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Anharmonic excited state frequencies of *para*-difluorobenzene, toluene and catechol using analytic RI-CC2 second derivatives†

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Analytic second nuclear derivatives for excited electronic state energies have been implemented for the resolution-of-the-identity accelerated CC2, CIS(D_∞) and ADC(2) models. Our efficient implementation with $\mathcal{O}(\mathcal{N}^2)$ memory demands enables the treatment of medium sized molecules with large basis sets and high numerical precision and thereby paves the way for semi-numerical evaluation of the higher-order derivatives required for anharmonic corrections to excited state vibrational frequencies. We compare CC2 harmonic and anharmonic excited state frequencies with experimental values for *para*-difluorobenzene, toluene and catechol. Basis set problems occur for out-of-plane bending vibrations due to intramolecular basis set superposition error. For non-planar molecules and in plane modes of planar molecules, the agreement between theory and experiment is better than 30 cm⁻¹ on average and we reassign a number of experimental bands on the basis of the *ab initio* predictions.

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1 Introduction

The characterisation of molecules in excited electronic states remains a challenge, both for experimental and theoretical chemistry. Electronic excitation is often accompanied by significant structural change and complex intramolecular vibrational energy redistribution processes,^{1–3} resulting in rich spectra that are sometimes difficult to interpret. Better theoretical treatments of electronic excited states are key to understanding photochemical phenomena and ultimately to harnessing photochemistry as a route to controlling molecular bond fission processes.^{4–7}

Coupled cluster methods are among the most accurate *ab initio* electronic structure methods.⁸ Coupled cluster methods for excited state properties have been developed extensively by Stanton and Gauss in the framework of equation of motion coupled cluster (EOM-CC) theory.^{9–11} Benchmark studies,^{12–15} found that EOM-CCSD ground and excited state harmonic frequencies agree with values derived from experiment with a root mean squared deviation (RMSD) of 20–30 cm⁻¹.

CC2 was designed¹⁶ as approximation to CCSD with an $\mathcal{O}(\mathcal{N}^5)$ scaling of the computational costs with system size $\mathcal{N}$ instead of $\mathcal{O}(\mathcal{N}^6)$. CC2 conserves the order in the fluctuation

potential through which single excitation dominated transitions are described correctly in CCSD theory. In the original implementations with exact four-index electron repulsion integrals the high prefactor of the computational costs and the high memory demands for transforming and storing the two-electron integrals severely limited system and basis set sizes for which calculations could be performed on commonly available computer hardware. This bottleneck is removed by combining CC2 with the resolution of the identity (RI) approximation for the two electron integrals.^{17,18} The RI approximation accelerates the calculations by one to two orders of magnitude, depending on the basis set, reduces memory demands to $\mathcal{O}(\mathcal{N}^2)$ and reduces disc storage demands to $\mathcal{O}(\mathcal{N}^3)$. These computational savings make it possible to study large molecules such as chlorophylls,¹⁹ which previously could only be studied with CIS or TDDFT.

The same technique can be used to accelerate other second-order methods for excited states, *e.g.* configuration interaction singles with a perturbative doubles correction,²⁰ CIS(D), and its iterative variant²¹ CIS(D_∞), and the algebraic diagrammatic construction through second-order,²² ADC(2), with similarly pivotal efficiency savings. Compared to the non-iterative perturbative CIS(D) approximation, the iterative methods CIS(D_∞), ADC(2), and CC2 have the advantage that they provide a consistent description of excited state potential energy surfaces (PES), even in the region of avoided crossings, and can thus be used more straightforwardly for searching and characterizing stationary points on the excited state PES.

Implementations of analytic excited state gradients of these approximate second order methods have been reported in

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ref. 23 and 24, where it is demonstrated that memory demands for first derivatives scale at most as $\mathcal{O}(\mathcal{N}^2)$ when using the RI approximation. Recently, second-order electronic response properties for ground and excited states and ground state nuclear Hessians as analytic second derivatives of the energy have been implemented.^{25–27} To preserve both the high computational efficiency and the low storage demands for second derivatives the RI approximation is combined with a numerical Laplace transformation of orbital energy denominators²⁸ for the contribution of double excitation amplitudes to first derivatives of the density matrices.

In the current work, we extend the theory and implementation to analytic geometrical second derivatives for CC2, CIS(D_∞) and ADC(2) excited state energies. Analytic second derivatives are particularly important for obtaining the high numerical accuracy required to calculate semi-numerical third and fourth derivatives for polyatomic molecules. Only recently has a similar route has been pursued for TDDFT.^{29,30} With the implementation of analytic Hessians for CC2 and ADC(2) it becomes possible to compute anharmonic vibrational spectra of polyatomic molecules with a correlated *ab initio* wavefunction method. We demonstrate the applicability of our implementation by computing harmonic and anharmonic excited state frequencies for medium sized molecules, which we compare to experimentally observed band centres.

2 Theory

2.1 Excited state Hessian for RI-CC2

The theory and implementation of orbital-relaxed electric second-order response properties for excited states at the RI-CC2 level has been presented in ref. 26. In the current work we focus on the additional theory required for geometric second derivatives. We use identical notation to that of ref. 26 and 27 and, rather than repeat them here, refer the reader to that work for the full definition of all of the terms and intermediates.

In coupled cluster theory, properties of an excited state *f* can be obtained as derivatives of the excited state quasienergy Lagrangian²⁶

$$\begin{aligned} \mathcal{L}^{\text{CC,rel,f}} = E_{\text{CC}} &+ \sum_{\mu} \bar{t}_{\mu}^{\text{f}} \Omega_{\mu} + \sum_{\mu\nu} L_{\mu}^{\text{f}} A_{\mu\nu} R_{\nu}^{\text{f}} \\ &+ \sum_{\mu_0} \bar{\kappa}_{\mu_0}^{\text{f}} F_{\mu_0} + \bar{\omega}^{\text{f}} \left(1 - \sum_{\mu} L_{\mu}^{\text{f}} R_{\mu}^{\text{f}} \right) \\ &+ \bar{\lambda}^{\text{f}} \left(\sum_{\mu} L_{\mu}^{\text{f}} R_{\mu}^{\text{f,(0)}} - \sum_{\mu} L_{\mu}^{\text{f,(0)}} R_{\mu}^{\text{f}} \right). \end{aligned} \quad (1)$$

The Lagrangian is composed of the ground state energy E_{CC} , the vector functions Ω_{μ} for the ground state cluster amplitudes t_{μ} , with Lagrange multipliers $\bar{t}_{\mu}^{\text{f}}$, and the excitation energies $\omega^{\text{f}} = \sum_{\mu\nu} L_{\mu}^{\text{f}} A_{\mu\nu} R_{\nu}^{\text{f}}$. The fourth term is the orbital-rotation constraint, which imposes vanishing Fock matrix elements F_{pq} for the relevant orbital pairs $pq \in \mu_0$. The last and second last terms determine the phase and normalisation of the left and

right eigenvectors of the Jacobi matrix **A**, by coupling them to the eigenvectors $L^{\text{f,(0)}}$ and $R^{\text{f,(0)}}$ of the unperturbed limit.²⁶

The hessian of the excited state is obtained by differentiating the Lagrangian twice with respect to the nuclear positions. The result can be obtained in the same way as for excited state polarisabilities, presented in ref. 26:

$$\begin{aligned} \left(\frac{\text{d} \mathcal{L}^{\text{CC,rel,f}}}{\text{d} \varepsilon_x \text{d} \varepsilon_y} \right)_{\varepsilon=0} &= \langle \hat{J}^{\text{xy}} \rangle^{\text{ex}} + \hat{P}^{\text{xy}} \sum_{pq} D_{pq}^{\text{F,ex,y}} F_{pq}^{\text{x}} \\ &+ \hat{P}^{\text{xy}} \frac{1}{2} \sum_{pqrs} \hat{d}_{pqrs}^{\text{nsep,ex,y}} (pq|rs)^{\text{x}} \\ &+ \sum_{\mu\nu} F_{\mu\nu}^{\text{ex}} t_{\mu}^{\text{x}} t_{\nu}^{\text{y}} + \hat{P}^{\text{xy}} \sum_{\mu\nu\kappa} L_{\mu}^{\text{f,(0)}} B_{\mu\nu\kappa} t_{\kappa}^{\text{x}} R_{\nu}^{\text{f,y}}. \end{aligned} \quad (2)$$

$\langle \hat{J}^{\text{xy}} \rangle^{\text{ex}}$ is the expectation value of an effective second order Hamiltonian (*vide infra*) evaluated with the density for the excited state. The second and third terms depend on first derivatives of integrals and first derivatives of the cluster amplitudes (t_{ν}^{y}) and right eigenvectors $R_{\nu}^{\text{f,y}}$ contained in the derivative one- and two-particle densities. The final two terms collect contributions that are bilinear in first derivatives of the cluster amplitudes and eigenvectors.

The only difference to polarizabilities are that for derivatives with respect to nuclear coordinates the AO basis and consequently all AO one- and two-electron integrals depend on the perturbation. This also gives rise to additional contributions to the derivatives of the MO coefficients related to the changes in the AO overlap. We write the first derivative of the molecular orbital (MO) coefficients with respect to nuclear displacements as

$$\mathbf{C}^{\text{x}} = \mathbf{C}^0 \mathbf{U}^{\text{x}} \quad (3)$$

where $\mathbf{U}^{\text{x}}$ contains the derivative of the orbital rotation parameters κ^{x} and the derivative of the overlap matrix $\mathbf{S}^{[\text{x}]}$ in the basis of the unmodified molecular orbitals (UMOs).

$$\mathbf{U}^{\text{x}} = \kappa^{\text{x}} - \frac{1}{2} \mathbf{S}^{[\text{x}]} \quad (4)$$

Here and throughout, a square bracket indicates a derivative of the UMO integrals, which includes the derivative of the AO integrals, but not the derivative of the MO coefficients.^{31,32} The first term in eqn (2) is the expectation value of an effective second-order Hamiltonian which collects several contributions to the second geometric derivatives of the Hamiltonian in the MO basis:

$$\begin{aligned} \hat{J}^{\text{xy}} &= H^{[\text{xy}]} + \hat{P}^{\text{xy}} \left(U^{\text{x}}, \hat{H}^{[\text{y}]} \right) + (U^{\text{x}}, U^{\text{y}}, \hat{H}) \\ &- \frac{1}{2} \hat{P}^{\text{xy}} \left((U^{\text{y}}, S^{[\text{x}]}, \hat{H}) - \frac{1}{2} (S^{[\text{xy}]}, \hat{H}) \right) \end{aligned} \quad (5)$$

For electric response properties only the second and third terms are present and $H^{[\text{y}]}$ has no two-electron part. The round brackets indicate a one-index transformation of the integrals.³² The evaluation of the expectation value $\langle \hat{J}^{\text{xy}} \rangle^{\text{ex}}$ with $\mathcal{O}(\mathcal{N}^2)$ memory demands will be described below.



The last two terms in eqn (2) depend only indirectly on the derivatives of the one- and two-electron integrals, through the derivatives of the cluster amplitudes, t_μ^x , and right eigenvectors, $R_\nu^{f,y}$. These are evaluated as described in ref. 26 for electronic derivatives, with the difference that now the first derivatives of the AO two-electron integrals have to be included in the calculation of the amplitude and eigenvector derivatives. Detailed expressions for the derivatives of the cluster amplitudes are given in ref. 27 and the changes to the expressions for the derivatives of the eigenvectors are obtained in an analogous way.

The second and third terms in eqn (2) combine (*via* the densities) first derivatives of the amplitudes t_ν^y and eigenvectors $R_\nu^{f,y}$ for a coordinate y with first derivatives of the MO Fock matrix F_{pq}^x and two-electron integrals $(pq|rs)^x$ for a coordinate x . Here, as for all two-electron integrals in the correlation treatment, we employ the RI approximation

$$(pq|rs) \approx (pq|rs)^{\text{RI}} = \sum_Q B_{Q,pq} B_{Q,rs}, \quad (6)$$

with three index intermediates $B_{Q,pq} = \sum_P (pq|P) \left[V^{-\frac{1}{2}} \right]_{PQ}$ composed of two-index $V_{PQ} = (P|Q)$ and three-index $(pq|P)$ electron repulsion integrals. The indices P and Q denote functions from an auxiliary basis set. For an efficient evaluation we rewrite the two-particle term as:

$$\sum_{pqrs} \hat{d}_{pqrs}^{\text{nsep,ex},y} (pq||rs)^x = \sum_{\alpha\beta Q} \Delta_{\alpha\beta}^{Q,y} (\alpha\beta|Q)^{[x]} - \sum_{PQ} \gamma_{PQ}^y V_{PQ}^{[x]} + \sum_{pq} F_{pq}^{\text{eff},y} U_{pq}^x \quad (7)$$

where $F_{pq}^{\text{eff},y}$ is an effective Fock matrix (*cf.* ref. 26 and 27) and $\Delta_{\alpha\beta}^{Q,y}$ and γ_{PQ}^y are, respectively, a three-index two-particle density in the AO basis (indices α, β) and a two-index two-particle density:

$$\Delta_{\alpha\beta}^{Q,y} = \sum_{pq} A_{xp}^p A_{\beta q}^h \sum_{rsQ} \hat{d}_{pqrs}^{\text{nsep},y} (rs|Q) [V^{-1}]_{PQ}, \quad (8)$$

$$\gamma_{PQ}^y = \frac{1}{2} \sum_{\alpha\beta R} \Delta_{\alpha\beta}^{P,y} (\alpha\beta|R) [V^{-1}]_{RQ}. \quad (9)$$

We highlight that $\Delta_{\alpha\beta}^{Q,y}$ is evaluated without ever building the four-index two-particle density $\hat{d}_{pqrs}^{\text{nsep},y}$. Full working expressions for the effective Fock matrix and for the two- and three-index densities are provided in the ESI.†

The one-electron densities $D^{\text{F,ex},y}$ require contractions of the doubles parts of the left eigenvectors and the derivatives of the right eigenvectors of the form $\sum_{abk} L_{jakb}^f R_{iakb}^{f,y}$ and $\sum_{ijc} L_{iajc}^f R_{ibjc}^{f,y}$. For undifferentiated eigenvectors and amplitudes the RI approximation is sufficient to implement these contractions efficiently with only $\mathcal{O}(\mathcal{N}^2)$ memory demands and without storing doubles vectors as four-index quantities on disc. For differentiated eigenvectors, however, this is not the case. There are additional terms that are not simply dressed two-electron integrals, but

come from contractions of the undifferentiated eigenvectors with derivatives of the Fock matrix:

$$R_{aibj}^{f,x} = \frac{1}{\varepsilon_a - \varepsilon_i + \varepsilon_b - \varepsilon_j + \omega^f} \hat{P}_{ab}^{ij} \times \left(\sum_Q \bar{B}_{Q,ai}^{f,x} \hat{B}_{Q,bj} + \sum_Q \bar{B}_{Q,ai} [R^f] \left(\bar{B}_{Q,bj} [t^x] + \hat{B}_{Q,bj}^x \right) + \sum_c R_{aicj}^f F_{bc}^x - \sum_k R_{aibk}^f F_{kj}^x - \bar{x}^f R_{aibj}^f \right) \quad (10)$$

For full definitions of the intermediates see ref. 26. To convert the last three terms into an expression that is separable in the index pairs ai and bj we apply a numerical Laplace transformation of the orbital energy denominators

$$\frac{1}{\varepsilon_a - \varepsilon_i + \varepsilon_b - \varepsilon_j + \omega^f} \approx \sum_m^{N_m} w_m e^{(\varepsilon_i - \varepsilon_a)\theta_m} e^{(\varepsilon_j - \varepsilon_b + \omega^f)\theta_m} \quad (11)$$

where θ_m and w_m are the integration points and weights, respectively. With this approximation the eigenvector elements can be written as

$$R_{aibj}^f = -\hat{P}_{ab}^{ij} \sum_{Qm} w_m K_{Qm,ai} \bar{K}_{Qm,bj}, \quad (12)$$

with

$$K_{Qm,ai} = \hat{B}_{Q,ai} e^{(\varepsilon_i - \varepsilon_a)\theta_m}, \quad (13)$$

$$\bar{K}_{Qm,ai} = \bar{B}_{Q,ai} [R^f] e^{(\varepsilon_i - \varepsilon_a + \omega^f)\theta_m}, \quad (14)$$

and the transformations with the derivatives of the Fock matrix can be performed on the three-index intermediates. This makes it possible to evaluate the contributions from R^f to the right hand side of $R^{f,y}$ within the same loops as the contributions from the dressed integrals, just with an additional summation over the Laplace grid points.

We now turn to the evaluation of the expectation value $\hat{f}^{xy}$ of the second order Hamiltonian for an excited state. The contributions to $\hat{f}^{xy}$ are grouped into three terms. The first term contains the second derivatives of the AO integrals for the Hamiltonian. It is evaluated in the same manner in the AO basis as the respective contribution from $H^{[x]}$ to the gradient described in ref. 23:

$$\begin{aligned} \langle H^{[xy]} \rangle^{\text{ex}} &= \sum_{\alpha\beta} D_{\alpha\beta}^{\text{eff,ex}} h_{\alpha\beta}^{[xy]} + \frac{1}{4} \sum_{\alpha\beta} \left(D_{\alpha\beta}^{\text{eff,ex}} - \frac{1}{8} D_{\alpha\beta}^{\text{SCF}} \right) \\ &\times \sum_{\gamma\delta} A_{\alpha\beta\gamma\delta}^{\text{CPHF},[xy]} D_{\gamma\delta}^{\text{SCF}} \\ &+ \sum_{\alpha\beta Q} \Delta_{\alpha\beta}^{Q,y} (\alpha\beta|Q)^{[xy]} - \sum_{PQ} \gamma_{PQ}^y V_{PQ}^{[xy]} \\ &+ \frac{1}{2} \hat{P}^{xy} \left(\sum_{\alpha\beta Q} \Delta_{\alpha\beta}^{Q,[x]} (\alpha\beta|Q)^{[y]} - \sum_{PQ} \gamma_{PQ}^{[x]} V_{PQ}^{[y]} \right) \end{aligned} \quad (15)$$

$D^{\text{eff,ex}}$ is the effective orbital-relaxed excited state and D^{SCF} the SCF one-particle density. $h_{pq}^{[xy]}$ and $A_{pqrs}^{\text{CPHF},[xy]}$ are, respectively,



the core Hamiltonian and CPHF matrices computed from the second derivatives of the AO integrals. $\Delta_{\alpha\beta}^Q$ and $\Delta_{\alpha\beta}^{Q[x]}$ are auxiliary three-index and γ_{pq} and $\gamma_{pq}^{[x]}$ two-index two-particle densities defined in the ESI.†

All other contributions to $\langle \hat{J}^{xy} \rangle^{\text{ex}}$ are rewritten as contractions of effective Fock matrices with derivatives of the overlap matrices and U^x . This is done to avoid the evaluations of two-electron integrals in the MO basis and with this any $\mathcal{O}(\mathcal{N}^5)$ scaling steps that depend on two perturbations.

The second and third contribution to $\hat{J}^{xy}$ are combined by introducing a modified first-order Hamiltonian,²⁷

$$\hat{\mathcal{H}}^x = \hat{H}^{[x]} + \frac{1}{2}(U^x, \hat{H}), \text{ to:}$$

$$\hat{P}^{xy} \left\{ \langle (U^x, \hat{H}^y) \rangle^{\text{ex}} + \frac{1}{2} \langle (U^x, (U^y, \hat{H})) \rangle^{\text{ex}} \right\} = \hat{P}^{xy} \langle (U^x, \hat{\mathcal{H}}^y) \rangle^{\text{ex}}. \quad (16)$$

Such expectation values of one-index transformed Hamiltonians can be evaluated as contraction of the transformation matrices U^x with effective Fock matrices:

$$\langle (U^x, \hat{\mathcal{H}}^y) \rangle^{\text{ex}} = \sum_{pq} U_{pq}^x F_{pq}^{\text{eff}} [\hat{\mathcal{H}}^y] \quad (17)$$

The definition and calculation of the effective Fock matrices for the modified first-order Hamiltonian $F_{pq}^{\text{eff}} [\hat{\mathcal{H}}^y]$ have been given in ref. 26 and 27. The last two contributions to $\hat{J}^{xy}$ are combined by introducing the second derivative of the overlap matrix $S^{xy} = S^{[xy]} + \hat{P}^{xy}(U^y, S^{[x]})$ to

$$(S^{[xy]}, \hat{H}) + \hat{P}^{xy}((U^y, S^{[x]}), \hat{H}) = (S^{xy}, \hat{H}) \quad (18)$$

and is evaluated as:

$$-\frac{1}{2} \langle (S^{xy}, \hat{H}) \rangle^{\text{ex}} = -\frac{1}{2} \sum_{pq} S_{pq}^{xy} F_{pq}^{\text{eff}, \text{ex}} \quad (19)$$

where $F_{pq}^{\text{eff}, \text{ex}}$ is the (unperturbed) excited state effective Fock matrix, also known as energy weighted density matrix. Its implementation for RI-CC2 has been described in ref. 23.

The above formulas have been implemented in the development version of the *ricc2* program of the TURBOMOLE package.³³ All contributions are evaluated using integral-direct algorithms in the AO basis with $\mathcal{O}(\mathcal{N}^2)$ memory demands and strictly avoiding any operations that scale as $\mathcal{O}(\mathcal{N}^5)$ with the basis set size and at the same time quadratically with the number of perturbations, so that the computation of the full Hessian still scales only as $\mathcal{O}(\mathcal{N}^6)$ with the system size. The time-determining steps are the solution of the linear equations for the first derivatives of the amplitudes and eigenvectors and the calculation of the first-order density matrices that have to be done for each perturbation.

The implementation of the excited state Hessians and the excited state polarizabilities enables also the calculation of the derivatives of the (excited state) dipole moment as mixed derivatives by differentiating once with respect to the strength of an electric field and once with respect to the nuclear coordinates.

2.2 CIS(D_∞) and ADC(2)

With only few modifications the implementation is easily adapted to CIS(D_∞) and ADC(2).^{21,22} In both methods the coupled cluster ground state amplitudes are replaced by the amplitudes from first-order Møller–Plesset perturbation theory (MPPT). We assume that the Brillouin condition is fulfilled and thus the singles ground state amplitudes $t^{(0)}$ and their derivatives t^x vanish in first order MPPT, as do the singles Lagrange multipliers $\bar{t}^{(0)}$. Discarding all contributions from these singles parameters turns the CC2 code into a CIS(D_∞) code. As a side effect, the equations for the derivatives of the cluster amplitudes can be inverted directly, in the canonical implementation, bypassing the iterative solution procedure.

The implementation for ADC(2) is closely related to the implementation CIS(D_∞). Here the secular matrix is symmetrised:^{22,24}

$$\mathbf{A}^{\text{ADC}(2)} = \frac{1}{2} \left(\mathbf{A}^{\text{CIS}(\text{D}_{\infty})} + \left(\mathbf{A}^{\text{CIS}(\text{D}_{\infty})} \right)^{\dagger} \right) \quad (20)$$

This only requires some small modifications in the singles-singles block of the Jacobian to symmetrise the contributions from $\langle \mu_1 | [[H, T_2^{(1)}], \tau_{\nu}] | \text{HF} \rangle$ and the corresponding contributions to the right hand side of the equation for R^{fy} and to the derivatives of the Jacobian. Due to the symmetric secular matrix, the left eigenvectors are identical with the right eigenvectors and only one set of eigenvalue equations has to be solved.

2.3 Thresholds and numerical accuracy

The numerical accuracy of the excited state Hessians, and also the computational costs, depend mainly on two thresholds: T_{Lap} and T_{LRE} . T_{Lap} controls the numerical Laplace transformation such that

$$\sqrt{F_{\text{Lap}}} = \sqrt{\int_{\varepsilon_{\min}}^{\varepsilon_{\max}} \left(\frac{1}{x} - \sum_{\alpha} w_{\alpha} e^{-x\theta_{\alpha}} \right)^2 dx} \leq T_{\text{Lap}} \quad (21)$$

T_{LRE} is the threshold for the residual of the equations for the first derivatives of the cluster amplitudes and eigenvectors. The grid points for the Laplace transformation were optimised for the interval from $\varepsilon_{\min} = 2(\varepsilon_{\text{LUMO}} - \varepsilon_{\text{HOMO}}) - \omega_f$ and $\varepsilon_{\max} = 2(\varepsilon_{\text{vir}, \max} - \varepsilon_{\text{occ}, \max})$, where ω_f is the excitation energy, by minimizing F_{Lap} with respect to θ_m and w_m .

We investigated the dependence of the results on the thresholds for a test set composed of the lowest excited states of glyoxal, methanethial, propinal, benzene and naphthalene at their equilibrium structures, the lowest two excited states of water and formaldehyde at the ground and the excited state equilibrium structure, and the third excited state of formaldehyde and the lowest two excited states of thiophene again at their equilibrium structures. For all 16 cases we computed the Hessian for the cc-pVDZ and the aug-cc-pVDZ basis sets with different values for both thresholds, T_{Lap} and T_{LRE} , and evaluated the deviations in the elements of the Hessian from a reference calculation with very tight thresholds. For the aug-cc-pVDZ result



the mean absolute deviations of the Hessian elements follow roughly the relation

$$\Delta_{\text{Hess,CC2}}^{\text{MAD}} \leq 5T_{\text{Lap}} + 10T_{\text{LRE}} \quad (22)$$

with the maximum deviations 20–50 times larger. For the cc-pVDZ basis set the deviations are between one and two orders of magnitude smaller. In general, we observe that, not unexpectedly, the thresholds have to be set about two orders of magnitude tighter than for ground state calculations and that tighter thresholds are needed if other states are close by.

In this work we used a tight threshold for the response equations $T_{\text{LRE}} = 10^{-10}$ and 11 grid points for the numerical Laplace integration, which corresponds to T_{Lap} between 2×10^{-5} and 2×10^{-10} . These tight values were required because the third and fourth derivatives for VPT2 theory are obtained by numerical differentiation of the hessian. We kept the number of Laplace points the same for all Hessians used in a finite difference formula to prevent any numerical noise from changes of the grid points.

2.4 Calculation of anharmonic corrections and frequencies

The analytical implementation of the Hessians enables a semi-numerical evaluation of third and fourth derivatives of the energy, which are sufficiently accurate to be used as cubic and quartic force constants for the calculation of anharmonic frequencies. The third and fourth derivatives are calculated by central finite differences of Hessians. Since only semi-diagonal quartic derivatives are required for VPT2 theory, the total number of Hessians required is $N_{\text{acc}}(3N - 6) + 1$, where N_{acc} is either two or four. $N_{\text{acc}} = 4$ is required when four- and five-point formulas are used for cubic and quartic derivatives to reduce the error to $\mathcal{O}(\delta^4)$ in the displacement δ .³⁴

We did not use the normal coordinates $\vec{l}_m$ of the equilibrium structure for the displacement vectors since they are normalized in the mass weighted coordinate system, which leads to unbalanced step sizes. Instead, we used rescaled coordinates

$$\vec{v}^m = \sqrt{\mu_m} \mathbf{M}^{-1/2} \vec{l}_m, \quad (23)$$

which ensures that the displacements become nearly independent of the atomic masses. Here, the vector $\vec{l}_m$ denotes the m th eigenvector of the mass weighted Hessian $\mathbf{M}^{-1/2} \mathbf{F} \mathbf{M}^{-1/2}$ and $\mathbf{M}$ is the diagonal matrix of the atomic masses, and $\mu_m = (\vec{l}_m^T \mathbf{M}^{-1} \vec{l}_m)^{-1}$ is the reduced mass of the normal mode m .

The anharmonic force constants are used for the calculation of anharmonic frequencies, by setting up the vibrational Hamiltonian as:³⁵

$$\mathbf{H}_{\text{vib}} = \mathbf{H}_{\text{vib}}^0 + \mathbf{H}_{\text{vib}}^1 + \mathbf{H}_{\text{vib}}^2 \quad (24)$$

$$\begin{aligned} \mathbf{H}_{\text{vib}} = & \frac{1}{2} \sum_r \omega_r (p_r^2 + q_r^2) + \frac{1}{6} \sum_{rst} \phi_{rst} q_r q_s q_t \\ & + \frac{1}{24} \sum_{rstu} \phi_{rstu} q_r q_s q_t q_u + \sum_{\alpha} B_{\alpha}^e j_{\alpha}^2 \end{aligned} \quad (25)$$

The zeroth order term $\mathbf{H}_{\text{vib}}^0$ is a quantum harmonic oscillator in reduced normal coordinates. The first-order correction $\mathbf{H}_{\text{vib}}^1$

consists of the cubic force constants ϕ_{rst} . The second-order correction $\mathbf{H}_{\text{vib}}^2$ contains the quartic force constants ϕ_{rstu} and a Coriolis term that depends on the equilibrium rotational constants B_{α}^e and the vibrational angular momenta j_{α} .

The anharmonic frequencies are calculated using second order vibrational perturbation theory (VPT2), as it is implemented in the program DYNAMOL.³⁶ Therein, the VPT2 equations and treatment of resonance effects follow the formulation described in the paper of Amos *et al.*³⁷

3 Results

We have used our new efficient CC2 excited state Hessian implementation to construct harmonic and quartic force fields of toluene, *para*-difluorobenzene and catechol in their first excited electronic state, from which we obtain the band centres of vibrational transitions using VPT2. These medium sized molecules have been studied experimentally and we compare the computed and experimental wavenumbers, assessing the accuracy of the CC2 predictions and correcting assignments where appropriate. The nomenclature of the normal modes is adopted from the systematic studies of the Wright group for substituted benzenes,^{38–40} which provides a more unique assignment than the Mulliken⁴¹ or Wilson⁴² nomenclature. For modes localised on the substituents, the common spectroscopic notation for stretches (ν), bends (δ) and torsions (τ) is used. Outputs from all of our VPT2 calculations are provided in the ESI.† These contain a full list of fundamentals, overtones, combination bands, all Fermi and Darling–Dennison resonances, and all effective Hamiltonians for the polyads used to treat these resonances variationally.

3.1 *para*-Difluorobenzene

The molecule *para*-difluorobenzene has been the focus of a considerable number of spectroscopic explorations to determine its ground and excited state structure and dynamics^{43–48} including the intramolecular vibrational redistribution pathways in the excited state, which are mediated by rotational coupling and through Fermi resonance.^{49,50} Here we examine the fundamental vibrational transitions of *para*-difluorobenzene in its first excited state 1^1B_{2u} using CC2 theory.

Spectroscopic studies reveal that *para*-difluorobenzene exhibits D_{2h} symmetry in both its ground and first excited state,⁴⁵ and the CC2/cc-pVTZ optimised geometries indeed retain D_{2h} symmetry. The optimised structures are listed in the ESI.† The computed 0–0 transition is 4.67 eV, which is close to the experimentally determined energy of 4.57 eV.⁴⁶

Table 1 reports harmonic and anharmonic wavenumbers for the fundamental vibrational transitions of *para*-difluorobenzene in its first excited state, computed using CC2/cc-pVTZ and CC2/cc-pVQZ levels of theory using the optimised CC2/cc-pVTZ excited state geometry. The harmonic wavenumbers computed using the cc-pV5Z basis set are also listed. Using the normal mode vectors, each transition has been assigned using the D_i labeling convention of ref. 39 for *para*-di-substituted benzenes.



Table 1 CC2 excited state vibrational wavenumbers for *para*-difluorobenzene

pD_i	Sym.	Exp.	Harm.				VPT2		Exp.	S_1
		S_0	TZ	QZ	5Z		TZ	QZ	S_1	
20	30(b _{3u})	158	120	121	126		120	121	120 ⁴⁷	120
14	8(a _u)	422	148	147	161		169	168	175 ⁴⁷	175
19	17(b _{2g})	374	271	276	280		307	305	274 ⁴⁷	274
30	22(b _{2u})	348	352	353	352		349	350	352 ⁴⁷	352
29	27(b _{3g})	446	388	388	392		380	381	403 ⁴⁷	403
13	9(b _{1g})	800	404	406	411		453	476	588 ⁴⁶	*
11	6(a _g)	450	411	412	423		407	408	410 ⁴⁷	410
18	29(b _{3u})	505	472	471	473		466	466	438 ⁴⁷	438
12	7(a _u)	945	486	490	503		612	676	583 ⁴⁷	*
17	16(b _{2g})	692	527	546	539		651	647	528 ⁴⁷	*
28	26(b _{3g})	635	558	558	558		541	542	558 ⁴⁷	558
15	15(b _{2g})	838	610	616	623		700	658	670 ⁴⁷	670
16	28(b _{3u})	928	614	621	624		636	813	619 ⁴⁷	*
10	14(b _{1u})	740	716	716	713		702	703	666 ⁴⁶	666
9	5(a _g)	859	824	825	819		812	811	818 ⁴⁷	818
8	13(b _{1u})	1014	960	962	961		949	951	937 ⁴⁷	937
27	21(b _{2u})	1085	1025	1025	1022		1014	1006	1100 ⁴⁸	
7	4(a _g)	1140	1119	1117	1116		1100	1102	1116 ⁴⁶	1116
6	12(b _{1u})	1228	1228	1222	1218		1203	1195	1015 ⁴⁶	
26	25(b _{3g})	1285	1254	1257	1256		1241	1244	933 ⁴⁶	
5	3(a _g)	1257	1273	1265	1260		1241	1234	1251 ⁴⁷	1251
25	20(b _{2u})	1306	1317	1319	1315		1297	1298	1591 ⁴⁸	
4	11(b _{1u})	1514	1449	1444	1439		1412	1408	1335 ⁴⁶	1335
23	24(b _{3g})	1595	1510	1509	1497		1475	1475	1516 ⁴⁸	1516
3	2(a _g)	1615	1557	1552	1540		1507	1503		
24	19(b _{2u})	1437	1680	1677	1664		1635	1632		1591
22	23(b _{3g})	3085	3253	3252	3246		3118	3116		
2	10(b _{1u})	3073	3257	3255	3249		3118	3116		
21	18(b _{2u})	3091	3267	3265	3258		3107	3103		
1	1(a _g)	3088	3271	3268	3261		3125	3123		

For ease of reference, the irreducible representation, Mulliken labeling and ground state experimental wavenumbers have also been included.

Examining the basis set convergence of the harmonic frequencies, we find that while the in plane modes converge rapidly with the basis set, the out of plane modes (a_u, b_{1g}, b_{2g} and b_{3u}) converge slowly, with deviations of more than 10 cm⁻¹ between cc-pVTZ and cc-pV5Z values for the low frequency vibrations. As highlighted by Martin, Taylor and Lee for benzene and acetylene,^{51,52} modes that break planarity suffer from basis set inconsistency errors that artificially lower the frequency and overestimate anharmonic terms in the quartic force field.

This artificial enhancement of the anharmonic couplings to out-of-plane vibrations complicates *ab initio* assignment the experimental bands. These difficulties notwithstanding, our calculations confirm the majority of the assignments of the 24 fundamental bands observed experimentally. The rightmost two columns of Table 1 list the previous assignment as collated in ref. 47 and our own *ab initio* assignment, respectively.

The fundamentals D_6 and D_{27} were tentatively assigned to a b_{1u} band at 1015 cm⁻¹ and a b_{2u} band at 1100 cm⁻¹, respectively, in gas phase two-photon spectroscopy measurements.⁴⁸ These assignments can be confidently discarded on the basis of our calculations and it is likely that these bands do not correspond to fundamental transitions. Similarly, Knight and Kable's tentative assignment of the b_{3g} band at 933 to the D_{26} fundamental can also be discarded. There is a somewhat larger

than expected discrepancy between the predicted and observed D_4 fundamental, which we cannot explain. Finally, we make a reassignment of the b_{2u} band at 1591 cm⁻¹ to D_{24} . The problems encountered with the low frequency out of plane modes prevent meaningful comment of the assignments of modes D_{12} , D_{13} , D_{16} and D_{17} , which is indicated by an asterisk in the table. For the in plane modes, the overall agreement between anharmonic CC2/cc-pVQZ and experimentally observed transitions is very good, with an RMSD of 26 cm⁻¹.

3.2 Toluene

The excited state vibrational frequencies of toluene have been studied using dispersed fluorescence spectroscopy,⁵³ UV-IR double resonance spectroscopy,⁵⁴ one-colour resonance-enhanced multiphoton ionization and two-colour zero kinetic energy spectroscopy.^{55,56} 21 of the 39 fundamentals have been assigned, as have the methyl group internal rotor bands, five further overtone bands and several combination bands in the range 200–1500 cm⁻¹.

The CC2/cc-pVTZ optimised structures have C_s symmetry in both the ground and first (¹B₂) excited states. In the ground state the carbon atoms fall in a plane, one hydrogen of the methyl group orientated perpendicular to the plane. In the excited state, the methyl group moves slightly out of plane and has a dihedral angle of 4 degrees with the benzene ring. The 0–0 transition energy for the first excited state is at 4.65 eV,⁵³ which is closely reproduced by CC2/cc-pVTZ theory, which yields 4.86 eV.

Table 2 summarises the calculated and experimental frequencies of toluene in its first excited state. We list harmonic frequencies in cc-pVTZ, cc-pVQZ and cc-pV5Z basis sets, computed at the optimised cc-pVTZ structure, and VPT2 anharmonic frequencies using the cc-pVTZ basis set. The normal mode vectors were analysed and classified using the M_i nomenclature of Gardner and Wright³⁸ for the ring modes, and given pseudo C_{2v} symmetry labels, where the methyl group is treated as a single pseudo-atom.

Concerning the basis set convergence of the harmonic wavenumbers, we find a similar pattern as for *para*-difluorobenzene. The in plane modes are converged to within 10 cm⁻¹ with the cc-pVTZ basis, which is well below the intrinsic error bar of 30 cm⁻¹ commonly ascribed to CC2 theory (for ground state frequencies) due to missing higher order correlation effects. As for *para*-difluorobenzene, the out-of-plane modes display a much slower basis set convergence, with differences of more than 30 cm⁻¹ between cc-pVTZ and cc-pV5Z values for the torsion and for modes M_{12} , M_{15} and M_{16} .

Overall, the CC2/cc-pVTZ anharmonic frequencies agree very well with the experimental band centres for the fundamental transitions. The only outlier is mode M_{16} , which appears to have artificially enhanced positive anharmonic corrections due to the basis set incompleteness errors. By the same token, we expect the predicted fundamentals for modes M_{12} and M_{15} to also lie above the experimental bands, should they be measured in the future. The methyl internal rotation is not expected to be well described through VPT2 theory and requires a more advanced vibrational treatment using a potential energy surface that exhibits the three equivalent minima, see ref. 57–59 for



Table 2 CC2 excited state vibrational wavenumbers for toluene

M_i	Sym.	Harm.			VPT2	Exp.
		TZ	QZ	5Z	TZ	S_1
τ		86	118	122	28	
20	b_1	141	148	151	140	145
14	a_2	234	238	238	236	226
19	b_1	319	325	325	331	314
30	b_2	323	328	329	325	331
18	b_1	432	441	445	460	
11	a_1	449	450	451	442	457
29	b_2	524	524	526	512	532
13	a_2	539	544	560	563	
17	b_1	556	560	566	552	
16	b_1	639	651	669	734	697
12	a_2	658	667	686	736	
15	b_1	717	733	747	839	
10	a_1	751	749	749	737	753
28	b_2	932	932	932	915	
8	a_1	943	939	938	928	934
9	a_1	969	969	972	957	966
δ_{as}^-		1003	1003	1002	972	
δ_{as}^-		1048	1047	1047	1024	1021
27	b_2	1157	1155	1156	1137	
7	a_1	1162	1160	1160	1143	
6	a_1	1220	1214	1214	1196	1193
26	b_2	1302	1304	1305	1281	
24	b_2	1386	1385	1386	1364	
δ_s^+		1393	1394	1395	1358	
5	a_1	1426	1423	1424	1394	
δ_{as}^+		1474	1472	1472	1432	
δ_{as}^+		1485	1484	1483	1443	
23	b_2	1534	1526	1524	1482	
4	a_1	1557	1547	1545	1505	
25	b_2	1685	1668	1661	1635	
ν_s		3024	3015	3013	2921	2893
ν_{as}		3098	3091	3088	2956	2956
ν_{as}		3142	3135	3133	3013	2988
22	b_2	3209	3203	3202	3076	3048
3	a_1	3209	3204	3203	3077	3063
2	a_1	3218	3212	3210	3079	3077
21	b_2	3231	3224	3222	3087	3087
1	a_1	3240	3233	3231	3098	3097
Overtones						
20	a_1	281	296	302	281	290
14	a_1	469	476	476	471	452
19	a_1	638	650	650	666	629
8	a_1	1885	1878	1876	1854	1868
9	a_1	1939	1939	1945	1914	1929
Combination bands						
14 + 20	b_2	375	386	389	374	371
19 + 20	a_1	460	473	476	473	462
14 + 19	b_2	553	563	563	566	539
18 + 19	a_1	751	766	770	796	734
29 + 30	a_1	847	852	855	837	864
11 + 29	b_2	973	974	977	956	988
12 + 14	a_1	892	905	924	975	916
10 + 29	b_2	1275	1273	1275	1248	1263
8 + 11	a_1	1392	1389	1389	1370	1390
8 + 29	b_2	1467	1463	1464	1439	1463
9 + 11	a_1	1418	1419	1423	1399	1426
9 + 29	b_2	1493	1493	1498	1468	1494
8 + 9	a_1	1912	1908	1910	1885	1900
6 + 29	b_2	1743	1738	1740	1706	1727
23 + 25	a_1	3219	3194	3185	3121	3101
4 + 25	b_2	3241	3215	3206	3140	

examples of such theories. The experimental assignment of the five aromatic C–H stretching frequencies M_1 , M_2 , M_3 , M_{21} and

M_{22} , is complicated by the presence of Fermi resonances, which have not yet been untangled. Our calculations predict that M_{21} is in resonance with M_4M_{25} and that both M_1 and M_2 are in resonance with $M_{23}M_{25}$. The computed M_4M_{25} and $M_{23}M_{25}$ combination bands are listed in Table 2 and lie at higher frequency than the principle C–H stretches. Since we lack predicted intensities or experimental band shape information, we pragmatically assign the experimental bands in order of increasing frequency, which results in excellent agreement between computed and experimental values.

In Table 2 we also list selected overtones and combination bands for comparison with those determined experimentally, collected from ref. 53, 54 and 56. The overtone and combination bands agree with experiment to within the error bar expected from the agreement found for the fundamentals. Some Fermi resonances observed in the spectra are not reproduced by our VPT2 calculations. Specifically, we do not find a resonance between M_{11} and $M_{19}M_{20}$ and we find no indication that mode M_6 is in resonance with other states. Others, however, are identified by our calculations. The resonance between M_{10} and $M_{18}M_{19}$ is found in both theory and experiment, but our calculations predict the fundamental M_{10} to be lower in energy than $M_{18}M_{19}$, which if reliable would interchanging Gardener *et al.*'s assignment of the resonant pair, which has $M_{18}M_{19}$ lower in energy than M_{10} . Our calculations also verify the resonance between $M_{10}M_{29}$ and $M_{18}M_{19}M_{29}$.

Overall, the RMSD between observed and predicted band centres using VPT2 theory with a CC2/cc-pVTZ force field is 16 cm^{-1} for the fundamental transitions in the excited electronic state.

3.3 Catechol

Catechol (1,2-dihydroxybenzene) is biochemically important since the catecholamines adrenaline, noradrenaline, and dopamine are active in neurotransmission. It has been the subject of extensive spectroscopic studies, many of which have focused on the structure and dynamics of the low energy rotamers formed through changing the relative orientations of the hydroxyl groups, and the resulting differing levels of intra- and intermolecular hydrogen bonding.⁶⁰ The vibrational frequencies of the excited state have been probed using resonant two-photon ionization, fluorescence emission techniques, and molecular-beam hole-burning experiments.^{61,62}

Catechol is planar in the lowest energy isomer of the ground electronic state and the hydroxyl groups form an intermolecular hydrogen bond, but the structure of the excited state is assumed to be slightly distorted out of plane.⁶² Our structural investigations, using CC2/cc-pVTZ theory to optimise geometries of the ground and excited states, confirm that the ground state is planar with an intermolecular hydrogen bond. We find that the excited state retains the intermolecular hydrogen bond, but is significantly distorted from planarity, puckering at the carbons with the hydroxyl groups. The optimised structures are reported in the ESI.† CC2/cc-pVTZ predicts the 0–0 transition at 4.53 eV which is in line with the experimentally determined 0–0 excitation energy of 4.42 eV.⁶²

The calculated and experimental frequencies of the ground and excited states are compiled in Table 3. Only 16 fundamentals



of the excited state have been reported. The previous assignment was predicated on the assumption that the selection rules and band shapes for the planar S_0 state can be transferred to guide assignment of the S_1 state.^{62,63} Our calculations indicate that this assumption was flawed and we report a completely fresh assignment for the observed S_1 vibrational bands. We therefore present frequencies also for the ground state, comparing our purely *ab initio* assignment procedure to the more comprehensive experimental assignments available for this state.^{64–67} We use the oD_i labeling for *ortho*-di-substituted benzene rings of ref. 40 and, following the convention of earlier works,⁶⁴ use ν_1 , δ_1 , τ_1 to refer to the motions of the hydrogen donating OH group, and ν_2 , δ_2 , τ_2 to refer to the motions of the hydrogen accepting OH group.

The procedure adopted for assigning the vibrational modes of the S_0 state was as follows: first the normal modes were analysed and categorised according to the oD_i nomenclature, resulting in approximate C_{2v} symmetry labels a_1 , b_2 , b_1 , a_2 with corresponding type A, B, C line shapes for the IR active bands a_1 , b_2 , b_1 , respectively; these were then assigned in frequency order in effective C_{2v} symmetry blocks to the bands observed by Wilson for vapour phase catechol;⁶⁴ the only exception to this are the four $\alpha(\text{CH})$ bending modes, where the a_1 and b_2 labels are swapped compared to Wilson's and where recent work casts doubt on the original assignment.⁴⁰ Where additional or more precise measurements are available, Wilson's band centres are replaced or supplemented by the modern values and the Raman active a_2 modes are assigned in frequency order using measurements from solid state catechol; the bands assigned to the OH bending (δ), stretching (ν) and torsional (τ) modes are identified from examination of the normal mode coordinates.

The overall agreement with experiment and CC2/cc-pVTZ theory is excellent, with an RMSD of 29 cm^{-1} . The difficulties associated with out-of-plane vibrations are much less pronounced here and there is only one outlier of this type, the ring puckering mode D_{26} . The only other significant outlier is mode D_9 , which is the mode distorting from a aromatic ring to three localised double bonds.

Turning now to the S_1 state, comparison with experiment is problematic. Only 16 frequencies have been assigned and the assignment appears flawed. A vibronic progression with a spacing of 113 cm^{-1} was observed in resonant two-photon ionisation spectra of catechol and was assigned to OH torsional overtones.⁶² Our calculations do not predict low frequency torsional modes, but in fact predict a significant redshift of the torsional frequencies upon electronic excitation. The more tetrahedral arrangement at the oxygen centres in the excited state structure leads to a stronger intermolecular hydrogen bond, higher torsional frequencies and lower OH stretching frequencies. Instead, our calculations suggest assigning the progression of 113 cm^{-1} to the low frequency D_{29} ring bend. Note that the selection rules based on spatial symmetry do not rigorously apply since the excited state structure is significantly distorted out of plane. Having discounted the presence of low-frequency torsional modes, many of the spectral features observed in the experimental works must be reassigned. Proceeding to match computed and experimental values, accounting for band shape information where available, we report a

Table 3 CC2/cc-pVTZ vibrational wavenumbers for catechol in the S_0 and S_1 states

oD_i	C_{2v}	S_0			S_1		
		Harm.	VPT2	Exp.	Harm.	VPT2	Exp.
30	a_2	178	220	199	148	145	
τ_2	a_2	209	254		475	425	461
29	b_1	293	295	299	130	127	113
21	a_1	303	300	320	295	293	299
τ_1	b_1	432	433		585	576	607
20	b_2	438	434	449	399	390	395
28	b_1	456	454	456	369	348	317
19	b_2	552	545	542	517	486	472
27	a_2	561	590	582	342	347	
18	a_1	578	569	564	544	495	488
26	a_2	674	802	721	444	420	
25	b_1	743	744	741	495	508	502
17	a_1	772	760	768	732	714	735
24	a_2	820	846	851	595	560	588
16	b_2	854	840	859	829	812	840
23	b_1	897	924	916	761	732	746
22	a_2	930	988	963	837	836	863
15	a_1	1043	1028	1030	968	950	950
14	b_2	1098	1079	1092	1044	1028	1061
13/ δ_2	a_1	1163	1143	1151	1149	1122	
13/ δ_2	a_1	1168	1150	1151	1166	1159	
12	b_2	1210	1190	1195	1122	1104	
11	b_2	1268	1240	1251	1254	1220	
10	a_1	1309	1278	1263	1286	1252	
δ_1	b_2	1367	1338	1365	1343	1312	
9	a_1	1458	1419	1324	1402	1368	
8	b_2	1494	1455	1479	1450	1415	
7	a_1	1540	1502	1504	1635	1581	
6	b_2	1646	1601	1607	1511	1466	
5	a_1	1651	1609	1616	1566	1517	
4	b_2	3186	3048	3051	3177	3046	
3	a_1	3211	3092	3051	3250	3102	
2	b_2	3224	3101	3060	3190	3065	
1	a_1	3235	3105	3081	3204	3065	
ν_1	b_1	3750	3553	3605	3612	3390	
ν_2	a_1	3813	3626	3663	3542	3252	

fresh *ab initio* assignment in Table 3. The RMSD between VPT2 fundamentals using a CC2/cc-pVTZ force field and the observed bands with our fresh assignment is 22 cm^{-1} .

4 Conclusion

Excited state Hessians have been implemented for CC2, with a focus on keeping the scaling of the main memory demands to at most $\mathcal{O}(N^2)$. This has been realized by exploiting the RI approximation for the two electron integrals and by choosing a Laplace decomposition of orbital energy denominators in the calculation of the first-order density matrices. The implementation is an extension of that for excited state polarizabilities and was extended straightforwardly to excited state Hessians for the CIS(D_∞) and ADC(2) methods. The code is parallelized with OpenMP and MPI to make use of modern computer hardware. The analytic implementation enables the semi-numerical calculation of third and fourth derivatives for anharmonic corrections.

We applied VPT2 theory based on CC2 quartic force fields to *para*-difluorobenzene, toluene and catechol and compared the computed frequencies to experimentally observed vibrational bands of the first excited states. In contrast to previous



benchmark studies, we find that the accuracy of CC2 frequencies for excited electronic states is comparable to MP2 or CC2 frequencies for the ground electronic state, with typical average deviations between theory and experiment of less than 30 cm^{-1} for fundamental transitions when using a cc-pVTZ basis. However, we find that out-of-plane modes carry a much larger uncertainty due to internal basis set superposition errors, which leads to a strong basis set dependence of the force field terms, and unphysically large anharmonic corrections to typically underestimated harmonic frequencies.

In addition to assessing the accuracy of CC2 theory, our calculations have revealed some anomalies in the assignment of some of the experimentally observed bands. Our calculations discount some of the more tentative assignments in the spectrum of *para*-difluorobenzene. More significantly, our calculations indicate that the assignment of the lowest frequency features in the vibrational bands of catechol to OH torsions is incorrect, and that these lie much higher in energy due to the stronger hydrogen bond in the distorted S_1 excited state than in the planar S_0 state. Instead, we assign these features to ring modes, which become symmetry allowed transitions due to the low symmetry of the relaxed excited state structure. Due to this re-interpretation of the experimental bands it was necessary to perform a completely fresh assignment, and our new assignment can be considered a pure *ab initio* assignment.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 R. N. Dixon, T. A. A. Oliver and M. N. R. Ashfold, *J. Chem. Phys.*, 2011, **134**, 194303.
- 2 T. N. V. Karsili, A. M. Wenge, B. Marchetti and M. N. R. Ashfold, *Phys. Chem. Chem. Phys.*, 2014, **16**, 588–598.
- 3 A. G. Sage, T. A. A. Oliver, G. A. King, D. Murdock, J. N. Harvey and M. N. R. Ashfold, *J. Chem. Phys.*, 2013, **138**, 164318.
- 4 A. L. Sobolewski, W. Domcke, C. Dedonder-Lardeux and C. Jouvet, *Phys. Chem. Chem. Phys.*, 2002, **4**, 1093.
- 5 T. N. V. Karsili, A. M. Wenge, D. Murdock, S. J. Harris, J. N. Harvey, R. N. Dixon and M. N. R. Ashfold, *Chem. Sci.*, 2013, **4**, 2434–2446.
- 6 T. A. A. Oliver, G. A. King, D. P. Tew, R. N. Dixon and M. N. R. Ashfold, *J. Phys. Chem. A*, 2012, **116**, 12444–12459.
- 7 M. N. R. Ashfold, G. A. King, D. Murdock, M. G. D. Nix, T. A. A. Oliver and A. G. Sage, *Phys. Chem. Chem. Phys.*, 2010, **12**, 1218–1238.
- 8 G. D. Purvis and R. J. Bartlett, *J. Chem. Phys.*, 1982, **76**, 1910–1918.
- 9 J. F. Stanton and J. Gauss, *J. Chem. Phys.*, 1995, **103**, 8931–8943.
- 10 J. F. Stanton and J. Gauss, *Theor. Chim. Acta*, 1995, **91**, 267–289.
- 11 J. F. Stanton and J. Gauss, *J. Chem. Phys.*, 1996, **104**, 9859–9869.
- 12 O. Christiansen, J. F. Stanton and J. Gauss, *J. Chem. Phys.*, 1998, **108**, 3987–4001.
- 13 K. W. Sattelmeyer, J. F. Stanton, J. Olsen and J. Gauss, *Chem. Phys. Lett.*, 2001, **347**, 499–504.
- 14 H. Larsen, J. Olsen, P. Jørgensen and O. Christiansen, *J. Chem. Phys.*, 2000, **113**, 6677–6686.
- 15 M. Musiał, K. Kowalska and R. J. Bartlett, *THEOCHEM*, 2006, **768**, 103–109.
- 16 O. Christiansen, H. Koch and P. Jørgensen, *Chem. Phys. Lett.*, 1995, **243**, 409–418.
- 17 O. Vahtras, J. Almlöf and M. Feyereisen, *Chem. Phys. Lett.*, 1993, **213**, 514–518.
- 18 C. Hättig and F. Weigend, *J. Chem. Phys.*, 2000, **113**, 5154–5161.
- 19 T. S. Balaban, P. Braun, C. Hättig, A. Hellweg, J. Kern, W. Saenger and A. Zouni, *Biochim. Biophys. Acta, Bioenerg.*, 2009, **1787**, 1254–1265.
- 20 M. Head-Gordon, R. J. Rico, M. Oumi and T. J. Lee, *Chem. Phys. Lett.*, 1994, **219**, 21–29.
- 21 M. Head-Gordon, M. Oumi and D. Maurice, *Mol. Phys.*, 1999, **96**, 593–602.
- 22 J. Schirmer, *Phys. Rev. A: At., Mol., Opt. Phys.*, 1982, **26**, 2395–2416.
- 23 A. Köhn and C. Hättig, *J. Chem. Phys.*, 2003, **119**, 5021–5036.
- 24 C. Hättig, *Adv. Quantum Chem.*, 2005, **50**, 37–60.
- 25 D. H. Friese, N. O. C. Winter, P. Balzerowski, R. Schwan and C. Hättig, *J. Chem. Phys.*, 2012, **136**, 174106.
- 26 N. K. Graf, D. H. Friese, N. O. C. Winter and C. Hättig, *J. Chem. Phys.*, 2015, **143**, 244108.
- 27 D. H. Friese, C. Hättig and J. Koßmann, *J. Chem. Theory Comput.*, 2013, **9**, 1469–1480.
- 28 J. Almlöf, *Chem. Phys. Lett.*, 1991, **181**, 319–320.
- 29 F. Egidi, D. B. Williams-Young, A. Baiardi, J. Bloino, G. Scalmani, M. J. Frisch, X. Li and V. Barone, *J. Chem. Theory Comput.*, 2017, **13**, 2789–2803.
- 30 J. Liu and W. Liang, *J. Chem. Phys.*, 2011, **135**, 184111.
- 31 T. U. Helgaker, in *Geometrical Derivatives of Energy Surfaces and Molecular Properties*, ed. P. Jørgensen and J. Simons, Reidel, Dordrecht, Netherlands, 1986, p. 1.
- 32 T. Helgaker and P. Jørgensen, *Adv. Quantum Chem.*, 1988, **19**, 183–245.
- 33 *TURBOMOLE V7.0 2015, a development of University of Karlsruhe and Forschungszentrum Karlsruhe GmbH*, 1989–2007, TURBOMOLE GmbH, since 2007, available from <http://www.turbomole.com>.
- 34 B. Fornberg, *Math. Comput.*, 1988, **51**, 699–706.
- 35 I. M. Mills, in *Molecular Spectroscopy: Modern Research*, ed. K. Narahari Rao and C. Weldon Mathews, Academic Press, Inc., New York and London, 1972, p. 115.
- 36 W. Mizukami and D. P. Tew, *J. Chem. Phys.*, 2013, **139**, 194108.
- 37 R. D. Amos, N. C. Handy, W. H. Green, D. Jayatilaka, A. Willetts and P. Palmieri, *J. Chem. Phys.*, 1991, **95**, 8323–8336.



- 38 A. M. Gardner and T. G. Wright, *J. Chem. Phys.*, 2011, **135**, 114305.
- 39 A. Andrejeva, A. M. Gardner, W. D. Tuttle and T. G. Wright, *J. Mol. Spectrosc.*, 2016, **321**, 28–49.
- 40 W. D. Tuttle, A. M. Gardner, A. Andrejeva, D. J. Kemp, J. C. Wakefield and T. G. Wright, *J. Mol. Spectrosc.*, 2018, **344**, 46–60.
- 41 R. S. Mulliken, *J. Chem. Phys.*, 1955, **23**, 1997–2011.
- 42 E. B. Wilson, *Phys. Rev.*, 1934, **45**, 706–714.
- 43 T. Den, H.-M. Frey, P. M. Felker and S. Leutwyler, *J. Chem. Phys.*, 2015, **143**, 144306.
- 44 R. Zimmerman and T. Dunn, *J. Mol. Spectrosc.*, 1985, **110**, 312–325.
- 45 T. Cvitaš and J. Hollas, *Mol. Phys.*, 1970, **18**, 793–800.
- 46 A. E. W. Knight and S. H. Kable, *J. Chem. Phys.*, 1988, **89**, 7139–7160.
- 47 H. J. Elston, E. R. Davidson, F. G. Todd and C. S. Parmenter, *J. Phys. Chem.*, 1993, **97**, 5506–5518.
- 48 M. Robey and E. Schlag, *Chem. Phys.*, 1978, **30**, 9–17.
- 49 K. L. Reid, T. A. Field, M. Towrie and P. Matousek, *J. Chem. Phys.*, 1999, **111**, 1438–1445.
- 50 J. Long, C. Qin, Y. Liu, S. Zhang and B. Zhang, *Phys. Rev. A: At., Mol., Opt. Phys.*, 2011, **84**, 063409.
- 51 J. M. Martin, P. R. Taylor and T. J. Lee, *Chem. Phys. Lett.*, 1997, **275**, 414–422.
- 52 J. M. L. Martin, T. J. Lee and P. R. Taylor, *J. Chem. Phys.*, 1998, **108**, 676–691.
- 53 C. G. Hickman, J. R. Gascooke and W. D. Lawrance, *J. Chem. Phys.*, 1996, **104**, 4887–4901.
- 54 C. Minejima, T. Ebata and N. Mikami, *Phys. Chem. Chem. Phys.*, 2002, **4**, 1537–1541.
- 55 A. M. Gardner, A. M. Green, V. M. Tamé-Reyes, V. H. K. Wilton and T. G. Wright, *J. Chem. Phys.*, 2013, **138**, 134303.
- 56 A. M. Gardner, A. M. Green, V. M. Tamé-Reyes, K. L. Reid, J. A. Davies, V. H. K. Parkes and T. G. Wright, *J. Chem. Phys.*, 2014, **140**, 114308.
- 57 D. P. Tew, N. C. Handy and S. Carter, *Phys. Chem. Chem. Phys.*, 2001, **3**, 1958–1964.
- 58 D. P. Tew, N. C. Handy and S. Carter, *Mol. Phys.*, 2004, **102**, 2217–2226.
- 59 D. P. Tew and W. Mizukami, *J. Phys. Chem. A*, 2016, **120**, 9815–9828.
- 60 T. Dunn, R. Tembreull and D. Lubman, *Chem. Phys. Lett.*, 1985, **121**, 453–457.
- 61 M. Gerhards, W. Perl, S. Schumm, U. Henrichs, C. Jacoby and K. Kleinermanns, *J. Chem. Phys.*, 1996, **104**, 9362–9375.
- 62 T. Bürgi and S. Leutwyler, *J. Chem. Phys.*, 1994, **101**, 8418–8429.
- 63 F. Ramírez and J. L. Navarrete, *Vib. Spectrosc.*, 1993, **4**, 321–334.
- 64 H. W. Wilson, *Spectrochim. Acta, Part A*, 1974, **30**, 2141.
- 65 S. J. Greaves and W. P. Griffith, *Spectrochim. Acta, Part A*, 1991, **47**, 133–140.
- 66 T. Y. Koh, S. J. Greaves and W. P. Griffith, *Spectrochim. Acta, Part A*, 1994, **50**, 857–873.
- 67 H. G. Kjaergaard, D. L. Howard, D. P. Schofield, T. W. Robinson, S.-I. Ishiuchi and M. Fujii, *J. Phys. Chem. A*, 2002, **106**, 258–266.

