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Nanozymes: a promising solution for dental antibacterial applications

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Dental diseases pose significant public health challenges globally, affecting millions with conditions exacerbated by microbial-induced inflammation. Traditional natural enzymes, despite their antibacterial and anti-inflammatory capabilities, are limited by operational stability and environmental sensitivity. This review explores the revolutionary realm of nanozyme-artificial enzymes made from nanomaterials—which offer enhanced stability, cost-effectiveness, and ease of modification. We discuss the advent of nanozymes since their first recognition in 2007, emphasizing their enzyme-mimicking capabilities and applications in dental medicine, particularly for dental caries, pulpitis, periodontitis and peri-implantitis. This paper presents a comprehensive analysis of nanozymes' classification, mechanisms, and emerging applications, shedding light on their potential to revolutionize dental antibacterial treatments and addressing current challenges and future perspectives in their development.

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1. Introduction

Dental diseases are among the most common diseases globally, posing significant health and economic challenges and substantially diminishing the quality of life for those impacted.¹ Microbial-induced inflammation is a common feature of dental diseases like dental caries, endodontic diseases, periodontitis, and peri-implantitis.² Mechanical decontamination is a commonly used clinical method that effectively removes plaque biofilms. Nevertheless, it has limitations in cleaning inaccessible areas and can damage tooth and implant surfaces, promoting bacterial aggregation and resulting in bleeding and injury to the alveolar bone and gums.³ Nonmechanical approaches, such as antiseptics like chlorhexidine and hydrogen peroxide, can effectively penetrate the microstructures of implant fixtures without damaging surfaces. However, chlorhexidine is less effective against plaque biofilms and is unsuitable for daily use due to adverse effects like tartar formation and tooth staining, while high-concentration hydrogen peroxide can harm normal tissues.^{4,5} The supplemental use of local antibiotics offers an additional treatment option for oral infectious diseases. Yet, eliminating bacterial biofilms *in vivo* at minimal antibiotic concentrations is challenging. Additionally, high doses of antibiotics can increase

biofilm tolerance and promote bacterial drug resistance.⁶ Promising research and applications of natural enzymes with antibacterial properties have been explored for treating these conditions.⁷ However, natural enzymes often suffer from intrinsic limitations, including low operational stability, sensitivity to temperature and pH variations, and difficulties in recycling.⁸ To overcome these deficiencies, researchers have explored enzyme mimics.

Nanozymes, as artificial enzymes, are a new type of functional nanomaterial with enzyme mimic activities.⁹ Compared to natural enzymes, nanozymes are easier to prepare, have adjustable catalytic activity, exhibit high stability, are cost-effective, and are more manageable.¹⁰ These attributes position nanozymes as viable substitutes for natural enzymes across industrial, biological, and medical fields.¹¹ Since the discovery of horseradish peroxidase(POD)-like activity in ferromagnetic nanoparticles in 2007,¹² numerous nanozymes have been synthesized and utilized in various fields.^{13,14} In 2013, Wei and colleagues described nanozymes as nanomaterials exhibiting enzyme-like activities.¹⁵ Currently, nanozymes are acknowledged for their ability to catalyze substrates into products under conditions similar to those of traditional artificial enzymes.¹⁶ The unique magnetic, fluorescent, and electrical properties of nanozymes further enhance their potential as substitutes for natural enzymes, making them a focal point of recent research and development.¹⁵

Over the past decade, nanozymes have driven significant technological advancements in oral medicine due to their exceptional physicochemical properties and intrinsic enzyme-like activities.¹⁷ They offer promising solutions for treating conditions such as caries, periodontitis, and peri-implantitis.

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Despite advancements, comprehensive reviews on nanozyme applications in dental antibacterial research are limited. Previous reviews have made substantial contributions but still exhibit some limitations. For instance, Chen *et al.* conducted an extensive review on the application of nanozymes in oral healthcare, discussing various aspects including antibacterial, anti-inflammatory, and tissue regeneration properties.¹⁷ Notably, their classification approach concentrated on diseases related to antibacterial applications; however, their work did not specifically address the use of nanozymes in periodontal disease treatment. Cai *et al.* focused their review on the use of nanozymes for treating oral infections, highlighting their antibacterial and anti-inflammatory properties.¹⁸ However, their review lacked a comprehensive overview of antibacterial applications, as it covered only a limited number of cases. This selective approach may be less informative for newcomers to the field who are looking for a more complete understanding. As nanozymes were recognized as one of the top ten chemical advancements of 2022,¹⁹ there has been rapid development and an influx of new research, especially in the field of dental antibacterial applications. It is worth noting that the reviews by Chen and Cai were both published before 2023, and no comprehensive review has yet covered the significant advancements made in this field over the past two years. To address these gaps and provide an up-to-date perspective, a thorough overview and analysis of recent research on nanozymes for dental antibacterial applications is necessary. Our work makes several key contributions compared to existing literature: first, we provide an extensive analysis of the use of nanozymes in dental antibacterial treatment, specifically focusing on applications in dental caries, pulpitis, periodontitis, and peri-implantitis, thereby offering a holistic view of their potential in managing diverse dental conditions. Furthermore, we incorporate the most recent advancements, including publications from the past two years, such as the emerging use of single-atom nanozymes for implant-related biofilm infections—an area that previous reviews have not adequately covered. In contrast to earlier reviews that often discussed antibacterial, anti-inflammatory, and tissue regeneration aspects together, our work focuses exclusively on the direct antibacterial action of nanozymes. This focused approach allows for a detailed and in-depth exploration of their antibacterial efficacy, without delving into broader combined applications or anti-inflammatory contexts.

This review elucidates the classification, catalytic mechanisms, and significant progress of nanozymes in dental medicine. It begins by categorizing nanozymes based on metal elements and enzymatic reaction mechanisms, and then explains their antibacterial mechanisms. Importantly, the review details the applications of nanozymes in dental antibacterial fields (Table 1), providing an in-depth overview of their potential in treating dental caries, pulpitis, periodontitis, and peri-implantitis. The paper concludes by outlining the challenges associated with nanozymes in dental medicine and suggesting directions for future research. This review aims to inspire new insights and technologies, providing readers with a comprehensive understanding and outlook for developing

more effective antibacterial nanozymes, thus advancing dental antibacterial research.

2. Classification of nanozymes

Nanozymes, a novel type of nanomaterial, exhibit inherent enzyme-like properties, making them particularly valuable in antibacterial applications. These materials are renowned for their catalytic activities, stability, and versatility. Depending on their composition, nanozymes can be categorized into four primary types: metal-based, metal oxide-based, carbon-based, and other varieties (Fig. 1). Metal-based nanozymes include materials such as Au,³⁹ Ag,⁴⁸ and Pt,⁴⁹ as well as their composites. They are known for their POD-like activities, which help disrupt bacterial biofilms and inhibit the growth of oral pathogens. Metal oxide-based nanozymes, including Fe₃O₄,⁵⁰ CeO₂,⁴⁵ MnO₂,⁴¹ and Co₃O₄,⁵¹ are composed of transition metal oxides and utilize Fenton reactions and charge transfer mechanisms to demonstrate enzyme-like activities. Carbon-based nanozymes, such as carbon dots, carbon nanotubes, graphdiyne, and MXenes, are favoured over metal-based nanozymes for their enhanced biocompatibility.⁵² However, there are currently no reports of their application in the field of oral antibacterial use. Other common types of nanozymes include metal-organic frameworks,⁴⁵ metal sulfides,³⁸ and Prussian blue⁵³ nanozymes, which also mimic natural enzyme functions.

Nanozymes are also classified based on the types of catalytic reactions they facilitate (Fig. 1). Nanozymes with POD activity, such as those based on Fe₃O₄, catalyze the breakdown of hydrogen peroxide (H₂O₂) into highly reactive hydroxyl radicals ([•]OH).⁵² Oxidase (OXD) nanozymes, such as those based on Co, Ru, Au, and CeO₂, efficiently catalyze substrate oxidation without requiring H₂O₂, even at low substrate concentrations.⁵⁴ Catalase (CAT) nanozymes break down H₂O₂ into water and oxygen, reducing reactive oxygen species (ROS) accumulation and protecting cells from oxidative stress.⁵⁵ Superoxide dismutase (SOD) nanozymes, including those based on CeO₂, MnO₂, Pt, and Prussian blue, effectively scavenge excess ROS, contributing to antioxidant defense. This activity can synergize with antibacterial effects, offering anti-inflammatory benefits for oral health.⁵⁶ Additionally, nanozymes with hydrolase activity, which degrade extracellular DNA (eDNA), also contribute to antibacterial effects.⁵⁷

3. Antibacterial mechanisms of nanozymes

The study of nanozymes has attracted significant attention from researchers. Understanding nanozyme-mediated antimicrobial mechanisms is crucial for improving infectious disease control and developing biomedical technologies. The antibacterial mechanisms of nanozymes remain underexplored due to the diversity of nanozyme types, their physical and chemical properties, and various interfering factors. Current findings categorize the antibacterial effects of nanozymes into two primary types: the generation of ROS and non-ROS mechanisms.



Table 1 Summary of nanozymes in dental antibacterial applications

Application	Nanozyme formulations	Enzyme-like activity	Functionality	Ref.
Caries	Fe ₃ O ₄	POD	Degrade exopolysaccharides and kill bacteria disrupt intractable oral biofilms and prevent tooth decay	20
	Ferumoxytol	POD		21
	Ferumoxytol	POD	Disrupt cell membrane and degrade EPS matrix	22
	Ferumoxytol/SnF ₂	POD	Inhibit biofilm accumulation and decrease enamel damage	23
	Graphdiyne/l-cys/Ag	POD and CAT	Remove plaque biofilm and remineralize teeth	24
	Fe ₃ O ₄ /dextran	POD	Target biofilm cells and degrade EPS matrix	25
	Fe ₃ O ₄ /dextran/GOx	GOx and POD	Increase H ₂ O ₂ , kill bacteria and degrade the EPS matrix	26
	CoPt@G@GOx	GOx and POD	Causes the death of bacteria both in planktonic states and within biofilms	27
	Iron oxide and iron sulfide	POD	Produce H ₂ O ₂ , degrade biofilm matrix and kill bacteria	28
	CaO ₂ /TA/Fe	POD	Blast the tight biofilm and proceed with cascade catalysis eradication of biofilm	29
Endodontic infections	Fe ₃ O ₄	POD	Enhance antibacterial activity on root canal surfaces and in dentinal tubules	30
	Fe ₃ O ₄ /CaO ₂	POD	Scavenge on root canal biofilm infection and prevent further inflammation expansion	31
	Fe ₃ O ₄ /GOx	GOx and POD	Eliminate <i>E. faecalis</i> and <i>C. albicans</i> and destructed the dense biofilm matrix	32
	Cu ²⁺	POD	Eradicate biofilms caused by <i>E. faecalis</i> and <i>C. albicans</i> in the root canals of infected teeth	33
	MPN-Pd	OXD	Inhibit biofilms formed by bacteria, fungi, and polymicrobial communities	34
Periodontal disease	Iron oxide	POD	Kill bacteria and degrade and remove biofilms	35
	Oxygenated nanodiamonds	POD	Destruct bacterial cell membranes and biofilms	36
	CN-PtNCs	OXD and POD	Alleviate inflammation and mitigate bone loss	37
	FeSN	POD	Decrease GSH and ATP and enhance bacterial killing efficiency	38
	Au/Pt NCs@GO _x	GOx and POD	Disrupt biofilms and kill bacteria	39
	Fe ₃ O ₄ @Ce6/C6@MnO ₂	CAT	Provide oxygen in infection sites and inhibit anaerobic pathogens	40
	CaO ₂ /MnO ₂	CAT	Provide a continuous oxygen supply and enhance the potential for periodontal healing	41
	CeO ₂ @Ce6 NPs	SOD and CAT	Eradicate bacteria and mitigate inflammation	42
	Lu-Bi ₂ Te ₃ @Fe ₃ O ₄	POD	Cause nitrosative stress on biomacromolecules and damage bacterial cell membranes and DNA	43
Peri-implantitis	MnO ₂	CAT	Entirely destroy biofilms without harming the surrounding mucosa or implant surfaces	44
	Ce-MOF	OXD	Consume extracellular ATP, inhibit bacterial adhesion and prevent biofilm formation	45
	CuN _x -CNS SAzyme	POD, OXD and CAT	Inhibit multidrug-resistant bacteria and eliminate stubborn biofilms	46
	Cu ₂ MoS ₄	POD, OXD and CAT	Kill bacteria, polarize macrophages and promotes healing of infected tissue	47

Nanozymes primarily exhibit antibacterial effects by mimicking the POD and OXD activities of natural enzymes, leading to the generation and regulation of ROS. ROS are intermediate chemical species formed during the partial reduction of oxygen, encompassing H₂O₂, [•]OH, superoxide anions ([•]O₂[−]), and singlet oxygen (¹O₂).⁵⁸ ROS can irreversibly damage bacterial structures, including cell walls, membranes,

DNA, proteins, polysaccharides, and nucleic acids.⁵⁹ Additionally, they can disintegrate mature biofilms and inhibit their formation. H₂O₂ has intrinsic antibacterial properties at high concentrations (166 mM to 1.0 M) but can also damage healthy tissues.⁵⁴ However, nanozymes with POD-like activity can transform low concentrations of H₂O₂ (<1 mM) into highly toxic [•]OH, effectively eliminating bacteria.⁶⁰ To further reduce

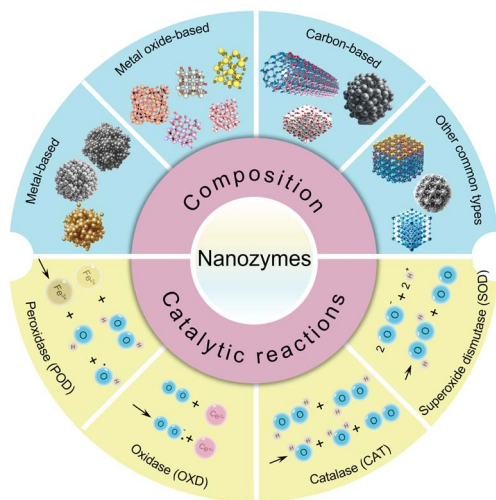


Fig. 1 Schematic presentation of nanozymes classifications (composition and catalytic reactions).

reliance on H₂O₂, nanozymes with OXD-like activity have been developed. These nanozymes catalyze oxygen into H₂O₂ and highly reactive [•]O₂⁻/[•]O₂, demonstrating potent antibacterial capabilities.³⁷

In addition to generating ROS, nanozymes can also generate reactive nitrogen species (RNS), which harm cells and demonstrate potent antibacterial effects against various bacteria, including resistant strains.⁴³ Furthermore, nanozymes can mimic DNase-like activity, accelerating the hydrolysis of DNA and eliminating biofilms.⁶¹ eDNA, a key component of the extracellular matrix (ECM), helps bacteria adhere to surfaces and connect with each other, maintaining biofilm integrity.⁶² DNase nanozymes disrupt the ECM, significantly enhancing traditional antibiotics' efficacy against enclosed bacteria and offering a promising strategy to combat drug-resistant bacteria. In addition to enzyme-activity mechanisms, nanozymes catalyze oxygen production from H₂O₂, creating an aerobic environment that enhances antibacterial effects against anaerobic bacteria.^{40,63} Moreover, the mechanical force from the rapid movement of oxygen produced through the catalysis of H₂O₂ can disrupt and decontaminate biofilms, thereby exerting an antibacterial effect.⁴⁴

4. Application of nanozymes in dental antibacterial field

According to research, there are approximately 700 types of microorganisms in the human mouth.^{64,65} Various microorganisms in the oral cavity form a closely related ecosystem with the host, and the microbial community is interconnected with the extracellular matrix to form a biofilm on the surface of the oral cavity.^{66,67} When the oral environment changes, the imbalance of the microbial ecosystem, in which the level of bacterial flora changes, may result in various oral diseases, including caries and periodontal disease, *etc.*^{68,69} However, conventional antibacterial agents have poor removal effect on

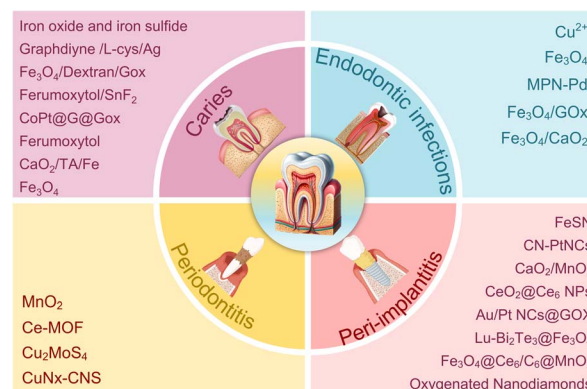


Fig. 2 Application of nanozymes in dental antibacterial field (caries, endodontic infections, periodontitis and peri-implantitis).

dental plaque biofilms, and studies have found that the proportion of oral antimicrobial resistant bacteria gradually increases.⁷⁰ Hence, there is a pressing need for novel clinical treatments to combat biofilms. Nanozyme-based antibacterial agents have gained significant attention in dentistry due to their cost-effectiveness, structural stability, exceptional antibacterial performance, and broad antibacterial spectrum.⁷¹ In this section, we explore how nanozymes can be utilized as a tool for the treatment and prevention of dental infections (Fig. 2).

4.1 Caries

Dental caries is a prevalent bacterial infection, affecting up to 90% of school children and nearly all adults worldwide.⁷² Dental caries primarily results from the colonization and biofilm formation of pathogenic microorganisms on tooth surfaces.⁷³ The protective extracellular matrix embedding bacteria makes dental biofilms difficult to remove or treat. Biofilms create acidic microenvironments that dissolve enamel apatite, resulting in dental caries.⁷⁴ Traditional antimicrobials, such as chlorhexidine, often fail due to their limited effectiveness against cariogenic biofilms.^{75,76} Therefore, more potent anti-biofilm treatments are necessary for caries prevention.

In 2016, Gao *et al.* first reported a novel strategy using Fe₃O₄ nanozymes to manage plaque biofilms and prevent dental caries (Fig. 3A). The study demonstrated that Fe₃O₄ nanoparticles possess POD-like activity, converting H₂O₂ into free radicals in acidic conditions, which degrades exopolysaccharides and kills bacteria. This process mitigates dental caries severity and can halt its progression while preserving normal tissue *in vivo*.²⁰ Both Fe₃O₄ and ferumoxytol are iron oxide nanoparticles (IONPs) with comparable catalytic activities crucial for their therapeutic effects. Ferumoxytol, approved by the US Food and Drug Administration for treating iron deficiency, has also been shown to inhibit tumor growth in mice by enhancing macrophage-associated ROS production.^{21,77} Inspired by the above research, Liu *et al.* discovered that ferumoxytol binds within the biofilm ultrastructure and generates free radicals from H₂O₂, which disrupt cell membranes and degrade the extracellular polymeric substances matrix,



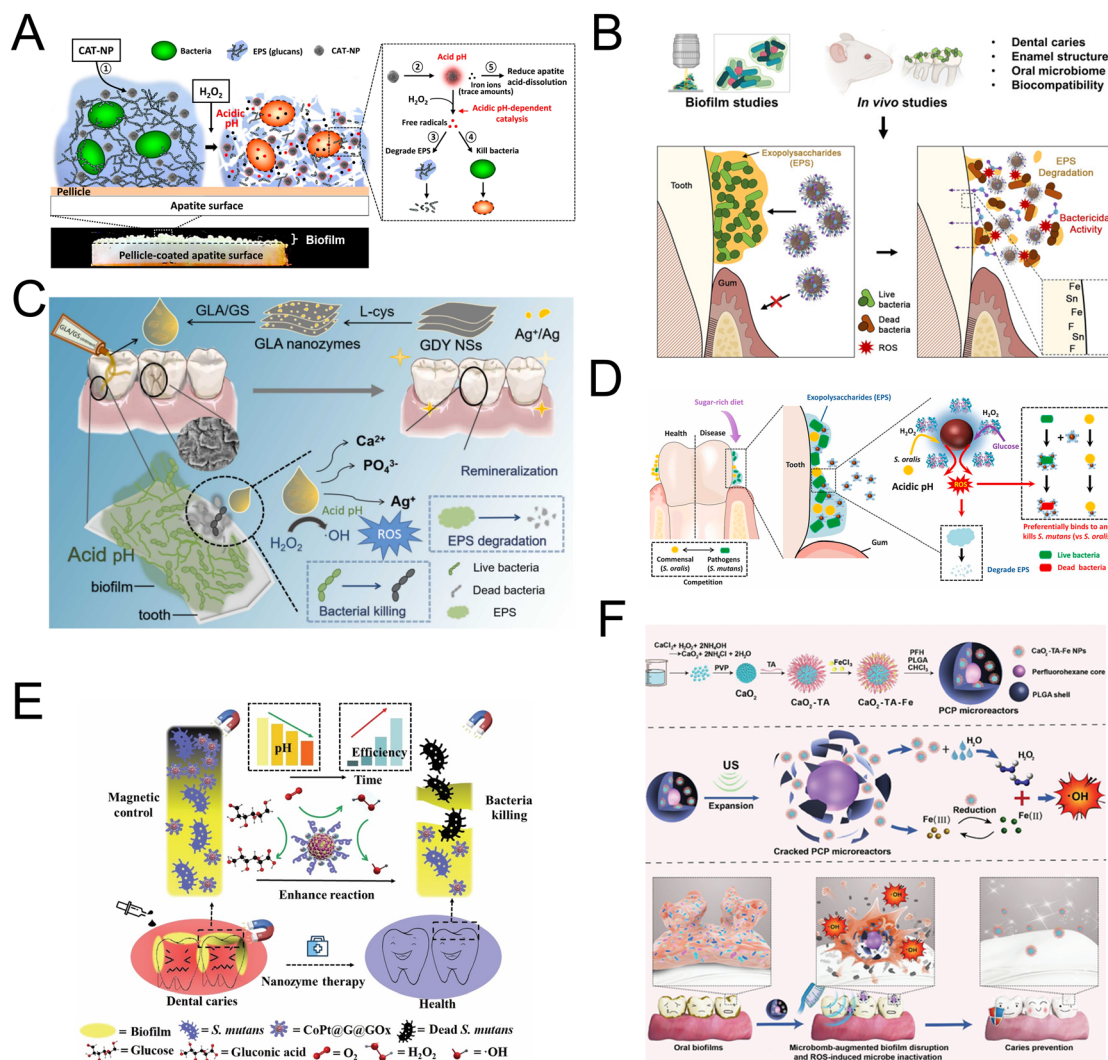


Fig. 3 (A) Schematic diagram of biofilm disruption under acidic conditions by CAT-NP/H₂O₂ *in situ*. Reproduced with permission from ref. 20. Copyright 2016, Elsevier. (B) Schematic diagram of enhanced bioactivity and caries-protective effects against biofilms using laboratory and *in vivo* models. Reproduced with permission from ref. 23. Copyright 2023, Springer Nature. (C) Schematic diagram of synthesis of GLA/GS ointment using coordination-reduction combined biomineralization strategy and the catalytic ion therapy function for caries prevention. Reproduced with permission from ref. 24. Copyright 2024, Elsevier. (D) Schematic diagram of the selective catalytic-therapeutic mechanism of Dex-IONP-GOx for treatment of virulent acidogenic biofilms. Reproduced with permission from ref. 26. Copyright 2024, Elsevier. (E) Proposed mechanism of a hybrid nanozyme targeting oral pathogenic biofilms with antibacterial and antibiofilm effects through enzyme-nanozyme cascade reaction. Reproduced with permission from ref. 27. Copyright 2022, Springer Nature. (F) Proposed concept of the ultrasound-activated ROS generating microbombs targeting dental biofilm. Reproduced with permission from ref. 29. Copyright 2023, John Wiley and Sons.

resulting in *in situ* bacterial death.²¹ Subsequent research presents initial human evidence supporting the therapeutic potential of catalytic IONPs as targeted nanomedicine for oral infectious diseases. It also demonstrates ferumoxytol's antimicrobial specificity against *Streptococcus mutans*, attributed to the interaction between ferumoxytol's carboxymethyl glucan and the glucan-binding protein of *Streptococcus mutans*.²² Ferumoxytol effectively disrupts caries-causing biofilms by catalytically activating H₂O₂, without affecting enamel acid demineralization. To improve the efficacy of ferumoxytol, Huang *et al.* discovered significant synergy when ferumoxytol is combined with stannous fluoride (SnF₂), greatly inhibiting biofilm accumulation and enamel damage more effectively than

either agent alone (Fig. 3B). Additionally, the study demonstrates that SnF₂ enhances ferumoxytol's catalytic activity, significantly boosting ROS generation and antibiofilm efficacy.²³ On the other hand, Liao *et al.* explored a different avenue for mineralization by developing a bioinspired ointment (Fig. 3C). They engineered graphdiyne/L-cysteine/Ag (GLA) nanozymes by anchoring low-dose Ag nanoparticles and ions on graphdiyne *via* a coordination-reduction strategy using L-cysteine. Encapsulated in a gelatin methacryloyl (GelMA) and sodium alginate (SA) base, the GLA/GS ointment activates under acidic conditions typical of dental plaque, converting low-dose H₂O₂ into highly reactive [•]OH. Furthermore, it facilitates enamel remineralization by providing nucleation sites that



attract calcium and phosphate ions from saliva, promoting the growth of hydroxyapatite, the main component of tooth enamel.²⁴

Considering that nanozymes lacked the stabilizing coating necessary for clinical applications, Naha *et al.* constructed bifunctional dextran-coated iron oxide nanozymes (Dex-NZM). Dex-NZM targets *Streptococcus mutans* in oral biofilms with high specificity through a dextran binding mechanism, achieving a selective antibacterial effect to prevent dental caries without affecting surrounding tissues.^{25,78} Pathogens like *Streptococcus mutans* flourish in sugar-rich environments and promote cariogenic biofilms.⁷⁹ Based on Naha's earlier study, Huang *et al.* developed a bi-functional nanohybrid system with glucose oxidase (GOx) covalently attached to dextran-coated iron oxide nanoparticles (Fig. 3D). This system exploits disease-related pathological conditions (*e.g.*, high sugar availability, low pH, and elevated EPS levels) to enhance H₂O₂ production and more effectively trigger ROS in a controlled, pH-dependent manner.²⁶ Building on the advancements made by Naha *et al.* and Huang *et al.* in utilizing dextran-coated iron oxide nanoparticles for targeted antibiofilm action, Dong *et al.* expanded the application of nanozymes into a more integrated system (Fig. 3E). They developed CoPt@graphene@GOx (CoPt@G@GOx), a complex that combines GOx with magnetic graphitic CoPt nanocrystals (CoPt@G). In this configuration, GOx converts glucose present in the environment into gluconic acid and H₂O₂. The CoPt@G component, acting as a POD mimic, then utilizes the H₂O₂ generated by GOx to create highly toxic [•]OH. This mechanism effectively causes the death of bacteria both in planktonic states and within biofilms, leveraging the ROS in a manner similar to the earlier studies but with a unique, integrated approach that enhances efficacy and specificity.²⁷

To address the issues of rapid H₂O₂ depletion and poor stability, which restrict sustained therapeutic effects in dental caries infected with biofilms, Wang *et al.* introduced an alternative strategy. Their research utilized *Streptococcus gordonii* themselves to produce H₂O₂. Utilizing iron oxide and iron sulfide nanozymes with POD-like activity, the researchers leveraged bacterial metabolism to produce H₂O₂, thereby introducing a novel approach to stabilize H₂O₂ and reduce *Streptococcus mutans* biofilm formation on human dentin surfaces.²⁸ Many studies on nanozymes in antimicrobial biofilms have demonstrated significant reductions in biofilm formation and damage to biofilm structures. However, most of these studies do not employ clinically convenient treatment protocols or address the specificities of biofilms in particular environments, restricting their clinical applicability and everyday use for oral biofilm elimination. To address this gap, Guo *et al.* developed an ultrasonically-activated microbomb for dental biofilm elimination, incorporating tannic acid-iron modified calcium peroxide (CaO₂-TA-Fe) nanoparticles and perfluorohexane (PFH) into poly(lactide-co-glycolide) (PLGA) (Fig. 3F). The ultrasonic toothbrush activates the PFH, inducing a phase shift that compromises the PLGA shell and triggers the release of H₂O₂ from CaO₂. The tannic acid-iron network converts H₂O₂ into highly toxic [•]OH *via* the Fenton reaction,

enhancing antibacterial effectiveness. This strategy is promising for cost-effective and widespread prevention of caries and treatment of biofilm-associated diseases.²⁹

4.2 Endodontic infections

Endodontic infections are a common issue in dental medicine, frequently causing discomfort and clinical conditions like pain, swelling, pulpitis, apical periodontitis, and root resorption.⁸⁰ The disinfection process is challenging due to the complex root canal system, which includes isthmuses, accessory canals, and dentinal tubules that can harbor bacteria and biofilms.⁸¹ Traditional disinfectants, including Ca(OH)₂, sodium hypochlorite, and chlorhexidine, often fall short in effectively eliminating biofilms and are known to have certain adverse effects.⁸² Recent advancements in nanotechnology have opened up promising avenues for effectively eliminating bacteria, disrupting biofilms, and managing infections within dentinal tubules.⁸³

Biocompatible iron oxide nanoparticles exhibit potent anti-biofilm properties without adverse effects on oral tissues *in vivo*.²⁰ Bukhari *et al.* utilized iron oxide nanoparticles, with intrinsic POD-like activity to catalyze H₂O₂ and generate ROS, offering a novel endodontic disinfection method to enhance bacterial elimination in dentinal tubules.³⁰ The effectiveness of Fe₃O₄ nanoparticles in biofilm control is often limited by low concentrations of H₂O₂ in microenvironments, and adding exogenous H₂O₂ could disrupt tissue healing. Addressing this, Song *et al.* developed a Fe₃O₄-CaO₂ hydrogel that produces ROS in response to the bacterial environment, effectively eradicating root canal biofilm without requiring additional excitation (Fig. 4A).³¹ To enhance the production of H₂O₂, Ji *et al.* utilized GOx, an enzyme that catalyzes the conversion of β-D-glucose into H₂O₂, using molecular oxygen as the electron acceptor (Fig. 4B). This innovative method not only enables localized production of H₂O₂ but also depletes crucial energy sources for bacterial survival. It has shown significant antibacterial effectiveness against the Gram-positive bacterium *Enterococcus faecalis* and the yeast *Candida albicans*, highlighting its potential as a targeted antimicrobial strategy.³² Ethylenediaminetetraacetic acid (EDTA) is widely used as an irrigation solution, but its antimicrobial properties are limited.⁸⁴ Aslan *et al.* engineered EDTA nanoformulations that exhibit catalytic and antimicrobial activities through a Fenton-like reaction with H₂O₂. EDTA nanofibers were used as an irrigation solution to eliminate biofilms formed by *Enterococcus faecalis* and *Candida albicans* in infected root canals.³³ Photothermal therapy (PTT) can disrupt pathogen integrity by inducing localized hyperthermia through noninvasive light irradiation. Chen *et al.* developed metal-phenolic networks with palladium nanoparticle nodes, integrating OXD-like and photothermal properties to effectively inhibit biofilms formed by bacteria, fungi, and polymicrobial communities (Fig. 4C).³⁴ By establishing root canal and oropharyngeal candidiasis models, they demonstrated the significant efficacy of this system in combating infections associated with such biofilms. In addressing the limitations inherent in both chemical and biological strategies, Hwang



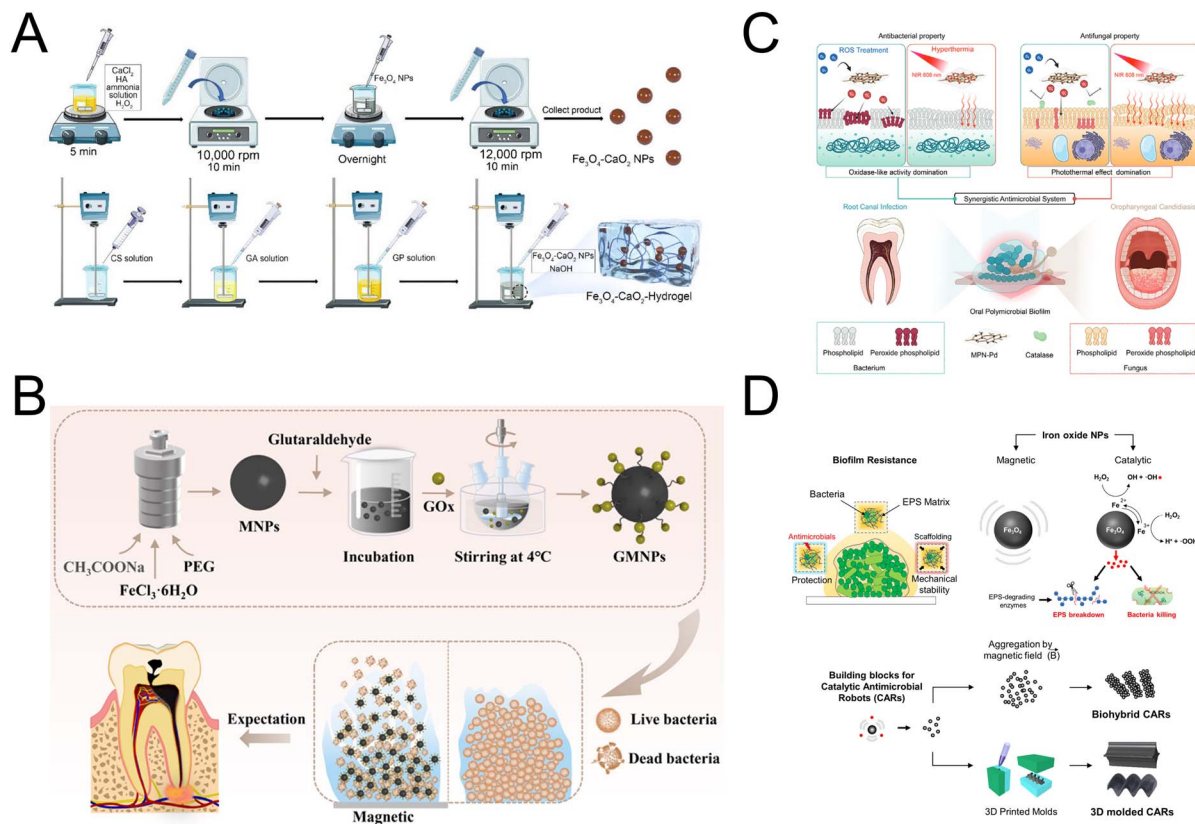


Fig. 4 (A) Schematic diagram of the preparation of $\text{Fe}_3\text{O}_4\text{-CaO}_2$ NPs and $\text{Fe}_3\text{O}_4\text{-CaO}_2\text{-Hydrogel}$. Reproduced with permission from ref. 31. Copyright 2022, Multidisciplinary Digital Publishing Institute. (B) Schematic diagram of the preparation and application of MNPs and GMNPs. Reproduced with permission from ref. 32. Copyright 2021, American Chemical Society. (C) Schematic diagram of the MPN-Pd-mediated synergistic antimicrobial system for treating oral polymicrobial biofilm-associated infections. Reproduced with permission from ref. 34. Copyright 2023, John Wiley and Sons. (D) Catalytic and magnetic iron oxide NPs as building blocks for small-scale robots designed for biofilm killing and removal. Reproduced with permission from ref. 35. Copyright 2019, The American Association for the Advancement of Science.

et al. introduced catalytic antimicrobial robots (CARs), an innovative fusion of chemical and mechanical methodologies (Fig. 4D). These CARs utilize the chemical potency of iron oxide nanoparticles combined with their mechanical disruption abilities to attack both the structural and biological defenses of biofilms. The magneto-catalytic capabilities of these robots are driven by iron oxide nanoparticles that mimic POD activity, catalyzing H_2O_2 into reactive molecules that dismantle biofilms. Additionally, these nanoparticles, whether free-floating or encapsulated within various matrices, can be magnetically maneuvered to precisely target biofilm accumulations. The robots navigate predetermined paths to dislodge, scrub away, and eliminate bacterial remnants and biofilm residues.³⁵ Thus, CARs offer a comprehensive approach to overcoming biofilm-associated infections in endodontic treatments.

4.3 Periodontal disease

Periodontal disease, a common inflammatory disorder, is primarily caused by chronic bacterial infection from periodontal pathogens.⁸⁵ It can result in tooth loss, affecting aesthetics, masticatory function, and overall quality of life, and may also be linked to systemic conditions such as diabetes, cardiovascular disease, and Alzheimer's disease.⁸⁶ Effective

management and timely removal of harmful oral microorganisms are essential for maintaining oral and overall health. Adjunctive antibiotic therapy with mechanical debridement is commonly used in periodontitis treatment to eliminate pathogenic microorganisms and reduce bacterial recolonization.⁸⁷ However, the rise of bacterial resistance due to antibiotic abuse has created an urgent need for alternative strategies.⁸⁸ Enzyme mimics have become a novel class of antibiotics, noted for their outstanding antibacterial properties, minimal systemic toxicity, and resistance to multi-drug resistance mechanisms.⁸⁹

Among novel materials for medical applications, nanodiamonds (NDs) and various carbon-based nanomaterials stand out due to their unique properties and functionalities.^{90,91} Fang *et al.* synthesized oxygenated nanodiamonds (O-NDs) with POD-like activity, capable of catalyzing the production of free radicals in the presence of low concentrations of H_2O_2 (Fig. 5A). These radicals enhance the destruction of bacterial cell membranes and biofilms, optimize periodontal inflammation management, and accelerate healing at periodontal infection sites.³⁶ Further innovations in carbon-based nanomaterials include the work of Wu *et al.*, who developed an injectable anti-biofilm ointment.³⁷ This formulation combines Pt nanoparticle clusters (PtNCs) and graphitic carbon nitride (CN) with a mixture of PEG₄₀₀/



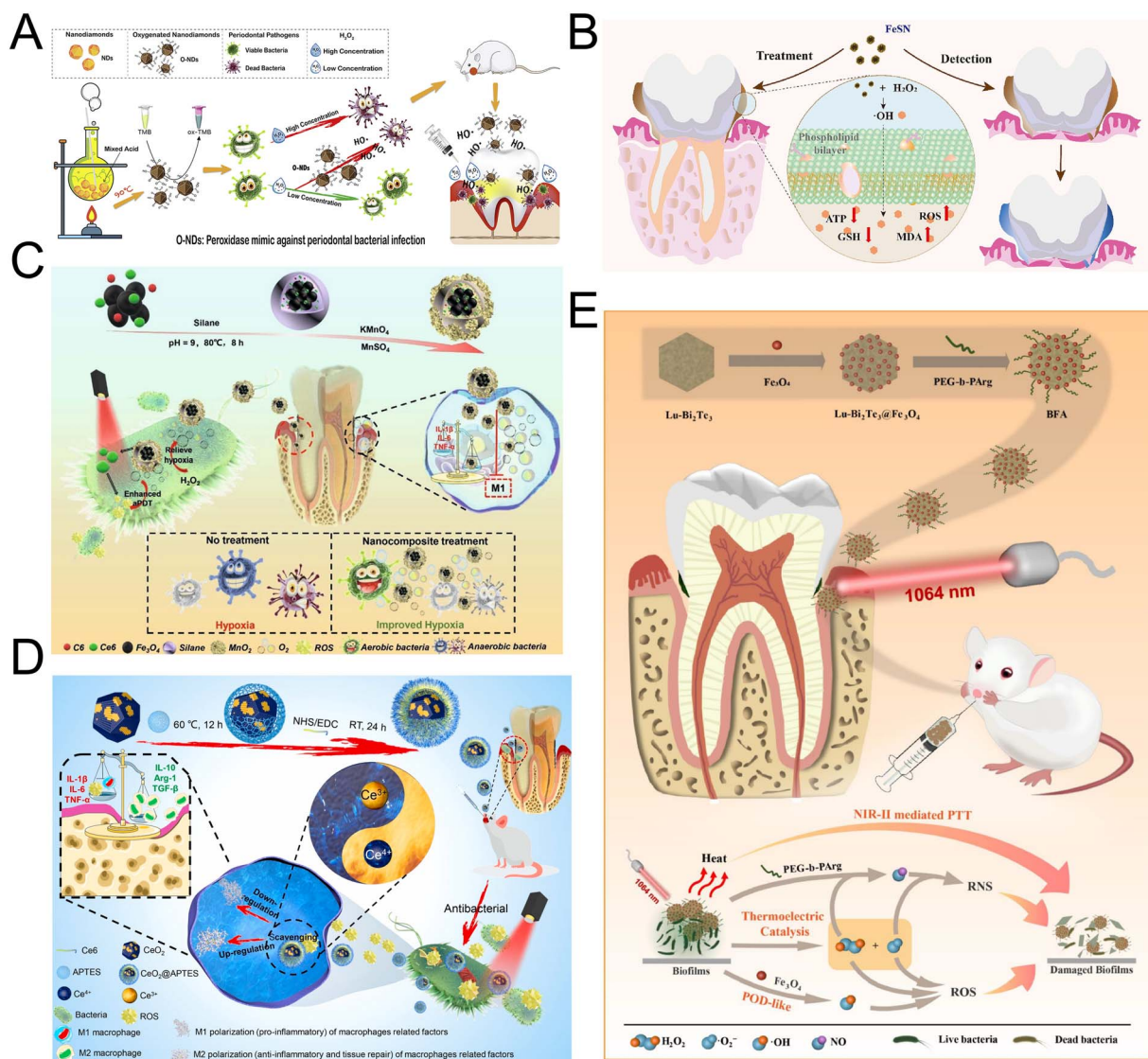


Fig. 5 (A) Schematic diagram for the synthesis of O-NDs and the O-NDs/H₂O₂ system for antibacterial defense in periodontal diseases. Reproduced with permission from ref. 36. Copyright 2020, Elsevier. (B) Schematic diagram of enhanced bioactivity and caries-protective effects against biofilms using laboratory and *in vivo* models by co-delivering fluoride, iron, and tin on the outer enamel surface. Reproduced with permission from ref. 38. Copyright 2023, Elsevier. (C) Schematic diagram of the synthesis of F@Ce6-M NCs and their application in enhanced and selective antibacterial properties and downregulation of the pro-inflammatory cytokines for the treatment of periodontal diseases by ameliorating hypoxia. Reproduced with permission from ref. 40. Copyright 2021, John Wiley and Sons. (D) Schematic illustration of CeO₂@Ce6 nanocomposite in synthesis, the antibacterial mechanism and modulating the polarization of macrophages for the treatment of periodontal diseases. Reproduced with permission from ref. 42. Copyright 2021, Elsevier. (E) Preparation and biomedical application of BFA. Reproduced with permission from ref. 43. Copyright 2023, Elsevier.

PEG₄₀₀₀. The CN-PtNC ointment exhibits both OXD-like and POD-like properties, enabling ROS production without the need for light, due to its efficient oxygen adsorption and activation capabilities. Notably, this ointment effectively treated periodontitis in rats, reducing inflammation and minimizing bone loss.³⁷ To enhance the utilization of H₂O₂ by nanozymes, Shen *et al.* created FeSN nanozymes that effectively mimic active site of POD, achieving this through the self-assembled coordination of cysteine, histidine, and iron ions, resulting in high catalytic efficiency (Fig. 5B). The FeSN nanozyme displays distinctive antibacterial properties, causing an increase in ROS levels and

a decrease in glutathione and ATP within *F. nucleatum* cells, which enhances the efficiency of bacterial eradication.³⁸ Besides, Au nanoclusters (Au NCs) have also been recognized for their intrinsic enzyme-like activities, which include high catalytic efficiency and superior biocompatibility suitable for *in vivo* applications. However, the limited POD-like activity of Au NCs significantly hindered their use in antibacterial therapies.³² To overcome this, Wang *et al.* addressed the limitation by creating bimetallic nanoclusters, incorporating Pt atoms into Au NCs to improve the catalytic sites of the clusterzyme. Additionally, they developed a cascade catalytic nanozyme by

chemically coupling GOx onto Au/Pt NCs, which transforms non-toxic glucose into gluconic acid and H_2O_2 . This configuration optimizes the environment and substrate for POD reactions, enhancing antibacterial and antibiofilm activity against *F. nucleatum*.³⁹

Antimicrobial photodynamic therapy (aPDT) employs light-activated photosensitizers to generate cytotoxic ROS, offering a promising sterilization method.^{93,94} However, its effectiveness is compromised in hypoxic conditions, typical of periodontal disease, where anaerobic bacteria thrive and oxygen scarcity leads to less ROS generation and increased inflammation.^{95,96} Addressing this challenge, Sun *et al.* developed a nanoplatform that features MnO_2 -coated, amphiphilic silane-modified nanoparticles with an Fe_3O_4 core, Chlorin e6, and Coumarin 6 (Fig. 5C). This innovative design enhances aPDT efficacy by catalyzing H_2O_2 into O_2 , improving oxygen availability. It also supports magnetic targeting and real-time treatment monitoring, potentially overcoming hypoxia limitations in periodontal therapy.⁴⁰ Building on this innovation, Santos *et al.* introduced another novel solution addressing the oxygen scarcity in periodontal treatments through the use of electrospinning technology. They developed composite fibrous membranes with a bead-on-string structure that effectively function as a controlled oxygen-release system. These membranes contain CaO_2 nanoparticles as an oxygen-generating precursor and MnO_2 nanosheets as nanozymes to catalyze H_2O_2 decomposition into oxygen.⁴¹ This approach not only complements the earlier advancements by providing a continuous oxygen supply but also significantly enhances the potential for periodontal healing by maintaining an oxygen-rich environment throughout the treatment process. However, aPDT can also precipitate pro-inflammatory effects due to excessive ROS, which disrupts the oxidant/antioxidant balance and attracts inflammatory cells, potentially damaging periodontal tissues.⁹⁷ To address these inflammatory challenges, Sun *et al.* further innovated by developing CeO_2 @Ce6 nanocomposites that not only possess antibacterial properties but also mitigate inflammation (Fig. 5D). CeO_2 NPs mimic SOD and CAT by catalytically reacting with superoxide and H_2O_2 through redox cycling between Ce^{3+} and Ce^{4+} ions. Under red light excitation at 630 nm, the CeO_2 @Ce6 nanocomposites achieve effective sterilization by enhancing ROS production during aPDT, followed by a swift reduction in ROS levels post-therapy due to the exceptional ROS-scavenging capacity of the CeO_2 . This dual function—boosting ROS for bacterial eradication and then curbing it to prevent inflammation—highlights the potential of CeO_2 @Ce6 nanocomposites to balance the therapeutic and inflammatory responses in periodontal disease treatment.⁴² In contrast, PTT operates on a different principle, utilizing photothermal agents to convert light into heat, effectively eliminating bacteria through hyperthermia. This heat disrupts cell membranes and denatures proteins, directly targeting free bacteria and biofilms.⁹⁸ Expanding on this technique, Dai *et al.* introduced a synergistic approach that combines PTT with heat-induced ROS and RNS to enhance biofilm eradication and aid in the infected tissue recovery (Fig. 5E). Their innovative platform involves decorating lutetium-doped Bi_2Te_3 nanoplates with

POD-like Fe_3O_4 and PEG-*b*-PArg. These nanoparticles catalyze reactions with H_2O_2 to produce highly reactive $\cdot\text{OH}$, which then interacts with $\cdot\text{O}_2^-$ and NO to generate the potent RNS, ONOO^- . The RNS and ROS generated are broad-spectrum antibacterial agents, causing nitrosative stress on biomacromolecules and damaging bacterial cell membranes and DNA.⁴³

4.4 Peri-implantitis

Dental implants have become the leading clinical method for restoring the structure and function of missing teeth over the past four decades.⁹⁹ Despite their high survival rate, the incidence of peri-implant diseases continues to rise.¹⁰⁰ Mechanical decontamination methods can severely alter the implant's microstructure and surface electrochemical properties,^{101,102} while nonmechanical antiseptics often fail to remove tightly bound extracellular polymeric substance structures effectively.⁴⁴ Systemic antibiotic administration has traditionally been preferred for treating implant-associated infections, but it frequently leads to antibiotic resistance.¹⁰³ In response to these challenges, nanozymes have garnered considerable interest for their ability to impart antibacterial and anti-inflammatory properties to implant surfaces.^{3,104} These nanozymes catalyze H_2O_2 decomposition, generating ROS that destroy bacterial DNA, proteins, and lipids, thus mitigating bacterial resistance.¹⁰⁵ This innovative approach promises to enhance antimicrobial efficacy while preserving the structural and functional integrity of dental implants, presenting a promising alternative to conventional treatments.

Lee *et al.* devised a novel and safe treatment method for peri-implantitis, utilizing the dynamic action of micro-sized oxygen bubbles (Fig. 6A). These bubbles are produced from a catalytic reaction involving H_2O_2 and MnO_2 nanozyme-doped silica diatom microparticles, referred to as diatom microbubblers (DM). The swift movement of these tiny DM particles allows them to navigate through the small spaces between implant screws, delivering just enough force to thoroughly eliminate biofilms without damaging the surrounding tissues or the surfaces of the implants.⁴⁴ In deep tissues, the catalytic ROS production by nanozymes is often reduced due to restricted substrate diffusion. Therefore, it is crucial for nanozymes to have enhanced activity to generate sufficient ROS at inhibitory levels, even under low substrate concentrations, when addressing deep infections. Metal-organic frameworks (MOFs) are porous crystalline materials made of organic ligands and metal ions or clusters, notable for their ordered pore structures and large surface areas.^{106,107} Zhang *et al.* developed cerium-based metal-organic framework (Ce-BTC) coatings on medical titanium surfaces through a solvothermal process, followed by hydrogen plasma immersion ion implantation. These coatings feature numerous coordinatively unsaturated metal sites (Fig. 6B).⁴⁵ The resulting Ce-BTC coatings exhibit a robust ATP deprivation capacity and OXD-like activity. This combination effectively inhibits biofilm formation and eradicates bacteria, especially in the acidic microenvironment induced by bacterial activity. Compared to conventional noble-metal or transition-metal oxide/sulfide-based nanozymes, single-atom nanozymes

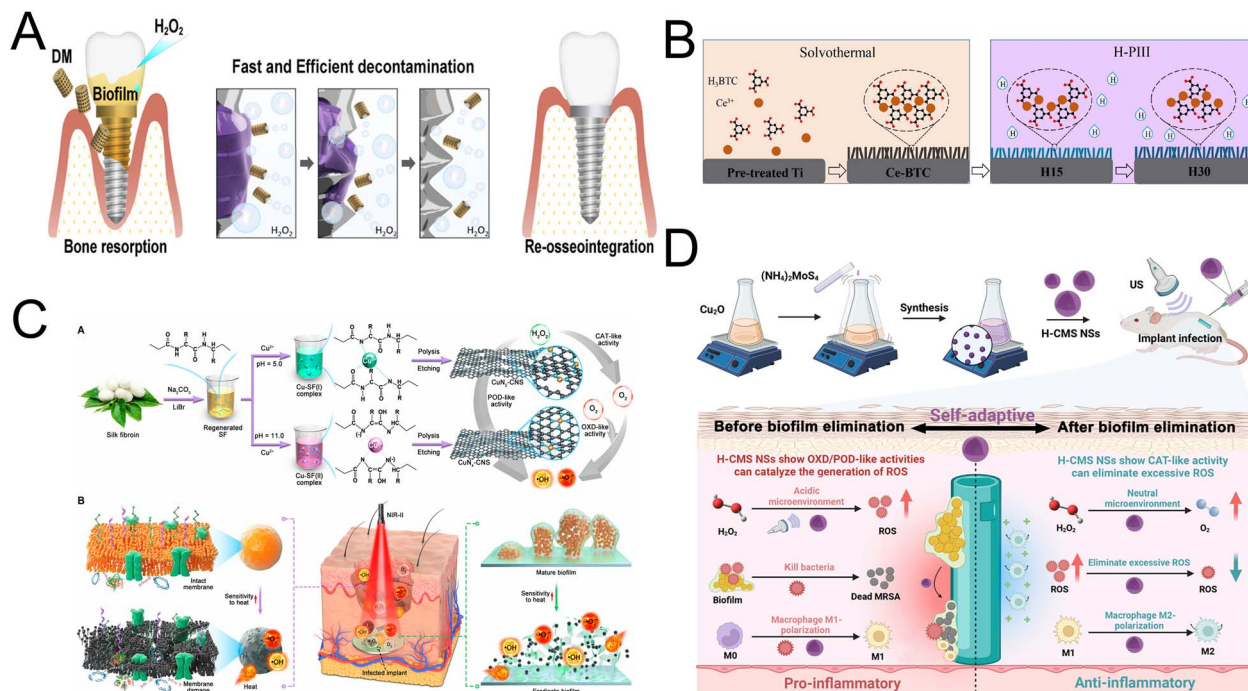


Fig. 6 (A) Hypothetical schematic diagram of DM decontaminating the peri-implantitis-affected implant. Reproduced with permission from ref. 44. Copyright 2022, American Chemical Society. (B) Schematic diagram of the preparation of Ce-BTC, H15, and H30. Reproduced with permission from ref. 45. Copyright 2022, Elsevier. (C) Synthesis procedure and antibacterial therapy mechanism of CuN_x-CNS SAzyme. Reproduced with permission from ref. 46. Copyright 2023, American Association for the Advancement of Science. (D) Preparing Hollow Cu₂MoS₄ Nanospheres with a self-adaptive antibiofilm effect and immune modulation for treating implant infections. Reproduced with permission from ref. 47. Copyright 2023, American Chemical Society.

(SAzymes) feature atomically dispersed metal atoms that maximize atom utilization efficiency, significantly boosting their enzyme-like activities.¹⁰⁸ Bai *et al.* developed a copper and silk fibroin complex to create copper SAzymes with atomically dispersed copper sites on ultrathin 2D porous N-doped carbon nanosheets (CuN_x-CNS) (Fig. 6C). These SAzymes demonstrate POD, CAT, and OXD activities, effectively converting H₂O₂ and O₂ into ROS *via* parallel and cascaded reactions. *In vitro* and *in vivo* experiments show that the optimized CuN₄-CNS effectively inhibits multidrug-resistant bacteria and eradicates persistent biofilms, presenting significant therapeutic potential for deep implant-related biofilm infections.⁴⁶ However, excessive inflammation can damage tissues surrounding the implant post-biofilm removal. Therefore, it is crucial to develop therapeutic agents that can modulate the inflammatory response throughout the various stages of treatment. To address this, Yang *et al.* prepared hollow Cu₂MoS₄ nanospheres (H-CMS NSs) with pH-responsive enzyme-like activities using an etching-precipitation method (Fig. 6D). These H-CMS NSs, with OXD/POD-like activities, generate ROS specifically in the acidic microenvironment of biofilms. In a neutral environment post-biofilm elimination, H-CMS NSs demonstrate CAT-like activities that inhibit M1 macrophage polarization and decrease proinflammatory cytokines. Yang's team has developed pH-responsive nanozymes capable of adaptively targeting biofilms while also modulating the macrophage-mediated inflammatory

response, offering an efficient approach for treating implant infections.⁴⁷

5. Challenges and perspectives

This review explores the classification, mechanisms, and dental applications of nanozymes, highlighting recent advances in antibacterial research that underscore their significant promise in dental care. The oral microbial community is closely related to both oral and systemic health.¹⁰⁹ It can lead to dental caries, apical periodontitis, periodontal disease, pericoronitis, and oral mucosal disorders, and other oral conditions, as well as contribute to many systemic diseases.¹¹⁰ Natural enzymes such as proteolytic and amylase, recognized for their antibacterial, anti-inflammatory, and immune-enhancing properties, have been utilized to treat periodontitis, oral ulcers, and dental caries. However, these enzymes face significant challenges, including poor stability, high costs, labor-intensive purification, and difficulties in long-term storage.¹¹¹ Advances in biomaterials, particularly nanozymes, have revolutionized oral health care by enhancing functionality and quality of life. Nanozymes, emerging as innovative alternatives, offer stability, scalability, and tunability.¹¹² They provide unique solutions to oral infectious diseases through their distinct physicochemical properties and functional advantages, exhibiting antibacterial, antioxidant, and anti-inflammatory effects.^{113,114} Despite these promising advances, the full potential of nanozymes remains



underutilized in dentistry. Many benefits observed in other medical fields have yet to be explored in dental applications, and numerous challenges highlighted in existing studies remain unresolved.

Firstly, the mechanisms of nanozymes are not well-defined, and their catalytic efficiency is insufficient. Although some studies have proposed potential catalytic mechanisms, their precise workings remain unclear.¹¹⁵ Collaborative efforts in computational simulation, theoretical calculation, and artificial intelligence are crucial for advancing knowledge of nanozyme functions, enhancing their catalytic activities, and expanding their potential applications.^{116–118} Furthermore, the catalytic activity of nanozymes is currently inferior to that of natural enzymes, limiting their effectiveness in *in vivo* antibacterial applications. Recent studies suggest that the catalytic performance of nanozymes is influenced by their intrinsic physico-chemical properties (shape, size, surface modifications) and external factors (temperature, pH, substrate concentration).¹¹⁹ The influence of the complex biological microenvironment on nanozyme activity and their long-term effects remains under explored. There is a pressing need to develop nanozymes with high catalytic efficiency that are well-suited for biological systems. Notably, SAzymes, which feature atomically dispersed active sites akin to those in natural metalloenzymes, represent a new frontier in cost-effective catalysis.¹²⁰

Secondly, nanozymes exhibit limited specificity. In dentistry, nanozymes exhibit enzyme-like activities, including OXD, POD, CAT, and SOD. However, unlike natural enzymes, nanozymes often lack complex substrate-binding pockets, leading to nonspecific substrate interactions and behavior similar to conventional catalysts, which can cause side effects in biological settings.^{14,17} Thus, developing precise nanozyme-based therapies is a pressing research focus. Efforts should prioritize enhancing nanozyme selectivity for targeted therapies. Current research indicates that specificity may be enhanced by modifying nanozymes with aptamers, chiral molecules, or molecularly imprinted polymers, or by integrating them with natural enzymes that have inherent substrate selectivity.¹¹ Future studies should focus on developing new types of nanozymes with increased specificity using these approaches, potentially leading to precise therapies that effectively treat specific dental conditions without adverse effects.

Thirdly, research on the antibacterial effects of nanozymes in dentistry lags behind their advancements in other areas of clinical medicine. While nanomaterials have addressed various therapeutic needs in dentistry, other medical fields have more thoroughly exploited the intrinsic properties of nanozymes—including light sensitivity, photothermal effects, magnetism, and synergistic chemodynamic properties—to enhance their catalytic activity.^{121,122} The emerging concept of nanocatalytic medicine, driven by extensive research, holds promise for monitoring, antibacterial treatments, tumor therapy, regeneration, and tissue protection.^{123–126} However, these innovations have yet to be widely applied in dentistry. By integrating the unique properties of nanozymes and leveraging their proven successes in other medical domains, substantial progress in dental treatments and patient outcomes could be achieved.

Despite the demonstrated efficacy of nanozymes in inhibiting bacterial growth *in vitro*, translating this innovative approach into clinical practice presents significant challenges due to the limited scope of current preclinical research. One of the primary challenges is the complexity of the oral cavity's microbial environment and dynamic conditions, which make it difficult to predict and control the effects of nanozymes on both pathogenic and beneficial microbes, as well as on host cells. Future studies should investigate the antibacterial activity of nanozymes and their unintended impacts on beneficial microbial communities, and develop nanozymes that can selectively target pathogenic bacteria while sparing beneficial microbes—key to maintaining oral health. Additionally, the unique interactions between nanozymes and host cells, such as their ability to achieve targeted intracellular delivery and controlled catalytic activity and magnetic field-driven drug release, represent a promising direction for future development.^{127,128} Therefore, carefully designed studies focusing on cellular uptake, intracellular degradation, and effects on cell viability are essential to assess the therapeutic potential and biocompatibility of nanozymes. Furthermore, evaluating potential nanozyme-induced inflammatory responses, alterations in immune signalling, and unintended cytotoxicity through comprehensive *in vitro* and *in vivo* immune assays is crucial to ensure their safety and understand their interactions with immune cells in clinical applications. Another primary challenge is the unclear nature of nanozymes' degradation products and their potential impacts on the body. Nanozymes often contain non-essential metal elements, whose accumulation in tissues can pose health risks.^{129,130} Strategies such as coating nanozymes with biocompatible polymers, using core-shell structures, and employing surface passivation, biocompatible coatings, and doping techniques can help mitigate adverse effects and enhance clinical viability. Furthermore, even at low concentrations, pH-dependent nanozymes can react strongly in gastric acid, potentially increasing the gastrointestinal burden, leading to weight loss, and elevating oxidative stress in blood and liver.^{131–133} Nanozymes' ability to be engineered for selective degradation or stability based on environmental triggers offers more precise control than traditional antimicrobials. Developing pH-responsive coatings or encapsulating materials could ensure their stability in acidic stomach environments, presenting a promising solution. Researchers should also design *in vitro* saliva simulation experiments to explore how natural enzymes affect nanozyme stability and catalytic activity, and develop appropriate animal models to study the distribution, accumulation, and clearance of nanozymes *in vivo*. This will help assess the potential impact of long-term enzyme interactions on nanozyme functionality loss or the emergence of toxic degradation products. In conclusion, although nanozymes offer an innovative and promising approach for managing oral infections, considerable efforts are needed to fully understand their interactions with the oral microbiome, ensure their long-term safety, address potential side effects, and determine their clinical applicability.



6. Conclusions

Antibacterial nanozymes have developed rapidly and hold significant potential for preventing and treating dental infectious diseases. However, they have not yet fully met clinical demands in dentistry, presenting numerous challenges and opportunities for further research and application. Unresolved issues require deeper exploration to understand their catalytic mechanisms and to develop new nanozyme varieties suitable for clinical trials. It is essential for dental researchers to collaborate on molecular studies, address specific clinical challenges, and evaluate the long-term effects of nanozymes in the oral environment. This review aims to spark interest and provide insights into the antibacterial properties of nanozymes, advocating for continued research to develop safe and effective nanozymes for future clinical applications.

Data availability

No primary research results, software or code have been included and no new data were generated or analysed as part of this review.

Author contributions

Lipeng Liu: conceptualization, investigation, writing – original draft, writing – review & editing. Yaoyuan Zhang: investigation, methodology, writing – original draft. Tianjuan Ju: writing – original draft. Xutao Chen: conceptualization, methodology. Xinwei Li: visualization. Li-an Wu: conceptualization, supervision, project administration, funding acquisition, writing – review & editing.

Conflicts of interest

The authors declare no competing interests.

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