



**Highlighting research from Professor Keiji Tanaka's group
from the Center for Polymer Interface and Molecular
Adhesion Science, Kyushu University, Japan**

Effect of interfacial local conformation of polymer chains on
adhesion strength

The first direct observation of the phase-separated structure
of PS/PMMA at the solid interface was achieved by SFG
imaging. The effects of local orientational distribution at
interfaces on adhesion was also investigated.

As featured in:



See Keiji Tanaka *et al.*, *Polym. Chem.*,
2024, 15, 4425.



Cite this: *Polym. Chem.*, 2024, **15**, 4425

Effect of interfacial local conformation of polymer chains on adhesion strength†

Tatsuki Abe, ^a Satoru Yamamoto ^a and Keiji Tanaka ^{*a,b}

The aggregation states of polymer chains at solid interfaces are strongly related to their adhesion properties. In this study, we focused on sum-frequency generation (SFG) vibrational spectroscopy, which offers the best depth resolution at the sub-nanometer level among available techniques, and explored its potential as an imaging method. A poly(methyl methacrylate) (PMMA) thin film with a line-and-space pattern was prepared on a quartz substrate. Imaging of the interfacial line-and-space structure in the film was successfully achieved based on SFG signals arising from ester methyl groups as well as carbonyl groups. Once SFG imaging was established, it was applied to blend films of polystyrene (PS) and PMMA with different compositions on the quartz substrate, enabling the direct and non-destructive observation of wetting layers and phase-separated structures buried at the substrate interface for the first time. The interfacial adhesion strength between the blend films and the quartz substrate, evaluated using the surface and interfacial cutting analysis system, showed a clear correlation with the interfacial structure of the blend. This study, which enables the analysis of the relationship between the local orientation of polymer chains at the interface and adhesion strength, is expected to contribute greatly to the design of next-generation adhesives and adhesion technologies.

Received 31st August 2024,
Accepted 2nd October 2024

DOI: 10.1039/d4py00951g

rsc.li/polymers

1. Introduction

The structure and properties of polymer chains at a buried solid interface are considered crucial for the adhesion properties of polymer-based adhesives.^{1–15} To better understand the emergence of adhesion strength, various theories have been proposed, including interlocking,^{16–19} chain diffusion,^{20–24} Coulomb interactions,^{25–27} van der Waals interactions,^{28–30} and hydrogen bonding^{31–33} at adhesion interfaces. These theories suggest that polymer chains directly contact the adherend surface, leading to both physical and, in some cases, chemical interactions. Thus, it is essential to understand and control the orientation of functional groups in polymer chains at the interface, the resulting local conformations, and the way these interfacial adsorbed chains interact with bulk polymer chains to form the interfacial layer. This understanding is key to optimizing adhesion performance.

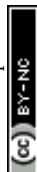
The interface is in a different energy state compared to the bulk because it is in contact with different media. Therefore, the aggregation state of molecular chains at the interface is unique compared to the bulk,^{34–44} necessitating direct evaluation. Interfacial selective spectroscopy is effective in analyzing the aggregation state of molecular chains at the interface. Specifically, techniques such as near edge X-ray absorption fine structure,^{45,46} energy loss near-edge structure,^{47,48} and soft X-ray absorption spectroscopy combined with etching using a gas cluster ion beam (GC-IB)⁴⁹ are notable. Additionally, vibrational spectroscopy is also powerful, among which sum frequency generation (SFG) vibrational spectroscopy based on the second-order nonlinear optical effect^{50–57} is used for analyzing the orientation of functional groups at the interface, due to its excellent depth resolution.

Evaluating the in-plane aggregation states of polymers at interfaces buried within a material is not straightforward. So far, imaging techniques such as secondary ion mass spectrometry^{58,59} and scanning force microscopy observation using solvent etching,^{60,61} and electron microscopy observation combined with computer tomography have been reported. However, these techniques are fundamentally destructive tests. On the other hand, if imaging using the aforementioned SFG spectroscopy,^{62,63} which has not yet been applied to synthetic polymers by other research groups,⁶⁴ were to become more commonly utilized, it would enable the non-destructive analysis of the aggregation states of polymer chains

^aCenter for Polymer Interface and Molecular Adhesion Science, Kyushu University, Fukuoka 819-0395, Japan

^bDepartment of Applied Chemistry, Kyushu University, Fukuoka 819-0395, Japan.
E-mail: k-tanaka@cstf.kyushu-u.ac.jp

†Electronic supplementary information (ESI) available: Local conformation of carbonyl groups by molecular dynamics simulation, SFG spectra in carbonyl and hydroxyl regions for PS/PMMA on SiO_x. See DOI: <https://doi.org/10.1039/d4py00951g>



2. Experimental

40 μm

quartz or SiO_x coated CaF_2

100 nm

PMMA

40 μm

quartz or SiO_x coated CaF_2

air

Then, an immiscible blend of PS and PMMA with various blend ratios at the quartz interface was characterized using SFG in conjunction with SAICAS. The M_n and PDI values of PS of PMMA, purchased from Tosoh (Tokyo, Japan) and Scientific Polymer Products Inc. (Ontario, NY, USA), respectively, were 1030k and 1100k, and 1.20 and 1.12. Films of the PS/PMMA blend were prepared by solvent-casting from toluene solutions onto CaF_2 prisms covered with a 10 nm-thick SiO_x layer. CaF_2 plates, also coated with a 10 nm-thick SiO_x layer, were then mounted on the films at 433 K for 48 h, creating a sandwich structure with the blend films. This procedure also served as the thermal annealing process for the films. For the interfacial adhesion tests using SAICAS, the blend films prepared on quartz substrates were used. The surface chemistry of quartz and SiO_x should be identical. The cutting velocity was 50 nm s^{-1} perpendicular to the interface and 500 nm s^{-1} parallel to the interface. The peeling strength was determined by the force required in the parallel direction as the blade progressed during parallel cutting.

3. Results and discussion

This journal is © The Royal Society of Chemistry 2024



Fig. 2 (a) SFG spectra in the CH region with the sp polarization combination for a thin PMMA L&S film before (top) and after (bottom) annealing at 433 K for 24 h. SFG images for OCH_{3s} and CH_{2as} signals with the sp polarization combination for the (b and c) as-spun and (d and e) annealed films. (f) A digital microscopic image of the film. The angle formed with the line pattern (ω) is shown.

p-polarized IR input) polarization combination. This setup was used to amplify signal intensity in our SFG microscope by avoiding the insertion of a polarizer between the SF output and the detector. In this configuration, the specimen was positioned on a stage such that the angle between the lines in the pattern and the incident plane of the beams (ω) was set to 0°. For clarity, each SFG spectrum has been vertically shifted. The most intense peak observed at 2955 cm⁻¹ could be attributed to the C-H symmetric stretching vibration of ester methyl groups (OCH_{3s}).⁶⁹ Peaks at 2912 and 2942 cm⁻¹ were assignable to the symmetric and antisymmetric stretching vibrations of methylene groups (CH_{2s} and CH_{2as}).⁶⁹ Although the types of SFG peaks observed were similar before and after the annealing treatment of the film, the intensity of each peak varied. Specifically, the intensity from CH_{2as} and OCH_{3s} increased after annealing. Thus, it is evident that the local conformation of PMMA chains at the quartz interface changed due to thermal annealing.

Panels (b) and (c) of Fig. 2 show two-dimensional maps of SFG signals at 2955 and 2942 cm⁻¹, respectively, for the as-spun PMMA L&S film. These correspond to the SFG images for OCH_{3s} and CH_{2as} signals, exhibiting horizontally aligned stripes with a period of approximately 40 μm. For comparison, Fig. 2(f) shows a digital microscopic image of the annealed PMMA L&S film. A similar stripe pattern aligned along the horizontal direction was observed. The width of a stripe in the SFG images is identical to that in the digital microscopic image. Additionally, when changing ω to 90°, the pattern in the SFG images rotated accordingly by 90° (not shown).

If successful imaging of the local conformation for interfacial chains is achieved, the signal intensity is expected to increase in proportion to the degree and/or the population of oriented functional groups at the interface. Panels (d) and (e) of Fig. 2 show corresponding SFG images to Fig. 2(b) and (c) but after annealing at 433 K for 24 h. The width of stripes shown in both panels was identical to that in Fig. 2(b) and (c). However, the intensity of stripe regions was enhanced after

thermal annealing, resulting in a clearer contrast between the brighter and darker regions.

The tilt angle of ester methyl groups (θ_{OCH_3}) relative to the substrate interface in the PMMA L&S films was determined from the peak intensity ratio obtained with the ssp and sps polarization combinations (not shown). Following the method by Chen *et al.*,⁷⁰ the θ_{OCH_3} values for the as-spun and annealed films were calculated to be 48 and 61°, respectively, with the assumption of no angular distribution. These values are comparable to our recent results obtained from the combination of phase-sensitive SFG spectroscopy and full-atomistic molecular dynamics (MD) simulation.⁷¹ This angle of approximately 60° is effective for the formation of hydrogen bonds between side chain carbonyl groups of PMMA, which are directly connected to ester methyl groups, and silanol groups on the quartz surface, as discussed later.

When polymer chains undergo structural relaxation at the quartz interface through thermal annealing, as shown in Fig. 2(a), the orientation of functional groups is expected to differ from that before annealing.^{72–77} Fig. 3(a) shows SFG spectra in the C=O region with the sp polarization combination for the PMMA L&S film before and after thermal annealing. A distinct peak at 1740 cm⁻¹, attributed to the stretching vibration of free carbonyl groups, was observed for the as-spun film. However, after annealing, the peak shifted to the lower wavenumber position of 1728 cm⁻¹. The red-shift of the peak after annealing, indicative of structural relaxation, can be explained by the formation of hydrogen bonding. This means that carbonyl groups bonded to SiOH groups at the quartz surface after annealing.^{69,78} Furthermore, the peak intensity was enhanced after thermal annealing, implying not only the enhanced orientation of carbonyl groups but also an increase in the number density of hydrogen-bonded carbonyl groups.⁷⁹ To address the transient change in the local conformation of PMMA at the quartz interface during thermal annealing, MD simulations were conducted. The population of carbonyl groups orienting towards the quartz surface, associ-





Fig. 3 (a) SFG spectra in the C=O region with the sp polarization combination for the thin PMMA L&S film before (top) and after (bottom) annealing at 433 K for 24 h. (b–e) SFG images for C=O signals with the sp polarization combination obtained at IR wavenumber of (b and d) 1740 cm^{-1} and (c and e) 1728 cm^{-1} for the as-spun and annealed films, respectively.

ated with the formation of hydrogen bonds, increased with annealing, as shown in Fig. S1.† This MD result is in excellent accordance with SFG spectra shown in Fig. 3(a).

SFG imaging was then applied to this vibration region. Panels (b) and (c) of Fig. 3 show two-dimensional maps of SFG intensity at 1740 and 1728 cm^{-1} , respectively, corresponding to the stretching of free and hydrogen-bonded carbonyl groups in PMMA L&S before thermal annealing. Similar stripe patterns, horizontally aligned with the identical width observed in Fig. 2, were noted, indicating that the signals in the SFG images can be attributed to PMMA. Panels (d) and (e) of Fig. 3 show SFG imaging at 1740 and 1728 cm^{-1} for the PMMA L&S after annealing. The contrast in the images became clearer at 1728 cm^{-1} than at 1740 cm^{-1} , which is opposite to the trend observed before annealing. The integrated intensity of the SFG signals from hydrogen-bonded carbonyl groups was larger in Fig. 3(e) (*i.e.* 4.4×10^7) than in Fig. 3(c) (*i.e.* 2.1×10^7). This result is consistent with the SFG spectra shown in Fig. 3(a). Thus, it can be claimed that the formation of hydrogen bonds between carbonyl groups of the PMMA side chain and the substrate surface has been successfully visualized.

We now turn to an examination of how the formation of hydrogen bonds of carbonyl groups to the quartz surface impacts the adhesion strength. The angle between the cutting direction and the L&S structure corresponding to ω was set to 90°, meaning that the cutting blade penetrated perpendicular to the L&S structure. Fig. 4(a) shows a schematic representation of the cutting geometry for the SAICAS measurement. In the early stage of cutting, a diamond blade proceeds through the film diagonally (regime I). Once it reaches the substrate surface, the direction changes to horizontal against the substrate surface (regime II). The latter cutting mode corresponds to peeling, providing information on the adhesion strength between PMMA and quartz. Fig. 4(b) shows the horizontal force (F_h) profile for the PMMA L&S film before and after



Fig. 4 (a) Schematic representation of SAICAS measurement, showing cutting modes and detecting forces along the directions vertical and horizontal to the interface. (b) Horizontal force (F_h) profile for the PMMA L & S film before and after annealing, measured in the direction perpendicular to the line pattern.

annealing. After thermal annealing, the F_h increased and the periodic fluctuation corresponding to the L&S structure was observed. Increased and decreased F_h regions correspond to interfacial peeling and the movement of the cutting blade on the substrate, respectively. Considering SFG results, the enhancement of interfacial adhesion strength can be attributed to the formation of hydrogen bonding by carbonyl groups, thereby changing the local conformation of PMMA at the adherend surface. Polymer chains in the as-spun film are frozen in an energetically unstable state, whereas side-chain carbonyl groups adopt an energetically preferable conformation after annealing. This conformation enables them to form a stronger bond with the adherend surface.

Once we established the mapping of functional groups at a buried interface using SFG with a sub-nanometer depth resolution, the technique was applied to examine the aggregation states of a polymer blend of PS and PMMA, which is in a phase-separated state in the bulk.^{60,61} On the other hand, it is widely accepted that a wetting layer, or at least an enriched layer, of a component might form at the free surface and the substrate interface to minimize the interfacial free energy, as is well established in block copolymers.^{80,81} To the best of our knowledge, it remains uncertain whether this blend system is in a phase-separated state even at the buried interface.

Panels (a) and (b) of Fig. 5 show SFG spectra for PS/PMMA blend films with different bulk compositions using the sp and pp polarization combinations. Since the film surface was covered with a SiO_x -covered CaF_2 plate, the spectra reflect the structural information at the substrate surface. The most prominent peak for the neat PMMA film was observed at 2955 cm^{-1} , assignable to OCH_{3s} .⁶⁹ This peak was also observed for the PS/PMMA (25/75), (50/50), and (75/25) films, but not for the neat PS film. These results make it clear that ester methyl groups of PMMA were oriented at the SiO_x interface due to the interaction through hydrogen bonding with silanol groups at the SiO_x surface. As shown in Fig. S2,† both peaks attributed to C=O and OH stretching vibrations shifted to lower wavenumber regions with increasing bulk PMMA ratio. This shift indicates the formation of hydrogen bonds between PMMA side chains and the substrate surface.⁸² In contrast, peaks in the range of 3000 to 3100 cm^{-1} , assignable to stretching





Fig. 5 SFG spectra for PS/PMMA blend films with different bulk compositions with (a) sp and (b) pp polarization combinations. SFG images for PS/PMMA (c) (25/75), (d) (50/50), and (e) (75/25), respectively. All blend films were annealed at 433 K for 48 h under vacuum.

vibrations of phenyl groups,^{83,84} were observed for the PS neat film. These peaks were not observed for the PMMA film or the PS/PMMA (25/75), (50/50), and (75/25) films with the sp polarization combination, but were present for the PS/PMMA (75/25) film with the pp polarization combination. Thus, it can be claimed that the substrate interface of the PS/PMMA (25/75) and (50/50) films was covered with the wetting layer of PMMA.^{85–87} In the case of the PS/PMMA (75/25) film, although both components of PS and PMMA were present at the substrate interface, it is difficult to determine whether they are in a phase-separated or a homogeneous state. It is interesting to note that for the PS/PMMA (75/25) film, SFG signals attributed to phenyl groups were observed not with the sp polarization combination, but with the pp polarization combination. This result likely indicates that phenyl groups were predominantly tilted significantly toward the substrate interface in this blend film. This interpretation is also consistent with previous reports.^{83,84}

To discuss the in-plane distribution of local conformation, SFG imaging was conducted. Panels (c–e) of Fig. 5 show two-dimensional mappings of SFG intensity at 2955 cm^{−1}, corresponding to OCH_{3s}, for the PS/PMMA (c) (25/75), (d) (50/50), and (e) (75/25) films at the SiO_x interface. In the case of the PS/PMMA (50/50) and (25/75) films, homogeneous images were obtained. Thus, it is clear that PMMA was preferentially adsorbed onto the substrate surface, forming the wetting layer of PMMA formed at the SiO_x surface, as discussed above. Since the degrees of polymerization for both the components were comparable, the conformational entropy loss of the components at the SiO_x interface cannot be considered an important factor in this structural formation.⁸⁸ Thus, the segregation of PMMA at the SiO_x interface was likely driven by the surface energy difference between the components.

In the case of the PS/PMMA (75/25) film, a heterogeneous structure with domains approximately 20–40 μm in diameter appeared, as shown in Fig. 5(e). Considering that SFG signals in the black regions were finite, not zero, the white and black

regions correspond to areas where ester methyl groups are more or less oriented, respectively. This indicates the presence of two phases with different aggregation states at the SiO_x interface. Thus, it seems most likely that the PS/PMMA (75/25) blend is separated into PS-rich and PMMA-rich phases at the SiO_x interface. It is noteworthy that the areal fraction of the white to black regions was 64% and was much larger than the bulk feed ratio of PMMA. This is simply because PMMA was preferentially segregated at the SiO_x interface. To the best of our knowledge, this is the first report of a non-destructive structural analysis of a polymer blend at a buried interface with a solid substrate, made possible by the combination of SFG spectroscopy and microscopy.

We finally address the relationship between aggregation states and adhesive properties at the substrate interface. Fig. 6 shows the dependence of F_h , corresponding to peeling force at the SiO_x interface, on the bulk blend composition, obtained through SAICAS measurements. The abscissa represents the bulk PMMA ratio. The F_h value increased with increasing bulk PMMA ratio because the extent and efficacy of hydrogen bonding

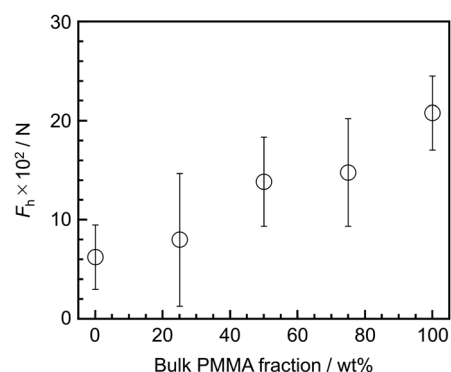


Fig. 6 Bulk PMMA composition dependence of F_h for PS/PMMA blend films at the SiO_x interface.



between PMMA side chains and the SiO_x surface increased with increasing bulk PMMA ratio, as shown in Fig. S2.† For example, for the PS/PMMA (25/75) and (50/50) blends, the SFG spectra and images appear similar, resulting in comparable peeling strength. Conversely, in the case of the PS/PMMA (75/25) blend film, although the interface was mainly covered with PMMA, the interfacial aggregation states were quite different from the others due to the phase-separated structure, specifically the presence of PS. This led to a lower peeling strength. As expected, the PS film exhibited the lowest peeling strength among samples employed owing to the lack of the formation of hydrogen bonding with the SiO_x surface. Therefore, it can be concluded that interaction such as hydrogen bonding between polymer segments and the substrate surface are crucial factors in controlling adhesive properties.

4. Conclusions

We examined the effect of the local orientational distribution of functional groups at the solid interface on adhesion properties by applying SFG imaging in conjunction with SAICAS measurements. Initially, orientational imaging based on SFG signals from ester methyl and carbonyl groups was conducted for PMMA L&S films. The structural reorganization of PMMA chains, specifically ester methyl groups and carbonyl groups, induced by thermal annealing was visualized at the quartz interface. Then, the interfacial distribution of local conformation in PS/PMMA blends with various compositions was examined. In the case of the PS/PMMA (50/50) and (25/75) films, SFG images from ester methyl groups were obtained, indicating that PMMA was preferentially segregated at the quartz surface, namely forming a wetting layer of PMMA at the interface. In contrast, the PS/PMMA (75/25) film exhibited a heterogeneous SFG image from ester methyl groups, with SFG signals from aromatic CH vibrations also clearly observed. These results make it clear that a phase-separated structure was formed at the quartz interface. Furthermore, SFG spectroscopy revealed that hydrogen bonding between side chain carbonyl groups of PMMA and hydroxy groups on quartz was promoted with increasing PMMA fraction. The interfacial adhesion strength concurrently increased with PMMA fraction due to effective hydrogen bonding at the interface. Thus, it can be concluded that the interfacial aggregation states of polymer chains, specifically hydrogen bonding in this case, are crucial to control their adhesion properties. We believe that by further studying the molecular chain interactions at adhesive interfaces, particularly the interactions between polymers and adherend surfaces at the functional group level, next-generation adhesives and adhesion technologies will be established.

Author contributions

Tatsuki Abe: data curation, formal analysis, methodology, writing – original draft. Satoru Yamamoto: data curation,

formal analysis. Keiji Tanaka: project administration, conceptualization, funding acquisition, methodology, writing – review & editing, supervision. All authors discussed the results and contributed to the final manuscript.

Data availability

The data supporting this article will be available upon request at <https://www.cstf.kyushu-u.ac.jp/~tanaka-lab/english/>.

Conflicts of interest

The authors have no conflicts to disclose.

Acknowledgements

This work was supported by JSPS KAKENHI for Scientific Research (B) (No. JP23H02014) to KT and for Early-Career Scientists (No. JP23K13794) to TA from the Ministry of Education, Culture, Sports, Science, and Technology, Japan. We are also thankful for the support from the JST-Mirai Program (JPMJMI18A2) (KT).

References

- 1 S. Wu, *Polymer Interface and Adhesion*, CRC Press, New York, 1982.
- 2 R. P. Wool, *Polymer Interfaces: Structure and Strength*, Hanser Publications, New York, 1994.
- 3 D. L. Allara, F. M. Fowkes, J. Noolandi, G. W. Rubloff and M. V. Tirrell, *Mater. Sci. Eng.*, 1986, **83**, 213–226.
- 4 H. R. Brown, *Annu. Rev. Mater. Sci.*, 1991, **21**, 463–489.
- 5 M. A. C. Stuart, W. T. S. Huck, J. Genzer, M. Müller, C. Ober, M. Stamm, G. B. Sukhorukov, I. Szleifer, V. V. Tsukruk, M. Urban, F. Winnik, S. Zauscher, I. Luzinov and S. Minko, *Nat. Mater.*, 2010, **9**, 101–113.
- 6 A. J. Crosby, K. R. Shull, H. Lakrout and C. Creton, *J. Appl. Phys.*, 2000, **88**, 2956–2966.
- 7 H.-J. Kim, K.-J. Lee and Y. Seo, *Macromolecules*, 2002, **35**, 1267–1275.
- 8 H. Lee, S. M. Dellatore, W. M. Miller and P. B. Messersmith, *Science*, 2007, **318**, 426–430.
- 9 L. Ge, S. Sethi, L. Ci, P. M. Ajayan and A. Dhinojwala, *Proc. Natl. Acad. Sci. U. S. A.*, 2007, **104**, 10792–10795.
- 10 E. P. Chan, E. J. Smith, R. C. Hayward and A. J. Crosby, *Adv. Mater.*, 2008, **20**, 711–716.
- 11 L. Imbernon, E. K. Oikonomou, S. Norvez and L. Leibler, *Polym. Chem.*, 2015, **6**, 4271–4278.
- 12 M. Aoki, A. Shundo, K. Okamoto, T. Ganbe and K. Tanaka, *Polym. J.*, 2019, **51**, 359–363.
- 13 G. Raos and B. Zappone, *Macromolecules*, 2021, **54**, 10617–10644.



- 14 S. Yamamoto, R. Kuwahara and K. Tanaka, *ACS Appl. Polym. Mater.*, 2022, **4**, 6038–6046.
- 15 M. Aizawa, H. Akiyama, Y. Matsuzawa and A. Shishido, *Polym. J.*, 2024, **56**, 401–408.
- 16 J. W. McBain and D. G. Hopkins, *J. Phys. Chem.*, 1925, **29**, 188–204.
- 17 T. Matsumoto, Y. Shimizu and T. Nishino, *Macromolecules*, 2021, **54**, 7226–7233.
- 18 K. Hagita, T. Murashima, T. Miyata and H. Jinnai, *Macromolecules*, 2024, **57**, 3862–3872.
- 19 K. Aragishi, Y. Takeda, Y. Suzuki and A. Matsumoto, *Polym. J.*, 2024, **56**, 529–540.
- 20 K. Jud, H. H. Kausch and J. G. Williams, *J. Mater. Sci.*, 1981, **16**, 204–210.
- 21 R. Schnell, M. Stamm and C. Creton, *Macromolecules*, 1998, **31**, 2284–2292.
- 22 Y. M. Boiko and R. E. Prud'Homme, *J. Appl. Polym. Sci.*, 1999, **74**, 825–830.
- 23 K. Akabori, D. Baba, K. Koguchi, K. Tanaka and T. Nagamura, *J. Polym. Sci., Part B: Polym. Phys.*, 2006, **44**, 3598–3604.
- 24 H. Kim, D. Kawaguchi, K. Tanaka and Y. Seo, *Langmuir*, 2018, **34**, 11027–11033.
- 25 B. V. Derjaguin and V. P. Smilga, *J. Appl. Phys.*, 1967, **38**, 4609–4616.
- 26 M. Erdmann, R. David, A. Fornof and H. E. Gaub, *Nat. Nanotechnol.*, 2010, **5**, 154–159.
- 27 S. Saeki, D. Kawaguchi, Y. Tsuji, S. Yamamoto, K. Yoshizawa and K. Tanaka, *Langmuir*, 2024, **40**, 9725–9731.
- 28 G. Fourche, *Polym. Eng. Sci.*, 1995, **35**, 957–967.
- 29 C. Zhang, Y. Fujii and K. Tanaka, *ACS Macro Lett.*, 2012, **1**, 1317–1320.
- 30 Y. Tsuji, Y. Kitamura, M. Someya, T. Takano, M. Yaginuma, K. Nakanishi and K. Yoshizawa, *ACS Omega*, 2019, **4**, 4491–4504.
- 31 J. L. Keddie, R. A. L. Jones and R. A. Cory, *Faraday Discuss.*, 1994, **98**, 219–230.
- 32 T. Semoto, Y. Tsuji and K. Yoshizawa, *J. Phys. Chem. C*, 2011, **115**, 11701–11708.
- 33 B. Cheng, J. Yu, T. Arisawa, K. Hayashi, J. J. Richardson, Y. Shibuta and H. Ejima, *Nat. Commun.*, 2022, **13**, 1892.
- 34 J. A. Forrest, K. Dalnoki-Veress and J. R. Dutcher, *Phys. Rev. E: Stat. Phys., Plasmas, Fluids, Relat. Interdiscip. Top.*, 1997, **56**, 5705–5716.
- 35 C. J. Ellison and J. M. Torkelson, *Nat. Mater.*, 2003, **2**, 695–700.
- 36 K. Akabori, K. Tanaka, T. Nagamura, A. Takahara and T. Kajiyama, *Macromolecules*, 2005, **38**, 9735–9741.
- 37 P. Gin, N. Jiang, C. Liang, T. Taniguchi, B. Akgun, S. K. Satija, M. K. Endoh and T. Koga, *Phys. Rev. Lett.*, 2009, **109**, 265501.
- 38 Y. Fujii, Z. Yang, J. Leach, H. Atarashi, K. Tanaka and O. K. C. Tsui, *Macromolecules*, 2009, **42**, 7418–7422.
- 39 S. Sun, H. Xu, J. Han, Y. Zhu, B. Zuo, X. Wang and W. Zhang, *Soft Matter*, 2016, **12**, 8348–8358.
- 40 N. G. Perez-de-Eulate, M. Sferrazza, D. Cangialosi and S. Napolitano, *ACS Macro Lett.*, 2017, **6**, 354–358.
- 41 D. Christie, R. A. Register and R. D. Priestley, *ACS Cent. Sci.*, 2018, **4**, 504–511.
- 42 W. Ren, Y. Hong, H. Wei, J. Xu, C. Zhang, X. Zhou and X. Wang, *Macromolecules*, 2023, **56**, 1410–1418.
- 43 J. H. Merrill, R. Li and C. B. Roth, *ACS Macro Lett.*, 2023, **12**(1), 1–7.
- 44 Y. Yang, H. Shen, S. Ge, Z. Yao and B. Zuo, *Adv. Mater. Interfaces*, 2024, 2400239, DOI: [10.1002/admi.202400239](https://doi.org/10.1002/admi.202400239).
- 45 L. R. Pattison, A. Hexemer, E. J. Kramer, S. Krishnan, P. M. Petroff and D. A. Fischer, *Macromolecules*, 2006, **39**, 2225–2231.
- 46 P. K.-H. Ho, L.-L. Chua, M. Dipankar, X. Gao, D. Qi, A. T.-S. Wee, J.-F. Chang and R. H. Friend, *Adv. Mater.*, 2007, **19**, 215–221.
- 47 Y. Kamikawa, K. Amezawa and K. Terada, *J. Phys. Chem. C*, 2021, **125**, 3890–3900.
- 48 T. Miyata, Y. K. Sato, Y. Kawagoe, K. Shirasu, H.-F. Wang, A. Kumagai, S. Kinoshita, M. Mizukami, K. Yoshida, H.-H. Huang, T. Okabe, K. Hagita, T. Mizoguchi and H. Jinnai, *Nat. Commun.*, 2024, **15**, 1898.
- 49 H. Yamane, M. Oura, D. Kawaguchi, K. Nitta, O. Sekizawa, T. Ishikawa, S. Yamamoto, K. Tanaka and T. Hatsui, *Polym. J.*, 2024, **56**, 215–220.
- 50 Y. R. Shen, *Nature*, 1989, **337**, 519–525.
- 51 Z. Chen, Y. R. Shen and G. A. Somorjai, *Annu. Rev. Phys. Chem.*, 2002, **53**, 437–465.
- 52 Y. Oda, A. Horinouchi, D. Kawaguchi, H. Matsuno, S. Kanaoka, S. Aoshima and K. Tanaka, *Langmuir*, 2014, **30**, 1215–1219.
- 53 H. Zhu, K. C. Jha, R. S. Bhatta, M. Tsige and A. Dhinojwala, *Langmuir*, 2014, **30**, 11609–11618.
- 54 N. Dhopatkhar, E. Anim-Danso, C. Peng, S. Singla, X. Liu, A. Joy and A. Dhinojwala, *Macromolecules*, 2018, **51**, 5114–5120.
- 55 L. Kong, M. Zhang, Y. Zhang, W. Zhang, X. Zhou, L. Zhang and X. Wang, *Polym. Chem.*, 2020, **11**, 7429–7437.
- 56 B. Li, J. S. Andre, X. Chen, B. Walther, R. Paradkar, C. Feng, C. Tucker, C. Mohler and Z. Chen, *Anal. Chem.*, 2020, **92**, 14145–14152.
- 57 Z. Chen, *Langmuir*, 2022, **38**, 4483–4489.
- 58 J. P. Thomas, L. Zhao, M. Abd-Ellah, N. F. Heinig and K. T. Leung, *Anal. Chem.*, 2013, **85**, 6840–6845.
- 59 J. Bailey, R. Havelund, A. G. Shard, I. S. Gilmore, M. R. Alexander, J. S. Sharp and D. J. Scurr, *ACS Appl. Mater. Interfaces*, 2015, **7**, 2654–2659.
- 60 K. Tanaka, A. Takahara and T. Kajiyama, *Macromolecules*, 1996, **29**, 3232–3239.
- 61 A. Karim, T. M. Slawewski, S. K. Kumar, J. F. Douglas, S. K. Satija, C. C. Han, T. P. Russell, Y. Liu, R. Overney, J. Sokolov and M. H. Rafailovich, *Macromolecules*, 1998, **31**, 857–862.
- 62 M. Flörsheimer, C. Brillert and H. Fuchs, *Langmuir*, 1999, **15**, 5437–5439.
- 63 S. A. Shah and S. Baldelli, *Acc. Chem. Res.*, 2020, **53**, 1139–1150.



- 64 T. Abe, H. Shimada, T. Hoshino, D. Kawaguchi and K. Tanaka, *Polym. J.*, 2022, **54**, 679–685.
- 65 N. Nagai, T. Imai, K. Terada, H. Seki, H. Okumura, H. Fujino, T. Yamamoto, I. Nishiyama and A. Hatta, *Surf. Interface Anal.*, 2002, **34**, 545–551.
- 66 I. Narayama, D. Baba, A. Takahara and K. Tanaka, *Appl. Surf. Sci.*, 2010, **257**, 1145–1148.
- 67 F. Saito, I. Nishiyama and T. Hyodo, *Surf. Coat. Technol.*, 2010, **205**, 419–422.
- 68 B. Son, M.-H. Ryou, J. Choi, T. Lee, H. K. Yu, J. H. Kim and Y. M. Lee, *ACS Appl. Mater. Interfaces*, 2014, **6**, 526–531.
- 69 Y. Tateishi, N. Kai, H. Noguchi, K. Uosaki, T. Nagamura and K. Tanaka, *Polym. Chem.*, 2010, **1**, 303–311.
- 70 J. Wang, C. Chen, S. M. Buck and Z. Chen, *J. Phys. Chem. B*, 2001, **105**, 12118–12125.
- 71 D. Kawaguchi, K. Sasahara, M. Inutsuka, T. Abe, S. Yamamoto and K. Tanaka, *J. Chem. Phys.*, 2023, **159**, 244902.
- 72 S. Sugimoto, M. Inutsuka, D. Kawaguchi and K. Tanaka, *ACS Macro Lett.*, 2018, **7**, 85–89.
- 73 B. Zuo, M. Inutsuka, D. Kawaguchi, X. Wang and K. Tanaka, *Macromolecules*, 2018, **51**, 2180–2186.
- 74 Y. Hong, Y. Li, F. Wang, B. Zuo, X. Wang, L. Zhang, D. Kawaguchi and K. Tanaka, *Macromolecules*, 2018, **51**, 5620–5627.
- 75 H. K. Nguyen, S. Sugimoto, A. Konomi, M. Inutsuka, D. Kawaguchi and K. Tanaka, *ACS Macro Lett.*, 2019, **8**, 1006–1011.
- 76 C. Cai, M. S. Azam and D. Hore, *J. Phys. Chem. C*, 2021, **125**, 22214–22222.
- 77 K. Uchida, K. Mita, S. Yamamoto and K. Tanaka, *Polym. J.*, 2023, **55**, 683–690.
- 78 M. Inutsuka, A. Horinouchi and K. Tanaka, *ACS Macro Lett.*, 2015, **4**, 1174–1178.
- 79 D. Rossi, Y. Dong, R. Paradkar, X. Chen, Y. Wu, C. Mohler, T.-C. Kuo and Z. Chen, *Langmuir*, 2024, **40**, 12689–12696.
- 80 A. M. Mayes, T. P. Russell, P. Bassereau, S. M. Baker and G. S. Smith, *Macromolecules*, 1994, **27**, 749–755.
- 81 K. Tanaka, A. Takahara and T. Kajiyama, *Acta Polym.*, 1995, **46**, 476–482.
- 82 S. Ye, S. Morita, G. Li, H. Noda, M. Tanaka, K. Uosaki and M. Osawa, *Macromolecules*, 2003, **36**, 5694–5703.
- 83 K. S. Gautam, A. D. Schwab, A. Dhinojwala, D. Zhang, S. M. Dougal and M. S. Yeganeh, *Phys. Rev. Lett.*, 2000, **85**, 3854–3857.
- 84 B. Li, L. Chen, S. Zhang, Q. Tao, Y.-H. Ma, P. Hu, X. Lu, K. C. Chou and Z. Chen, *Appl. Surf. Sci.*, 2023, **611**, 155715.
- 85 H. Wang and R. J. Composto, *J. Chem. Phys.*, 2000, **113**, 10386–10397.
- 86 D. U. Ahn, Z. Wang, I. P. Campbell, M. P. Stoykovich and Y. Ding, *Polymer*, 2012, **53**, 4187–4194.
- 87 Z. Zhang, Z. Wang and Y. Ding, *Polymer*, 2014, **55**, 4150–4155.
- 88 T. Kajiyama, K. Tanaka and A. Takahara, *Macromolecules*, 1998, **31**, 3746–3749.

