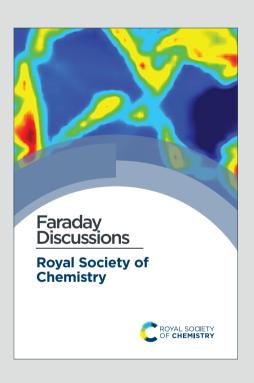
Faraday Discussions



Accepted Manuscript



This is an Accepted Manuscript, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about Accepted Manuscripts in the <u>Information for Authors</u>.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard Terms & Conditions and the Ethical guidelines still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this Accepted Manuscript or any consequences arising from the use of any information it contains.

This article can be cited before page numbers have been issued, to do this please use: D. Chevrier, S. Gautam and A. Scheffel. *Faraday Discuss.*, 2025, DOI: 10.1039/D5FD00021A.



Nanobeam-scanning X-ray Fluorescence Microscopy Reveals the Elemental Composition of Dense Intracellular Bodies in Biomineralizing Coccolithophores

Daniel M. Chevrier,1,* Shristy Gautam,1 André Scheffel2

¹Aix-Marseille Université, CNRS, CEA, BIAM, UMR7265 Institut de Biosciences and Biotechnologies d'Aix-Marseille, CEA Cadarache, F-13108 Saint-Paul-lez-Durance, France

²University of Greifswald, Institute of Aquatic Microbiology, Felix-Hausdorff-Straße 8, 17489 Greifswald, Germany

Coccolithophore microalgae intracellularly produce nanostructured calcitic platelets, known as coccoliths, through a biologically-controlled mineralization process. Mature coccoliths are secreted to the cell surface and assembled into a shell that envelops the cell. The large-scale global production of coccoliths, followed by their sedimentation to the ocean floor, significantly contributes to carbon cycling. Despite progress in understanding the biomineralization pathway of coccoliths, we are still limited in our ability to predict how future climate conditions will impact coccolith formation and thus ocean carbon fluxes. Investigating coccolith biomineralization at the single-cell level is therefore critical to advance our understanding but remains challenging since current imaging techniques lack the combined spatial and temporal resolution coupled with element-specific detection to follow processes in situ. In light of this gap, nanobeam-scanning X-ray fluorescence microscopy (nano-XRF) in the hard X-ray regime is employed here to investigate the intracellular elemental distribution of the coccolithophore Gephyrocapsa huxleyi (formerly Emiliania huxleyi) achieving a resolution of 100 nm and elemental detection from phosphorus (P) to zinc (Zn). Calcium- and phosphorus-rich intracellular bodies, previously proposed to be involved in coccolith biomineralization, were observed in cells initially prepared ex situ by drying. Interestingly, nano-XRF imaging reveals metal species (e.g., Mn, Fe, Zn) within these bodies that were not detected in earlier studies, suggesting multiple biological roles for these structures. Moving towards native-state imaging, G. huxleyi was then imaged in hydrated state using a dedicated liquid cell device. Measurements were performed on G. huxleyi cells both with and without coccolith shell in sea water medium and compared to those of dried cells, demonstrating comparable image quality. The future potential and limitations of liquid cell nano-XRF imaging for coccolithophores and other microorganisms are further discussed.

Intro

Coccolithophore microalgae are single-cell phytoplankton that produce calcite platelets called coccoliths. Coccolithophores and detached coccoliths are found in the surface waters of most marine environments. Coccoliths are intricate, nanostructured calcium carbonate (CaCO₃) biominerals, which vary in shape, crystal organization and size according to species. They are formed inside the cell and eventually exocytosed to the cell surface, creating an exterior layer known as a coccosphere, which covers the microalgal cell.

Although coccoliths are relatively small, ranging from 0.5-15 μ m in size,¹ the immense quantity and wide distribution of coccolithophores makes them one of the most productive calcium carbonate builders on the planet.² Therefore, in addition to their ecological importance as primary producers (i.e., converting sunlight and CO_2 into organic material), calcification activity of coccolithophores (e.g., production and sedimentation of coccoliths) has a major impact on the oceanic carbon cycle.³ A large number of studies

have revealed mixed responses of coccolithophores to future climate change conditions, specifically ocean acidification.^{4–6} Therefore, understanding calcite biomineralization in coccolithophores is not only of fundamental research interest but also of environmental importance.

Coccoliths are typically produced one by one within an intracellular space known as the coccolith vesicle.³ Here, nanocrystals of calcite precipitate and grow under strict (bio-)chemical conditions and spatial confinement that direct the final coccolith architecture. Electron microscopy studies are now shedding light on crystal growth mechanisms.^{7,8} Before these processes take place in the coccolith vesicle, the coccolithophore cell must uptake and transport the precursor ions for calcite formation (e.g., HCO₃-, Ca²⁺) to the site of mineralization. The recent discovery of calcium- and phosphorus-rich intracellular bodies in coccolithophores has inspired the hypothesis that these bodies serve as concentrated reservoirs of Ca²⁺, assisting in the storage and transport of calcium for calcification.^{9,10} Such Ca-rich bodies have been observed in *Gephyrocapsa huxleyi* (formerly *Emiliania huxleyi*) and *Chrysotila carterae* (formerly *Pleurochrysis carterae*) within vacuole-like compartments and as small Golgi-derived vesicles, respectively.^{11–13} These bodies are also concentrated in (poly-)phosphate or acidic polysaccharides, which likely acts as a chelator of Ca²⁺ and other cations.^{11,14} It has been proposed that these bodies may resemble acidocalcisomes¹⁵ though the biological link between them is undetermined.

A remaining gap in understanding coccolithophore calcification is how the intracellular composition, in particular the composition and presence of the reported Ca-rich bodies, changes in response to mineralization activity or external environmental factors. While previous research efforts employ cryoelectron and cryo-X-ray microscopy techniques to provide high spatial resolution imaging, 7,11,13 they are relatively low-throughput techniques considering the dedicated sample preparation required, which impedes investigations on single-cell level dynamics of biomineralization. At the lower spatial resolution regime, confocal microscopy can follow individual cells with high temporal resolution at timescales of coccolith formation. 16,17 However, confocal microscopy relies on fluorophore staining and has limited sensitivity to detect biological and chemical changes. Recent studies on microalgal cell composition and structure are turning to X-ray fluorescence-based microscopy techniques that provide full cell imaging with subcellular resolution and a broad range of detectable elements. 18-21 X-ray-based imaging methods generally allow for deeper penetration into biological samples compared to electron-based imaging. However, radiation damage should still be considered, especially for imaging at high spatial resolution.²² For instance, X-ray beam-scanning imaging at high resolution (i.e., scanning step equal to X-ray beam size) requires longer acquisition times to maintain adequate signal-to-noise levels compared to lower resolution imaging, which in turn increases the radiation dose.

The work presented herein demonstrates how synchrotron-based nanobeam-scanning X-ray fluorescence microscopy (nano-XRF) can reveal the elemental composition from P to Zn of individual coccolithophores with 100 nm spatial resolution and under ambient conditions dried or in liquid. Samples of *G. huxleyi* were first prepared from actively calcifying cultures for *ex situ* measurements to identify detectable elements pertaining to coccolith structures and the microalgae. Particular emphasis was made on ion-rich intracellular bodies in *G. huxleyi* and how the composition changes across cells and sample condition. After demonstrating the imaging possibilities of nano-XRF on dry coccolithophore cells, samples were measured in their hydrated native-state by employing a liquid cell device. *G. huxleyi* was measured with and without exterior coccoliths to demonstrate single-cell imaging in liquid and to measure the dense intracellular body composition. Further discussion is then given on employing liquid cell nano-XRF for the study of coccolithophores and other microorganisms.

Materials and Methods

Coccolithophore culturing and sample preparation for nano-XRF

Gephyrocapsa huxleyi (strain AWI1516 and RCC1731) were cultivated in the artificial sea water medium Aquil at 20 °C with a 12/12 h light/dark cycle using a full spectrum LED light source (Sansi) at 100 μ E. Cultures were grown in batch condition using sterile and stoppered glass flasks with a large head space. Coccolithophore samples were deposited on either TEM grids or Si₃N₄ substrates for nano-XRF imaging. Specifically, 5 μ L of coccolithophore culture was deposited on the sample substrate and left to sediment for at least 10 min before the drop of media was removed by filter paper. A washing solution of 50 mM NaHCO₃ (pH 8) was then used to dilute the salt concentration by successive washing steps of the substrate surface. Finally, alkaline Milli-Q water (18.2 Ω cm) was used to wash the substrates before final drying.

Decalcification treatment and re-calcification of G. huxleyi

Cells without extracellular coccoliths were prepared according to previous protocols. 10,11 Briefly, a 0.5 M ethylenediaminetetraacetic acid (EDTA) solution (pH 8.0) was added to a culture of *G. huxleyi* at 1/100 volume to obtain a final concentration of 5 mM EDTA. EDTA is a chelating agent that binds to divalent cations such as calcium (Ca^{2+}), effectively sequestering them and thus dissolving calcite minerals. After 5-10 min of incubation at 20 °C with gentle mixing, the culture is virtually free of all exterior coccoliths. Samples were centrifuged at 1000 g for 10 min, resuspended in fresh Aquil medium and prepared for nano-XRF measurement using the abovementioned protocol. For samples at early stage coccosphere formation, the EDTA-treated culture was washed three times with fresh Aquil medium and left to incubate at 20 °C and 100 μ E for 4 h.

Liquid cell device for native-state measurements of G. huxleyi

Silicon nitride membrane (SiN) materials for preparing liquid phase measurements were acquired from Protochips (Echips, Poseidon Select). These materials are composed of a Si wafer base with a 50 nm layer of SiN_x (x \approx 1.3-1.5) on top. Distinct top and bottom SiN materials are used (see Figure 6), both having an imaging window of 50x550 µm² (etched from Si substrate side, leaving exposed SiN). The bottom layer had a spacer height of 5 µm. The spacer material is made of SU-8 photoresist. A custom-made liquid cell holder fabricated by the I14 beamline of Diamond Light Source (Oxfordshire, UK) was used to assemble and house the two SiN materials. SiN materials were first plasma cleaned (Fischione NanoClean, Model 1070) before preparing the liquid cell. Directly after plasma cleaning, the SiN material with SU-8 spacer is placed spacer side-up in the custom liquid cell holder. 1-2 μ L of coccolithophore culture (cell concentration ~1x10⁷ cells mL⁻¹, concentrated by centrifugation at 1000 g for 10 min) is deposited on the bottom SiN material. The second SiN material is then placed on top of the deposited drop of coccolithophores and aligned with the imaging window region of the SiN material below. Once sufficient overlap of the imaging windows is made, which ensures X-ray and ambient light transmission through the liquid cell, a faceplate is secured over the two SiN materials to create a seal between them. The liquid cell device is first imaged with an optical microscope before mounting on the sample stage in the beamline. Coccolithophores were measured under static liquid cell condition, meaning no media flow was applied.

Nanobeam-scanning X-ray fluorescence microscopy

Measurements were conducted at the I14 nanoprobe of Diamond Light Source (Oxfordshire, UK). Samples measured in dried condition were prepared on transmission electron microscopy grids or Si_3N_4 membranes

DOI: 10.1039/D5FD00021A

(Silson Ltd). All measurements were conducted under ambient pressure and temperature using an incident photon energy of 8, 10 or 11.5 keV. In particular, dried EDTA-treated and early stage calcification samples were imaged with 8 keV incident X-ray energy. As a result, Zn Kα XRF signal could not be measured for these datasets (Figures 3 and 4). The focused X-ray beam was ~60 nm (FWHM) in diameter. X-ray fluorescence (XRF) was collected from the front of the sample using a four-element silicon drift detector (RaySpec, UK).²³ A raster scanning step size of 50–100 nm was used with an exposure time of 30–60 ms per pixel to collect high-resolution XRF maps. High-resolution mapping of liquid cell samples used 100 nm and 15-30 ms exposure time. A photon-counting Merlin detector (Quantum Detectors, UK) was used in transmission configuration to collect X-ray scattering for generating phase gradient images (referred to as phase contrast imaging below). Detailed information on how the transmitted signal was transformed, including masking of the beam, background intensity adjustment, and phase integration, is described by Quinn et al.²⁴ PyMCA 5.6.7 software was used to energy calibrate XRF spectra, background subtraction, fit XRF sum-spectra and export XRF maps for individual emission lines (e.g., Ca Kα). ImageJ was used to further render XRF maps. ImageJ was also used to measure the diameter of dense intracellular bodies using the full width half maximum distance from line profiles. All XRF maps are presented with linear-based intensity scales unless otherwise mentioned in the figure caption.

Scanning electron microscopy

Scanning electron microscopy (SEM) imaging was conducted with a Zeiss Sigma 300 VP. G. huxleyi was prepared for imaging according to the protocol mentioned above for nano-XRF except for the use of silicon as substrate. A thin layer of carbon (~10 nm) was deposited on the samples before loading samples into the instrument. Imaging was conducted using secondary electron detection mode under vacuum (1x10-6 bar) at 3 kV.

Results

Intact coccospheres and detached coccoliths of *Gephyrocapsa huxleyi* are shown in Figure 1a. From this scanning electron microscopy image, only coccoliths are discernible since the microalgal cell is fully enveloped by them. When *G. huxleyi* cells were examined by nanobeam-scanning X-ray fluorescence microscopy (nano-XRF) a combination of structural and elemental information can be collected. Figure 1b outlines a simplified experimental set-up for nano-XRF imaging. An incident X-ray beam $^{\sim}60$ nm in diameter is focused on a sample mounted in the normal position to the beam. The sample is raster-scanned in two dimensions using continuous on-the-fly scanning where data is collected at each recorded pixel or step size. X-ray fluorescence events from core-level excitation/de-excitation processes (e.g., Ca Kα fluorescence) are collected in the forward scattering direction. Deflected X-rays from elastic scattering are detected behind the sample to produce a phase contrast image. The total XRF spectrum from a coccolithophore cell is presented below in Figure 1b with all major elemental fluorescence contributions. While many elements are detected, some are from the materials used to construct the beamline (e.g., Cr, Al), elements found in the atmosphere (e.g., Ar), from sample substrate (e.g., Si, Cu) or from salt impurities (e.g., Cl). Nevertheless, the change of XRF intensity of each element across the sample reveals the elements that pertain to coccolithophores.

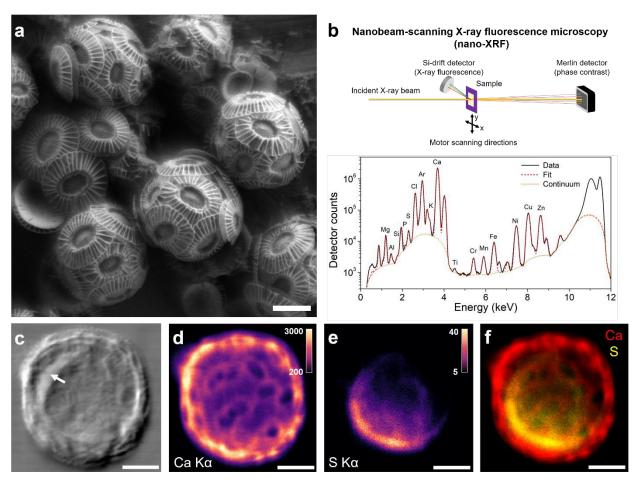


Figure 1. (a) Scanning electron microscopy image of *G. huxleyi*. (b) Experimental set-up for nanobeam-scanning X-ray fluorescence microscopy and phase contrast measurements with example of fitted X-ray fluorescence (XRF) spectrum of *G. huxleyi* below. Dried *G. huxleyi* cell with coccosphere shown with (c) X-ray phase contrast, (d) Ca Kα

Faraday Discussions Accepted Manuscript

XRF, (e) S $K\alpha$ XRF and (f) a composite XRF map of Ca and S XRF. The arrow in (c) points to high contrast material which is likely to represent the majority of the collapsed cell, which has shrunk upon drying. Intensity scale shown as XRF counts. Scale bar 2 μ m.

Figure 1c presents the X-ray phase contrast image of a single G. huxleyi cell. Elliptical structures resembling coccoliths within a circular ring representing the coccosphere are evident due to their high contrast, along with high contrast material within the coccosphere likely representing the dry algal cell. Note that the high energy X-ray beam penetrates the entire sample. For this reason, the phase contrast image is convoluted by overlapping coccoliths opposite to each other in the coccosphere and the dried microalgal cell. Figures 1d and 1e present Ca K α and S K α X-ray fluorescence (XRF) of the same cell as in Figure 1c, respectively. The XRF intensity of each pixel is presented on a continuous color scale, which was extrapolated from the fitted XRF peak for each element (see Figure 1b for example). The Ca K α XRF map shows coccoliths from opposite sides of the coccosphere as explained for the phase contrast image. The S K α XRF shows a weaker emission and distribution pattern that is distinct from that of calcium. As sulfur is present in proteins and also in certain lipids and polysaccharides, the S K α XRF signal provides a marker for cellular material that would otherwise give low contrast at hard X-ray energies. A composite image of these two XRF signals is shown in Figure 1f to distinguish the coccosphere outside and the cellular material inside.

When more cells were imaged additional structures were found. Figure 2 shows XRF maps of an aggregate of four coccolith-bearing G. huxleyi. Once more, Ca Kα XRF (Figure 2a) shows coccoliths and S Kα XRF (Figure 2b) the distribution of cellular material. For two of the four cells, XRF signals from P, K, Mn, Fe and Zn are detected (Figures 2c-g), pertaining mainly to small round bodies. The overlay of Mn, Fe and Zn XRF maps reveals significant co-localization (Figure 2h). This pattern aligns with the characteristics of acidocalcisomes in other organisms, which are known to store such ions together. 20 We therefore propose that the P, K, Mn, Fe, and Zn XRF signals originate from the same subcellular structures, and, due to their diverse metal ion content, we designate them as "ion-rich" bodies in this study. Nano-XRF cannot distinguish the valence of transition metal ion species, therefore the charge of each metal ion is not specifically indicated but is assumed to be divalent (e.g., Mn²⁺, Zn²⁺) or trivalent (Fe³⁺) under ambient conditions. A composite XRF image of S and a few of the detected metal ions (Figure 2h) shows that the location of these bodies coincides with the organic material of individual cells, which suggests them to be of cellular origin rather than extracellular contaminations. Previous cytological studies of G. huxleyi have not revealed bodies with metal ions such as Mn, Fe and Zn.11,13,25 Additional structures are observed in the K Kα XRF map (Figure 2d) that coincide with parts of the S signal, suggesting a cellular origin of the signal. Indeed, K is an important component of cells, used alongside other ions to regulate osmotic pressure and pH.^{26,27} The presence of Ca in the ion-rich bodies could not be determined for this sample due to the strong XRF intensities from calcite crystals in coccoliths that made extracting the weaker Ca XRF signal from other structures unreliable. This raised the question of whether Ca could also be a component of the detected ion-rich bodies.

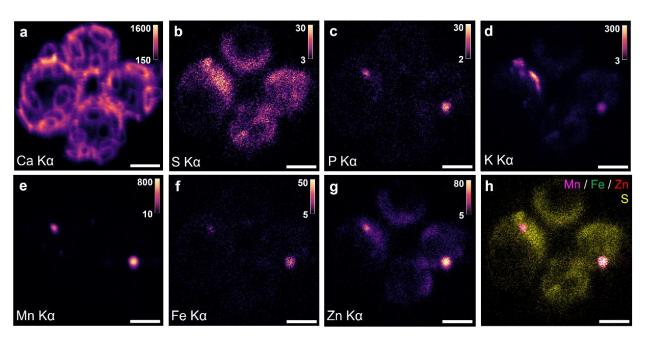


Figure 2. Nano-XRF analysis of dried, coccolith-bearing *G. huxleyi* showing (a) Ca Kα XRF, (b) S Kα XRF, (c) P Kα XRF, (d) K Kα XRF, (e) Mn Kα XRF, (f) Fe Kα XRF, and (g) Zn Kα XRF maps. (h) Composite map of Mn, Fe, Zn, and S XRF signals. Intensity scale shown as XRF counts. Scale bar 2 μm.

To clarify this question, a chemical treatment with EDTA (see Materials and Methods) was performed to dissolve the coccoliths in the coccosphere. Figure 3 presents X-ray phase contrast and XRF maps for an aggregate of 10 coccosphere-free (i.e., naked) G. huxleyi cells. The S Kα XRF (Figure 3b) signal shows the position of the cells and coincides with the high density material in the phase contrast image (Figure S1 displays fitted XRF spectrum). In the Ca Kα XRF map, four ring-like structures are visible (Figure 3c), which look reminiscent to coccoliths. Coccoliths can escape EDTA dissolution only when being still within coccolith vesicle, as EDTA cannot pass intact membranes, and therefore these rings are presumably nascent coccoliths. The arrow in Figure 3c points to one ring structure that appears to be in the protococcolith stage, containing only small calcite nanocrystals, as shortly after nucleation phase and before the crystals start growing. Additionally, several small bodies are visible in the Ca Kα XRF map, all at locations suggesting them to be intracellular components. Examining P, K and Mn Kα XRF maps (Figure 3c-f), several small bodies are identified, being present at the same position in all maps and also containing calcium as revealed by the Ca map. The average diameter of these ion-rich bodies was measured to be 550 +/- 150 nm from nano-XRF data (see Figure 5 for example of line profile plots). This size corresponds with Ca-Prich bodies previously imaged with cryo-FIB-SEM for G. huxleyi. 11 This complex elemental composition of the Ca-P-rich bodies was not recognized before with other techniques, demonstrating the potential of nano-XRF in advancing our knowledge on ion-storage pathways and their involvement in calcite formation in coccolithophores. The K Kα XRF map (Figure 3e) shows strong signal intensity over the area at which the cells are located, as concluded from co-localization with the S XRF signal, and from ion-rich bodies. Fe is weakly detected in the sample and only slightly concentrated in ion-rich bodies compared to the XRF intensity of cell material (Figure S2).

View Article Online DOI: 10.1039/D5FD00021A

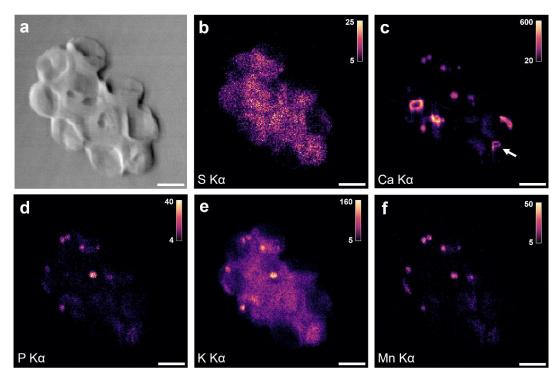


Figure 3. A cluster of dried EDTA-treated G. huxleyi cells shown with (a) X-ray phase contrast imaging, (b) S Kα XRF, (c) Ca Kα XRF, (d) P Kα XRF, (e) K Kα XRF, and (f) Mn Kα XRF. Intensity scale shown as XRF counts. A logarithmic-base intensity scale is used for (c). Arrow in (c) points to proto-coccolith. Scale bar 3 µm.

Previous work hypothesized that Ca- and P-rich bodies in coccolithophores assist in the delivery and/or storage of calcium ions for calcification in the coccolith vesicle. 10,11 How these bodies change in composition due to external factors or during calcite formation in the coccolith vesicle is left to be investigated. Nano-XRF could serve as a tool in this regard to probe such intracellular transformations. To investigate this, EDTA-decalcified G. huxleyi cells were returned to standard medium and growth conditions, then left to start rebuilding their coccolith shell for a few hours (see Materials and Methods). Figure 4 shows G. huxleyi cells after 4 h of calcification. The X-ray phase contrast image shows two intact cells and a third likely damaged cell, which is concluded from its horseshoe-like shape (top left of Figure 4a). In between these three cells there appears to be intracellular material that likely leaked out from the damaged cell, as evidenced by the S XRF signal (Figure 4b). In Figure 4c (Ca K α XRF), coccolith structures are seen in various states of completion and orientation; most of them seem fully formed, due to the presence of shield elements (see larger arrow in Figure 4c, coccolith from side view), while one lacks shield elements and thus must be an early proto-coccolith stage coccolith (see smaller arrow). Three intracellular bodies rich in Ca are visible in Figure 4c. For P, K and Mn (Figure 4d-f), higher XRF intensities can be found in these three Ca-rich bodies but also in two others. These five ion-rich bodies are numbered in Figure 4d. Similar to the XRF signals from ion-rich bodies in Figure 3, the relative amount of P, K and Mn appears to differ from one body to another. Additional unidentifiable structures are found at the bottom of the map for Ca, P, K and Mn XRF signals (see asterisks in Figure 4c). No appreciable Fe XRF signal pertaining to coccolithophore cells or ion-rich bodies was found.

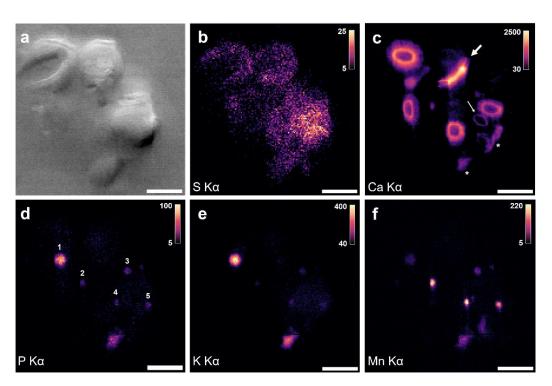


Figure 4. A collection of dry *G. huxleyi* cells decalcifying and allowed to recalcify for 4 h shown with (a) X-ray phase contrast imaging, (b) S K α XRF, (c) Ca K α XRF (top arrow, coccolith on its side; bottom arrow, proto-coccolith), (d) P K α XRF (ion-rich bodies are numbered), (e) K K α XRF and (f) Mn K α XRF. Intensity scale shown as XRF counts. A logarithmic-base intensity scale is used for (c). Scale bar 3 μ m.

To survey the change of ion-rich body elemental composition, select bodies were analyzed from EDTA-treated and calcifying samples (data from Figures 3 and 4). Bodies with minimal interference from the Ca XRF signal of coccoliths were selected. Figure 5 displays composite images of these samples with Ca Kα XRF in red and K Kα XRF in yellow, with inspected bodies numbered. XRF intensities are extracted via line profiles associated with each numbered body (see horizontal dotted lines in Figure 5a and b) and are plotted beside each composite map. Although a quantitative analysis of elemental amounts cannot be made here, since a calibration standard was not measured together with the samples, the relative XRF intensities between elements can be compared because the samples were measured under identical experimental conditions (e.g., the same incident energy, sample substrate, exposure time and step size). Of the mapped elements, Ca, K, Mn and P are consistently found at higher concentrations in ion-rich bodies compared to the surrounding cell material. Fe and S XRF intensities do not vary significantly across ion-rich bodies.

The relative composition between Ca, K, Mn and P elements varies from body to body. This is inferred by comparing the maximum XRF intensity for each element across the line profiles in Figure 5. Therefore, the proportionality of XRF intensities is examined for each body. For EDTA-decalcified *G. huxleyi* (Figure 5a), all bodies are rich in Ca and K except for one body having lower amount of Ca (body 2). Mn and P quantities relative to K appear to be similar for this sample. For early stage calcifying cells (Figure 5b), two out of three bodies are rich in Ca and K except for body 2, which has a lower K XRF intensity compared to Ca. In this body, the relative quantity of Mn is higher compared to all other bodies. The relative intensities of P and Ca XRF appear to be the most conserved across the investigated bodies. From this comparison (Figure 5a and b), the relative amount of K appears to decrease for ion-rich bodies in the early stage calcifying

DOI: 10.1039/D5FD00021A

This article is licensed under a Creative Commons Attribution 3.0 Unported Licence Open Access Article. Published on 05 2025. Downloaded on 2025/5/28 11:56:14.

cells. A conservative conclusion is that bodies are generally rich in Ca, K, Mn and P but that changes in elemental stoichiometry are possible. Whether this is due to body size, the biological state of the coccolithophore before sample preparation, biological heterogeneity or an effect of preparation is to be clarified.

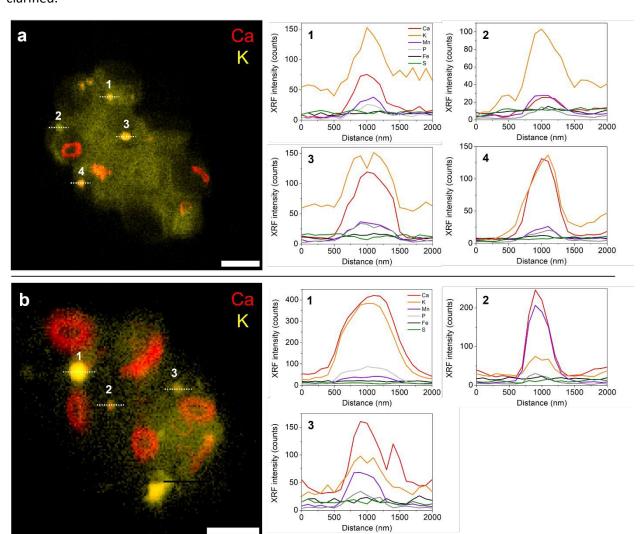


Figure 5. Composite maps of Ca and K XRF for (a) EDTA-decalcified G. huxleyi and (b) early-stage recalcifying G. huxleyi. Inspected ion-rich bodies are numbered along with a horizontal dotted line that indicates the pixels used to create XRF intensity line profile plots found beside each composite map. Scale bar 3 µm.

Liquid cell X-ray microscopy is an underdeveloped experimental approach when compared to liquid cell transmission electron microscopy.²⁸ Nevertheless, one advantage for hard X-ray nanoprobe beamlines is that measurements can be performed under ambient conditions with a relatively unencumbered sample environment. This means custom sample environments and liquid cell devices can be implemented at the beamline for in situ measurements.^{29,30} The advantages of measuring coccolithophores directly in liquid (i.e., native-state) would avoid uncertainties regarding effects of sample preparation. Furthermore, it would allow for the intracellular composition to be directly measured when coccolithophores are exposed to controlled media conditions in the liquid cell environment. Figure 6 presents the preparation of a liquid cell device for nano-XRF measurement with silicon nitride membrane (SiN) materials (full details in

Materials and Methods). Importantly, one of the SiN materials used to prepare the liquid cell has a spacer height of 5 μ m, which is similar to the typical diameter of a *G. huxleyi* cell. This ensured minimal cell sedimentation when the liquid cell device is mounted for measurement and minimizes XRF signal from the medium. Coccolithophores are first deposited on a SiN material (containing the spacer) placed horizontally and positioned in the liquid cell device holder (Figure 6a). A second SiN material is then placed on top and aligned so that the two window regions overlap for light transmission (Figure 6b). A faceplate cover is then secured and the assembly is turned upright (90°) (Figure 6c). The liquid cell device is further installed in the beamline (Figures 6d).

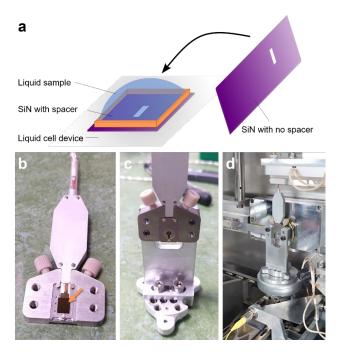


Figure 6. Sample preparation for liquid cell nano-XRF analysis. (a) Schematic illustrating the addition of liquid sample to SiN material situated in liquid cell device holder and a second SiN material being placed on top afterwards. (b) Photo of SiN materials (arrow) assembled in liquid cell sample holder without faceplate cover. (c) Liquid cell holder with faceplate secured and mounted on stage holder. (d) Liquid cell device mounted for measurement at the I14 hard X-ray nanoprobe beamline.

Figure 7a-d presents X-ray phase contrast image and XRF maps for a single coccosphere-bearing *G. huxleyi* cell in sea water medium at low scanning resolution. Phase contrast (Figure 7a) and Ca Kα XRF (Figure 7b) reveal the coccosphere along with detached coccoliths. Due to the high concentration of chloride salts in the sea water medium, the Cl Kα XRF signal can be used to ensure the liquid cell is retaining the medium by inspecting the background signal (Figure 7c). The gradient of Cl XRF signal in the medium is likely due to the bulging of SiN window region under positive pressure from the sealed liquid cell device. Importantly, the Cl XRF signal can also be used to determine the location of hydrated coccolithophore cells since they displace the sea water medium and exhibit a lower Cl concentration in the cytosol than has the medium (Figure 7c). The P Kα XRF signal (Figure 7d) shows low intensity in the medium but reveals an intracellular body within the coccolithophore cell with higher signal intensity. Figure S3 shows K and Mn XRF correspond with the intracellular body and S XRF more with the coccolithophore cell. When the same coccolithophore was measured again at higher scanning resolution (Figures 7e-h), phase contrast and Ca Kα XRF images show a similar level of detail for coccolith structures as found for samples measured in the

DOI: 10.1039/D5FD00021A

dried condition (e.g., Figure 1c and d). The intracellular body is further revealed by the P K α XRF signal (Figure 7h). This first liquid cell nano-XRF analysis demonstrates the recovery of ample XRF signal from both coccolith and intracellular body structures in saline medium. It shows that the stability of the liquid cell device on a nano-scanning sample stage can yield datasets of similar quality to those obtained from dried samples.

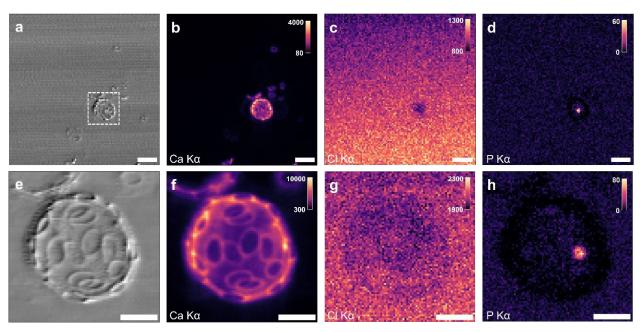


Figure 7. A coccosphere-bearing *G. huxleyi* cell in sea water medium measured at 300 nm scanning resolution with (a) X-ray phase contrast imaging, (b) Ca K α XRF, (c) Cl K α XRF, and (d) P K α XRF. The same coccolithophore cell measured with 100 nm scanning resolution shown with (e) X-ray phase contrast imaging, (f) Ca K α XRF, (g) Cl K α XRF and (h) P K α XRF. Intensity scale shown as XRF counts. Scale bars (a-d) 5 μ m and (e-h) 2 μ m.

To further examine ion-rich bodies of *G. huxleyi* in the native-state, EDTA-decalcified *G. huxleyi* cells were prepared for liquid cell nano-XRF imaging. *G. huxleyi* cells loaded in the liquid cell device were first confirmed to be in the measurement region by optical microscopy (Figure S4a). Once mounted for X-ray measurement, a low resolution XRF map with fast exposure time (500 nm step, 15 ms exposure time) was collected to confirm individual cell locations (Figure S4b and c). Figure 8 presents X-ray phase contrast image and XRF maps for several *G. huxleyi* cells at higher resolution (Figure S5 shows the fitted XRF spectrum). The phase contrast image in Figure 8a shows both intracellular coccoliths and dense bodies, but not the cell bodies as when imaging dry cells. For naked, hydrated coccolithophore cells, the cell body is not apparent likely due to the similar density of the cytosol and the medium. The Cl K α XRF map (Figure 8b), however, shows the location of individual cells by the displacement of the sea water medium. Due to the smaller size of decalcified cells (~3 μ m) and the larger spacer height (5 μ m), there appears to be some overlap between cells, which is also evident from the optical micrograph (Figure S4a).

The Ca K α XRF map (Figure 8c) shows several nascent coccoliths and Ca-rich bodies in the measured region. Some coccolith structures have lower contrast in the phase gradient image, possibly due to the smaller size of the proto-coccoliths compared to mature coccoliths, orientation to the X-ray beam, or the additional background contribution from the liquid cell. The P K α XRF map shows higher concentrations in six features, with five similar in appearance to ion-rich bodies and the other more dilute and elongated.

There is a slightly higher concentration of K in some features (Figure 8e), although the contrast is weak due to the high background XRF signal from the medium. Features observed in the Mn K α XRF map align with those in the P K α XRF map from intracellular bodies (see line profiles), as do Zn XRF signals (Figure S6). Fe was also detected, but not in bodies containing Mn, Zn, and P; instead, it appeared as irregular structures. An overlay of Fe and Cl XRF maps suggests this Fe-rich material to be extracellular (Figure S6), indicating that they may have been introduced during sample preparation. The Zn signal coincides with the ion-rich bodies. Line profiles for three intracellular bodies were made to extract the XRF intensity for the fitted elements (shown in Figure 8f, with associated XRF intensity plots presented on the right). For this dataset the relative Ca, P and Mn quantities appear to be similar across the three bodies. Body 1 shows a higher concentration of K compared to the background whereas bodies 2 and 3 do not show higher K content.

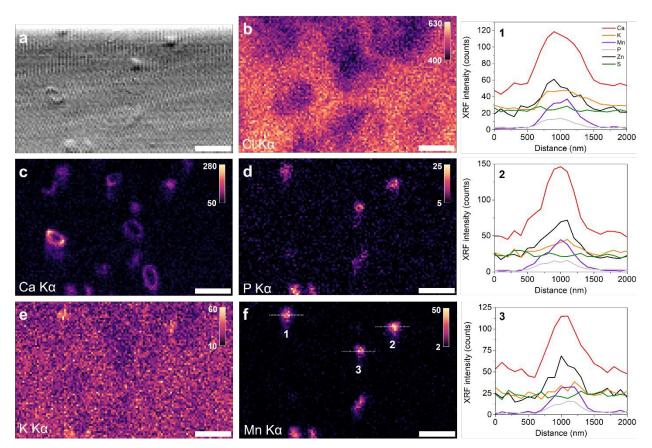


Figure 8. A collection of EDTA-treated *G. huxleyi* cells measured in sea water medium at 100 nm scanning resolution with (a) X-ray phase contrast imaging (top of image cropped because of artifact), (b) CI K α XRF, (c) Ca K α XRF, (d) P K α XRF, (e) K K α XRF, and (f) Mn K α XRF. Intensity scale shown as XRF counts. Line profiles and numbered intracellular bodies are shown in (f) with corresponding XRF intensity plots in the right-side of the figure. Figure S6 shows S K α XRF map. Scale bar 2 μ m.

Faraday Discussions Accepted Manuscript

Discussion

While nano-XRF has been used to investigate the elemental composition of coccoliths, 31,32 it is just beginning to be used to measure entire coccolithophore cells.33 Compared to electron-based spectromicroscopy techniques, the hard X-ray incident beam can penetrate entire coccolithophore cells, circumventing sample thickness limitations and the need for thin-sectioning. The elemental composition of dense intracellular bodies in G. huxleyi was further revealed in this work thanks to the large range of detectable biologically-relevant elements. Notably, the incident hard X-ray energy provided evidence of high Mn content for several of the imaged ion-rich bodies. We note that Mn was in fact detected in Carich bodies of G. huxleyi by Gal et al. using scanning transmission electron microscopy with energy dispersive X-ray spectroscopy (STEM-EDX) but was not discussed in their results. 10 Considerable Fe content was found in ion-rich bodies for dried coccosphere-bearing G. huxleyi (Figure 2f) but was mostly absent in the other samples. At incident X-ray energies >10 keV, Zn was detected in the ion-rich bodies (Figures 2g and S6b). The modest Zn K\alpha XRF counts in these maps are due to the high efficiency of core-level electron excitation (i.e., small difference between incident X-ray energy and Zn K-edge absorption energy). Therefore, the quantity of Zn in these structures is likely low. Careful XRF measurements with a calibrated standard (e.g., AXO standard material) will reveal more on the quantitative amounts of these metal species in ion-rich bodies.^{33,34} The presence of several metal ions in dense intracellular bodies for *G. huxleyi* strongly suggests these bodies to be involved in biological processes besides calcification (e.g., metal ion and phosphate storage, osmoregulation, pH homeostasis). In fact, the presence of Fe and Zn in addition to Ca, P and K shows a stronger resemblance to acidocalcisome composition observed in other microorganisms. 15,20,35

Nano-XRF was able to detect intracellular coccoliths at early proto-coccolith stage and ion-rich bodies for G. huxleyi in both dried and liquid condition. For the most part, one ion-rich body was observed per cell. For decalcified G. huxleyi imaged dried (Figure 3), however, more than one ion-rich body per cell was observed. Previous studies of G. huxleyi have identified the presence of dilute amounts of Ca and P in contact with the more dense ion-rich body.11 and a second, presumably polyphosphate-rich body.16 Nevertheless, the drying effects from sample preparation complicate this observation in the presented data. For liquid cell imaging of decalcified cells (Figure 8), the lower exposure time makes it challenging to detect these additional features. To avoid overlapping cells in the scanning beam direction, which could help distinguish structures from individual cells, the spacer material could be reduced for experiments with G. huxleyi (e.g., 3 μm) to improve the monolayer arrangement of cells. Cryogenic nano-XRF imaging of ice-vitrified G. huxleyi should be pursued to determine whether these additional features are detectable by enhancing weaker XRF signals with longer exposure times, increasing scanning resolution and measuring under vacuum conditions. Applying ptychography-based approaches with nano-XRF would also improve the spatial resolution to better distinguish intracellular structures.³⁶ Using nano-XRF to characterize the intracellular elemental composition of larger coccolithophore species such as C. carterae, which have several ion-rich bodies per cell and produce more than one coccolith at a time,³⁷ would offer a chance to capture the more complex organization of these structures during coccolith formation.³³

This work demonstrates that nano-XRF is capable of imaging microalgae in native-state using a liquid cell device, which can avoid potential artifacts from dried or fixed samples. The relative XRF signals of Ca, P and Mn for ion-rich bodies were more consistent under hydrated conditions compared to dried condition, for example. Another advantage for the hydrated condition was that X-ray phase contrast imaging produced images where nascent coccoliths and intracellular bodies were discernible (comparing Figure 3a

and 4a with Figure 8a). One limitation for liquid cell imaging is the salt water medium typically used for coccolithophores. The high concentration of K in the medium will make it challenging to follow intracellular composition changes of K under different conditions.

Intracellular bodies and nascent coccoliths could be imaged for *G. huxleyi* in liquid using relatively short exposure times (e.g., 20 ms per pixel used in Figure 8) to minimize X-ray damage. While the XRF signal is weaker compared to dry measurements because of faster scanning, the relevant elements for a single coccolithophore can be detected with 100 nm spatial resolution in less than one minute (i.e., 5000 x 5000 nm² with 100 nm step size and 20 ms exposure time). To increase XRF signal counts but maintain total exposure time or to reduce total scanning time, a sparse-scanning approach could be implemented.³⁸

Although measuring directly in liquid opens up possibilities for in situ experimentation with living cells, the effects of X-ray exposure from a typical nano-XRF measurement need to be assessed and, most likely, reduced. For example, when the coccosphere-bearing cell in Figure 7 was imaged again, but at higher resolution and longer exposure time (e.g., 50 ms), K and Mn XRF signals were no longer associated with the ion-rich body (Figure S3). In response, more conservative imaging parameters were used when EDTAtreated G. huxleyi cells were imaged (i.e., lower scanning resolution and shorter exposure time). Therefore, the total exposure time should be managed carefully to avoid changes to the sample. Brighter fourth generation X-ray sources and faster readouts of future detector technology should decrease scanning times for nano-XRF measurements with similar or better XRF detection levels. This will reduce the amount of time the X-ray beam impinges on the sample, which could help preserve cell viability within the liquid cell device. With faster data collection a higher number of cells can be measured per condition, improving statistical and biological significance. Optimization of the liquid cell device design and construction should be pursued to provide media flow to the measurement region, reduce reactive oxygen species generation and isolate different measurement regions, thereby safeguarding the viability of living cells in the liquid device and further advancing the in situ capabilities of liquid cell nano-XRF imaging. The choice of incident X-ray beam energy and resultant energy of photoelectrons that dissipate through the sample could also be considered to lower beam-related damage to biological material.³⁹

A limitation of liquid cell nano-XRF imaging of coccolithophores, and most microalgae, is the representation of a 3D object in 2D datasets. The orientation of the coccolithophore and the organization of its intracellular structures can complicate the interpretation of the generated image. For example, the Ca XRF signal from an intracellular coccolith can overlap with an intracellular body. Tilt-series image collection could provide additional perspectives to help build a 3D understanding of the imaged object. However, the time for collecting these additional nano-XRF datasets and the limited number of cells in the central axis of rotation are setbacks. Developments in new X-ray imaging modalities could offer alternative strategies for collecting 3D information. An example would be multi-slice ptychographic tomography, where the depth of field is moved through the sample (i.e., in the z-direction or X-ray beam direction) and no sample rotation is needed.

As mentioned above, there are improvements to be made to reduce beam dosage before *in situ* experiments can be conducted. Nevertheless, this work demonstrates that imaging can be conducted for hydrated, living microorganisms in high salinity conditions (i.e., strong background signal, complex media) to retrieve elemental and physical information at the single-cell level. Hard X-ray nanoprobe techniques are particularly advantageous for measuring biomineralizing microorganisms in liquid in this regard as many relevant elements and heavier metal ion species from the mineral and mineral precursors can be

DOI: 10.1039/D5FD00021A

detected via XRF. X-ray phase contrast imaging can simultaneously add structural information from the biomineral and dense intracellular bodies to the collected nano-XRF datasets. The liquid cell design can be modified to accommodate smaller or larger objects by changing the spacer height of SiN materials. The window size could also be enlarged to provide more imaging regions of microorganisms.

Conclusion

Nano-XRF and complementary X-ray phase contrast imaging were used to investigate the elemental composition and structure of coccolithophore *G. huxleyi* at subcellular resolution. Coccolithophores imaged without exterior coccoliths enabled identification of intracellular nascent coccolith structures and dense ion-rich bodies. Nano-XRF analysis showed additional metal ions (Mn, Fe and Zn) are present in ion-rich bodies, which could put the function of these bodies in a larger context beyond biomineralization. Harnessing the high energy X-ray beam, whole *G. huxleyi* cells were directly imaged in sea water medium using a liquid cell device. The stability of the liquid cell device in the nanobeam-scanning sample stage was demonstrated, yielding elemental and structural information of coccolithophores similar in fidelity to dried samples. For single-cell level studies on coccolithophore biomineralization, nano-XRF provides a balance between elemental sensitivity and spatial-temporal resolution to image entire cells without lengthy sample preparation. Liquid cell nano-XRF imaging offers direct native-state information that could be extended to other biomineralizing microorganisms. Future work should assess the effects of beam damage and develop strategies to minimize them.

Data availability statement

The data supporting this article are within the manuscript and included as part of the Supplementary Information. Datasets for this article, including optical, electron and X-ray microscopy imaging, will also be available at [name of repository to be determined] at [URL – format https://doi.org/DOI to be determined].

Conflicts of interest

There are no conflicts of interest to declare.

Acknowledgements

We would like to acknowledge the staff at the I14 hard X-ray nanoprobe of Diamond Light Source for technical support and helpful scientific discussions. We acknowledge awarded synchrotron beamtimes at I14 of the Diamond Light Source (Standard proposal MG23602 and Long term proposal MG28688). The authors would like to thank the Laboratoire de Maîtrise de la Contamination, de la Chimie des Caloporteurs et du Tritium (LMCT) at CEA Cadarache (DES/IRESNE) for scanning electron microscopy access. This work was supported by the French National Research Agency (microCOCCO: ANR-21-CE42-0022). This project has received financial support from the CNRS through the MITI interdisciplinary programs (PhysChemCocco).

References

- 1 J. Young, M. Geisen, L. Cros, A. Kleijne, C. Sprengel, I. Probert and J. Østergaard, *Journal of Nannoplankton Research*, 2003, **Special Issue**, 1–125.
- 2 J. D. Milliman, Global Biogeochemical Cycles, 1993, 7, 927–957.
- 3 A. R. Taylor, C. Brownlee and G. Wheeler, Annual Review of Marine Science, 2017, 9, 283–310.
- 4 L. Beaufort, I. Probert, E. M. Bendif, N. Metzl, C. Goyet, N. Buchet, P. Coupel, M. Grelaud, B. Rost, R. E. M. Rickaby and C. D. Vargas, *Nature*, 2011, **476**, 80–83.
- 5 S. A. O. Dea, S. J. Gibbs, P. R. Bown, J. R. Young, A. J. Poulton, C. Newsam and P. A. Wilson, *Nature Communications*, 2014, **5**, 5363.
- 6 J. Meyer and U. Riebesell, *Biogeosciences*, 2015, **12**, 1671–1682.
- 7 Y. Kadan, F. Tollervey, N. Varsano, J. Mahamid and A. Gal, *Proceedings of the National Academy of Sciences*, 2021, **118**, e2025670118.
- 8 E. M. Avrahami, Z. Eyal, N. Varsano, I. Zagoriy, J. Mahamid and A. Gal, *Advanced Materials*, 2024, **36**, 2309547.
- 9 K. Kahil, S. Weiner, L. Addadi and A. Gal, *Journal of the American Chemical Society*, 2021, **143**, 21100–21112.
- 10 A. Gal, S. Sviben, R. Wirth, A. Schreiber, B. Lassalle-Kaiser, D. Faivre, A. Scheffel, A. Gal, S. Sviben, A. Scheffel, D. Faivre, R. Wirth, A. Schreiber and B. Lassalle-Kaiser, *Advanced Science*, 2017, **4**, 1700088.
- 11 S. Sviben, A. Gal, M. A. Hood, L. Bertinetti, Y. Politi, M. Bennet, P. Krishnamoorthy, A. Schertel, R. Wirth, A. Sorrentino, E. Pereiro, D. Faivre and A. Scheffel, *Nature Communications*, 2016, **7**, 11228.
- 12 M. E. Marsh, Protoplasma, 1996, 190, 181-188.
- 13 A. Gal, A. Sorrentino, K. Kahil, E. Pereiro, D. Faivre and A. Scheffel, *Proceedings of the National Academy of Sciences*, 2018, **115**, 11000–11005.
- 14 M. E. Marsh, D. K. Chang and G. C. King, Journal of Biological Chemistry, 1992, 267, 20507–20512.
- 15 R. Docampo, W. de Souza, K. Miranda, P. Rohloff and S. N. J. Moreno, *Nature Reviews Microbiology*, 2005, **3**, 251–261.
- 16 H. Peled-Zehavi and A. Gal, Advanced Biology, 2021, 5, 2000296.
- 17 E. Fox, E. Meyer, N. Panasiak, A. R. Taylor and A. R. Taylor, Frontiers in Marine Science, 2018, 5, 326.
- 18 J. Deng, D. J. Vine, S. Chen, Y. S. G. Nashed, Q. Jin, N. W. Phillips, T. Peterka, R. Ross, S. Vogt and C. Jacobsen, *Proceedings of the National Academy of Sciences*, 2015, **112**, 2314–2319.
- 19 J. Deng, Y. H. Lo, M. Gallagher-Jones, S. Chen, A. Pryor, Q. Jin, Y. P. Hong, Y. S. G. Nashed, S. Vogt, J. Miao and C. Jacobsen, *Science Advances*, 2018, **4**, eaau4548.
- 20 S. Schmollinger, S. Chen, D. Strenkert, C. Hui, M. Ralle and S. S. Merchant, *Proceedings of the National Academy of Sciences*, 2021, **118**, e2026811118.
- 21 S. Schmollinger, S. Chen and S. S. Merchant, Metallomics, 2023, 15, mfad025.
- 22 C. Y. J. Hémonnot and S. Köster, ACS Nano, 2017, 11, 8542–8559.
- 23 P. D. Quinn, L. Alianelli, M. Gomez-Gonzalez, D. Mahoney, F. Cacho-Nerin, A. Peach and J. E. Parker, *Journal of Synchrotron Radiation*, 2021, **28**, 1–8.
- 24 P. D. Quinn, F. Cacho-Nerin, M. A. Gomez-Gonzalez, J. E. Parker, T. Poon and J. M. Walker, *J Synchrotron Rad*, 2023, **30**, 200–207.
- 25 Y. Kadan, L. Aram, E. Shimoni, S. Levin-Zaidman, S. Rosenwasser and A. Gal, *Journal of Structural Biology*, 2020, **210**, 107465.
- 26 R. A. Leigh and R. G. Wyn Jones, New Phytologist, 1984, 97, 1–13.
- 27 H.-H. Kunz, M. Gierth, A. Herdean, M. Satoh-Cruz, D. M. Kramer, C. Spetea and J. I. Schroeder, *Proceedings of the National Academy of Sciences*, 2014, **111**, 7480–7485.
- 28 H. G. Liao and H. Zheng, Annual Review of Physical Chemistry, 2016, 67, 719-747.

View Article Online DOI: 10.1039/D5FD00021A

- 29 D. M. Chevrier, E. Cerdá-Doñate, L. Gandarias, M. A. Gomez-Gonzalez, S. Swaraj, P. E. D. Soto-Rodriguez, A. Fraisse, T. Robinson and D. Faivre, ChemRxiv, 2024, preprint, DOI: 10.26434/chemrxiv-2024-cm6nd.
- 30 J. E. Parker, M. Gomez-Gonzalez, Y. Van Lishout, H. Islam, D. D. Martin, D. Ozkaya, P. D. Quinn and M. E. Schuster, J. Synchrotron Rad., 2022, 29, 431-438.
- 31 C. Bottini, M. Dapiaggi, E. Erba, G. Faucher and N. Rotiroti, Scientific Reports, 2020, 10, 9825.
- 32 J. M. Walker, H. J. M. Greene, Y. Moazzam, P. D. Quinn, J. E. Parker and G. Langer, Environmental Science: Processes & Impacts, 2024, 26, 966-974.
- 33 A. Triccas, D. M. Chevrier, M. Verezhak, J. Ihli, M. Guizar-Sicairos, M. Holler, A. Scheffel, N. Ozaki, V. Chamard, R. Wood, T. Grünewald and F. Nudelman, bioRxiv, 2024, preprint, DOI: 10.1101/2024.11.15.623571.
- 34 L. Lemelle, A. Simionovici, T. Schoonjans, R. Tucoulou, E. Enrico, M. Salomé, A. Hofmann and B. Cavalazzi, TrAC - Trends in Analytical Chemistry, 2017, 91, 104–111.
- 35 D. A. Scott, R. Docampo, J. A. Dvorak, S. Shi and R. D. Leapman, Journal of Biological Chemistry, 1997, **272**, 28020–28029.
- 36 J. Deng, D. J. Vine, S. Chen, Q. Jin, Y. S. G. Nashed, T. Peterka, S. Vogt and C. Jacobsen, Scientific Reports, 2017, 7, 445.
- 37 M. E. Marsh, Comparative Biochemistry and Physiology Part B: Biochemistry and Molecular Biology, 2003, 136, 743-754.
- 38 P. D. Quinn, M. Sabaté Landman, T. Davis, M. Freitag, S. Gazzola and S. Dolgov, Chemical & Biomedical Imaging, 2024, 2, 283-292.
- 39 K. Sauer, I. Zizak, J.-B. Forien, A. Rack, E. Scoppola and P. Zaslansky, Nature Communications, 2022, 13,
- 40 P. Li and A. Maiden, Sci Rep, 2018, 8, 2049.

View Article Online DOI: 10.1039/D5FD00021A

Faraday Discussions Accepted Manuscript

Open Access Article. Published on 05 2025. Downloaded on 2025/5/28 11:56:14.

This article is licensed under a Creative Commons Attribution 3.0 Unported Licence.

The data supporting this article are within the manuscript and included as part of the Supplementary Information. Datasets for this article, including optical, electron and X-ray microscopy imaging, will also be available at [name of repository] at [URL – format https://doi.org/DOI].