters for the microring resonator and
21 UV response curves are found in Table 3. As for the
22 ELSD, the response is non-linear (Figure 4C) and the
23 fitting parameters are provided in Table 4. As
24 mentioned this non-linearity makes quantification
25 difficult to determine mass concentrations. Calibration
26 curve points represent the average of three replicate
27 experiments, and error bars represent the standard
28 deviation from visible over the plot range presented. As
29 mentioned above, we attribute the increased error of
30 the microring resonator platform to inherent noise of
31 the detector and the reduced precision of baseline
32 fitting for peak integration. Finally, limits of detection
33 (LOD) and limits of quantification (LOQ) were
34 calculated for each detector (Table 5). LOD and LOQ
35 are dependent on the standard composition since
36 detection differs slightly due to differing RI contrast for
37 the microrings, though this changes depending on the
38 analyte/mobile phase pairing. Therefore in cases
39 where RI contrast is low increasing sample size can be
40 advantageous especially in polymer separations where
41 sample size is rarely a limitation. The LODs and LOQs
42 for UV also depend on standard composition or more
43 specifically chromophore content, this however is a
44 more difficult challenge when the analyte of interest
45 lacks such chemical signature.

41 Polymer Blend Separations

42
43 Polymer blends were prepared as mixtures of
44 the various PS-PMMA copolymers at different ratios.
45 These included varying amounts of 100%, 82%, and
46 54% PS with the 31% PS standard used as an internal
47 control having constant concentration constant across
48 all 4 synthetic blends. For blend separations a total
49 mass of 0.75 mg was injected on the column and
50 detection was carried out using all three detectors.
51 Representative chromatograms for n=3 blend
52 separations are shown in Figure 5 (peak areas
53 presented in Figure S5), with the same elution order as
54 in Figure 2 (highest to lowest PS). Using the
55 compositional calibration presented in Figure 3 we can
56 identify each individual component of the blends, the
57 first eluted peak is 100%, second 82%, third 54% and
58 last eluted is the 31%. The 31% component was used

as an internal standard and the consistency of the
sample preparation across blends is verified by looking
at the overlapping 31% blend component.

Quantification and Analysis of Polymer Blend

Samples

The quantification of mass injected was done
using the corresponding calibration curve for each
peak component. Peak components were identified by
retention time. A comparison of quantification across
detectors for each blend can be found in Figure 6 and
Table 6, additionally actual mass are provided in Table
S1. In this comparison, similar values are observed for
each method, verifying the quantitative ability of the
microring resonators. The microring resonator platform
appears to offer some significant advantages over
traditional HPLC detectors. Its large dynamic range of
response enables detection of analytes on top of a
strongly sloping gradient baseline. Observation of the
gradient baseline (i. e., the gradient composition) itself
is another advantage, which is not possible with
traditional LC detectors (Figure 7). Real-time
monitoring of the solvent gradient can account for
fluctuations in pump performance and detection of non-
ideal gradients serving as a diagnostic tool to monitor
run integrity. Additionally, observation of the gradient
baseline enables detection of gradient distortion which
can compromise resolution. This in practice is
demonstrated in Figure 7, where a 100% cyclohexane
to 100% THF gradient was utilized without injections.
Using the programmed methods of Figure 7A the raw
gradient traces in Figure 7B were transformed into
relative shift (Δpm) as a function of %THF Figure 7C.
Here (Figure 7C) it is observed that with a steep
gradient (i.e. 8 minutes), which is equivalent to
approximately 1 column volume, there is a nonlinear
distortion in the raw gradient traces that is not observed
in methods of multiple column volumes. The 100 min
trace covers >10 column volumes and illustrates a
linear gradient which is ideal for separations of better
resolution. Gritti and Guiochon observed these same
distortion trends using reverse phase gradients.⁴⁸ On a
side note, comparing raw gradient traces of different
lengths (Figure 7B) there is the observation of different
gradient slopes due to varied rates of solvent mixing
however the overall change in RI is consistent.
Therefore, if run integrity was compromised it would be
obvious early on before translating to traces into a
function of solvent composition. In fact closer
examination of Figure 1A, which utilizes a truncated
gradient (95% (95:5 cyclohexane: THF) to 10%
cyclohexane) and has a gradient length of 20 minutes,
reveals a slight bit of curvature in a nominally linear
gradient. This curvature may indicate that our gradient
was slightly steeper than optimal, however further
investigation implied that time is not a factor since the
curvature was nearly the same across methods of
various lengths (Figure S6).⁴⁸ Similar curvature among
the various gradients does not appear to be an artifact

of the method itself but is likely due to the use of 5% strong solvent in reservoir A or inadequate solvent mixing. The use of strong solvent in reservoir A was necessary to maintain complete solubility of all analytes on injection. Although its presence appears to offer an advantage in terms of gradient ideality, it limits retention of copolymers with larger % styrene. The reason for minimal gradient distortion is that THF is always present in the column so there is minimal loss of THF as the gradient begins due to interaction of THF with the column. Further, UV absorbing solvents (e. g., toluene) are compatible with microring resonator platform, meaning direct monitoring of the gradient is possible. Finally, although all detectors used in this study required calibration for mass concentration determination, the microring resonator platform is believed to have an advantage over ELSD. Although not obvious for the system studied here, the ELSD typically exhibits both a solvent and polymer composition dependence to its response,^{24,49} and both of these dependencies are typically nonlinear. On the other hand, the microring resonator response is only dependent on the polymer composition eluting at a particular time in the gradient, and the response to that component is linear with concentration.

Conclusions

Commercial detectors compatible with gradient elution HPLC, such as ELSD and UV, have limitations for quantitative CCD measurements of industrially relevant polymers. Using microring resonators as gradient detectors for the separation of various PS-PMMA copolymers demonstrates advantages of the platform for complex polymer analysis. Foremost, the gradient compatibility of the microring resonator platform was exhibited, showing that refractive index based detection of gradient elution LC can be achieved unlike commercial dRI detectors. Here it was also observed that polymers like PMMA, which do not have a chromophore, can be detected by the microrings, providing an advantage over the commonly used UV detector. Additionally, various calibrations were performed for composition and mass injected, where mass calibrations illustrated the linearity of the microring resonator response. Linearity is where commonly used detectors like ELSD struggle since linearity is important for the ability to quantify mass concentrations. The quantitative results further support the applicability of the microring resonator platform for quantitative polymer analysis in solvent gradients.

However, it is important to also point out the challenges of the microring resonator platform as a solvent gradient chromatography detector. In comparison to commercial detectors, the baseline noise is larger. Because of the increased noise, there are additional processing steps for data analysis. Also, the decreased surface sensitivity with high molecular weights is another limitation. Commercial development of this platform for chromatographic separations would

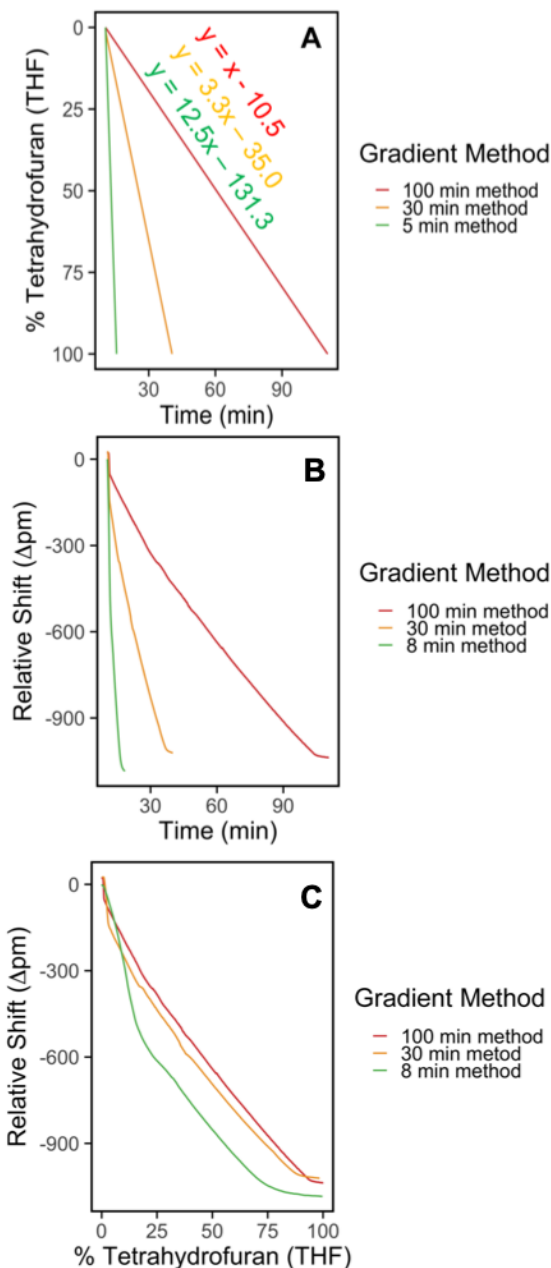


Figure 7: Real Time Monitoring of Solvent Gradient Baseline

(A) Here we wrote three gradient methods (100% cyclohexane to 100 % THF) of varying length. By plotting zoomed in traces (B) as relative shift versus time and (C) as a function of solvent composition demonstrates how gradient ideality can be directly evaluated and optimized. For example, the 8 min method (equivalent to 1 column volume) shows a non-ideal distorted trace which will limit resolution. An optimized trace is represented by the 100 min method which covers over 10 column volumes.

likely require improvements in each of those areas, which is an active area of continued research.

In summary, microring resonators offer much applicability to polymer analysis with broad versatility in a single detector. A universal linear mass concentration detector for use in solvent gradient HPLC is a longstanding challenge for separations. Despite the described limitations of the platform, this work represents an advancement toward a new detector technology for industrial copolymer analysis.

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Supplemental Information

Please find supplemental figures for interface and flow diagrams, peak integration comparisons and a table of actual polymer blend compositions.

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