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**Removal and Growth of Microorganisms across Treatment
and Simulated Distribution at a Pilot-Scale Direct Potable
Reuse Facility**

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1 Title: Removal and Growth of Microorganisms across Treatment and Simulated Distribution at a
2 Pilot-Scale Direct Potable Reuse Facility

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14

15 **Abstract**

16 Multi-barrier advanced treatment trains have been shown to purify wastewater to drinking water
17 standards, but improved methods are needed to better understand microbial concentrations,
18 viability, and growth potential throughout treatment and distribution systems. In this study, bulk
19 water cell counts, adenosine triphosphate concentrations, and assimilable organic carbon were
20 measured throughout a pilot-scale direct potable reuse facility and three parallel chlorinated
21 simulated distribution systems fed with the pilot's finished water. We also investigated the impacts
22 of treatment operations (e.g., membrane cleanings) and perturbations (e.g., incomplete wastewater
23 nitrification) on microbial water quality. Intact cell counts and total adenosine triphosphate

24 concentrations were reduced to near or below method quantification limits (22 cells/mL and 10^{-4}
25 nM, respectively) by reverse osmosis and advanced oxidation. Total cell counts and assimilable
26 organic carbon concentrations were consistently above method quantification limits (12 cells/mL
27 and 10 $\mu\text{g/L}$, respectively) in finished water (i.e., granular activated carbon filtrate). However,
28 assimilable organic carbon levels in finished water were lower than values typically measured in
29 conventional drinking water systems in the United States, and cell counts were dramatically
30 reduced by batch chlorination. Occasional ammonia passage through advanced treatment caused
31 large variations in chlorine demand in finished water, and a strong inverse correlation was
32 observed between cell counts and residual free chlorine in the simulated distribution systems. We
33 show that multi-barrier advanced treatment reliably rejects wastewater-derived bacterial cells, and
34 that the methods applied herein can provide complementary information to traditional monitoring
35 approaches to evaluate microbial water quality throughout treatment and distribution.

36

37 **Water Impact Statement:** Purification of wastewater can provide water-stressed regions a
38 reliable drinking water supply. We evaluated and provided recommendations for the use of
39 enhanced analytical tools to assess microbial water quality in an advanced treatment train and
40 simulated distribution systems, with the aim to further improve the microbial safety of potable
41 reuse.

42

43 **Key words:** advanced water treatment, log removal values, flow cytometry, ATP, assimilable
44 organic carbon, chlorine residual

45

46 **1. Introduction**

47 Increasing pressures from population growth, urbanization, and climate change are challenging
48 many cities' water security. For arid cities in the southwestern United States (e.g., El Paso, Texas),
49 aggravated droughts in recent years have intermittently reduced freshwater reservoir levels to less
50 than 10% of maximum capacity [1]. There is growing recognition that direct potable reuse, i.e., the
51 deliberate introduction of purified wastewater into a drinking water treatment facility or
52 distribution system, can produce high-quality water to diversify municipal water portfolios, reduce
53 imports, and meet future demand. Direct potable reuse requires extensive advanced treatment of
54 wastewater, which most often involves combinations of ozonation, granular activated carbon
55 filtration (GAC), microfiltration (MF), reverse osmosis (RO) or nanofiltration (NF), and a H₂O₂-
56 UV advanced oxidation process (AOP) [2].

57 Finished water from advanced treatment trains has been shown to meet water quality
58 regulations for potable reuse [3-5], but knowledge gaps remain regarding how individual treatment
59 processes alter microbial water quality. In particular, it is unclear how final treatment steps (e.g.,
60 chemical conditioning or GAC filtration) influence microbial abundance and regrowth of finished
61 and distributed waters. Furthermore, advanced treatment facilities almost always monitor total
62 organic carbon, but rarely measure assimilable organic carbon (AOC), which is the fraction of
63 labile dissolved carbon that can be most rapidly utilized by heterotrophic microorganisms [6]. AOC
64 has been linked to microbial growth in drinking water [7] and non-potable reuse systems [8], but
65 the role of residual AOC in potable reuse systems has not been specifically assessed. Measurement
66 of AOC and the numbers and viability of microorganisms across different potable reuse treatment
67 trains could provide insight into the performance and ideal sequencing of treatment processes to
68 manage microbial numbers and growth potential.

69 Recent sequencing-based studies of potable reuse treatment trains by Stamps et al. (2018)
70 and Kantor et al. (2019) reported that advanced treatment substantially reduces the proportion of
71 the treated water microbial taxa that is shared with wastewater, but there have been no studies to
72 date investigating microbial abundance and viability in continuous-flow drinking water
73 distribution systems fed with advanced treated wastewater [9,10]. The results herein complement
74 the findings of Kantor et al. (2019), which are from the same pilot system. Previous studies of
75 distribution systems fed with conventionally-sourced waters treated by NF or RO report substantial
76 microbial growth despite oligotrophic conditions in the treated water [11-13]. A recent lab-scale
77 study that simulated non-continuous, build-scale plumbing in the context of direct potable reuse
78 reported that conventional drinking water and potable reuse scenarios yielded equivalent microbial
79 regrowth, but the proportion of organisms that were viable was not quantified [14]. There remains
80 a need to evaluate microbial growth and viability in distribution systems under continuous-flow
81 conditions and under more potable reuse blend ratios. There is also a research gap regarding the
82 effectiveness of different chlorine residual types (i.e., free vs. combined chlorine) and
83 concentrations at suppressing microbial numbers and activity in potable reuse distribution systems.
84 In general, lower growth and microbial numbers in distributed water may be expected in potable
85 reuse scenarios due to lower concentrations of AOC and suspended solids.

86 Several emerging methods for evaluating microbial water quality, including fluorescence-
87 based flow cytometric assays for enumerating total and intact microbial cells, a luminescence-
88 based assay for measuring adenosine tri-phosphate (ATP), and a flow cytometry-based bioassay
89 for quantifying AOC, have successfully been used for evaluating microbial growth and viability
90 in conventional drinking water treatment and distribution systems [15-19]. Previous studies have
91 shown cell counts determined by flow cytometry to be more reproducible and informative, and to

92 correlate more strongly with ATP, than traditional microbial quantification methods such as
93 heterotrophic plate counts [17,20-22]. Moreover, as cultivation-independent methods, flow
94 cytometry and intracellular ATP quantify nearly all organisms present in a sample and are more
95 applicable to the low-biomass effluents of treatment processes like MF, RO, AOP, and
96 chlorination. Lastly, the flow cytometry-based bioassay for measuring AOC [23] has been applied
97 by researchers to drinking water treatment trains [24-27], but to our knowledge has not yet been
98 applied across an entire advanced treatment facility.

99 In this study we aimed to determine the effects of advanced water treatment processes on
100 microbial numbers, viability, and growth potential in the context of direct potable reuse.
101 Specifically, we quantified bulk water cell counts, ATP, and AOC across a pilot-scale advanced
102 treatment facility and three bench-scale simulated drinking water distribution systems. The impact
103 of operations (e.g., membrane cleanings) and treatment perturbations (e.g., incomplete nitrification
104 at the wastewater facility) on microbial water quality were also investigated. Finally, we consider
105 the utility and feasibility of deploying the assays used herein to full-scale advanced treatment
106 facilities. This work is unique in applying enhanced analytical methods to understand advanced
107 treatment processes that will be integral components of many future potable reuse treatment trains.
108 We also discuss implications for the management of microbial water quality during distribution,
109 which has been a crucial and understudied aspect of potable reuse systems.

110

111 **2. Study Site, Methods, and Materials**

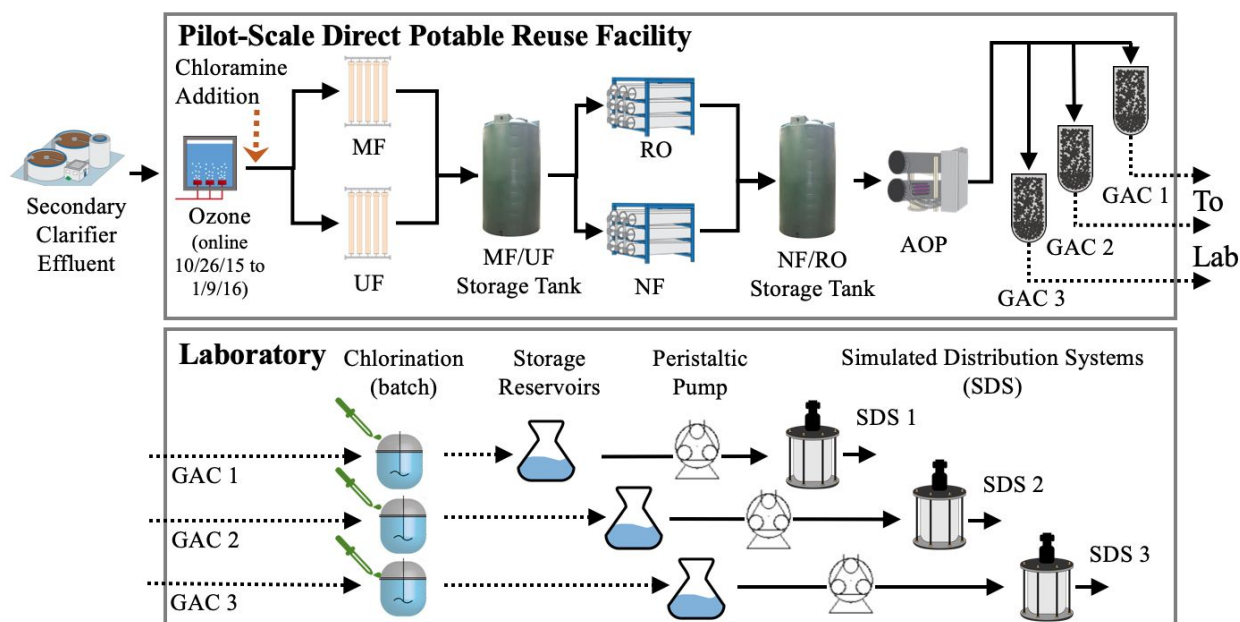
112 **2.1 Layout of the pilot-scale direct potable reuse facility and bulk water sampling**

113 We studied a pilot-scale advanced treatment facility in El Paso, Texas that was operated from June
114 8, 2015 to January 29, 2016 with a treatment capacity of 5.3×10^5 L/day. To determine optimal

115 full-scale treatment train design, several treatment steps were evaluated with parallel treatment
116 processes that were fed equal flows. The advanced treatment train is illustrated in **Figure 1**, and
117 major treatment processes are described in detail in the **Supplementary Methods**. Briefly, raw
118 wastewater was treated by a conventional activated sludge facility designed to fully nitrify influent
119 ammonia; however, the facility often failed to fully nitrify ammonia during high loading periods.
120 Unchlorinated secondary wastewater effluent was then treated sequentially by ozonation (target
121 concentration of 3.5 mg/L), chloramination to limit membrane biofouling (target concentration 2
122 – 4 mg/L as Cl₂), parallel microfiltration and ultrafiltration (MF and UF; 0.1 and 0.04 µm nominal
123 pore sizes, respectively), parallel reverse osmosis (RO) or nanofiltration (NF), a H₂O₂-UV
124 advanced oxidation process (AOP; UV dose of 840 mJ/cm² and H₂O₂ dose of 4 mg/L), and granular
125 activated carbon filtration (GAC). We also collected, transported, and chlorinated GAC filtrate
126 that was then fed to three bench-scale, parallel storage reservoirs (“Reservoir”) and parallel
127 simulated distribution systems (SDS), as discussed further below. We collected grab bulk water
128 samples from the pilot facility’s secondary wastewater feed, after every major treatment process
129 (including after each individual process that were in parallel; e.g., both MF and UF filtrates), and
130 from the bench-scale reservoirs and SDS from October 2015 through January 2016. Data for
131 parallel processes are often aggregated and presented herein as “MF/UF”, “NF/RO”, “GAC”,
132 “Reservoir”, or “SDS”. It should be noted that samples from the bench-scale reservoirs and SDS
133 were taken ~48 hours after collection and chlorination of GAC filtrate (see Section 2.3).

134 For comparison with results from the SDS, bulk water samples from El Paso’s full-scale
135 drinking water distribution system were collected at 12 sampling locations on January 26, 2016.
136 These 12 sampling locations were selected to cover a range of chlorine concentrations. El Paso
137 targets a free chlorine residual of 1.8 mg/L as Cl₂ entering the full-scale distribution system and a

138 minimum free chlorine residual of 0.5 mg/L as Cl₂ at all sampling locations throughout the city
 139 [28]. These 12 sampling locations were serviced by two drinking water treatment facilities that
 140 treated surface water. Each facility employed coagulation, flocculation, sedimentation, filtration
 141 (granular activated carbon), and disinfection by either free chlorine or chlorine dioxide. Depending
 142 on season, water ages in the full-scale distribution system range from <24 hours to >240 hours,
 143 with a typical water age of ~48 hours.



144
 145 **Figure 1: Design of the pilot-scale facility and experimental setup of the simulated chlorination and distribution**
 146 **system.** Secondary-treated wastewater was fed to a pilot-scale advanced water treatment train that treated water
 147 sequentially using ozonation, parallel microfiltration (MF) and ultrafiltration (UF), parallel nanofiltration (NF) and
 148 reverse osmosis (RO), a H₂O₂-UV advanced oxidation process (AOP), and three parallel granular activated carbon
 149 (GAC) filters. GAC filtrate was collected, batch chlorinated, stored in reservoirs, and fed into annular reactors that
 150 acted as simulated distribution systems (SDS) [9]. Differences in arrow lengths for the laboratory system are purely
 151 for visual effect and do not indicate a difference in tubing length or storage times.

152
 153 Collection of operational data (e.g., MF/UF transmembrane pressure, membrane cleaning
 154 and backwash times) and sampling for physical and chemical water quality parameters (including,
 155 but not limited to, pH, temperature, total organic carbon, nitrite, total iron and manganese,
 156 chloramine, and total ammonia-N {“TAN” = NH₄⁺ + NH₃}) at various stages throughout treatment
 157 were conducted by Arcadis and student researchers from the University of El Paso, Texas. Samples

158 were dated according to the number of days from pilot start-up (June 8, 2015). We flushed pilot
159 facility sample taps for the following durations prior to collection of bulk water samples: secondary
160 wastewater feed, ozone, and chloramination (>5 min), MF/UF filtrates and storage tank (>15 min),
161 NF/RO permeates and storage tank, and AOP effluent (>30 min), and GAC filtrate (>15 min).

162

163 **2.2 Annular reactors as simulated distribution systems (SDS): setup and operation**

164 Three biofilm rotating annular reactors (Model 1320 LS; Biosurface Technologies Corporation,
165 Bozeman, MT) were operated as parallel simulated distribution systems (SDS) beginning on
166 October 6, 2015 at the Robertson-Umbenhauer Water Treatment Plant in El Paso, Texas. Annular
167 reactors have been used in many previous studies to simulate drinking water distribution systems
168 [29-32]. Prior to operation, the SDS were sterilized and prepared as described previously [30].
169 Briefly, storage reservoirs (borosilicate glass), SDS, and associated tubing were sterilized by
170 autoclave, and then 20 ethanol-cleaned polycarbonate biofilm coupons were inserted into each
171 annular reactor. After full assembly, the storage reservoirs and SDS were rinsed by pumping sterile
172 ultrapure water through the systems for ~24 h at the intended operational flow rate. Storage
173 reservoirs and SDS were operated in the dark (i.e., wrapped up tinfoil) at a hydraulic residence
174 time of ~18 hours (flow rate ~0.92 mL/min) and an ambient temperature of 21.5 ± 1.1 °C. A typical
175 water age in El Paso's full-scale distribution system is ~48 hours (data not shown). However, a
176 lower hydraulic retention time (18 hours) was chosen for the SDS to increase the volume of water
177 that could be collected from the SDS for sequencing analyses, results of which are presented in
178 Kantor et al. (2019) [33]. The inner cylinder rotation speed was set to 50 rpm to create an estimated
179 shear stress of ~0.25 N/m² on the inner cylinder surface. This shear stress is reported as common
180 in drinking water pipes in a review of simulation distribution systems by Gomes et al. (2014) [29].

181

182 2.3 Simulated distribution systems (SDS): collection and chlorination of feed water

183 Each SDS was fed chlorinated filtrate from a different pilot facility GAC filter. The three GAC
184 filters were acclimated for approximately five months prior to SDS setup. GAC filtrate was
185 collected approximately every two days in 2-L borosilicate glass bottles. Initially, the bottles were
186 cleaned to remove AOC (Hammes and Egli, 2005), then autoclaved and stored in the dark between
187 uses. GAC filtrate was chlorinated to breakpoint at bench-scale with sodium hypochlorite (target
188 CT value of ~30 mg-min/L) and stirred to ensure full and rapid mixing of chlorine. The storage
189 reservoirs were drained of all water and then immediately filled with freshly chlorinated GAC
190 filtrate.

191 The SDS were acclimated for 69 days (pilot operation days 120 – 189) with an average
192 initial feed free chlorine concentration of 0.89 ± 0.07 mg/L as Cl_2 . After the acclimation period,
193 the impact of increasing free chlorine concentrations on cell counts and ATP concentrations in the
194 SDS was evaluated by targeting increasingly greater initial free chlorine residuals (in mg/L as Cl_2)
195 in the reservoirs for the following periods of time: 0.8 mg/L (days 189 – 205); 0.9 mg/L (days 206
196 – 214); 1.0 mg/L (days 215 – 228), 1.6 mg/L (days 229 – 234). Free and total chlorine
197 measurements were taken using a DR/850 Portable Colorimeter (Hach USA, Loveland, CO) using
198 DPD 10 mL Free Chlorine (#97009-454; VWR International) and DPD 10 mL Total Chlorine
199 (#97009-464; VWR International, Radnor, PA) Reagent Powder Pillows, respectively.

200 Chlorine demand was calculated by subtracting the initial free chlorine residual
201 concentration (i.e., after achieving a CT value of 30 mg-min/L) measured in the reservoirs from
202 the calculated concentration of chlorine added during chlorination. Ammonia concentration was
203 measured using an online analyzer (ChemScan[®] UV-6101, ASA Analytics; Waukesha WI) located

204 after NF/RO. Ammonia in the GAC filtrates was estimated to be the average of five online analyzer
205 measurements (each taken 5 minutes apart), accounting for GAC empty bed contact times, as well
206 as the hydraulic retention times of the NF/RO storage tank and AOP, but assuming negligible
207 losses and transformations of ammonia through storage, AOP [34,35], and GAC. Chlorine demand
208 from ammonia was calculated using a weight ratio of 8:1 (Cl:N) previously reported to achieve
209 complete oxidation of ammonia to nitrogen gas by chlorine [36].

210

211 **2.4 Cell counts by fluorescent staining and flow cytometry**

212 Total and intact cell counts were measured in triplicate by flow cytometry using slight
213 modifications from a previously optimized protocol presented in Prest et al., 2013 [37]. This
214 modification included pre-heating greater sample volumes (1 mL or 1.5 mL, instead of 500 uL, to
215 account for greater analytical run volumes for low cell count matrices) at 35°C for a longer duration
216 (10 min, instead of 5 min, to account for greater sample volumes). The protocol thereafter followed
217 Prest et al., 2013 [37]. Pre-heated samples were then stained with 10 µL of working dye solution
218 per mL of sample (see below), and then incubated in the dark at 35°C (10 minutes) before
219 measurement by flow cytometry. Working dye solutions for total cell counts were created by
220 diluting SYBR[®] Green I (10,000x in DMSO, S9430; Sigma-Aldrich, St. Louis, MO) 100x in
221 buffer (10 mM TRIS in 0.1-µm filtered nanopure water); working dye solutions for intact cell
222 counts followed that of total cell counts, but included propidium iodide (30 mM. P1304MP; Life
223 Technologies, Carlsbad, CA) at a working dye concentration of 0.6 mM. Prior to preheating,
224 secondary wastewater, ozonation, and chloramine samples were diluted 10- or 100-fold into
225 syringe-filtered (0.1 µm, Millex-VV[®] Syringe Filter Unit; Millipore, Billerica, MA) bottled Evian
226 mineral water to achieve final cell counts less than 2×10^5 cells/mL.

227 Sample measurements were performed in technical triplicate (i.e., three measurements
228 taken from the same sample) on an Accuri™ C6 flow cytometer (BD Biosciences, San Jose, CA)
229 equipped with a 50-mW laser emitting a fixed wavelength of 488 nm and volumetric counting
230 software. The cytometer was calibrated (BD Accuri™ Spherotech 6-Peak and 8-Peak Validation
231 Beads; Cat. #653145 and #653144, respectively) according to manufacturer protocol to measure
232 the number of fluorescent particles in a user-defined fluid volume. Bead calibration was performed
233 twice; once prior to any field sampling, and once two months into the sampling campaign. Event
234 counts were triggered on the green fluorescence channel (i.e., FL1) with a threshold of 800 (for
235 FL1-H) [38]. Measurements were performed at a flow rate setting of 66 $\mu\text{L}/\text{min}$. Lower cell count
236 samples (e.g., NF/RO permeate and AOP) were measured with high run volumes (i.e., 1,000 μL)
237 to increase the number of quantified cells. Microbial cell signals were distinguished from
238 background and instrument noise and enumerated on density plots of green (FL1; 533 ± 30 nm)
239 and red (FL3; >670 nm) fluorescence using an electronic gate template previously optimized for
240 water samples [38]. Gate positions for MF, UF, NF, RO, and AOP were modified to reduce the
241 impact of background noise on cell counts (see **Supplementary Figure S1**). The precision of
242 primary measurements is shown in **Supplemental Table S2**.

243 Calculation of \log_{10} reduction values for cell counts across individual advanced treatment
244 processes were based on process feed and effluent data from each individual sampling day.
245 However, \log_{10} reduction value calculations for the bench-scale reservoir used average data from
246 the GAC filtrates, and calculations for the SDS used average data from the reservoirs. For
247 statistical analyses and \log_{10} removal calculations, all values below the quantification limit were
248 set at the quantification limits (12 and 22 cells/mL for total and intact cell counts, respectively) to
249 reflect the conservative philosophy of drinking water management. Results for cell counts and

250 ATP were lognormally distributed (all Shapiro-Wilk tests yield $p < 0.05$), and therefore results are
251 presented as geometric means and geometric standard deviations (“geometric SD”).

252

253 **2.5 Adenosine tri-phosphate concentrations**

254 Total and intracellular ATP concentrations were measured in technical triplicate via a previously
255 optimized protocol [17] using BacTiter-Glo™ Microbial Cell Viability Assay reagents (G8231,
256 Promega Corporation, Madison, WI) with a GloMax^R 20/20 Luminometer (Turner BioSystems,
257 Sunnyvale, CA). BacTiter-Glo™ reagent was prepared according to manufacturer
258 recommendations, with a >2 hour incubation period of reconstituted reagent at room temperature
259 to diminish background noise. The prepared reagent was then stored at -20 °C and used within 24
260 hours. Sample analysis, in brief, included simultaneously incubating 500 µL of water sample and
261 50 µL of prepared reagent in separate tubes at 38 °C for 3 min. Then, 50 µL of the reagent was
262 transferred to the water sample, followed by further incubation at 38 °C for 20 s and then immediate
263 measurement on the luminometer. Luminescence was measured as an integral over 10 s, expressed
264 as relative light units, that were converted to ATP concentrations using an experimentally
265 determined calibration curve (**Supplementary Figure S2**) prepared with a pure ATP standard
266 (P1132; Promega Corporation, Madison, WI). Extracellular ATP was differentiated from total
267 ATP by syringe filtration (0.1 µm, Millex-VV[®] Syringe Filter Unit; Millipore, Billerica, MA) to
268 remove microbial cells prior to ATP analysis. Intracellular ATP was calculated by subtracting
269 extracellular ATP from total ATP. The accuracy of primary measurements is shown in
270 **Supplemental Table S2**. Samples did not require dilution to remain within the linear range of
271 quantification. Calculation of \log_{10} removal values for ATP followed that of log removal
272 calculations for cell counts. For statistical analyses and \log_{10} removal calculations, all values below

273 the quantification limit were set at the quantification limit (1×10^{-4} nM and 1.82×10^{-5} nM for
274 total and intracellular ATP, respectively) to reflect the conservative philosophy of drinking water
275 management.

276

277 **2.6 Assimilable organic carbon concentrations**

278 AOC-free glassware, filters, and materials were prepared as described previously (Hammes and
279 Egli, 2005). AOC concentrations were determined via batch growth assays in biological triplicate
280 as described previously [25]. Briefly, 50 mL of syringe-filtered (0.22 μ m, Millex-GP® Syringe
281 Filter Unit, PES; Millipore, Billerica, MA) water samples were inoculated with 1 mL of unfiltered
282 bottled Evian mineral water to yield initial total cell counts of approximately 1×10^4 cells/mL.
283 Previous studies have measured very low concentrations of AOC (i.e., <10 μ g/L) in Evian water
284 [39], and use of an Evian microbial inoculum has been shown to obtain similar AOC values in
285 water as using other environmentally-sourced inocula [23]. All samples were supplemented with
286 phosphate, nitrogen, iron, and trace element solution buffers (**Supplementary Table S3**) to create
287 a carbon-limiting environment; nutrient and mineral solution recipes were provided by the
288 Drinking Water Microbiology Group at Eawag, Switzerland [40]. Suspensions were then split into
289 triplicate 20-mL vials via direct pour, and the vials were capped, vortexed, and incubated for 3-5
290 days at 30°C with continuous gentle shaking. Cell growth was measured by quantifying total cell
291 counts by flow cytometry in technical triplicate at the end of the 3-5 batch period (initial cell counts
292 were assumed to be negligible). AOC concentrations were determined using a conversion factor
293 (9.63×10^6 cells per μ g of AOC) derived experimentally from growth of Evian microbial inoculum
294 on known acetate concentrations (**Supplementary Figure S3**). All assays were performed with a
295 negative control (a newly purchased bottle of Evian water) that was processed identically to

296 samples. AOC concentrations measured in the negative control were subtracted out from sample
297 results. The accuracy of primary measurements is shown in **Supplemental Table S2**.

298

299 **3. Results and Discussion**

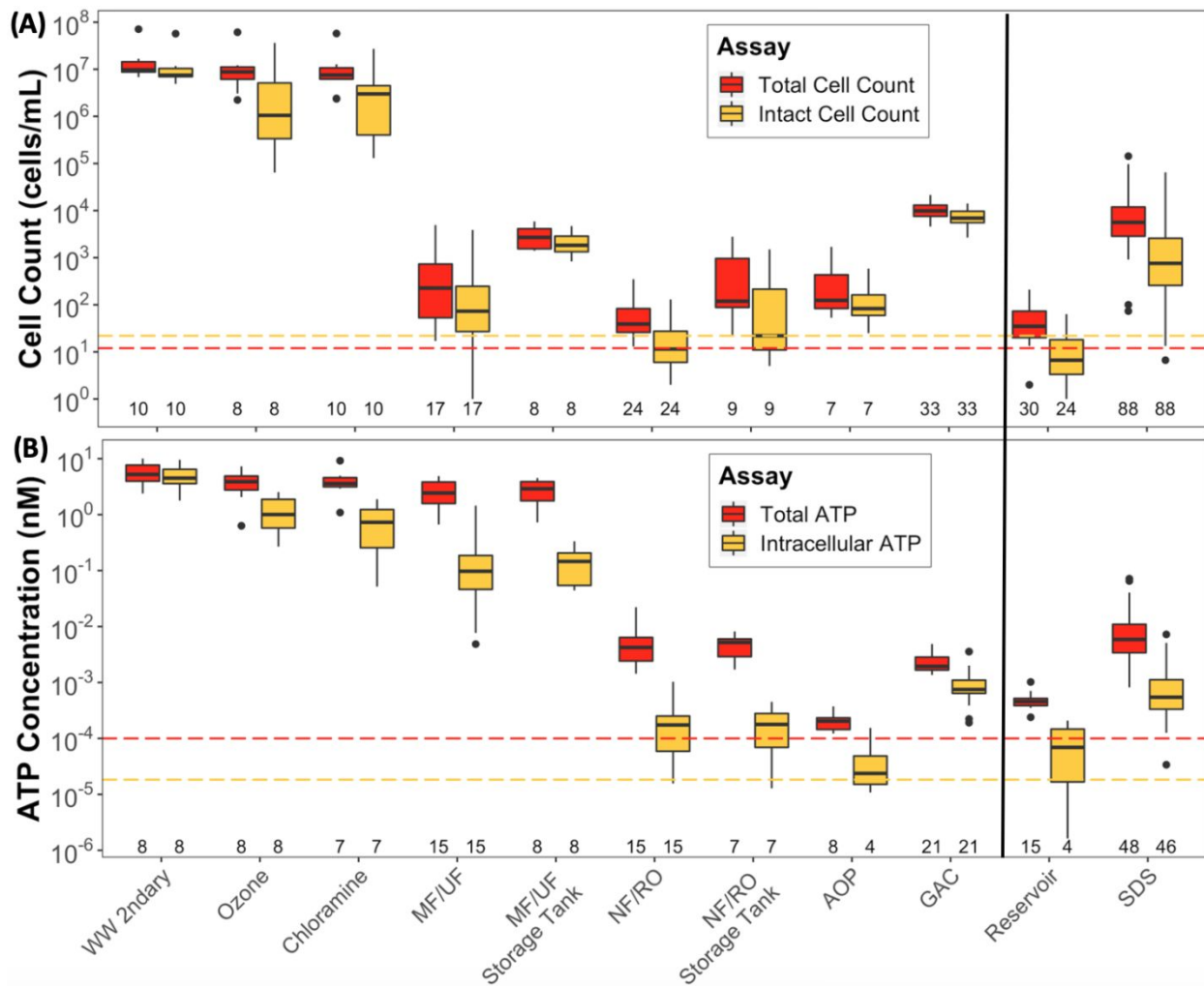
300 **3.1 Cell counts throughout the pilot treatment train**

301 Flow cytometry was a sensitive, rapid, and reproducible method that allowed quantification of
302 planktonic cell concentrations over nearly seven orders of magnitude. All measurements of total
303 and intact cell counts taken throughout the pilot treatment train and three bench-scale SDS are
304 presented in **Figure 2A**, and calculated \log_{10} removal values are presented in **Figure 3A**. Summary
305 statistics and log removal data for cell counts across each treatment process can be found in **Tables**
306 **S5** and **S7**, respectively. The geometric mean total and intact cell counts in the secondary
307 wastewater feed to the pilot were 1.26×10^7 cells/mL (geometric SD = 1.95) and 6.31×10^6
308 cells/mL (geometric SD = 1.97), respectively. Ozonation reduced total and intact cell counts by an
309 average of 0.20 \log_{10} and 0.91 \log_{10} , changes that were not significant ($p = 0.79$ and 0.38,
310 respectively). These relatively low removal values by ozone were unsurprising, given the relatively
311 low initial target ozone concentration of 3.5 mg/L and presumably high demand exerted by
312 organics in the secondary wastewater feed (total organic carbon = 9.3 ± 1.7 mg/L). It should be
313 noted that the primary purpose of ozonation in the El Paso pilot was not to inactivate
314 microorganisms, but to possibly improve the performance and operation of MF/UF membranes.

315 The MF/UF processes provided the greatest average reduction of cells of any treatment
316 process, reducing total and intact cell counts by a combined average of 4.60 and 4.28 \log_{10} ,
317 respectively. Geometric mean total and intact cell counts in the MF/UF effluent were 2.09×10^2
318 cells/mL (geometric SD = 5.33) and 1.12×10^2 cells/mL (geometric SD = 5.13), respectively.

319 These large removal values for cell counts across MF/UF were expected, because the MF and UF
320 membranes had nominal pore sizes of 0.1 μm and 0.04 μm , respectively, which are considerably
321 smaller than the typical size range of 0.3 – 3 μm for wastewater bacteria [41]. The UF achieved
322 greater log removal than MF for both total (4.97 \log_{10} vs. 4.18 \log_{10} ; $p = 0.046$) and intact cell
323 counts (4.83 \log_{10} vs. 3.67 \log_{10} ; $p = 0.040$), possibly due to the smaller nominal pore size of the
324 UF membranes.

325 Cells observed in the MF/UF effluent may be a combination of organisms that passed
326 through MF/UF and organisms that originated from biofilms on the backside of the membranes or
327 other downstream surfaces. Cells can pass through MF/UF via abnormalities in the membranes
328 (e.g., enlarged pores or surface defects) caused by manufacturing imperfections and operational
329 stresses [42], especially if they are members of the “ultra-small” class of bacteria ($<0.2 \mu\text{m}$) [43].
330 There were no apparent visual differences in the side and forward scatter cytometric fingerprints
331 of the MF/UF feed and effluent streams that might have indicated substantial passage of small
332 bacteria through the membranes (data not shown).

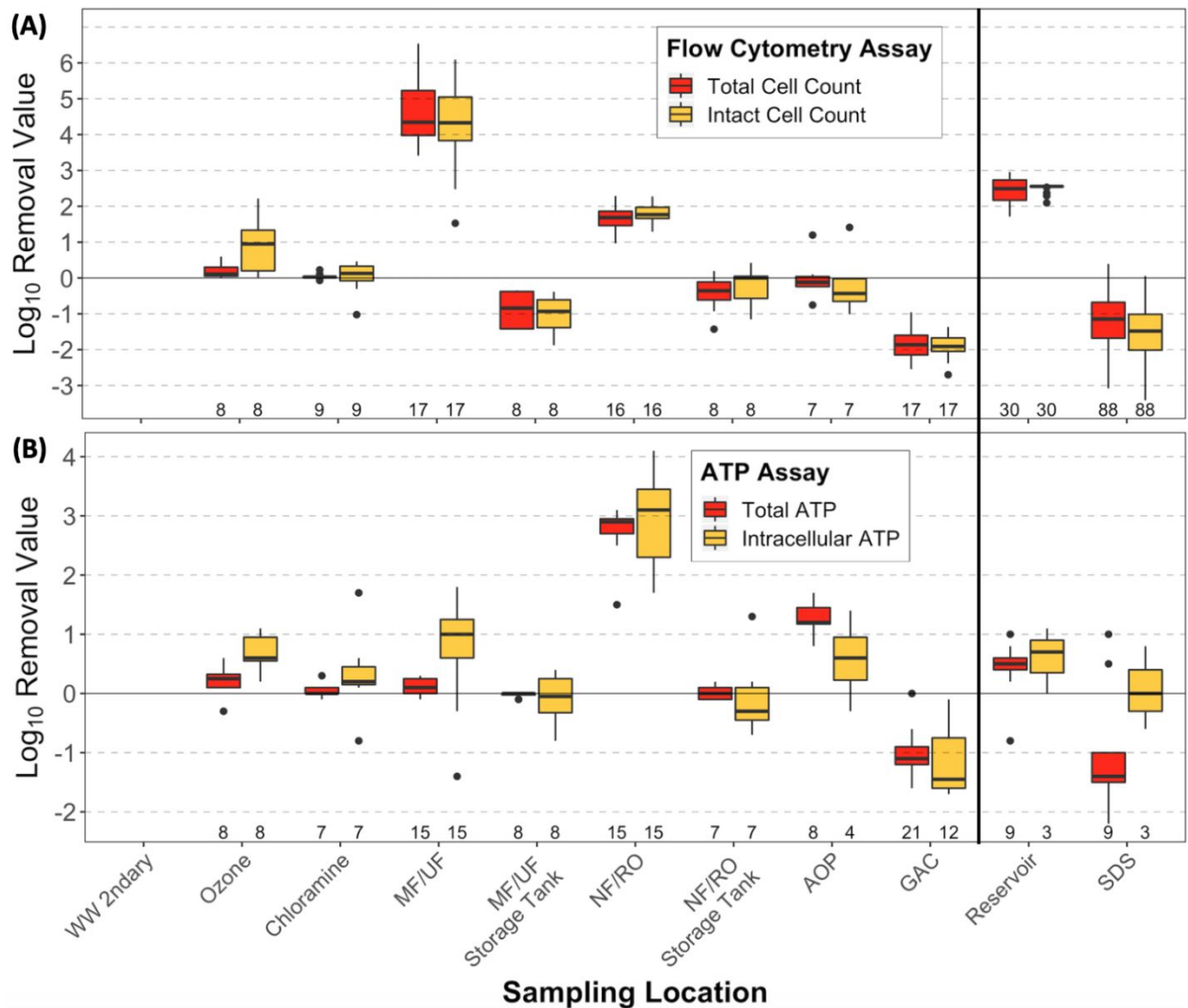


333

334 **Figure 2. Boxplot of (A) cell counts and (B) ATP concentrations across the pilot facility and into the bench-**
 335 **scale reservoirs and SDS. Data shown for MF/UF, NF/RO, GAC, Reservoir, and SDS are combined measurements**
 336 **from the respective parallel treatment process effluents. The vertical black line indicates batch chlorination of GAC**
 337 **filtrate. The lower limits of quantification for total and intact cell counts (12 and 22 cells/mL) and total and intracellular**
 338 **ATP (1×10^{-4} and 1.82×10^{-5} nM) are indicated by the red and blue colored lines, respectively. The total number of**
 339 **samples taken (n) at each location is located immediately above the x-axes. All samples were analyzed in technical**
 340 **triplicate.**

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Figure 3. Boxplots of \log_{10} removal values for (A) cell counts and (B) ATP concentrations across the pilot facility and into the bench-scale reservoirs and SDS. Data shown for MF/UF, NF/RO, GAC, Reservoir, and SDS are combined measurements from parallel units. The vertical black line indicates batch chlorination of GAC filtrate. The total number of samples taken (n) at each location is located immediately above the x-axes.

Total and intact cell counts rose in the MF/UF effluent storage tank by an average of 0.88

and 1.08 \log_{10} , respectively, and then fell by 1.67 \log_{10} and 1.80 \log_{10} across NF/RO. Geometric

mean total and intact cell counts in the NF/RO permeate were 48 cells/mL (geometric SD = 2.44)

and 28 cells/mL (geometric SD = 1.66), respectively. Differences in removal of total ($p = 0.272$)

and intact cell counts ($p = 0.254$) by NF and RO were not significant. Previous challenge studies

of RO membranes have yielded removal values for bacteria of 2.9 to 5.4 \log_{10} [42]. However,

355 assessment of the actual removal capacity of RO/NF was not possible because 71% of intact cell
356 count measurements ($n = 24$) were below the method quantification limit. Total and intact cell
357 counts increased by an average of 0.45 and 0.13 \log_{10} , respectively, between NF/RO permeates
358 and the NF/RO permeate storage tank. Total and intact cell counts rose by 0.01 and 0.20 \log_{10} ,
359 respectively, through the AOP, but these changes were not significant ($p = 0.39$ and 0.25,
360 respectively). Wunsch et al. (2019) reported a slight but significant decrease in intact cell counts
361 in river water that was subjected to a UV- H_2O_2 AOP (treatment conditions: 4 mg/L H_2O_2 with 600
362 mJ/cm² low pressure UV light) [44]. Regardless, damage to cellular membranes in UV- H_2O_2 AOPs
363 has been shown to be minimal as compared to the inactivation achieved via direct photolysis to
364 genomic DNA (e.g., formation of pyrimidine dimers and adducts) by UV light; this type of damage
365 will not be detected by an intact cell count assay [44,45].

366 Total and intact cell counts increased by an average of 1.80 and 1.91 \log_{10} , respectively,
367 across the GAC filter

368 s, indicating that a microbial community grew on the filter media that provided a consistent
369 source of cells to the filtrate. There were no significant differences in the change of total or intact
370 cell counts between the three GAC filters ($p > 0.05$ for all comparisons). Microbial growth on the
371 GAC filters may have reached steady-state, based on low variability in the total and intact cell
372 counts (geometric SD = 1.53 and 1.58, respectively) in the GAC filtrate across the pilot sampling
373 period (days 140 – 227). Additionally, we observed consistent microbial community profiles of
374 the GAC filtrates for days 184 – 224 of plant operation based on 16S rRNA gene sequencing
375 results [9]. It is highly unlikely that the major community members originated with the GAC when
376 it was installed. GAC media is produced using very high temperatures ($T \gg 100$ °C) and stored
377 dry. Cells in the GAC filtrate were most likely from a microbial community that colonized and

378 continuously multiplied in the GAC media. Subsequent batch chlorination and storage in the
379 reservoir reduced total and intact cell counts by greater than 2.44 and 2.51 \log_{10} , respectively,
380 given that 80% of intact cell count measurements were below the quantification limit ($n = 30$).

381 Concerns exist that media filters in water treatment facilities might harbor and release
382 opportunistic pathogens, with previous studies reporting increases of *Legionella* spp. [9,46,47] and
383 nontuberculosis mycobacteria [48,49] through sand and GAC filters. However, the methods
384 employed in these papers didn't distinguish between pathogenic and non-pathogenic strains of
385 *Legionella* and mycobacteria, and concentrations of *Legionella* spp. and *Mycobacterium* spp. in
386 the pilot's GAC filtrate was previously reported to be similar to other values reported for
387 conventional drinking water systems [9]. Furthermore, the combination of GAC filtration and
388 chlorination is very similar to what is practiced in conventional drinking water treatment. A recent
389 study by Cheswick et al. (2019) evaluated full-scale chlorination in numerous drinking water
390 treatment facilities and identified a CT value of >30 mg-min/L as a critical threshold for reducing
391 the proportion of intact cell counts to less than 1% [50]. We targeted a CT value of ~ 30 mg-min/L
392 in bench-scale chlorination; however, we did not measure cell count counts immediately after
393 chlorination (all measurements from the bench-scale storage reservoirs were taken after two days
394 of retention time).

395 The application of flow cytometry to advanced water treatment necessitated the
396 development of lower method quantification limits. Previous studies have reported quantification
397 limits of $\sim 2 \times 10^2$ to 1×10^3 cells/mL, which would usually permit satisfactory evaluation of
398 drinking water distribution systems, but restricts assessment of treatment processes like MF/UF,
399 NF/RO, AOP, and chlorination [20,26]. Therefore, lower quantification limits for total and intact
400 cell counts were established for low-cell count matrices by using modified electronic gates (see

401 Figure S1), increasing the sample run volume from 50 to 1,000 μL , and evaluating cell counts from
402 repeated measurements (11x) of un-stained and stained blanks (0.1 μm -filtered bottled mineral
403 Evian water). We set the lower limits of quantification as
404 $\Sigma(\text{average and } 3 * \text{standard deviation})$ of the stained blanks, yielding 12 and 22 cells/mL for
405 the total and intact cell count assays, respectively (**Supplementary Table S4**) [51]. Despite low
406 quantification limits, we observed substantial fractions of intact cell count measurements below
407 the quantification limit after MF/UF, NF/RO, and chlorination (i.e., in the simulated reservoir)
408 (**Supplementary Table S6**).

409

410 **3.2 Adenosine triphosphate concentrations throughout the pilot treatment train**

411 Similar to flow cytometry, measurement of ATP concentrations has been promoted as a rapid,
412 cultivation-independent alternative to quantify and assess microbial biomass and viability in
413 engineered water systems. In this study, ATP measurement was found to be less sensitive than
414 flow cytometry for detecting changes in microbial biomass through MF/UF and storage tanks, but
415 more sensitive for NF/RO and AOP. All measurements of total and intracellular ATP
416 concentrations taken throughout the pilot treatment train and three bench-scale SDS are presented
417 in **Figure 2B**, and calculated \log_{10} removal values are presented in **Figure 3B**. Summary statistics
418 and log removal data for ATP across all treatment processes can be found in **Tables S6** and **S8**.
419 Geometric mean total and intracellular ATP concentrations in the secondary wastewater feed were
420 0.727 (geometric SD = 1.62) and 0.663 nM (geometric SD = 1.69), respectively. Extracellular ATP
421 constituted 13% of total ATP in the pilot feed, and this ratio increased to 57% after ozonation and
422 to 75% after chloramination, indicating release of intracellular ATP by oxidative destruction of
423 microbial cells that is consistent with modest decreases in intact cell counts.

424 MF/UF reduced total and intracellular ATP by an average of 0.11 and 0.79 \log_{10} ,
425 respectively. These log reductions determined by ATP are $\sim 3 - 4 \log_{10}$ lower than those determined
426 by flow cytometry. The low reductions in intracellular ATP across MF/UF can be partially ascribed
427 to difficulty in measuring changes in intracellular ATP in water samples with high ratios of
428 extracellular-to-total ATP concentrations (90% for MF/UF) that confound the calculation of
429 intracellular ATP values (intracellular = total – extracellular), a challenge noted previously by a
430 study of ATP concentrations through ozonation and UF treatment of drinking water [20]. There
431 were no statistically significant differences in the removal of total ($p = 0.624$) and intracellular
432 ATP ($p = 0.904$) between the MF and UF membranes.

433 The average log removal values for total and intracellular ATP through NF/RO (2.76 and
434 2.92 \log_{10} , respectively) were significantly greater ($p < 0.001$) than removal values for total and
435 intact cell counts by $\sim 1 \log_{10}$. There were no statistically significant differences in the removal of
436 total ($p = 0.834$) and intracellular ATP ($p = 0.834$) between the NF and RO membranes. Significant
437 destruction of total ATP (1.28 \log_{10} ; $p = 0.002$) was observed across AOP, whereas no significant
438 changes were observed for intracellular ATP (0.58 \log_{10} , $p = 0.07$) or cell counts by flow
439 cytometry. Removal of total ATP was likely achieved via reactions with hydroxyl radicals and
440 other oxidants; a previous study reported insignificant changes in ATP concentrations by UV
441 photolysis alone [52].

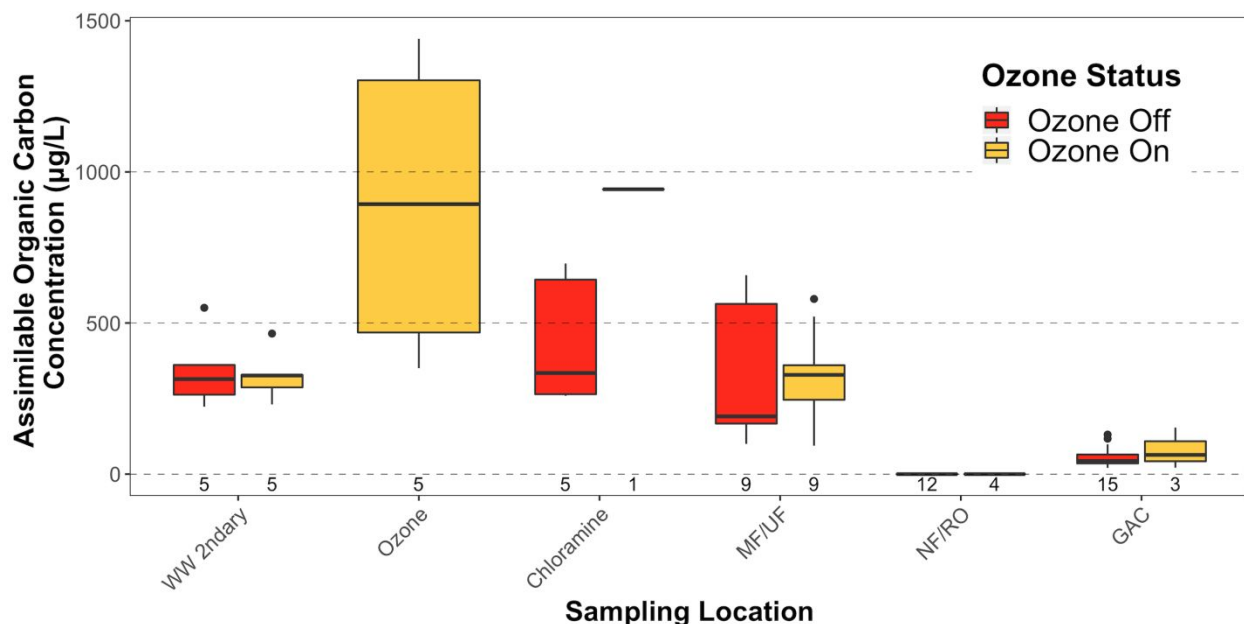
442

443 **3.3 Assimilable organic carbon throughout the pilot treatment train**

444 Despite high concentrations of AOC in the secondary treated wastewater feed, the GAC filtrate
445 contained AOC concentrations (average = 66 $\mu\text{g/L}$) on the lower end of what has been reported in
446 drinking water in the United States [53]. The ozone process was operated from October 26, 2015

447 – January 9, 2016 to evaluate potential benefits to MF/UF operation and was found to impact AOC
448 concentrations across the treatment train; therefore, changes in AOC across treatment processes
449 were analyzed separately for days when ozone was online versus offline (**Figure 4**). AOC data for
450 all individual processes are available in the supplementary information (**Supplementary Table**
451 **S9**). AOC concentrations presented herein are the values obtained after subtracting measurements
452 in respective negative controls (mean growth in negative controls = $14 \pm 2.5 \mu\text{g/L}$; $n = 8$).
453 Furthermore, for each sampling day, percent changes in AOC across treatment process (e.g.,
454 removal efficiencies) were calculated only if influent and effluent data for the respective treatment
455 process were available. AOC in the secondary wastewater was stable over the study period and
456 did not differ significantly ($p = 0.835$) between days the ozone unit was online ($327 \pm 86 \mu\text{g/L}$; n
457 $= 5$) or offline ($342 \pm 127 \mu\text{g/L}$; $n = 5$); therefore, differences in subsequent measurements between
458 dates when ozonation did or did not occur were likely not the result of unexpectedly high
459 differences in pilot feed AOC between those dates. The relatively low target ozone concentration
460 of 3.5 mg/L increased AOC by $276 \pm 157\%$, yielding fluctuating post-ozone AOC concentrations
461 with an average of $891 \pm 485 \mu\text{g/L}$ ($n = 5$). The influence of chloramination (target concentration
462 2 – 4 mg/L) may have depended on whether the upstream ozone process was in operation. When
463 ozone was offline, chloramination increased AOC by $35 \pm 50\%$ ($n = 5$). However, only one
464 chloramine sample taken when ozone was online passed QA/QC, in which chloramination
465 increased AOC by only 5%. More sampling is required to determine the impact of chloramination
466 on AOC in advanced water treatment.

467



468

469 **Figure 4: Boxplot of AOC concentrations throughout the pilot treatment facility.** MF/UF, NF/RO, and GAC
 470 represent combined data from respective parallel treatment process effluents. The total number of samples taken (n)
 471 at each location is located immediately above the x-axes. All samples were analyzed in technical triplicate, and data
 472 shown are results after subtracting out results from negative controls.

473

474 AOC rejection by MF and UF was low immediately following a maintenance clean or

475 recovery clean and increased with time after membrane cleanings (**Figure 5**), a phenomenon

476 discussed further below. Considering all samples taken over the study period, MF and UF were

477 not significantly different in AOC reduction ($p = 0.229$), with AOC reductions of $35 \pm 23\%$ for

478 MF ($n = 8$) and $51 \pm 32\%$ for UF ($n = 10$). However, considering only samples taken 9 hours or

479 more after membrane cleanings, the measured AOC removal efficiencies for MF and UF were 51

480 $\pm 10\%$ ($n = 5$) and $70 \pm 14\%$ ($n = 7$), resulting in a significantly higher removal

481 efficiency for UF as compared to MF ($p = 0.018$). These removal efficiencies for AOC fall within

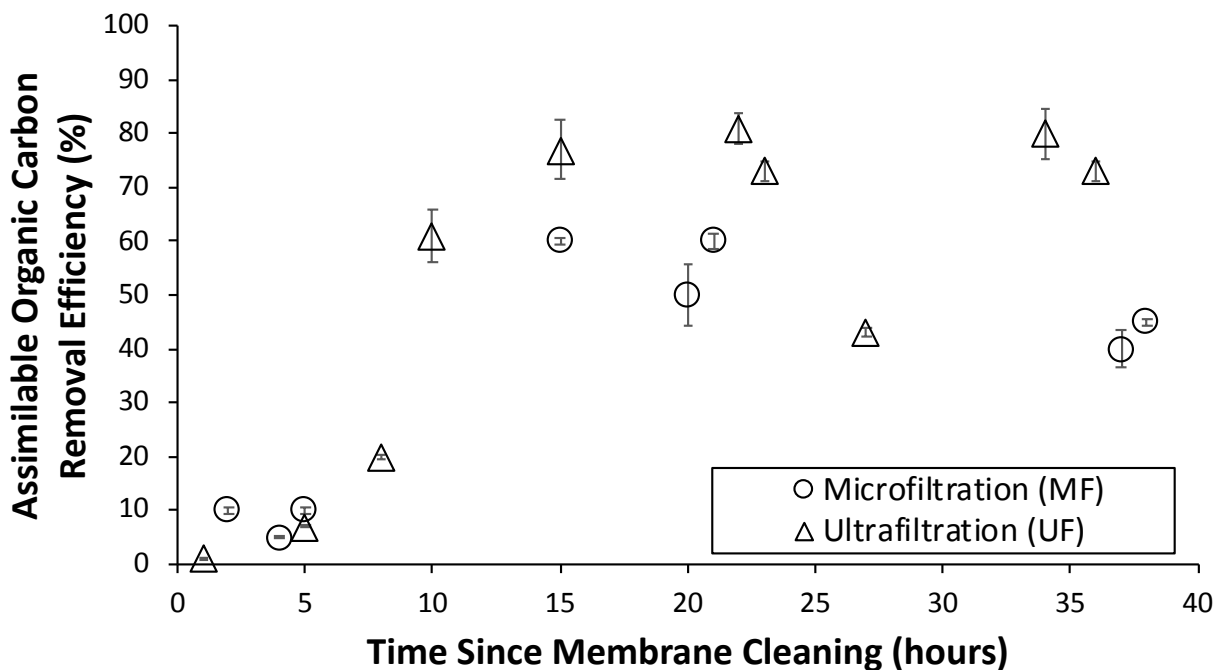
482 the range of reported total organic carbon rejection efficiencies for MF and UF of 45 – 65% and

483 50 – 75%, respectively [54]. Furthermore, considering only samples taken 9 hours or more after

484 membrane cleanings, we observed that combined MF/UF AOC concentrations were significantly

485 higher ($p = 0.043$) when the ozone unit was online ($323 \pm 172 \mu\text{g/L}$; $n = 8$) as compared to offline
486 ($156 \pm 39 \mu\text{g/L}$; $n = 4$).

487 The MF and UF units underwent maintenance cleans (to reverse moderate fouling)
488 approximately every 48 and 36 hours, respectively, and recovery cleans (to reverse heavy fouling)
489 about once per month (**Figure 5**). MF/UF rejection efficiencies for AOC were $8.8 \pm 6.4\%$ ($n = 6$)
490 when measured between hours 0 and 9 after a membrane cleaning, but rejection efficiencies rose
491 to $62 \pm 15\%$ ($n = 12$) when measured 9 hours or more after a cleaning. Fouling of the MF/UF
492 membranes (as indicated by increases in transmembrane pressures between cleanings; data not
493 shown) resulted in greater rejection efficiencies of AOC. These results are consistent with previous
494 studies of MF and UF that report increasing rejection efficiencies for natural organic matter and
495 dissolved organic carbon as membranes foul via pore plugging, pore size reduction, and cake
496 formation [55,56].



497
498 **Figure 5: Impact of time since membrane cleanings on the removal efficiency of assimilable organic carbon by**
499 **MF and UF.** All samples were analyzed in technical triplicate. Error bars show standard error of the mean.

500

501 As discussed previously, total and intact cell counts rose by 1.10 and 1.62 \log_{10} in the
502 MF/UF storage tank, which were likely driven by high concentrations of AOC in the MF/UF
503 effluents and a 30-min hydraulic retention time in the storage tank. Assuming $\sim 1 \mu\text{g/L}$ of AOC
504 yields $\sim 10^7$ cells/L [57], we estimate that the substantial regrowth observed in the MF/UF storage
505 tank would only have required $\sim 0.25 \mu\text{g/L}$. Furthermore, constituents of AOC have not been
506 reported as direct causes of fouling, but rather they serve as a substrate for growth of
507 microorganism that could cause significant biofouling over time [58,59]. Because planktonic
508 bacteria in RO feed waters have been shown to colonize and proliferate across membrane surfaces
509 [60], it may be advantageous to minimize the hydraulic retention time of inter-process storage
510 tanks to limit the numbers of bacteria that regrow in RO feed streams and could contribute to
511 fouling.

512 All AOC measurements of the NF/RO permeate samples yielded negative results (i.e., after
513 subtracting out results for the negative controls), which suggests growth inhibition of the Evian
514 inoculum in NF/RO permeate. Growth inhibition was likely not due to nutrient limitations, because
515 all AOC assays were amended with nutrient buffers containing 14 macro- and micronutrients (e.g.,
516 P, N, S, Fe, Mn, Co, etc.) previously designed to force carbon limitation [61]. Rather, growth was
517 likely inhibited in NF/RO permeate samples by low pH ($\sim 5.3 - 5.5$) that can hinder transport of
518 compounds across cellular membranes [62] and force bacteria to expend excess energy to maintain
519 homeostasis [63]. A recent study of bacterial growth methods by Sousi et al. (2018) reported
520 inhibition of growth in RO samples at low pH (5.5) but significant growth following adjustment
521 of RO sample pH to 7.8 with a sodium bicarbonate solution [64]. Future studies applying microbial
522 regrowth methods to measure AOC in NF and RO permeates should explore adjustment of sample
523 pH in addition to mineral and nutrient amendments to overcome possible growth inhibition.

524 The average AOC concentration in the GAC filtrate was $60 \pm 37 \mu\text{g/L}$. Filtrate AOC
525 concentrations for GAC 1 were $67 \pm 55 \mu\text{g/L}$ and for GAC 2 were $66 \pm 35 \mu\text{g/L}$. GAC 3 had lower
526 filtrate AOC concentrations ($47 \pm 26 \mu\text{g/L}$) than GAC filters 1 or 2, which could be due to the
527 longer empty bed contact time (15 min vs. 5 min, respectively), but this difference in filtrate AOC
528 was not significant ($p = 0.257$). Sources of AOC in the GAC filtrates may have been partially
529 composed of low molecular weight, uncharged carbon compounds (e.g., alcohols and aldehydes)
530 that passed through RO but were below the AOC method detection limit (possibly due to inhibition
531 by low pH) at the NF/RO sampling location [65]. This possibility is supported by metagenomes
532 recovered from the GAC media-associated biofilm and filtrates, in which genes associated with
533 the metabolisms of methanol and formaldehyde were identified [9]. If RO permeate was a source,
534 we would expect higher AOC in the GAC filtrate when ozonation was online ($80 \pm 68 \mu\text{g/L}$; $n =$
535 3) versus offline ($56 \pm 33 \mu\text{g/L}$; $n = 15$), a trend that was observed but was not statistically
536 significant ($p = 0.613$). Another potential source of carbon was autotrophic production by algae,
537 consistent with observations of green growth on the walls of the uncovered GAC media columns
538 and metagenomic detection of genes in the GAC media-associated biofilm likely corresponding to
539 the green algal order Sphaeropleales [9]. A final potential carbon source may have the GAC media
540 itself, but to our knowledge the leaching of labile carbon from GAC media has not been reported
541 in the literature.

542 The average measured AOC concentration of $66 \mu\text{g/L}$ in the GAC filtrates is on the lower
543 end of values previously reported in treated and distributed drinking waters in the United States.
544 A study of 95 drinking water treatment facilities by Volk and LeChevallier (2000) reported average
545 treated effluent AOC concentrations of $\sim 100 \mu\text{g/L}$, with only 9% of facilities producing a treated
546 effluent with AOC $< 50 \mu\text{g/L}$ [53]. The El Paso pilot facility produces a finished water (i.e., GAC

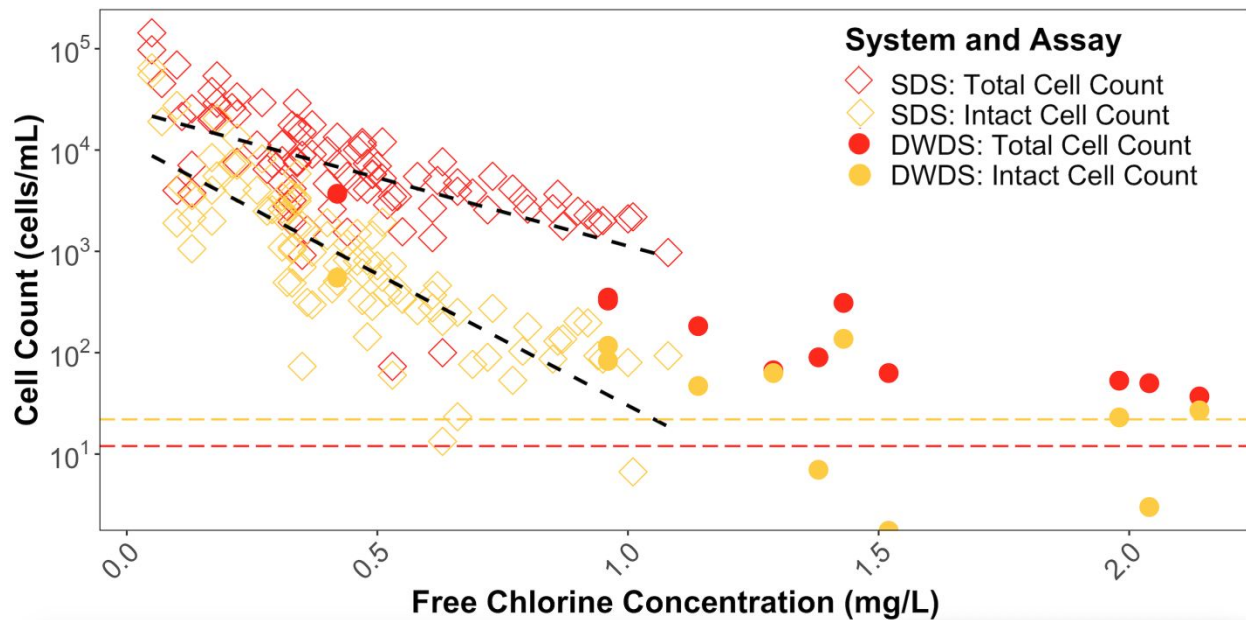
547 filtrate) more similar to that reported for drinking water treatment in Zürich, Switzerland (32 µg/L)
548 [25] and Rotterdam, the Netherlands (~30 µg/L) [18]. These treatment systems employ ozonation
549 followed by biological filtration to reduce concentrations of labile carbon and maintain biological
550 stability in the distribution system without the application of a residual disinfectant.

551

552 **3.4 Microbial abundance dynamics in the simulated distribution systems (SDS)**

553 Residual free chlorine exhibited a strong influence on planktonic cells in the SDS, in which
554 we observed a clear inverse correlation for both total and intact cell counts with free chlorine
555 concentrations (**Figure 6**). Increasing free chlorine residual was expected to lower cell counts,
556 given previous reports of lower heterotrophic plant counts and total coliforms at higher chlorine
557 residual concentrations in distribution systems [7,66,67]. Measurements of flow cytometric cell
558 counts and free chlorine residual from the three SDS indicate that an inverse relationship also
559 exists between the entire population of planktonic cells and free chlorine, a finding that is
560 consistent with studies of full-scale drinking water distribution systems in Scotland [51], Latvia
561 [19], and France [68]. However, this is the first demonstration of cell counts and chlorine dynamics
562 in the distribution of advanced treated water, for which we expect the relatively lower suspended
563 solids and organic matter content to increase disinfection efficacy (relative to conventionally-
564 sourced drinking water) by reducing protection afforded to microorganisms [69]. Several other
565 parameters have been linked to planktonic cell concentrations in distribution systems, such as
566 temperature [18,70], water age [19,71,72], concentrations of labile carbon [6,18,73], and
567 concentrations of nutrients [74,75], but we did not test for these relationships.

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Figure 6: Effect of free chlorine concentration on total and intact cell counts in the SDS and in El Paso's full-scale drinking water distribution system (DWDS). Regression models (dashed black lines) were developed from cell counts in the SDS only. SDS data only includes samples taken after the SDS acclimation period (i.e., after Day 189 of pilot operation). The limits of quantification for total and intact cell counts (12 and 22 cells/mL) are indicated by the dashed red and blue colored lines, respectively. All samples were analyzed in technical triplicate.

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The inverse relationship of cell counts and free chlorine concentration in the SDS may be stronger than in typical full-scale distribution systems due to enhanced free chlorine disinfection conditions prior to, and in the SDS (i.e., relatively low pH and efficient mixing), less established biofilms, and lower amounts of suspended solids and loose deposits that can harbor microorganisms [76]. The average pH of the NF and RO permeates was 5.61 and 5.71, respectively, which rose to 6.53 ± 0.18 in the SDS ($n = 204$) without adjustment, and is lower than the typical pH range of 7.3 – 8.1 for El Paso's full-scale distribution system [77]. Because free chlorine is a weak acid (HOCl/OCl^- , $\text{pK}_A = 7.54$), at a pH of 6.53 about 90% of the SDS chlorine residual would have existed in the form of HOCl , which is estimated to be 40-80x more effective at inactivating bacteria than its counterion OCl^- [78,79]. It should be noted that for full-scale membrane-based potable reuse facilities, the lack of minerals and low pH of NF/RO permeates will necessitate pH adjustment and remineralization (e.g., decarbonation and lime addition) of

588 finished water to limit corrosion of distribution infrastructure [80]. Therefore, the lack of pH
589 adjustment and remineralization of GAC filtrate in this study somewhat limits the applicability of
590 the SDS results to full-scale potable reuse systems.

591 Free chlorine exerted more influence on intact cell counts than total cell counts, as
592 evidenced by log-linear slopes of -1.50 ± 0.13 for total cell counts ($R^2 = 0.675$, $n = 68$) and -2.71
593 ± 0.17 for intact cell counts ($R^2 = 0.787$, $n = 68$) (**Figure 6**). This greater decline in intact cell
594 counts with increasing free chlorine residual may be attributable to more widespread oxidative
595 damage to cellular membranes without complete destruction of cells, a finding previously observed
596 during free chlorine disinfection [81]. We were unable to find regression models in the literature
597 for other simulated or full-scale drinking water distribution systems.

598 The relationship between cell counts in the SDS and free chlorine residual can also be
599 visualized as a time series (**Supplementary Figure S4**). Significant decay of the free chlorine
600 residual occurred in the SDS, as measured by large differences in free chlorine ($59 \pm 16\%$, $n =$
601 102) between the reservoirs and the SDS. In general, total and intact cell counts rose as the free
602 chlorine residual fell in the SDS, and vice versa. Throughout the study, the storage reservoirs
603 contained high concentrations of free chlorine (>0.5 mg/L as Cl_2) and low geometric mean total
604 (39 cells/mL; geometric SD = 2.18) and intact cell counts (24 cells/mL; geometric SD = 1.28).
605 Despite a low inflow of planktonic cells, the SDS harbored large quantities of total and intact cells,
606 likely due to a combination of bulk water regrowth and release of cells from tubing and reactor
607 surface biofilms. A recently published study of a full-scale system fed with ultrafiltered (i.e., low-
608 cell count) drinking water estimated that $>50\%$ of planktonic cells in distribution originated from
609 pipe biofilm (the remaining fraction originated from the treatment facility's finished water), but
610 the researchers assumed negligible regrowth of bacteria in the bulk water due to short residence

611 times (<25 hour) and low temperatures (~7.5 °C) [82]. However, the higher temperatures (~21.5
612 °C) in the SDS described here could be more conducive to microbial growth.

613 Twelve samples were collected from El Paso's drinking water distribution system on
614 January 26, 2015 for comparison to the SDS results (**Figure 6**). These samples had an average free
615 chlorine concentration of 1.4 mg/L as Cl₂ (range: 0.42 – 2.14 mg/L), pH of 7.9 ± 0.2, and
616 temperature of 20.6 ± 0.25 °C. It is difficult to draw comparisons between the full-scale and
617 simulated systems given substantial differences in residual free chlorine and the limited number
618 of full-scale samples. It should be noted that the target free chlorine residual (~1.0 mg/L as Cl₂)
619 for the SDSs was chosen to increase total biomass available (compared to higher residual
620 concentrations) for 16S ribosomal and metagenomic sequencing analyses; these results were
621 published separately [9].

622 Previous studies have calculated ATP content per cell ("ATP-per-cell") to assess changes
623 in metabolic activity [83,84] and to propose conversion factors for estimating bacterial cell
624 concentrations from ATP data [17,21,85]. It is crucial for researchers to note the specific
625 methodology used for calculating ATP-per-cell, because results will vary depending on use of total
626 or intracellular ATP, viable or total cell counts, and cell units or cell biovolume. A correlation (R^2
627 = 0.47; $p < 0.001$; $n = 43$) was observed between intracellular ATP and intact cell count
628 measurements in the SDS (**Supplementary Figure S5**), resulting in a geometric mean ATP-per-
629 cell value of 5.50×10^{-10} nmol/cell (geometric SD = 3.23; $n = 43$), which is about 4.5 times greater
630 than the geometric mean 1.23×10^{-10} nmol/cell (geometric SD = 2.12; $n = 21$) found in the GAC
631 filtrates from this study, and about three times greater than the average 1.74×10^{-10} nmol/cell
632 reported previously by Hammes et al. (2010) [17]. Interestingly, Hammes et al. (2010) tested many
633 water types (i.e., groundwater, lake and river water, bottled water, unchlorinated drinking water,

634 wastewater effluent) and found their ATP-per-cell value applied to all tested waters; however, their
635 dataset did not include chlorinated drinking water. ATP-per-cell values can be influenced by the
636 physiological state, size, and composition of microorganisms in a sample [17,83,84,86], all of
637 which in turn can be influenced by the presence of chlorine [87-89]. Considering that the GAC
638 filtrate and SDS shared a large portion of their microbial communities, we speculate that the
639 presence of a free chlorine residual in the SDS elevated the average metabolic activity of
640 microorganisms as compared to the GAC filtrate.

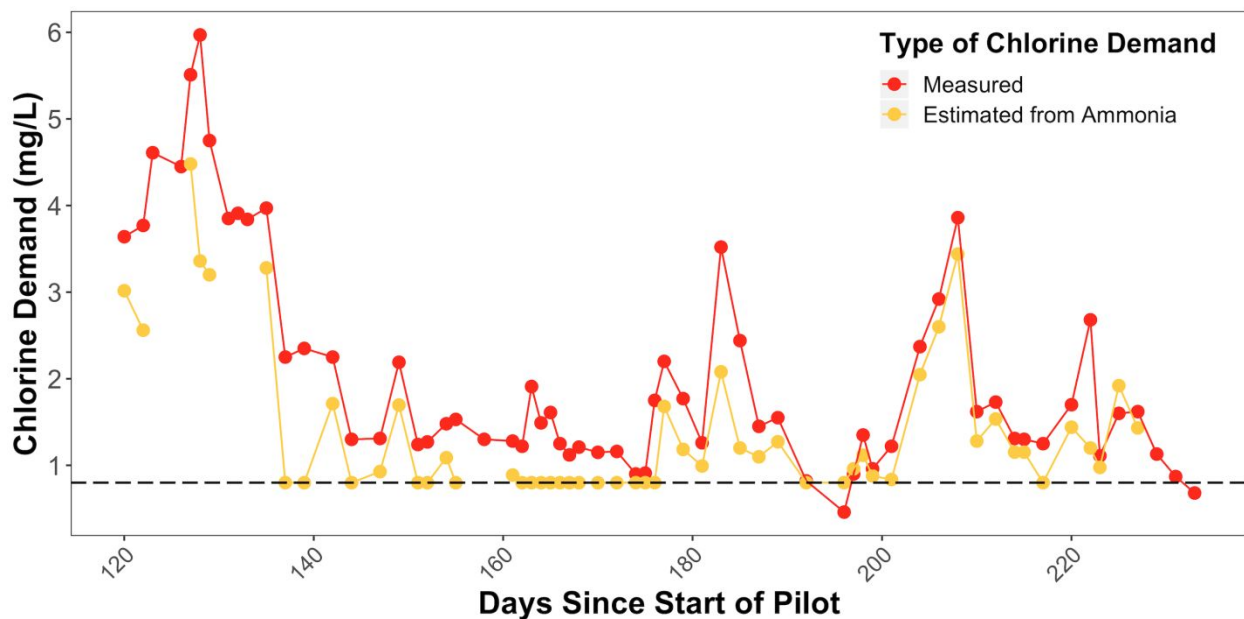
641

642 **3.5 Dynamics of chlorination and chlorine demand**

643 A high variability in chlorine demand of the GAC filtrates was observed across the study
644 period, which tracked closely with estimated demand from ammonia (**Figure 7**). Total ammonia-
645 N (TAN) fluctuations in the pilot were linked to diurnal variations in nitrification efficiency in the
646 upstream secondary wastewater treatment facility. The activated sludge treatment process
647 frequently failed to nitrify influent TAN fully during peak loading periods (e.g., mid-morning),
648 leading to average and 90th percentile concentrations of TAN in the pilot feed of 3.0 and 7.5 mg/L
649 as N, respectively (n = 54,592). The NF/RO membranes were expected to achieve the majority of
650 TAN removal. Because sulfuric acid addition in the NF/RO feed dropped the pH to ~6.5 to limit
651 mineral scaling on the membrane, the vast majority of TAN likely existed in the form of
652 ammonium (pKa = 9.24), which has previously reported rejection efficiencies by RO of >95%
653 [90]. Despite a cumulative removal efficiencies of ~93% for TAN through NF/RO, the average
654 and 90th percentile concentrations of TAN in the RO permeate were 0.20 and 0.40 mg/L as N,
655 respectively (n = 47,394), although 71% of measurements were below the method detection limit
656 (0.1 mg/L as N). However, the average TAN concentration in the NF/RO permeate rises to $0.37 \pm$

657 0.35 mg/L as N when considering only days for which measurements of TAN in the NF/RO
 658 permeate were above the detection limit. TAN would likely pass through the AOP and GAC filters
 659 and ultimately cause an estimated chlorine demand of 3.0 ± 2.8 mg/L during breakpoint
 660 chlorination of GAC filtrate, assuming a 8:1 weight ratio of Cl:N [36]. Zeng et al. (2015) observed
 661 higher TAN concentrations in the finished water from five potable reuse advanced treatment trains
 662 (range: 0.19 – 2.41 mg/L as N), which would exert 1.5 – 19.3 mg/L of chlorine demand [5]. It
 663 should be noted that conventional wastewater treatment facilities have been designed and operated
 664 to meet specific standards (e.g., environmental protection) that differ from what may be desired
 665 for optimal preparation of wastewater for advanced treatment [91].

666



667

668 **Figure 7: Chlorine demand in bench-scale breakpoint chlorination of the pilot's GAC filtrates.** Measured
 669 chlorine demand for each day is the average from the three GAC filtrates. The limit of detection for estimated demand
 670 from ammonia (based on a 0.1 mg/L limit of detection for measurement of ammonia-N in the NF/RO permeate) is
 671 indicated by the horizontal dashed blue line.

672

673 Considering all days the GAC filtrate was breakpoint chlorinated, the average measured

674 chlorine demand in GAC filtrate was 2.04 mg/L (median = 1.54 mg/L; n = 192). However, an

675 measured chlorine demand surpassed 3.0 mg/L on 20% of batch chlorination days (n = 64), with

676 a maximum measured chlorine demand of 5.97 mg/L. Neglecting days for which TAN
677 concentrations were below the online analyzer's detection limit (0.1 mg/L as N), we estimated that
678 demand from TAN accounted for an average of $79 \pm 15\%$ of measured chlorine demand ($n = 34$).
679 Other possible sources of chlorine demand in the GAC filtrates included organic material, metals,
680 and nitrite. In the NF/RO permeate grab samples, however, average and 90th percentile
681 concentrations were low and relatively stable for total organic carbon (0.37 and 0.49 mg/L; $n =$
682 14), nitrite-N (0.04 and 0.06 mg/L; $n = 122$), total iron, (<0.04 and <0.04 ug/L; $n = 14$) and total
683 manganese (<2.5 and <3.5 ug/L; $n = 14$).

684

685 **4. Implications for Practice and Future Research**

686 The combination of flow cytometry with ATP measurements more completely tracked
687 microbial abundance and viability throughout advanced treatment than either method alone,
688 supporting previous studies that have proposed pairing these two cultivation-independent
689 approaches to assess drinking water systems [15-19]. The choice of which method to employ
690 depends on what treatment process is being evaluated. Regarding NF/RO, we observed greater
691 removal values of total ATP ($2.71 \pm 0.39 \log_{10}$) as compared to total cell counts by flow cytometry
692 ($1.67 \pm 0.40 \log_{10}$). Moreover, credited removal of pathogens by RO membranes is typically based
693 on ambient indicators such as conductivity or total organic carbon, but removal of these
694 compounds by RO is typically low ($1 - 2.3 \log_{10}$) [92] when compared to the removal of pathogens
695 such as adenovirus ($2.7 - 6.5 \log_{10}$) [4]. Measured removal of ambient ATP may be a better
696 indicator for pathogen removal by NF/RO than currently accepted parameters. In contrast to
697 NF/RO, we observed much greater removal of total cell counts ($4.60 \pm 0.72 \log_{10}$) as compared to
698 total ($0.11 \pm 0.14 \log_{10}$) or intracellular ATP ($0.79 \pm 0.79 \log_{10}$) across the MF/UF membranes.

699 Observed removal of cells by flow cytometry was similar to and frequently higher than the $4 \log_{10}$
700 removal credited for *Cryptosporidium* and *Giardia* by MF/UF pressure decay tests [93]. However,
701 these decay tests are indirect, are typically only conducted once or twice per day, and yield scant
702 information about performance trends that are time-dependent (e.g., effects of membrane cleanings
703 on pathogen removal). Online and continuous measurements of total cells across MF/UF or total
704 ATP across NF/RO could provide enhanced and more accurate monitoring of microbial removal
705 in advanced treatment trains. However, the benefits and drawbacks to full-scale implementation
706 of flow cytometry and ATP analysis must also be compared with other emerging techniques. For
707 example, the IMD-WTM bacteriological counter (BioVigilant, Tuscon, AZ) has recently been used
708 to quantify cells in RO feed and permeate without the use of reagents, which could significantly
709 lower operating costs for continuous quantification of bacteria as compared to stain-intensive flow
710 cytometric methods [94].

711 Continuous flow cytometry has recently been applied for evaluating the microbial water
712 quality of unchlorinated tap water [95,96] and evaluating simulations of shock chlorination [89],
713 but has yet to be applied to advanced treatment trains. Future research is warranted to explore the
714 potential for online, continuous flow cytometry to enhance monitoring of advanced treatment
715 processes, specifically MF/UF, full-scale chlorination, regrowth in inter-process storage tanks, and
716 evaluating the biological activity of GAC filters. More research is also needed to determine the
717 utility of flow cytometry, either with grab samples or on-line configuration, to monitor microbial
718 water quality in both conventional and potable reuse distribution systems.

719 However, application of enhanced microbial evaluation tools also highlighted the
720 continued challenges in microbial monitoring of advanced treated water. For example, although
721 we detected low total and intact cell counts (e.g., <50 cells/mL) in advanced treated and chlorinated

722 water, evaluation of actual removal efficiencies was constrained by the limits of quantification.
723 Furthermore, removal efficiencies of ATP concentrations were low through ozonation and MF/UF
724 as compared to removal observed for cell counts. This is similar to a finding by Hammes et al.
725 (2010), who cautioned the use of ATP measurements in these water treatment matrices [20].
726 Finally, more work is needed to overcome inhibition due to low pH, minerals, and nutrients when
727 applying growth-based bioassays for waters treated by NF or RO. A recent study by Sousi et al.
728 (2018) successfully applied a growth-based bioassay to groundwater treated by RO, but such
729 methods still need to be successfully applied to RO/NF and AOP matrices in potable reuse [64].

730 The large fluctuations in ammonia-based chlorine demand in the finished water (i.e., GAC
731 filtrate) observed in this study highlights one potential challenge of producing purified water with
732 a stable chlorine residual at full-scale facilities. Some advanced treatment facilities rely on free
733 chlorine disinfection for large removal credits for viruses ($4 - 6 \log_{10}$) to meet stringent potable
734 reuse regulations for pathogens [3], which will be more difficult to achieve with fluctuating
735 chlorine demand. Moreover, reducing chlorine demand can minimize the formation of disinfection
736 byproducts (e.g., haloacetonitriles and haloacetaldehydes), which have been observed in higher
737 concentrations following chlorine-based advanced oxidation of NF/RO permeate [97,98]. It should
738 be noted that greater total ammonia-N concentrations have been observed in other studies of
739 advanced treatment trains [99], which would cause considerably greater chlorine demand than that
740 reported herein. Thus, we recommend ensuring complete nitrification upstream in conventional
741 wastewater treatment, which could have additional benefits for advanced treatment by reducing
742 biofouling and operating pressures of MF membranes [100]. Future research is needed to
743 investigate the impacts chemical conditioning of RO and NF permeates (e.g., pH adjustment and
744 remineralization to prevent pipe corrosion) may have on disinfection processes. There also remains

745 a need to evaluate the efficacy and stability of chlorine residual in potable reuse distribution
746 systems.

747 Lastly, more research is needed to evaluate the impact of different final treatment steps
748 (e.g., GAC filtration) on finished water quality, as well as characterizing the implications of
749 introducing advanced treated wastewater into existing distributions systems. Although not
750 intended to be operated in biological mode, the GAC filters in this study harbored a microbial
751 community that was both active (as indicated by increases in intact cell counts through filtration)
752 and previously reported to be consistent over time [9]. From a treatment process perspective, a last-
753 stage GAC filter can act as an additional barrier for organic pollutants [101], such as through
754 microbially-mediated breakdown of disinfection byproduct precursors [102], while not increasing
755 concentrations of opportunistic pathogens relative to conventional water systems [9]. From an
756 ecological perspective, a diverse microbial community in distribution should lead to greater
757 biological stability and be more capable of outcompeting frank and opportunistic pathogens
758 [103,104]. Because advanced treatment decreases bacterial presence and diversity, seeding of
759 microbes from GAC filters could increase finished water biostability. In a complementary study
760 of this pilot system we observed some microbial diversity shared between GAC filtrates and the
761 SDS [9], but more research is needed to determine the impacts of final treatment steps on microbial
762 ecology in distribution.

763

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777

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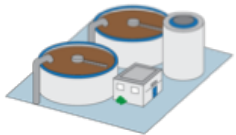
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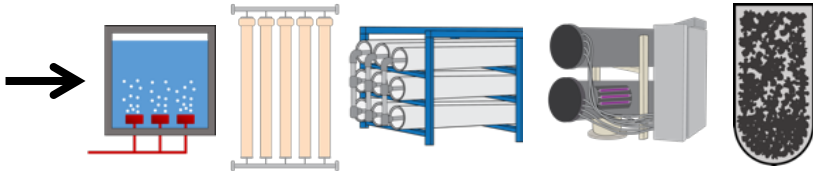
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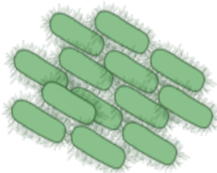
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Wastewater Treatment



Environmental Science, Water, Health & Technology
Advanced Treatment



Distribution System



How many cells? How viable are they?
How much potential for growth?

