



**Environmental
Science**
Nano

Graphite Nanoparticle Addition to Fertilizers Reduces Nitrate Leaching in Growth of Lettuce (*Lactuca sativa*)

Journal:	<i>Environmental Science: Nano</i>
Manuscript ID	EN-ART-08-2019-000890.R2
Article Type:	Paper

SCHOLARONE™
Manuscripts

1
2
3 Agricultural runoff is a leading cause of nitrate contaminated water which can lead to
4 eutrophication and human health impacts. Graphetic 2-D carbon nanoparticles combined
5 with fertilizer were found to reduce the amount of nitrate being leached through the soil and
6 increased the average lettuce yield over the non-nanoparticle dose in some of the treatments.
7 Lettuce is a high value crop grown in areas such as the Salinas Valley, California, which
8 have problems with nitrate contaminated groundwater and over fertilization. Additions of
9 nanoparticles to fertilizer blends can allow for a reduction in nutrient leaching in turn
10 preventing nutrient runoff and increasing nutrient availability to the plant.
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 **Graphite Nanoparticle Addition to Fertilizers Reduces Nitrate Leaching in Growth of**
4 **Lettuce (*Lactuca sativa*)**
5

6 Madelyn Pandorf^a, Leila Pourzahedi^b, Leanne Gilbertson^{c,d}, Gregory V. Lowry^b, Pierre Herckes^e,
7 and Paul Westerhoff^{a,*}
8
9

10
11 ^aSchool of Sustainable Engineering and the Built Environment (SSEBE)
12 Arizona State University
13 P.O. Box 3005, Tempe, Arizona 85287, USA
14
15

16
17 ^bCivil and Environmental Engineering Department
18 Carnegie Mellon University
19 5000 Forbes Avenue, Pittsburgh, Pennsylvania 15213, USA
20

21 ^cDepartment of Civil and Environmental Engineering
22 University of Pittsburgh
23 Pittsburgh, Pennsylvania 15261, USA
24
25

26 ^dDepartment of Chemical and Petroleum Engineering
27 University of Pittsburgh
28 Pittsburgh, Pennsylvania 15261, USA
29
30

31 ^e School of Molecular Sciences
32 Arizona State University
33 P.O. Box 871604, Tempe, Arizona 85287-1604, USA
34
35
36
37

38
39 *Corresponding author.
40 Tel.: 1-480-965-2885
41 E-mail address: p.westerhoff@asu.edu
42
43

44 **In Preparation for: Environmental Science: Nano**
45
46
47
48

49 **Date of Last revision: September 30, 2019**
50
51
52
53
54
55
56
57

Abstract

Nitrogen leaching into groundwater occurs in nearly all intensively-fertilized agriculture applications and poses growing environmental and human health risks such as eutrophication and drinking water contamination. This potential for contamination will intensify as the population grows. This study focused on nitrate leaching through soil during growth of romaine lettuce (*Lactuca sativa*), a high value crop in a region (Salinas Valley, CA) suffering from nitrate-contaminated water. 2-D graphite carbon nanoparticles (CNPs) produced via an electrochemical exfoliation process, resulting in ~8 nm thickness and 250–850 nm width, were combined with fertilizer and applied to the lettuce in soil to test the CNP effect on yield, nitrate leaching, and plant nutrient uptake. Greenhouse experiments were conducted under different nutrient loadings and soil matrices. CNP addition did not inhibit the lettuce leaf yield, and decreased nitrate leaching in several scenarios. When fertilizer was reduced to 70% of the recommended dose and combined with less than 1%wt CNPs, nitrate leaching decreased by 57%. Furthermore, there was no significant difference in yield compared to the 100% recommended fertilizer dose without CNPs. Increasing the soil's hydraulic conductivity enhanced the ability of CNPs to reduce nitrate leaching and increase plant nitrogen uptake. CNP addition to mineral fertilizer blends may allow lower fertilizer doses and thus decrease nitrate infiltration through the soil without comprising yields.

1. Introduction

Population growth is increasing the demand for food, thereby increasing agricultural activities and putting greater stress on the environment. Currently, fertilizers are often over-applied to maximize yield, which deteriorates sensitive environments.^{1,2} Nitrogen loading to surface water causes eutrophication and toxic algae blooms throughout the USA, and it is also a factor in the hypoxic region in the Gulf of Mexico at the Mississippi River delta.^{3,4} For example, Salinas Valley (California, USA) cultivates numerous high-value crops, and the associated nitrate contamination from fertilizers is a leading cause of coastal eutrophication in Monterey Bay.⁵ Nitrate is regulated in drinking water to prevent methemoglobinemia, but recent studies associated lower nitrate levels (0.5 to 5 mgNO₃-N/L) with a variety of cancers, and potentially in endocrine disruption.^{6,7} Nationally, adding nitrogen-based fertilizers for agriculture is also a primary factor for nitrate being the most frequently occurring groundwater pollutant. 20% of rural drinking water wells have nitrate above the Environmental Protection Agency's Maximum Contaminant Level (MCL),⁸ and over 250,000 people in Salinas Valley and Tulare Lake Basin are at risk of nitrate contamination to their drinking water source, where croplands have contributed 90% of the nitrate groundwater loading.⁵

Harter and Lund (2012) estimate that nitrogen fertilizer additions would need to decrease by 443 kg N/ha/crop for field crops in the Salinas Valley in order to reduce groundwater nitrate leaching to benchmark levels (35 kg N/ha/crop).⁵ Several practices exist to mitigate nitrogen runoff such as constructed wetlands, bioretention facilities, cover crops, and conservation tillage.^{9,10} However, the current best management practices can require combining multiple techniques and can drastically range in their effectiveness of reducing nutrient leaching.

1
2
3 45 Nanotechnology is an alternative to nitrogen fertilizers that can potentially improve
4
5 46 yields while also reducing the nutrient leaching.^{11, 12} Nanotechnology has shown to improve crop
6
7 47 growth, nutrient uptake, and seed germination.¹³⁻¹⁵ Nanomaterials are different from bulk
8
9 48 material in both their physical and chemical properties and have large surface to volume ratios.^{16,}
10
11 49 ¹⁷ The nanoparticle's effect on a plant will differ depending on the plant's growth stage, method
12
13 50 of application, exposure time and concentration, and the physical and chemical composition of
14
15 51 the nanoparticle.¹⁸ Carbon nanoparticles applied to plants can benefit or adversely influence soil
16
17 52 microbes and crop yields.¹⁹⁻²⁶ Many carbon nanoparticles have been studied for crop application,
18
19 53 including carbon nanotubes, fullerene, fullerol, graphene, carbon nanohorns, and carbon nano-
20
21 54 onions.^{19-21, 25-27} Carbon nanotubes had a positive effect on tobacco, maize, and alfalfa when
22
23 55 using a growth medium.^{24, 26, 28} However, negative effects on *Arabidopsis* growth and roots were
24
25 56 observed with water soluble fullerene C₇₀, revealing a disruption in cell division, mitochondrial
26
27 57 activity, and microtubule organization.²⁹
28
29
30
31
32

33 58 The effects of nanoparticles on crops are variable and require research to study specific
34
35 59 nanoparticle and crop interactions. This paper focuses on carbon-based nanomaterials because
36
37 60 there is limited work on the simultaneous effects of carbon nanoparticles on plant yield (i.e.,
38
39 61 increased food production) and nutrient leaching (i.e., environment impacts). Studies suggest
40
41 62 carbon nanoparticle additions to an artificial growth medium, such as Murashige and Skoog
42
43 63 medium, improve seed germination, but few studies use real soils or allow plants to grow for the
44
45 64 full harvest period. Lettuce, a high value crop for the southwestern United States, was chosen
46
47 65 due to its economic importance and nitrogen fertilizer requirements. The study focused on the
48
49 66 effects of graphitic carbon nanoparticles (CNPs) on nitrate leaching and lettuce yield using local
50
51 67 Arizona soil as the growth medium. The specific objectives were: (1) determine the effect of
52
53
54
55
56
57
58
59
60

1
2
3 68 CNPs on lettuce yield, (2) use a nutrient balance to identify whether nanoparticles impact
4
5 69 nutrient uptake and leaching, and (3) assess the effect of soil hydraulic conductivity on CNP and
6
7 70 nutrient mobility. Three trials were conducted; trials 1 and 2 varied nutrient loads, and trial 3
8
9 71 varied the soil hydraulic conductivity. Lettuce leaf yield, nutrient leaching, and nutrient uptake
10
11 72 into the plant tissue were collected for each trial.
12
13

14 73 **2. Materials and Methods**

15 74 **2.1. Planting Strategy and Harvesting**

16
17 75 Lettuce seeds (*Lactuca sativa*, var. Green Towers; High Mowing Organic Seeds) were
18
19 76 planted in 7.5 L plastic pots on June 13, 2017 (trial 1), September 8, 2017 (trial 2), and January
20
21 77 19, 2018 (trial 3) (Table S1). Figure 1 shows a photograph of the pots and diagram of the
22
23 78 leachate collection system equipped to each pot. Pots were lined with fiberglass mesh (Saint-
24
25 79 Gobain ADFORS) followed by 0.6 kg of gravel to filter soil from the leachate. Each pot received
26
27 80 3.3 kg of dry soil that was sieved using a U.S. Standard Sieve Series No. 10 (2000-micron mesh)
28
29 81 for a homogenous soil composition. Soil was saturated prior to planting, and fertilizer was
30
31 82 applied as a liquid to the soil surface (Section 2.5). To address germination issues, four seeds
32
33 83 were placed per pot in trial 1 and ten seeds were placed per pot in trial 2. In trial 2 due to
34
35 84 ongoing germination issues, pots that did not germinate received one transplant that were grown
36
37 85 in perlite three weeks after seeding. Seeds for trial 3 were grown in perlite (Vigoro Organic
38
39 86 Perlite) and then transplanted one plant per pot on February 9, 2018. In trial 1 and trial 2
40
41 87 experiments, seeds were thinned to one seedling per pot approximately 10 days after planting,
42
43 88 when the true leaves had developed. All trials had only one plant per pot that developed fully
44
45 89 until harvest.
46
47
48
49
50
51
52
53
54 90



91
92 **Figure 1.** Photographs of the leachate collection system equipped on each pot and the pots
93 arranged in the greenhouse (from left to right). On the right side, the layout of each treatment
94 within a randomized block design is shown for Trial 2.

96 Plants were grown in a temperature-controlled greenhouse ($24\pm 2^\circ\text{C}$) receiving natural
97 light located at Arizona State University ($33^\circ 25' 12.3''\text{N}$ and $111^\circ 55' 58.4''\text{W}$). Temperature
98 (Table S2) and photosynthetic active radiation (PAR) (Figures S1–S5) were monitored
99 continuously within the greenhouse, with the temperature controlled to remain in the optimum
100 range for lettuce growth.³⁰ To sufficiently saturate the soil near the root zone and to generate
101 leachate, each lettuce plant was drip irrigated twice daily with tap water for 1–2 minutes using a
102 poly tubing drip irrigation system and one Rain Bird emitter (3.79 L h^{-1}) per pot. The lettuce was
103 harvested after 7–8 weeks, and the wet and dry weight of the leaf and fully extracted root mass
104 were recorded. The roots were rinsed thoroughly to remove attached soil.

105 2.3. Carbon Nanoparticle Preparation and Characterization

106 The source of graphite carbon nanoparticles (CNPs) were based upon prior work by
107 collaborators in China who observed improved crop production when CNPs were blended with
108 fertilizers.³⁷ The CNPs (Hualong Fertilizer Technology Company, China) were developed for

1
2
3 109 agricultural use were produced via an electrochemical exfoliation process by applying an electric
4
5 110 pulse of 3–5 V through an inert cathode and a pure graphite anode in an electrolyte solution.³¹
6
7
8 111 Nanoparticles as-received were characterized for size and shape using transmission electron
9
10 112 microscopy (Philips CM200-FEG TEM). Energy dispersive spectroscopy (EDS) elemental
11
12 113 mapping was used for surface composition (JEOL 2010 FEG TEM), and a carbon hydrogen
13
14 114 nitrogen (CHN) analyzer (Perkin Elmer PE2400) was used for elemental composition. Pore size,
15
16 115 pore volume, and surface area were determined by measuring and plotting the gas adsorbed
17
18 116 versus the relative equilibrium (Micrometrics Tristar II 3020).³² The Brunauer–Emmett–Teller
19
20 117 (BET) equation was used to determine surface area, and Barrett–Joyner–Halenda (BJH)
21
22 118 methodology was used for pore size and volume. Surface thickness was analyzed by atomic
23
24 119 force microscopy (Bruker MultiMode 8 AFM). The topographical images were taken in peak
25
26 120 force tapping mode with non-contact cantilevers (NCHV) with a spring constant of 42 N/m
27
28 121 (Bruker, Camarillo, CA). Image analysis used the Nanoscope Analysis version 1.7 software
29
30 122 (Figure S10).

35 123 **2.4. Soil Types**

37 124 Local Arizona top soil (15–30 cm from surface) collected from the same plot at the
38
39 125 Maricopa Agricultural Center (33°04' 22' N, 111° 58' 26.5' W) was sieved and then used in all the
40
41 126 experiments. In trial 3, two additional soil blends were created by blending the Arizona soil with
42
43 127 sand to make a 30% sand (30S) and 70% sand (70S) by dry weight in order to increase the
44
45 128 saturated hydraulic conductivity. The Arizona soil was characterized as a Casa Grande clay loam
46
47 129 composed of 34.7% sand, 32.8% silt, and 32.5% clay using the United States Geological Survey
48
49 130 (USGS) Web Soil survey. All three soils had pH between 8.4 and 8.9, and the Arizona soil had
50
51 131 an organic matter content of 0.54%. General pre-planting soil sample compositions can be found
52
53
54
55
56
57
58
59
60

1
2
3 132 in Table S3. The Arizona soil, 30S, and 70S blends had characteristics similar to some of the soil
4
5 133 properties found in the Salinas Valley (USGS Web Soil Survey). The saturated hydraulic
6
7 134 conductivity for each soil type was determined using a UMS KSAT saturated hydraulic
8
9 135 conductivity meter and the method from its Operation Manual.³³ The saturated hydraulic
10
11 136 conductivity (k_{sat}) for Arizona soil, 30S, and 70S was 4.62×10^{-6} m/s, 5.23×10^{-6} m/s, and 3.35×10^{-5}
12
13 m/s, respectively (Table S3). A 15 cm long tubular soil sampler was used to take ten to twelve
14
15 137 soil cores from each pot post harvesting.
16
17 138

19 139 **2.5. Fertilizer Treatments**

20
21
22 140 Fertilizer treatments were configured in a randomized complete block design (Figures
23
24 141 S6–S8). Trials had a different number of treatments, dictated by the purpose of each trial:
25
26 142 varying one fertilizer dose (trial 1), varying three fertilizer doses (trial 2), and varying soil
27
28 143 hydraulic conductivity (trial 3). Supplied nutrients (ratio of N-P-K) were ammonium nitrate (34-
29
30 0-0) from ESKS and triple superphosphate (0-45-0) and muriate potash (0-0-60) both from
31
32 144 Fertilzona. Micronutrients were in the form of zinc sulfate, supplied from Fertilzona, and were
33
34 145 applied separately to prevent precipitation with the NPK fertilizer.
35
36 146

37
38 147 Fertilizer application rates (196 kg N/ha (175 lbs./acre), 67 kg P_2O_5 kg/ha (60 lbs./acre),
39
40 148 135 kg K_2O /ha (120 lbs./acre), and 3.4 kg ZnS/ha (3 lbs./acre)) were based on the Lettuce
41
42 149 Production in California guidebook and applied proportionally to each pot (Table 1).³⁴ Based on
43
44 150 successful experiments in China³⁷, CNPs were dosed at 3000 mg CNP/kg fertilizer to treatments
45
46 151 with a “+CNP” in the name. Fertilizer blends were applied as a liquid by dissolving the granular
47
48 152 fertilizer into tap water. The CNPs were combined with the fertilizer nutrient blends using
49
50 153 sonication and stir plates (details in the SI under Fertilizer Treatments).
51
52
53

54 154
55
56
57
58
59
60

1
2
3 **Table 1.** Nutrient application rates (kg/ha) for nitrogen, phosphorus, potassium, and CNPs for trial
4
5
6 1, trial 2, and trial 3 growing periods. To convert application rates from actual mass added per pot
7
8 157 (mg/pot) to field relative fertilizer applications, multiple values below by 4.1 ($[\text{mg/pot}] / [\text{kg/ha}]$).
9
10 158 Number of replicates per treatment that grew successfully were out of 6 plants for trial 1 and 4
11
12 159 plants for trails 2 and 3. †All fertilizer treatments in trial 3 were planted in 3 different soil types
13
14
15 160 ‡The CNP30S and 30-30S only had three replicates due to one plant dying mid-growing season.

Trial #	Fertilizer Treatment	Experiment Acronym	Number of Reps	Application Rates (kg/ha)			
				N	P ₂ O ₅	K ₂ O	CNP
1	No fertilizer	NT	6	0	0	0	0
1	CNP dosing only	CNP	6	0	0	0	2.85
1	100% recommended NPK dosing	100	4	196	67	135	0
1	CNP plus 100% recommended NPK	100+CNP	3	196	67	135	2.85
1	CNP plus 70% recommended NPK	70+CNP	6	137	47	94	2.00
2	No fertilizer	NT	3	0	0	0	0
2	CNP dosing only	CNP	4	0	0	0	2.85
2	100% recommended NPK dosing	100	2	196	67	135	0
2	CNP plus 100% recommended NPK	100+CNP	3	196	67	135	2.85
2	70% recommended NPK dosing	70	3	137	47	94	0
2	CNP plus 70% recommended NPK	70+CNP	2	137	47	94	2.00
2	50% recommended NPK dosing	50	3	98	34	68	0
2	CNP plus 50% recommended NPK	50+CNP	3	98	34	68	1.44
2	30% recommended NPK dosing	30	4	59	20	41	0
2	CNP plus 30% recommended NPK	30+CNP	1	59	20	41	0.85
3	No fertilizer (AZ, 30S, 70S soil) †	NT	4	0	0	0	0
3	CNP dosing only (AZ, 30S, 70S soil) †	CNP	4‡	0	0	0	0.85
3	30% recommended NPK dosing (AZ, 30S, 70S soil) †	30	4‡	59	20	41	0
3	CNP plus 30% recommended NPK (AZ, 30S, 70S soil) †	30+CNP	4	59	20	41	0.85

161
162 **2.5.1 Trial 1 & 2 Experimental Setup**
163
164 Trial 1 had five nutrient treatments replicated six times (Table 1) as a proof of concept
165 experiment to evaluate if CNPs had an effect on plant growth and nitrogen leaching through the
166 soil (Figure S6). Trial 2 had ten fertilizer treatments replicated four times to test the effect of
167 varying the fertilizer dose (Figure S7). The fertilizer treatments were as follows: no fertilizer
control (NT), carbon nanoparticles only (CNP), 100% mineral fertilizer (100), 100% mineral

1
2
3 168 fertilizer with CNP (100+CNP), 70% mineral fertilizer (70), 70% mineral fertilizer with CNP
4
5 169 (70+CNP), 50% mineral fertilizer (50), 50% mineral fertilizer with CNP (50+CNP), 30%
6
7 170 mineral fertilizer (30), and 30% mineral fertilizer with CNP (30+CNP). The 70+CNP, 50+CNP,
8
9 171 and 30+CNP treatments received 30%, 50%, and 70% less nutrients and nanoparticles than
10
11 172 100+CNP, respectively. The CNP-only dose received the same amount of carbon nanoparticles
12
13 173 as 100+CNP. Due to germination issues in trial 2, pots that did not germinate received one
14
15 174 transplanted lettuce plant that were grown in perlite three weeks after initial seeding. Plants that
16
17 175 were transplanted were staggered by three weeks (harvest 2) from the initial seedlings (harvest
18
19 176 1). Harvest 1 plants were all grown and harvested in the same greenhouse; however, harvest 2
20
21 177 plants were moved to another greenhouse due to greenhouse maintenance for the final three
22
23 178 weeks of their growing cycle. No effect on yield from moving greenhouses was observed
24
25 179 between harvest 1 and harvest 2 plants.

30 180 **2.5.2 Trial 3 Experimental Setup**

31
32
33 181 Trial 3 tested the effect of hydraulic conductivity (increasing drainage by adding sand) on
34
35 182 CNP performance using four fertilizer treatments (NT, CNP, 30, and 30+CNP) and three soil
36
37 183 blends. The three soil blends were 100% Arizona soil (AZ), Arizona soil blended with 30% sand
38
39 184 by dry weight (30S), and Arizona soil blended with 70% sand by dry weight (70S). The sand
40
41 185 (DecoRock Paver Sand) was dried and sieved through a 2000 micron mesh before blending.
42
43 186 Thirty percent of the recommended nutrient requirements were used for all the mineral fertilizer
44
45 187 doses. All fertilizer treatments were added at a 30% fertilization rate to examine if the effects on
46
47 188 yield could be attributed to changes in hydraulic conductivity and CNP addition and not over
48
49 189 fertilization. Trial 3 fertilizer treatments were: no treatment in Arizona soil (NT AZ), no
50
51 190 treatment in 30% sand blend (NT 30S), no treatment in 70% sand blend (NT 70S), carbon
52
53
54
55
56
57
58
59
60

1
2
3 191 nanoparticles only in Arizona soil (CNP AZ), carbon nanoparticles only in 30% sand blend
4
5 192 (CNP 30S), carbon nanoparticles only in 70% sand blend (CNP 70S), 30% mineral fertilizer in
6
7 193 Arizona soil (30 AZ), 30% mineral fertilizer with CNP in Arizona soil (30+CNP AZ), 30%
8
9 194 mineral fertilizer in 30% sand blend (30-30S), 30% mineral fertilizer with CNP in 30% sand
10
11 195 blend (30+CNP 30S), 30% mineral fertilizer in 70% sand blend (30-70S), and 30% mineral
12
13 196 fertilizer with CNP in 70% sand blend (30+CNP 70S).
14
15
16

17 197 **2.5.3 Nitrogen Mass Balances**

18
19 198 The nitrogen in leachate, plant tissue, and soil were used to formulate a nutrient balance
20
21 199 for each trial. Equation 1 was used to calculate the percentage and mass of applied nitrogen from
22
23 200 fertilizer that ended up in the leachate and plant tissue.
24
25

26 201 Equation 1 ***Mass or Percent of applied N*** = $\frac{N - N_o}{TF}$

27
28
29 202 Where N is the amount of nitrogen in the leachate or plant tissue, N_o is the amount of nitrogen in
30
31 203 the NT (i.e., no fertilizer treatment) leachate or NT plant tissue, and TF is the total amount of
32
33 204 nitrogen applied from fertilizer. Details are provided in the Supplemental Information; Tables
34
35 205 S13–S20 contain the data used for the calculations.
36
37
38
39
40

41 207 **2.6. Leachate Collection and Measurement**

42
43 208 Lysimeters were used to collect leachate. The lysimeters used a funnel at the bottom of
44
45 209 each pot that flowed into a leachate collection reservoir (Figure S9). Leachate was collected bi-
46
47 210 weekly or as needed. Volume, conductivity, and pH were measured upon collection using a
48
49 211 graduated cylinder, Oakton ECTestr 11+ meter, and Oakton pHTestr 30 meter, respectively.
50
51 212 Both probes were calibrated prior to leachate collection. Leachate was then filtered using a 0.2
52
53
54
55
56
57
58
59
60

213 μm Nylon membrane filter (Environmental Express) and analyzed for anions and cations (Table
214 S4).

215 **2.7. Analytical Methods**

216 The wet and dry weights of the lettuce leaf and extracted root mass were recorded for
217 each pot. Lettuce leaf yield is the recorded wet weight of the leaves. Plant tissue was dried at
218 60°C for one week, ground (Thomas Scientific 3383-L10 Wiley Mill), and sent to Waters
219 Agricultural Labs, Inc. for nutrient analysis (Table S4).³⁵ The soil composition was digested
220 using the Mehlich 3 acid extraction method by Waters Agricultural Labs, Inc. and analyzed for
221 nutrients, metals, soil pH, and cation exchange capacity (CEC) (Table S4). Metals and
222 phosphorus in the plant tissue and soil were measured on a iCAP TQ ICP-MS (Thermo-Fisher
223 Scientific) using a Mehlich 3 acid soil extraction method and a plant tissue wet digestion Digi
224 Block 3000 method. Cations (ammonium, calcium, sodium, potassium, magnesium) and anions
225 (nitrate, phosphate, chloride, sulfate) in the leachate aqueous solutions were measured at Arizona
226 State University on an ion chromatograph (Dionex ICS-5000 DC) equipped with an IONPAC
227 column AS18 for anion and CS12-A for cation (Table S4).

229 **2.8. Data Analysis**

230 Statistical analysis was conducted on the lettuce leaf yield for trial 1 and trial 2 combined
231 and trial 3 experiments. Data was analyzed using IBM SPSS 24.0 software. A one-way ANOVA
232 was used with treatment as the independent variable and yield as the dependent variable. A
233 confidence interval of $p < 0.10$ was chosen due to variability within plant growth, which
234 increased data uncertainty. The leaf yield for trial 1/trial 2 combined data and trial 3 data were
235 log-transformed to a normal distribution. A Welch correction was used on the combined trial

1
2
3 236 1/trial 2 data due to a violation in the assumption of homogeneity of variances with analysis
4
5 237 using a Games-Howell post-hoc test ($p < 0.10$) for yields considered significantly different. Trial
6
7 238 3 data passed the Shapiro-Wilks and Levene's Test, and a one-way ANOVA was then conducted
8
9 239 with a Tukey's HSD post-hoc test ($p < 0.10$).

240 **3. Results**

241 **3.1 CNP Characterization**

242 SEM and TEM were used to characterize the CNP size and structure (Figures S10–S12).
243 SEM images exhibited a range of aggregated and single particle sizes (Figure S11). The TEM
244 images showed a combination of a crystalline structure made of parallel graphitic sheets and
245 amorphous carbon defects (Figure S10a and Figure S12a-c). The CNP sizes ranged from 250 to
246 850 nm wide and < 5 to ~ 40 nm in thickness (Figures S10a, S10d, S11). The average particle
247 thickness was 8 nm determined from 100 particle cross sections using AMF in conjunction with
248 the section function in the Nanoscale software (Figures S10c and S10d). The Raman spectrum
249 (Figure S10b) showed a E_{2g} peak at 1573 cm^{-1} (G-band), which indicates a graphitic
250 composition, and a disorder-induced peak at 1354 cm^{-1} (D-band), which is attributed to defects in
251 the structure.^{36, 37} The defects are in the form of amorphous carbon and do not have crystalline
252 structures seen in Figure S12a. The CHN results showed that the CNPs are approximately 96%
253 carbon, primarily composed of graphite and amorphous carbon structures seen in the TEM
254 images. Copper, calcium, potassium, sodium, oxygen, and other elements compose the
255 remaining 4% (Figure S12d). The average BET surface area was $1.82\text{ m}^2/\text{g}$, and the Barrett–
256 Joyner–Halenda (BJH) pore volume was $0.0072\text{ cm}^3/\text{g}$. The CNPs had negative zeta potentials
257 between pH 4 to 10 (Table S5).

258 **3.2 Effect of NPK Loading and Hydraulic Conductivity on Leaf Yield**

3.2.1. *Effect of Nutrient Loading on Yield*

One hundred percent nutrient addition without CNP (labeled as “100”) and 70+CNP treatments (i.e., 70% of the recommended nutrients plus CNP) produced the two highest average leaf yields (Figure 2a and Tables S6-S7). Compared against 100+CNP, the 70+CNP treatment was the optimum treatment to lower fertilizer inputs with minimal reduction to leaf yield. Alone amongst the treatments, the 70+CNP treatment enhanced the average leaf yield by 17% compared with the non-CNP treatment at the same fertilizer dose (Figure 2a and Table S8). The average leaf yield declined significantly with the 50+CNP and 30% treatments compared to the 100%, 70%, and non-CNP 50% doses (Figure 2a and Table S8). In the combined trial 1 and 2 statistical analysis, there was no statistical difference in yield between plants fertilized with 100% and 70% treatments ($p>0.10$), implying less fertilizer can be applied and still achieve adequate growth. There was no statistical difference ($p>0.10$) in leaf yield between treatments with and without CNPs except for the 50% treatment (Table S8). All plants treated with NPK had significantly higher leaf yields ($p<0.10$) than the controls (no fertilizer added), showing that the soil alone is not supplying sufficient nutrients for optimal growth, and an additional nutrient source is necessary.

3.2.2. *Effect of Hydraulic Conductivity on Yield*

A 30% fertilizer treatment for trial 3 was used to understand if changes in yield are due to differences in hydraulic conductivity and CNP addition and not from over fertilization. The 30+CNP 30S and 30+CNP AZ treatments produced the highest average leaf yields followed by 30 AZ and 30-30S (Figure 2b and Table S9). The 30+CNP 30S treatment increased the average leaf yield by 24% over the non-CNP treatment, making it the optimum blend for CNPs to enhance growth. Increasing the hydraulic conductivity of the soil allowed for increased average

leaf yield in the plants treated with CNPs versus the non-CNP plants (Figure 2b). All fertilized treatments were significantly higher in leaf yield than the controls.

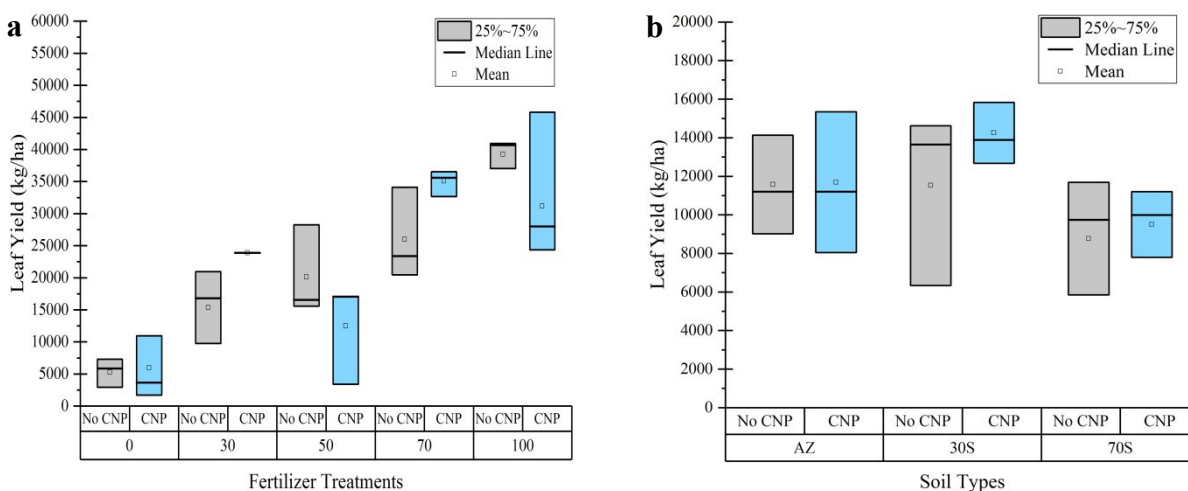
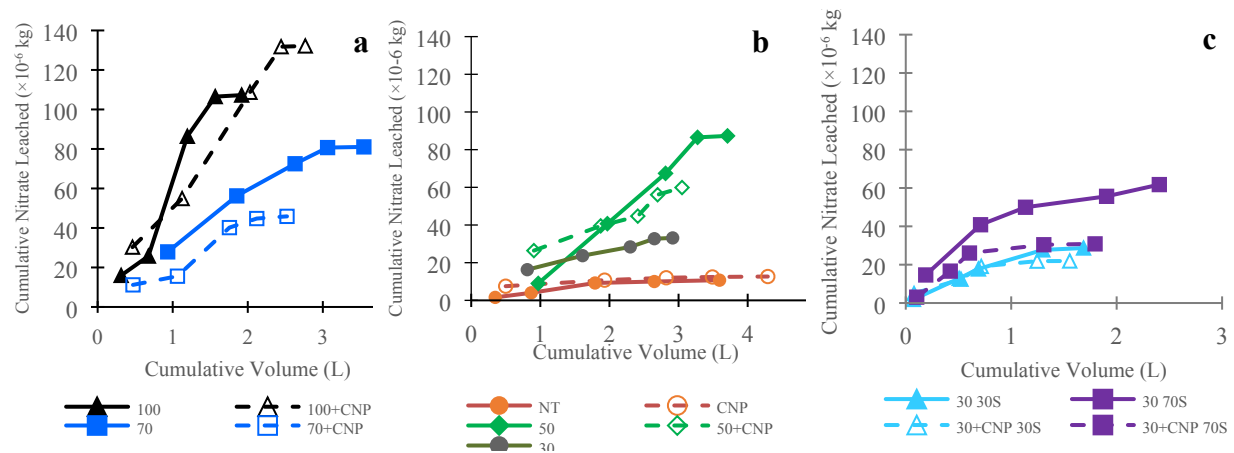


Figure 2. Aggregate leaf yield data across all growing seasons when varying a) fertilizer treatments as 0, 30%, 50%, 70% or 100% of the recommended NPK dose and, b) soil hydraulic conductivity with 0, 30% (30S) or 70% (70S) sand mixed into the soil. Bar and whisker diagrams show mean, median and quartile ranges (box). Blue boxes are for treatments that include nutrients and CNP; grey boxes are for nutrient addition without CNP.

3.3 Nitrogen Leaching

This study investigated whether CNP addition with fertilizers can reduce the amount of nutrients leaching from the soil with nitrogen being the main concern. Of the nitrogen species analyzed, >95% of the leached nitrogen was in the form of nitrate, and thus nitrate was the focus in the results. Leachate data for all other nutrients can be found in Tables S10–S12. The 30+CNP data was not included because only one replicate grew and was not considered a representative sample (Table 1).



297 **Figure 3.** Cumulative nitrate leached ($\times 10^{-6}$ kg) versus cumulative volume leached (L) for: a)
 298 100, 100+CNP, 70, 70+CNP treatments (averages from trials 1 and 2), b) NT, CNP, 50,
 299 50+CNP, 30 treatments (averages from trials 1 and 2), and c) 30S and 70S soils in trial 3

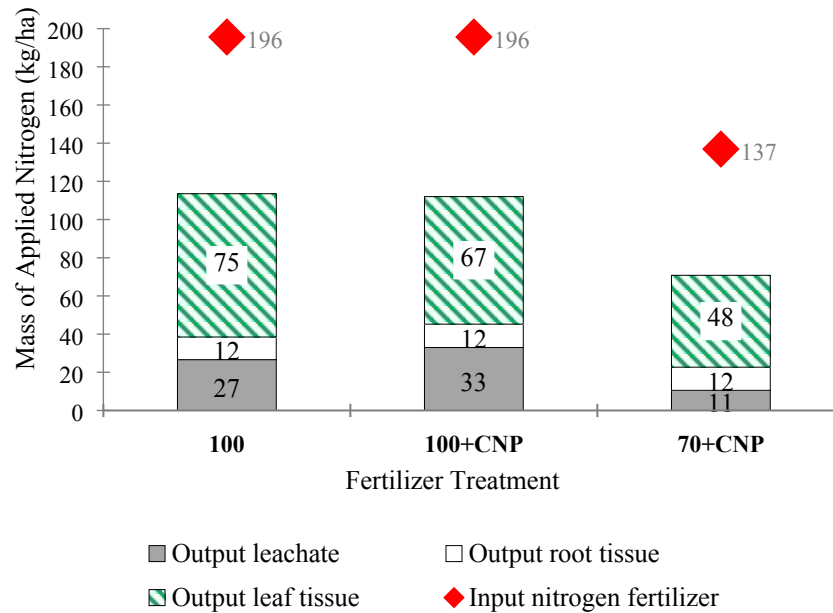
300
 301 There was a trend of lower nitrate leaching in the 70+CNP treatment compared to the
 302 70% fertilizer dose without CNP. In experiments where the nutrient loading varied (Figure 3a
 303 and b), the 70+CNP treatment on average had 43% less nitrate leached than the 70 treatment.
 304 This translates to an average of 19.7 kg NO_3^-/ha leached for the 70 treatment and 10.8 kg NO_3^-
 305 /ha leached for the 70+CNP treatment in trial 2 (Table S11). The CNPs did not have an effect on
 306 the 100% treatments possibly due to an oversaturation of nitrogen or phosphorous (Figure 3a and
 307 Table S10 and S11). The 50% and 70% fertilizer treatments with CNPs had a lower average
 308 nitrate leached compared to non-CNP treatments (Figures 3a and 3b).

309 When the hydraulic conductivity varied (trial 3), the 30+CNP 30S and 30+CNP 70S
 310 treatments leached less nitrate compared to the non-CNP treatments, indicating the CNPs had a
 311 role in reducing nitrate leaching (Figure 3c). The CNPs did not have an effect in the Arizona soil
 312 in trial 3 and leached more nitrate while having similar yields to the non-CNP treatment (Table
 313 S12). The amount of nitrate leached between the NT and CNP treatments was similar between all

1
2
3 314 soil types. Furthermore, there was also less nitrogen leaching from the 70S soil type, because
4
5 315 with 70% sand in the matrix the 70S initially contained less ambient or naturally-occurring
6
7 316 nitrogen (Table S12). In both the 30-30S and 30-70S treatments, the plants with CNP had a
8
9 317 higher average yield and a lower amount of nitrate leached. There was a 22% reduction in nitrate
10
11 318 leached between the 30-30S and 30+CNP 30S treatments and a 55% reduction between the 30-
12
13 319 70S and 30+CNP 70S treatments, when comparing treatments with and without CNPs added to
14
15 320 the fertilizer.
16
17
18

19 321 **3.4 Nutrient Balance**

20
21 322 To understand nitrogen mobility and effect of CNPs, a nutrient balance was conducted.
22
23 323 Nitrogen (N) was the only nutrient considered due to the concern of nitrate contamination in
24
25 324 water bodies. In trial 1, over 80% of the applied nitrogen (from fertilizer) was recovered in the
26
27 325 leachate, plant tissue (leaf and root), and soil (Table S13 Part A and B). Of the N supplied
28
29 326 through fertilization, 14%, 17%, and 8% was recovered in the leachate, and 44%, 40%, and 44%
30
31 327 was in the plant tissue (leaves and root) for treatments 100, 100+CNP, and 70+CNP, respectively
32
33 328 (Figure 4 and Table S13 Part C). None of the treatments fell within the optimum nitrogen content
34
35 329 range (33–48 g N/kg lettuce, according to Hartz and Johnstone (2007) for high lettuce leaf yield
36
37 330 production (Table S14).¹ The 70+CNP had approximately half of the average amount of applied
38
39 331 nitrogen leached compared to the 100+CNP and 100 fertilizer treatments (Figure 4). All
40
41 332 treatments had approximately the same amount of nitrogen uptake into the root tissue. For the
42
43 333 100% and 70% treatments, the amount of leached nitrogen decreased as plant yield increased
44
45 334 (Figure S13–S14).
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60



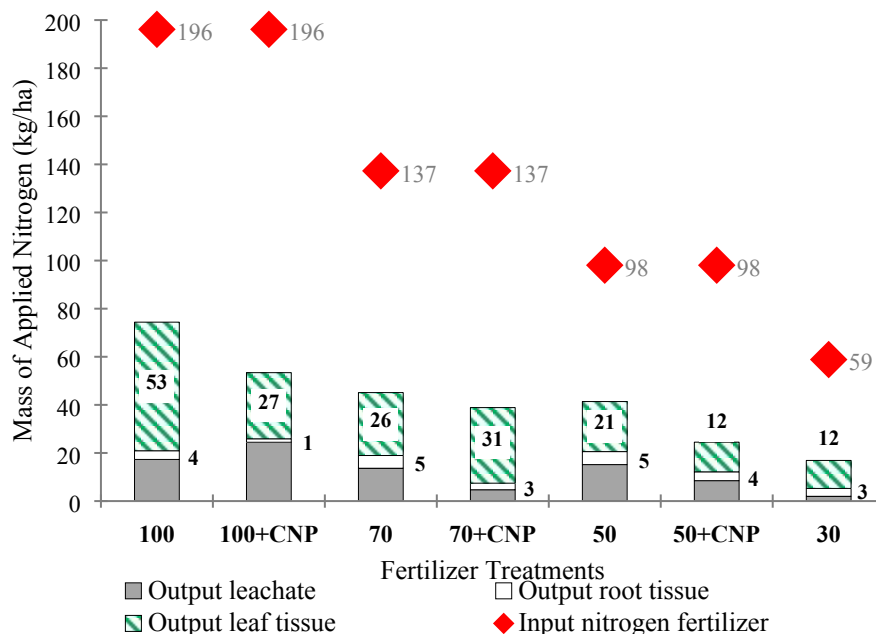
336

337 **Figure 4.** Trial 1 nutrient balance comparisons for the mass of applied nitrogen (kg/ha) against
 338 the nitrogen outputs from the pot in the form of leachate (gray bar), leaf tissue (hashed-marked
 339 green bar), and root tissue (white bar).

340

341 The largest portion of *applied* nitrogen was found in the plant tissue for trial 2, which
 342 follows the trial 1 data trends (Figures 4 and 5). The plant tissue nitrogen of CNP treatments was
 343 higher than the non-CNP treatments for the 70% treatment only (Figure 5 and Table S15). In
 344 Table S15, the 100, 70, 70+CNP, 50, and 30 experiments all had similar uptake efficiencies of
 345 applied nitrogen from fertilizer into the plant tissue averaging ~25%. However, only treatments
 346 100, 100+CNP, 70+CNP, and 50 fell within optimum nitrogen content concentrations for high
 347 yield lettuce production (Table S16).¹ The 100+CNP showed an increase in nitrogen leached
 348 compared to the 100 treatment; in addition, the 100+CNP nitrogen tissue content was lower than
 349 the 100 dose, indicating nitrogen was leached through the soil profile instead of being adsorbed
 350 by the roots. The 70+CNP and 50+CNP treatments reduced the average amount of applied

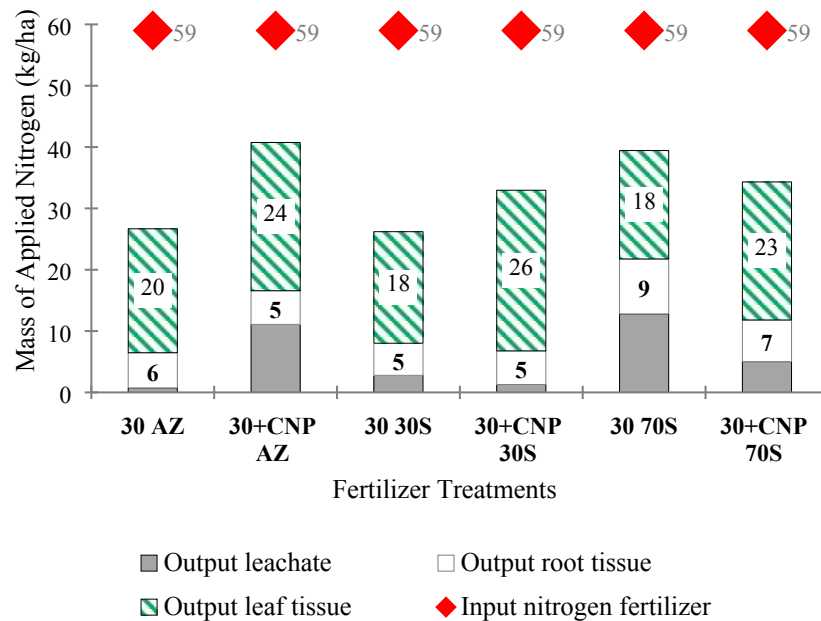
351 nitrogen leached by half compared to the 70 and 50 treatments (Figure 5). There was no
 352 consistent trend of CNP treatments having higher nitrogen uptake in the plant tissue, implying
 353 that for this trial CNPs played a larger role in nitrogen leaching rather than in uptake.



354 **Figure 5.** Trial 2 nutrient balance comparisons for the mass of applied nitrogen (kg/ha) against
 355 the nitrogen outputs from the pot in the form of leachate (gray bar), leaf tissue (hashed-marked
 356 green bar), and root tissue (white bar).

357
 358
 359 In experiments with varying soil hydraulic conductivity (trial 3), less nitrogen leached and more
 360 nitrogen was in the plant tissue for the 30+CNP 30S and 30+CNP 70S treatments compared with
 361 the non-CNP treatments (Figure 6 and Table S18). In Figures 4 through 6, less nitrogen was
 362 accounted for in the sum of leachate plus root and leaf tissue compared against the mass of
 363 nitrogen fertilizer added. Part of the unaccounted for nitrogen is due to biogeochemical processes
 364 that produce N-gases by soil microbes. Because the mass of nitrogen added as fertilizer, to the
 365 mass of soil, represented a small fraction of the nitrogen in the soil before or after the

366 experiments, there was no statistical differences in soil nitrogen content across the duration of
 367 the growth experiment. Compared against experiments without CNP addition to the fertilizers,
 368 adding CNPs had on average 12–13% more nitrogen in the plant tissue (Table S19). The 70S
 369 treatments had a lower nutrient content due to less naturally-available nutrients in the soil. The
 370 nitrogen leached was reduced by over half between the 30+CNP and 30 treatments in the 30S
 371 and 70S soil compositions. In the soils with increased hydraulic conductivity, the CNPs
 372 increased yield and nitrogen in the plant tissue and decreased the amount of nitrogen being
 373 leached compared to non-CNP treatments (Figure S15 and Table S20).



374

375 **Figure 6.** Trial 3 nutrient balance comparisons for the mass of applied nitrogen (kg/ha) against
 376 the nitrogen outputs from the pot in the form of leachate (gray bar), leaf tissue (hashed-marked
 377 green bar), and root tissue (white bar). 30% recommended NPK fertilizer loading rates, both with
 378 (+CNP) or without graphitic nanoparticles, in three soil matrices: 1) Arizona soil (AZ), 2) 30%
 379 sand blended with 70% Arizona soil (30S), and 3) 70% sand blended with 30% Arizona soil
 380 (70S).

1
2
3 3814
5 3826
7
8 383 **4. Discussion**9
10 384 ***4.1 Effect of Nutrient Loading and CNPs on Yield***

11
12 385 The 70+CNP and 30+CNP 30S treatments were the two optimal scenarios to increase
13
14 386 lettuce yield compared with the non-CNP treatments (Figure 3). There were no statistically lower
15
16 387 leaf yields between CNP and non-CNP treated plants, except in the 50% treatment, implying that
17
18 388 CNPs did not hinder growth (Tables S8–S9). The average leaf yields were highest in the 100%
19
20 389 and 70% treatments and began to decline for the 50% and 30% treatments. This decline in
21
22 390 average yield for lower nutrient doses implies that at least 70% of the recommended nutrients are
23
24 391 needed for adequate growth. The 100 non-CNP and 70+CNP treatments were the only treatments
25
26 392 with average yields (39,227 and 35,085 kg/ha, respectively) above the average romaine lettuce
27
28 393 yield in the Salinas Valley (~34,800 kg/ha).³⁸ The CNPs in this study have been used previously
29
30 394 at the lab scale and in field trials and have shown a significant positive impact on growth through
31
32 395 the following suggested mechanisms: increased nutrient adsorption and active transport of
33
34 396 nutrients to the roots; enhanced synthesis of starch, which in turn increased the carbohydrate
35
36 397 production and photosynthesis; and increased mitochondria within the plants.³⁷ These
37
38 398 mechanisms could be a contributing factor in the increase in average leaf yield found in *some* of
39
40 399 the CNP fertilized plants in this study.

41
42 400 The Arizona soil structure and composition were one of the factors believed to hinder the
43
44 401 effects of CNPs on yield due to germination issues caused by mechanical resistance where the
45
46 402 seedling had difficulty emerging through the soil.³⁹ The complexity of using a soil matrix as a
47
48 403 growth medium could potentially also lower the CNP effectiveness. Other studies involving
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 404 different types of carbon nanomaterials showed a positive effect on germination and yield when
4
5 405 using an artificial growth medium or directly applying the CNPs to the seeds.⁴⁰⁻⁴² A study on
6
7 406 tomato plants found that nanoparticle size, shape, crystallinity, surface chemistry, and charge
8
9
10 407 played an important role on physiological responses of tomato plants.^{41, 43} The higher the
11
12 408 negative surface charge and better dispersed the functionalized CNTs were, the larger the tomato
13
14 409 growth.⁴¹ The CNPs used in our study did not stay well dispersed in water, and potentially
15
16
17 410 favored instability and aggregation in the fertilizer solution, which may have impacted their
18
19 411 effect on yield. The alkaline nature of Arizona soils may also impact attachment of negatively
20
21 412 charged CNPs during infiltration. In addition, the CNPs in this study varied greatly in size and
22
23 413 had crystalline and non-crystalline structure, which may have caused the variability in increasing
24
25
26 414 yield in only some treatments.

28 415 ***4.2 Effect of Soil Properties on Yield, Nutrient Mobility, and CNP Effectiveness***

30
31 416 Soil conductivity can affect CNP performance and nutrient availability. In addition, the
32
33 417 high pH (pH 8.4) and calcareous nature of the Arizona soil are not conducive for plant growth
34
35 418 because they can impede nutrient availability (Table S3).^{44, 45} Because CNPs did not consistently
36
37 419 improve yield and lower nitrate leaching for all treatments in trials 1 and 2, it can be concluded
38
39 420 that soil composition is an important factor in the effectiveness of CNPs. Soil properties, such as
40
41
42 421 the high clay and silt contents in the Arizona soil, impact the behavior and mobility of
43
44 422 nanoparticles and affect their homo- and hetero-aggregation due to their high surface area and
45
46
47 423 small particle size.⁴⁶⁻⁴⁸ Nanoparticles can hetero-aggregate with natural soil colloids, which
48
49 424 reduces their mobility within the soil matrix.^{46, 47} The Arizona soil had a high clay and silt
50
51 425 content that could bind the CNPs and therefore lower their effectiveness. This was evident in
52
53 426 trial 3; the 30S and 70S soil blends, which had lower amounts of clay and silt particles and
54
55
56
57
58
59
60

1
2
3 427 organic matter content, showed CNPs to have an improved effectiveness in leaf yield and
4
5 428 nitrogen uptake and reduced nitrate leaching compared to the AZ soil with CNPs (Figure 2b and
6
7 429 Table S20). This trend may indicate that the increase in hydraulic conductivity allowed for
8
9 430 increased movement of the CNPs from the soil surface to the root zone and therefore enhanced
10
11 431 plant growth and reduced leaching. However, the 70S soil blend also had less naturally-available
12
13 432 nutrients, which may be why those treatments had a lower average leaf yield (8,770–9,500
14
15 433 kg/ha) compared to the AZ (11,500–11,700 kg/ha) and 30S (11,500–14,250 kg/ha) soil types
16
17 434 (Table S20). The increase in hydraulic conductivity will also increase the leaching ability of
18
19 435 nutrients; however, CNP adsorption properties can reduce nutrient leaching. This was observed
20
21 436 in the 70S soil blend; the CNPs most likely increased adsorption of nitrogen species and
22
23 437 therefore reduced the amount of nitrate being leached. This implies that increasing the saturated
24
25 438 hydraulic conductivity through sand addition can allow for better mobility of the nanoparticles to
26
27 439 adsorb ions and less hetero-aggregation with soil particles with the tradeoff of less naturally-
28
29 440 available soil nutrients.

35 441 ***4.3 Effect of CNP Properties on Nutrient Mobility***

36
37 442 The nitrate leaching data between CNP and non-CNP treatments in trials 1 and 2 had
38
39 443 varying results in which the 70+CNP treatment was the optimal treatment for increasing yield
40
41 444 while also lowering nitrate leached compared to the non-CNP treatment (Table S17 and Figure
42
43 445 S14). The 70+CNP results are important in that there was over a 56% reduction of the nitrate
44
45 446 leached between the 70+CNP and 100 treatments without compromising the leaf yield in both
46
47 447 trial 1 and trial 2 (Figure 2a). In addition, trial 3 showed that increasing the hydraulic
48
49 448 conductivity (drainage) of the soil made CNPs more effective in reducing nitrate leaching,
50
51 449 improving yield and nitrogen uptake into the plant (Figure 2b and S15). We hypothesize for
52
53
54
55
56
57
58
59
60

1
2
3 450 future research that the improved CNP mobility can allow them to act as a potential slow-release
4
5 451 nitrogen fertilizer that increases nutrient delivery to the plant's root zone and decreases nutrient
6
7 452 leaching. In trial 3, the 30S soil blend showed the largest increase in leaf yield (24%) while the
8
9
10 453 70S soil blend had the largest reduction in nitrate leached (54% reduction) when comparing the
11
12 454 CNP and non-CNP treatments (Table S20).

13
14 455 The reduction in nitrate leached with CNP addition in this study could be due to the
15
16 456 adsorption properties and surface characteristics of the CNPs, which can have a significant
17
18 457 impact on the nutrient fate. The adsorptive properties of carbon materials for nutrients and
19
20 458 contaminants are well established, hence the reason they are used in filters and in water and
21
22 459 wastewater treatment processes.⁴⁹ Many nutrient leaching studies using biochar and charcoal
23
24 460 found that carbon additions reduced nutrient leaching.⁵⁰⁻⁵³ These conclusions were further
25
26 461 established in a laboratory adsorption study that showed CNPs have an affinity for nitrogen
27
28 462 species and removed ~18% of ammonium and nitrate at the 1,500 mg CNP/L dose (Figure S16).
29
30 463 This translates to ~0.035 mg N/mg CNP for the 300 mg CNP/L dose and ~0.015 mg N/mg CNP
31
32 464 for the 1,500 mg CNP/L dose. This was also seen in a study of CNPs produced from biochar in
33
34 465 which the presence of functional groups led to higher selectivity of ammonium over nitrate ions
35
36 466 due to negative surface charge.⁵⁴ Sonkar et al. (2012) investigated water-soluble carbon nano-
37
38 467 onions and concluded they adsorb anions through hydrogen bonding and electrostatic
39
40 468 interactions and then slowly release them.⁵⁵ Charcoal addition to the soil reduced nitrogen
41
42 469 leaching due to electrostatic adsorption of nutrients and retention of soil water containing
43
44 470 nutrients.⁵² A study on biochar found it reduced the nitrate leached due to the adsorption of
45
46 471 ammonium and other soluble compounds, which in turn prevented mineralization and
47
48 472 nitrification of the ammonium.⁵⁰ This could have been the case for CNPs that then acted as a
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 473 slow-release fertilizer for nitrogen; therefore, decreasing the amount of nitrogen leaching into the
4
5 474 water and instead adsorbed by the lettuce roots, however further investigation is needed.
6

7
8 475 This study showed mixed effects of CNPs on leaching and plant nutrient content due to
9
10 476 germination issues, lower yields, variability between replicates, and properties of the Arizona
11
12 477 soil (pH, calcareous, etc.). The yield, nutrient leaching, and plant tissue composition are
13
14 478 interconnected and showed for some treatments that the smaller the yield, the less nutrient uptake
15
16 479 and more nutrients leached (Figures S13–15). The problem of nitrate leaching can also be
17
18 480 intensified when fertilizers are over-applied, which can be a contributing factor to the lowered
19
20 481 effectiveness in the 100% mineral fertilizer treatments. Additionally, the amount of PAR sunlight
21
22 482 reaching the plants was lower in trial 2, which can affect yields and nutrient uptake (Figure S3–
23
24 483 S4) because there is a direct correlation between nutrient uptake and solar radiation.^{56, 57} Trial 2
25
26 484 data demonstrates this effect; the yields and plant nutrient content in the fertilized treatments
27
28 485 were lower compared with trial 1 (Table S17).
29
30

31 486 **5. Conclusion**

32
33 487 There is a need for new technologies, such as combining CNPs with mineral fertilizer, to
34
35 488 help reduce the impacts that agricultural production has on the environment. Without fertilizer
36
37 489 addition, lower plant yields were observed and clearly demonstrated the necessity to add
38
39 490 nutrient-based fertilizers. We found that compared against fertilizer without CNP added, CNP
40
41 491 fertilizer treatments did not have a negative impact on lettuce yield and decreased nitrate
42
43 492 leaching in some fertilizer treatments. Two strategies were tested to reduce nitrate leaching:
44
45 493 reducing the amount of fertilizer and changing the saturated hydraulic conductivity of the soil.
46
47 494 To maintain a lettuce yield of over 34,000 kg/ha (i.e., similar to the average in the Salinas
48
49 495 Valley³⁸), the amount of mineral fertilizer could be reduced by 30% to result in 25% less nitrate
50
51
52
53
54
55
56
57
58
59
60

1
2
3 496 leaching, or the amount of fertilizer could be reduced by 30% and combined with less than 1wt%
4
5 497 of CNPs to result in 57% less nitrate leaching compared with using 100% of the recommended
6
7 498 fertilizer dose. Treatments using CNPs did not change the nutritional value of the leaf tissue,
8
9
10 499 based upon elemental analysis (Tables S14-S19). Additionally, increasing the soil hydraulic
11
12 500 conductivity improved the effect of CNPs on reducing nitrate leaching and increasing plant
13
14 501 nutrient uptake. Adding CNP to fertilizer blends could allow farmers to add less fertilizer while
15
16
17 502 achieving high production yields and reducing the amount of nitrate reaching surrounding water
18
19 503 bodies.

20
21 504 In addition to understanding the economic and life cycle implications of CNPs as
22
23 505 fertilizer additives, future research should also consider the farm worker, environmental health,
24
25 506 and safety precautions of working with CNPs, as could be leaching of CNPs into ground or
26
27 507 surface waters or uptake into plants. To help explain the observations herein that CNPs have the
28
29 508 potential to reduce nitrogen leaching from agricultural soils, research is needed to better
30
31 509 understand how CNPs influence biogeochemical mechanisms (e.g., plant root microbiota, [CNP-](#)
32
33 510 [nitrogen adsorption](#)) on nitrogen processes in the soil.
34
35
36
37

38 511

39 512 **6.0 Acknowledgements**

40
41
42 513 This work was partially funded from the US Environmental Protection Agency through the
43
44 514 STAR program (RD83558001) and the National Science Foundation (EEC-1449500)
45
46 515 Nanosystems Engineering Research Center on Nanotechnology-Enabled Water Treatment.
47
48 516 Technical assistance from Dr. Yuqiang Bi and technical editing by Laurel Passantino was greatly
49
50 517 appreciated.
51
52
53
54
55
56
57
58
59
60

1
2
3 518 **7.0 References**
4
5

- 6 519 1. Hartz, T. K.; Johnstone, P. R.; Williams, E.; Smith, R. F., Establishing Lettuce Leaf
7
8 520 Nutrient Optimum Ranges Through DRIS Analysis. *HortScience* **2007**, *42*, (1), 143-146.
9
10 521 2. Hartz, T. K.; Bendixen, W. E.; Wierdsma, L., The Value of Presidedress Soil Nitrate
11
12 522 Testing as a Nitrogen Management Tool in Irrigated Vegetable Production. *HortScience* **2000**,
13
14 523 *35*, (4), 651-656.
15
16 524 3. Council, N. R., *Nutrient control actions for improving water quality in the Mississippi*
17
18 525 *River Basin and Northern Gulf of Mexico*. National Academies Press: 2009.
19
20 526 4. Council, N. R., *Improving water quality in the Mississippi river basin and northern Gulf*
21
22 527 *of Mexico: Strategies and priorities*. National Academies Press: 2012.
23
24 528 5. Harter, T.; Lund, J. R., *Addressing Nitrate in California's Drinking Water: With a Focus*
25
26 529 *on Tulare Lake Basin and Salinas Valley Groundwater: Report for the State Water Resources*
27
28 530 *Control Board Report to the Legislature*. 2012.
29
30 531 6. Pennino, M. J.; Compton, J. E.; Leibowitz, S. G., Trends in Drinking Water Nitrate
31
32 532 Violations Across the United States. *Environmental Science & Technology* **2017**, *51*, (22),
33
34 533 13450-13460.
35
36 534 7. Ward, M. H.; Jones, R. R.; Brender, J. D.; de Kok, T. M.; Weyer, P. J.; Nolan, B. T.;
37
38 535 Villanueva, C. M.; van Breda, S. G., Drinking Water Nitrate and Human Health: An Updated
39
40 536 Review. *International Journal of Environmental Research and Public Health* **2018**, *15*, (7), 31.
41
42 537 8. Burow, K. R.; Nolan, B. T.; Rupert, M. G.; Dubrovsky, N. M., Nitrate in Groundwater of
43
44 538 the United States, 1991–2003. *Environmental Science & Technology* **2010**, *44*, (13), 4988-4997.
45
46 539 9. Manuel, J., Nutrient Pollution: A Persistent Threat to Waterways. *Environmental Health*
47
48 540 *Perspectives* **2014**, *122*, (11), A304-A309.
49
50
51
52
53
54
55
56
57
58
59
60

- 1
2
3 541 10. Vymazal, J., Removal of nutrients in various types of constructed wetlands. *Sci. Total*
4
5 542 *Environ.* **2007**, *380*, (1), 48-65.
6
7
8 543 11. Rodrigues, S. M.; Demokritou, P.; Dokoozlian, N.; Hendren, C. O.; Karn, B.; Mauter, M.
9
10 544 S.; Sadik, O. A.; Safarpour, M.; Unrine, J. M.; Viers, J.; Welle, P.; White, J. C.; Wiesner, M. R.;
11
12 545 Lowry, G. V., Nanotechnology for sustainable food production: promising opportunities and
13
14 546 scientific challenges. *Environmental Science: Nano* **2017**, *4*, (4), 767-781.
15
16
17 547 12. White, J. C.; Gardea-Torresdey, J., Achieving food security through the very small. *Nat.*
18
19 548 *Nanotechnol.* **2018**, *13*, (8), 627-629.
20
21
22 549 13. Pandey, K.; Lahiani, M. H.; Hicks, V. K.; Hudson, M. K.; Green, M. J.; Khodakovskaya,
23
24 550 M., Effects of carbon-based nanomaterials on seed germination, biomass accumulation and salt
25
26 551 stress response of bioenergy crops. *PLOS ONE* **2018**, *13*, (8), e0202274.
27
28
29 552 14. Capaldi Arruda, S. C.; Diniz Silva, A. L.; Moretto Galazzi, R.; Antunes Azevedo, R.;
30
31 553 Zezzi Arruda, M. A., Nanoparticles applied to plant science: A review. *Talanta* **2015**, *131*, 693-
32
33 554 705.
34
35
36 555 15. Mukherjee, A.; Majumdar, S.; Servin, A. D.; Pagano, L.; Dhankher, O. P.; White, J. C.,
37
38 556 Carbon Nanomaterials in Agriculture: A Critical Review. *Front. Plant Sci.* **2016**, *7*, (172).
39
40
41 557 16. Rossi, M.; Cubadda, F.; Dini, L.; Terranova, M.; Aureli, F.; Sorbo, A.; Passeri, D.,
42
43 558 Scientific basis of nanotechnology, implications for the food sector and future trends. *Trends in*
44
45 559 *Food Science & Technology* **2014**, *40*, (2), 127-148.
46
47
48 560 17. Roduner, E., Size matters: why nanomaterials are different. *Chemical Society Reviews*
49
50 561 **2006**, *35*, (7), 583-592.
51
52
53
54
55
56
57
58
59
60

- 1
2
3 562 18. Aslani, F.; Bagheri, S.; Muhd Julkapli, N.; Juraimi, A. S.; Hashemi, F. S. G.; Baghdadi,
4
5 563 A., Effects of Engineered Nanomaterials on Plants Growth: An Overview. *The Scientific World*
6
7 564 *Journal* **2014**, 2014, 28.
- 8
9
10 565 19. Begum, P.; Ikhtiari, R.; Fugetsu, B., Graphene phytotoxicity in the seedling stage of
11
12 566 cabbage, tomato, red spinach, and lettuce. *Carbon* **2011**, 49, (12), 3907-3919.
- 13
14 567 20. Cañas, J. E.; Long, M.; Nations, S.; Vadan, R.; Dai, L.; Luo, M.; Ambikapathi, R.; Lee,
15
16 568 E. H.; Olszyk, D., Effects of functionalized and nonfunctionalized single-walled carbon
17
18 569 nanotubes on root elongation of select crop species. *Environmental Toxicology and Chemistry*
19
20 570 **2008**, 27, (9), 1922-1931.
- 21
22 571 21. Lin, C.; Fugetsu, B.; Su, Y.; Watari, F., Studies on toxicity of multi-walled carbon
23
24 572 nanotubes on Arabidopsis T87 suspension cells. *Journal of Hazardous Materials* **2009**, 170, (2),
25
26 573 578-583.
- 27
28 574 22. Jackson, P.; Jacobsen, N. R.; Baun, A.; Birkedal, R.; Kühnel, D.; Jensen, K. A.; Vogel,
29
30 575 U.; Wallin, H., Bioaccumulation and ecotoxicity of carbon nanotubes. *Chemistry Central*
31
32 576 *Journal* **2013**, 7, (1), 154.
- 33
34 577 23. Jin, L.; Son, Y.; Yoon, T. K.; Kang, Y. J.; Kim, W.; Chung, H., High concentrations of
35
36 578 single-walled carbon nanotubes lower soil enzyme activity and microbial biomass.
37
38 579 *Ecotoxicology and environmental safety* **2013**, 88, 9-15.
- 39
40 580 24. Khodakovskaya, M. V.; de Silva, K.; Biris, A. S.; Dervishi, E.; Villagarcia, H., Carbon
41
42 581 nanotubes induce growth enhancement of tobacco cells. *ACS nano* **2012**, 6, (3), 2128-2135.
- 43
44 582 25. Sawant, D., Effect of carbon nanotubes on seed germination, growth and yield of hybrid
45
46 583 Bt cotton Var. 7383 BG II. **2016**.
- 47
48
49
50
51
52
53
54
55
56
57
58
59
60

- 1
2
3 584 26. Miralles, P.; Johnson, E.; Church, T. L.; Harris, A. T., Multiwalled carbon nanotubes in
4
5 585 alfalfa and wheat: toxicology and uptake. *Journal of The Royal Society Interface* **2012**, *9*, (77),
6
7 586 3514-3527.
- 8
9
10 587 27. De La Torre-Roche, R.; Hawthorne, J.; Deng, Y.; Xing, B.; Cai, W.; Newman, L. A.;
11
12 588 Wang, C.; Ma, X.; White, J. C., Fullerene-enhanced accumulation of p, p'-DDE in agricultural
13
14 589 crop species. *Environmental science & technology* **2012**, *46*, (17), 9315-9323.
- 15
16
17 590 28. Tiwari, D. K.; Dasgupta-Schubert, N.; Villaseñor Cendejas, L. M.; Villegas, J.; Carreto
18
19 591 Montoya, L.; Borjas García, S. E., Interfacing carbon nanotubes (CNT) with plants: enhancement
20
21 592 of growth, water and ionic nutrient uptake in maize (*Zea mays*) and implications for
22
23 593 nanoagriculture. *Applied Nanoscience* **2014**, *4*, (5), 577-591.
- 24
25
26 594 29. Liu, Q.; Zhao, Y.; Wan, Y.; Zheng, J.; Zhang, X.; Wang, C.; Fang, X.; Lin, J., Study of
27
28 595 the Inhibitory Effect of Water-Soluble Fullerenes on Plant Growth at the Cellular Level. *ACS*
29
30 596 *Nano* **2010**, *4*, (10), 5743-5748.
- 31
32
33 597 30. Seginer, I.; Shina, G.; Albright, L. D.; Marsh, L. S., Optimal temperature setpoints for
34
35 598 greenhouse lettuce. *Journal of Agricultural Engineering Research* **1991**, *49*, 209-226.
- 36
37
38 599 31. Liu, J. Preparation of nano graphite powder from nano-graphite sol. 2003.
- 39
40 600 32. Solanki, A.; Boyer, T. H., Pharmaceutical removal in synthetic human urine using
41
42 601 biochar. *Environmental Science: Water Research & Technology* **2017**, *3*, (3), 553-565.
- 43
44 602 33. UMS *Operation Manual KSAT*; Germany, 2013.
- 45
46
47 603 34. Smith, R.; Cahn, M.; Daugovish, O.; Koike, S., *Leaf lettuce production in California*.
48
49 604 UCANR Publications: 2011.
- 50
51 605 35. Waters Agricultural Laboratories, I. <https://watersag.com/>

- 1
2
3 606 36. Reich, S.; Thomsen, C., Raman spectroscopy of graphite. *Philosophical Transactions of*
4
5 607 *the Royal Society of London. Series A: Mathematical, Physical and*
6
7 608 *Engineering Sciences* **2004**, *362*, (1824), 2271-2288.
- 9
10 609 37. Song, G.; Pandorf, M.; Westerhoff, P.; Ma, Y., Carbon Nanomaterial-Based Fertilizers
11
12 610 Can Improve Plant Growth. In *Nanotechnology Applications in the Food Industry*, Rai, V. R.;
13
14 611 Bai, J. A., Eds. Taylor & Francis: 2018; pp 22-44.
- 16
17 612 38. Geisseler, D.; Horwath, W. R., Lettuce production in California. *Fertilizer Research and*
18
19 613 *Education Program*. http://apps.cdffa.ca.gov/frep/docs/Lettuce_Production_CA.pdf. Accessed
20
21 614 *May 2014*.
- 23
24 615 39. Letey, J., Relationship between Soil Physical Properties and Crop Production. In
25
26 616 *Advances in Soil Science*, Stewart, B. A., Ed. Springer New York: New York, NY, 1985; pp 277-
27
28 617 294.
- 30
31 618 40. Kole, C.; Kole, P.; Randunu, K. M.; Choudhary, P.; Podila, R.; Ke, P. C.; Rao, A. M.;
32
33 619 Marcus, R. K., Nanobiotechnology can boost crop production and quality: first evidence from
34
35 620 increased plant biomass, fruit yield and phytomedicine content in bitter melon (*Momordica*
36
37 621 *charantia*). *BMC biotechnology* **2013**, *13*, (1), 1.
- 39
40 622 41. Khodakovskaya, M. V.; Kim, B. S.; Kim, J. N.; Alimohammadi, M.; Dervishi, E.;
41
42 623 Mustafa, T.; Cernigla, C. E., Carbon nanotubes as plant growth regulators: effects on tomato
43
44 624 growth, reproductive system, and soil microbial community. *Small* **2013**, *9*, (1), 115-123.
- 46
47 625 42. Lahiani, M. H.; Chen, J.; Irin, F.; Poretzky, A. A.; Green, M. J.; Khodakovskaya, M. V.,
48
49 626 Interaction of carbon nanohorns with plants: uptake and biological effects. *Carbon* **2015**, *81*,
50
51 627 607-619.

- 1
2
3 628 43. Villagarcia, H.; Dervishi, E.; Silva, K. d.; Biris, A. S.; Khodakovskaya, M. V., Surface
4
5 629 Chemistry of Carbon Nanotubes Impacts the Growth and Expression of Water Channel Protein
6
7 630 in Tomato Plants. *Small* **2012**, *8*, (15), 2328-2334.
8
9
10 631 44. Hopkins, B.; Ellsworth, J. In *Phosphorus availability with alkaline/calcareous soil*,
11
12 632 Western Nutrient Management Conference, 2005; 2005; pp 88-93.
13
14 633 45. Thomas, G., Soil pH and soil acidity. *Methods of Soil Analysis Part 3—Chemical*
15
16 634 *Methods* **1996**, (methodsofsoilan3), 475-490.
17
18
19 635 46. Ma, C.; White, J. C.; Zhao, J.; Zhao, Q.; Xing, B., Uptake of Engineered Nanoparticles
20
21 636 by Food Crops: Characterization, Mechanisms, and Implications. *Annual Review of Food*
22
23 637 *Science and Technology* **2018**, *9*, (1), 129-153.
24
25
26 638 47. Rodrigues, S. M.; Trindade, T.; Duarte, A. C.; Pereira, E.; Koopmans, G. F.; Römkens, P.
27
28 639 F. A. M., A framework to measure the availability of engineered nanoparticles in soils: Trends in
29
30 640 soil tests and analytical tools. *TrAC Trends in Analytical Chemistry* **2016**, *75*, 129-140.
31
32
33 641 48. Cornelis, G.; Hund-Rinke, K.; Kuhlbusch, T.; Van den Brink, N.; Nickel, C., Fate and
34
35 642 Bioavailability of Engineered Nanoparticles in Soils: A Review. *Crit. Rev. Environ. Sci. Technol.*
36
37 643 **2014**, *44*, (24), 2720-2764.
38
39
40 644 49. Przepiórski, J., Chapter 9 Activated carbon filters and their industrial applications. In
41
42 645 *Interface Science and Technology*, Bandosz, T. J., Ed. Elsevier: 2006; Vol. 7, pp 421-474.
43
44 646 50. Laird, D.; Fleming, P.; Wang, B.; Horton, R.; Karlen, D., Biochar impact on nutrient
45
46 647 leaching from a Midwestern agricultural soil. *Geoderma* **2010**, *158*, (3), 436-442.
47
48
49 648 51. Yao, Y.; Gao, B.; Zhang, M.; Inyang, M.; Zimmerman, A. R., Effect of biochar
50
51 649 amendment on sorption and leaching of nitrate, ammonium, and phosphate in a sandy soil.
52
53 650 *Chemosphere* **2012**, *89*, (11), 1467-1471.
54
55
56
57
58
59
60

- 1
2
3 651 52. Lehmann, J.; Pereira da Silva, J.; Steiner, C.; Nehls, T.; Zech, W.; Glaser, B., Nutrient
4
5 652 availability and leaching in an archaeological Anthrosol and a Ferralsol of the Central Amazon
6
7 653 basin: fertilizer, manure and charcoal amendments. *Plant and Soil* **2003**, *249*, (2), 343-357.
8
9
10 654 53. Major, J.; Rondon, M.; Molina, D.; Riha, S. J.; Lehmann, J., Maize yield and nutrition
11
12 655 during 4 years after biochar application to a Colombian savanna oxisol. *Plant and Soil* **2010**,
13
14 656 333, (1), 117-128.
15
16
17 657 54. Saxena, M.; Maity, S.; Sarkar, S., Carbon nanoparticles in 'biochar' boost wheat
18
19 658 (*Triticum aestivum*) plant growth. *RSC Advances* **2014**, *4*, (75), 39948-39954.
20
21
22 659 55. Sonkar, S. K.; Roy, M.; Babar, D. G.; Sarkar, S., Water soluble carbon nano-onions from
23
24 660 wood wool as growth promoters for gram plants. *Nanoscale* **2012**, *4*, (24), 7670-7675.
25
26
27 661 56. Wurr, D. C. E.; Fellows, J. R., The influence of solar radiation and temperature on the
28
29 662 head weight of crisp lettuce. *Journal of Horticultural Science* **1991**, *66*, (2), 183-190.
30
31 663 57. Campillo, C.; Fortes, R.; del Henar Prieto, M., Solar radiation effect on crop production.
32
33 664 In *Solar Radiation*, InTech: 2012.
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3
4
5
6

Versus



NO_3^- NO_3^- NO_3^-
 NO_3^- NO_3^-

NPK +
Carbon Nanoparticles



NO_3^- NO_3^- NO_3^- NO_3^-
 NO_3^- NO_3^- NO_3^-

NPK