

Chemical Science

Volume 15
Number 30
14 August 2024
Pages 11669–12150

rsc.li/chemical-science



ISSN 2041-6539

EDGE ARTICLE

Rachel Codd *et al.*

A first-in-class dual-chelator theranostic agent designed for use with imaging-therapy radiometal pairs of different elements

Cite this: *Chem. Sci.*, 2024, 15, 11748

All publication charges for this article have been paid for by the Royal Society of Chemistry

A first-in-class dual-chelator theranostic agent designed for use with imaging-therapy radiometal pairs of different elements†

James L. Wood,^a Saikat Ghosh,^b Zachary H. Houston,^b Nicholas L. Fletcher,^b James Humphries,^b Karine Mardon,^b Dewan T. Akhter,^b William Tieu,^c Alesia Ivashkevich,^d Michael P. Wheatcroft,^d Kristofer J. Thurecht^b and Rachel Codd^{a*}

A covalent adduct of DFOB and DOTA separated by a L-lysine residue (DFOB-L-Lys-N⁶-DOTA) exhibited remarkable regioselective metal binding, with ¹H-¹³C NMR spectral shifts supporting Zr(IV) coordinating to the DFOB unit, and Lu(III) coordinating to the DOTA unit. This first-in-class, dual-chelator theranostic design could enable the use of imaging-therapy radiometal pairs of different elements, such as ⁸⁹Zr for positron emission tomography (PET) imaging and ¹⁷⁷Lu for low-energy β⁻-particle radiation therapy. DFOB-L-Lys-N⁶-DOTA was elaborated with an amine-terminated polyethylene glycol extender unit (PEG4) to give DFOB-N²-(PEG4)-L-Lys-N⁶-DOTA (compound **D2**) to enable installation of a phenylisothiocyanate group (Ph-NCS) for subsequent monoclonal antibody (mAb) conjugation (mAb = HuJ591). **D2**-mAb was radiolabeled with ⁸⁹Zr or ¹⁷⁷Lu to produce [⁸⁹Zr]Zr-**D2**-mAb or [¹⁷⁷Lu]Lu-**D2**-mAb, respectively, and *in vivo* PET/CT imaging and *in vivo/ex vivo* biodistribution properties measured with the matched controls [⁸⁹Zr]Zr-DFOB-mAb or [¹⁷⁷Lu]Lu-DOTA-mAb in a murine LNCaP prostate tumour xenograft model. The ⁸⁹Zr-immuno-PET imaging function of [⁸⁹Zr]Zr-**D2**-mAb and [⁸⁹Zr]Zr-DFOB-mAb showed no significant difference in tumour accumulation at 48 or 120 h post injection. [⁸⁹Zr]Zr-**D2**-mAb and [¹⁷⁷Lu]Lu-**D2**-mAb showed similar *ex vivo* biodistribution properties at 120 h post-injection. Tumour uptake of [¹⁷⁷Lu]Lu-**D2**-mAb shown by SPECT/CT imaging at 48 h and 120 h post-injection supported the therapeutic function of **D2**, which was corroborated by similar therapeutic efficacy between [¹⁷⁷Lu]Lu-**D2**-mAb and [¹⁷⁷Lu]Lu-DOTA-mAb, both showing a sustained reduction in tumour volume (>80% over 65 d) compared to vehicle. The work identifies **D2** as a trifunctional chelator that could expand capabilities in mixed-element radiometal theranostics to improve dosimetry and the clinical outcomes of molecularly targeted radiation.

Received 30th April 2024
Accepted 21st May 2024

DOI: 10.1039/d4sc02851a

rsc.li/chemical-science

Introduction

The use of radiometals for targeted imaging and therapy in nuclear medicine is expanding,^{1–7} as evident by recent U.S. Food and Drug Administration approvals for ¹⁷⁷Lu *endo*-radiotherapy agents Lutathera® for neuroendocrine tumours (NET)† and Pluvicto™ for metastatic castration-resistant prostate cancer

(mCRPC). A major clinical advantage of these agents is afforded by the ability to individualise the dose of β⁻-particle radiation therapy by first measuring the tumour burden using an imaging agent containing the same targeting vector as the therapeutic agent (*e.g.*, octreotate, which targets somatostatin receptors in NET; or vipivotide tetraxetan (PSMA-617), which targets prostate-specific membrane antigen (PSMA) in mCRPC). However, pairing metal ions of different elements with different coordination chemistries in a theranostic approach requires the use of different chelators (*e.g.*, HBED-CC in ⁶⁸Ga-gozetotide paired with DOTA in ¹⁷⁷Lu-based Pluvicto™), which can result in different biodistribution properties between the imaging and therapy pair. Using different isotopes of the same element (*e.g.*, ⁶⁴Cu for imaging, ⁶⁷Cu for therapy) is one solution to this problem, since the requirement for a unique chelator (*e.g.*, sarcophagine)^{8,9} guarantees the properties of the imaging and therapy complexes are identical. Same-element imaging-therapy radiometal pairs, however, are few which limits

^aThe University of Sydney, School of Medical Sciences, New South Wales 2006, Australia. E-mail: rachel.codd@sydney.edu.au

^bCentre for Advanced Imaging (CAI), Australian Institute for Bioengineering and Nanotechnology (AIBN) and ARC Training Centre for Innovation in Biomedical Imaging Technology, The University of Queensland, Brisbane, Queensland 4072, Australia

^cMolecular Imaging and Therapy Research Unit (MITRU), South Australian Health and Medical Research Institute (SAHMRI), Adelaide, Australia

^dTelix Pharmaceuticals Limited, North Melbourne, Victoria 3051, Australia

† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d4sc02851a>



Chart 1 Structure of desferrioxamine B (DFOB, 1) and DOTA (2).

opportunities to use mixed-element radiometal pairs with useful decay properties. Theranostic systems compatible with radiometals of different elements are being explored using discrete chelators with capacity for cross-metal binding.^{10–14}

We have taken a different approach by grafting together two chelators, desferrioxamine B (DFOB, 1) and 1,4,7,10-tetraazacyclododecane-1,4,7,10-tetraacetic acid (DOTA, 2) (Chart 1), which have been established experimentally and/or clinically as useful chelators for ⁸⁹Zr(IV) (positron-emission tomography (PET) imaging) or ¹⁷⁷Lu(III) (β-radiation therapy), respectively.

The risk of this approach is that most chelators have broad metal selectivity,^{1–4} forming complexes with a range of metal ions with variable affinities and stabilities, which could complicate robust verification and function of the radiolabelled product. The polyaminocarboxylic acid chelator 2 is a case in point, which can coordinate many radiometal ions, including but not limited to ⁶⁸Ga, ¹⁷⁷Lu, ⁴⁴Sc, ²¹²Pb, ²²⁵Ac and ²¹¹At.^{1–4,15,16} The hydroxamic acid chelator 1 can complex ⁸⁹Zr, ⁶⁸Ga, ⁹⁰Nb, and others.^{1–4,17–22} Considering the metal selectivity of 1 and 2 as discrete compounds (Table 1) indicated the possibility that a covalent DFOB-DOTA adduct might have an unusual and useful regioselective metal binding function towards sets of imaging-therapy radiometal pairs of different elements, such as ⁸⁹Zr ($t_{1/2} = 3.3$ d, $E(\beta_{\max}^+) = 901$ keV (22.8%), EC = 77%, $E_{\gamma} = 909$ keV (100%))^{18,23} and ¹⁷⁷Lu ($t_{1/2} = 6.6$ d, $E(\beta_{\max}^-) = 497$ keV (78.6%), $E_{\gamma} = 208$ keV (11.1%), 113 keV (6.6%)).²⁴

This work describes the preparation and characterisation of the dual chelator DFOB-*N*²-(PEG4)-*L*-Lys-*N*⁶-DOTA (3) (named ‘D2’ = DFOB and DOTA, Scheme 1). Compound 3 (D2) has been designed for regioselective binding of ^{nat}/⁸⁹Zr(IV) in the 1 region and ^{nat}/¹⁷⁷Lu(III) in the 2 region, and includes an amine-terminated polyethylene glycol (PEG4) unit to enable further chemistry, including biomolecule conjugation. Radiolabelling D2-mAb (mAb = monoclonal antibody = HuJ591) with ⁸⁹Zr(IV)

or ¹⁷⁷Lu(III) gave [⁸⁹Zr]Zr-D2-mAb (4) or [¹⁷⁷Lu]Lu-D2-mAb (5) complexes, respectively (Scheme 1). The *in vitro* and *in vivo* properties of [⁸⁹Zr]Zr-D2-mAb and [¹⁷⁷Lu]Lu-D2-mAb were evaluated together with the matched single chelator control compounds [⁸⁹Zr]Zr-1-mAb and [¹⁷⁷Lu]Lu-2-mAb in LNCaP cell-based assays and a murine LNCaP prostate cancer xenograft model. [⁸⁹Zr]Zr-D2-mAb and [⁸⁹Zr]Zr-1-mAb showed similar PET/CT imaging function and tumour uptake (about 15% ID g^{−1}) at 48 h and 120 h post injection, with SPECT/CT imaging showing tumour uptake of [¹⁷⁷Lu]Lu-D2-mAb to support therapeutic function, which was corroborated by therapeutic efficacy data. The work identifies D2 as a trifunctional dual-chelator theranostic compound with potential for use with ⁸⁹Zr for imaging or ¹⁷⁷Lu for therapy, and other mixed-element radiometal pairs, to expand capabilities in nuclear medicine.

Results and discussion

Metal selectivity of discrete chelators DFOB and DOTA

DFOB (1). Hydroxamic acids have a rich coordination chemistry.^{25–27} This includes the characterisation of many complexes between ^{nat}Zr(IV) or ⁸⁹Zr(IV) and 1 or higher denticity analogues designed to saturate the octadentate Zr(IV) coordination sphere to maximise the stability of ⁸⁹Zr PET imaging agents.^{19,28–37} Despite the broad metal selectivity of hydroxamic acids, there are no reports of structurally characterised complexes between hydroxamic acids and Lu(III).

The ability of 1 to bind Zr(IV) and not Lu(III) was confirmed from {¹H}–¹³C NMR spectroscopic measurements (Fig. 1) from D₂O solutions of 1 alone, or in the presence of a stoichiometric excess of Zr(IV) or Lu(III). The diamagnetism of Zr(IV) and Lu(III) made this approach feasible with minimal metal-dependent line broadening.

Consistent with previous work on complexes between 1 and Al(III) or Ga(III),³⁹ or Zr(IV),⁴⁴ the cluster of signals between 173–175 ppm attributable to the C=O (amide) and C=O (hydroxamic acid) carbon atoms in the spectrum from 1 (Fig. 1a) were resolved in the presence of Zr(IV) (Fig. 1b) into two sets, with the C=O (hydroxamate) signals involved in metal coordination appearing upfield at 164.1–164.9 ppm. The two C=O signals in the Zr(IV)-1 complex at about 174 ppm that did not shift appreciably compared to 1 could be assigned as the two C=O (amide) carbon atoms, since these would experience less change in chemical environment. The signals at 47.6–47.8 ppm attributable to the methylene groups adjacent to the hydroxamic acid N–OH in free 1 moved downfield to 50.2–50.9 ppm in the presence of Zr(IV). The signal attributable to the terminal CH₃ group in 1 shifted upfield from 19.2 to 16.3 ppm in the presence of Zr(IV), supporting Zr(IV)-1 complexation. In addition to these readily assigned signals, there were other spectral shifts in the Zr(IV)-1 system that demonstrated complex formation, which would likely include the presence of isomers.³⁹

The spectra from free 1 or in solution with Lu(III) (Fig. 1c) were close to coincident, indicating the absence of Lu(III)-1 complexation. Metal complexation experiments were conducted at room temperature or at mildly elevated temperatures (≤37 °C) to mimic radiolabelling conditions and provide results

Table 1 Log *K* values of 1 : 1 complexes formed between DFOB (1) or DOTA (2) and selected metal ions with corresponding theranostic radioisotopes in use/development

Chelator	Log <i>K</i>				
	Zr(IV)	Lu(III)	Ga(III)	La(III)	Pb(II)
DFOB (1)	40.0 ^{a,38}	NR ^b	27.6 ^{c,39}	NR ^b	10.0 ^{d,40}
DOTA (2)	NR ^b	25.4 ^{e,41}	26.1 ^{f,42}	22.9 ^{e,41}	24.3 ^{g,43}

^a 25 °C, I = 1 M NaClO₄. ^b NR = not reported. ^c 25 °C, I = 1 M KCl. ^d 25 °C, I = 0.1 M KCl. ^e 25 °C, I = 0.1 M NaCl. ^f 25 °C, I = 0.1 (CH₃)₄N⁺Cl[−]. ^g 25 °C, I = 0.1 M NaClO₄.





Scheme 1 Structure of compound D2 (3), [⁸⁹Zr]Zr-D2-mAb (4), and [¹⁷⁷Lu]Lu-D2-mAb (5).

relevant to procedures used for mAbs and other thermally labile biological vectors.

DOTA (2). The complexation of Lu(III) or ¹⁷⁷Lu with 2 is well established, with 2 present in both Lutathera® and Pluvicto™. While a Zr(IV)-DOTA complex and Zr(IV) complexes with related polyaminocarboxylic acid chelators have been characterised^{45,46} the elevated temperature (≥65 °C) necessary for complexation with 2 suggests that the formation of Zr(IV)-DOTA species is less favourable at temperatures used for mAb radiolabelling (≤37 °C). The {¹H}-¹³C NMR spectra from solutions of 2 alone or in the presence of Zr(IV) or Lu(III) acquired at room temperature or -10 °C gave broad signals, as typically observed for 2 due to the presence of multiple interchanging conformational isomers.^{47–49} The compound DOTA-N²-acetyl-L-lysine methyl ester (8a) was prepared as a lower-symmetry 2 analogue and surrogate of the 2 region in D2. Spectra from solutions of Lu(III) and 8a showed a set of broad signals between 55.5–66 ppm consistent with Lu(III) coordination *via* the *endo*- and *exo*-

methylene groups of the 2 motif. These signals were not evident in the Zr(IV) or free ligand system (Fig. S17†).

Synthesis of DFOB-L-Lys-N⁶-DOTA (3a) and regioselective coordination of Zr(IV) and Lu(III)

The metal selectivity results from the {¹H}-¹³C NMR spectroscopic data together with established Zr(IV)-1 and Lu(III)-2 chemistry (Table 1) supported that under mild temperature conditions, a DFOB-DOTA covalent adduct would preferentially bind ^{nat}Zr(IV) or ⁸⁹Zr(IV) at the DFOB region and not at the DOTA region, and bind ^{nat}Lu(III) or ¹⁷⁷Lu(III) at the DOTA region and not at the DFOB region. A construct useful for targeted radio-metal delivery requires a functional group to graft peptides and/or mAbs, which led to the design of DFOB-L-Lys-N⁶-DOTA (Scheme 2, 3a) as the preliminary target compound.

A protected precursor of 3a, DFOB-N²-(Fmoc)-L-Lys-N⁶-DOTA(O^tBu)₃ (3e), was prepared from an HBTU-mediated



Fig. 1 {¹H}-¹³C NMR spectra from solutions (D₂O) of (a) 1, (b) Zr(IV) : 1 (1.8 : 1), or (c) Lu(III) : 1 (1.3 : 1). The signal at 38.4 ppm in (a) and (c) is due to a decoupler zero frequency spike.



Scheme 2 Synthesis of DFOB- N^2 -(PEG₄)-L-Lys- N^6 -DOTA (3) (D2), DFOB-L-Lys- N^6 -DOTA (3a), and DFOB- N^2 -(PEG₄-Ph-NCS)-L-Lys- N^6 -DOTA (3b) (D2-Ph-NCS).

amide coupling reaction between **1** and N^2 -(Fmoc)-L-Lys- N^6 -DOTA(O^tBu)₃ (**8**), which itself was prepared from N^2 -(Fmoc)-L-Lys (**6**) and DOTA(O^tBu)₃ (**7**). Global deprotection of **3e** generated **3a** in an overall yield of 21% (Scheme 2). The $\{^1\text{H}\}$ - ^{13}C NMR spectrum of **3a** (Fig. 2b), assigned using data from the composite fragments **1** (Fig. 2a)³⁹ and **8a**, showed signals between 47.9–52 ppm and 54.8–57.1 ppm ascribed to the *endo*- and *exo*-methylene carbon atoms of the DOTA region, respectively. The carbonyl carbon atoms in the DOTA region of **3a** were resolved at 169.6–169.8 ppm (C45, C47) and 169.9 ppm (C46). Compared to **2** or **8a**, the nature and conformation of the extended structure present in **3a** modulated the molecular tumbling to enable resolution of the signals ascribable to the DOTA region at room temperature.

The $\{^1\text{H}\}$ - ^{13}C NMR spectra from **3a** alone (Fig. 3a) or in the presence of Zr(IV) or Lu(III) supported the regioselective

coordination of Zr(IV) to the **1** region and Lu(III) to the **2** region. The spectral changes in the Zr(IV)-**3a** system (Fig. 3b) were similar to those observed in the Zr(IV)-**1** system (Fig. 1b), with an upfield shift in the three signals ascribed to the carbonyl carbon atoms of the Zr(IV) coordinating hydroxamic acid groups (164.1–164.8 ppm) and the terminal methyl group (16.3 ppm), and a downfield shift (50.1–51.3 ppm) in the three signals attributable to the methylene carbon atoms adjacent to the N-OH groups. These latter sharp signals were overlaid with the signals for the *endo*-methylene carbon atoms in the **2** region, and although broadened, their appearance supported the non-coordinating role of the **2** region in solutions of Zr(IV) and **3a**.

Signals ascribed to the **1** region in the spectrum from a solution of **3a** and Lu(III) (Fig. 3c) mapped closely to the same ascribed signals in **3a** alone, supporting the **1** region was non-coordinating towards Lu(III). In the presence of Lu(III), the



Fig. 2 $\{^1\text{H}\}$ - ^{13}C NMR spectra from solutions (D₂O) of (a) **1** or (b) **3a**.



Fig. 3 $\{^1\text{H}\}-^{13}\text{C}$ NMR spectra from solutions (D_2O) of (a) **3a**, (b) $\text{Zr(IV)} : \text{3a}$ (0.8 : 1), or (c) $\text{Lu(III)} : \text{3a}$ (0.8 : 1).

signals for the *endo*- and *exo*-methylene carbon atoms of the **2** region in **3a** were significantly broadened, with low intensity signals appearing at 63.3 ppm and 65.7 ppm (inset) and an upfield shift of a signal ascribed to one of the carbonyl carbon atoms in the **2** region in **3a**. Together, these results supported that at room temperature, Zr(IV) preferentially coordinated the **1** region and not the **2** region in **3a**, and Lu(III) preferentially coordinated the **2** region and not the **1** region in **3a**.

Stoichiometry of complexes between Zr(IV) , Lu(III) or Ga(III) and **3a**

Having established the regioselectivity of $\text{Zr(IV)}/\text{Lu(III)}-\text{3a}$ coordination from $\{^1\text{H}\}-^{13}\text{C}$ NMR spectroscopy, complementary experiments to support selective binding were undertaken using liquid chromatography-mass spectrometry (LC-MS) to report metal:**3a** stoichiometry. In addition to Zr(IV) and Lu(III) ,



Fig. 4 LC-MS traces (total ion current (TIC)) from solutions of **3a** in the absence (a–c, solid line) or presence (broken line) of (a) Zr(IV) , (b) Lu(III) , or (c) Ga(III) . EIC traces set to report protonated double-charged 1 : 1 or 2 : 1 metal : **3a** complexes for (d) Zr(IV) (**9**) (blue) or (**10**) (black), (e) Lu(III) (**11**) (red) or (**12**) (black), or (f) Ga(III) (**13a**, **13b**) (green) or (**14**) (olive), with the isotope pattern experimental (coloured line) or calculated (white column) for (g) 1 : 1 $\text{Zr(IV)} : \text{3a}$ (**9**) ($[\text{M}^+ + \text{H}]^{2+}$ adduct), (h) 1 : 1 $\text{Lu(III)} : \text{3a}$ (**11**) ($[\text{M} + \text{H}]^+$ adduct), (i) 1 : 1 $\text{Ga(III)} : \text{3a}$ (**13a**, **13b**) ($[\text{M} + \text{H}]^+$ adduct), or (j) 2 : 1 $\text{Ga(III)} : \text{3a}$ (**14**) ($[\text{M} + \text{H}]^+$ adduct).

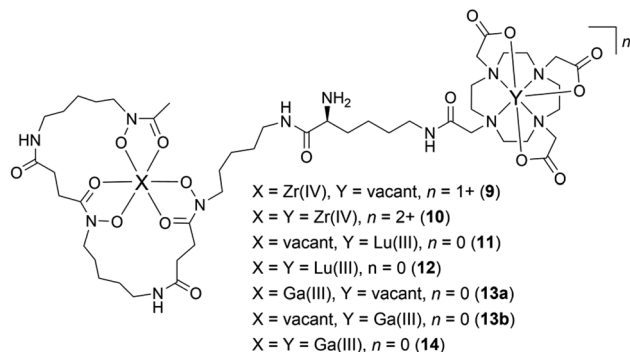


Chart 2 Complexes between Zr(IV), Lu(III), or Ga(III), and **3a**, with a 1 : 1 (9, 11, 13a, 13b) or 2 : 1 (10, 12, 14) metal:**3a** stoichiometry.

these experiments included Ga(III), which has an affinity towards **1** and **2** (Table 1), and was useful in evaluating a potential case of cross-chelator coordination. Solutions of **3a** were incubated with a molar equivalent (Fig. 4) of Zr(IV), Lu(III) or Ga(III) at 37 °C and analysed by LC-MS. The $\{^1\text{H}\}$ - ^{13}C NMR spectroscopic data predicted the formation of 1:1 Zr(IV):**3a** (Chart 2), $[\text{Zr(IV)-3a}_{(\text{DFOB})}]^+$ (**9**), and not 2:1 Zr(IV):**3a** $[(\text{Zr(IV)})_2\text{-3a}_{(\text{DFOB})(\text{DOTA})}]^{2+}$ (**10**). This expected trend in stoichiometry would be matched for Lu(III), albeit with binding in the 2 region, with the formation of 1:1 Lu(III) : **3a** $[\text{Lu(III)-3a}_{(\text{DOTA})}]$ (**11**) and not 2:1 Lu(IV) : **3a** $[(\text{Lu(III)})_2\text{-3a}_{(\text{DFOB})(\text{DOTA})}]$ (**12**). The affinity of Ga(III) for **1** and **2** could result in the formation of two 1:1 Ga(III):**3a** isomers indistinguishable by LC-MS $[\text{Ga(III)-3a}_{(\text{DFOB})}]$ (**13a**) and $[\text{Ga(III)-3a}_{(\text{DOTA})}]$ (**13b**), and the 2:1 Ga(III):**3a** complex $[(\text{Ga(III)})_2\text{-3a}_{(\text{DFOB})(\text{DOTA})}]$ (**14**).

The intensity of the signal at 7.3 min in the LC-MS traces of the free ligand **3a** (Fig. 4a-c, solid line) (total ion chromatogram (TIC)) was reduced in the presence of Zr(IV), Lu(III), or almost completely in the presence of Ga(III) (Fig. 4a-c, respectively, broken line), as consistent with the formation of complexes between Zr(IV), Lu(III) or Ga(III) and **3a**.

Signals at $t_R = 6.1$ min or 7.4 min were detected from the extracted ion chromatogram (EIC) traces set to report 1:1 Zr(IV):**3a** (**9**) (Fig. 4d, blue) or Lu(III):**3a** (**11**) (Fig. 4e, red), respectively (Table 2). No signals were observed for the 2:1 Zr(IV):**3a** (**10**) (Fig. 4d, black) or 2:1 Lu(III):**3a** (**12**) (Fig. 4e, black) complexes. The experimental isotope patterns for **9**

Table 2 Metal:ligand complexes in 1:1 or 2:1 stoichiometries between Zr(IV), Lu(III) or Ga(III) and **3a** (observed and theoretical)

	X^a	Y^a	n	Adduct (1+) m/z	Adduct (2+) m/z	Observed
9	Zr(IV)	Vacant	1+	1161.5 ^b	581.3 ^c	Y
10	Zr(IV)	Zr(IV)	2+	N/A	624.2 ^d	N
11	Vacant	Lu(III)	0	1247.6 ^e	624.3 ^f	Y
12	Lu(III)	Lu(III)	0	1419.5 ^e	710.2 ^f	N
13a	Ga(III)	Vacant	0	1141.5 ^e	571.3 ^f	Y
13b	Vacant	Ga(III)	0	1141.5 ^e	571.3 ^f	Y
14	Ga(III)	Ga(III)	0	1207.4 ^e	604.2 ^f	Y

^a Refer Chart 2. ^b $[M]^+$. ^c $\{([M]^+ + H)\}^{2+}$. ^d $[M]^{2+}$. ^e $[M + H]^+$. ^f $[M + 2H]^{2+}$.

(Fig. 4g, blue) and **11** (Fig. 4h, red) were consistent with calculated patterns (white columns) in support of the assignment of these species.

The solution of Ga(III) and **3a** gave EIC signals at $t_R = 6.0$ min and 5.8 min correlating with complexes with 1 : 1 (**13a**, **13b**) (Fig. 4f, green) and 2 : 1 (**14**) (Fig. 4f, olive) stoichiometries, respectively, showing that, unlike Zr(IV) and Lu(III), Ga(III) was able to simultaneously coordinate to the **1** and **2** regions of **3a**. The assignment of the 1 : 1 isomer pair (**13a**, **13b**) (Fig. 4i) and the 2 : 1 complex (**14**) (Fig. 4j) was supported by the match between the experimental and theoretical isotope patterns.

The selectivity of coordination of **3a** as established with natural abundance $^{nat}\text{Zr(IV)}$ and $^{nat}\text{Lu(III)}$ metal solutions supported progressing to radiolabelling studies and evaluating the potential of the **D2** dual-chelator system to deliver molecularly targeted imaging/therapy radiation.

Synthesis and characterisation of DFOB- N^2 -(PEG4)-L-Lys- N^6 -DOTA (3) (D2) and DFOB- N^2 -(PEG4-Ph-NCS)-L-Lys- N^6 -DOTA (3b) (D2-Ph-NCS)

A polyethylene glycol unit (PEG4) was appended to the internal L-Lys residue in **3a** to increase water solubility and to increase the distance between the dual-chelator scaffold and the mAb to maintain mAb-antigen recognition properties. PEG units and simpler ether-containing constructs have been incorporated in other 1- and 2-type chelators to improve water solubility and other performance measures.^{30,32,50–53} Compound **3 (D2)** (Scheme 2) was prepared from **3e** by removing the Fmoc group to give **3d** followed by an amide coupling reaction with *N*-Boc-amine-PEG4-acid to give **3c**, and a final global deprotection step to give **3 (D2)**. A sample of **3 (D2)** was reacted with 1,4-phenylenediisothiocyanate and the resultant **3b (D2-Ph-NCS)** purified by HPLC and lyophilised. Other types of activating groups§ would be compatible with **D2**,^{2,29,54,55} with Ph-NCS selected here as one in common use, and useful for benchmarking performance against the discrete chelators DFOB-Ph-NCS (1-Ph-NCS) and DOTA-Bn-NCS (2-Bn-NCS). The overall yield of **3b (D2-Ph-NCS)** starting from the reaction between **6** and **7** was approximately 10.4%. The most significant loss of yield occurred from the reaction between **3 (D2)** and 1,4-phenylene diisothiocyanate, due in part to the poor solubility of the latter reagent in organic solvents.

Antibody conjugation with D2-Ph-NCS (3b) and radiolabelling D2-mAb

The monoclonal antibody HuJ591 (mAb) was conjugated to **D2-Ph-NCS (3b)**, **1-Ph-NCS** or **2-Bn-NCS**, and after buffer exchange, the resulting **D2-mAb**, **1-mAb** or **2-mAb** complexes were radio-labelled with ^{89}Zr or ^{177}Lu . The chelator : mAb ratio determined from MALDI data (Fig. S26[†]) was about 2 for **1-mAb**, 3 for **2-mAb** and 4 for **D2-mAb** (Table S1[†]). HuJ591 is a de-immunised variant of the antibody J591 that targets PSMA expressed on LNCaP cells, which is a cell line established from metastatic prostate cancer.⁵⁶⁻⁶¹

The radio-iTLC traces from **D2**-mAb radiolabelled with ^{89}Zr or ^{177}Lu gave well-resolved signals characteristic of ^{89}Zr / ^{177}Lu -**D2**-

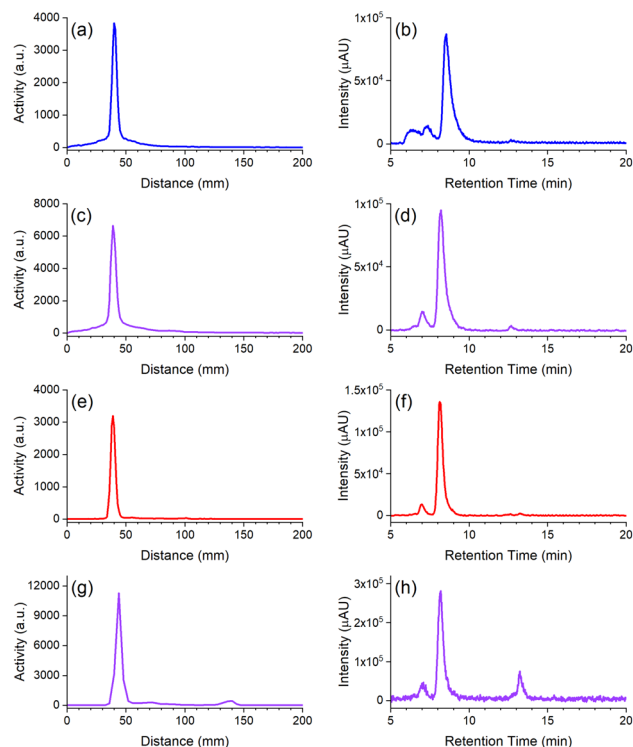


Fig. 5 Radio-iTLC (left column) or radio-SEC-HPLC (right column) traces from 1-mAb (a and b) or D2-mAb (c and d) radiolabelled with ^{89}Zr , and 2-mAb (e and f) or D2-mAb (g and h) radiolabelled with ^{177}Lu . Radio-iTLC reaction mixtures were mixed with 1 : 1.5 mM EDTA (^{177}Lu) or DTPA (^{89}Zr) and developed in a 1 : 1 EtOH : H₂O solvent system (Agilent iTLC-SG Glass microfiber chromatography paper impregnated with silica gel).

mAb (Fig. 5c) or [^{177}Lu]Lu-D2-mAb (Fig. 5g) and correlated with the respective signals for [^{89}Zr]Zr-1-mAb (Fig. 5a) or [^{177}Lu]Lu-2-mAb (Fig. 5e) as matched controls. An earlier radiolabelling study using D2 conjugated to the mAb girentuximab showed similar results with D2-mAb and showed minimal (<1%) radiolabelling of 2 with ^{89}Zr and no detectable radiolabelling of 1 with ^{177}Lu at mild temperatures (Fig. S25[†]), in accord with the chelator-metal selectivity observed using $^{\text{nat}}\text{Zr}(\text{IV})$ and $^{\text{nat}}\text{Lu}(\text{III})$.

In vitro properties and radiochemical stability of [^{89}Zr]Zr-D2-mAb (4) and [^{177}Lu]Lu-D2-mAb (5)

The *in vitro* cell membrane binding and internalisation of [^{89}Zr]Zr-D2-mAb, [^{177}Lu]Lu-D2-mAb, and the single-chelator controls [^{89}Zr]Zr-1-mAb and [^{177}Lu]Lu-2-mAb, were assessed in LNCaP prostate cancer cells, in which the binding specificity of HuJ591 and cell-surface PSMA is well established.^{58,59,62}

In vitro cell-associated activity at 1 h was similar between [^{89}Zr]Zr-D2-mAb and [^{89}Zr]Zr-1-mAb (Fig. 6a), and between [^{177}Lu]Lu-D2-mAb and [^{177}Lu]Lu-2-mAb (Fig. 6c), showing the mAb was functional in D2-mAb. Levels of cell-membrane and internalised activity were similar between [^{89}Zr]Zr-D2-mAb and [^{89}Zr]Zr-1-mAb (Fig. 6b), with a slight reduction in internalisation of [^{177}Lu]Lu-D2-mAb compared to [^{177}Lu]Lu-2-mAb (Fig. 6d). The capacity of radiolabelled antibodies to accumulate and remain intracellularly (residualisation) can be useful in

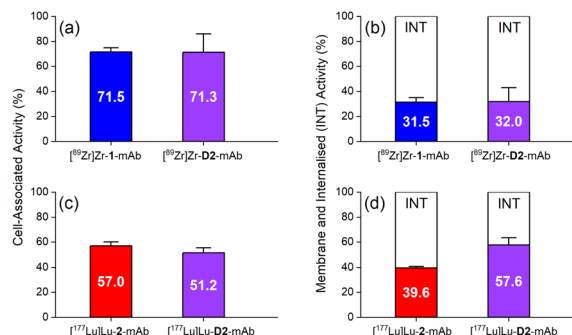


Fig. 6 Average percentage of cell-associated activity (LNCaP cells) at 1 h of (a) [^{89}Zr]Zr-1-mAb (blue), [^{89}Zr]Zr-D2-mAb (purple), and (c) [^{177}Lu]Lu-2-mAb (red), [^{177}Lu]Lu-D2-mAb (purple); and membrane (coloured) or internalised (INT) fractions of (b) [^{89}Zr]Zr-1-mAb (blue), [^{89}Zr]Zr-D2-mAb (purple), and (d) [^{177}Lu]Lu-2-mAb (red), [^{177}Lu]Lu-D2-mAb (purple).

maximising DNA radiation damage in endo-radiotherapy.^{63,64} The radiochemical stability of [^{89}Zr]Zr-D2-mAb and [^{89}Zr]Zr-1-mAb was measured using radio-iTLC and radio-SEC-HPLC in human serum and in phosphate-buffered saline (Fig. S27–S30 and Table S3[†]), with sampling at day 0, 2, 4 and 7. [^{89}Zr]Zr-D2-mAb maintained radiochemical purity in human serum at day 7 to 81% (radio-SEC-HPLC) or 88% (radio-iTLC) with the respective values for [^{89}Zr]Zr-1-mAb at 61% or 76%.

In vivo properties of [^{89}Zr]Zr-D2-mAb (4) and [^{177}Lu]Lu-D2-mAb (5)

The biodistribution in tumour and other tissues of [^{89}Zr]Zr-D2-mAb, [^{177}Lu]Lu-D2-mAb, and the single-chelator controls [^{89}Zr]Zr-1-mAb and [^{177}Lu]Lu-2-mAb, were assessed *in vivo* in a murine LNCaP xenograft model by PET/CT imaging and SPECT/CT imaging or *ex vivo* by gamma counting.¶ [^{89}Zr]Zr-D2-mAb and [^{89}Zr]Zr-1-mAb showed similar tumour/tissue bio-distribution at 4 h post-injection (Fig. 7a), with [^{89}Zr]Zr-D2-mAb showing significantly higher accumulation in the tumour. This trend continued at 24 h post-injection (Fig. 7b).

From 48 h to 120 h post-injection, the level of accumulation in the tumour showed no difference between [^{89}Zr]Zr-D2-mAb and [^{89}Zr]Zr-1-mAb (Fig. 7c and d). At 120 h post-injection, there was no statistical difference in bone uptake between [^{89}Zr]Zr-D2-mAb and [^{89}Zr]Zr-1-mAb, suggesting the stability of ^{89}Zr binding was similar between the two conjugates (Fig. 7d).

Tumour xenografted mice imaged by PET/CT using [^{89}Zr]Zr-D2-mAb (upper) or [^{89}Zr]Zr-1-mAb (lower) (Fig. 8) or by SPECT/CT using [^{177}Lu]Lu-D2-mAb (Fig. 9) showed a discernible tumour at all timepoints, with signals isolated from background and able to be delineated without interference.

Ex vivo measurements of [^{89}Zr]Zr-D2-mAb or [^{89}Zr]Zr-1-mAb correlated similarly to *in vivo* biodistribution and the PET/CT images at 48 h (Fig. 7c, 8 and 10a) and 120 h (Fig. 7d, 8 and 10b) post-injection. There was a significant difference in *ex vivo* biodistribution in the kidney and liver tissues at 48 h post-injection (Fig. 10a), with these differences reducing at 120 h post-injection (Fig. 10b). Most other collected tissues showed no



Fig. 7 *In vivo* biodistribution ($n = 3$) of [⁸⁹Zr]Zr-1-mAb (blue) and [⁸⁹Zr]Zr-D2-mAb (purple) ($n = 3$) at (a) 4 h, (b) 24 h, (c) 48 h, or (d) 120 h post-injection as determined by ROI analysis of PET/CT images. * $p \leq 0.05$, ** $p \leq 0.01$, *** $p \leq 0.001$.



Fig. 8 Representative 3D maximum intensity projections for all mice ([⁸⁹Zr]Zr-D2-mAb upper; [⁸⁹Zr]Zr-1-mAb, lower); in ⁸⁹Zr PET/CT imaging cohort at 4, 24, 48 and 120 h post-injection. Signal normalised to same thresholding (% ID g⁻¹).

significant difference in biodistribution between [⁸⁹Zr]Zr-D2-mAb or [⁸⁹Zr]Zr-1-mAb, with trends similar to *in vivo* biodistribution at the same time points (Fig. 7c, d; 10a and b).



Fig. 9 Representative 3D maximum intensity projections for all mice ([¹⁷⁷Lu]Lu-D2-mAb) in ¹⁷⁷Lu SPECT/CT imaging cohort at 48 and 120 h post-injection. Signal normalised to same thresholding (% ID g⁻¹).

Ex vivo biodistribution of [⁸⁹Zr]Zr-D2-mAb and [¹⁷⁷Lu]Lu-D2-mAb showed a significant difference in accumulation only within the gastrointestinal tract tissue, with all other tissues showing similar biodistribution (Fig. 10c). Both [⁸⁹Zr]Zr-D2-mAb and [¹⁷⁷Lu]Lu-D2-mAb showed minimal uptake into bone tissue (shoulder and knee) at 120 h post-injection, which correlated to the minimal uptake observed in the *in vivo* biodistribution (Fig. 8 and 9). This suggests that the [⁸⁹Zr]Zr-D2-mAb complex remained stable up to 120 h *in vivo*, with bone tissue signals a typical marker of complex instability arising from the osteophilic nature of ⁸⁹Zr(IV).^{35,65} The similar biodistribution properties of [⁸⁹Zr]Zr-D2-mAb and [¹⁷⁷Lu]Lu-D2-mAb indicate D2 could have potential utility for improved imaging-therapy dosimetry. The biodistribution of [⁸⁹Zr]Zr-D2-mAb and [⁸⁹Zr]Zr-1-mAb, and [¹⁷⁷Lu]Lu-D2-mAb and [¹⁷⁷Lu]Lu-



Fig. 10 *Ex vivo* biodistribution from gamma counting ($n = 3$) of [⁸⁹Zr]Zr-1-mAb (blue) and [⁸⁹Zr]Zr-D2-mAb (purple) at (a) 48 h p. i. or (b) 120 h p. i., or (c) [⁸⁹Zr]Zr-D2-mAb (purple, blue diagonal line) and [¹⁷⁷Lu]Lu-D2-mAb (purple, red diagonal line) at 120 h p. i. * $p \leq 0.05$, ** $p \leq 0.01$.



Fig. 11 Therapeutic efficacy of $[^{177}\text{Lu}]\text{Lu-D2-mAb}$ (purple) and $[^{177}\text{Lu}]\text{Lu-2-mAb}$ (red) in a murine LNCaP xenograft model, monitored over a 65 day period, with vehicle only PBS as control (green). Therapeutic efficacy in each cohort was measured by changes in tumour volume (a) and weight (b).

2-mAb, supported that D2-mAb formulated for either imaging or therapy could function similar to the matched single-chelator controls.⁶⁶

Therapeutic efficacy of $[^{177}\text{Lu}]\text{Lu-D2-mAb}$ (5) and $[^{177}\text{Lu}]\text{Lu-2-mAb}$. Differences in tumour volume and weight in a murine LNCaP xenograft model were monitored over a 65 day period to assess therapeutic efficacy of $[^{177}\text{Lu}]\text{Lu-D2-mAb}$ in comparison to $[^{177}\text{Lu}]\text{Lu-2-mAb}$, using phosphate-buffered saline (PBS) as a vehicle only control (Fig. 11).

Therapeutic efficacy of $[^{177}\text{Lu}]\text{Lu-D2-mAb}$ was found to closely match that of $[^{177}\text{Lu}]\text{Lu-2-mAb}$, with both compounds showing >80% reduction in tumour volume (<40 mm^3 after 65 days, Fig. 11a) and a slight increase in weight over the 65 day monitoring period (Fig. 11b). Both treatment cohorts exhibited a sustained reduction in tumour volume and sustained increase in weight over the observation period. The vehicle only cohort displayed an increase in tumour volume until ethical cut-offs were reached and mice were euthanised in accordance with the ethics approval guidelines (Fig. 11a). A trend of increasing weight loss was exhibited in the vehicle only cohort (Fig. 11b). These efficacy data showed the therapeutic potential of $[^{177}\text{Lu}]\text{Lu-D2-mAb}$, with similar performance to $[^{177}\text{Lu}]\text{Lu-2-mAb}$.

Conclusions

A covalent adduct between desferrioxamine B (DFOB, 1) and 1,4,7,10-tetraazacyclododecane-1,4,7,10-tetraacetic acid (DOTA, 2) separated by an L-lysine unit (DFOB-L-Lys- N^6 -DOTA, 3a) and containing an amine-terminated PEG motif for further chemistry and biomolecule conjugation was prepared and named 'D2' (3). The dual chelator D2-mAb complex was selectively radiolabelled with ^{89}Zr or ^{177}Lu to form $[^{89}\text{Zr}]\text{Zr-D2-mAb}$ (4) or $[^{177}\text{Lu}]\text{Lu-D2-mAb}$ (5), respectively, with the regioselectivity of $^{89}\text{Zr}(\text{IV})$ binding to the 1 region or $^{177}\text{Lu}(\text{III})$ to the 2 region, consistent with $\{^1\text{H}\}$ - ^{13}C NMR spectroscopic data, and complementary LC-MS

measurements supporting the 1 : 1 metal : ligand stoichiometry. $[^{89}\text{Zr}]\text{Zr-D2-mAb}$ and $[^{89}\text{Zr}]\text{Zr-1-mAb}$ showed similar PET/CT imaging function and $[^{177}\text{Lu}]\text{Lu-D2-mAb}$ and $[^{177}\text{Lu}]\text{Lu-2-mAb}$ showed similar therapeutic efficacy in a murine LNCaP prostate cancer xenograft model, with tumour localisation of $[^{177}\text{Lu}]\text{Lu-D2-mAb}$ shown by SPECT/CT imaging.

The functional properties of $[^{89}\text{Zr}]\text{Zr-D2-mAb}$ and $[^{177}\text{Lu}]\text{Lu-D2-mAb}$ demonstrates the potential of D2 as a single ^{89}Zr - ^{177}Lu theranostic agent which could streamline development pathways and improve patient dosimetry. D2 could also open opportunities to use other mixed-element radiometal pairs, including ^{89}Zr (imaging) and ^{225}Ac (therapy) or ^{212}Pb (therapy), which forms ongoing work, together with studies to examine properties of D2 co-labelled with ^{89}Zr and ^{nat}Lu (imaging) or with ^{177}Lu and ^{nat}Zr (therapy), with this iso-structural and iso-charge imaging-therapy pair predicted to have identical bio-distribution properties.

Experimental

Materials and methods

O-(1H-Benzotriazol-1-yl)-N,N',N'-tetramethyluronium hexafluorophosphate (HBTU, 98%) was obtained from Alfa Aesar. 2,2-Dimethyl-4-oxo-3,8,11,14,17-pentaoxa-5-azaicosan-20-oic acid (N-Boc-PEG4-CO₂H, 97%) was obtained from AmBeed. Ammonium acetate (97%) was obtained from APS Finechem. Sodium chloride (99.7%) was obtained from Chemsupply. HEPES (99.7%) was obtained from Formedium. DOTA-tri(^tBu) ester ($\geq 95\%$), DFOB-Ph-NCS ($\geq 94\%$) and DOTA-Bn-NCS ($\geq 94\%$) were obtained from Macrocyclics. Dichloromethane ($\geq 99.9\%$), N-ethyl-diisopropylamine ($\geq 98\%$), sodium hydrogen carbonate (99.7%), formic acid (98–100%) and magnesium sulfate, anhydrous ($\geq 98\%$) were obtained from Merck. Acetonitrile (99.9%) was obtained from RCI Labscan. Hydrochloric acid (36%) and toluene (99.5%) were obtained from Ajax Finechem. Desferrioxamine mesylate ($\geq 92.5\%$), N,N-dimethylformamide, anhydrous (99.8%), piperidine (99%), sodium hydroxide, gallium(III) acetylacetonate (99.99%), lutetium(III) chloride (99.99%), zirconium(IV) chloride ($\geq 99.9\%$), sodium hydroxide ($\geq 98\%$), methanol, anhydrous (99.8%), triethylamine ($\geq 99\%$), p-phenylene diisothiocyanate (98%), diethyl ether, anhydrous ($\geq 99.7\%$) and trifluoroacetic acid (99%) were obtained from Sigma-Aldrich. N $_{\alpha}$ -(9H-Fluoren-9-ylmethoxy) carbonyl-L-lysine hydrochloride ($>98\%$) was obtained from TCI. Milli-Q water was prepared using a Millipore Q-pod system and used as required. Healthy male Balb/c nude mice from 8 weeks old (weight ~20 g) were obtained from Ozgene. Human serum used for stability studies was obtained from Sigma-Aldrich (product number H4522). Human LNCaP clone FGC cells were obtained from the American Type Culture Collection (ATCC, product number CRL-1740) and used for xenograft tumour growth and cell assays.

Synthesis and characterisation

N^2 -(Fmoc)-L-Lys- N^6 -DOTA(O^tBu)₃ (8). To a solution of DOTA(O^tBu)₃ (7) (1.02 g, 1.78 mmol) in DMF (50 mL) was added



DIPEA (2.07 mL, 11.9 mmol). The mixture was stirred at r. t for 10 min before the addition of HBTU (652.2 mg, 1.72 mmol). After a further 30 min period of stirring, N^2 -(Fmoc)-L-lysine HCl (**6**) (707.2 mg, 1.92 mmol) was added, and the mixture was stirred for 2 h at r. t. before removing the solvent *in vacuo*. The residue was purified *via* solid phase extraction and the collected fractions were combined and the solvent was removed *in vacuo* to yield **8** (862.2 mg, 0.93 mmol, 52.2%) as a yellow-white solid. LRMS (ESI) m/z : $[M + H]^+$ calculated for $C_{49}H_{75}N_6O_{11}^+$ 923.55, found 923.35; $[M + 2H]^{2+}$ calculated for $C_{49}H_{76}N_6O_{11}^{2+}$ 462.28, found 462.35.

DFOB- N^2 -(Fmoc)-L-Lys- N^6 -DOTA(O^tBu)₃ (3e**).** To a solution of **8** (485.0 mg, 0.53 mmol) in DMF (10 mL) was added DIPEA (158.0 μ L, 0.91 mmol). The mixture was left to stir at r. t for 10 min before the addition of HBTU (202.2 mg, 0.53 mmol). This mixture was stirred at r. t for a further 30 min before the addition of desferrioxamine mesylate (**1**) (358.2 mg, 0.55 mmol), with the solution stirred at 50 °C for 1 h and the solvent removed *in vacuo*. The residue was dissolved in DCM (200 mL) and washed with saturated sodium bicarbonate (3 \times 100 mL) and saturated brine (100 mL) before drying the organic layer with anhydrous magnesium sulphate. The organic layer was filtered, and the solvent was removed *in vacuo* to yield **3e** as a yellow oil (623.5 mg, 0.43 mmol, 81.1%). LRMS (ESI) m/z : $[M + H]^+$ calculated for $C_{74}H_{121}N_{12}O_{18}^+$: 1465.89, found 1465.90; $[M + 2H]^{2+}$ calculated: 733.45, found 733.70; $[M + 3H]^{3+}$ calculated: 489.30, found 489.50.

DFOB-L-Lys- N^6 -DOTA(O^tBu)₃ (3d**).** Compound **3e** (623.5 mg, 0.43 mmol) was dissolved in a solution of DMF : piperidine (4 mL : 1 mL) and the solution was stirred at r. t for 1 h, and the solvent removed *in vacuo*. Cold ether (approx. 10 mL) was added to the residue and after stirring for 15 min at 0 °C, the ether was decanted, and the residue dried *in vacuo* to yield **3d** as a pale-yellow oil. LRMS (ESI) m/z : $[M + H]^+$ calculated for $C_{59}H_{111}N_{12}O_{16}^+$: 1243.83, found 1243.60; $[M + 2H]^{2+}$ calculated: 622.42, found 622.50; $[M + 3H]^{3+}$ calculated: 415.28, found 415.30. This intermediate was used in subsequent steps without further purification.

DFOB-L-Lys- N^6 -DOTA (3a**).** The sample of **3d** was dissolved in a solution of TFA : DCM (4.5 mL : 0.5 mL) and left to stir for 16 h at r. t before removing the solvent *in vacuo*. The residue was resuspended in toluene (10 mL) and the solvent was removed *in vacuo*. The residue was further purified *via* HPLC and lyophilised to yield **3a** (95.4 mg, 88.8 μ mol, 20.7%) as a white powder. ^{13}C NMR (600 MHz, D₂O) δ 176.93, 174.79, 174.77, 173.80 (2C), 173.55, 172.09, 169.89, 169.75, 169.72, 56.90, 56.31 (2C), 55.06, 53.08, 51.78 (2C), 50.52, 50.39, 48.54, 48.39, 48.29, 48.01, 47.85, 47.82, 47.66, 39.27, 39.15, 39.08, 38.46, 30.42, 30.40, 30.06, 27.88, 27.85, 27.81, 27.61, 27.59, 27.20, 25.40 (3C), 23.08, 22.99 (2C), 21.00, 19.21. HRMS (H-ESI II) m/z : $[M + H]^+$ calculated for $C_{47}H_{87}N_{12}O_{16}^+$: 1075.63575, found 1075.63498; $[M + 2H]^{2+}$ calculated: 538.321514, found 538.320867; $[M + 3H]^{3+}$ calculated: 359.216768, found 359.216712.

DFOB- N^2 -(N-Boc-PEG4)-L-Lys- N^6 -DOTA(O^tBu)₃ (3c**).** To a solution of **3d** (532.9 mg, 0.43 mmol) in DMF was added DIPEA (124.5 μ L, 0.71 mmol). The mixture was stirred at r. t for 10 min before the addition of HBTU (163.4 mg, 0.43 mmol).

This mixture was stirred at r. t for a further 30 min before the addition of *N*-Boc-PEG4-CO₂H (approx. 146 mg, 0.40 mmol). The mixture was stirred for 2 h at r. t. and the solvent was removed *in vacuo*. The residue was dissolved in DCM (200 mL) and washed with saturated sodium bicarbonate (3 \times 100 mL) and saturated brine (100 mL) before drying the organic layer with anhydrous magnesium sulphate. The organic layer was filtered, and the solvent was removed *in vacuo* to yield **3c** (574.9 mg, 0.40 mmol, 100%) as a yellow oil. LRMS (ESI) m/z : $[M + 2H]^{2+}$ calculated for $C_{75}H_{141}N_{13}O_{23}^{2+}$: 796.0, found 796.0; $[M + 3H]^{3+}$ calculated: 497.65, found 497.80. *N*-Boc fragmentation was typically observed during LC-MS analysis.

DFOB- N^2 -(PEG4)-L-Lys- N^6 -DOTA (3**) (**D2**).** The sample of **3c** was dissolved in a mixture of TFA : DCM (9 : 1) which was stirred at r. t for 16 h and the solvent was removed *in vacuo*. The residue was resuspended in toluene (10 mL) and the solvent was removed *in vacuo*. The residue was purified *via* solid phase extraction (100% ACN for 4 min, 100% H₂O for 2.4 min, load, 100% ACN for 2 min, 20–80% ACN : H₂O in 5% steps at 3 min each step, 95% ACN : H₂O for 3 min, 10 mL min^{−1}), and the collected fractions were combined, and the solvent was removed *in vacuo*. A fraction of the semi-pure product was further purified *via* HPLC and lyophilised to yield **D2** (**3**) as a white-brown solid (3.0 mg, 2.27 μ mol). 1H NMR (600 MHz, D₂O) δ 8.42 (s, 0H), 8.42 (s, 0H), 4.01 (t, J = 6.7 Hz, 1H), 3.82 (qd, J = 16.8, 7.8 Hz, 5H), 3.64 (d, J = 10.3 Hz, 5H), 3.61 (d, J = 6.8 Hz, 5H), 3.48–3.41 (m, 14H), 3.18 (q, J = 8.7, 6.8 Hz, 10H), 3.11 (s, 7H), 3.01 (d, J = 15.5 Hz, 3H), 2.91 (d, J = 15.3 Hz, 3H), 2.80 (t, J = 7.1 Hz, 5H), 2.50 (t, J = 7.1 Hz, 5H), 2.15–2.10 (m, 4H), 1.90 (q, J = 7.0 Hz, 3H), 1.66–1.41 (m, 21H), 1.31 (t, J = 7.8 Hz, 8H). ^{13}C NMR (600 MHz, D₂O) δ 172.39, 172.34 (3C), 172.10 (2C), 171.88, 171.84, 170.6 (2C), 170.3, 70.22 (3C), 70.18, 70.11, 70.05, 69.97, 67.28, 58.92 (2C), 56.29, 55.93 (3C), 53.16, 51.55 51.22, 50.3 (2C), 47.54 (2C), 47.24 (3C), 38.99, 38.90 (2C), 39.77, 38.71, 36.30, 30.49, 30.43, 29.28 (3C), 29.09 (3C), 29.02, 28.13, 28.09, 23.97 (2C), 23.90, 23.21, 20.82 (2C). LRMS (ESI) m/z : $[M + 2H]^{2+}$ calculated for $C_{58}H_{109}N_{13}O_{21}^{2+}$: 661.89, found 662.0; $[M + 3H]^{3+}$ calculated: 441.60, found 441.80; $[M + 4H]^{4+}$ calculated: 331.45, found 331.60.

DFOB- N^2 -(PEG4-Ph-NCS)-L-Lys- N^6 -DOTA (3b**) (**D2-Ph-NCS**).** To a solution of semi-pure **3** (**D2**) (114.9 mg, 86.93 μ mol) in DMF (4.2 mL) was added triethylamine (242.0 μ L, 1.74 mmol). This solution was added to a solution of 1,4-phenylenediisothiocyanate (165.9 mg, 0.86 mmol) in DMF (12.5 mL). After centrifugation (800 rpm, 90 min), the reaction mixture was aliquoted into 6 equal fractions and diethyl ether (9 mL) was added to each fraction. The fractions were cooled at 4 °C for 2 h and after centrifugation (4000 rpm, 5 min), the diethyl ether was decanted, and each fraction was washed with diethyl ether (10 mL). Excess diethyl ether was decanted, and the pellets were dried with N₂. To the dry pellets was added DMF (196.3 μ L), MeOH (1.47 mL) and diethyl ether (9 mL) before refrigerating for 16 h. Upon completion, the ether was removed, and the pellet was washed with diethyl ether (10 mL) and dried with N₂. The pellets were dissolved in a mixture of ACN : H₂O (3 : 7) and combined before further purification using HPLC. Fractions containing the product were lyophilised to yield **D2-Ph-NCS** (**3b**)



as a white solid (32.5 mg, 21.47 μmol , 24.7%). ^1H NMR (700 MHz, D_2O) δ 7.36 (q, J = 8.7 Hz, 4H), 4.19 (dd, J = 8.7, 5.7 Hz, 1H), 3.76 (t, J = 6.2 Hz, 4H), 3.74–3.56 (m, 31H), 3.23–3.12 (m, 11H), 2.79 (td, J = 7.1, 2.7 Hz, 5H), 2.58–2.50 (m, 4H), 2.49 (t, J = 7.1 Hz, 5H), 2.13 (s, 3H), 1.72–1.61 (m, 5H), 1.61 (dd, J = 9.6, 5.7 Hz, 5H), 1.51 (ddt, J = 14.3, 10.6, 7.2 Hz, 12H), 1.34 (s, 3H), 1.29 (tp, J = 14.9, 7.9, 6.6 Hz, 8H). Acquired using water suppression. ^{13}C NMR (700 MHz, D_2O) δ 179.82, 174.84, 174.04, 173.93, 173.85, 173.63, 134.81, 126.91, 126.37, 69.77, 69.73, 69.66, 66.73, 55.33, 54.24, 51.59, 47.97, 47.93, 47.78, 44.32, 39.26, 39.20, 35.77, 30.84, 30.56, 28.05, 28.01, 27.98, 27.91, 27.77, 27.73, 25.57, 25.53, 23.14, 22.99, 22.63, 19.66, 19.34. HRMS (H-ESI II) m/z : $[\text{M} + \text{H}]^+$ calculated for $\text{C}_{66}\text{H}_{112}\text{N}_{15}\text{O}_{21}\text{S}_2^+$: 1514.75931, found 1514.75500; $[\text{M} + 2\text{H}]^{2+}$ calculated: 757.883295, found 757.881144; $[\text{M} + 3\text{H}]^{3+}$ calculated: 505.591289, found 505.593039. A larger-scale sample of **D2-Ph-NCS** provided by AusPep as a custom synthesis (90% pure by RP-HPLC) was used for selected studies (Fig. 5–10; S12–S15, S26–S30 and Tables S1–S7†), with all remaining work using compounds synthesized in-house (Fig. 2–4, S1–S11 and S16–S25†).

Data availability

All relevant data are presented in the main text and ESI† (general information, NMR spectra, LC-MS data, preparation of **D2-mAb**, **1-mAb**, **2-mAb**, radiolabelling, biology).

Author contributions

The study was conceptualized by RC, MPW, and JLW. All authors contributed to the design of the study methods and the analysis of results. Experimental data was generated by JLW, ZHH, SG, NLF, JH, DTA, KM and WT. Clinical grade HuJ591 was provided by MPW and AI. The first draft of the manuscript was written by RC and JLW, with all authors contributing to manuscript review, editing, and data presentation, and approving the final submission.

Conflicts of interest

The authors declare the following competing financial interest(s). Intellectual property related to this research includes Rachel Codd and James L. Wood as listed inventors. Michael P. Wheatcroft and Alesia Ivashkevich are employees of Telix Pharmaceuticals.

Acknowledgements

Acknowledgement is made to The University of Sydney (Industry Engagement Seed Fund, Commercialisation CDIP Fund) and Telix Pharmaceuticals for funding, and to the Australian Government with the support of an Australian Government Research Training Program (RTP) Scholarship (J. L. W). Prof. T. Pukala (University of Adelaide) is acknowledged for initial MALDI measurements of **D2-mAb**. This research was facilitated by access to Sydney Mass Spectrometry, a core research facility

at the University of Sydney. We thank Dr B. Mohanty for assistance with NMR spectroscopy carried out at Sydney Analytical, a core research facility at the University of Sydney. We thank the National Imaging Facility (NIF) at the University of Queensland and the National Collaborative Research Infrastructure Strategy (NCRIS) for use of the imaging facilities.

Notes and references

‡ Abbreviations: mCRPC, metastatic castration-resistant prostate cancer; DFOB, desferrioxamine B; DOTA, 1,4,7,10-tetraazacyclododecane-1,4,7,10-tetraacetic acid; HBED-CC, 3,3'-((ethane-1,2-diylbis(carboxymethyl)azanediyl)) bis(methylene) bis(4-hydroxy-3,1-phenylene)dipropionic acid; NET, neuroendocrine tumours; PSMA, prostate-specific membrane antigen.

§ Examples of other activating groups are NHS-, maleimide-, or ethyl squaramide groups installed from reactions of **D2** with bis-*N*-succinimidyl glutarate, *N*-succinimidyl-3-maleimidopropionate, or diethyl squarate, respectively.

¶ All studies involving animals were conducted in accordance with the guidelines of the Animal Ethics Committee of The University of Queensland, and the Australian Code for the Care and Use of Animals for Scientific Purposes (AEC approval number: 2022/AE000135).

- 1 T. J. Wadas, E. H. Wong, G. R. Weisman and C. J. Anderson, *Chem. Rev.*, 2010, **110**, 2858–2902.
- 2 E. W. Price and C. Orvig, *Chem. Soc. Rev.*, 2014, **43**, 260–290.
- 3 T. I. Kostelnik and C. Orvig, *Chem. Rev.*, 2019, **119**, 902–956.
- 4 E. Boros and A. B. Packard, *Chem. Rev.*, 2019, **119**, 870–901.
- 5 L. Bodei, K. Herrmann, H. Schöder, A. M. Scott and J. S. Lewis, *Nat. Rev. Clin. Oncol.*, 2022, **19**, 534–550.
- 6 G. Sgouros, L. Bodei, M. R. McDevitt and J. R. Nedrow, *Nat. Rev. Drug Discovery*, 2020, **19**, 589–608.
- 7 K. A. Morgan, S. E. Rudd, A. Noor and P. S. Donnelly, *Chem. Rev.*, 2023, **123**, 12004–12035.
- 8 R. J. Hicks, P. Jackson, G. Kong, R. E. Ware, M. S. Hofman, D. A. Pattison, T. A. Akhurst, E. Drummond, P. Roselt, J. Callahan, R. Price, C. M. Jeffery, E. Hong, W. Noonan, A. Herschtal, L. J. Hicks, A. Hedt, M. Harris, B. M. Paterson and P. S. Donnelly, *J. Nucl. Med.*, 2019, **60**, 777–785.
- 9 C. Cullinane, C. M. Jeffery, P. D. Roselt, E. M. van Dam, S. Jackson, K. Kuan, P. Jackson, D. Binns, J. van Zuylenkom, M. J. Harris, R. J. Hicks and P. S. Donnelly, *J. Nucl. Med.*, 2020, **61**, 1800–1805.
- 10 I. Captain, G. J.-P. Deblonde, P. B. Rupert, D. D. An, M.-C. Illy, E. Rostan, C. Y. Ralston, R. K. Strong and R. J. Abergel, *Inorg. Chem.*, 2016, **55**, 11930–11936.
- 11 K. P. Carter, G. J.-P. Deblonde, T. D. Lohrey, T. A. Bailey, D. D. An, K. M. Shield, W. W. Lukens and R. J. Abergel, *Commun. Chem.*, 2020, **3**, 1–7.
- 12 J. Xu, F. Cai, Z. Luo, W. Fan, J. Dai, J. Cui, S. Li, C. Geng, Q. Zheng, Z. Wang and X. Tang, *Eur. J. Nucl. Med. Mol. Imaging*, 2022, **49**, 2618–2633.
- 13 L. Wharton, C. Zhang, H. Yang, J. Zeisler, V. Radchenko, C. Rodríguez-Rodríguez, M. Osooly, B. O. Patrick, K.-S. Lin, F. Bénard, P. Schaffer and C. Orvig, *Bioconjugate Chem.*, 2022, **33**, 505–522.
- 14 B. A. Vaughn, C. S. Loveless, S. J. Cingoranelli, D. Schlyer, S. E. Lapi and E. Boros, *Mol. Pharmaceutics*, 2021, **18**, 4511–4519.



- 15 Z. Baranyai, G. Tircsó and F. Rösch, *Eur. J. Inorg. Chem.*, 2020, 36–56.
- 16 R. C. Mease, C. M. Kang, V. Kumar, S. R. Banerjee, I. Minn, M. Brummet, K. L. Gabrielson, Y. Feng, A. Park, A. P. Kiess, G. Sgouros, G. Vaidyanathan, M. R. Zalutsky and M. G. Pomper, *J. Nucl. Med.*, 2022, **63**, 259–267.
- 17 J. R. Dilworth and S. I. Pascu, *Chem. Soc. Rev.*, 2018, **47**, 2554–2571.
- 18 N. B. Bhatt, D. N. Pandya and T. J. Wadas, *Molecules*, 2018, **23**, 638.
- 19 M. Chomet, G. A. M. S. van Dongen and D. J. Vugts, *Bioconjugate Chem.*, 2021, **32**, 1315–1330.
- 20 V. Radchenko, S. Busse and F. Roesch, *Nucl. Med. Biol.*, 2014, **41**, 721–727.
- 21 P. M. Smith-Jones, B. Stolz, C. Bruns, R. Albert, H. W. Reist, R. Fridrich and H. R. Macke, *J. Nucl. Med.*, 1994, **35**, 317–325.
- 22 C. J. Mathias, M. R. Lewis, D. E. Reichert, R. Laforest, T. L. Sharp, J. S. Lewis, Z.-F. Yang, D. J. Waters, P. W. Snyder, P. S. Low, M. J. Welch and M. A. Green, *Nucl. Med. Biol.*, 2003, **30**, 725–731.
- 23 J. P. Holland, in *Handbook of Radiopharmaceuticals: Methodology and Applications*, ed. M. R. Kilbourn and P. J. Scott, John Wiley & Sons Ltd, Hoboken, NJ, 2nd edn, 2021, ch. 11, pp. 343–374.
- 24 U. Pandey, N. Gamre, Y. Kumar, P. Shetty, H. Dev Sarma and A. Dash, *J. Radioanal. Nucl. Chem.*, 2016, **307**, 187–194.
- 25 C. J. Marmion, D. Griffith and K. B. Nolan, *Eur. J. Inorg. Chem.*, 2004, 3003–3016.
- 26 R. Codd, *Coord. Chem. Rev.*, 2008, **252**, 1387–1408.
- 27 L. Biró, P. Buglyó and E. Farkas, *Curr. Med. Chem.*, 2021, **28**, 7209–7237.
- 28 M. Patra, A. Bauman, C. Mari, C. A. Fischer, O. Blacque, D. Haussinger, G. Gasser and T. L. Mindt, *Chem. Commun.*, 2014, **50**, 11523–11525.
- 29 S. E. Rudd, P. Roselt, C. Cullinane, R. J. Hicks and P. S. Donnelly, *Chem. Commun.*, 2016, **52**, 11889–11892.
- 30 T. Richardson-Sanchez, W. Tieu, M. P. Gotsbacher, T. J. Telfer and R. Codd, *Org. Biomol. Chem.*, 2017, **15**, 5719–5730.
- 31 W. Tieu, T. Lifa, A. Katsifis and R. Codd, *Inorg. Chem.*, 2017, **56**, 3719–3728.
- 32 C. J. M. Brown, M. P. Gotsbacher and R. Codd, *Aust. J. Chem.*, 2020, **73**, 969–978.
- 33 J. P. Holland, *Inorg. Chem.*, 2020, **59**, 2070–2082.
- 34 A. K. Salih, S. J. Raheem, M. Dominguez Garcia, W. K. Ahiahonu and E. W. Price, *Inorg. Chem.*, 2022, **61**, 20964–20976.
- 35 R. Raavé, G. Sandker, P. Adumeau, C. B. Jacobsen, F. Mangin, M. Meyer, M. Moreau, C. Bernhard, L. Da Costa, A. Dubois, V. Goncalves, M. Gustafsson, M. Rijpkema, O. Boerman, J.-C. Chambron, S. Heskamp and F. Denat, *Eur. J. Nucl. Med. Mol. Imaging*, 2019, **46**, 1966–1977.
- 36 H. Damerow, X. Cheng, V. von Kiedrowski, R. Schirrmacher, B. Wängler, G. Fricker and C. Wängler, *Pharmaceutics*, 2022, **14**, 2214.
- 37 E. Khozeimeh Sarbisheh, K. L. Summers, A. K. Salih, J. J. H. Cotelesage, A. Zimmerling, I. J. Pickering, G. N. George and E. W. Price, *Inorg. Chem.*, 2023, **62**, 2637–2651.
- 38 Y. Toporivska and E. Gumienna-Kontecka, *J. Inorg. Biochem.*, 2019, **198**, 110753.
- 39 B. Borgias, A. D. Hugl and K. N. Raymond, *Inorg. Chem.*, 1989, **28**, 3538–3545.
- 40 B. J. Hernlem, L. M. Vane and G. D. Sayles, *Inorg. Chim. Acta*, 1996, **244**, 179–184.
- 41 W. P. Cacheris, S. K. Nickle and A. D. Sherry, *Inorg. Chem.*, 1987, **26**, 958–960.
- 42 V. Kubiček, J. Havlíčková, J. Kotek, G. Tircsó, P. Hermann, E. Tóth and I. Lukeš, *Inorg. Chem.*, 2010, **49**, 10960–10969.
- 43 C. G. Pippin, T. J. McMurphy, M. W. Brechbiel, M. McDonald, R. Lambrecht, D. Milenic, M. Roselli, D. Colcher and O. A. Gansow, *Inorg. Chim. Acta*, 1995, **239**, 43–51.
- 44 M. T. Ma, L. K. Meszaros, B. M. Paterson, D. J. Berry, M. S. Cooper, Y. M. Ma, R. C. Hider and P. J. Blower, *Dalton Trans.*, 2015, **44**, 4884–4900.
- 45 D. N. Pandya, N. Bhatt, H. Yuan, C. S. Day, B. M. Ehrmann, M. Wright, U. Bierbach and T. J. Wadas, *Chem. Sci.*, 2017, **8**, 2309–2314.
- 46 D. N. Pandya, K. E. Henry, C. S. Day, S. A. Graves, V. L. Nagle, T. R. Dilling, A. Sinha, B. M. Ehrmann, N. B. Bhatt, Y. Menda, J. S. Lewis and T. J. Wadas, *Inorg. Chem.*, 2020, **59**, 17473–17487.
- 47 S. Aime, M. Botta and G. Ermondi, *Inorg. Chem.*, 1992, **31**, 4291–4299.
- 48 S. Aime, M. Botta, M. Fasano, M. P. M. Marques, C. F. G. C. Geraldies, D. Pubanz and A. E. Merbach, *Inorg. Chem.*, 1997, **36**, 2059–2068.
- 49 J. Blahut, P. Hermann, Z. Tošner and C. Platas-Iglesias, *Phys. Chem. Chem. Phys.*, 2017, **19**, 26662–26671.
- 50 M. Briand, M. L. Aulsebrook, T. L. Mindt and G. Gasser, *Dalton Trans.*, 2017, **46**, 16387–16389.
- 51 A. Guillou, D. F. Earley, S. Klingler, E. Nisli, L. J. Nüesch, R. Fay and J. P. Holland, *Bioconjugate Chem.*, 2021, **32**, 1263–1275.
- 52 D. Novak, T. Tomašič, M. Krošelj, U. Javornik, J. Plavec, M. Anderluh and P. K. Peitl, *ChemMedChem*, 2021, **16**, 155–163.
- 53 M. W. Giese, R. H. Woodman, G. T. Hermanson and P. D. Davis, in *Chemical Linkers in Antibody–Drug Conjugates (ADCs)*, ed. F. L. van Delft and J. M. Lambert, The Royal Society of Chemistry, 2022, vol. 81, ch. 9, pp. 286–376.
- 54 H. D. King, D. Yurgaitis, D. Willner, R. A. Firestone, M. B. Yang, S. J. Lasch, K. E. Hellström and P. A. Trail, *Bioconjugate Chem.*, 1999, **10**, 279–288.
- 55 G. Liao, Z. Zhou, S. Burgula, J. Liao, C. Yuan, Q. Wu and Z. Guo, *Bioconjugate Chem.*, 2015, **26**, 466–476.
- 56 C. Abate-Shen and F. Nunes de Almeida, *Cancer Res.*, 2022, **82**, 1689–1691.
- 57 D. M. Nanus, M. I. Milowsky, L. Kostakoglu, P. M. Smith-Jones, S. Vallabahajosula, S. J. Goldsmith and N. H. Bander, *J. Urol.*, 2003, **170**, S84–S88.



- 58 N. T. Viola-Villegas, K. K. Sevak, S. D. Carlin, M. G. Doran, W. H. Evans, D. W. Bartlett, A. M. Wu and J. S. Lewis, *Mol. Pharmaceutics*, 2014, **11**, 3965–3973.
- 59 S. Vallabhajosula, P. M. Smith-Jones, V. Navarro, S. J. Goldsmith and N. H. Bander, *Prostate*, 2004, **58**, 145–155.
- 60 R. I. J. Merkx, D. Lobeek, M. Konijnenberg, L. D. Jiménez-Franco, A. Kluge, E. Oosterwijk, P. F. A. Mulders and M. Rijpkema, *Eur. J. Nucl. Med. Mol. Imaging*, 2019, **48**, 3277–3285.
- 61 R. I. J. Merkx, M. Rijpkema, G. M. Franssen, A. Kip, B. Smeets, A. Morgenstern, F. Bruchertseifer, E. Yan, M. P. Wheatcroft, E. Oosterwijk, P. F. A. Mulders and S. Heskamp, *Pharmaceutics*, 2022, **15**, 570.
- 62 S. Vallabhajosula, A. Nikolopoulou, Y. S. Jhanwar, G. Kaur, S. T. Tagawa, D. M. Nanus, N. H. Bander and S. J. Goldsmith, *Curr. Radiopharm.*, 2016, **9**, 44–53.
- 63 J. M. White, F. E. Escorcía and N. T. Viola, *Theranostics*, 2021, **11**, 6293–6314.
- 64 G. L. DeNardo, S. J. Kennel, J. A. Siegel and S. J. Denardo, *Clin. Lymphoma*, 2004, **5**, S5–S10.
- 65 M. Chomet, M. Schreurs, M. J. Bolijn, M. Verlaan, W. Beaino, K. Brown, A. J. Poot, A. D. Windhorst, H. Gill, J. Marik, S. Williams, J. Cowell, G. Gasser, T. L. Mindt, G. A. M. S. van Dongen and D. J. Vugts, *Eur. J. Nucl. Med. Mol. Imaging*, 2021, **48**, 694–707.
- 66 U. Eberlein, M. Cremonesi and M. Lassmann, *J. Nucl. Med.*, 2017, **58**(Suppl 2), 97S–103S.

