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Effect of the heteroatom on the magnetic and luminescence properties of hybrid lanthanide-substituted Keggin-type polyoxometalates†

Janire Bustamante-Fernández,^{a,b} Estibaliz Ruiz-Bilbao,^b
Corina Rodríguez-Esteban,^b Mathieu Gonidec,^c José A. García,^d
Luis Lezama,^a Juan M. Gutiérrez-Zorrilla,^{a,b} Itziar Oyarzabal^{*b,e} and
Beñat Artetxe^{id} ^{*a}

Replacement of the heteroatom from Si to Ge has a strong influence on the luminescence properties of a series of hybrid, sandwich-type $K_5[Ln(\alpha\text{-GeW}_{11}O_{39})(C_{20}H_{22}Br_2N_2O_4)]\cdot 14H_2O$ (1_{Ge-Ln} , $Ln = Sm$ to Lu) anions. Interestingly, the Gd and Yb derivatives retain their ability to display slow relaxation of magnetisation.

The combination of lacunary polyoxometalates (POMs) that act as inorganic multidentate O donor ligands with lanthanide(III) ions^{1,2} (Ln) constitutes a suitable approach for the preparation of systems exhibiting interesting catalytic,^{3,4} optical^{5,6} and magnetic properties.^{7,8} In fact, the high oxophilicity and coordination numbers of 4f metal ions allow the generation of some of the largest and most outstanding POMs known to date.^{9–11} In spite of their structural simplicity, small sandwich-type species that enclose a single Ln^{III} ion between two monolacunary fragments are some of the most explored families. This interest mainly originates from their potential applications in molecular magnetism, including fields like molecular spintronics,¹² data-storage¹³ and quantum computing.¹⁴ In close analogy to double-decker type coordination complexes,¹⁵ the strong magnetic anisotropy together with the large ground-state magnetic moments of Ln ions has allowed Peacock-Weakley type anions to show single-molecule magnet (SMM) behaviour. Examples of these sandwich-type species

with monolacunary polyoxotungstates enclosing a central 4f metal in a square antiprismatic D_{4d} geometry include $[Ln(W_5O_{18})_2]^{9-}$ ($Ln = Tb, Dy, Ho$ and Er) and $[Ln(\beta_2\text{-SiW}_{11}O_{39})_2]^{13-}$ ($Ln = Dy, Ho, Er$ and Yb) anions.^{16,17} These Lindqvist- or Keggin-type fragments display some advantages in comparison with classical organic ligands used in coordination complexes: (i) POMs show higher thermal and chemical stability both in solution and the solid state; (ii) the rigidity of the ligand can lead to highly symmetric environments for Ln centres, and (iii) their large size and diamagnetism ensure magnetic isolation over the neighbouring species. Furthermore, 4f metal ions can be sensitised *via* photoexcitation of the $O \rightarrow W$ ligand-to-metal charge-transfer states to result in intense light emission.^{18,19}

Despite their potential, there are only a few reports on families of mononuclear complexes with the Ln ion displaying simultaneous coordination with lacunary polyoxotungstates and organic ligands, to the best of our knowledge. Monolacunary Keggin-type anions were reacted with different Ln /organic ligand combinations such as Dy /phthalocyanine and Sm /benzoate leading to systems exhibiting slow relaxation of magnetisation and tuneable luminescence, respectively.^{20,21} It is also noteworthy that the $[n\text{-NBu}_4]_3[LnH(PW_{11}O_{39})(phen)_2]\cdot H_2O$ ($phen = \text{phenanthroline}$) family exhibits interesting magneto-luminescence properties.^{22,23}

Very recently, some of us published a series of hybrid anions formed by Ln -containing Keggin-type anions and the compartmental organic ligand N,N' -dimethyl- N,N' -bis(2-hydroxy-3-formyl-5-bromobenzyl)-ethylenediamine (H_2L , Fig. 1).²⁴ This molecule displays an external O_4 site which can accommodate large oxophilic 4f centres and an inner N_2O_2 pocket available for the coordination of smaller 3d metal ions. This combination can result in heterometallic 3d–4f complexes, which represents a suitable way to improve the SMM behaviour of a given system.²⁵ In a first approach, we selected the monolacunary Keggin type tungstosilicate and reacted it with mid-to-late Ln cations, leading to a family of ten isostructural compounds, namely $K_5[Ln(\alpha\text{-SiW}_{11}O_{39})(H_2L)]\cdot 14H_2O$

^aDepartamento de Química Orgánica e Inorgánica, Facultad de Ciencia y Tecnología, Universidad del País Vasco UPV/EHU, P.O. Box 644, 48080 Bilbao, Spain.

E-mail: benat.artetxe@ehu.eus

^bBCMaterials, Edificio Martina Casiano, 3rd Floor, UPV/EHU Science Park, Barrio Sarriena s/n, 48940 Leioa, Spain. E-mail: itziar.oyarzabal@bcmaterials.net

^cUniv. Bordeaux, CNRS, Bordeaux INP, ICMCB, UMR 5026, F-33600 Pessac, France

^dDepartamento de Física Aplicada II, Facultad de Ciencia y Tecnología. Universidad del País Vasco UPV/EHU, P.O. Box 644, 48080 Bilbao, Spain

^eIKERBASQUE, Basque Foundation for Science, 48009 Bilbao, Spain

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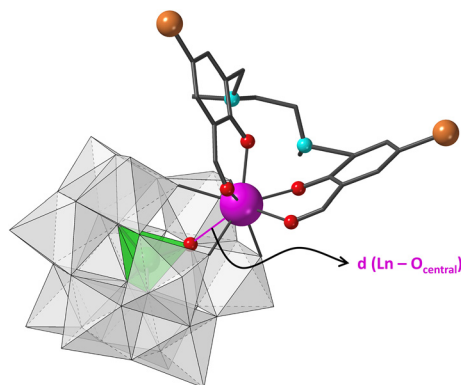


Fig. 1 Molecular structure of hybrid $[\text{Ln}(\text{H}_2\text{L})(\alpha\text{-GeW}_{11}\text{O}_{39})]^{5-}$ anions in $\mathbf{1}_{\text{Ge}}\text{-Ln}$. The $\text{Ln}\cdots\text{O}_{\text{central}}$ bond is represented as a purple line and discussed in the text. Colour code: WO_6 , grey; GeO_4 , green; Ln, purple; C, black, O, red; N, blue; Br, orange. H atoms have been omitted for clarity.

($\mathbf{1}_{\text{Si}}\text{-Ln}$, Ln = Sm to Lu). Encouraged by the fact that the Gd and Yb derivatives showed slow relaxation of magnetisation and that efficient emission in the visible and NIR region was reported for different members of the family, we decided to evaluate the influence of the modification of its constituents on the final magneto-luminescence properties.

First, different first row transition metal salts were included in the original reaction, but no incorporation of 3d metal was observed regardless of the synthetic conditions (nature of the 3d metal and the counterion, reaction temperature, aqueous reaction media). Then, we tried to evaluate the effect of the heteroatom without modifying the total charge of the anion by replacing monolacunary Keggin type tungstosilicates with tungstogermanates. Following the same synthetic procedure, nine isostructural compounds namely $\text{K}_5[\text{Ln}(\alpha\text{-GeW}_{11}\text{O}_{39})(\text{H}_2\text{L})]\cdot 14\text{H}_2\text{O}$ ($\mathbf{1}_{\text{Ge}}\text{-Ln}$, Ln = Sm to Yb) were isolated (Fig. S1–S7 and Tables S1–S6†). Single crystal X-ray diffraction studies revealed that the molecular structure of each hybrid polyanion is similar to those of the $\mathbf{1}_{\text{Si}}\text{-Ln}$ derivatives (Fig. 1). These are composed of a trivalent Ln cation simultaneously coordinated by (i) the Keggin-type monolacunary fragment through the four O atoms delimiting the vacant site and (ii) the outer O_4 site from the organic ligand formed by two aldehyde and two phenoxy groups. Regarding the crystal packing, the weak supramolecular interactions including π – π stacking between aromatic rings and $\text{Br}\cdots\text{Br}$ bonds involving H_2L ligands give rise to hexameric assemblies formed by different hybrid units (Fig. S7†).

The geometry of the LnO_8 coordination polyhedra was studied through continuous shape measures (CShM),²⁶ which show that the lowest distortion values were obtained using the biaugmented trigonal prism (BTPR, C_{2v} , CShM: 0.556–0.707) and square antiprism (SAPR, D_{4d} , CShM: 0.852–1.061) as reference shapes (Table S7†). The BTPR vs. SAPR shape map (Fig. S8†) displays low deviation values for all the 4f ions from the minimal distortion pathway between both reference shapes (<0.3). This fact confirms that their geometry can be

best described as BTPR distorted towards SAPR. The scatter of LnO_8 polyhedra together with those from compounds $\mathbf{1}_{\text{Si}}\text{-Ln}$ in the shape map evidences that they can be easily classified into two different groups, with those reported in this work being closer to ideal BTPR. This difference arises from the out-of-pocket coordination mode of Ln ions toward the Keggin skeleton. The larger size of the heteroatom in $\mathbf{1}_{\text{Ge}}\text{-Ln}$ allows 4f ions to be better incorporated into the lacunary site, in such a way that the distances to central O atoms are shortened in ca. 0.2 Å when moving from Si to Ge ($\text{Ln}\cdots\text{O}_{\text{Ge}} = 3\text{--}3.1$ Å, Fig. 1). This fact decreases the average folding angle between the aromatic rings of the organic ligand in ca. 2°.

In order to explore if the changes in the local geometry of the Ln impact the magnetic properties, direct-current (dc) magnetic susceptibility measurements were carried out by using single crystals of $\mathbf{1}_{\text{Ge}}\text{-Ln}$ (Fig. S9 and S10, Table S8†). Starting with $\mathbf{1}_{\text{Ge}}\text{-Gd}$, the nearly constant value of the χ_{MT} product in the studied temperature range implies the absence of significant interactions between Gd ions, which is in agreement with the large distances between the Ln ions (>8 Å) within the crystal packing. The fitting of the Q-band EPR spectra at room temperature (Fig. S11†) yields $D = 0.0831\text{ cm}^{-1}$, $E = 0.0244\text{ cm}^{-1}$, $g_x = 1.993$, $g_y = 1.992$ and $g_z = 1.983$, which effectively reproduce the susceptibility and magnetisation data when $g = 2$. Compared to $\mathbf{1}_{\text{Si}}\text{-Gd}$, the modifications in the Ln coordination environment lead to negligible differences in the zero field splitting parameters (D and E) and energy levels (Fig. S12, Table S9†), in agreement with the isotropic nature of gadolinium ions. As in the case of $\mathbf{1}_{\text{Si}}\text{-Gd}$, $\mathbf{1}_{\text{Ge}}\text{-Gd}$ displays slow relaxation of magnetisation at low temperatures (Fig. 2 and S13–S15†). The Arrhenius fit of the relaxation times at higher temperatures leads to an energy barrier of 8.1 cm^{-1} with $\tau_0 = 6.5 \times 10^{-6}\text{ s}$, which is significantly larger than the separation between the ground and first excited states (0.40 cm^{-1}). This, together with the deviation of the relaxation times from linearity and the relatively large α values obtained in the Cole–Cole plots (0.17 (2 K)–0.02 (8 K)), suggests the pres-

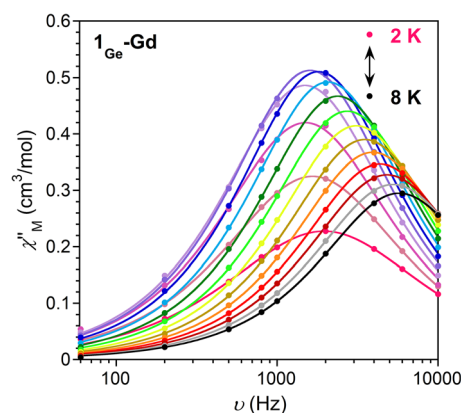


Fig. 2 Frequency dependence of the out-of-phase component of the ac susceptibility at $H_{\text{dc}} = 2500\text{ Oe}$ for $\mathbf{1}_{\text{Ge}}\text{-Gd}$. The experimental data are denoted by circles; solid lines represent the best fit to the data.



ence of multiple relaxation processes different from Orbach (*i.e.*, Raman).²⁷

In the case of **1_{Ge-Sm}** and **1_{Ge-Eu}**, the $\chi_M T$ products at room temperature (0.35 and 1.39 cm³ K mol⁻¹, respectively) are higher than expected (0.09 and 0 cm³ K mol⁻¹, respectively), which implies the existence of thermally populated excited states (Fig. S16†). Upon cooling, the $\chi_M T$ values decrease until reaching 0.022 and 0.019 cm³ K mol⁻¹ at 2 K, respectively. The fitting of the magnetic susceptibility of **1_{Ge-Eu}** to the equation proposed by Kahn²⁸ yields $\lambda = 332$ cm⁻¹ and $\delta = 0.13\%$, with λ being the spin-orbit coupling parameter and δ the percentage of Eu^{II} paramagnetic impurities. The value of λ is close to that extracted from luminescence measurements ($\lambda = 314$ cm⁻¹, see below) and suggests a slightly larger separation between the ground and first excited states than in **1_{Si-Eu}** (332 cm⁻¹ for **1_{Ge-Eu}** *vs.* 321 cm⁻¹ for **1_{Si-Eu}**, Fig. S18†). The same behaviour is observed for **1_{Ge-Sm}**, in which the separation between the ground and first excited states is increased from 125 cm⁻¹ (**1_{Si-Sm}**) to 140 cm⁻¹ (**1_{Ge-Sm}**) according to the photoluminescence measurements (see below).

The room temperature $\chi_M T$ products of the remaining compounds are close to the expected values (Fig. S9, Table S8†). As the temperature decreases, the $\chi_M T$ products exhibit relative stability and, in most cases, they experience a sudden drop below ~50 K due to the depopulation of the M_J sublevels of the Ln ions. The presence of populated low-lying states prevents these compounds from achieving the expected saturation values at 2 K and 7 T (Fig. S10, Table S8†).

Regarding the ac susceptibility, **1_{Ge-Dy}** exhibits a modest frequency dependence with the maxima of the χ''_M signals below ~2.7 K (Fig. S19–S21†), while **1_{Ge-Yb}** displays clear slow relaxation of magnetisation below ~9 K (Fig. S22–S24†). Compared to the silicon-based counterparts, the maxima of χ''_M signals are shifted towards higher temperatures in both compounds, enabling the extraction of the energy barrier for **1_{Ge-Dy}** ($U_{\text{eff}} = 14.3$ cm⁻¹, $\tau_0 = 6.3 \times 10^{-9}$ s). Even though the observed improvements could be associated with an increase in the energy gap between the ground and first excited states of Dy and Yb, the large α values (0.26 (2 K)–0.40 (2.7 K) for **1_{Ge-Dy}**; 0.12 (5 K)–0.05 (9 K) for **1_{Ge-Yb}**) and the deviation of the relaxation times from linearity indicate that the magnetic relaxation does not occur exclusively *via* the Orbach mechanism. In fact, photoluminescence studies of **1_{Ge-Yb}** reveal a larger separation between the ground and first excited states (343 cm⁻¹ for **1_{Ge-Yb}** *vs.* 261 cm⁻¹ for **1_{Si-Yb}**), but the energy barriers obtained by the Arrhenius fitting (26.4 cm⁻¹ for **1_{Ge-Yb}** and 14.8 cm⁻¹ for **1_{Si-Yb}**) are notably lower. Thus, the relaxation of magnetisation in **1_{Ge-Yb}** occurs most likely through a Raman mechanism, as observed in related Yb^{III}-based compounds.²⁹

When it comes to photophysical properties, solid state photoluminescence was studied for **1_{Ge-Sm}**, **1_{Ge-Eu}** and **1_{Ge-Tb}** in the visible region and for **1_{Ge-Er}** and **1_{Ge-Yb}** in the near infrared (NIR) region. First, the excitation of **1_{Ge-Sm}** and **1_{Ge-Eu}** (Fig. S25†) was monitored around their most intense bands at 614 nm and 599 nm, respectively. The resulting spectra

display broad bands in the 300–450 nm range, which imply that the antenna effect is operative in these compounds.

Emission spectra were recorded at different temperatures (from 15 K to room temperature) after irradiation at 375 and 430 nm. Excitation wavelengths correspond to the maxima of either the UV-Vis spectrum of the H₂L ligand (Fig. S26†) or the excitation spectrum of **1_{Ge-Eu}**, respectively. In both cases, when the temperature increases, not only the intensity of the signals decreases, but the distinction between sublevels also appears less differentiated. The emission spectra of **1_{Ge-Eu}** (Fig. S27 and S28†) exhibit five bands centred at 580, 596, 614, 653 and 701 nm, which are attributed to the ⁵D₀ → ⁷F_{*J*} (*J* = 0, 1, 2, 3 and 4) manifolds. The relative intensity of the emission bands is virtually identical to that of **1_{Si-Eu}** with the strongest transition corresponding to the hypersensitive ⁵D₀ → ⁷F₂, which provides intense reddish luminescence (Fig. 3, S29 and S30†). The additional splitting of this specific transition when going from **1_{Ge-Eu}** (five signals) to **1_{Si-Eu}** (four signals) is in line with the lower symmetry of {EuO₈} centres in the former compound. Observation of the forbidden ⁵D₀ → ⁷F₀ and ⁵D₀ → ⁷F₃ transitions indicates that Eu^{III} ions occupy a low-symmetry site,³⁰ as expected from the biaugmented trigonal prismatic coordination geometry. The average λ parameter was extracted from the positions of the emission bands and it is similar to that found in **1_{Si-Eu}** (314 cm⁻¹ for **1_{Ge-Eu}** *vs.* 310 cm⁻¹ for **1_{Si-Eu}**).

In the case of **1_{Ge-Sm}** and **1_{Ge-Tb}**, spectra were collected under excitation with a continuous HeCd laser at 325 nm due to the lower intensity of the emission. Analogous behaviour to that shown by **1_{Si-Tb}** was observed for **1_{Ge-Tb}**, which exhibits a very weak metal-centred emission (Fig. S31†) due to the out-of-pocket coordination of the Tb^{III} ion. Even though the energy transfer from the ligand to the Ln is not very efficient in **1_{Ge-Tb}**, the spectrum of **1_{Ge-Sm}** (Fig. S32†) shows three emission bands that are associated with the transitions ⁴G_{5/2} → ⁶H_{*J*} (*J* = 5/2, 7/2 and 9/2) located at 563, 599 and 646 nm, respectively.³¹ The positions of the bands indicate a slightly higher λ parameter in **1_{Ge-Sm}** (293 cm⁻¹) than that in **1_{Si-Sm}** (280 cm⁻¹). The ratio of the intensities for the first (*J* = 5/2) and the most intense bands (*J* = 7/2) increases from **1_{Si-Sm}** (*ca.* 1 : 3) to **1_{Ge-Sm}** (*ca.* 1 : 5), in such a way that a clear colour shift to orange-yellowish is observed with the replacement of the heteroatom (Fig. 3, S29 and S30†). These emission bands almost vanished at room temperature, in line with the low absolute quantum

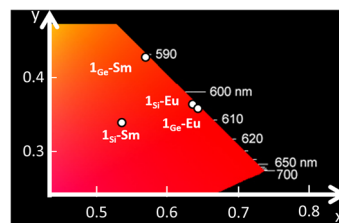


Fig. 3 CIE 1931 *x, y* chromaticity coordinates as a function of the emission wavelengths for **1_{Si-Sm}**, **1_{Ge-Sm}**, **1_{Si-Eu}** and **1_{Ge-Eu}**.



yield (<0.01%) measured for **1_{Ge-Sm}**, which is similar to that for **1_{Si-Sm}**.

The luminescence decay curve of the most intense line was measured for **1_{Ge-Eu}** upon excitation at 375 nm and 430 nm (Fig. S33†). Data were fitted to single and double exponential functions: $I = A_0 + A_1 \exp(-t/\tau_1)$ (15–150 K) and $I = A_0 + A_1 \exp(-t/\tau_1) + A_2 \exp(-t/\tau_2)$ (298 K), with τ_n being the luminescence lifetime, A_0 background and A_n weighting parameters (Table S10†). At lower temperatures, values of $\tau_1 = ca. 800 \mu s$ are achieved at both wavelengths, whereas a double exponential function was needed to fit decay curves at room temperature, leading to a significant decrease in the lifetime with $\tau_1 \sim 100 \mu s$ and $\tau_2 = 600 \mu s$, which suggests that there is a thermally activated nonradiative deactivation process of the excited states. Although the obtained lifetimes are similar to those obtained for **1_{Si-Eu}**, the emission decay curves obtained for **1_{Ge-Eu}** at low temperatures are better fitted to a single exponential function in contrast to that for **1_{Si-Eu}**, for which two decay components were required. Interestingly, the absolute quantum yields at room temperature are three times higher for **1_{Ge-Eu}** than those for **1_{Si-Eu}**, with an average value of 1.18% in contrast to the 0.36% found for the silicon-based counterpart.

Additionally, the emission in the near infrared region was studied for **1_{Ge-Er}** and **1_{Ge-Yb}** upon excitation at 325 nm (Fig. S34†). The emission spectrum of **1_{Ge-Er}** shows the most intense line at 1535 nm, which can be assigned to the $^4I_{13/2} \rightarrow ^4I_{15/2}$ transition. In this case, the intensity of the emission was very weak even at low temperatures, so the measurements were performed only at 15 K. However, the high intensity of the quadruplet arising from the $^2F_{5/2} \rightarrow ^2F_{7/2}$ transition in **1_{Ge-Yb}** and expanding between 970 and 1050 nm allows the collection of the spectrum from 15 K up to room temperature. The high intensity of the emission allowed the decay curves to be measured (Fig. S35†). The calculated values are on the order of a few μs , ranging from 5302(9) ns at 20 K to 2865(12) ns at room temperature. As in the case of **1_{Ge-Eu}**, the room temperature lifetime is lower because of the presence of more than one thermally activated non-radiative deactivation mechanism. Nevertheless, in contrast to **1_{Si-Yb}**, **1_{Ge-Yb}** emits at room temperature, constituting the first POM-based system with this behaviour and one of the few Yb^{III}-based coordination complexes in the literature.^{32–34} It is noteworthy that these systems are of high interest for bioimaging applications (human tissues are relatively transparent to NIR light),³⁵ as well as for telecommunication devices and lasers.

In summary, our study shows how the replacement of the heteroatom of a POM provokes a significant change in the photoluminescence properties of several lanthanide-based hybrid organic-inorganic compounds. The slight modifications in the bond lengths of the Ln ions lead to a clear colour shift to orange in **1_{Ge-Sm}** upon photoirradiation, whereas **1_{Ge-Yb}** is one of the few Yb-based compounds exhibiting slow relaxation of magnetisation (below ~ 9 K) and NIR emission at room temperature. Regarding the magnetic properties, the frequency dependent ac susceptibility signals of

the Gd, Dy and Yb derivatives are slightly shifted towards higher temperatures. This trend implies that the in-pocket coordination mode could be favourable for Ln ions to exhibit more efficient slow relaxation of magnetisation and luminescent emission. Thus, we plan to extend these systematic studies to different lacunary POM fragments, including mono-lacunary Keggin anions with larger heteroatoms.

Author contributions

J. B. F., E. R. B., I. O., and B. A.: formal analysis, writing – original draft and review and editing. J. B. F., C. R. E., J. A. G., M. G. and L. L.: investigation and data curation. I. O., B. A., and J. M. G. Z.: supervision, conceptualisation and funding acquisition.

Data availability

Data available on request from the authors.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 C. Boskovic, *Acc. Chem. Res.*, 2017, **50**, 2205–2214.
- 2 C. M. Granadeiro, D. Juliao, S. O. Ribeiro, L. Cunha-Silva and S. S. Balula, *Coord. Chem. Rev.*, 2023, **476**, 214914.
- 3 G. Trautwein, B. El Bakkali, J. Alcañiz-Monge, B. Artetxe, S. Reinoso and J. M. Gutiérrez-Zorrilla, *J. Catal.*, 2015, **331**, 110–117.
- 4 G.-P. Yang, S.-X. Shang, B. Yu and C.-H. Hu, *Inorg. Chem. Front.*, 2018, **5**, 2472–2477.
- 5 K. Zheng and P. Ma, *Dalton Trans.*, 2024, **53**, 3949–3958.
- 6 C. M. Granadeiro, R. A. S. Ferreira, P. C. R. Soares-Santos, L. D. Carlos, T. Trindade and H. I. S. Nogueira, *J. Mater. Chem.*, 2010, **20**, 3313–3318.



- 7 J. M. Clemente-Juan, E. Coronado and A. Gaita-Ariño, *Chem. Soc. Rev.*, 2012, **41**, 7464–7478.
- 8 Z. X. Yang, F. Gong, D. Lin and Y. Huo, *Coord. Chem. Rev.*, 2023, **492**, 215205.
- 9 B. S. Bassil, M. H. Dickman, I. Römer, B. von der Kammer and U. Kortz, *Angew. Chem., Int. Ed.*, 2007, **46**, 6192–6195.
- 10 S. Reinoso, M. Giménez-Marqués, J. R. Galán-Mascarós, P. Vitoria and J. M. Gutiérrez-Zorrilla, *Angew. Chem., Int. Ed.*, 2010, **49**, 8384–8388.
- 11 F. Hussain, F. Conrad and G. R. Patzke, *Angew. Chem., Int. Ed.*, 2009, **48**, 9088–9091.
- 12 J. J. Baldoví, S. Cardona-Serra, A. Gaita-Ariño and E. Coronado, *Adv. Inorg. Chem.*, 2017, **69**, 213–249.
- 13 Z.-X. Yang, F. Gong, D. Lin and Y. Huo, *Coord. Chem. Rev.*, 2023, **492**, 215205.
- 14 S. G. McAdams, A.-M. Ariciu, A. K. Kostopoulos, J. P. S. Walsh and F. Tuna, *Coord. Chem. Rev.*, 2017, **346**, 216–239.
- 15 N. Ishikawa, M. Sugita, T. Ishikawa, S. Koshihara and Y. Kaizu, *J. Am. Chem. Soc.*, 2003, **125**, 8694–8695.
- 16 M. A. AlDamen, J. M. Clemente-Juan, E. Coronado, C. Martí-Gastaldo and A. Gaita-Ariño, *J. Am. Chem. Soc.*, 2008, **130**, 8874–8875.
- 17 M. A. AlDamen, S. Cardona-Serra, J. M. Clemente Juan, E. Coronado, A. Gaita-Ariño, C. Martí-Gastaldo, F. Luis and O. Montero, *Inorg. Chem.*, 2009, **48**, 3467–3479.
- 18 Z. Li, Z.-H. Lv, Y.-Q. Sun, X.-X. Li and S.-T. Zheng, *CCS Chem.*, 2022, **4**, 2938–2945.
- 19 B. Artetxe, S. Reinoso, L. San Felices, J. M. Gutiérrez-Zorrilla, J. A. García, F. Haso, T. Liu and C. Vicent, *Chem. – Eur. J.*, 2015, **21**, 7736–7745.
- 20 H. Wu, B. Yan, R. Liang, V. Singh, P. Ma, J. Wang and J. Niu, *Dalton Trans.*, 2019, **49**, 388–394.
- 21 S. Sarwar, S. Sanz, J. van Leusen, G. S. Nichol, E. K. Brechin and P. Kögerler, *Dalton Trans.*, 2020, **49**, 16638–16642.
- 22 W. A. Cañón-Mancisidor, M. Zapata-Lizama, P. Hermosilla-Ibáñez, C. Cruz, D. Venegas-Yazigi and G. Mínguez Espallargas, *Chem. Commun.*, 2019, **55**, 14992–14995.
- 23 M. Zapata-Lizama, P. Hermosilla-Ibáñez, D. Venegas-Yazigi, G. Mínguez Espallargas, L. J. Queiroz Maia, G. Gasparotto, R. C. De Santana and W. A. Cañón-Mancisidor, *Inorg. Chem. Front.*, 2020, **17**, 3049–3062.
- 24 E. Ruiz-Bilbao, M. Pardo-Almanza, I. Oyarzabal, B. Artetxe, L. San Felices, J. A. García, J. M. Seco, E. Colacio, L. Lezama and J. M. Gutiérrez-Zorrilla, *Inorg. Chem.*, 2022, **61**, 2428–2443.
- 25 A. Zabala-Lekuona, J. M. Seco and E. Colacio, *Coord. Chem. Rev.*, 2021, **441**, 213984.
- 26 D. Casanova, M. Llunell, P. Alemany and S. Alvarez, *Chem. – Eur. J.*, 2005, **11**, 1479–1494.
- 27 G. Handzlik, M. Magott, M. Arczyński, A. M. Sheveleva, F. Tuna, M. Sarewicz, A. Osyczka, M. Rams, V. Vieru, L. F. Chibotaru and D. Pinkowicz, *J. Phys. Chem. Lett.*, 2020, **11**, 1508–1515.
- 28 O. Kahn, *Molecular Magnetism*, Dover Publications, New York, 2021.
- 29 I. Oyarzabal, B. Artetxe, A. Rodríguez-Diéguez, J. Á. García, J. M. Seco and E. Colacio, *Dalton Trans.*, 2016, **45**, 9712–9726.
- 30 K. Binnemans, *Coord. Chem. Rev.*, 2015, **295**, 1–45.
- 31 Y. Zheng, J. Lin and Q. Wang, *Photochem. Photobiol. Sci.*, 2012, **11**, 1567–1574.
- 32 R. Jankowski, J. J. Zakrzewski, O. Surma, S. Ohkoshi, S. Chorazy and B. Sieklucka, *Inorg. Chem. Front.*, 2019, **6**, 2423–2434.
- 33 P. F. Muldoon, G. Collet, S. V. Eliseeva, T. L. Luo, S. Petoud and N. L. Rosi, *J. Am. Chem. Soc.*, 2020, **142**, 8776–8781.
- 34 C. Liu, S. V. Eliseeva, T. L. Luo, P. F. Muldoon, S. Petoud and N. L. Rosi, *Chem. Sci.*, 2018, **9**, 8099–8102.
- 35 I. Martinić, S. V. Eliseeva and S. Petoud, *J. Lumin.*, 2017, **189**, 19–43.

