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Triple action Pt(IV) derivatives of cisplatin: a new class of potent anticancer agents that overcome resistance†

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A series of triple action Pt(IV) prodrugs was designed to test the hypothesis that multi-action compounds, where each bioactive moiety intervenes in several cellular processes, might be more effective than a single agent at killing cancer cells. In particular, "triple action" Pt(IV) derivatives of cisplatin, where the axial ligands are inhibitors of cyclooxygenase (COX), histone deacetylase (HDAC) or pyruvate dehydrogenase kinase (PDK) were developed. All compounds, $\text{Pt}(\text{NH}_3)_2(\text{COX})(\text{PDK})\text{Cl}_2$, $\text{Pt}(\text{NH}_3)_2(\text{COX})(\text{HDAC})\text{Cl}_2$ and $\text{Pt}(\text{NH}_3)_2(\text{HDAC})(\text{PDK})\text{Cl}_2$, where COX = aspirin or ibuprofen, PDK = dichloroacetate and HDAC = valproate or phenylbutyrate, were significantly more cytotoxic than cisplatin against all cell lines of an in-house panel of human cancer cells. They were particularly effective against thyroid and pancreatic cancer cells in monolayer cytotoxicity tests. Remarkably, in 3D spheroid cancer cell cultures, some triple action compounds showed an antitumor potency up to 50-fold higher than cisplatin against a KRAS mutated pancreatic cancer cell line (PSN-1 cells). Standard biochemical assays classically employed to explore structure activity relationships of platinum drugs, such as cellular uptake and binding to potential biological targets (DNA, HDAC, mitochondria, and COX), do not provide linear correlations with the overall cytotoxicity data. We observed a preferential induction of ROS production and of an anti-mitochondrial effect in cancer cells compared to rapidly dividing non-cancerous cells. Thus, we propose that these new triple action Pt(IV) derivatives of cisplatin are a novel and interesting class of potent and selective cytotoxic agents.

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Introduction

Cisplatin, carboplatin and oxaliplatin (Fig. 1) are square planar Pt(II) anti-cancer drugs that are in widespread clinical use.¹ The first two compounds are particularly effective against testicular and ovarian cancers, but are also widely used for many other indications in combination with other drugs, while oxaliplatin is used to treat colorectal cancers. Platinum drugs are used in 50% of all clinical regimens.² The platinum drugs are believed to trigger cancer cell death by losing their non-am(m)ine ligands and covalently binding to two adjacent guanines on the same strand of the nuclear DNA. Binding to the DNA creates a significant distortion of the double helical structure, and the responses of the cancer cells to the distortion determine the fate of the cancer cells.³

Despite its efficacy and success, severe side effects and resistance limit the usage of cisplatin.⁴ In order to overcome

resistance, clinicians administer the platinum drugs in combination with other drugs.⁵ The main advantage of combination chemotherapy is that several anti-proliferative agents that have different mechanisms of action and different cellular targets, attack the tumors thereby increasing the chances of killing the cancer cells and of overcoming resistance to a single drug. The problem with co-administration of several drugs is that each drug has a different pharmacokinetic profile, thus complicating the prediction of the overall therapeutic outcome.

Although currently all the Pt drugs used in the clinic are Pt(II) complexes, Pt(IV) complexes have recently attracted a lot of attention since their chemical properties allow great flexibility in the design of novel drugs, including multi-target drugs.⁶ Pt(IV) complexes, such as satraplatin (Fig. 1), are low spin d^6 octahedral complexes that are kinetically more inert than their Pt(II) precursors and may be administered orally, a feature that can improve the quality of life of patients and reduce hospitalization costs.⁷ Pt(IV) complexes are believed to act as prodrugs that are activated inside the cancer cells by reductive elimination resulting in the concurrent release the two axial ligands as well as the square planar Pt(II) drug (Scheme 1).⁸ "Dual action" Pt(IV) prodrugs that are Pt(IV) derivatives of cisplatin, carboplatin or oxaliplatin with bioactive axial ligands

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Fig. 1 Structures of the three Pt(II) drugs approved by FDA as well as satraplatin (top) and examples of dual action Pt(IV) prodrugs with DPKi (DCA), COXis (ASP and Ibu) and HDACis (Val and PhB) (bottom).



Scheme 1 The Pt(IV) prodrugs were prepared from cisplatin that was oxidized to oxoplatin with H₂O₂. Two successive carboxylations of oxoplatin with the different anhydrides yield "triple action" prodrugs that are reduced in the cell releasing cisplatin as well as the two bioactive axial ligands.

have gained popularity, unfolding interesting horizons for improving Pt-based chemotherapy.⁹

Mitaplatin is the Pt(IV) derivative of cisplatin that has two dichloroacetate (DCA) moieties in the axial positions (Fig. 1).¹⁰ DCA is an orphan drug that inhibits the pyruvate dehydrogenase kinase (PDK), an enzyme that phosphorylates the pyruvate dehydrogenase complex (PDHC), a key enzyme in the cellular respiration process. The inhibition of PDHC by PDK shifts the cellular metabolism from glucose oxidation to glycolysis (Warburg effect).¹¹ The inhibition of PDK reverses the Warburg effect, compromising the survival of the tumour cells. Mitaplatin combines the action of cisplatin that binds to the DNA with that of DCA that acts at mitochondrial level.

Pt(IV) complexes with axial histone deacetylase (HDAC) inhibitors are another example of dual action compounds. Histones are proteins that control the structure of chromatin and nucleosomes. Their deacetylation results in a more open chromatin structure, exposing DNA to potential platination.

HDAC inhibitors can reactivate dormant tumor-suppressor genes.¹² HDAC inhibitors, such as vorinostat or belinostat, are best characterized as anticancer agents. Some HDACis were approved by the FDA for treatment of lymphomas, and others are in clinical trials.¹³ Tethering the HDACis phenylbutyrate (PhB) or valproate (Val) to the axial positions of the Pt(IV) derivative of cisplatin yielded very potent cytotoxic agents being up to 100-fold more potent than cisplatin.^{14–16} PhB is also an inhibitor of PDK, and its combination with DCA led to an increase of PDHC activity that is greater than the sum of the effects of the single drugs alone.¹⁷

Another class of dual action prodrugs are Pt(IV) complexes with cyclooxygenase (COX) inhibitors in the axial positions. The rationale for the design of such complexes stems from the observation that cancer is often associated with chronic inflammation and that COX-2 is overexpressed in many tumors and is involved in their initiation and progression.¹⁸ Also, there are reports that combining cisplatin with COXis enhances the



potency of cisplatin and/or reduces the side effects.^{19,20} The nonsteroidal anti-inflammatory drugs (NSAIDs) that were conjugated to the axial positions of Pt(IV) complexes include, ibuprofen and indomethacin. The Pt(IV) derivative of cisplatin with one axial aspirin and one hydroxido ligand is a potent cytotoxic agent and also exhibits anti-inflammatory activity.^{21,22}

The Pt(IV) derivative of cisplatin with two ibuprofens in the axial positions, *cis*-Pt(Ibu)₂ (Fig. 1) is a very active compound, with IC₅₀ values in the nM range.²³ Moreover, combining HDAC and COX-2 inhibitors increases the efficacy by a factor of 10 compared with the single drugs.²⁴

The hypotheses underlying the preparation of triple action compounds were also reinforced by the high effectiveness of a quadruple action Pt(IV) prodrug that simultaneously releases four different bioactive moieties inside the cancer cell. Its cytotoxicity against cancer cells was up to 450-fold more potent than cisplatin against KRAS mutated cancer cells.²⁵ Surprisingly, we are not aware of any reports on systematic studies on triple action Pt(IV) based prodrugs.

Herein we report on the design, synthesis and biological studies on eight triple action Pt(IV) compounds, that upon reduction release three different bioactive moieties in the cancer cell. There are a lot of possible rationales and options for choosing the two different bioactive axial ligands. We elected to begin by choosing the above mentioned bioactive ligands because of the demonstrated ability of each one, by a different mechanism, to act synergistically with cisplatin to enhance cytotoxicity. The eight compounds reported here represent some possible combinations of these ligands. The compounds

are Pt(IV) derivatives of cisplatin that can be divided to three classes; (a) those with one PDKi (DCA) and one COXi (Asp or Ibu), (b) those with one PDKi and one HDACi (PhB or Val) and (c) those with one HDACi and one COXi.

Results and discussion

Synthesis

The complexes presented in this work are Pt(IV) derivatives of cisplatin with two different bioactive ligands in the axial positions making them “triple action” Pt(IV) anticancer prodrug candidates. They were obtained from oxoplatin by two successive carboxylations with the anhydrides of the axial ligands (Fig. 2). The first carboxylation was performed in DMSO using a low concentration of oxoplatin (~10 mg mL⁻¹)²⁶ and the second carboxylation was performed in acetonitrile at higher concentrations.

The efficiency of monocarboxylations depends of the solubility of the Pt(IV) dihydroxido complex and on the reactivity of the anhydride. Oxoplatin is poorly soluble in common solvents making it difficult to isolate the monocarboxylated complexes, especially with highly reactive anhydrides. Initially, the reaction mixtures were pale yellow suspensions and upon completion of the first carboxylation the reaction mixtures turned into pale yellow solutions. The ¹⁹⁵Pt NMR resonances at ~1050 ppm, are consistent with monocarboxylato derivatives of oxoplatin. The monocarboxylation derivatives with aspirin (CAOH), ibuprofen (CIOH), phenylbutyrate (CPOH), and valproate (CVOH) were

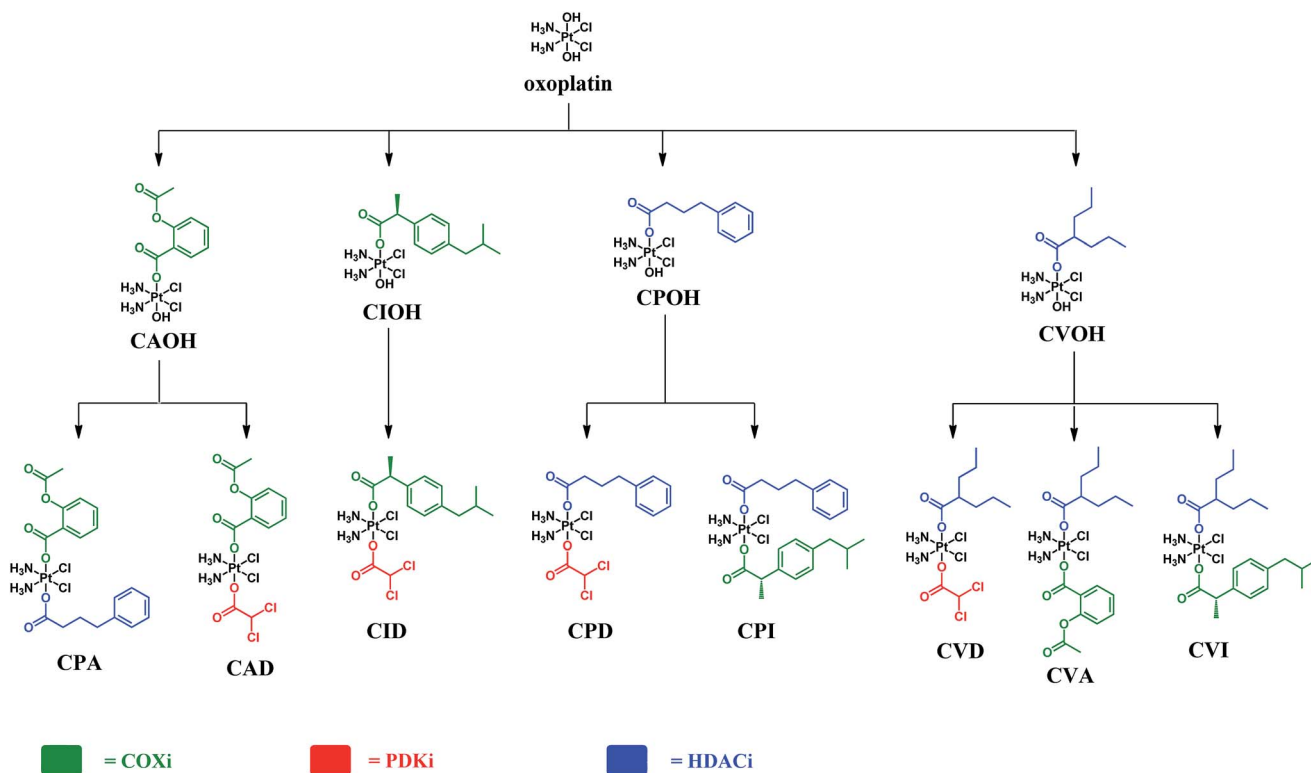


Fig. 2 Schemes of the synthetic procedures of the triple action Pt(IV) derivatives of cisplatin.



used, without further purification, as intermediates for the following step.

Because of the high reactivity of the DCA anhydride we were not able to selectively obtain the monocarboxylato derivative $\text{ctc}[\text{Pt}(\text{NH}_3)_2(\text{DCA})(\text{OH})\text{Cl}_2]$. Aspirin anhydride reacted with oxoplatin very slowly, so the concentration of oxoplatin was raised to $\sim 15 \text{ mg mL}^{-1}$, to increase the speed of reaction. The second carboxylation reactions were performed in acetonitrile and monitored by RP-HPLC. All the biscarboxylato derivatives were characterized by ^1H , ^{195}Pt NMR and HPLC and their purity was ascertained by elemental analysis (Fig. S1–S8 and Table S1†). The compounds did not fly in the ESIMS.

These compounds can be divided into three classes based on the combination of the biological activity of their axial ligands: the first has one COXi and one PDKi (**CAD** and **CID**); the second has one HDACi and one COXi (**CPA**, **CPI**, **CVA** and **CVI**) and the third has one HDACi and one PDKi (**CPD** and **CVD**) – Fig. 2.

Stability studies

The stability of the eight triple action compounds in phosphate buffer solution at 37°C was monitored by HPLC. **CID**, **CPD**, **CVD** and **CVI** were stable under these conditions. After one day, **CPA**, **CVA** and **CPI** underwent slow ~ 10 , 20 and 25% degradation, respectively. Degradation continued in the following three days. **CAD** was the least stable compound, with $>50\%$ degradation after the first day, and complete degradation in the following days. The nature of these degradation processes is currently under investigation in order to elucidate their mechanism. Interestingly, compounds with an aspirin ligand were less stable, leading us to suspect that this ligand may play a central role in the degradation process.

Cytotoxicity

The cytotoxicity of the eight triple action compounds was assessed in a panel of human cancer cell lines comprised of colorectal (HCT-15), thyroid (BCPAP), pancreatic (PSN-1), and colon (LoVo) cancer cell lines by the MTT test, using cisplatin and oxaliplatin as references. IC_{50} values were calculated from

the dose–survival curves obtained after 72 h of drug treatment (Table 1). In addition, the cytotoxic properties of the complexes were evaluated in a pair of tumor cell lines selected for their sensitivity and resistance to cisplatin (ovarian adenocarcinoma cells 2008/C13*). We calculated cross-resistance profiles by means of the resistance factor (RF), defined as the ratio between IC_{50} values calculated for the resistant cells and those obtained with the sensitive ones. Interestingly, regardless of the composition of the triple action compounds, all were more potent than cisplatin in all the cell lines tested and they were more effective than oxaliplatin against thyroid (BCPAP), pancreatic (PSN-1), colon (LoVo) and ovarian (2008) cancer cells. While the average IC_{50} of cisplatin over the six cell lines was $12.46 \mu\text{M}$, those for the triple action compounds range from 0.37 – $1.46 \mu\text{M}$. Three cell lines are more sensitive than the others to the triple action compounds. The average IC_{50} values of all eight compounds against BCPAP ($0.14 \mu\text{M}$), PSN-1 ($0.26 \mu\text{M}$) and C13* ($1.13 \mu\text{M}$) are 51, 71 and 20 times lower than those of cisplatin while against HCT-15, 2008 and LoVo they are about 6, 3 and 24 fold better.

The triple action compounds comprise three groups, based on the nature of their bioactive axial moieties. The first group combines COX and PDK inhibitors and consists of **CAD** and **CID**. These two compounds have average IC_{50} values of 0.70 and $0.41 \mu\text{M}$, respectively, about 18 and 31 times lower than cisplatin (average $\text{IC}_{50} = 12.47 \mu\text{M}$) and roughly 5 and 8 times lower than oxaliplatin (average $\text{IC}_{50} = 3.23 \mu\text{M}$). Both compounds were extremely effective against the thyroid (BCPAP, $\text{IC}_{50} = 0.06$ and $0.14 \mu\text{M}$, respectively) and pancreatic cancer (PSN-1, $\text{IC}_{50} = 0.06$ and $0.08 \mu\text{M}$, respectively) cell lines. **CAD** and **CID** were somewhat less active against the resistant ovarian cancer C13* but still much better than cisplatin. They were slightly cross-resistant with cisplatin (RF = 2.7 and 3.0, respectively).

The second group of triple action compounds combines HDAC and COX inhibitors, and consists of four compounds: **CPA**, **CPI**, **CVA**, and **CVI**. The average IC_{50} against all tested cell lines of **CPA** ($0.46 \mu\text{M}$) is 27 fold lower than that of cisplatin, while those for **CPI**, **CVA**, and **CVI** are *ca.* 10 times more potent

Table 1 Cytotoxicity assessed by the MTT test. Cells (3 – $8 \times 10^4 \text{ mL}^{-1}$) were treated for 72 h with increasing concentrations of tested compounds. IC_{50} values were calculated by a four parameter logistic model ($P < 0.05$). Resistant factor (RF) is defined as IC_{50} resistant/parent line

Class	Cmpd	IC_{50} (μM)						RF	Average
		HCT-15 colon	BCPAP thyroid	PSN-1 pancreas	LoVo colon	2008 ovary	C13* ovary		
COXi, PDKi	CAD	1.03 ± 0.25	0.06 ± 0.005	0.06 ± 0.008	0.755 ± 0.06	0.61 ± 0.19	1.66 ± 0.09	2.7	0.70
	CID	0.65 ± 0.17	0.14 ± 0.04	0.08 ± 0.01	0.285 ± 0.02	0.32 ± 0.09	0.97 ± 0.22	3.0	0.41
HDACi, COXi	CPA	1.86 ± 0.41	0.06 ± 0.004	0.09 ± 0.02	0.055 ± 0.01	0.29 ± 0.11	0.43 ± 0.12	1.5	0.46
	CPI	4.98 ± 1.25	0.08 ± 0.01	0.92 ± 0.2	0.211 ± 0.08	0.89 ± 0.19	1.65 ± 0.11	1.9	1.46
	CVA	4.51 ± 0.85	0.01 ± 0.003	0.13 ± 0.04	0.97 ± 0.08	0.69 ± 0.08	0.77 ± 0.04	1.1	1.18
	CVI	3.98 ± 0.89	0.68 ± 0.08	0.07 ± 0.01	0.034 ± 0.03	1.35 ± 0.29	1.61 ± 0.42	1.2	1.29
HDACi, PDKi	CPD	1.06 ± 0.13	0.11 ± 0.03	0.10 ± 0.02	0.019 ± 0.01	0.33 ± 0.04	0.58 ± 0.06	1.8	0.37
	CVD	3.11 ± 1.02	0.01 ± 0.004	0.61 ± 0.1	0.757 ± 0.23	1.12 ± 0.20	1.39 ± 0.37	1.2	1.17
Average IC_{50}		2.65	0.14	0.26	0.39	1.13	0.71		
Ref.	CDDP	15.28 ± 2.63	7.38 ± 1.53	18.25 ± 3.11	9.15 ± 2.07	2.22 ± 1.02	22.52 ± 3.15	10.1	12.47
	OXP	1.15 ± 0.43	4.37 ± 1.07	8.25 ± 3.42	1.01 ± 0.34	1.53 ± 0.88	3.06 ± 1.00	2.0	3.23



than cisplatin. CPA is 7 times more effective than oxaliplatin and CPI, CPV, and CVI are about 3 times more potent than oxaliplatin. These compounds are less potent against colorectal cancer HCT-15 cells (low μM IC_{50}s) but extremely potent against thyroid and pancreatic cancers, with some IC_{50}s under 100 nM. Against the colorectal HCT-15 cancer cells, the compounds were more potent than cisplatin. In the ovarian cancer cell pair (2008 and C13* cells), they showed no cross resistance with cisplatin, eliciting RF values lower than 2.

The last class of compounds consists of CPD and CVD that combine HDAC and PDK inhibitors. The average IC_{50} values for CPD (0.37 μM) are 34 and 9-fold more potent than cisplatin and oxaliplatin, respectively, and 3 times more potent than CVD. Excluding HCT-15 cells, CPD has sub-micromolar IC_{50} values against all tested cell lines. Replacing PhB with Val (as the HDACi) resulted in reduced potency of CVD compared to CPD in all cell lines, with the exception of thyroid tumor cells, against which CVD has a low nM IC_{50} . Both compounds retained their activity against cisplatin-resistant ovarian cancer cells with RF values lower than 2.

In order to assess whether triple action compounds are more active than their dual action analogues, we synthesized and screened five dual action compounds that have one bioactive ligand in an axial position and an inert acetato group in the other. We used HDACis (PhB and Val), PDKi (DCA) or COXis (Asp and Ibu) as the bioactive ligands. Complexes CPAc, CVAc, CDAc, CAAc, and CIAc, are depicted in Fig. 3. Although they are not the exact analogues of the triple action compounds (differ in lipophilicity, solubility *etc.*), the dual action compounds are all capable of delivering the cisplatin and one bioactive ligand into the cancer cells. Thus, a comparison with the triple action compounds could be helpful for assessing the relevance of adding the second bioactive ligand.

Comparing the average IC_{50} values of the five dual action compounds to the eight triple action compounds and to cisplatin (Table S2[†]), we see that in general the triple action compounds are more potent than the dual action compounds that in turn are more effective than cisplatin and oxaliplatin. Only in some cases, dual action compounds were slightly more effective than their corresponding triple action complexes (Table S2[†]).

All the triple action compounds are potent cytotoxic agents and hence it is difficult to select lead compounds from these data. Although the average IC_{50} values of the compounds are very impressive, we are interested primarily in identifying specific compounds that are effective against specific cancers. Judging from their IC_{50} values, CID, CPD, CPA and CAD seem like the most potent compounds.

To assess whether combining the three bioactive components in a single triple action prodrug is superior to exposing the cells to a mixture of its components, we evaluated the cytotoxicity of a combination of cisplatin, PhB and DCA at 1 : 1 : 1 ratio against five cancer cell lines. The results reported in Table S4[†] clearly confirm that co-treatment of cancer cells with cisplatin, PhB and DCA, yielded IC_{50} values similar to those of cisplatin itself (significantly higher than CPD) and did not result in an synergistic or additive cytotoxic effect.

To assess further the potential of these compounds, we screened them against 3D spheroids of PSN-1 pancreatic cancer cells. Conventional 2D cell cultures cannot mimic the complexity of *in vivo* tumors and hence they are not always good predictors of *in vivo* activity. On the other hand, 3D cell cultures possess several features such as cell-cell interactions, hypoxia, drug penetration, response and resistance, and production/deposition of the extracellular matrix that more closely mimic *in vivo* tumors, being potentially more predictive for *in vivo* results than conventional 2D cell cultures.²⁷

We tested the triple action Pt(IV) cisplatin derivatives on spheroids of PSN-1 pancreatic tumor cells, using cisplatin as reference (Table 2). We chose to conduct the 3D spheroids screen on the pancreatic cancer cell line (rather than the thyroid cancer cell line) since pancreatic cancer survival rates are dramatically lower than those for thyroid cancer, and there is interest in agents that are effective against pancreatic cancer. All the newly synthesized triple action compounds were significantly more active than cisplatin. Three compounds, CVA, CVI and CPD, have sub- μM IC_{50} values, being more than 50 times more active than cisplatin. The other five compound have IC_{50}s ranging from *ca.* 3 to 8 μM , being 15 to 8 fold more potent than cisplatin.

Since all the triple action compounds are not or just slightly cross-resistant with cisplatin we decided to check whether they



Fig. 3 Chemical structures of acetato dual action compounds CPAc, CVAc, CDAc, CAAc and CIAc.



Table 2 Cytotoxicity assessed in 3D cultures. Spheroids (2.5×10^3 cells per well) were treated with tested compounds. The growth inhibitory effect was evaluated after 72 hours by means of APH test. IC_{50} values were calculated from the dose–survival curves by four parameter logistic model ($P < 0.05$)

Class	IC_{50} (μM)	
	Compound	PSN-1 3D
COXi, PDKi	CAD	3.73 ± 0.35
	CID	5.25 ± 0.71
HDACi, COXi	CPA	7.75 ± 0.01
	CPI	3.63 ± 0.53
	CVA	0.79 ± 0.06
	CVI	0.93 ± 0.46
HDACi, PDKi	CPD	0.95 ± 1.12
	CVD	3.38 ± 0.88
Reference	CDDP	52.56 ± 3.78

can also circumvent oxaliplatin resistance. We screened them against LoVo colon cancer cells and the oxaliplatin resistant form LoVo OXP (Table S3†). Four compounds circumvented resistance to oxaliplatin having RF values < 2 : the two compounds combining HDAC and PDK inhibitors (**CPD** and **CVD**) and two combining the HDACis (PhB and Val) with the COXi aspirin. Interestingly, all compounds containing ibuprofen had very high RF values. **CPD** and **CPA** exhibit remarkably low IC_{50} values, and are the most active among the all eight triple action compounds.

One of the major drawbacks of chemotherapeutics, including platinum drugs, are the side effects originating in part from toxic effects to non-cancerous cells. We measured the cytotoxicity of the compounds against non-cancerous cells (HEK293) and calculated the selectivity index (SI) defined as the ratio of the IC_{50} s of the non-cancerous cells to those of the cancer cells that are more sensitive to triple action complexes, namely PSN-1, BCPAP and LoVo cells. The cytotoxicity data for HEK293 cells and the selectivity indices appear in Table 3.

CPD has high selectivity indices against PSN-1 (15.6), BCPAP (14.2) and LoVo (82.1) cancer cells. **CVA** and **CVD** have high SI against BCPAP cancer cells (59 and 198 respectively) and showed a moderate selectivity against PSN-1 cells (SI 4.5 and 3.2 respectively). **CVI** has SI of 6.7 and 13.8 against PSN-1 and LoVo respectively. The remaining complexes have no selectivity towards those cancer cells.

Among the eight compounds studied, **CPD** is the most promising compound since it is one of the most active compounds in 2D and 3D cell cultures, is active against cisplatin- and oxaliplatin-resistant cell lines, and has high selectivity towards BCPAP and the KRAS mutated PSN-1 and LoVo cancer cell lines.

Biochemical assays

When trying to kill a cancer cell with a single agent such as cisplatin it is intuitively appealing to assume that increased cellular accumulation should correlate with enhanced platination of nuclear DNA resulting in higher potency. Many reports describe attempts to increase the cellular accumulation of cisplatin, including using Pt(IV) prodrugs with lipophilic

Table 3 Cytotoxicity assessed by the MTT test in HEK293. Cells ($5 \times 10^3 \text{ mL}^{-1}$) were treated for 72 h with increasing concentrations of tested compounds. The cytotoxicity was assessed by the MTT test. IC_{50} values were calculated by a four parameter logistic model ($P < 0.05$). S.D. = standard deviation. Selectivity index (SI) is defined as IC_{50} non-tumor/tumor cell line

Class	Compound	IC_{50} (μM) HEK293	Selectivity index (SI)		
			PSN-1	BCPAP	LoVo
COXi, PDKi	CAD	0.09 ± 0.02	1.5	1.5	0.1
	CID	0.07 ± 0.03	0.9	0.5	0.2
HDACi, COXi	CPA	0.11 ± 0.01	1.2	1.8	2.0
	CPI	0.13 ± 0.03	0.1	1.6	0.6
	CVA	0.59 ± 0.21	4.5	59.0	0.6
	CVI	0.47 ± 0.08	6.7	0.7	13.8
HDACi, PDKi	CPD	1.56 ± 0.22	15.6	14.2	82.1
	CVD	1.98 ± 0.26	3.2	198.0	2.6
Reference	CDDP	19.62 ± 2.33	1.1	2.6	2.1

ligands.²⁸ The presence of lipophilic ligands in the axial positions increases the lipophilicity of the whole Pt(IV) complex, improving the cellular internalization of both cisplatin and of the anionic axial ligands (“synergistic accumulation”).¹⁴ This in turn is expected to increase the efficiency of DNA platination by cisplatin.

We measured the cellular uptake and the DNA platination levels of the triple action compounds and cisplatin in PSN-1 cells (Fig. 4A and B). Clearly, there is no correlation between the cellular accumulation and the potency. The most potent compound, **CAD**, has the lowest accumulation level among the triple action compounds, significantly lower than **CID** and **CVI** whose IC_{50} values are similar to that of **CAD**. This suggests that **CAD** might be the most effective compound at killing the PSN-1 cancer cells. Although we might expect that higher cellular accumulation would result in higher platination levels of nuclear DNA, this is not the case here (Fig. 4B). Interestingly, there is no correlation between platination levels of the nuclear DNA and the cytotoxicity. **CPI** is the least potent triple action compound yet has the highest levels of DNA platination. One of the most potent compounds, **CAD**, has the lowest platination level of all compounds, 5.6 fold lower than cisplatin although it is 304 fold more cytotoxic than cisplatin. DNA platination by the triple action compounds suggests that the prodrugs were reduced in the cell releasing cisplatin that modified the DNA. In summary, with the triple action compounds we find no correlation between the cytotoxicity and cellular accumulation or DNA platination.

Although we realize that each bioactive component can interact with several cellular targets, it is interesting to see whether they actually inhibit in the cancer cells the enzymes they were designed to inhibit.

We investigated their ability to inhibit HDAC activity (Fig. 5A). As expected, only the compounds with a PhB or Val ligands inhibited HDAC activity. PhB and Val in the axial positions were equally effective at inhibiting HDAC activity. Again, there is no correlation between inhibition of HDAC activity and cytotoxicity.



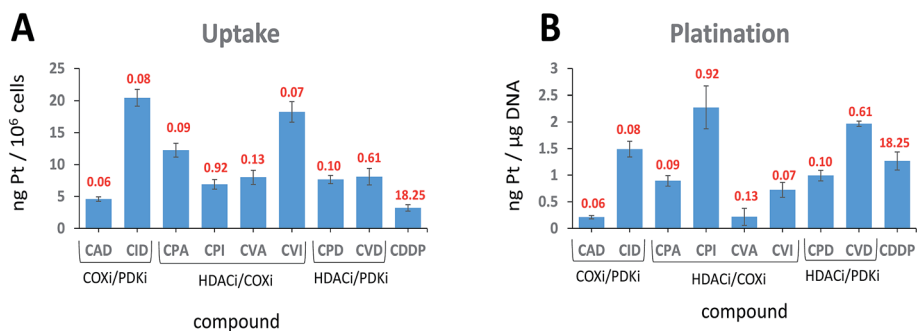


Fig. 4 Cellular uptake (A) and DNA platination (B). (A) PSN-1 cells were incubated for 24 h with 1 μ M of tested complexes. (B) Platination levels of nuclear DNA extracts. PSN-1 cells were treated for 24 h with 1 μ M of tested complexes. DNA was extracted, quantified and the amount of Pt bound to DNA was estimated by GF–AAS. The IC₅₀ values of the compounds against the PSN-1 cells are depicted in red.

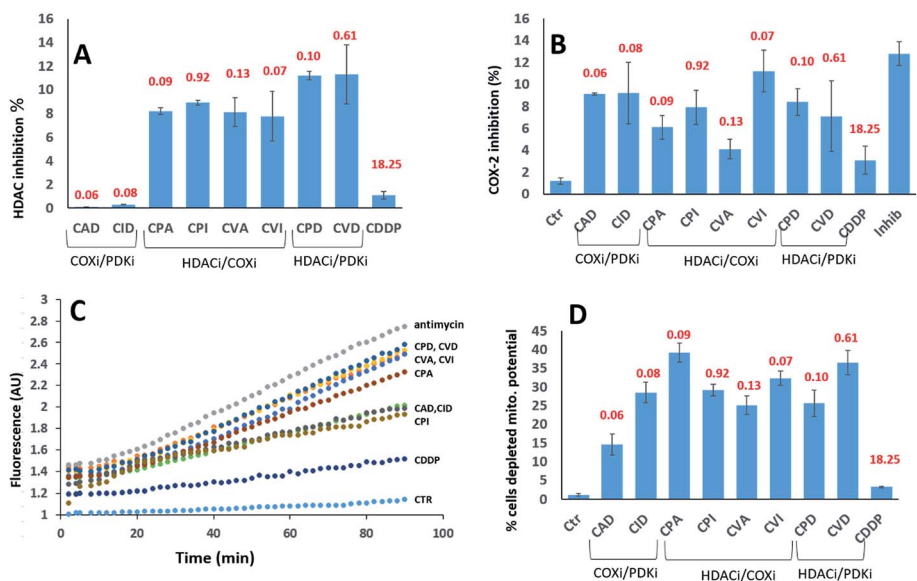


Fig. 5 (A) HDAC inhibition. PSN-1 cells were incubated for 24 h with 1 μ M of tested complexes. The HDAC activity was determined in cells by FLUOR DE LYS® HDAC fluorometric activity assay kit (Enzo Life Sciences) following the manufacturers' instructions. In all cases, data are the means of at least three independent experiments. (B) COX-2 expression. PSN-1 cells were incubated for 24 h with 1 μ M of tested complexes. The inhibition of COX-2 was measured by using COX activity assay kit (Cayman Chemical, Ann Arbor, USA) following the manufacturers' instructions. In all cases, data are the means of at least three independent experiments. Error bars indicate SD. (C) ROS production. PSN-1 cells were pre-incubated in PBS/10 mM glucose medium for 20 min at 37 °C in the presence of 10 μ M CM–DCFDA and then treated with equimolar doses (10 μ M) of Pt(IV) derivatives. The fluorescence of DCF was measured. In all cases, data are the means of three independent experiments. (D) Effects on cellular mitochondrial membrane potential. PSN-1 cells were treated for 24 h with 1 μ M of tested compounds. The percentage of cells with hypopolarized mitochondrial membrane potential was determined by Mito-ID® Membrane Potential Kit. The IC₅₀ values of the compounds against the PSN-1 cells are depicted in red.

We assayed their ability to inhibit the expression of COX-2 in PSN-1 treated cells (Fig. 5B). Previous reports indicated that aspirin–Pt(IV) derivatives maintained their COX inhibition effect,²¹ while ibuprofen–Pt(IV) derivatives did not.²² In the case of triple action Pt(IV) compounds, CVI was the most effective COX-2 inhibitor (11.2% inhibition), while its aspirin analogue CVA was the least effective (4.1% inhibition). A similar trend was observed for CPI and CPA, causing 7.9 and 6.1% inhibition respectively. COX-2/PDKi functionalized compounds CAD and CID caused ~9% COX-2 inhibition, with the aspirin derivative slightly more potent than the ibuprofen analogue. Surprisingly, CPD and CVD complexes that do not have any COX-2 moiety caused 8.4 and

7.1% COX-2 inhibition, respectively. These results suggest that also complexes not containing any COX-2 moiety can somehow inhibit the activity of COX-2 in cancer cells.

It is well known that DCA treatment causes an imbalance in the redox homeostasis, leading to increased production of reactive species, de-structuration of mitochondrial pathophysiology and loss of mitochondrial membrane potential.²⁷ Therefore, we assayed the effects induced by tested complexes on mitochondria by measuring the cellular ROS production (Fig. 5C) and mitochondrial membrane potential (Fig. 5D). All tested complexes were able to induce a time-dependent increase in the cellular basal ROS production. The ROS production



induced by triple action complexes was significantly superior than that induced by cisplatin but it was lower than that provoked by a classical inhibitor of the mitochondrial respiratory chain at the level of complex III, namely antimycin A. The most effective complexes were **CVD** and **CPD**, being able to induce an increase in cellular basal ROS production that is only slightly inferior to that elicited by antimycin A. It is well known that a substantial increase of ROS production at mitochondrial level leads to the hypopolarization of the mitochondrial membrane. Consistently, all the triple action compounds are significantly more active than their cisplatin precursor in inducing mitochondrial membrane depolarization in cancer cells. Complexes **CPA**, **CVD** and **CVI** induced hypopolarization in 30–40% of the cancer cell population. **CID**, **CPD**, **CPI** and **CVA** induced hypopolarization in 25–30%, while **CAD** in ~15%. These results clearly showed that all complexes increase the cancer cell population with depolarized mitochondria. Interestingly, a mitochondrial effect was observed also for complexes lacking of DCA and PhB moieties that are known to act on mitochondria. This behaviour can be explained by taking into account that COX-2 modulators can affect mitochondrial homeostasis by raising the cellular basal ROS production, resulting in mitochondrial membrane potential loss. It is important to note that in rapidly dividing non-cancerous HEK293 cells, all triple action Pt(IV) complexes, and in particular **CPD** and **CVD**, were barely effective in inducing cellular ROS production and in affecting the mitochondrial membrane potential (Fig. S13, † panel A and B). This is in sharp contrast to the results obtained with the pancreatic cancer cells (PSN-1), demonstrating their preferential activity toward cancer cells compared with non-tumor cells.

Conclusions

In this work we studied three classes of triple action Pt(IV) derivatives of cisplatin having; (1) COXi/PDKi (**CAD** and **CID**); (2) HDACi/COXi (**CPA**, **CPI**, **CVA** and **CVI**); (3) HDACi/PDKi (**CPD** and **CVD**). All compounds were more cytotoxic than cisplatin against all cell lines tested and were particularly effective against thyroid and pancreatic cancer cells.

As 2D cytotoxicity studies indicate that all eight compounds were potent cytotoxic agents, further experiments, such as 3D cytotoxicity and selectivity studies were performed to help select lead compounds. Among the eight compounds studied, **CPD** is the most promising compound, being active against cisplatin- and oxaliplatin-resistant cell lines, and having high selectivity towards PSN-1, BCPAP and LoVo cancer cell lines.

The triple action Pt(IV) prodrugs were designed to test the hypothesis that multi-action compounds, where each bioactive moiety intervenes in several cellular processes, might be more effective than a single agent as several mechanisms may be involved in killing the cells.

Thus, it is not surprising that the standard biochemical essays employed for platinum drugs such as investigating the cellular uptake and the possible biological targets (DNA, HDAC, and COX) do not provide correlations with the overall cytotoxicity data. As all compounds are active, it is unlikely that there is

a single mechanism of action common to all compounds. The potency of the triple action compounds probably depends primarily on the specific cellular interactions of each of the components and on possible synergistic effects to which the cell responds, and not necessarily just on the amount of the drug that accumulates in the cell.

Interestingly, the most promising triple action compound, **CPD**, shares the bioactive axial ligands (DCA and PhB) with the potent and selective dinuclear quadruple action Pt(IV) complex we recently described.²⁵

In light of the cellular studies reported above, we believe that triple action Pt(IV) derivatives of cisplatin represent a novel and interesting class of potent and selective cytotoxic agents.

Conflicts of interest

There are no conflicts to declare.

List of abbreviations

Asp	Aspirin
CDDP	Cisplatin
DCA	Dichloroacetic acid/dichloroacetate
Ibu	Ibuprofen
OAc	Acetate
PhB	Phenylbutyric acid/phenylbutyrate
Val	Valproic acid/valproate

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References

- 1 S. Dasari and P. B. Tchounwou, *Eur. J. Pharmacol.*, 2014, **740**, 364–378.
- 2 N. J. Wheate, S. Walker, G. E. Craig and R. Oun, *Dalton Trans.*, 2010, **39**, 8113–8127.
- 3 D. Wang and S. J. Lippard, *Nat. Rev. Drug Discovery*, 2005, **4**, 307–320.
- 4 A. M. Florea and D. Busselberg, *Cancers*, 2011, **3**, 1351–1371.
- 5 (a) T. Boulikas and M. Vougiouka, *Oncol. Rep.*, 2004, **11**, 559–595; (b) L. Galluzzi, L. Senovilla, I. Vitale, J. Michels, I. Martins, O. Kepp, M. Castedo and G. Kroemer, *Oncogene*, 2012, **31**, 1869–1883.
- 6 E. Gabano, M. Ravera and D. Osella, *Dalton Trans.*, 2014, **43**, 9813–9820.
- 7 H. Choy, *Expert Rev. Anticancer Ther.*, 2006, **6**, 973–982.
- 8 (a) M. D. Hall and T. W. Hambley, *Coord. Chem. Rev.*, 2002, **232**, 49–67; (b) E. Wexselblatt and D. Gibson, *J. Inorg. Biochem.*, 2012, **117**, 220–229.



- 9 (a) W. H. Ang, I. Khalaila, C. S. Allardyce, L. Juillerat-Jeanneret and P. J. Dyson, *J. Am. Chem. Soc.*, 2005, **127**, 1382–1383; (b) L. L. Ma, R. Ma, Y. P. Wang, X. Y. Zhu, J. L. Zhang, H. C. Chan, X. F. Chen, W. J. Zhang, S. K. Chiu and G. Y. Zhu, *Chem. Commun.*, 2015, **51**, 6301–6304; (c) K. R. Barnes, A. Kutikov and S. J. Lippard, *Chem. Biol.*, 2004, **11**, 557–564; (d) V. Novohradsky, L. Zerzankova, J. Stepankova, O. Vrana, R. Raveendran, D. Gibson, J. Kasparkova and V. Brabec, *J. Inorg. Biochem.*, 2014, **140**, 72–79; (e) A. Curci, N. Denora, R. M. Iacobazzi, N. Ditaranto, J. D. Hoeschele, N. Margiotta and G. Natile, *Inorg. Chim. Acta*, 2018, **472**, 221–228; (f) A. R. Z. Almotaairy, V. Gandin, L. Morrison, C. Marzano, D. Montagner and A. Erxleben, *J. Inorg. Biochem.*, 2017, **177**, 1–7.
- 10 S. Dhar and S. J. Lippard, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 22199–22204.
- 11 R. Ferriero and N. Brunetti-Pierri, *Oncotarget*, 2013, **4**, 804–805.
- 12 A. Villar-Garea and M. Esteller, *Int. J. Cancer*, 2004, **112**, 171–178.
- 13 (a) M. R. Acharya, A. Sparreboom, J. Venitz and W. D. Figg, *Mol. Pharmacol.*, 2005, **68**, 917–932; (b) T. Beckers, C. Burkhardt, H. Wieland, P. Gimmnich, T. Ciossek, T. Maier and K. Sanders, *Int. J. Cancer*, 2007, **121**, 1138–1148; (c) A. C. West and R. W. Johnstone, *J. Clin. Invest.*, 2014, **124**, 30–39.
- 14 R. Raveendran, J. P. Braude, E. Wexselblatt, V. Novohradsky, O. Stuchlikova, V. Brabec, V. Gandin and D. Gibson, *Chem. Sci.*, 2016, **7**, 2381–2391.
- 15 M. Alessio, I. Zanellato, I. Bonarrigo, E. Gabano, M. Ravera and D. Osella, *J. Inorg. Biochem.*, 2013, **129**, 52–57.
- 16 J. Yang, X. Sun, W. Mao, M. Sui, J. Tang and Y. Shen, *Mol. Pharm.*, 2012, **9**, 2793–2800.
- 17 R. Ferriero, C. Iannuzzi, G. Manco and N. Brunetti-Pierri, *J. Inherited Metab. Dis.*, 2015, **38**, 895–904.
- 18 N. Ghosh, R. Chaki, V. Mandal and S. C. Mandal, *Pharmacol. Rep.*, 2010, **62**, 233–244.
- 19 (a) D. W. Knapp, N. W. Glickman, W. R. Widmer, D. B. DeNicola, L. G. Adams, T. Kuczek, P. L. Bonney, A. E. DeGortari, C. Han and L. T. Glickman, *Cancer Chemother. Pharmacol.*, 2000, **46**, 221–226; (b) P. A. Boria, D. J. Murry, P. F. Bennett, N. W. Glickman, P. W. Snyder, B. L. Merkel, D. L. Schlittler, A. J. Mutsaers, R. M. Thomas and D. W. Knapp, *J. Am. Vet. Med. Assoc.*, 2004, **224**, 388–394; (c) K. Hattori, R. Matsushita, K. Kimura, Y. Abe and E. Nakashima, *Biol. Pharm. Bull.*, 2001, **24**(10), 1214–1217; (d) H. Endo, M. Yano, Y. Okumura and H. Kido, *Cell Death Dis.*, 2015, **5**, e1027.
- 20 M. N. Bijman, C. A. Hermelink, M. P. van Berkel, A. C. Laan, M. L. Janmaat, G. J. Peters and E. Boven, *Biochem. Pharmacol.*, 2008, **75**, 427.
- 21 R. K. Pathak, S. Marrache, J. H. Choi, T. B. Berding and S. Dhar, *Angew. Chem., Int. Ed.*, 2014, **53**, 1963–1967.
- 22 Q. Cheng, H. Shi, H. Wang, Y. Min, J. Wang and Y. Liu, *Chem. Commun.*, 2014, **50**, 7427–7430.
- 23 W. Neumann, B. C. Crews, M. B. Sarosi, C. M. Daniel, K. Ghebreselasie, M. S. Scholz, L. J. Marnett and E. Hey-Hawkins, *ChemMedChem*, 2015, **10**, 183–192.
- 24 O. Peulen, A. Gonzalez, P. Peixoto, A. Turtoi, D. Mottet, P. Delvenne and V. Castronovo, *PLoS One*, 2013, **8**, e75102.
- 25 E. Petruzzella, J. P. Braude, J. R. Aldrich-Wright, V. Gandin and D. Gibson, *Angew. Chem., Int. Ed.*, 2017, **56**, 11539–11544.
- 26 (a) C. F. Chin, Q. Tian, M. I. Setyawati, W. Fang, E. S. Tan, D. T. Leong and W. H. Ang, *J. Med. Chem.*, 2012, **55**, 7571; (b) S. Q. Yap, C. F. Chin, A. H. Hong Thng, Y. Y. Pang, H. K. Ho and W. H. Ang, *ChemMedChem*, 2017, **12**, 300.
- 27 S. Kankotia and P. W. Stacpoole, *Biochim. Biophys. Acta*, 2014, **1846**, 617–629.
- 28 I. Zanellato, I. Bonarrigo, D. Colangelo, E. Gabano, M. Ravera, M. Alessio and D. Osella, *J. Inorg. Biochem.*, 2014, **140**, 219–227.

