Physical Chemistry Chemical Physics





The Infuence of Model Building Schemes and Molecular Dynamics Sampling on QM-cluster Models: The Chorismate Mutase Case Study

Journal:	Physical Chemistry Chemical Physics
Manuscript ID	CP-ART-12-2023-006100.R2
Article Type:	Paper
Date Submitted by the Author:	21-Mar-2024
Complete List of Authors:	Agbaglo, Donatus; The University of Memphis,, Chemistry Summers, Thomas; The University of Memphis, Department of Chemistry Cheng, Qianyi; The University of Memphis,, Department of Chemistry DeYonker, Nathan; The University of Memphis,, Chemistry

SCHOLARONE[™] Manuscripts

The Influence of Model Building Schemes and Molecular Dynamics Sampling on QM-cluster Models: The Chorismate Mutase Case Study

Donatus A. Agbaglo,[†] Thomas J. Summers,^{†,‡} Qianyi Cheng,[†] and Nathan J. DeYonker^{*,†}

[†]Department of Chemistry, University of Memphis, Memphis, TN 38152, U.S.A. [‡]Theoretical Division, Los Alamos National Laboratory, Los Alamos, NM 87545, U.S.A.

E-mail: ndyonker@memphis.edu

¹ Abstract

Most QM-cluster models of enzymes are constructed based on X-ray crystal structures, which 2 limits comparison to *in vivo* structure and mechanism. The active site of chorismate mutase 3 from *Bacillus subtilis* and the enzymatic transformation of chorismate to prephenate is used 4 as a case study to guide construction of QM-cluster models built first from the X-ray crystal 5 structure, then from molecular dynamics (MD) simulation snapshots. The Residue Interac-6 tion Network-based ResidUe Selector (*RINRUS*) software toolkit, developed by our group to 7 simplify and automate the construction of QM-cluster models, is expanded to handle MD to 8 QM-cluster model workflows. Several options, some employing novel topological clustering 9 from Residue Interaction Network (RIN) information, are evaluated for generating conforma-10 tional clustering from MD simulation. *RINRUS* then generates a statistical thermodynamic 11 framework for QM-cluster modeling of the chorismate mutase mechanism via refining 250 12

MD frames with Density Functional Theory (DFT). The 250 QM-cluster models sampled 13 provide a mean ΔG^{\ddagger} of 10.3 \pm 2.6 kcal mol⁻¹ compared to the experimental value of 15.4 14 kcal mol⁻¹ at 25 ^oC. While the difference between theory and experiment is consequential, 15 the level of theory used is modest and therefore "chemical" accuracy is unexpected. More 16 important are the comparisons made between QM-cluster models designed from the X-ray 17 crystal structure versus those from MD frames. The large variations in kinetic and ther-18 modynamic properties arise from geometric changes in the ensemble of QM-cluster models, 19 rather from the composition of the QM-cluster models or from the active site-solvent inter-20 face. The findings open the way for further quantitative and reproducible calibration in the 21 field of computational enzymology using the model construction framework afforded with 22 the *RINRUS* software toolkit. 23

24 Introduction

Through multiscale QM/MM or QM-only "cluster model" studies, stationary points along a 25 reaction mechanism can be optimized, which allows a structural probe of the enzyme kinet-26 ics that is impossible to directly observe experimentally.¹ As the reliability of computational 27 enzymology and the tractable size of QM-regions increase, a greater focus on cyberinfrastruc-28 ture is required for building consistent and reproducible atomic-level enzyme models. Our 29 group has developed the Residue Interaction Network Residue Selector (RINRUS) software 30 toolkit to facilitate studying the reaction mechanisms of enzymes with quantum chemistry.^{2–4} 31 Instead of relying on chemical intuition or distance-based criteria to prioritize the critical 32 fragments within the enzyme active site, RINRUS algorithmically constructs enzyme models 33 based on several possible qualitative and quantitative criteria. RINRUS infrastructure was 34 first developed to build QM-cluster models of enzymes, but adapting the code to also build 35 QM/MM enzyme models is in progress. In this work, we explore the enzyme chorismate mu-36 tase in conjunction with a proof-of-concept expansion of *RINRUS* capabilities: interfacing 37



³⁸ QM-cluster modeling with Molecular Dynamics (MD) techniques.

Scheme 1: Schematic representation of the Claisen rearrangement of chorismate to prephenate

Chorismate mutase (CM) catalyzes the reaction of chorismate to prephenate, participat-39 ing in the shikimate pathway that biologically produces phenylalanine and tyrosine amino 40 acids (Scheme 1).⁵⁻¹⁴ The shikimate pathway does not occur in the animal kingdom, and 41 thus provides a target for the development of new antibiotics, fungicides, and herbicides.¹⁵ 42 While chorimsate mutase has been widely studied experimentally and computationally, there 43 are still mysteries to be unraveled with respect to the extraordinary kinetic enhancement of 44 its active site. The chorismate mutase enzymatic reaction promotes a 10⁶-fold rate accelera-45 tion of prephenate production through a Claisen rearrangement in the catalytic elementary 46 step.^{16,17} This Claisen rearrangement is one of the few known examples of a naturally-47 occurring catalyzed pericyclic reaction.¹⁸ In 1993, Lipscomb and coworkers published an 48 X-ray crystal structure of *Bacillus subtilis* chorismate mutase (BsCM, PDB: 2CHT) at 2.2 49 Å resolution that forms the basis of most theoretical works.¹⁹ This structure contains an 50 endo-oxabicyclic transition state analogue (TSA), 8-hydroxy-2-oxa-bicyclo[3.3.1]non-6-ene-51 3,5-dicarboxylic acid, which offered structural insight into the enzyme mechanism. Since the 52 pericyclic reaction does not involve covalent substrate-protein bonding or acid-base chem-53 istry, CM makes an intriguing, and in some respects, simplified case study of enzyme catal-54 vsis.^{20,21} 55

⁵⁶ Mutagenesis, computational enzymology, and biochemical kinetics have been indispens-

able tools to study the mechanism of the CM reaction, especially for exploring the transition 57 state stabilization (TSS) and near-attack conformation (NAC) hypotheses or for describ-58 ing manifestations of their complementary kinetic and thermodynamic behavior.^{9,18,22–25} 59 Mutagenesis experiments of *Escherichia coli* chorismate mutase (EcCM)^{26–30} and BsCM 60 revealed the catalytic importance of many charged active site residues for establishing hy-61 drogen bonding with the negatively charged substrate. For example, replacement of Arg90 62 with a positively-charged lysine still decreases the catalytic efficiency by at least three orders 63 of magnitude in BsCM.^{19,31} 64

Theoretical studies of chorismate mutase with QM/MM-MD first emphasized the im-65 portance of a near attack conformation (NAC) as the main catalytic driving power behind 66 the proposed mechanism.^{20,32,33} Studies done by Bruice and co-worker showed that NAC 67 rearrangement of chorismate structure is a result of activated carbon and oxygen ligand 68 atoms approaching within the van der Waals contact distance at very small bond angles, 69 creating a favorable orientation of π -orbital overlap.^{20,32} The proponents of the NAC hy-70 pothesis focus on geometric distortion of the substrate in the active site. However, those 71 who argue for the TSS hypothesis indicate that positively charged residues like Lys39 in 72 EcCM and Arg90 in BsCM stabilize the developing negative charge during bond breaking 73 at the ether oxygen.^{25,31,34} Bond-breaking then leads to electrostatic stabilization of active 74 site residues, lowering the activation energy. Subsequent QM/MM and QM-cluster model 75 calculations have provided evidence that catalysis is due to both near attack conformation 76 and transition state stabilization, but with TSS being the main driving force of the proposed 77 mechanism. 21,22,35 78

While computational enzymology has advanced rapidly over the last two decades, ^{6,36–38} one persistent challenge in this research area is designing effective QM-regions that reliably predict catalytic activity, with kinetic and thermodynamic properties that can converge quickly with respect to model size. *Ad hoc* methods of selecting residues for inclusion in the QM regions of QM/MM models or in QM-cluster models are poorly reproducible and

not well calibrated. One technique for QM region selection is to include all residues that 84 are within a specific radial distance from the center of the active site or from the center of 85 mass of the substrate. This construction paradigm is based on the idea that spherical active 86 site models are appropriate. Several studies by our group and others reveal that this is not 87 always the case, ^{3,4,39–46} though CM active sites are known to be fairly compact and spherical. 88 To facilitate improved benchmarking in computational enzymology, our group has created 89 the *RINRUS* software toolkit to automate the process of generating QM-cluster models. Our 90 goal is to address various community-wide challenges in computational enzymology, such 91 as standardizing QM-cluster (and eventually QM/MM) model construction, lowering the 92 learning curve for new users, and reducing trial and error caused by *ad hoc* model-building 93 schemes. *RINRUS* uses an automated approach to trim and cap the active site fragments. 94 With a given protein structure and a user-defined "seed", which consists of the substrate 95 and any active site fragments necessary to describe the chemical reaction, RINRUS identifies 96 proximal fragments that have important non-covalent interactions with the seed using the 97 graph theory concept of the Residue Interaction Network (RIN).^{47,48} 98

To summarize the *RINRUS* procedure, a protein structure is converted into a RIN graph 99 composed of only a subset of the fragments (referred to as "nodes" in graph theory) that have 100 an identifiable electrostatic and/or steric interaction (referred to as "edges" in graph theory) 101 with the seed nodes.^{47,48} The RIN is then processed using one or more user-selected schemes 102 that identify qualitative interaction types (Structural Interaction Fingerprints, SIFs)⁴⁹ or 103 quantitative schemes that utilize first-principles interaction energies like symmetry-adapted 104 perturbation theory (SAPT or F/I-SAPT, see below).^{50–53} RINRUS can also be used to rank 105 fragments via distance-based criteria. Once a ranking scheme is chosen and fragment rank is 106 enumerated, *RINRUS* will algorithmically construct QM-cluster models and provide input 107 files formatted appropriately for several commercial and open-source quantum chemistry 108 software packages. 109

110

This work has two major objectives. First, we analyze how specific residues influence the

enzymatic reaction and contribute to the convergence of *RINRUS*-built QM-cluster mod-111 els of chorismate mutase. Multiple fragment ranking schemes are explored and compared, 112 with models built incrementally, growing by one fragment at a time. Second, we explore 113 QM-cluster modeling in a quasi time-dependent fashion by sampling MD snapshots with 114 refined QM-cluster models to account for conformational averaging. Thermally stable con-115 formational change is one of the most important aspects of regulating protein structure and 116 activity, and conformational sampling of enzymes is typically probed on the micro-second 117 time scale via MD simulations.⁵⁴ We have selected 250 snapshots from a 20 ns MD simula-118 tion of BsCM and processed each with *RINRUS* to obtain 250 different QM-cluster models. 119 The catalytic transition state for each of the 250 QM-cluster models is optimized, and via 120 computation of the connected reactant and product structures, kinetic and thermodynamic 121 data is obtained. 122

123 Methods

All computations were based on the X-ray crystal structure of the *Bacillus subtilis* choris-124 mate mutase taken from PDB entry 2CHT. The 2CHT enzyme is trimeric with three active 125 sites formed at the interface of adjacent monomer chains. The active site of the crystallo-126 graphic A/C chain was used for QM-model construction in this work. Further justification 127 of using the chain A/C interface is provided in the Supporting Information. Hydrogen atoms 128 were added to the enzyme using the *reduce* program.⁵⁵ For all QM-cluster models and MD 129 simulations, the TSA found in the crystallographic active sites was replaced with the native 130 substrate (chorismate). 131

¹³² Incremental QM-cluster model building with *RINRUS*

RINRUS identifies and ranks inter-residue interactions based upon two existing packages
 that compute the RIN and output node/edge information in machine and human-readable

formats. *Probe*⁵⁶ rolls a small sphere over the internal van der Waals surface of a protein structure to identify and classify non-covalent interatomic interactions between fragments of a protein structure; *arpeggio*⁵⁷ uses interatomic distance and angle criteria to identify and classify interactions. Throughout this work, "seed", "substrate", and "ligand" synonymously refer to the chorismate molecule shown in Figure S1.

A good fragment ranking scheme is needed to design reliable QM-cluster models, which 140 is a core feature of the open-source *RINRUS* package.⁵⁸ There are three different fragment 141 ranking schemes being tested in this work. The RINRUS-probe workflow ranks the impor-142 tance of active site fragments based on the number of contact counts between each fragment 143 and the seed. When incrementally building models, fragments (categorized as residue side 144 chains, residue main chains, or solvent water molecules) are added to the model one at a time 145 in order from the fragment with the highest number of contacts with substrate to the low-146 est. While probe parses interaction types into five simple SIF categories, arpeggio classifies 147 fourteen different chemical interaction type, based on the CREDO set of protein-substrate in-148 teractions.⁵⁹ While *arpeggio* also accounts for typical interaction types like hydrogen bonding 149 and hydrophobic contacts, it can also more flexibly account for weaker inter-residue inter-150 actions such as aromatic π -stacking or less common interactions such as halogen bonds. It 151 should be noted that the proximal interactions computed by *arpeggio* are ignored in this 152 study because the focus is on fragments that have recognized intermolecular forces with the 153 chorismate substrate, rather than distance-based metrics. 154

Symmetry adapted perturbation theory (SAPT) has become an increasingly popular approach for computing non-covalent interaction energies between two molecules or fragments. ^{50–52,60–63} SAPT calculations are especially useful in that the interaction energies are readily decomposed into electrostatic, exchange-repulsion, induction, and dispersion components. Functional-group SAPT (F-SAPT)⁵² is an extension of SAPT that provides an effective secondary two-body partition of the SAPT components. This additional partitioning allows computation of interaction energy between a fragment A (in this case study, the chorismate ligand) and user-defined sub-fragments of a fragment B (the various side chain and backbone fragments of the active site). F-SAPT is leveraged to decompose the interaction energy between chorismate and individual residue main chains or side chains, without cutting or capping fragments differently from what is used in the parent QM-cluster models. We will use the F-SAPT interaction energies between chorismate and surrounding residue fragments to rank incremental QM-cluster model building. This work uses the zeroth-order formulation of F-SAPT, F-SAPT0, described by the equation:

$$E_{int} = E_{elec}^{(1)} + E_{exh}^{(1)} + [E_{ind}^{(2)} + E_{exch-ind}^{(2)} + \delta E_{HF}^{(2)}]_{ind} + [E_{disp} + E_{exch-disp}^{(2)}]_{disp}$$
(1)

F-SAPT0 computations employed the jun-cc-pVDZ basis set ^{51,52} for all atoms and frozen core electrons via the *PSI4* v1.3 package.⁶⁴ The jun-cc-pVDZ basis set has been demonstrated to provide reliable SAPT interaction energies.⁶⁵

In recent work, a poor correlation between number of probe contacts and F-SAPT interaction energies was observed. 66,67 We then hypothesized that F-SAPT interaction energies will be a more quantitatively reliable metric for ranking the importance of active site residues. However, SAPT calculations are computationally expensive (days of CPU time) compared to the near-negligible effort required to compute and parse a RIN from *probe* or *arpeggio* ranking (< 20 seconds of CPU time).

QM-cluster models were generated using the *RINRUS* software.⁵⁸ Trimming of residue 178 fragments is performed algorithmically by *RINRUS* depending on if the backbone NH, back-179 bone CO, and/or side chain of a residue has interatomic contacts with chorismate. Where 180 covalent bonds are broken in the trimming procedure (typically across C_{α} atoms), RINRUS 181 automatically adds hydrogen atoms to satisfy carbon valency. We refer throughout to the 182 QM-cluster model that contains all fragments with a quantifiable interaction with the cho-183 rismate ligand as a "maximal model". Trimming details for the maximal model of the X-ray 184 crystal structure active site are shown in Table S1. To maintain the general shape and mimic 185

the semi-rigid character of the protein tertiary structure, all C_{α} atoms, along with the C_{β} atoms of any Arg, Lys, Glu, Met, Trp, and Phe side chains were frozen to their crystallographic positions (if obtained from the X-ray crystal structure) or frozen at their positions in the respective MD frame (if obtained from MD simulation). All chorismate atoms were unconstrained in the QM-cluster model computations.

The QM computations were carried out using the Gaussian16 software package.⁶⁸ The 191 geometries of the models were optimized using density functional theory (DFT) with the 192 B3LYP exchange-correlation functional.^{69,70} The 6-31G(d') basis set was used for N, O, and 193 S⁷¹ and the 6-31G basis set was used for C and H atoms.⁶⁰ The Grimme D3 (Becke-Johnson) 194 dispersion correction (GB3BJ) was also included,⁷² along with a conductor-like polarizable 195 continuum model (CPCM) using UAKS sets of atomic radii, a non-default electronic scaling 196 factor of 1.2, and default cavity parameters for water but with an attenuated dielectric 197 constant of $\varepsilon = 4.^{73,74}$ Transition states were located for the elementary step of the proposed 198 mechanism, and the reactants and products were then confirmed by following the intrinsic 199 reaction coordinate (IRC).¹ The zero-point energies (ZPE) and thermal enthalpy/free energy 200 corrections were calculated at 1 atm and 298.15 K. 201

²⁰² MD trajectory-based QM-cluster models

For the MD simulations, some pre-processing of the X-ray crystal structure was necessary.⁶⁷ Missing residues in the 2CHT X-ray crystal structure were added from the C-terminus using PDB entry 1DBF,⁷⁵ a BsCM structure without substrate or TSA in complex with the protein. The two structures were globally aligned and atomic coordinates from 1DBF were added to the 2CHT structure based upon the point where the two structures begin a common structural alignment. Specifically, residues 1 and 116-127 from 1DBF were added to 2CHT for chain A, residues 1 and 115-127 were added for chain B, and residues 1-2 and 115-127

¹It is important to note that our group employs the "freeze code" scheme in Gaussian16, in which all Hessian elements are zero when two frozen Cartesian coordinates are involved. The phenomenon in which several small magnitude imaginary vibrational frequencies appear in thermochemical analysis does not occur in our treatment of the Hessian matrix.

were added for chain C (with residues 2 and 115-119 of 2CHT chain C being replaced with 210 the corresponding coordinates from 1DBF). Hydrogen atoms were added to this structure 211 via the H++ server using default parameters.⁷⁶ The native substrate chorismate in a pre-212 reactive conformation was used in MD simulations instead of the TSA. The AMBER18 MD 213 package⁷⁷ was used to run the MD simulations, and the AMBER force field ff14SB was used 214 with periodic boundary conditions and a cutoff value of 9 Å for non-bonded interactions. 215 The Antechamber package was employed to parameterize the chorismate substrate with the 216 Generalized Amber Force Field (GAFF).^{77,78} The protonated structure with chorismate was 217 solvated in a cubic 10 Å box of water with the explicit solvent model TIP3P.⁷⁹ The MD 218 model charge was neutralized by adding 9 Na^+ ions.⁸⁰ 219

An energy minimization of the system was first carried out with protein heavy atoms 220 constrained to their crystallographic coordinates using a harmonic positional restraint (k_{pos}) 221 of 200 kcal $\text{mol}^{-1}/\text{Å}^2$ allowing the solvent bath to be initially relaxed and the hydrogen 222 bonding networks to be established. The protein heavy atom constraints were then iteratively 223 relaxed over five 20 ps simulations using Langevin dynamics under constant-temperature, 224 constant-pressure (NPT) conditions at 300 K and 1 atm; the SHAKE algorithm 81 was used 225 to constrain all bonds involving hydrogen atoms for the initial equilibration simulation. The 226 protein was then allowed to move freely for a 20 ns production-level run. The timescale of 227 each frame was 1 ps, for a total of 20,000 frames. The protein RMSDs of MD trajectories 228 were calculated using the *cpptraj* module of AMBER18.⁸² 229

Schemes for selection of frames for the QM-cluster models from MD trajectories

Designing QM-cluster models from a large number of MD frames will allow consideration of conformational influence on kinetic and thermodynamic quantities. Eight schemes are considered in an attempt to cover a diverse sampling of conformations and non-equilibrium structures. From each scheme, 20 to 40 MD frames are selected and then used to construct

a QM-cluster model of the active site. The first scheme considered (\mathbf{S}_1) is perhaps the most 236 common scheme for MD simulation sampling, and involves selecting MD frames at equal 237 intervals over the course of an equilibrated simulation. This approach is effectively random 238 and unbiased. For the next set of schemes $(S_2, S_3, and S_4)$ we chose frames similar to the X-239 ray crystal structure. Furthermore, it may be better to consider only the structural variations 240 of the active site residues rather than of the whole protein, and this idea is incorporated into 241 \mathbf{S}_3 , \mathbf{S}_4 , \mathbf{S}_6 , \mathbf{S}_7 , and \mathbf{S}_8 . For the final set of schemes (\mathbf{S}_5 , \mathbf{S}_6 , \mathbf{S}_7 , and \mathbf{S}_8) frames were grouped 242 by a specific metric and then k-means clustering divided the frames into 3 or 4 clusters. These 243 schemes should increase the structural diversity of QM-cluster model refinement. Again, note 244 that the Chain A/C interface was used to construct the QM-cluster models from each selected 245 MD frame. Detailed frame selection criteria are as follows. 246

 \mathbf{S}_{1} - Twenty frames were selected from the MD simulation at equal intervals of 1,000 ps over the entire 20 ns equilibrated simulation.

S₂ - The RMSD of the backbone atoms (C, O, C_{α}, N, and H) of the entire protein structure compared to the X-ray crystal structure was measured for each frame. Frames with an RMSD within \pm 1 standard deviation (0.76 Å) of the mean RMSD (2.66 Å) were isolated, and a random number generator was used to select 30 frames from this data set.

S₃ - The RMSD of the backbone atoms of a selection of active site residues compared to the X-ray crystal structure was measured for each frame. The subset of active site residues was defined as all residues present in any of the QM-cluster models obtained from S₁: Arg7, Glu78, Arg90, Tyr108, Leu115, Phe57, Ala59, Lys60, Arg63, Val73, Thr74, and Cys75. Frames with an active site backbone RMSD within \pm 1 standard deviation (0.09 Å) of the mean RMSD (0.84 Å) were isolated, and a random number generator was used to select 30 frames from this data set.

 \mathbf{S}_4 - This scheme used the RMSD of the side chain atoms of the active site residues (listed in \mathbf{S}_3) compared to the X-ray crystal structure. Frames with a side chain backbone RMSD within ± 1 standard deviation (0.16 Å) of the mean RMSD (1.66 Å) were isolated, and a ²⁶³ random number generator was used to select 30 frames from this data set.

 S_5 - The RMSD of all heavy (non-hydrogen) atoms of protein and chorismate compared to the X-ray crystal structure was measured for each frame. K-means clustering was used to group the frames into three distinct clusters based on the gap statistic and elbow plots shown in Figure S2, and a random number generator was used to select 10 frames from each of the three clusters.

 \mathbf{S}_{6} - The RMSD of the backbone atoms of only the active site residues (from \mathbf{S}_{3}) compared to the X-ray crystal structure was measured for each frame. Based on analysis of the RMSD using the gap statistic and elbow plots in Figure S3, it became apparent that there is only one unique k-means cluster. We then subdivided the data into four clusters and randomly selected 10 frames from each of the four clusters.

 \mathbf{S}_{7} - This scheme used the RMSD of the side chain atoms of the active site residues compared to the X-ray crystal structure instead of the backbone atoms. Similar to \mathbf{S}_{6} , kmeans clustering with the active site side chain atom RMSD values was not a useful technique (Figure S4). The MD frames were still split into another four arbitrary clusters and randomly selected to provide an unbiased sampling of 40 additional MD frames.

 S_8 - The number of *probe* contacts between chorismate and surrounding residues was measured for each frame of the MD trajectory. K-means clustering grouped the frames into distinct clusters. However, the gap statistics and elbow plots shown in Figure S5 indicate our MD frames are not easily clustered into less than 10 sets, so the clustering is truncated at k = 3. A random number generator was used to randomly select 10 frames from each of the three clusters.

From the eight selection schemes, a total of 250 unique MD frames were chosen and then refined into QM-cluster models generated by *RINRUS*. Note that the composition of the QM-cluster models is not uniform. Each QM-cluster model constructed from MD includes all fragments recognized by the *probe* software as having inter-residue interactions with chorismate *for that specific MD frame*. Interestingly, nearly adjacent and even adjacent frames that were selected by the various schemes showed non-uniform RIN composition [frames 159 (\mathbf{S}_8) and 161 (\mathbf{S}_7); 1218 (\mathbf{S}_6) and 1221 (\mathbf{S}_7); 2473 (\mathbf{S}_8) and 2475 (\mathbf{S}_6); 7603 (\mathbf{S}_6) and 7607 (\mathbf{S}_5); 9748 (\mathbf{S}_6) and 9750 (\mathbf{S}_2); 12378 (\mathbf{S}_6) and 12379 (\mathbf{S}_5), 19719 (\mathbf{S}_2) and 19721 (\mathbf{S}_7)]. Active site RIN composition of the adjacent frames 12378 and 12379 is shown in Table S2.

²⁹⁵ Results and Discussion

²⁹⁶ Building QM-cluster models with different ranking schemes

We began by examining how different schemes to prioritize residue interactions affect the 297 construction of QM-cluster models and the convergence of predicted reaction properties. Full 298 information about model size, model charge, and kinetic and thermodynamic properties are 299 provided for all iterative building schemes in Tables S1 and S3. Using the X-ray crystal 300 structure, 12 QM-cluster models were built by incrementally adding one residue based on 301 their cumulative *probe* contact counts with chorismate within the X-ray crystal structure. 302 The maximal probe-derived model, which includes all residues with any probe contact with 303 the chorismate, contains 203 atoms and is shown in Figure 1. 304

The computed ΔG^{\ddagger} and ΔG_{rxn} values for the Claisen rearrangement reaction are plotted in Figure 2 for the *probe*-based model building scheme. First, the highest *probe*-ranked fragment, Phe57 side chain, is added to the chorismate to constitute the first QM-cluster model, containing 42 atoms and a total model charge of -2. The second QM-cluster model is generated by adding the Arg7 side chain to the first model. The second model now contains 64 atoms and a total model charge of -1. The list of fragments provided in Table S1 gives details about subsequent models, culminating in our "maximal model" of 203 atoms.

For this study, we define models as being converged for a given building scheme if *both* ΔG^{\ddagger} and ΔG_{rxn} of a model *and all subsequent larger models* are within ± 1 (tight convergence criteria) or ± 3 kcal mol⁻¹ (loose convergence criteria) of their reference values (ΔG^{\ddagger} and

 ΔG_{rxn} of the maximal model), respectively. The maximal probe-based RINRUS-designed 315 QM-cluster model has values of $\Delta G^{\ddagger} = 9.1$ kcal mol⁻¹ and $\Delta G_{rxn} = -16.3$ kcal mol⁻¹. As the 316 size of the model increases, the predicted ΔG^{\ddagger} and ΔG_{rxn} become converged at the 155-atom 317 model within the defined metric of convergence of ± 3 kcal mol⁻¹, after Arg63 was added to 318 the 133-atom model. None of the *probe* models converge both ΔG^{\ddagger} and ΔG_{rxn} to within 1.0 319 kcal mol⁻¹ of the maximal QM-cluster model. Overall, the ΔG_{rxn} value qualitatively agrees 320 with other QM-cluster model and QM/MM studies of the chorismate mutase catalytic step 321 that exhibited strongly exergonic reaction free energies.^{5,12,16,22,35,83,84} Below, we will explain 322 why the computed ΔG^{\ddagger} of converged and maximal QM-cluster models is significantly lower 323 than the known experimental value. 324

The Arpequive interaction classification may be more robust than probe in that it is not 325 inherently limited to only the local interatomic contacts. Indeed, all residues identified in 326 the probe ranking scheme are included in the arpeggio ranking scheme in addition to Glu78, 327 as well as the side chains of Val73, where only backbone atoms had been included in the 328 probe-based models as shown in Table S4. Due to differences in which specific residue atoms 329 interact with chorismate, some fragments in the *arpeggio*-based models are trimmed and 330 capped differently. Additional details of the *arpeggio* trimming scheme are shown in Table 331 S1. The maximal *arpeggio*-based model has 245 atoms, which makes it somewhat larger than 332 the maximal *probe*-based model (203 atoms). Figure 3 shows the computed values of ΔG^{\ddagger} 333 and ΔG_{rxn} when employing the *arpeggio*-based RIN to construct the QM-cluster models. 334 The maximal *arpeggio*-based model (used as the reference for convergence tests) has $\Delta G^{\ddagger} =$ 335 10.2 kcal mol⁻¹ and $\Delta G_{rxn} = -16.1$ kcal mol⁻¹. 336

³³⁷ Arpeggio-based models predict satisfactory convergence for ΔG^{\ddagger} and ΔG_{rxn} (Figure 3, ³³⁸ magenta plot) once QM-cluster models are larger than 200 atoms. However, we see a dra-³³⁹ matic disruption of convergence in the reaction free energy when Thr74 is added to form ³⁴⁰ the 136-atom model. The computed ΔG_{rxn} of -36.3 kcal mol⁻¹ is artificially too negative ³⁴¹ because the chorismate translates far out of the active site in the optimized product struc-

Physical Chemistry Chemical Physics

ture. Once Arg90 is added to form the 158-atom model, the chorismate is properly posed; all substrate-arginine hydrogen bonds seen in the maximal model are accounted for. While no *arpeggio*-based models have both ΔG_{rxn} and ΔG^{\ddagger} converged within \pm 1 kcal mol⁻¹ of the maximal model, convergence to the looser \pm 3 kcal mol⁻¹ threshold appears once the 177-atom model is constructed.

One limitation with the *arpequio* scheme is the more frequent occurrence of tie scores 347 for the number of interaction counts. While the number of *probe* contacts can vary over 348 3–4 orders of magnitude as it is linked to the continuous inter-residue surface area, *arpeggio* 349 interaction count scores will be much smaller as values arise from summing the categorical 350 presence/absence of interaction types. The *RINRUS* code does not yet preferentially discern 351 between fragments with tied rankings, so there is no chemical significance to the output 352 ordering for those residues. However, depending on which residues are selected in a tie 353 situation, the convergence of ΔG^{\ddagger} and ΔG_{rxn} values can be affected. 354

In situations where there is a tie in the number of *arpeggio* contact counts, we have 355 manually reordered the *RINRUS* ranking list. First, the number of *arpeggio* contact types 356 are used to break the tie. However, if there are fragments where the number of contact 357 types is also tied, the following convention was used to manually prioritize ranking: charged 358 residues > polar > non-polar residues. In situations where there is still a tie between residues 359 of the same category, the *probe*-based contact count ranking was used to break the tie, as 360 in the case of Thr74 being added before Tyr108. Improvements to the RINRUS code to 361 automatically account for tie-breaking in either *probe* or *arpequio* rankings are currently 362 in development. Further details about probe, arpeggio, and tie-broken arpeggio QM-cluster 363 models is given in Table S1. 364

The tie-broken *arpeggio*-based models (Figure 3, brown plot) show quicker convergence to the ΔG^{\ddagger} value of the maximal model (10.2 kcal mol⁻¹) than in the original *arpeggio* building scheme. Adding Arg90 before Thr74 via the tie-breaking scheme also eliminates the odd disruption of ΔG_{rxn} convergence. However, there is still no model where both the ΔG_{rxn} and ΔG^{\ddagger} are converged to within ± 1 kcal mol⁻¹ of the maximal model. Tie-broken arpeggio-based models have kinetic and thermodynamic values converging to the loose ± 3 kcal mol⁻¹ threshold starting with the 191-atom model. The tie-broken models do not have a significant effect on kinetic or thermodynamic convergence beyond fixing the spurious ΔG^{\ddagger} value. However, avoiding random ordering of fragments that have tied contact count or contact type values seems prudent until an automated approach is available.

A third scheme using quantitative chorismate-residue interaction energies as a ranking 375 method was evaluated. As observed in previous work,^{66,67} the F-SAPT interaction energies 376 prioritize important charged residues which play a key role in transition state stabilization. 377 Our analysis of several proteins (including chorismate mutase) indicated no apparent corre-378 lation between number of probe contact counts and E_{int} between noncovalently interacting 379 biochemical fragments, raising concern that *probe* may de-emphasize residues that have a 380 strong, but directional electrostatic interaction with seed fragments. The substrate-residue 381 interaction energies were computed using F-SAPT0, and a series of 11 QM-cluster mod-382 els were first constructed by adding fragments ranked from largest negative E_{int} with the 383 chorismate substrate to the largest positive E_{int} value (Table S5). It must be recognized 384 that a negative total F-SAPT interaction energy signifies a favorable interaction between a 385 residue fragment and chorismate, while a positive total F-SAPT interaction energy describes 386 a repulsive interaction. Given a dianionic chorismate substrate, it was expected that posi-387 tively charged residues will be ranked first, then polar residues, then nonpolar residues, then 388 negatively charged residues. The initial F-SAPT scheme ranked the four positively charged 389 residues highest; Arg7 is first ($E_{int} = -140.5 \text{ kcal mol}^{-1}$), followed by Arg63 ($E_{int} = -133.2$ 390 kcal mol⁻¹), then Arg90 ($E_{int} = -113.0$ kcal mol⁻¹), and Lys60 ($E_{int} = -78.1$ kcal mol⁻¹). For 391 comparison to a few polar residues, the E_{int} of Tyr108 and Thr74 are -15.5 kcal mol⁻¹ and 392 +10.9 kcal mol⁻¹, respectively. 393

Matching literature precedence, the *probe* and *arpeggio* schemes for constructing QMcluster models frequently de-prioritize charged residues compared to F-SAPT.⁶⁷ While Arg7

is ranked first or second in all three schemes, Arg90 is ranked 3rd by F-SAPT, 5th by 396 probe, and 6th by arpeggio, as illustrated in Tables S4 and S3. Arg63 (1st by F-SAPT) was 397 ranked 8th by probe and 3rd by arpeggio. A visual relationship between probe contacts and 398 the orientation of important charged active site arginine residues can be seen in Figure S6. 399 Phe57 is ranked first in the probe ranking scheme with a total of 288 contacts with chorismate 400 (highlighted in grey in Figure S6), but only has an F-SAPT E_{int} of -2.0 kcal mol⁻¹ and is 401 ranked 10th. Arg63 has only 97 probe (highlighted in yellow) interaction counts, making 402 it the 8th ranked fragment, but again has the second largest negative F-SAPT interaction 403 energy. Charged active site amino acid residues are crucial for both NAC and TSS of 404 the chorismate substrate. Yet Arg7 is the only one of four positively charged residues in 405 the BsCM active site that is ranked consistently high in the probe, arpeggio, and F-SAPT 406 schemes. Our F-SAPT results strongly suggest large residue side chains can be oriented in 407 such a way that they provide strong hydrogen bonds within an active site, but have low RIN 408 contact count values. 409

In a recent analysis of glycine-N-methyltransferase,⁸⁵ we recognized that residues with 410 strongly unfavorable (positive) interaction energies should be ranked higher than residues 411 with near-zero F-SAPT interaction energies. Ranking fragments by $|E_{int}|$ will thus prioritize 412 negatively charged active site fragments that have a large, but unfavorable interaction with 413 the dianionic substrate before fragments that have a small or negligible interaction with the 414 substrate. Semantically, the difference between F-SAPT schemes is subtle, but the quality 415 of QM-cluster models could be substantially affected by this choice. The ΔG^{\ddagger} and ΔG_{rxn} 416 values for the two F-SAPT ranking schemes (signed in magenta and unsigned in brown) are 417 overlaid in Figure 4. Both schemes overlap until the 139-atom model, where Ala59 is next 418 added in the signed scheme and Thr74 is added in the unsigned scheme. 419

As the F-SAPT calculations were derived from the maximal *probe* model, the *probe*, signed and unsigned F-SAPT schemes will all have an equivalent maximal model ($\Delta G^{\ddagger} =$ 9.1 kcal mol⁻¹ and $\Delta G_{rxn} = -16.3$ kcal mol⁻¹) that does not need to be recomputed. The

unsigned F-SAPT ranking scheme exhibits slightly improved convergence over the signed 423 scheme, as the last three unsigned models were within ± 3 kcal mol⁻¹ of the maximal model 424 for both ΔG^{\ddagger} and ΔG_{rxn} values (Table S3). Despite the expectation that QM-cluster models 425 derived from F-SAPT rankings would be optimal, none of the truncated F-SAPT models 426 are within ± 1 kcal mol⁻¹ of both the ΔG^{\ddagger} and ΔG_{rxn} values. Thus, there is little quali-427 tative difference between the largest QM-cluster models built with the F-SAPT, probe, or 428 arpeggio ranking schemes. The F-SAPT scheme is also quite computationally expensive on 429 the front end compared to probe and arpeggio schemes. Generally, only QM-cluster models 430 of chorismate mutase that closely resemble the maximal models are reliable. To ascertain 431 how more liberally truncated models can appropriately reproduce NAC or TSS phenomena. 432 a brute force or combinatorial approach (like the RINRUS-based investigation of Catechol-433 O-Methyltransferase)⁴ would need to be carried out on the chorismate mutase active site. 434

Previous eznymology studies done by our group have shown that B3LYP generally under-435 estimates free energies of activation compared to experiment.^{2-4,85} Accordingly, all ranking 436 schemes had maximal QM-cluster models of the chorismate mutase active site that exhibited 437 ΔG^{\ddagger} values significantly lower than the experimental value⁸⁶ of 15.4 \pm 0.5 kcal mol⁻¹. The 438 maximal F-SAPT / probe-based model predicted an activation free energy of 9.1 kcal mol⁻¹, 439 while the maximal *arpeggio*-based model predicted 10.2 kcal mol⁻¹. QM-cluster models 440 reported by Burschowsky and coauthors at the B3LYP/6-31G(d)//B3LYP/6-311+G(d,p) 441 level of theory arrived at an even lower ΔG^{\ddagger} value for the chorismate mutase catalysis (8.6 442 kcal mol⁻¹).³⁵ It is important to stress that this work is not concerned with accuracy of the 443 QM-cluster models, but focused on understanding how kinetics and thermodynamics are 444 influenced by the decisions involved in QM-cluster model construction. 445

Our lab (and others) are exploring much-needed benchmarks of one-electron basis set and density functional on enzyme models.^{45,46,87–92} To avoid model construction contributing to kinetic and thermodynamic errors, the current study demonstrates that QM-cluster models require, at minimum, over \sim 150 atoms. This lower bound to model size unfortunately guarantees that employing large basis sets and double-hybrid density functionals will be intractable for most production-level exploration of enzyme chemical mechanisms. Ideally, the community will arrive at a consensus on methodological best practices in QMcluster modeling to accurately and efficiently compare to experimental observation. Until then, dispersion-corrected B3LYP with small Pople-style basis sets is an efficient and mostly reliable level of theory for calibrating the error arising from QM-cluster model composition.

456 Building QM-cluster models from MD frames

Next, we explore the impact that fluctuations of residue and substrate positioning can have 457 on both the design of QM-cluster modeling and the resulting kinetic and thermodynamic 458 properties. First, 250 frames from a 20 ns MD simulation of solvated chorismate mutase 459 were sampled to construct maximal QM-cluster models of the active site using *probe* contacts. 460 Structures from MD simulations can be advantageous over crystallographic structures in their 461 unambiguous hydration shells and energy relaxation of the active site structure based on *in* 462 vivo substrates rather than inhibitors or transition state analogues. However, building QM-463 models from MD simulations will incorporate statistical uncertainty, as sampling many MD 464 frames are required to represent the diversity of structural conformations.^{93–95} In particular, 465 we examine three features particularly relevant for QM-cluster modeling that are expected to 466 cause variation in the predicted reaction properties: 1) the number and identity of residues 467 included in the model, 2) the number of waters included in the model, and 3) the statistical 468 ensemble of sampled frames. 460

In plotting the activation and reaction free energies for all 250 MD-derived QM-cluster models (Figure 5, Figure S7, and Table 1), there is a wide range of values wherein the mean activation free energy is 10.3 ± 2.6 kcal mol⁻¹ and the mean free energy of reaction is -15.4 ± 3.4 kcal mol⁻¹. These ranges encompass the converged values observed for QMcluster models built from the X-ray crystal structure, though this is unsurprising given the large standard deviation observed in the ensemble of refined MD frames. The size of the maximal QM-cluster models ranges from 158 to 240 atoms, with the five smallest models
containing only 8 residues and 5 or 6 waters and the largest model containing 13 residues
and 3 waters.

Using *probe* to identify active site fragments, a total of 22 residues were identified as 479 having at least one contact interaction with the substrate in at least one frame over the course 480 of the entire MD simulation. Table S6 shows the mean interaction counts of each identified 481 residue with chorismate. There is precedence that crystal packing leads to an increase in 482 protein-substrate contact counts.^{67,96} However, replacement of the TSA with chorismate in 483 the X-ray crystal structure without a subsequent geometry relaxation does not create steric 484 clashes with the protein, which might have nonphysically amplified the contact counts. As 485 expected, the Arg90 and Arg7 residues have the highest mean contact counts, 116.3 and 486 78.3, respectively. Several residues appear in RINs during the entire MD run with very low 487 mean interaction counts (< 0.02) such as Ala9, Pro117, and residues 242-245. None of these 488 residues have inter-residue contacts with the TSA in the X-ray crystal structure. Pro117 is 489 the only "rare" residue from the entire MD simulation that also appears in the 250 selected 490 frames that were refined to QM-cluster models. The mean interaction counts of residues 491 modeled in the 250 QM-cluster models is similar to those observed in the 20000 RINs of 492 the MD simulation (Table S6). This similarity affirms that the selection schemes used to 493 refine MD frames into QM-cluster models are representative of the entire MD simulation. 494 From Tables S6 and S7, we find that consistently high-ranking active site residues common 495 to probe, arpeggio, and F-SAPT schemes can occasionally be missing entirely from specific 496 MD frames. 497

Surprisingly, QM-cluster models with atypical composition do not necessarily create kinetic or thermodynamic outliers. Frame 394 is the only member of the 250-frame subset to not have any *probe* contacts with the Arg90 side chain. It also does not contain an Arg63 fragment, making it the only QM-cluster model with net -2 charge. The missing fragments result in a spuriously high free energy of activation (see below). The QM-cluster models made from frames 9464, 14007, 16450 are the only three of the 250 that have no *probe* contacts between substrate and Leu115, yet all three have kinetic/thermodynamic properties within the uncertainty range of the total set. Frames with rare residues have a small impact on the overall kinetic and thermodynamic values. For example, the five QM-cluster models that contain Pro117 have mean ΔG^{\ddagger} and ΔG_{rxn} values of 11.2 \pm 2.9 kcal mol⁻¹ and -15.1 \pm 4.3 kcal mol⁻¹, respectively.

Mean probe contact counts of the 250 QM-cluster models arising from MD sampling em-509 phasize charged residues more than the X-ray crystal structure, but interestingly, Figure S8 510 still shows a lack of correlation with F-SAPT $|E_{int}|$ values computed at the X-ray crystal 511 structure. MD-averaged *probe* counts rank the first five residues as Arg90, Arg7, Leu115, 512 Ala59, and Arg63. The Lys60 residue has a mean contact count of only 2.9, but as demon-513 strated earlier, has the 4th-largest $|E_{int}|$ with the substrate. The mean probe contact counts 514 for Leu115 are large (72.3), but it has the smallest absolute F-SAPT interaction energy. 515 Of the uncharged side chain fragments, Tyr108 has the smallest mean *probe* count (28.8)516 and the largest $|E_{int}|$ value. These conflicting results demonstrate how various schemes rank 517 residue importance differently. Great challenges remain in quantifying the impact of specific 518 amino acid fragments on protein-substrate reactivity. 519

The catalytic activity of chorismate mutase is particularly driven by charge stabilization 520 interactions, which might be susceptible to differences in net model charge. Thus, it is of 521 interest to examine whether differences in model charge of QM-cluster models refined from 522 individual MD frames can account for the broad range of activation and reaction free energies 523 observed. Figure 6 shows the distribution of the net model charges for the 250 QM-cluster 524 models compared to the range of ΔG^{\ddagger} and ΔG_{rxn} values for each model. The net charge of 525 our 250 QM-cluster models varies from -2 to +2, with the majority (200 models) having an 526 overall neutral charge. QM-cluster models with a neutral model charge had mean ΔG^{\ddagger} and 527 ΔG_{rxn} values of 10.1 \pm 2.4 and -15.7 \pm 3.3 kcal mol⁻¹, respectively. Only one MD-based 528 QM-cluster model (frame 394) has a -2 net charge model and it provides anomalously high 529

values of ΔG^{\ddagger} and ΔG_{rxn} , 20.0 and -7.7 kcal mol⁻¹, respectively. The outlying energetics of 530 frame 394 are likely due to missing Arg90 and Arg63 fragments, which have proven to be 531 critical for the enzyme catalysis.^{19,31} The 33 QM-cluster models with net +1 charge show 532 the largest range of ΔG^{\ddagger} values, encompassing the highest (19.1 kcal mol⁻¹, frame 4114) 533 and lowest (4.1 kcal mol⁻¹, frame 8310) values. However, the mean energetic values are in 534 reasonable agreement with the complete set, 11.1 ± 3.6 kcal mol⁻¹ for ΔG^{\ddagger} and $-14.0 \pm$ 535 3.3 kcal mol⁻¹ for ΔG_{rxn} . The net charge of the QM-cluster models do not systematically 536 influence the ΔG^{\ddagger} and ΔG_{rxn} values. 537

We have shown the maximal QM-cluster models based on the X-ray crystal structure, 538 from any of our building schemes, are expected to provide kinetics and reaction thermody-539 namics that are reliably converged at a given level of theory (Figure 2). The 250 maximal 540 QM-cluster models derived from MD will have significant variations in the residues that are 541 included in each RIN. This heterogeneity opens the question: when comparing QM-cluster 542 models with the same fragment composition but with different active site conformation 543 and/or relative frozen atom positions, will the computed reaction kinetics and thermody-544 namics show consistent values or large variance? To disentangle model composition from 545 model structure, the dataset is trimmed to only include MD-derived QM-cluster models 546 that have an identical composition. This data filtering ignores distinguishing models with 547 different water molecule positioning. The subset contained 144 total models in 37 different 548 bins (Figure S9). Among the groups of models with identical designs but taken from different 549 snapshots, the groups still show a wide distribution of ΔG^{\ddagger} and ΔG_{rxn} values, with ranges 550 from 4.1 to 16.4 kcal mol⁻¹ for ΔG^{\ddagger} and -28.8 to -6.7 kcal mol⁻¹ for ΔG_{rxn} . No patterns seem 551 to emerge from this data. If the bins in Figure S9 showed a narrow distribution of kinetics 552 and thermodynamics, we would conclude that the observed wide distribution of values in 553 the 250 QM-cluster models manifested from differences in active site fragment composition. 554 However, data in Figure S9 match the large variation of the total set of QM-cluster models 555 refined from the MD simulation. The variation must be due to conformational fluctuation 556

⁵⁵⁷ of active site residues and water molecules during the course of the MD trajectory.

The active site RIN from the X-ray crystal structure contains only a single crystallo-558 graphically resolved water molecule shown to have interactions with the substrate captured 559 by probe. The chorismate mutase active site is small and quite solvent-exposed, but the lack 560 of crystallographically resolved water molecules is unsurprising (though rarely quantified in 561 the literature). The 3D protein structure is typically of greater interest than the poorly 562 resolved oxygen nuclei of the bulk solvent. In contrast, the QM-cluster models generated 563 from the MD simulation encompass a comprehensive hydration shell. In the 250 MD frames 564 selected for QM-cluster model refinement, 2 to 10 water molecules are identified by probe 565 as having an interaction with chorismate (Figure 7). Intriguingly, the *RINRUS*-built QM-566 cluster models of chorismate mutase derived from MD frames have on average 5.6 water 567 molecules interacting with the substrate. Frame 6981 is the only QM-cluster model with 2 568 waters in the active site, and ΔG^{\ddagger} is predicted to be 10.8 kcal mol⁻¹. At the other extreme, 569 the two QM-cluster models with 10 waters have a mean ΔG^{\ddagger} value of 11.3 kcal mol⁻¹. Only 570 29 models total have 2, 3, 8, 9, or 10 water molecules in the RIN. Despite low occurrence in 571 the sampled MD frames, these models have mean predicted ΔG^{\ddagger} and ΔG_{rxn} values of 10.9 572 \pm 3.0 kcal mol⁻¹ and -14.8 \pm 2.9 kcal mol⁻¹, respectively; kinetics and thermodynamics are 573 within uncertainties of the total set of 250 models. The 221 QM-cluster models with 4 to 574 7 water molecules are qualitatively similar, 10.2 \pm 2.6 kcal mol⁻¹ for ΔG^{\ddagger} and -15.4 \pm 3.4 575 kcal mol⁻¹ for ΔG_{rxn} . Clearly, the number of waters in the BsCM active site has minimal 576 influence on the kinetic and thermodynamic properties of QM-cluster models. However, the 577 inclusion of any type of water network at the active site-solvent boundary in our MD-derived 578 QM-cluster models may be a factor in the ~ 2 kcal mol⁻¹ higher free energies of activation 579 observed compared to models constructed from the X-ray crystal structure. 580

Finally, we analyze groupings of the statistical ensemble of QM-cluster models (Table 1), which showed minimal statistical difference with the overall mean kinetic and thermodynamic values ($\Delta G^{\ddagger} = 10.3 \pm 2.6$ kcal mol⁻¹ and $\Delta G_{rxn} = -15.4 \pm 3.4$ kcal mol⁻¹). Schemes labeled ⁵⁸⁴ \mathbf{XS}_2 to \mathbf{XS}_8 , are expanded versions of \mathbf{S}_2 to \mathbf{S}_8 , and include all frames from the 250 QM-⁵⁸⁵ cluster models that fit the criteria of each Scheme. For example, \mathbf{XS}_2 includes the 30 frames ⁵⁸⁶ from \mathbf{S}_2 and the additional 118 frames from the 250 frame set that have an RMSD within 0.76 ⁵⁸⁷ Å of the mean backbone atom RMSD. Kinetic and thermodynamic results for the expanded ⁵⁸⁸ schemes are given in Table 2.

The first scheme, S_1 , contains 20 frames and should be representative of a random and 589 unbiased distribution of activation and reaction free energies over the course of the entire 590 MD simulation. Mean ΔG^{\ddagger} and ΔG_{rxn} values of the 20 frames used in \mathbf{S}_1 are lower than 591 the total set, but in reasonable agreement. Establishing that k-means clustering of \mathbf{S}_6 and 592 \mathbf{S}_7 was invalid, these two schemes also represent a random selection of frames. We combined 593 the frames of S_1 , S_6 , and S_7 (100 total) into an expanded Scheme ($S_1 + S_6 + S_7$) in Table 594 2. Interestingly, the kinetic and thermodynamic values of $S_1 + S_6 + S_7$ are within 0.10 595 kcal mol⁻¹ of the entire data set. This improved agreement suggests 20 randomly selected 596 frames (8% of the total data set) may not be a robust amount. Since most of the expanded 597 schemes have mean kinetic and thermodynamic values very similar to the total set of 250 598 MD frames, then a sample of 100 frames (40%) of the data points in total set) may be an 599 upper bound needed to emulate the total set. 600

The next sets of schemes (\mathbf{S}_2 , \mathbf{S}_3 , and \mathbf{S}_4), take into account the fluctuation of the active site residues and discard MD frames geometrically dissimilar to the X-ray crystal structure. All three schemes predict mean ΔG^{\ddagger} values slightly lower than the entire dataset. \mathbf{S}_2 and \mathbf{S}_4 mean ΔG_{rxn} values are lower than the total mean, while the \mathbf{S}_4 mean is slightly higher than \mathbf{S}_2 and \mathbf{S}_3 . The extended \mathbf{XS}_3 and \mathbf{XS}_4 schemes (Table 2) are closer to the total mean statistics than \mathbf{XS}_2 .

⁶⁰⁷ The \mathbf{S}_5 scheme used k-means clustering of the RMSDs (ranging from 1.46 to 4.22 Å ⁶⁰⁸ shown in Table S8) of the active site residues to group similar frames into clusters. The ⁶⁰⁹ three clusters for \mathbf{S}_5 are ordered from largest centroid RMSD value (\mathbf{S}_5 - \mathbf{C}_1) to the lowest ⁶¹⁰ (\mathbf{S}_5 - \mathbf{C}_3). The (\mathbf{S}_5 - \mathbf{C}_1) and (\mathbf{S}_5 - \mathbf{C}_3) clusters have nearly the same mean $\Delta \mathbf{G}^{\ddagger}$ value, below ⁶¹¹ the mean ΔG^{\ddagger} value of the total data set. The (\mathbf{S}_5 - \mathbf{C}_2) cluster in contrast, is higher (11.0 ⁶¹² kcal mol⁻¹) than the total data set. Values of ΔG_{rxn} become less negative as the centroid ⁶¹³ RMSD value decreases from \mathbf{C}_3 to \mathbf{C}_1 , and the extended Scheme \mathbf{XS}_5 - \mathbf{C}_3 to \mathbf{XS}_5 - \mathbf{C}_1 follows ⁶¹⁴ the same pattern. Scheme \mathbf{S}_5 gives a mean ΔG^{\ddagger} value closest to that of the total data set. ⁶¹⁵ The mean ΔG_{rxn} value for \mathbf{S}_5 is also quite close, but effectively random sampling in \mathbf{S}_6 and ⁶¹⁶ \mathbf{S}_7 give a slightly better match to the total set.

The last scheme (\mathbf{S}_8) classified MD frames with k-means clustering according to probe 617 interatomic contacts between the chorismate ligand and surrounding residues. All three 618 clusters of S_8 predicted the mean ΔG^{\ddagger} value to be 0.46 - 1.15 kcal mol⁻¹ more negative than 619 the mean of the total dataset. The statistics of the expanded clusters of \mathbf{XS}_8 are much closer 620 to the total dataset. Notwithstanding, the largest magnitude differences between any frame 621 selection scheme and mean values of the 250 QM-cluster models are 0.62 kcal mol^-1 for ΔG^{\ddagger} 622 and 1.56 kcal mol⁻¹ for ΔG_{rxn} . For the expanded schemes, the largest absolute differences 623 decrease to 0.17 kcal mol⁻¹ for mean ΔG^{\ddagger} and 0.78 kcal mol⁻¹ for mean ΔG_{rxn} . 624

In summary, efforts to find a subset of MD frame selection schemes that best reflect 625 the kinetic and thermodynamic values of a large statistical ensemble were inconclusive, yet 626 promising. All eight schemes shown in Table 1, with 20 - 40 MD frames in each refined 627 to QM-cluster models, give reasonable approximations to the larger set of 250 MD frames. 628 Expanded schemes with 69 - 186 selected MD frames give mean values even closer to the 629 larger data set. Schemes employing k-means clustering to partition frames via structural 630 metrics did not perform better than schemes with completely random selected MD frames. 631 However, the QM-cluster models were built from one of three trimeric BsCM active sites (the 632 Chain A/C interface) that exhibited the least conformational fluctuation during the course 633 of the 20 ns MD simulation. Machine-learned selection procedures like k-means clustering 634 may be more beneficial for enzymes with more disordered regions or that undergo substantial 635 conformational changes during the simulation time. 636

637 Conclusions

Over 50 QM-cluster models of *Bacillus subtilis* chorismate mutase based on the X-ray crystal structure, and an additional 250 QM-cluster models obtained from sampling MD frames were extensively tested with the *RINRUS* software package being developed by our group. *RINRUS* automatically identifies and trims fragments that interact with a substrate, allowing quantitative and reproducible analysis of how the active site fragments affect enzyme catalysis.

The smallest QM-cluster models built with probe, arpeggio and F-SAPT schemes showed 644 critical differences in how the kinetic and thermodynamics were altered by subsequent addi-645 tion of residues. Once model building schemes approach the size of the "maximal" model, 646 all three iterative schemes behaved similarly. We have seen some methodological issues with 647 the *arpequio* ranking scheme where ties can occur in the number of contact counts or con-648 tact types. The tie issue in *arpequio* was resolved manually, and fixed an outlying reaction 649 free energy that was observed in one of the smaller QM-cluster models. The solution to tie 650 interaction counts or types will require more automation to be incorporated into RINRUS 651 functionality. 652

The F-SAPT-based interaction energies highlight the importance of active site charged 653 residues. We recommend always using absolute values of F-SAPT interaction energies to 654 rank active site fragments in QM-cluster model construction. Rankings via signed inter-655 action energies may de-prioritize important active site fragments that exhibit electrostatic 656 repulsion with a substrate. The unsigned F-SAPT ranking scheme showed slight improve-657 ment of convergence compared to probe and arpeggio schemes, but no truncated models in 658 any of the schemes converged to within 1 kcal mol⁻¹ of the respective maximal models. We 659 again validate that there is no correlation between the number of *probe* contact counts and 660 E_{int} obtained from F-SAPT computations. More case studies are required to determine 661 if the small performance differences between schemes is related to the compact size of the 662 BsCM active site. Nevertheless, probe-based models, arpeggio and F-SAPT maximal models 663

⁶⁶⁴ are similar, providing evidence that the largest *RINRUS*-generated QM-cluster models are ⁶⁶⁵ robust and reliable.

As is widely known in the community and seen in our previous studies, B3LYP-GD3BJ with small Pople-style basis sets and implicit solvation with CPCM systematically underestimates the free energies of activation of enzyme mechanisms compared to the experimental kinetic value. A focus on the quality of the quantum chemical level of theory is purposefully avoided in this work, to instead efficiently provide insight about QM-cluster model building approaches.

The crystallographic protein structure was then solvated within an explicit water bath 672 and, over a 20 ns equilibrated MD simulation, 250 frames were selected to construct 250 QM-673 cluster models of the active site. The proposed catalyzed Claisen rearrangement mechanism 674 was computed for all QM-cluster 250 models, and the reaction thermodynamics are observed 675 to fluctuate, with the activation free energy spanning 10.34 ± 2.62 kcal mol⁻¹ and the reaction 676 free energy spanning -15.38 ± 3.40 kcal mol⁻¹. The variation is shown to be primarily 677 due to the changes in residue/solvent/ligand positioning and conformation that occur over 678 the MD simulation, rather than differences in residue composition among the models. For 679 example, we noted that some active site residues highly ranked in the *probe*, *arpeqqio*, and 680 F-SAPT schemes can be absent from specific MD frames when the residues shift to different 681 placements, but the computed kinetic and thermodynamic properties of those complexes 682 can still be reasonable given the QM-cluster model is suitably constructed. Furthermore, 683 while the catalytic mechanism is largely derived from charge stabilization interactions, and 684 we thus might expect the QM-cluster models to be very sensitive to changes in net model 685 charge. The results show most of the variation in ΔG^{\ddagger} and ΔG_{rxn} values is largely among 686 models with neutral net charge and a general insensitivity in predicted values with net charge 687 between ± 1 was observed. The active site interface with bulk solvent is shown to influence 688 kinetics and thermodynamics of the QM-cluster models. However, the number of explicit 689 water molecules included in the models appear to be inconsequential. 690

Collectively, results from the MD to QM-cluster model refinement point to the changing 691 molecular positioning rather than model composition as the main source for changing reac-692 tion thermodynamics over the sampled times. We attempted to trace the thermodynamic 693 differences to simple, easily quantifiable structural differences among the models, specifically 694 by grouping models based upon RMSDs in backbone or side chain atoms. Ultimately, none of 695 the metrics were better than random selection for acceptably sampling a statistical ensemble 696 of structures. A more multifaceted technique will be required to efficiently cluster MD frames 697 for QM-cluster model refinement, especially if the enzyme undergoes major conformational 698 changes during the MD simulation. 690

This study exemplifies diverse features of the *RINRUS* toolkit by comparing the struc-700 tural variation between X-ray crystal structure-based models and MD-based models of bacte-701 rial chorismate mutase. Composition of QM-cluster models, or the QM region of a QM/MM 702 model is an essential part of reliability and accuracy in computational enzymology. For 703 far too long, a lack of automated model building techniques and software has hampered 704 advancement of the field as well as the reproducibility of seminal work. Here, QM-cluster 705 modeling provided insight into the enzymatic activity of chorismate mutase by connecting 706 the model composition, the contribution of charged residues, the influence of explicit solvent 707 water molecules, and positioning and orientation of active site residues to the computed 708 kinetic and thermodynamic values. Accompanying data can be easily used to perform fur-709 ther cheminformatic analysis or to calibrate accuracy with more reliable quantum chemistry 710 methodologies; *RINRUS* was designed with reproducibility as a core feature. 711

$_{^{712}}$ Acknowledgments

This work is supported by the National Science Foundation (NSF) CAREER BIO-1846408
(for NJD and DAA), the Department of Energy (DOE) BES SBIR DE-SC0021568 (for NJD),
the National Institute of General Medical Sciences of the National Institutes of Health under

⁷¹⁶ award number 1R35GM145206-01 (for QC), and the NSF Graduate Research Fellowship
⁷¹⁷ Program under Grant No. 1451514 (for TJS). The authors are grateful for the support
⁷¹⁸ provided by the University of Memphis High Performance Computing Center.

719 Author Contributions

Donatus A. Agbaglo: Conceptualization (equal); Formal analysis (lead); Investigation 720 (lead); Methodology (equal); Visualization (lead); Writing – original draft (lead); Writing 721 - review & editing (equal). Thomas J. Summers: Conceptualization (equal); Formal 722 analysis (supporting); Methodology (equal); Visualization (supporting); Writing – original 723 draft (supporting); Writing – review & editing (equal). Qianyi Cheng: Writing – review 724 & editing (equal); Investigation (supporting). Nathan J. DeYonker: Conceptualization 725 (equal); Resources (lead); Writing – original draft (supporting); Writing – review & editing 726 (equal). 727

728 References

- (1) Kraut, D. A.; Carroll, K. S.; Herschlag, D. Challenges in enzyme mechanism and energetics. Annu. Rev. Biochem. 2003, 72, 517–571.
- (2) Cheng, Q.; DeYonker, N. J. Acylation and deacylation mechanism and kinetics of penicillin G reaction with Streptomyces R61 DD-peptidase. J. Comput. Chem. 2020, 41, 1685–1697.

(3) Cheng, Q.; DeYonker, N. J. QM-Cluster Model Study of the Guaiacol Hydrogen Atom
Transfer and Oxygen Rebound with Cytochrome P450 Enzyme GcoA. J. Phys. Chem.
B 2021, 125, 3296–3306.

⁷³⁷ (4) Summers, T. J.; Cheng, Q.; Palma, M. A.; Pham, D.-T.; Kelso III, D. K.; Web-

738		ster, C. E.; DeYonker, N. J. Cheminformatic quantum mechanical enzyme model design:
739		A catechol-O-methyltransferase case study. Biophys. J. 2021, 120, 3577–3587.
740	(5)	Woodcock, H. L.; Hodošček, M.; Sherwood, P.; Lee, Y. S.; Schaefer III, H. F.;
741		Brooks, B. R. Exploring the quantum mechanical/molecular mechanical replica path
742		method: a pathway optimization of the chorismate to prephenate Claisen rearrange-
743		ment catalyzed by chorismate mutase. Theor. Chem. Acc. 2003, 109, 140–148.
744	(6)	Lee, Y. S.; Worthington, S. E.; Krauss, M.; Brooks, B. R. Reaction mechanism of choris-
745		mate mutase studied by the combined potentials of quantum mechanics and molecular
746		mechanics. J. Phys. Chem. B 2002, 106, 12059–12065.
747	(7)	Wiest, O.; Houk, K. On the transition state of the chorismate-prephenate rearrange-
748		ment. J. Org. Chem. 1994, 59, 7582–7584.
749	(8)	Wiest, O.; Houk, K. Stabilization of the transition state of the chorismate-prephenate
750		rearrangement: An ab initio study of enzyme and antibody catalysis. J. Am. Chem.
751		Soc. 1995, 117, 11628–11639.
752	(9)	Lyne, P. D.; Mulholland, A. J.; Richards, W. G. Insights into chorismate mutase catal-
753		ysis from a combined QM/MM simulation of the enzyme reaction. J. Am. Chem. Soc.
754		1995 , <i>117</i> , 11345–11350.
755	(10)	Davidson, M.; Guest, J.; Simon Craw, J.; Hillier, I.; Vincent, M. Conformational and
756		solvation aspects of the chorismate–prephenate rearrangement studied by ab initio elec-
757		tronic structure and simulation methods. J. Chem. Soc., Perkin Trans. 2 1997, 1395-
758		1400.
759	(11)	Hall, R. J.; Hindle, S. A.; Burton, N. A.; Hillier, I. H. Aspects of hybrid QM/MM
760		calculations: the treatment of the $\rm QM/MM$ interface region and geometry optimization

with an application to chorismate mutase. J. Comput. Chem. 2000, 21, 1433–1441.

30

762	(12)	Khanjin, N. A.; Snyder, J. P.; Menger, F. Mechanism of chorismate mutase:	Contri-
763		bution of conformational restriction to catalysis in the Claisen rearrangement.	J. Am.
764		Chem. Soc. 1999, 121, 11831–11846.	

- (13) Madurga, S.; Vilaseca, E. SCRF study of the conformational equilibrium of chorismate
 in water. *Phys. Chem. Chem. Phys.* 2001, *3*, 3548–3554.
- (14) Crespo, A.; Scherlis, D. A.; Martí, M. A.; Ordejón, P.; Roitberg, A. E.; Estrin, D. A. A
 DFT-based QM-MM approach designed for the treatment of large molecular systems:
 Application to chorismate mutase. J. Phys. Chem. B 2003, 107, 13728–13736.
- (15) Dewick, P. M. The biosynthesis of shikimate metabolites. *Nat. Prod. Rep.* 1995, *12*, 101–133.
- Andrews, P.; Smith, G. D.; Young, I. Transition-state stabilization and enzymic catalysis. Kinetic and molecular orbital studies of the rearrangement of chorismate to prephenate. *Biochem.* 1973, *12*, 3492–3498.
- (17) Gorisch, H. Mechanism of chorismate mutase reaction. *Biochem.* **1978**, *17*, 3700–3705.
- (18) Freindorf, M.; Tao, Y.; Sethio, D.; Cremer, D.; Kraka, E. New mechanistic insights into
 the Claisen rearrangement of chorismate-a Unified Reaction Valley Approach study. *Mol. Phys.* 2019, 117, 1172–1192.
- (19) Chook, Y. M.; Ke, H.; Lipscomb, W. N. Crystal structures of the monofunctional
 chorismate mutase from Bacillus subtilis and its complex with a transition state analog. *Proc. Natl. Acad. Sci. USA* **1993**, *90*, 8600–8603.
- (20) Bruice, T. C. A view at the millennium: the efficiency of enzymatic catalysis. Acc.
 Chem. Res. 2002, 35, 139–148.
- (21) Claeyssens, F.; Ranaghan, K. E.; Lawan, N.; Macrae, S. J.; Manby, F. R.; Harvey, J. N.;
- ⁷⁸⁵ Mulholland, A. J. Analysis of chorismate mutase catalysis by QM/MM modelling of

- enzyme-catalysed and uncatalysed reactions. Org. & Biomol. Chem. 2011, 9, 1578–
 1590.
- ⁷⁸⁸ (22) Burschowsky, D.; van Eerde, A.; Ökvist, M.; Kienhöfer, A.; Kast, P.; Hilvert, D.; Kren⁷⁸⁹ gel, U. Electrostatic transition state stabilization rather than reactant destabilization
 ⁷⁹⁰ provides the chemical basis for efficient chorismate mutase catalysis. *Proc. Natl. Acad.*⁷⁹¹ Sci. 2014, 111, 17516–17521.
- (23) Zhang, X.; Zhang, X.; Bruice, T. C. A definitive mechanism for chorismate mutase. *Biochem.* 2005, 44, 10443–10448.
- (24) Lamb, A. L. Pericyclic reactions catalyzed by chorismate-utilizing enzymes. *Biochem.*2011, 50, 7476–7483.
- ⁷⁹⁶ (25) Strajbl, M.; Shurki, A.; Kato, M.; Warshel, A. Apparent NAC effect in chorismate
 ⁷⁹⁷ mutase reflects electrostatic transition state stabilization. J. Am. Chem. Soc. 2003,
 ⁷⁹⁸ 125, 10228–10237.
- (26) Galopin, C. C.; Zhang, S.; Wilson, D. B.; Ganem, B. On the mechanism of chorismate
 mutases: clues from wild-type E. coli enzyme and a site-directed mutant related to
 yeast chorismate mutase. *Tetrahedron Lett.* **1996**, *37*, 8675–8678.
- (27) Liu, D. R.; Cload, S. T.; Pastor, R. M.; Schultz, P. G. Analysis of active site residues
 in Escherichia coli chorismate mutase by site-directed mutagenesis. J. Am. Chem. Soc. **1996**, 118, 1789–1790.
- (28) Zhang, S.; Kongsaeree, P.; Clardy, J.; Wilson, D. B.; Ganem, B. Site-directed muta genesis of monofunctional chorismate mutase engineered from the E. coli P-protein.
 Bioorg. Med. Chem. 1996, 4, 1015–1020.
- (29) Schnappauf, G.; Sträter, N.; Lipscomb, W. N.; Braus, G. H. A glutamate residue in

80	9	the catalytic center of the yeast chorismate mutase restricts enzyme activity to acidic
81	D	conditions. Proc. Natl. Acad. Sci. USA 1997, 94, 8491–8496.
81	1 (30)	Lassila, J. K.; Keeffe, J. R.; Kast, P.; Mayo, S. L. Exhaustive mutagenesis of six sec-
81	2	ondary active-site residues in Escherichia coli chorismate mutase shows the importance
81	3	of hydrophobic side chains and a helix N-capping position for stability and catalysis.
81	4	Biochem. 2007, 46, 6883–6891.
81	5 (31)	Kienhöfer, A.; Kast, P.; Hilvert, D. Selective stabilization of the chorismate mutase
81	6	transition state by a positively charged hydrogen bond donor. J. Am. Chem. Soc. 2003,
81	7	125, 3206 - 3207.
81	B (32)	Hur, S.; Bruice, T. C. Enzymes do what is expected (chalcone isomerase versus choris-
81	9	mate mutase). J. Am. Chem. Soc. 2003, 125, 1472–1473.
82	o (33)	Zhang, X.; Bruice, T. C. The proficiency of a thermophilic chorismate mutase enzyme
82	1	is solely through an entropic advantage in the enzyme reaction. Proc. Natl. Acad. Sci.
82	2	2005 , <i>102</i> , 18356–18360.
82	3 (34)	Shurki, A.; Štrajbl, M.; Villa, J.; Warshel, A. How much do enzymes really gain by
82	4	restraining their reacting fragments? J. Am. Chem. Soc. 2002, 124, 4097–4107.
82	5 (35)	Burschowsky, D.; Krengel, U.; Uggerud, E.; Balcells, D. Quantum chemical modeling of
82	5	the reaction path of chorismate mutase based on the experimental substrate/product $% \mathcal{A}$
82	7	complex. FEBS. Open Bio. 2017, 7, 789–797.
82	₃ (36)	Mulholland, A. J. Computational enzymology: modelling the mechanisms of biological

- catalysts. 2008.
- (37) Ahmadi, S.; Barrios Herrera, L.; Chehelamirani, M.; Hostaš, J.; Jalife, S.;
 Salahub, D. R. Multiscale modeling of enzymes: QM-cluster, QM/MM, and
 QM/MM/MD: A tutorial review. *Intl. J. Quant. Chem.* 2018, *118*, e25558.

33

- (38) Guo, H.; Rao, N. Chorismate-Mutase-Catalyzed Claisen Rearrangement. The Claisen
 Rearrangement: Methods and Applications 2007, 1–23.
- (39) Kulik, H. J.; Zhang, J.; Klinman, J. P.; Martínez, T. J. How large should the QM
 region be in QM/MM calculations? The case of catechol O-methyltransferase. J. Phys. *Chem. B* 2016, 120, 11381–11394.
- (40) Kulik, H. J.; Luehr, N.; Ufimtsev, I. S.; Martinez, T. J. Ab initio quantum chemistry
 for protein structures. J. Phys. Chem. B 2012, 116, 12501–12509.
- (41) Karelina, M.; Kulik, H. J. Systematic quantum mechanical region determination in
 QM/MM simulation. J. Chem. Theory Comput. 2017, 13, 563–576.
- ⁸⁴² (42) Sumner, S.; Soderhjelm, P.; Ryde, U. Effect of geometry optimizations on QM-cluster
 ⁸⁴³ and QM/MM studies of reaction energies in proteins. J. Chem. Theory Comput. 2013,
 ⁸⁴⁴ 9, 4205–4214.
- (43) Hu, L.; Eliasson, J.; Heimdal, J.; Ryde, U. Do quantum mechanical energies calculated
 for small models of protein-active sites converge? J. Phys. Chem. A 2009, 113, 11793–
 11800.
- (44) Delcey, M. G.; Pierloot, K.; Phung, Q. M.; Vancoillie, S.; Lindh, R.; Ryde, U. Accurate
 calculations of geometries and singlet-triplet energy differences for active-site models
 of [NiFe] hydrogenase. *Phys. Chem. Chem. Phys.* 2014, 16, 7927–7938.
- (45) Wappett, D. A.; Goerigk, L. A guide to benchmarking enzymatically catalysed reactions: the importance of accurate reference energies and the chemical environment. *Theor. Chem. Acc.* 2021, 140, 1–15.
- (46) Kromann, J. C.; Christensen, A. S.; Cui, Q.; Jensen, J. H. Towards a barrier height
 benchmark set for biologically relevant systems. *PeerJ* 2016, *4*, e1994.

856	(47)	Di Paola, L.; De Ruvo, M.; Paci, P.; Santoni, D.; Giuliani, A. Protein contact networks:
857		an emerging paradigm in chemistry. Chem. Rev. 2013, 113, 1598–1613.
858	(48)	Doncheva, N. T.; Klein, K.; Domingues, F. S.; Albrecht, M. Analyzing and visualizing
859		residue networks of protein structures. Trends Biochem. Sci. 2011, 36, 179–182.
860	(49)	Deng, Z.; Chuaqui, C.; Singh, J. Structural interaction fingerprint (SIFt): a novel
861		method for analyzing three-dimensional protein-ligand binding interactions. J. Med.
862		Chem. 2004 , 47, 337–344.
863	(50)	Szalewicz, K. Symmetry-adapted perturbation theory of intermolecular forces. $Wiley$
864		Interdiscip. Rev. Comput. Mol. Sci. 2012, 2, 254–272.
865	(51)	Parker, T. M.; Burns, L. A.; Parrish, R. M.; Ryno, A. G.; Sherrill, C. D. Levels of
866		symmetry adapted perturbation theory (SAPT). I. Efficiency and performance for in-
867		teraction energies. J. Chem. Phys. 2014, 140, 094106.
868	(52)	Parrish, R. M.; Parker, T. M.; Sherrill, C. D. Chemical assignment of symmetry-adapted
869		perturbation theory interaction energy components: the functional-group SAPT parti-
870		tion. J. Chem. Theory Comput. 2014, 10, 4417–4431.
871	(53)	Parrish, R. M.; Thompson, K. C.; Martinez, T. J. Large-scale functional group
872		symmetry-adapted perturbation theory on graphical processing units. J. Chem. Theory $% \mathcal{L}^{(1)}$
873		Comput. 2018 , <i>14</i> , 1737–1753.
874	(54)	Heilmann, N.; Wolf, M.; Kozlowska, M.; Sedghamiz, E.; Setzler, J.; Brieg, M.; Wen-
875		zel, W. Sampling of the conformational landscape of small proteins with Monte Carlo
876		methods. Sci. Rep. 2020 , 10, 1–13.

⁸⁷⁷ (55) Word, J. M.; Lovell, S. C.; Richardson, J. S.; Richardson, D. C. Asparagine and glu⁸⁷⁸ tamine: using hydrogen atom contacts in the choice of side-chain amide orientation. J.
⁸⁷⁹ Mol. Biol. 1999, 285, 1735–1747.

35

- (56) Word, J. M.; Lovell, S. C.; LaBean, T. H.; Taylor, H. C.; Zalis, M. E.; Presley, B. K.;
 Richardson, J. S.; Richardson, D. C. Visualizing and quantifying molecular goodnessof-fit: small-probe contact dots with explicit hydrogen atoms. J. Mol. Biol. 1999, 285,
 1711–1733.
- ⁸⁸⁴ (57) Jubb, H. C.; Higueruelo, A. P.; Ochoa-Montaño, B.; Pitt, W. R.; Ascher, D. B.; Blundell, T. L. Arpeggio: a web server for calculating and visualising interatomic interactions
 ⁸⁸⁵ in protein structures. J. Mol. Biol. 2017, 429, 365–371.
- (58) Cheng, Q.; DeYonker, N. J.; Summers, T. J.; Agbaglo, D. A.; Suhagia, T.; Palma, M. A.
 GitHub natedey/RINRUS: Residue Interaction Network ResidUe Selector (RINRUS)
 public release. https://github.com/natedey/RINRUS (accessed 2022-09-14). 2022.
- (59) Schreyer, A.; Blundell, T. CREDO: a protein-ligand interaction database for drug
 discovery. *Chem. Biol. Drug Des.* 2009, 73, 157–167.
- (60) Hehre, W. J.; Ditchfield, R.; Pople, J. A. Self—consistent molecular orbital methods.
 XII. Further extensions of Gaussian—type basis sets for use in molecular orbital studies
 of organic molecules. J. Chem. Phys. 1972, 56, 2257–2261.
- ⁸⁹⁵ (61) Misquitta, A. J.; Podeszwa, R.; Jeziorski, B.; Szalewicz, K. Intermolecular potentials
 ⁸⁹⁶ based on symmetry-adapted perturbation theory with dispersion energies from time⁸⁹⁷ dependent density-functional calculations. J. Chem. Phys. 2005, 123, 214103.
- ⁸⁹⁸ (62) Bukowski, R.; Cencek, W.; Jankowski, P.; Jeziorska, M.; Jeziorski, B.; Kucharski, S.;
 ⁸⁹⁹ Lotrich, V.; Misquitta, A.; Moszynski, R.; Patkowski, K., et al. SAPT2008: An *ab*⁹⁰⁰ *initio* program for many-body symmetry-adapted perturbation theory calculations of
 ⁹⁰¹ intermolecular interaction energies. University of Delaware and University of Warsaw
 ⁹⁰² 2008,
- 903 (63) Spronk, S. A.; Glick, Z. L.; Metcalf, D. P.; Sherrill, C. D.; Cheney, D. L. A quantum

904	chemical interaction energy dataset for a	ccurately modeling	protein-ligand	interactions
905	Sci. Data 2023 , 10, 619.			

- (64) Parrish, R. M.; Burns, L. A.; Smith, D. G.; Simmonett, A. C.; DePrince III, A. E.;
 Hohenstein, E. G.; Bozkaya, U.; Sokolov, A. Y.; Di Remigio, R.; Richard, R. M.,
 et al. Psi4 1.1: An open-source electronic structure program emphasizing automation,
 advanced libraries, and interoperability. J. Chem. Theory Comput. 2017, 13, 3185–
 3197.
- (65) Burns, L. A.; Faver, J. C.; Zheng, Z.; Marshall, M. S.; Smith, D. G.; Vanommeslaeghe, K.; MacKerell Jr, A. D.; Merz Jr, K. M.; Sherrill, C. D. The BioFragment
 Database (BFDb): An open-data platform for computational chemistry analysis of
 noncovalent interactions. J. Chem. Phys. 2017, 147, 161727.
- (66) Summers, T. J.; Daniel, B. P.; Cheng, Q.; DeYonker, N. J. Quantifying Inter-Residue
 Contacts through Interaction Energies. J. Chem. Inf. Model 2019, 59, 5034–5044.
- ⁹¹⁷ (67) Summers, T. J.; Hemmati, R.; Miller, J. E.; Agbaglo, D. A.; Cheng, Q.; DeYonker, N. J.
 ⁹¹⁸ Evaluating the Active Site-Substrate Interplay Between X-ray Crystal Structure and
 ⁹¹⁹ Molecular Dynamics in Chorismate Mutase. J. Chem. Phys. 2023, 158, 065101.
- (68) Frisch, M.; Trucks, G.; Schlegel, H.; Scuseria, G.; Robb, M.; Cheeseman, J.; Scalmani, G.; Barone, V.; Petersson, G.; Nakatsuji, H., et al. Gaussian 16. 2016.
- (69) Beck, A. D. Density-functional thermochemistry. III. The role of exact exchange. J. *Chem. Phys.* 1993, 98, 5648–6.
- (70) Lee, C.; Yang, W.; Parr, R. G. Development of the Colle-Salvetti correlation-energy
 formula into a functional of the electron density. *Phys. Rev. B* 1988, *37*, 785.
- ⁹²⁶ (71) Petersson, G.; Al-Laham, M. A. A complete basis set model chemistry. II. Open-shell

systems and the total energies of the first-row atoms. J. Chem. Phys. 1991, 94, 6081–
6090.

- (72) Grimme, S.; Ehrlich, S.; Goerigk, L. Effect of the damping function in dispersion corrected density functional theory. J. Comput. Chem. 2011, 32, 1456–1465.
- (73) Barone, V.; Cossi, M. Quantum calculation of molecular energies and energy gradients
 in solution by a conductor solvent model. J. Phys. Chem. A 1998, 102, 1995–2001.
- (74) Cossi, M.; Rega, N.; Scalmani, G.; Barone, V. Energies, structures, and electronic
 properties of molecules in solution with the C-PCM solvation model. J. Comput. Chem.
 2003, 24, 669–681.
- (75) Worthington, S. E.; Roitberg, A. E.; Krauss, M. An MD/QM study of the chorismate
 mutase-catalyzed Claisen rearrangement reaction. J. Phys. Chem. B 2001, 105, 7087–
 7095.
- (76) Gordon, J. C.; Myers, J. B.; Folta, T.; Shoja, V.; Heath, L. S.; Onufriev, A. H++: a
 server for estimating p K as and adding missing hydrogens to macromolecules. *Nucleic Acids Res.* 2005, *33*, W368–W371.
- ⁹⁴² (77) Case, D.; Cerutti, D.; Cheatham, T.; Darden, T.; Duke, R.; Giese, T.; Gohlke, H.;
 ⁹⁴³ Goetz, A.; Greene, D.; Homeyer, N., et al. Amber18 (University of San Francisco).
 ⁹⁴⁴ 2017,
- (78) Wang, J.; Wolf, R. M.; Caldwell, J. W.; Kollman, P. A.; Case, D. A. Development and
 testing of a general amber force field. J. Comput. Chem. 2004, 25, 1157–1174.
- 947 (79) Jorgensen, W. L.; Chandrasekhar, J.; Madura, J. D.; Impey, R. W.; Klein, M. L.
 948 Comparison of simple potential functions for simulating liquid water. J. Chem. Phys.
 949 1983, 79, 926–935.

- (80) Joung, I. S.; Cheatham III, T. E. Determination of alkali and halide monovalent ion
 parameters for use in explicitly solvated biomolecular simulations. J. Phys. Chem. B
 2008, 112, 9020–9041.
- (81) Andersen, H. C. Rattle: A "velocity" version of the shake algorithm for molecular
 dynamics calculations. J. Comput. Phys. 1983, 52, 24–34.
- (82) Roe, D. R.; Cheatham III, T. E. PTRAJ and CPPTRAJ: software for processing and
 analysis of molecular dynamics trajectory data. J. Chem. Theory Comput. 2013, 9,
 3084–3095.
- (83) Ishida, T. Probing protein environment in an enzymatic process: All-electron quantum
 chemical analysis combined with *ab initio* quantum mechanical/molecular mechanical
 modeling of chorismate mutase. J. Chem. Phys. 2008, 129, 09B618.
- (84) Ishida, T. Effects of point mutation on enzymatic activity: correlation between protein
 electronic structure and motion in chorismate mutase reaction. J. Am. Chem. Soc.
 2010, 132, 7104–7118.
- (85) Cheng, Q.; DeYonker, N. J. The Glycine N-Methyltransferase Case Study: Another
 Challenge for QM-Cluster Models? J. Phys. Chem. B 2023,
- (86) Kast, P.; Asif-Ullah, M.; Hilvert, D. Is chorismate mutase a prototypic entropy trap?
 Activation parameters for the Bacillus subtilis enzyme. *Tetrahedron lett.* 1996, 37, 2691–2694.
- (87) Wappett, D. A.; Goerigk, L. Toward a quantum-chemical benchmark set for enzymatically catalyzed reactions: important steps and insights. J. Phys. Chem. A 2019, 123,
 7057-7074.
- 972 (88) Goerigk, L.; Grimme, S. A thorough benchmark of density functional methods for gen-

eral main group thermochemistry, kinetics, and noncovalent interactions. *Phys. Chem. Chem. Phys.* 2011, *13*, 6670–6688.

975 (89) Jurečka, P.; Šponer, J.; Černý, J.; Hobza, P. Benchmark database of accurate (MP2
976 and CCSD (T) complete basis set limit) interaction energies of small model complexes,
977 DNA base pairs, and amino acid pairs. *Phys. Chem. Chem. Phys.* 2006, *8*, 1985–1993.

- (90) Antony, J.; Grimme, S. Density functional theory including dispersion corrections for
 intermolecular interactions in a large benchmark set of biologically relevant molecules. *Phys. Chem. Chem. Phys.* 2006, *8*, 5287–5293.
- (91) Kesharwani, M. K.; Karton, A.; Martin, J. M. Benchmark *ab initio* conformational
 energies for the proteinogenic amino acids through explicitly correlated methods. Assessment of density functional methods. J. Chem. Theory Comput. 2016, 12, 444–454.
- (92) Paiva, P.; Ramos, M. J.; Fernandes, P. A. Assessing the validity of DLPNO-CCSD (T)
 in the calculation of activation and reaction energies of ubiquitous enzymatic reactions.
 J. Comput. Chem. 2020, 41, 2459–2468.
- (93) Ribeiro, A. J.; Santos-Martins, D.; Russo, N.; Ramos, M. J.; Fernandes, P. A. Enzymatic flexibility and reaction rate: a QM/MM study of HIV-1 protease. ACS Catal.
 2015, 5, 5617–5626.
- (94) Ryde, U. How many conformations need to be sampled to obtain converged QM/MM
 energies? The curse of exponential averaging. J. Chem. Theory Comput. 2017, 13,
 5745–5752.
- (95) Dokainish, H.; Gauld, J. Computational Approach Choice in Modeling Flexible Enzyme
 Active Sites. 2019,
- 995 (96) Mei, Z.; Treado, J. D.; Grigas, A. T.; Levine, Z. A.; Regan, L.; O'hern, C. S. Analyses

- of protein cores reveal fundamental differences between solution and crystal structures.
- ⁹⁹⁷ Proteins **2020**, 88, 1154–1161.



Figure 1: 3D representation of the *RINRUS* maximal model, from the X-ray crystal structure of *Bacillus subtilis* chorismate mutase, using the *probe* ranking scheme. Substrate carbon atoms are colored in magenta. Except for those of the crystallographically resolved water molecule, hydrogen atoms are omitted for clarity.



Figure 2: Kinetics and thermodynamics of the iteratively grown QM-cluster models using the *probe* ranking scheme. Computed ΔG^{\ddagger} values are represented by circles and ΔG_{rxn} values by triangles. The black dashed line shows the experimental ΔG^{\ddagger} value from reference.⁸⁶



Figure 3: Kinetics and thermodynamics of the iteratively grown QM-cluster models using the *arpeggio* ranking scheme. Computed ΔG^{\ddagger} values are represented by circles and stars, and ΔG_{rxn} values by crosses and triangles. The original ranking is given in magenta, while values from the tie-breaking scheme are given in brown. The black dashed line shows the experimental ΔG^{\ddagger} value from reference.⁸⁶



Figure 4: Kinetics and thermodynamics of the iteratively grown QM-cluster models using the F-SAPT ranking scheme. Computed ΔG^{\ddagger} values are represented by circles and stars, and ΔG_{rxn} values by crosses and triangles. The signed ranking order is given in magenta, while the unsigned ranking order is given in brown. The black dashed line shows the experimental ΔG^{\ddagger} value from reference.⁸⁶



Figure 5: Computed values of ΔG^{\ddagger} (circle) and ΔG_{rxn} (triangle) for the 250 maximal QMcluster models plotted against the select frame number (each representing a time scale of 1 ps). The black dashed line at the top is the experimental value from reference.⁸⁶



Figure 6: Charge distribution for the 250 QM-cluster models refined from MD frames. The corresponding number of QM-cluster models for each net model charge is: charge -2 = 1 QM-cluster model, charge -1 = 14 QM-cluster models, charge 0 = 200 QM-cluster models, charge +1 = 33 QM cluster models, and charge +2 = 2 QM-cluster models.



Figure 7: Mean activation free energy (brown), reaction free energy (magenta), and number of QM-cluster models with a given number of explicit water molecules (cyan) identified as having interatomic contacts with the chorismate for the 250 QM-cluster models built from selected MD frames.

Scheme	Cluster	# of frames	ΔG^{\ddagger}	σ	ΔG_{rxn}	σ
\mathbf{S}_1		20	10.07	± 2.87	-16.23	± 3.90
\mathbf{S}_2		30	10.12	± 2.39	-16.28	± 3.82
\mathbf{S}_3		30	10.03	± 2.83	-14.99	± 3.06
\mathbf{S}_4		30	10.06	± 1.88	-15.92	± 2.86
\mathbf{S}_{5}		30	10.29	\pm 3.05	-15.57	\pm 3.09
	C_1	10	9.90	± 2.74	-16.78	± 3.12
	C_2	10	11.03	± 3.98	-15.54	± 3.48
	C_3	10	9.95	± 1.92	-14.40	± 2.02
\mathbf{S}_6		40	10.23	± 2.38	-15.24	± 3.04
\mathbf{S}_7		40	10.74	± 2.69	-15.35	± 3.52
\mathbf{S}_8		30	10.96	± 2.69	-13.82	± 3.36
	C_1	10	10.83	± 1.95	-13.85	± 2.30
	C_2	10	10.80	± 3.94	-13.57	± 4.15
	C_3	10	11.49	± 1.41	-14.06	± 3.34
Combined		250	10.34	± 2.62	-15.38	± 3.40

Table 1: Mean free energies of activation and reaction for the various MD frame selection schemes. K-means clusters are labelled with a C (all in kcal mol^{-1}).

Table 2: Mean free energies of activation and reaction for the expanded schemes. The individual k-means clusters are labelled XC (all in kcal mol^{-1}).

Scheme	Cluster	# of frames	ΔG^{\ddagger}	σ	ΔG_{rxn}	σ
$S_1 + S_6 + S_7$		100	10.40	± 2.63	-15.48	± 3.44
\mathbf{XS}_2		148	10.17	± 2.75	-15.64	± 3.44
\mathbf{XS}_3		173	10.35	± 2.70	-15.46	± 3.50
\mathbf{XS}_4		186	10.39	± 2.64	-15.36	± 3.37
\mathbf{XS}_5						
	XC_1	92	10.25	± 2.63	-16.16	\pm 3.92
	XC_2	89	10.30	± 2.85	-15.00	± 3.05
	XC_3	69	10.50	± 2.28	-14.85	± 2.83
\mathbf{XS}_8						
	XC_1	77	10.42	± 2.52	-15.38	± 3.54
	XC_2	81	10.27	± 2.86	-15.05	± 3.39
	XC_3	92	10.32	± 2.49	-15.69	± 3.24
Combined		250	10.34	± 2.62	-15.38	± 3.40